

Effect of ionic strength on shear-thinning nanoclay-polymer composite hydrogels

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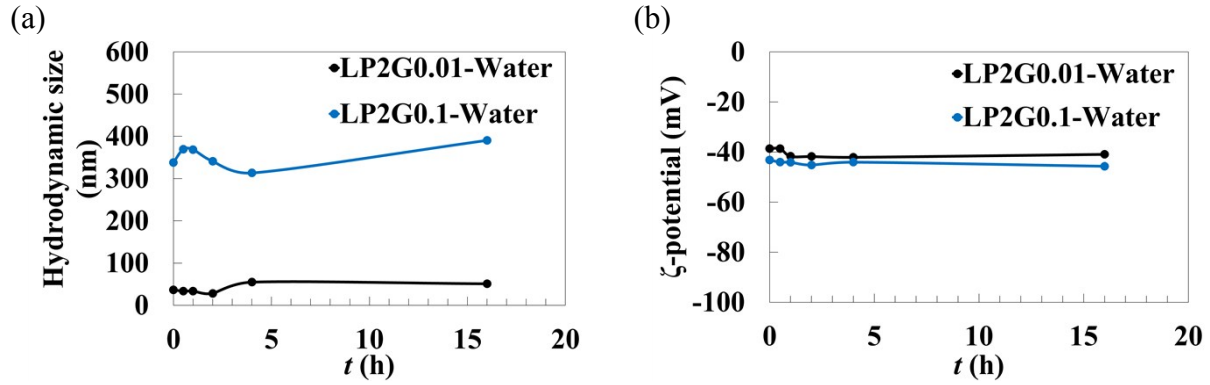


Figure S1. Dynamic light scattering (DLS) and electrokinetic analyses of LAPONITE®-gelatin (LP-G) suspensions, yielding hydrodynamic size (a) and ζ -potential (b). At a constant gelatin concentration, increasing LAPONITE® concentration (e.g., 2% compared to 1%, Figure 1c), increases the hydrodynamic size of LAPONITE®-gelatin particles as a result of polymer-mediated nanoplatelet bridging. The stable size of particles suggests that the gelatin-stabilized LAPONITE® clusters form a stable colloidal suspension.

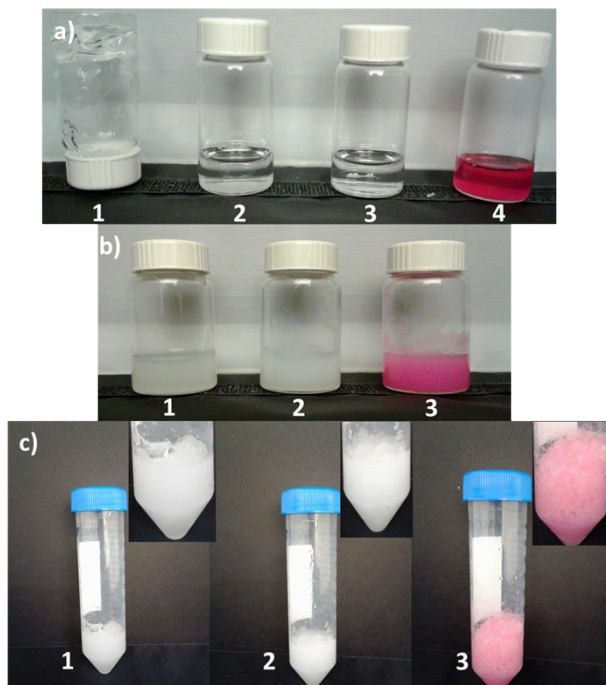


Figure S2. Images of LAPONITE®, gelatin, and different compositions of LAPONITE®-gelatin mixtures. **(a-1)** LAPONITE® (6 wt%) in Milli-Q water, **(a-2)** Gelatin (2 wt%) in Milli-Q water, **(a-3)** Gelatin (2 wt%) in PBS, **(a-4)** Gelatin (2 wt%) in DMEM. STBs prepared by mixing **(b-1)** LAPONITE® (6 wt%) in Milli-Q water and gelatin (2 wt%) in Milli-Q water, **(b-2)**: LAPONITE® (6 wt%) in Milli-Q water and gelatin (2 wt%) in PBS, and **(b-3)** LAPONITE® (6 wt%) in Milli-Q water and gelatin (2 wt%) in DMEM. STBs prepared by mixing **(c-1)** LAPONITE® (12 wt%) in Milli-Q water and gelatin (6 wt%) in Milli-Q water (1:1), **(c-2)** LAPONITE® (12 wt%) in Milli-Q water and gelatin (6 wt%) in PBS, and **(c-3)** LAPONITE® (12 wt%) in Milli-Q water and gelatin (6 wt%) in DMEM.

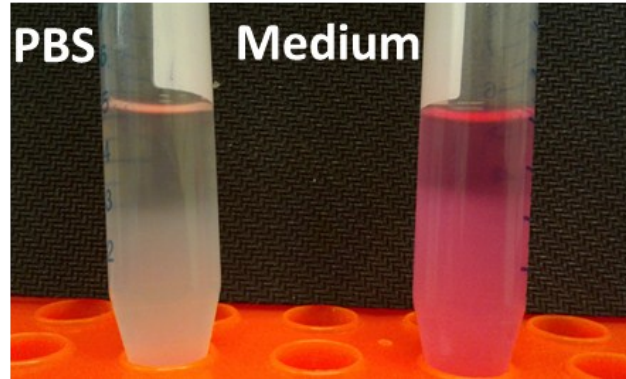


Figure S3. Images of nanocomposites prepared by mixing (1:1 v/v) water-dispersed LAPONITE® (2 wt%) with gelatin (0.1 wt% in PBS or DMEM). The dispersions undergo phase separation.

LAPONITE®, $T=37^{\circ}\text{C}$

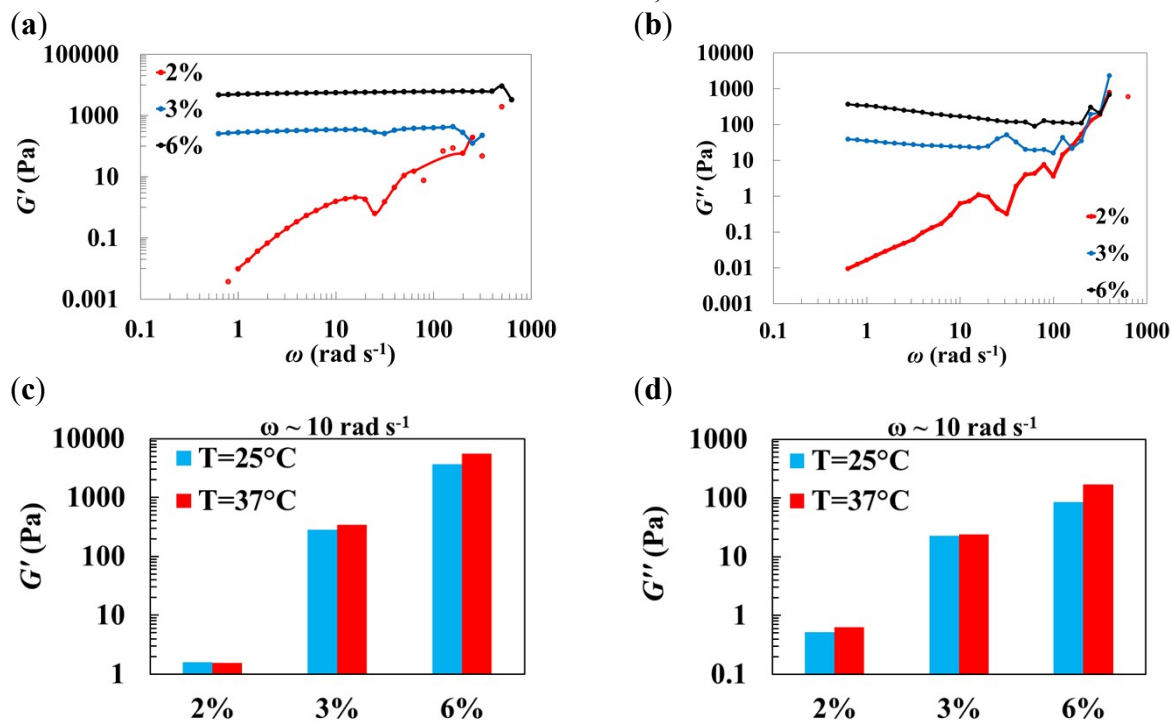


Figure S4. Rheological properties of LAPONITE® dispersions/gels at $T = 37^{\circ}\text{C}$, prepared in Milli-Q water. Storage moduli (a), and loss moduli (b) versus angular frequency. Increasing LAPONITE® concentration increases the viscoelastic moduli. Effect of temperature and LAPONITE® concentration on the storage (at $\omega \sim 10$ rad s^{-1} , c) and loss (at $\omega \sim 10$ rad s^{-1} , d) moduli of LAPONITE® dispersions/gels.

Gelatin 3%, $T=37^{\circ}\text{C}$

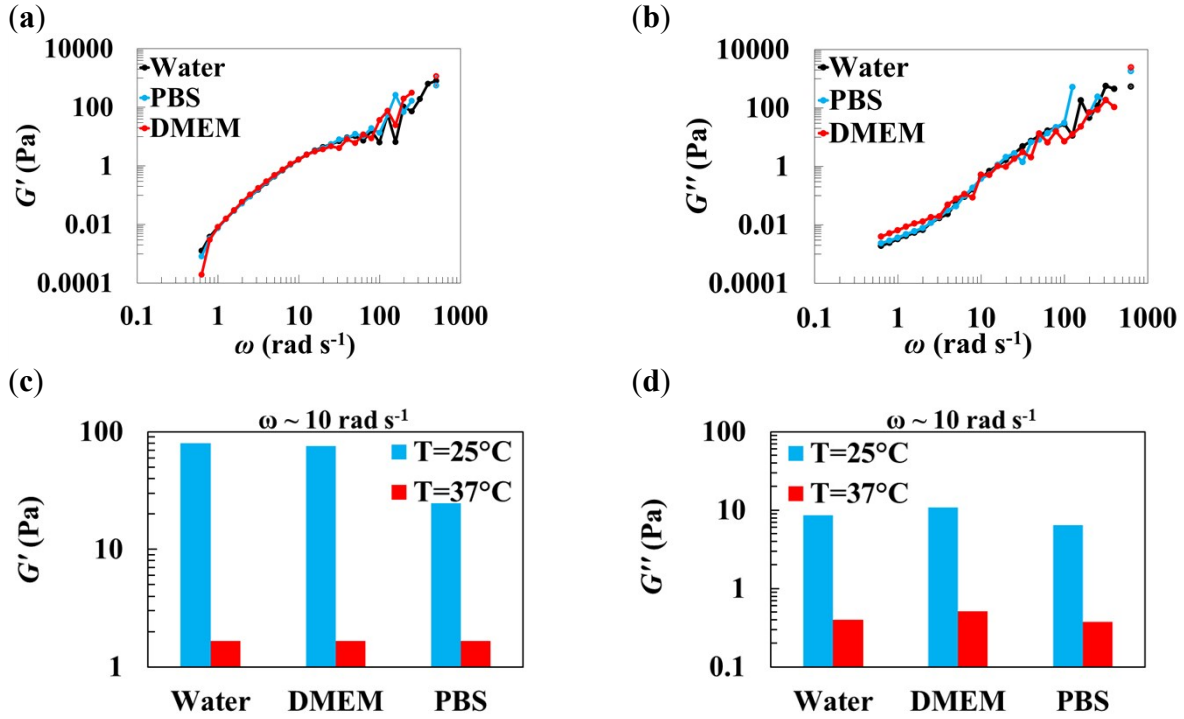


Figure S5. Rheological properties of gelatin solutions at $T = 37^{\circ}\text{C}$, prepared in Milli-Q water, PBS, and media. Storage moduli (a), and loss moduli (b) versus angular frequency. Effect of temperature and media on the storage moduli (at $\omega \sim 10 \text{ rad s}^{-1}$, c) and loss moduli (at $\omega \sim 10 \text{ rad s}^{-1}$, d) of gelatin solutions.

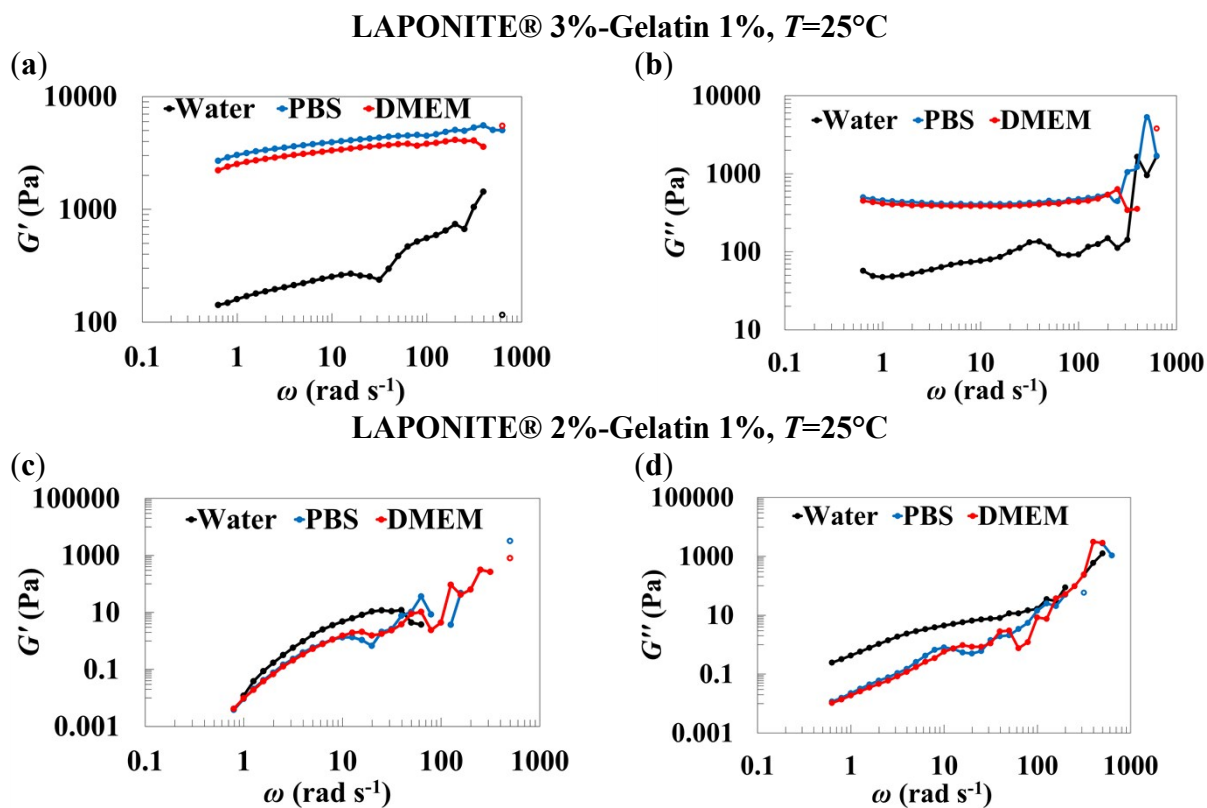


Figure S6. Rheological properties of LAPONITE®-gelatin (1%) shear-thinning hydrogels prepared by mixing exfoliated LAPONITE® with gelatin dissolved in Milli-Q water, PBS, and DMEM at 25°C . Storage (a,c) and loss moduli (b,d) versus angular frequency.

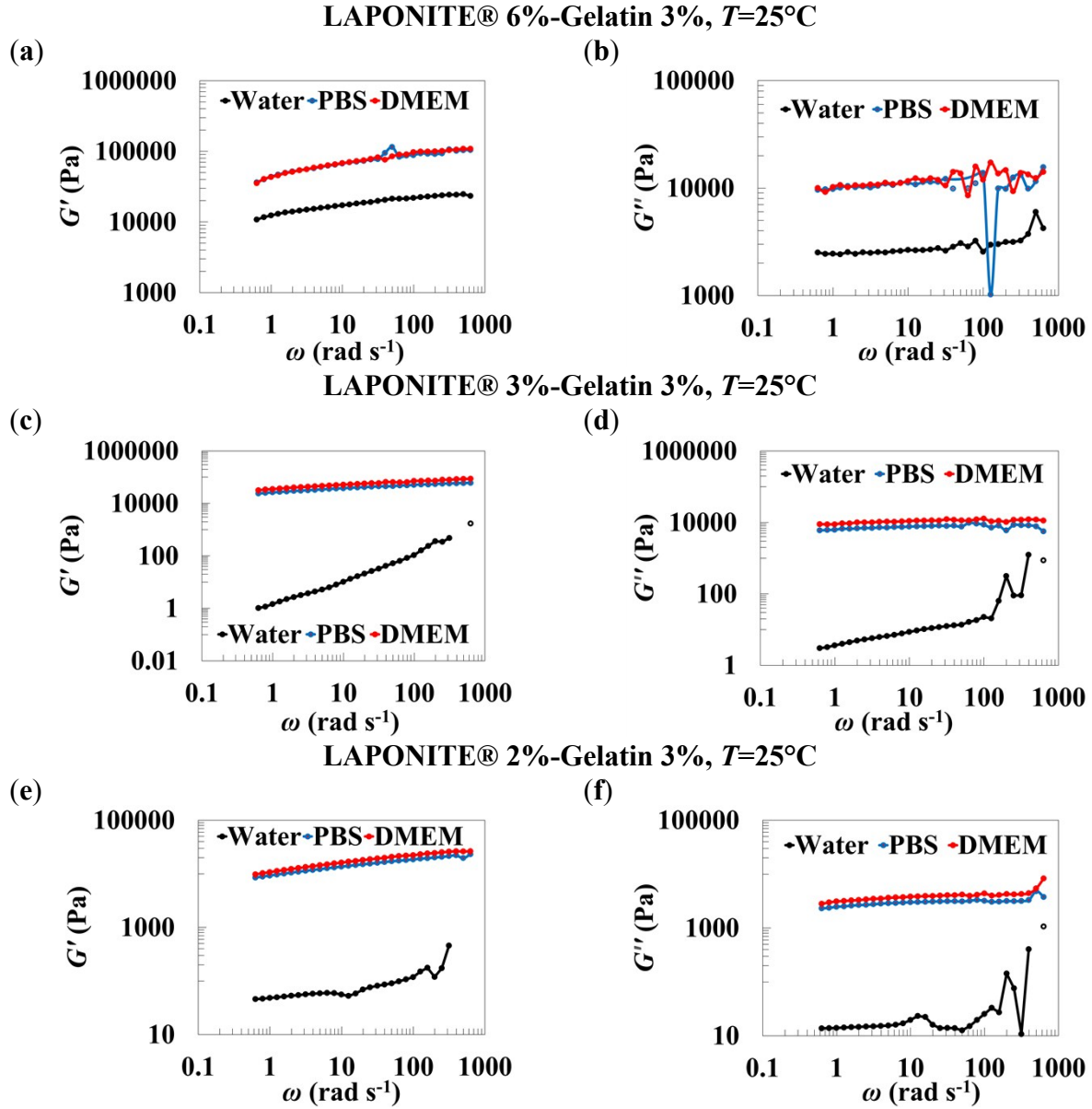


Figure S7. Rheological properties of LAPONITE®-gelatin (3%) shear-thinning hydrogels prepared by mixing exfoliated LAPONITE® with gelatin dissolved in Milli-Q water, PBS, and DMEM at 25°C . Storage (a,c,e) and loss moduli (b,d,f) versus angular frequency.

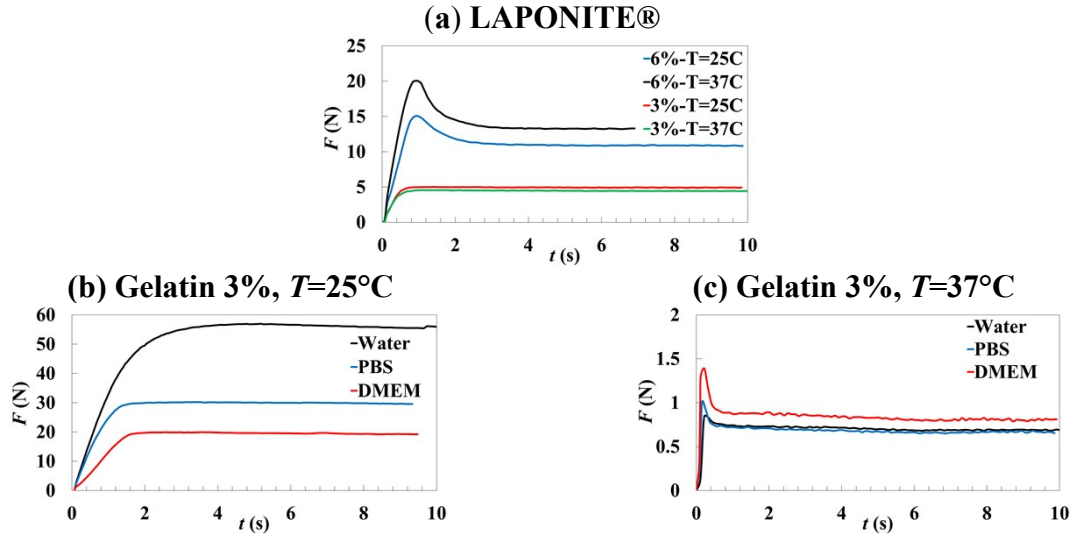


Figure S8. Dynamics of injection force through the 5F catheter for LAPONITE® (a) and gelatin (b,c). The higher the LAPONITE® concentration, the higher the injection force, regardless of temperature. Gelatin, a heat-sensitive biopolymer, undergoes phase transition at room temperature, resulting in an increase in the injection force.

***Movies** showing the injection of LAPONITE®-gelatin STBs prepared in water (**M1**), PBS (**M2**), and DMEM (**M3**) through a 3 mL Luer lock syringe, equipped with a 23 G needle (BD Precision Glide 23G×1” thin wall M, with $L = 38$ mm and $ID = 0.34$ mm). Phase separation was observed in PBS and DMEM.*