## Electronic Supplementary Information (ESI)

# A Lead-Free Perovskite-like Hybrid with Above-room-temperature Switching of Quadratic Nonlinear Optical Properties 

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## Experimental section <br> Synthesis

All reagents and solvents used in experiment were commercial products without further purification.
Compound 1 was easily synthesized by the reaction of hexamethyleneimine and antimony pentachloride with the molar ratio 2:1. Powder X-ray diffraction (PXRD) was performed on Miniflex 600 powder X-ray diffractometer.

## Thermal measurements

The differential scanning calorimetry (DSC) ( $10 \mathrm{~K} / \mathrm{min}, 5 \mathrm{~K} / \mathrm{min}$ and $2 \mathrm{~K} / \mathrm{min}$ ) data was recorded by Netzsch DSC 200 F3 calorimeter. And a Mettler Toledo TGA/SDTA 851e was used to thermogravimetric analysis data.

## Dielectric measurements

The complex permittivity of powder was performed on a Tonghui TH2828A analyzer in the temperature range of $310-349 \mathrm{~K}$. The complex dielectric permittivity $\varepsilon$ and $\varepsilon^{\prime \prime}\left(\varepsilon=\varepsilon^{\prime}-\mathrm{i} \varepsilon^{\prime \prime}\right)$ were measured in different frequency ( 500 kHz and 1000 kHz ).

## Linear and NLO properties

The UV/vis diffuse reflection spectrum of $\mathbf{1}$ was measured on Lambda 950 spectrometer with the range of 800-250 nm at 298 K . The FLS 920, Edinburgh Instruments was used to measure the variable-temperature NLO properties in the temperature range of 306 K to 348 K , by using an unexpanded laser beam with low divergence (pulsed by Nd:YAG laser at a wavelength of 1064 nm ).

## Single-crystal X-ray diffraction

Single-crystal X-ray were collected on a SuperNova diffractometer with Mo-K radiation ( $\lambda=$ $0.71073 \AA$ ) at different temperatures ( 290 K and 350 K ). All of crystal data were performed by the Crystalclear software package (Rigaku, 2005). The crystal structures were processed by the direct method and then refined by the full-matrix least-squares refinements on $F^{2}$, which use the

SHELXLTL software package. Non-H atoms were refined anisotropically according to all reflections with $I>2 \sigma(I)$ and all of hydrogen atoms were obtained geometrically. The supplementary crystallographic data for this paper were contained in CCDC (1588041, LTP) and CCDC (1588042, HTP).


Figure S1. Experimental and calculated powder X-ray diffraction patterns of $\mathbf{1 .}$


Figure S2. The TG curves of $\mathbf{1}$ with heating rate $10 \mathrm{~K} / \mathrm{min}$.


Figure S3. a) DSC curves of $\mathbf{1}$ measured at $10 \mathrm{~K} / \mathrm{min}$; b) DSC traces measured with the different heating/cooling rates ( 10,5 and $2 \mathrm{~K} / \mathrm{min}$ ).


Figure S4 The real part of the dielectric constant of $\mathbf{1}$ measured on 500 kHz and 1000 kHz . Inset: The corresponding imaginary part ( $\varepsilon^{\prime \prime}$ ).


Figure S5. UV-vis diffuse reflection spectrum of 1.


## Switching Cycles

Figure S6 Completely reversible and recoverable switching of NLO effects.


Figure S7. Symmetry transformation of $\mathbf{1}$ during the phase transition.

Table S1 Crystal structure and refinement detail of $\mathbf{1}$ at different temperatures.

| Empirical formula | $\left(\mathrm{C}_{6} \mathrm{H}_{14} \mathrm{~N}\right)_{2} \mathrm{SbCl}_{5}$ | $\left(\mathrm{C}_{6} \mathrm{H}_{14} \mathrm{~N}\right)_{2} \mathrm{SbCl}_{5}$ |
| :--- | :--- | :--- |
| Formula weight | 499.36 | 499.36 |
| Temperature/K | $350.00(10)$ | $290.00(10)$ |
| Crystal system | orthorhombic | orthorhombic |
| Space group | Pnma | $P 2_{1} 2_{1} 2_{1}$ |
| Unit cell dimensions $(\AA)$ | $13.0708(7)$ | $7.5861(3)$ |
|  | $7.9573(5)$ | $13.2521(5)$ |
|  | $20.1605(10)$ | $20.2623(8)$ |
| $V\left(\AA^{3}\right)$ | $2096.9(2)$ | $2037.00(14)$ |
| $Z, \rho_{\text {cal }}\left(\right.$ gcm $\left.^{-3}\right)$ | $4,1.582$ | $4,1.628$ |
| $\mathrm{~F}(000)$ | 1000.0 | 1000.0 |
| Theta range $\left({ }^{\circ}\right)$ | 6.818 to 58.986 | 6.772 to 58.8 |
| Limiting indices | $-9 \leq \mathrm{h} \leq 17$ | $-10 \leq \mathrm{h} \leq 10$ |
|  | $-8 \leq \mathrm{k} \leq 11$ | $-17 \leq \mathrm{k} \leq 13$ |
|  | $-22 \leq 1 \leq 26$ | $-19 \leq 1 \leq 25$ |
| Reflections collected/unique | $7734 / 2718\left(R_{\text {int }}=0.0303\right)$ | $8223 / 4546\left(R_{\text {int }}=0.0238\right)$ |
| Completeness $(\%)$ | 99.1 | 98.9 |
| Data/restraints/parameters | $2718 / 20 / 118$ | $4546 / 0 / 181$ |
| Goodness | 1.029 | 1.031 |
| Final $R$ indexes $[I>=2 \sigma(I)]$ | $R_{1}=0.0623, w R_{2}=0.1652$ | $R_{1}=0.0349, w R_{2}=0.0647$ |

Table S2 Bond Lengths for 1 at 290 K.

| Atom | Atom | Length $/ \AA$ | Atom | Atom | Length $/ \AA$ |
| :--- | :--- | :--- | :--- | :--- | :--- |
| $\mathrm{Sb}_{1}$ | $\mathrm{Cl}_{2}$ | $2.5298(15)$ | $\mathrm{C}_{7}$ | $\mathrm{C}_{8}$ | $1.486(10)$ |
| $\mathrm{Sb}_{1}$ | $\mathrm{Cl}_{5}$ | $2.7712(16)$ | $\mathrm{C}_{2}$ | $\mathrm{C}_{3}$ | $1.479(12)$ |
| $\mathrm{Sb}_{1}$ | $\mathrm{Cl}_{4}$ | $2.3845(15)$ | $\mathrm{C}_{2}$ | $\mathrm{C}_{1}$ | $1.468(11)$ |
| $\mathrm{Sb}_{1}$ | $\mathrm{Cl}_{1}{ }^{1}$ | $3.3523(16)$ | $\mathrm{C}_{3}$ | $\mathrm{C}_{4}$ | $1.484(11)$ |
| $\mathrm{Sb}_{1}$ | $\mathrm{Cl}_{1}$ | $2.6800(16)$ | $\mathrm{C}_{6}$ | $\mathrm{C}_{5}$ | $1.456(10)$ |
| $\mathrm{Sb}_{1}$ | $\mathrm{Cl}_{3}$ | $2.5884(16)$ | $\mathrm{C}_{8}$ | $\mathrm{C}_{9}$ | $1.435(11)$ |
| $\mathrm{N}_{2}$ | $\mathrm{C}_{7}$ | $1.483(8)$ | $\mathrm{C}_{4}$ | $\mathrm{C}_{5}$ | $1.473(10)$ |
| $\mathrm{N}_{2}$ | $\mathrm{C}_{12}$ | $1.495(8)$ | $\mathrm{C}_{9}$ | $\mathrm{C}_{10}$ | $1.442(13)$ |
| $\mathrm{N}_{1}$ | $\mathrm{C}_{6}$ | $1.472(10)$ | $\mathrm{C}_{10}$ | $\mathrm{C}_{11}$ | $1.335(13)$ |
| $\mathrm{N}_{1}$ | $\mathrm{C}_{1}$ | $1.419(9)$ | $\mathrm{C}_{11}$ | $\mathrm{C}_{12}$ | $1.382(13)$ |

${ }^{1}=1 / 2+\mathrm{X}, 1 / 2-\mathrm{Y}, 1-\mathrm{Z}$

## Computational methods

The dipole moments of constituent groups in the crystal were calculated with density functional theory (DFT) and the B3LYP extended ex-change functions employing the finite field (FF) method with the GAUSSIAN 09 electronic structure package. ${ }^{1}$ In order to take into account the relativistic effect, Sb atoms adopted the effective core potential (ECP) double- $\zeta$ (DZ) basic set of LanL2DZ, while the nonmetal elements took standard $6-311+\mathrm{g}(\mathrm{d})$ all-electron basis set. Average polarizability (a) and total intrinsic first hyperpolarizability $\left(\beta_{\mathrm{tot}}\right)$ can be estimated according to the following equations:

$$
\bar{\alpha}=\left(\alpha_{x x}+\alpha_{y y}+\alpha_{z z}\right) / 3
$$

and

$$
\beta_{t o t}=\left[\beta_{x}{ }^{2}+\beta_{y}{ }^{2}+\beta_{z}{ }^{2}\right]^{1 / 2}
$$

where

$$
\beta_{i}=\frac{1}{3} \sum_{j}\left[\beta_{i j j}+\beta_{j i j}+\beta_{j i j}\right] \quad i=x, y, z
$$

The electronic structure and optical properties calculations were performed by the plane-wave pseudopotential method implemented in the CASTEP software package. ${ }^{2,}{ }^{3}$ The exchangecorrelation potential was calculated using Perdew-Burke-Ernzerhof for solids (PBEsol) ${ }^{4}$ functional within the generalized gradient approximation (GGA) with the scheme. To achieve energy convergence, the kinetic energy cutoff of 600 eV within normal-conserving pseudopotential (NCP) ${ }^{5}$ was adopted. The following orbital electrons were treated as valence electrons: $\mathrm{Sn} 5 \mathrm{~s}^{2} 5 \mathrm{p}^{3} ; \mathrm{Cl} 3 \mathrm{~s}^{2}$ $3 p^{5}$; C $2 s^{2} 2 p^{2} ; \mathrm{N} 2 s^{2} 2 p^{3}$ and H $1 s^{1}$. To achieve the accurate density of the electronic states, the $k$ space integrations were done with Monkhorst-Pack grids with a $5 \times 3 \times 3$, including more than 50 empty bands in the optical properties calculations. The so-called length-gauge formalism ${ }^{6,7}$ derived by Aversa and Sipe was adopted to calculate NLO coefficients, which has been proved to be successful in calculating second-order susceptibility for NLO crystal.

Table S3 Calculated dipole moment ( $u_{\mathrm{g}}$, Debye), linear polarizability ( $\alpha_{\text {tot }}, \times 10^{-24} \mathrm{esu}$ ) and first hyperpolarizability $\left(\boldsymbol{\beta}_{\text {total }}, \times 10^{-30} \mathrm{esu}\right)$ values.

|  | $\mathrm{C}_{6} \mathrm{H}_{14} \mathrm{~N}$ cation 1 | $\mathrm{C}_{6} \mathrm{H}_{14} \mathrm{~N}$ cation 2 | $\mathrm{SbCl}_{5}$ | $\left(\mathrm{C}_{6} \mathrm{H}_{14} \mathrm{~N}\right)_{2} \mathrm{SbCl}_{5}$ |
| :---: | :---: | :---: | :---: | :---: |
| $\mu_{x}$ | -1.196 | 0.655 | 14.284 | 2.798 |
| $\mu_{y}$ | -2.277 | -2.425 | 15.536 | 3.941 |
| $\mu_{z}$ | -3.929 | 3.953 | 13.002 | -28.669 |
| $u_{g}$ | 4.696 | 4.683 | 24.788 | 29.074 |
| $\alpha_{x x}$ | 7.546 | 7.792 | 59.626 | 78.641 |
| $\alpha_{x y}$ | 0.051 | -0.220 | -1.315 | 4.001 |
| $\alpha_{y y}$ | 9.159 | 9.101 | 61.549 | 72.642 |
| $\alpha_{x z}$ | 0.269 | 0.382 | -2.707 | -5.856 |
| $\alpha_{y z}$ | 0.004 | 0.134 | -4.724 | -4.483 |
| $\alpha_{z z}$ | 9.063 | 9.189 | 59.515 | 85.642 |
| $\alpha_{\text {tot }}$ | 8.590 | 8.694 | 60.230 | 78.975 |
| $\boldsymbol{\beta}_{x x x}$ | -0.017 | -0.033 | 14.439 | -0.607 |
| $\beta_{x x y}$ | -0.025 | -0.005 | 17.048 | -9.711 |
| $\boldsymbol{\beta}_{x y y}$ | 0.051 | -0.017 | 22.458 | -5.608 |
| $\beta_{y y y}$ | 0.273 | 0.207 | 20.472 | 11.687 |
| $\boldsymbol{\beta}_{z x x}$ | 0.081 | -0.136 | 16.137 | 14.645 |
| $\beta_{x y z}$ | -0.013 | -0.015 | 10.705 | -1.999 |
| $\boldsymbol{\beta}_{z y}{ }^{\text {l }}$ | 0.147 | -0.087 | 21.000 | 9.239 |
| $\boldsymbol{\beta}_{x z z}$ | 0.033 | -0.023 | 24.689 | 9.412 |
| $\boldsymbol{\beta}_{y z z}$ | 0.120 | 0.088 | 1.765 | 12.614 |
| $\boldsymbol{\beta}_{z z z}$ | 0.427 | -0.371 | 54.620 | 84.592 |
| $\beta_{\text {total }}$ | 0.754 | 0.665 | 117.285 | 109.499 |

Table S4 Theoretical calculated SHG coefficients matrix for 1.

| Phase | HTP | LTP |  |
| :--- | :--- | :--- | :--- |
| Point group | $m m m$ | 222 |  |
| SHG coefficients matrix | -- | $\left[\begin{array}{cccccc}0 & 0 & 0 & 0.145 & 0 & 0 \\ 0 & 0 & 0 & 0 & 0.145 & 0 \\ 0 & 0 & 0 & 0 & 0 & 0.145\end{array}\right]$ |  |
|  |  |  |  |
|  |  | $d_{14}=d_{25}=d_{36}=0.145 \mathrm{pm} / \mathrm{V}$ |  |

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