

Supporting information

Reduced Graphene Oxide Nanosheet modified NiMn-LDH nanoflake arrays for High- Performance Supercapacitors

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Experimental Section

Synthesize of the NiMn-LDH@NF electrode. A homogeneous solution consisting of 4 mmol $\text{Ni}(\text{NO}_3)_2 \cdot 6\text{H}_2\text{O}$, 4 mmol $\text{Mn}(\text{NO}_3)_2 \cdot 4\text{H}_2\text{O}$, 4 mmol urea and 25 mL DI water was transferred into a 40 mL Teflon-lined stainless-steel autoclave containing one pieces of cleaned Ni foam ($1 \times 4 \text{ cm}^2$). The autoclave was first sealed for 1 h to ensure sufficient wetting of Ni foam by the reactant solution, and then gradually heated to 140°C and maintained for 14 h. After the autoclave cooled down, the Ni foams were washed under ultrasonication with deionized water and ethanol each for 3 times and dried at 80°C. The loading mass of active material was about 2.5 mg cm^{-2} .

Reduced graphene oxide (rGO) modification. Graphene oxide (GO) was deposited via a simple “dip-coating” process for 3 times. The NiMn-LDH@NF electrode was immersed into a GO suspension (2 mg mL^{-1}) for 5 minutes, and then dried at 50 °C. The resultant electrode was washed with DI water and dried in air at room temperature. The as-synthesized

GO@NiMn-LDH@NF hybrid is then immersed in a hydrazine hydrate aqueous solution to obtain the rGO@NiMn-LDH@NF electrode. The mass loading of rGO was about 0.2 mg cm⁻². As a control sample, rGO@NF was also prepared by conducting the same procedure on Ni foam.

Material Characterizations and Electrochemical measurements. The structures of the electrodes were characterized by X-ray powder diffraction (XRD, Rigaku-D/max 2500 V), scanning electron microscopy (SEM, Sirion 200) and transmission electron microscopy (TEM, JEM 2100 F). The electrochemical performance of the electrodes was evaluated using a conventional three-electrode with Pt as the counter electrode, Ag/AgCl as the reference electrode, and 3 M KOH solution as the electrolyte. The asymmetric supercapacitor was constructed with rGO@NiMn-LDH@NF as the positive electrode and commercially obtained active carbon as the negative electrode. The asymmetric supercapacitor was assembled in a CR2016-type coin cell using 3 M KOH solution as the electrolyte, and one piece of cellulose paper as the separator. All the electrochemical tests were carried out on a CHI760E electrochemical workstation.

Supplementary Figures

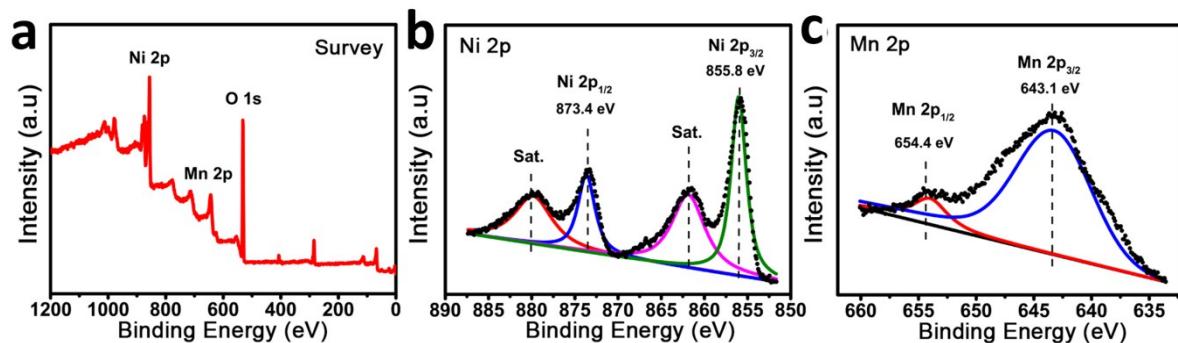


Figure S1. XPS (a) survey spectrum, and high resolution (b) Ni, (c) Mn spectra of NiMn-LDH.

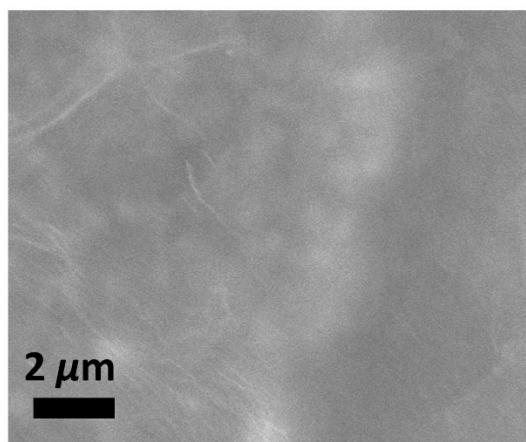


Figure S2. SEM image of rGO@NiMn-LDH@NF at high magnification.

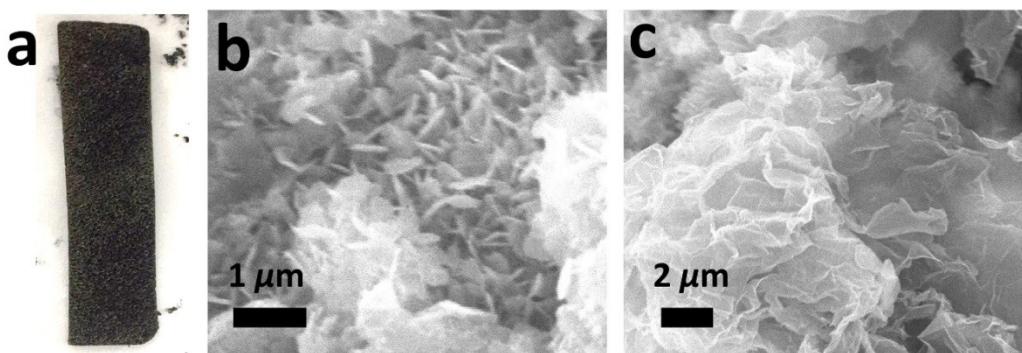


Figure S3. (a) Photo and (b, c) SEM images of the electrode surface directly modified with rGO by dip-coating method.

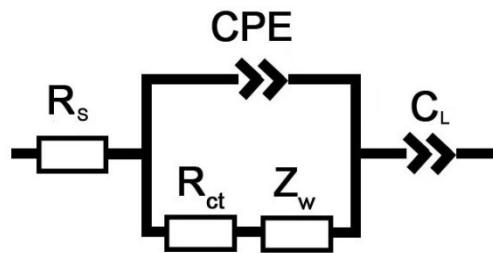


Figure S4. Equivalent circuit model of the electrodes.

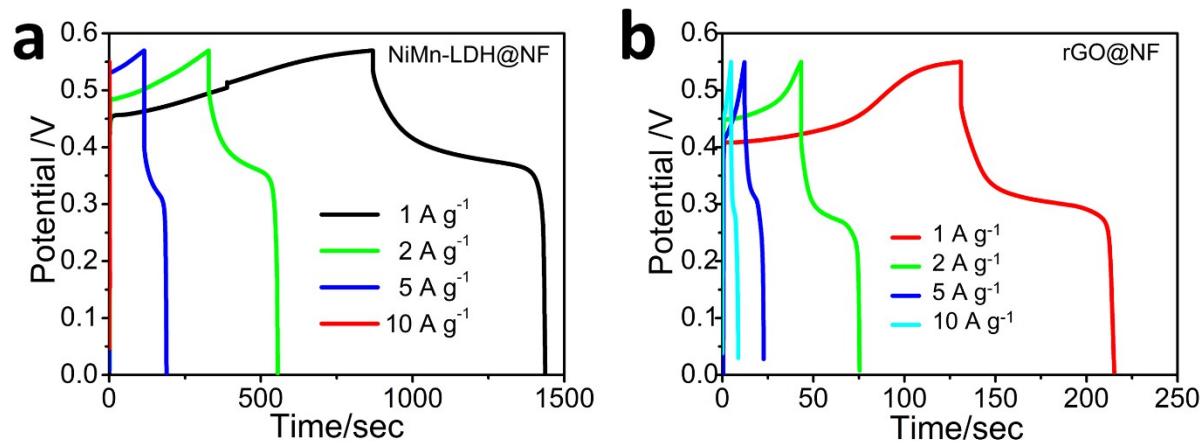


Figure S5. The galvanostatic charge-discharge curves of (a) NiMn-LDH@NF and (b) rGO@NF.

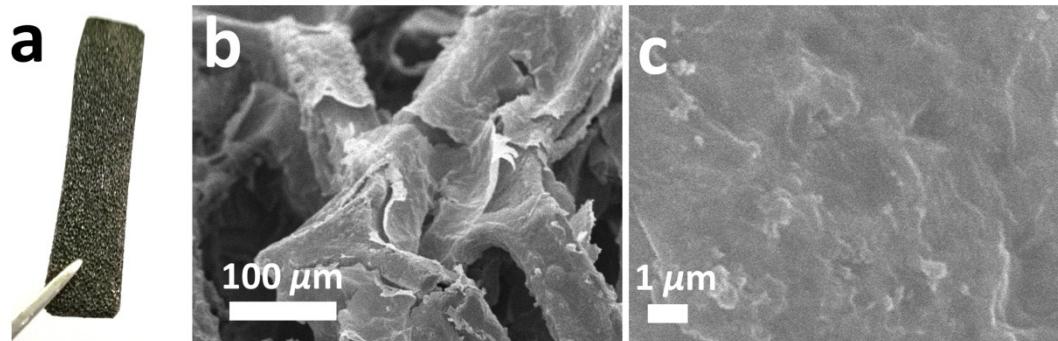


Figure S6. (a) Photo and (b, c) SEM images of the rGO@NiMn-LDH@NF electrode surface after cycling.

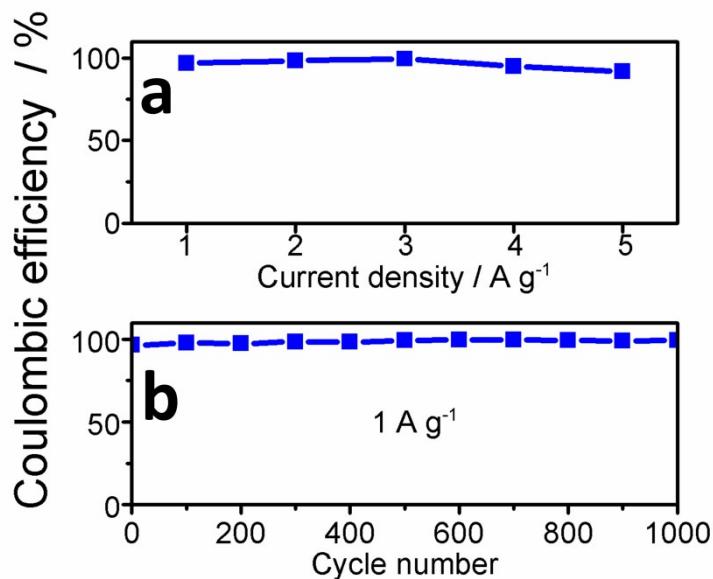


Figure S7. Coulombic efficiency of the ASC device as a function of (a) current densities and (b) cyclic numbers at 1 A g^{-1} .

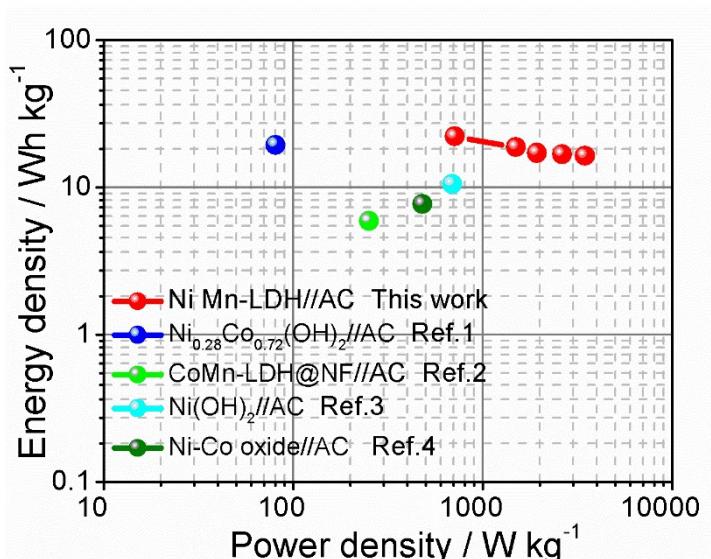


Figure S8. Energy and power densities of the ASC device. Below is the list of references:

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Supplementary Table

Table S1. Impedance parameters calculated from the equivalent circuit.

Electrodes	$R_s(\Omega)$	$R_{ct}(\Omega)$
rGO@NiMn-LDH@NF	0.61	0.91
NiMn-LDH@NF	1.44	16.66

Table S1. Comparison of capacitive performances in reported LDH electrodes.

Electrode materials	Specific capacitance at 1 A g^{-1} (F g^{-1})	Year	Reference
rGO@NiMn-LDH@NF	1696		This work
NiCo-LDH	1200	2018	1
CoAl-LDH/FGN	1222	2017	2
NiCo-LDH/CFC	2767	2017	3
NiMn-LDH/Porous carbon	1634	2017	4
GP@LDH	2395	2017	5
Ni-Co@Ni-Co LDH	2200	2017	6
Ti ₃ C ₂ /Ni-Co-Al-LDH	748	2017	7
rGO@MgAl-LDH	1334	2016	8
MXene/LDH	1061	2016	9
NiMn-LDH	1511	2016	10
CoMn LDH	1063	2016	11
CBC-N@LDH	1950	2016	12
MnCo-LDH@Ni(OH) ₂	2320	2016	13
NiCo-LDH/CNTs	1151	2015	14
EG-Co-Ni LDH	1313	2015	15
3DCGNC	1760	2015	16
Co ₃ O ₄ @NiAl-LDHs	1772	2014	17
LDHs-Nps/CH-NWs	1297	2014	18
Ni-Co LDH hybrid film	2682	2013	19
GNS/LDH	781	2011	20

Reference

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