

## Supporting Information of

# Site-Selective Conversion of Azido Groups at Carbonyl $\alpha$ -Positions into Oxime Groups Leading Triazide to Triple Click Conjugation Scaffold

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## [1] General Information Including Important Notices

**Caution!**: Organic azides, especially multiple azido compounds as well as diazo compounds are potentially hazardous and explosive. Although we have never experienced such an explosion with those used in this study, all manipulation should be carefully conducted behind a safety shield in a hood. Sodium azide should be handled with plastic spatula. At azidation stage of azido compound preparation, complete removal of residual halogenated solvent used in the last steps or extractions should be in mind to avoid generation of explosive species such as diazidomethane from dichloromethane.<sup>1</sup>

**Storage of TBAF:** Tetrabutylammonium fluoride (TBAF) was purchased from TCI (Tokyo Chemical Industry, Co. Ltd) as 1 mol/L solution of THF. Because we encountered irreproducible results when we used old solution, probably due to the decomposition of TBAF by the reagent itself or contaminated water from moisture,<sup>2</sup> the newly purchased bottle of TBAF solution was repacked in small subsection vial bottles. These were filled with nitrogen gas, and were stored in the refrigerator. With these small batch bottles, we successfully obtained reproducible results as described both in the manuscript and this supporting information.

**Analysis and Reagents:** <sup>1</sup>H and <sup>13</sup>C NMR spectra were recorded using a JEOL JNM-ECP500 spectrometer (500 MHz for <sup>1</sup>H NMR, and 126 MHz for <sup>13</sup>C NMR). Chemical shifts are reported as δ values in ppm and calibrated with respect to the residual solvent peak (CDCl<sub>3</sub>: δ 7.26 for <sup>1</sup>H NMR and δ 77.00 for <sup>13</sup>C NMR, CD<sub>3</sub>OD: δ 3.30 for <sup>1</sup>H NMR and δ 49.0 for <sup>13</sup>C NMR). The abbreviations used are as follows: s (singlet), d (doublet), t (triplet), q (quartet), br (broad), and m (multiplet). Melting points were measured using a Yanaco Micro melting point apparatus. Infrared spectra were measured using a JASCO FT-IR-4200 spectrometer. Mass spectra were recorded using a JEOL JMS-700 MStaion [EI-magnetic sector (70 eV), CI-magnetic sector, and ESI-TOF], and Bruker Autoflex II (MALDI-spiral TOF). All measurements of single crystal X-ray diffraction analysis were made on a Rigaku R-AXIS RAPID diffractometer using multi-layer mirror monochromated Mo-Kα radiation. The data were collected at a temperature of -150 or -170 °C. UV-visible spectra were recorded using JASCO V-630. The progress of the reactions was monitored by silica gel thin layer chromatography (TLC) (Merck TLC Silica gel 60 F<sub>254</sub>). Phosphomolybdic acid-cerium(IV) sulfate sulfuric acid solution was used for the TLC stains, and TLC was also monitored with UV lamp. Flash column chromatography was performed using neutral silica gel N60 from Kanto Chemical Co. Inc. If not specified as neutral silica gel column chromatography, Merck Silica gel 60 and packed column of Biotage® SNAP Ultra with HP-Sphere™ 25μm were used. If necessary, further purification of the crude materials was performed using a LC-908 recycling gel permeation chromatography (GPC) equipped with a JAIGEL 2H-40 column (chloroform elution) made by Japan Analytical Industry Co., Ltd. All reagents were purchased from Sigma-Aldrich, Wako Pure Chemical Industries, Ltd, TCI (Tokyo Chemical Industry, Co. Ltd), Kanto Chemical Co. Inc., and Nacalai Tescque. Anhydrous solvents such as tetrahydrofuran (THF), toluene, acetonitrile, and dichloromethane were purchased from Kanto Chemical and Wako Pure Chemical. Dimethyl sulfoxide (DMSO) was distilled under reduced pressure after refluxing in the presence of calcium hydride.

## [2] Synthesis of Oxime Compounds

### Preparation of Azido Substrates

Mono-, di-, and triazides **1a–l**, **3a–d**, **5a,b** were prepared in accordance with our previous report.<sup>3</sup>

### General Procedure for Synthesis of Oximes

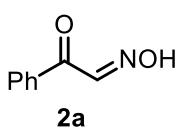
Stereochemistry of oximes (*E/Z*) was not determined. However, the products were obtained as a single stereoisomer unless otherwise noted. Chloroform-*d* should be passed through alumina prior to NMR experiments, otherwise the products isomerized to *E/Z* mixtures.

**From ketones:** TBAF (1.8 equiv, 1.0 M in THF) was added dropwise to a stirred solution of  $\alpha$ -azido ketone (1.0 equiv) and hydroxylammonium chloride (1.2 equiv) in DMSO (0.1 M based on azido substrate) at 25 °C (set by water bath) under the nitrogen gas atmosphere. After completion of the reaction checked by TLC, the mixture was diluted with ether and quenched with water. The solution was extracted three times with ether, and was washed with water and brine. The collected organic layer was dried over sodium sulfate. Concentration and purification by flash neutral silica gel column chromatography gave the oxime product.

**From amides:** TBAF (3.2 equiv, 1.0 M in THF) was added to a stirred solution of  $\alpha$ -azido amide (1.0 equiv) and hydroxylammonium chloride (1.2 equiv) in DMSO (0.1 M based on azido substrate) at 25 °C (set by water bath) under the nitrogen gas atmosphere. After completion of the reaction checked by TLC, the mixture was diluted with ether and quenched with water. The solution was extracted three times with ether, and was washed with water and brine. The collected organic layer was dried over sodium sulfate. Concentration and purification by flash neutral silica gel column chromatography gave the oxime product.

**NOTE:** Volume and the ratio of DMSO solvent in the reaction mixture is very important to succeed. However, we still used THF solution of TBAF, because use of commercial solid TBAF hydrate or removal of THF from the reagent solution prior to use gave low yields or irreproducible results. For storage of used TBAF, see General Information section.

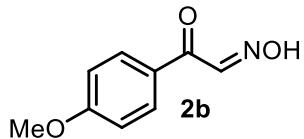
### **2-Oxo-2-phenylacetaldehyde oxime (2a)**



A total of 11.9 mg of **2a** (80%) was obtained from the reaction with azide **1a** (16.1 mg, 0.10 mmol), *N*-hydroxylammonium chloride (8.3 mg, 0.12 mmol, 1.2 equiv), and TBAF (0.18 mL, 1.0 M in THF, 0.18 mmol, 1.8 equiv) in DMSO (1.0 mL, 1.0 M) for 1 h followed by neutral silica gel chromatography (hexane/ethyl acetate = 15/1 to 8/1).

Yellow powder; *R*<sub>f</sub> value 0.60 (hexane/ethyl acetate = 1/1); m.p. 124–125 °C; IR (NaCl, neat)  $\nu_{\text{max}}$  3272, 2893, 1676, 1594, 1460, 1240 cm<sup>-1</sup>; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) δ 8.09–8.03 (m, 4H), 7.61 (t, 1H, *J* = 7.3 Hz), 7.48 (dd, 2H, *J* = 8.0, 8.0 Hz); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub>) δ 188.6, 148.4, 135.7, 133.7, 129.9, 128.5; LRMS (EI, M = C<sub>8</sub>H<sub>7</sub>NO<sub>2</sub>) *m/z* 149 (M<sup>+</sup>, 36%), 105 (100), 77 (63); HRMS (EI) calcd for C<sub>8</sub>H<sub>7</sub>NO<sub>2</sub> [M<sup>+</sup>] 149.0477, found 149.0475.

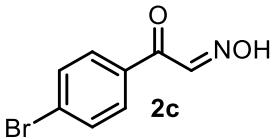
### **2-(4-Methoxyphenyl)-2-oxoacetaldehyde oxime (2b)**



A total of 15.3 mg of **2b** (86%) was obtained from the reaction with azide **1b** (19.0 mg, 0.10 mmol), *N*-hydroxylammonium chloride (8.3 mg, 0.12 mmol, 1.2 equiv), and TBAF (0.18 mL, 1.0 M in THF, 0.18 mmol, 1.8 equiv) in DMSO (1.0 mL, 1.0 M) for 5 h followed by neutral silica gel chromatography (hexane/ethyl acetate = 10/1 to 5/1).

Yellow powder;  $R_f$  value 0.50 (hexane/ethyl acetate = 1/1); m.p. 119–121 °C; IR (NaCl, neat)  $\nu_{\max}$  3225, 3176, 3044, 2984, 1597, 1571, 1259 cm<sup>-1</sup>; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) δ 8.56 (s, 1H), 8.09–8.05 (m, 3H), 6.95 (d, 2H, *J* = 8.5 Hz), 3.89 (s, 3H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub>) δ 186.6, 164.1, 148.7, 132.4, 128.6, 113.8, 55.5; LRMS (EI, M = C<sub>9</sub>H<sub>9</sub>NO<sub>3</sub>) *m/z* 179 (M<sup>+</sup>, 28%), 135 (100%); HRMS (EI) calcd for C<sub>9</sub>H<sub>9</sub>NO<sub>3</sub> [M<sup>+</sup>] 179.0582, found 179.0582.

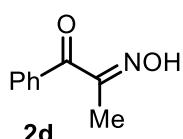
### **2-(4-Bromophenyl)-2-oxoacetaldehyde oxime (2c)**



A total of 14.9 mg of **2c** (65%) was obtained from the reaction with azide **1c** (24.0 mg, 0.10 mmol), *N*-hydroxylammonium chloride (8.3 mg, 0.12 mmol, 1.2 equiv), and TBAF (0.18 mL, 1.0 M in THF, 0.18 mmol, 1.8 equiv) in DMSO (1.0 mL, 1.0 M) for 1 h followed by neutral silica gel chromatography (hexane/ethyl acetate = 15/1 to 8/1).

Pale yellow powder;  $R_f$  value 0.5 (hexane/ethyl acetate = 2/1); m.p. 155–156 °C; IR (KBr, disc)  $\nu_{\max}$  3242, 3092, 1673, 1583, 1248 cm<sup>-1</sup>; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) δ 7.97 (s, 1H), 7.95 (d, 2H, *J* = 9.0 Hz), 7.62 (d, 2H, *J* = 9.0 Hz); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub>) δ 187.6, 148.9, 134.3, 131.7, 131.5, 129.0; LRMS (EI, M = C<sub>8</sub>H<sub>6</sub>BrNO<sub>2</sub>) *m/z* 229 (23%, M<sup>+</sup> of <sup>81</sup>Br), 227 (24, M<sup>+</sup> of <sup>79</sup>Br), 185 (96), 183 (100), 157 (37), 155 (37), 84 (84); HRMS (EI) calcd for C<sub>8</sub>H<sub>6</sub><sup>79</sup>BrNO<sub>2</sub> (M<sup>+</sup>) 226.9582, found 226.9588.

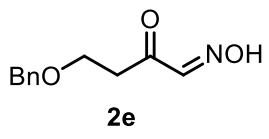
### **2-(Hydroxyimino)-1-phenylpropan-1-one (2d)**



A total of 11.8 mg of **2d** (73%) was obtained from the reaction with azide **1d** (17.4 mg, 0.10 mmol), *N*-hydroxylammonium chloride (8.3 mg, 0.12 mmol, 1.2 equiv), and TBAF (0.18 mL, 1.0 M in THF, 0.18 mmol, 1.8 equiv) in DMSO (1.0 mL, 1.0 M) for 1 h followed by neutral silica gel chromatography (hexane/ethyl acetate = 15/1 to 8/1).

White solid;  $R_f$  value 0.50 (hexane/ethyl acetate = 2/1); m.p. 113–114 °C; IR (NaCl, neat)  $\nu_{\max}$  3252, 2925, 1661, 1447, 1365, 1000 cm<sup>-1</sup>; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) δ 8.43 (s, 1H), 7.89 (d, 2H, *J* = 7.0 Hz), 7.57 (t, 1H, *J* = 7.5 Hz), 7.45 (dd, 2H, *J* = 7.0, 7.5 Hz), 2.17 (s, 3H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub>) δ 191.8, 156.9, 136.3, 132.8, 130.2, 128.2, 10.2; LRMS (EI, M = C<sub>9</sub>H<sub>9</sub>NO<sub>2</sub>) *m/z* 163 (M<sup>+</sup>, 30%), 105 (100), 77 (78); HRMS (EI) calcd for C<sub>9</sub>H<sub>9</sub>NO<sub>2</sub> (M<sup>+</sup>) 163.0633, found 163.0624.

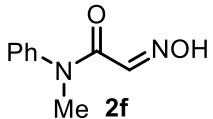
### **4-(Benzylxyloxy)-2-oxobutanal oxime (2e)**



A total of 11.2 mg of **2e** (54%) was obtained from the reaction with azide **1e** (21.9 mg, 0.10 mmol), *N*-hydroxylammonium chloride (8.3 mg, 0.12 mmol, 1.2 equiv), and TBAF (0.20 mL, 1.0 M in THF, 0.20 mmol, 2.0 equiv) in DMSO (1.0 mL, 1.0 M) for 2 h followed by neutral silica gel chromatography (hexane/ethyl acetate = 10/1 to 6/1).

Colorless oil;  $R_f$  value 0.33 (hexane/ethyl acetate = 2/1); IR (NaCl, neat)  $\nu_{\text{max}}$  3270, 3209, 3061, 2871, 1684, 1454  $\text{cm}^{-1}$ ;  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  9.64 (br, 1H), 7.56 (s, 1H), 7.37–7.31 (m, 5H), 4.58 (s, 2H), 3.87 (t, 2H,  $J$  = 7.0 Hz), 3.22 (m, 2H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$ )  $\delta$  196.6, 149.0, 137.0, 128.5, 128.1, 73.6, 65.0, 38.0; HRMS (ESI) calcd for  $\text{C}_{11}\text{H}_{12}\text{NO}_3$  [M–H] $^-$  206.0817, found 206.0812.

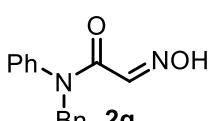
### **2-(Hydroxyimino)-N-methyl-N-phenylacetamide (2f)**



A total of 16.1 mg of **2f** (91%) was obtained from the reaction with azide **1f** (19.0 mg, 0.10 mmol), *N*-hydroxylammonium chloride (8.3 mg, 0.12 mmol, 1.2 equiv), and TBAF (0.32 mL, 1.0 M in THF, 0.32 mmol, 3.2 equiv) in DMSO (1.0 mL, 1.0 M) for 1 h followed by neutral silica gel chromatography (hexane/ethyl acetate = 10/1 to 3/1 to 1/1 to 1/2).

White solid;  $R_f$  value 0.20 (hexane/ethyl acetate = 1/1); m.p. 168–170 °C; IR (NaCl, neat)  $\nu_{\text{max}}$  3165, 1654, 1590, 1474, 1389, 1065  $\text{cm}^{-1}$ ;  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  10.23 (s, 1H), 7.48 (s, 1H), 7.43 (dd, 2H,  $J$  = 7.5, 7.5 Hz), 7.37 (t, 1H,  $J$  = 7.3 Hz), 7.19 (d, 2H,  $J$  = 7.5 Hz), 3.38 (s, 3H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$ )  $\delta$  161.5, 142.2, 141.8, 130.0, 128.3, 127.0, 37.5; LRMS (EI, M =  $\text{C}_9\text{H}_{10}\text{N}_2\text{O}_2$ )  $m/z$  178 ( $\text{M}^+$ , 69%), 161 (70), 106 (100); HRMS (EI) calcd for  $\text{C}_9\text{H}_{10}\text{N}_2\text{O}_2$  ( $\text{M}^+$ ) 178.0742, found 178.0736.

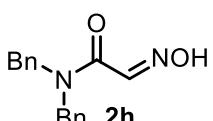
### **N-Benzyl-2-(hydroxyimino)-N-phenylacetamide (2g)**



A total of 25.1 mg of **2g** (99%) was obtained from the reaction with azide **1g** (26.5 mg, 0.10 mmol), *N*-hydroxylammonium chloride (8.3 mg, 0.12 mmol, 1.2 equiv), and TBAF (0.27 mL, 1.0 M in THF, 0.27 mmol, 2.7 equiv) in DMSO (1.0 mL, 1.0 M) for 1 h followed by neutral silica gel chromatography (hexane/ethyl acetate = 10/1 to 3/1 to 1/1).

Colorless amorphous solid;  $R_f$  value 0.4 (hexane/ethyl acetate = 1/1); IR (NaCl, neat)  $\nu_{\text{max}}$  3298, 3063, 1656, 1593, 1495, 1454, 1253  $\text{cm}^{-1}$ ;  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  7.45 (s, 1H), 7.34–7.32 (m, 3H), 7.26–7.24 (m, 3H), 7.21–7.19 (m, 2H), 6.99–6.47 (m, 2H), 4.98 (s, 2H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$ )  $\delta$  161.3, 142.4, 140.1, 136.3, 129.8, 128.9, 128.5, 128.4, 128.2, 127.6, 53.1; HRMS (CI) calcd for  $\text{C}_{15}\text{H}_{15}\text{N}_2\text{O}_2$  [M+H] $^+$  255.1134, found 255.1140.

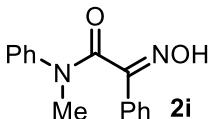
### **N,N-Dibenzyl-2-(hydroxyimino)acetamide (2h)**



A total of 25.8 mg of **2h** (96%) was obtained from the reaction with azide **1h** (28.0 mg, 0.10 mmol), *N*-hydroxylammonium chloride (8.3 mg, 0.12 mmol, 1.2 equiv), and TBAF (0.32 mL, 1.0 M in THF, 0.32 mmol, 3.2 equiv) in DMSO (1.0 mL, 1.0 M) for 1 h followed by neutral silica gel chromatography (hexane/ethyl acetate = 10/1 to 3/1).

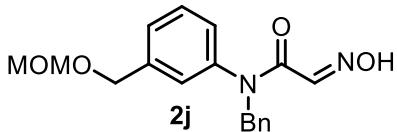
White amorphous solid;  $R_f$  value 0.53 (hexane/ethyl acetate = 1/1); IR (NaCl, neat)  $\nu_{\text{max}}$  3061, 3030, 2925, 1644, 1604, 1496, 1454  $\text{cm}^{-1}$ ;  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  7.92 (s, 1H), 7.38–7.26 (m, 7H), 7.26–7.18 (m, 3H), 4.60 (s, 2H), 4.59 (s, 2H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$ )  $\delta$  163.0, 143.4, 136.1, 135.8, 128.9, 128.7, 128.5, 127.9, 127.7, 127.0, 50.0, 48.0; HRMS (CI) calcd for  $\text{C}_{16}\text{H}_{17}\text{N}_2\text{O}_2$  [M+H] $^+$  269.1290, found 269.1295.

### 2-(Hydroxyimino)-N-methyl-N,2-diphenylacetamide (2i)



Total of 23.2 mg of **2i** (91%) was obtained from the reaction with azide **1i** (26.6 mg, 0.10 mmol), *N*-hydroxylammonium chloride (8.3 mg, 0.12 mmol, 1.2 equiv), and TBAF (0.32 mL, 1.0 M in THF, 0.32 mmol, 3.2 equiv) in DMSO (1.0 mL, 0.1 M) for 2 h. However in this case, the reaction mixture was diluted with diethyl ether and quenched with saturated aqueous solution of sodium bicarbonate. This quenching method was important to prevent the generation of hydrolyzed material of oxime **2i**. The mixture were extracted three times with ether and was washed with water and brine. The collected organic layer was dried over sodium sulfate. Concentration and purification by neutral silica gel column chromatography (hexane/ethyl acetate = 5/1 to 2/1 with 1% of triethylamine) afforded the product as above. Colorless oil;  $R_f$  value 0.4 (hexane/ethyl acetate = 1/1); IR (NaCl, neat)  $\nu_{\text{max}}$  3238, 3061, 1649, 1594, 1496  $\text{cm}^{-1}$ ;  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  9.60 (br, 1H), 7.66 (br, 1H), 7.55–7.41 (m, 2H), 7.40 (dd, 1H,  $J$  = 7.5, 7.0 Hz), 7.33 (dd, 1H,  $J$  = 7.5, 7.5 Hz), 7.20–7.15 (m, 3H), 7.01 (br, 2H), 3.45 (s, 3H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$ )  $\delta$  173.3, 167.8, 141.8, 134.9, 131.1, 129.4, 129.3, 128.4, 127.8, 127.3, 126.5, 36.7; HRMS (EI) calcd for  $\text{C}_{15}\text{H}_{14}\text{N}_2\text{O}_2$  ( $\text{M}^+$ ) 254.1055, found 254.1055.

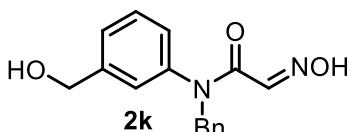
### *N*-Benzyl-2-(hydroxyimino)-*N*-(3-((methoxymethoxy)methyl)phenyl)acetamide (2j)



A total of 31.5 mg of **2j** (96%, ratio of isomers = 6.3 : 1 based on  $^1\text{H}$  NMR) as isomeric mixture was obtained from the reaction with azide **1j** (34.0 mg, 0.10 mmol), *N*-hydroxylammonium chloride (8.3 mg, 0.12 mmol, 1.2 equiv), and TBAF (0.32 mL, 1.0 M in THF, 0.32 mmol, 3.2 equiv) in DMSO (1.0 mL, 1.0 M) for 1 h followed by neutral silica gel chromatography (hexane/ethyl acetate = 6/1 to 2/1 to 1/2).

Colorless amorphous solid;  $R_f$  value 0.33 (hexane/ethyl acetate = 1/1); IR (NaCl, neat)  $\nu_{\text{max}}$  3316, 2935, 2887, 1660, 1453, 1047  $\text{cm}^{-1}$ ;  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ ) peaks from major isomer are listed.  $\delta$  7.45 (s, 1H), 7.30 (d, 2H,  $J$  = 5.0 Hz), 7.26–7.18 (m, 5H), 7.02 (s, 1H), 6.88 (m, 1H), 4.97 (s, 2H), 4.65 (s, 2H), 4.53 (s, 2H), 3.36 (s, 3H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$ ) Peaks from major isomer are listed.  $\delta$  161.5, 142.1, 140.2, 140.1, 136.3, 129.7, 128.9, 128.4, 127.6, 127.4, 127.1, 95.7, 68.1, 55.4, 53.1; LRMS (EI, M =  $\text{C}_{18}\text{H}_{20}\text{N}_2\text{O}_4$ )  $m/z$  328 ( $\text{M}^+$ , 2%), 311 (9), 194 (16), 91 (100); HRMS (EI) calcd for  $\text{C}_{18}\text{H}_{20}\text{N}_2\text{O}_4$  ( $\text{M}^+$ ) 328.1423, found 328.1427.

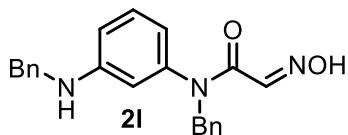
### *N*-Benzyl-2-(hydroxyimino)-*N*-(3-(hydroxymethyl)phenyl)acetamide (2k)



A total of 25.5 mg of **2k** (90%) as single isomer was obtained from the reaction with azide **1k** (29.5 mg, 0.10 mmol), hydroxylammonium chloride (8.3 mg, 0.12 mmol, 1.2 equiv), and TBAF (0.32 mL, 1.0 M in THF, 0.32 mmol, 3.2 equiv) in DMSO (1.0 mL, 1.0 M) for 1 h followed by neutral silica gel chromatography (hexane/ethyl acetate = 2/1 to 1/1 to 1/2 with 2% methanol).

White powder;  $R_f$  value 0.3 (hexane/ethyl acetate = 1/2); m.p. 137–138 °C; IR (NaCl, neat)  $\nu_{\text{max}}$  2870, 1655, 1602, 1586, 1453  $\text{cm}^{-1}$ ;  $^1\text{H}$  NMR (500 MHz,  $\text{CD}_3\text{OD}$ )  $\delta$  7.36–7.32 (m, 3H), 7.28–7.20 (m, 5H), 7.14 (s, 1H), 6.95–6.94 (m, 1H), 5.01 (s, 2H), 4.56 (s, 2H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CD}_3\text{OD}$ )  $\delta$  163.9, 145.2, 143.0, 141.7, 138.0, 130.7, 129.7, 129.5, 128.7, 128.2, 127.9, 127.4, 64.3, 54.0; HRMS (CI) calcd for  $\text{C}_{16}\text{H}_{17}\text{N}_2\text{O}_3$  [ $\text{M}+\text{H}$ ]<sup>+</sup> 285.1239, found 285.1230.

**N-Benzyl-N-(3-(benzylamino)phenyl)-2-(hydroxyimino)acetamide (2I)**

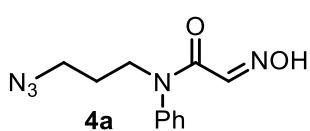


A total of 34.4 mg of **2I** (96%, isomeric ratio = 4.9 : 1 based on  $^1\text{H}$  NMR) as isomeric mixture was obtained from the reaction with azide **11** (37.0 mg, 0.10 mmol), *N*-hydroxylammonium chloride (8.3 mg, 0.12 mmol, 1.2 equiv), and TBAF (0.32 mL, 1.0 M in THF, 0.32 mmol, 3.2 equiv)

in DMSO (1.0 mL, 1.0 M) for 2 h followed by neutral silica gel chromatography (hexane/ethyl acetate = 5/1 to 3/1 to 2/1 to 1/1).

Yellow amorphous solid;  $R_f$  value 0.3 (hexane/ethyl acetate = 1/1); IR (NaCl, neat)  $\nu_{\text{max}}$  3354, 3061, 6029, 1656, 1602, 1495, 1453  $\text{cm}^{-1}$ ;  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ ) Peaks from major isomer are listed.  $\delta$  10.7 (br, 1H), 7.54 (s, 1H), 7.35 (dd, 2H,  $J$  = 7.5, 6.5 Hz), 7.31–7.26 (m, 3H), 7.25–7.19 (m, 5H), 7.07 (dd, 1H,  $J$  = 8.0, 8.0 Hz), 6.54 (dd, 1H,  $J$  = 8.0, 1.5 Hz), 6.30 (d, 1H,  $J$  = 6.5 Hz), 6.20 (s, 1H), 4.90 (s, 2H), 4.20 (s, 2H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$ ) Peaks from major isomer are listed.  $\delta$  161.4, 149.1, 142.4, 141.4, 138.4, 136.7, 130.3, 128.9, 128.7, 128.3, 127.45, 127.42, 127.39, 116.7, 112.9, 111.9, 52.9, 47.9; HRMS (MALDI-TOF) calcd for  $\text{C}_{22}\text{H}_{21}\text{N}_3\text{O}_2\text{Na}$  [ $\text{M}+\text{Na}]^+$  382.1531, found 382.1526.

**N-(3-Azidopropyl)-2-(hydroxyimino)-N-phenylacetamide (4a)**

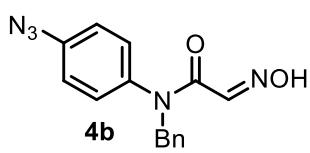


A total of 22.7 mg of **4a** (92%) as single isomer was obtained from the reaction with azide **3a** (25.9 mg, 0.10 mmol), *N*-hydroxylammonium chloride (8.3 mg, 0.12 mmol, 1.2 equiv), and TBAF (0.32 mL, 1.0 M in THF, 0.32 mmol, 3.2 equiv) in DMSO (1.0 mL, 1.0 M) for 1 h followed by neutral

silica gel chromatography (hexane/ethyl acetate = 5/1 to 1/1).

White solid;  $R_f$  value 0.15 (hexane/ethyl acetate = 1/1); m.p. 98–100 °C; IR (NaCl, neat)  $\nu_{\text{max}}$  3260, 2936, 2098, 1659, 1593, 1493, 1254  $\text{cm}^{-1}$ ;  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  10.35 (s, 1H), 7.45–7.37 (m, 4H), 7.16 (d, 2H,  $J$  = 7.5 Hz), 3.87 (t, 2H,  $J$  = 7.5 Hz), 3.34 (t, 2H,  $J$  = 7.0 Hz), 1.85 (tt, 2H,  $J$  = 7.5, 7.0 Hz);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$ )  $\delta$  161.4, 142.2, 140.2, 130.1, 128.6, 127.8, 49.0, 47.2, 27.0; LRMS (EI,  $\text{M} = \text{C}_{11}\text{H}_{13}\text{N}_5\text{O}_2$ )  $m/z$  247 ( $\text{M}^+$ , 5%), 164 (26), 147 (21), 119 (28), 106 (100); HRMS (EI) calcd for  $\text{C}_{11}\text{H}_{13}\text{N}_5\text{O}_2$  [ $\text{M}^+$ ] 247.1069, found 247.1078.

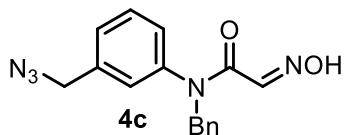
**N-(4-Azidophenyl)-N-benzyl-2-(hydroxyimino)acetamide (4b)**



A total of 27.4 mg of **4b** (93%) was obtained from the reaction with azide **3b** (30.6 mg, 0.10 mmol), *N*-hydroxylammonium chloride (8.3 mg, 0.12 mmol, 1.2 equiv), and TBAF (0.32 mL, 1.0 M in THF, 0.32 mmol, 3.2 equiv) in DMSO (1.0 mL, 1.0 M) for 1 h followed by neutral silica gel chromatography (hexane/ethyl acetate = 6/1 to 3/1 to 1/1).

Pale yellow amorphous solid;  $R_f$  value 0.43 (hexane/ethyl acetate = 1/1); IR (NaCl, neat)  $\nu_{\text{max}}$  3298, 3063, 2930, 2122, 1663, 1505, 1298, 1280  $\text{cm}^{-1}$ ;  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  7.44 (s, 1H), 7.26–7.25 (m, 3H), 7.19 (dd, 2H,  $J$  = 7.5, 3.5 Hz), 6.99–6.94 (m, 4H), 4.96 (s, 2H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$ )  $\delta$  161.3, 142.5, 140.4, 136.6, 136.1, 129.7, 129.0, 128.6, 127.8, 120.2, 53.1; LRMS (EI,  $\text{M} = \text{C}_{15}\text{H}_{13}\text{N}_5\text{O}_2$ )  $m/z$  295 (2%,  $\text{M}^+$ ), 282 (9), 267 (38), 223 (11), 91 (100); HRMS (EI) calcd for  $\text{C}_{15}\text{H}_{13}\text{N}_5\text{O}_2$  ( $\text{M}^+$ ) 295.1069, found 295.1078.

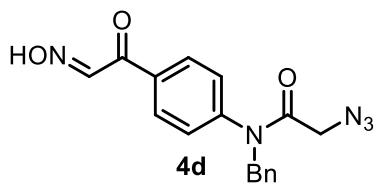
**N-(3-(Azidomethyl)phenyl)-N-benzyl-2-(hydroxyimino)acetamide (4c)**



A total of 28.9 mg of **4c** (94%) was obtained from the reaction with azide **3c** (32.0 mg, 0.10 mmol), *N*-hydroxylammonium chloride (8.3 mg, 0.12 mmol, 1.2 equiv), and TBAF (0.32 mL, 1.0 M in THF, 0.32 mmol, 3.2 equiv) in DMSO (1.0 mL, 1.0 M) for 1 h followed by neutral silica gel chromatography (hexane/ethyl acetate = 5/1 to 3/1 to 1/1).

Colorless amorphous solid;  $R_f$  value 0.43 (hexane/ethyl acetate = 1/1); IR (NaCl, neat)  $\nu_{\max}$  3298, 3062, 2932, 2101, 1656, 1451, 1258  $\text{cm}^{-1}$ ;  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  10.4 (br, 1H), 7.44 (s, 1H), 7.35 (dd, 1H,  $J$  = 8.5, 7.5 Hz), 7.29–7.26 (m, 3H), 7.26–7.24 (m, 2H), 7.20–7.15 (m, 2H), 6.95–6.94 (m, 2H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$ )  $\delta$  161.3, 142.2, 140.6, 137.4, 136.1, 130.2, 128.9, 128.51, 128.46, 128.14, 128.11, 127.77, 127.67, 53.8, 53.1; LRMS (EI, M =  $\text{C}_{16}\text{H}_{15}\text{N}_5\text{O}_2$ )  $m/z$  309 (7%,  $\text{M}^+$ ), 292 (20), 237 (19), 209 (30), 91 (100); HRMS (EI) calcd for  $\text{C}_{16}\text{H}_{15}\text{N}_5\text{O}_2$  ( $\text{M}^+$ ) 309.1226, found 309.1225.

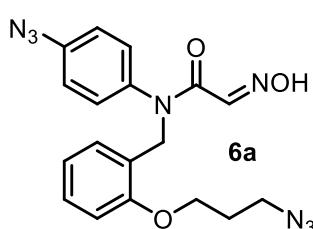
**2-Azido-N-benzyl-N-(4-(2-(hydroxyimino)acetyl)phenyl)acetamide (4d)**



A total of 21.8 mg of **4d** (65%) was obtained from the reaction with azide **3d** (34.9 mg, 0.10 mmol), *N*-hydroxylammonium chloride (8.3 mg, 0.12 mmol, 1.2 equiv), and TBAF (0.18 mL, 1.0 M in THF, 0.18 mmol, 1.8 equiv) in DMSO (1.0 mL, 1.0 M) for 1 h followed by neutral silica gel chromatography (hexane/ethyl acetate = 10/1 to 6/1 to 2/1).

Yellow oil;  $R_f$  value 0.55 (hexane/ethyl acetate = 1/1); IR (NaCl, neat)  $\nu_{\max}$  3282, 3060, 2107, 1657, 1600, 1426, 1260  $\text{cm}^{-1}$ ;  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  9.38 (br, 1H), 8.06 (d, 2H,  $J$  = 8.5 Hz), 7.96 (s, 1H), 7.26–7.25 (m, 3H), 7.17–7.15 (m, 2H), 7.06 (d, 2H,  $J$  = 8.0 Hz), 4.92 (s, 2H), 3.62 (s, 2H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$ )  $\delta$  188.1, 167.5, 148.4, 144.0, 135.8, 135.5, 131.8, 128.7, 128.6, 127.9, 127.8, 53.3, 50.9; HRMS (CI) calcd for  $\text{C}_{17}\text{H}_{16}\text{N}_5\text{O}_3$  [ $\text{M}+\text{H}]^+$  338.1253, found 338.1255.

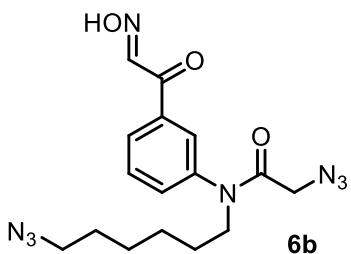
**N-(4-Azidophenyl)-N-(2-(3-azidopropoxy)benzyl)-2-(hydroxyimino)acetamide (6a)**



A total of 33.0 mg of **6a** (84%) was obtained from the reaction with azide **5a** (40.6 mg, 0.10 mmol), *N*-hydroxylammonium chloride (8.3 mg, 0.12 mmol, 1.2 equiv), and TBAF (0.32 mL, 1.0 M in THF, 0.32 mmol, 3.2 equiv) in DMSO (1.0 mL, 1.0 M) for 1 h followed by neutral silica gel chromatography (hexane/ethyl acetate = 6/1 to 3/1 to 2/1 to 1/1).

Pale yellow amorphous solid;  $R_f$  value 0.3 (hexane/ethyl acetate = 1/1); IR (NaCl, neat)  $\nu_{\max}$  2932, 2877, 2099, 1663, 1505, 1297, 1247  $\text{cm}^{-1}$ ;  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  10.0 (br, 1H), 7.43 (s, 1H), 7.21 (dd, 1H,  $J$  = 7.5, 7.0 Hz), 7.17 (d, 1H,  $J$  = 7.5 Hz), 6.97–6.93 (m, 4H), 6.85 (dd, 1H,  $J$  = 7.5, 7.5 Hz), 6.78 (d, 1H,  $J$  = 8.0 Hz), 5.03 (s, 2H), 3.91 (t, 2H,  $J$  = 5.5 Hz), 3.40 (t, 2H,  $J$  = 6.5 Hz), 1.89 (t, 2H,  $J$  = 6.5, 6.0 Hz);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$ )  $\delta$  161.1, 156.5, 142.5, 140.2, 136.7, 130.9, 129.7, 129.2, 124.0, 120.8, 119.9, 110.9, 64.2, 48.0, 47.6, 28.6; LRMS (EI, M =  $\text{C}_{18}\text{H}_{18}\text{N}_8\text{O}_3$ )  $m/z$  394 (M<sup>+</sup>, 2%), 366 (9), 162 (23), 134 (100), 105 (35); HRMS (EI) calcd for  $\text{C}_{18}\text{H}_{18}\text{N}_8\text{O}_3$  ( $\text{M}^+$ ) 394.1502, found 194.1505.

**2-Azido-N-(6-azidohexyl)-N-(3-(2-(hydroxyimino)acetyl)phenyl)acetamide (6b, CCDC 1879115)**

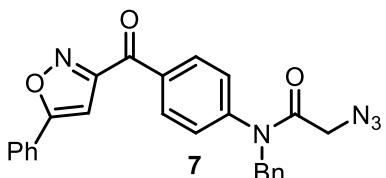


A total of 1.12 g of **6b** (65%) as single isomer was obtained from the reaction with azide **5b** (1.78 g, 4.64 mmol), hydroxylammonium chloride (387.4 mg, 5.57 mmol, 1.2 equiv), and TBAF (8.4 mL, 1.0 M in THF, 8.40 mmol, 1.8 equiv) in DMSO (1.0 mL, 1.0 M) for 1 h followed by neutral silica gel chromatography (hexane/ethyl acetate = 4/1 to 3/1). Recrystallization for X-ray analysis was performed with hexane/dichloromethane.

Pale yellow solid;  $R_f$  value 0.2 (hexane/ethyl acetate = 2/1); m.p. 85–86 °C; IR (NaCl, neat)  $\nu_{\text{max}}$  3265, 2935, 2861, 2105, 1655, 1444, 1256  $\text{cm}^{-1}$ ;  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  10.1 (s, 1H), 8.05 (d, 1H,  $J$  = 8.0 Hz), 7.98 (s, 1H), 7.95 (s, 1H), 7.59 (dd, 1H,  $J$  = 8.5, 7.0 Hz), 7.42–7.40 (m, 1H), 3.76 (t, 2H,  $J$  = 7.5 Hz), 3.66 (s, 2H), 3.26 (t, 2H,  $J$  = 7.0 Hz), 1.55 (tt, 4H,  $J$  = 7.0, 7.0 Hz), 1.40–1.32 (m, 4H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$ )  $\delta$  188.0, 167.6, 149.0, 139.8, 137.6, 131.9, 130.7, 130.4, 130.0, 51.5, 51.3, 49.7, 28.6, 27.2, 26.3, 26.1; HRMS (CI) calcd for  $\text{C}_{16}\text{H}_{21}\text{N}_8\text{O}_3$  [ $\text{M}+\text{H}]^+$  373.1737, found 373.1729.

**Chemoselective conjugation with azido oxime 4d**

**2-Azido-N-benzyl-N-(4-(5-phenylisoxazole-3-carbonyl)phenyl)acetamide (7)**



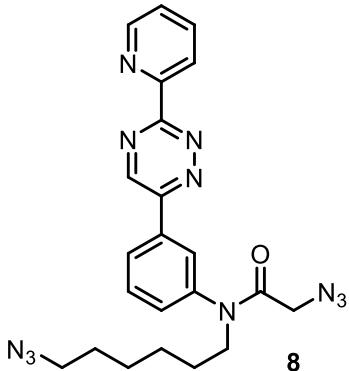
To a solution of **4d** (16.9 mg, 0.050 mmol) and phenyl acetylene (27.5  $\mu\text{L}$ , 0.25 mmol, 5.0 equiv) in methanol/water (0.50 mL, 5/1, 0.1 M) was added [bis(trifluoroacetoxy)iodo]benzene (33.2 mg, 0.075 mmol, 1.5 equiv) at room temperature. After 1 h, the solvent was removed under reduced pressure to obtain crude material, which was purified by neutral silica gel column chromatography (hexane/ethyl acetate = 8/1 to 5/1) followed by GPC purification to afford 9.1 mg of **7** (42%).

Colorless oil;  $R_f$  value 0.57 (hexane/ethyl acetate = 2/1); IR (NaCl, neat)  $\nu_{\text{max}}$  2105, 1669, 1600, 1441, 1251  $\text{cm}^{-1}$ ;  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  8.37 (dd, 2H,  $J$  = 7.0, 6.5 Hz), 7.85 (dd, 2H,  $J$  = 8.0, 3.0 Hz), 7.54–7.51 (m, 3H), 7.30–7.28 (m, 3H), 7.21 (dd, 2H,  $J$  = 7.5, 7.0 Hz), 7.17 (d, 2H,  $J$  = 8.0 Hz), 7.06 (s, 1H), 4.96 (s, 2H), 3.67 (s, 2H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$ )  $\delta$  184.4, 171.1, 166.9, 162.1, 145.3, 136.1, 135.6, 132.4, 130.9, 129.2, 128.9, 128.7, 128.3, 128.0, 126.4, 126.0, 100.1, 53.3, 50.9; LRMS (EI, M =  $\text{C}_{25}\text{H}_{19}\text{N}_5\text{O}_3$ )  $m/z$  437 ( $\text{M}^+$ , 1%), 353 (17), 277 (11), 208 (10), 146 (20), 91 (100); HRMS (EI) calcd for  $\text{C}_{25}\text{H}_{19}\text{N}_5\text{O}_3$  ( $\text{M}^+$ ) 437.1488, found 437.1485.

### [3] Synthesis of 1,2,4-Triazines

Synthesis of 3-(2-pyridyl)-1,2,4-triazines from  $\alpha$ -hydroxyimino ketones was performed in accordance with Vrabel's report.<sup>4</sup>

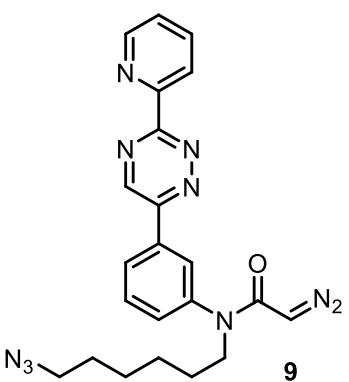
#### 2-Azido-N-(6-azidohexyl)-N-(3-(3-(pyridin-2-yl)-1,2,4-triazin-6-yl)phenyl)acetamide (8)



To a stirred solution of oxime **6b** (933 mg, 2.5 mmol) and hydrazine monohydrate (0.12 mL, 2.5 mmol, 1.0 equiv) in ethanol (3.8 mL, 0.67 M) was added one drop of acetic acid, and the mixture was heated at 45 °C for 16 h. Then, concentration of the reaction mixture *in vacuo* gave crude hydrazone (1.10 g) which was submitted to the next step without further purification.

To a stirred solution of crude hydrazone and 2-pyridinecarboxyaldehyde (0.29 mL, 3.0 mmol, 1.2 equiv) in ethanol (16.7 mL, 0.15 M) was added three drops of acetic acid. After 23 h, the reaction mixture was concentrated *in vacuo*. The obtained crude material was dissolved in acetic acid (4.2 mL, 0.60 M), and the mixture stirred was heated at 100 °C. After 1 h, the reaction mixture was diluted with water, and was quenched with saturated aqueous solution of sodium bicarbonate. The organic components were extracted three times with dichloromethane and washed with brine, and then was dried over sodium sulfate. Concentration of the organic layer followed by purification by silica gel column chromatography (hexane/ethyl acetate = 1/1 to 1/2 with 2% methanol to dichloromethane/methanol = 20/1) gave 959 mg of **8** (84% for 3 steps). Brown oil; R<sub>f</sub> value 0.13 (hexane/ethyl acetate = 1/1); IR (NaCl, neat) ν<sub>max</sub> 2934, 2103, 1669, 1584, 1403, 1258 cm<sup>-1</sup>; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) δ 9.21 (s, 2H), 8.91 (d, 1H, J = 4.0 Hz), 8.75 (d, 1H, J = 8.0 Hz), 8.14–8.13 (m, 2H), 7.96 (ddd, 1H, J = 8.0, 7.5, 2.0 Hz), 7.69 (dd, 1H, J = 8.0, 8.0 Hz), 7.51 (dd, 1H, J = 7.5, 7.0 Hz), 7.38 (d, 1H, J = 8.5 Hz), 3.80 (t, 2H, J = 8.0 Hz), 3.64 (s, 2H), 3.23 (t, 2H, J = 7.0 Hz), 1.58 (tt, 4H, J = 7.5, 7.5 Hz), 1.36 (t, 4H, J = 3.5 Hz); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub>) δ 166.9, 162.2, 154.6, 152.0, 150.6, 147.0, 142.0, 137.3, 135.2, 131.2, 130.6, 126.8, 126.5, 125.9, 124.1, 51.2, 50.9, 49.7, 28.6, 27.4, 26.3, 26.2; HRMS (MALDI-TOF) calcd for C<sub>22</sub>H<sub>23</sub>N<sub>11</sub>ONa [M+Na]<sup>+</sup> 480.1983, found 480.1973.

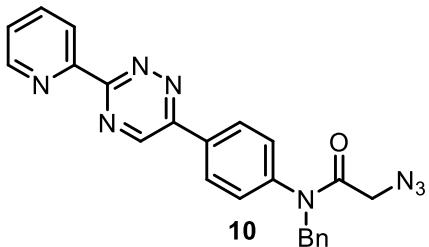
#### N-(6-Azidohexyl)-2-diazo-N-(3-(3-(pyridin-2-yl)-1,2,4-triazin-6-yl)phenyl)acetamide (9)



To a stirred solution of diazide **8** (666 mg, 1.46 mmol), 4-toluenesulfonyl hydrazide (1.36 g, 7.28 mmol, 5.0 equiv), and pyrrolidine (0.61 mL, 7.28 mmol, 5.0 equiv) in DMSO (15 mL, 0.1 M) was added TBAF (8.7 mL, 8.74 mmol, 6.0 equiv) dropwise at 25 °C. After 4 h, the resulting mixture was diluted with dichloromethane and was quenched with water. Organic components were extracted 3 times with dichloromethane and washed twice with water. Drying collected organic layer over sodium sulfate followed by concentration *in vacuo* and purification by neutral silica gel column chromatography (hexane/ethyl acetate = 1/1 to dichloromethane elution to dichloromethane/methanol = 20/1) and GPC gave 256 mg of **9** (40%).

Yellow amorphous solid;  $R_f$  value 0.33 (dichloromethane/methanol = 15/1); IR (NaCl, neat)  $\nu_{\text{max}}$  2934, 2860, 2102, 1621, 1584, 1402  $\text{cm}^{-1}$ ;  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  9.20 (s, 1H), 8.92 (d, 1H,  $J$  = 4.5 Hz), 8.76 (d, 1H,  $J$  = 8.5 Hz), 8.12–8.10 (m, 2H), 7.97 (ddd, 1H,  $J$  = 8.0, 8.0, 2.0 Hz), 7.66 (dd, 1H,  $J$  = 8.5, 8.0 Hz), 7.52 (ddd, 1H,  $J$  = 7.5, 7.5, 1.0 Hz), 7.42–7.40 (m, 1H), 4.50 (s, 1H), 3.83 (t, 2H,  $J$  = 7.5 Hz), 3.23 (t, 2H,  $J$  = 7.0 Hz), 1.60–1.56 (m, 4H), 1.37 (t, 4H,  $J$  = 4.0, 3.0 Hz);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$ )  $\delta$  165.4, 162.0, 154.9, 152.0, 150.6, 147.0, 142.9, 137.3, 134.8, 131.1, 130.9, 126.8, 126.2, 125.9, 124.0, 51.2, 49.1, 47.7, 28.7, 28.0, 26.4, 26.2; HRMS (ESI) calcd for  $\text{C}_{22}\text{H}_{23}\text{N}_{11}\text{O}\text{Na}$  [ $\text{M}+\text{Na}$ ]<sup>+</sup> 465.1876, found 465.1857.

### **2-Azido-N-benzyl-N-(4-(3-(pyridin-2-yl)-1,2,4-triazin-6-yl)phenyl)acetamide (10)**



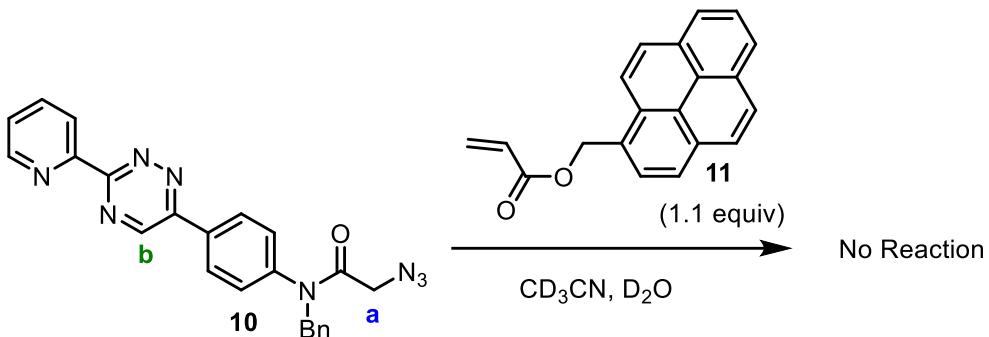
To a stirred solution of azido oxime **4d** (101 mg, 0.30 mmol) and hydrazine monohydrate (14.6  $\mu\text{L}$ , 0.30 mmol, 1.0 equiv) in ethanol (0.45 mL, 0.67 M) was added one drop of acetic acid, and the mixture was heated at 45 °C for 16 h. Then, concentration of the reaction mixture *in vacuo* gave crude hydrazone (110 mg) which was submitted to the next step without further purification.

To a stirred solution of crude hydrazone and 2-pyridine-carboxaldehyde (34.5  $\mu\text{L}$ , 0.359 mmol, 1.2 equiv) in ethanol (2.0 mL, 0.15 M) was added three drops of acetic acid. After 23 h, the reaction mixture was concentrated *in vacuo*. The obtained crude material was dissolved in acetic acid (0.50 mL, 0.60 M), and the stirred mixture was heated at 100 °C. After 1 h, the reaction mixture was diluted with water, and was quenched with saturated aqueous solution of sodium bicarbonate. The organic components were extracted three times with dichloromethane and washed with brine, and then was dried over sodium sulfate. Concentration of the collected organic layer *in vacuo* and purification by silica gel column chromatography (hexane/ethyl acetate = 1/1 to 1/2 with 2% methanol to dichloromethane/methanol = 20/1) gave 119 mg of **10** (94% for 3 steps).

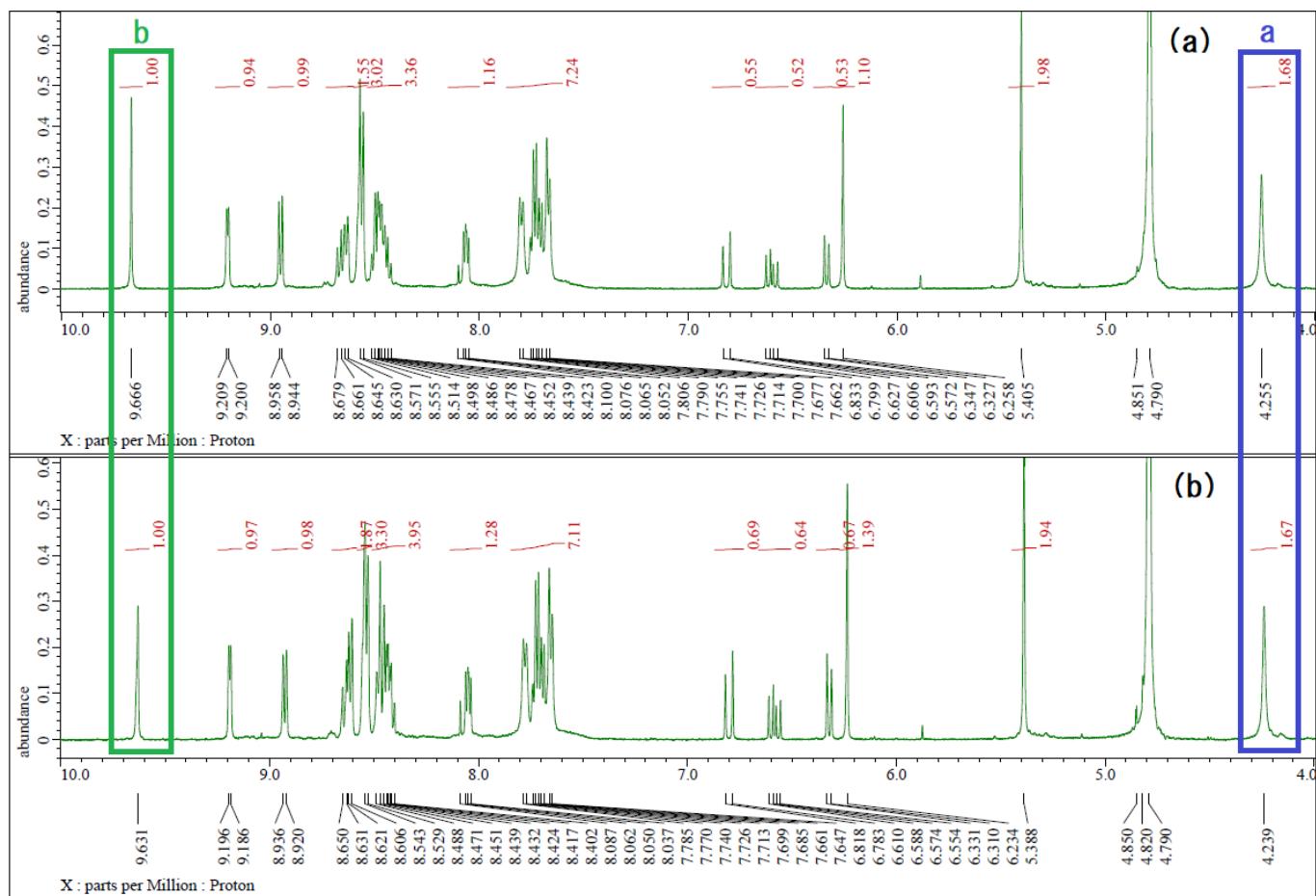
Brown amorphous solid;  $R_f$  value 0.13 (hexane/ethyl acetate = 1/1); IR (NaCl, neat)  $\nu_{\text{max}}$  2105, 1670, 1402, 1255  $\text{cm}^{-1}$ ;  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ )  $\delta$  9.17 (s, 1H), 8.90 (m, 1H), 8.74 (d, 1H,  $J$  = 8.0 Hz), 8.19 (ddd, 2H,  $J$  = 9.5, 2.5, 2.5 Hz), 7.96 (ddd, 1H,  $J$  = 8.0, 7.5, 2.0 Hz), 7.52–7.50 (m, 1H), 7.32–7.28 (m, 3H), 7.24–7.21 (m, 4H), 4.97 (s, 2H), 3.68 (s, 2H);  $^{13}\text{C}$  NMR (126 MHz,  $\text{CDCl}_3$ )  $\delta$  167.0, 162.0, 154.8, 152.0, 150.6, 146.9, 143.0, 137.3, 136.1, 133.4, 129.3, 129.0, 128.7, 128.5, 128.0, 125.9, 124.0, 53.4, 51.0; LRMS (EI, M =  $\text{C}_{23}\text{H}_{18}\text{N}_8\text{O}$ ) *m/z* 422 ( $\text{M}^+$ , 2%), 394 (12), 338 (23), 206 (26), 91 (100); HRMS (EI) calcd for  $\text{C}_{23}\text{H}_{18}\text{N}_8\text{O}$  ( $\text{M}^+$ ) 422.1604, found 422.1602.

## [4] Model Study on Chemoselectivity

- Azido triazine **10** with acrylate **11**

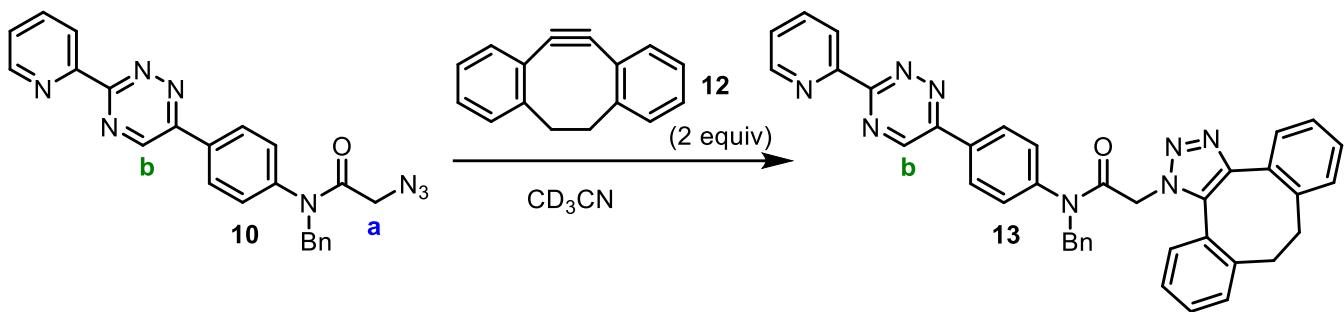


Azido triazine **10** (15.5 mg, 0.0375 mmol) was dissolved in acetonitrile-*d*<sub>3</sub>/D<sub>2</sub>O (0.75 mL, 1:1, 0.05 M). Then, 1-pyrenemethyl acrylate **11**<sup>5</sup> (11.2 mg, 0.041 mmol, 1.1 equiv) was added to the mixture. The reaction was monitored for 5 h by <sup>1</sup>H NMR (Figure S1).

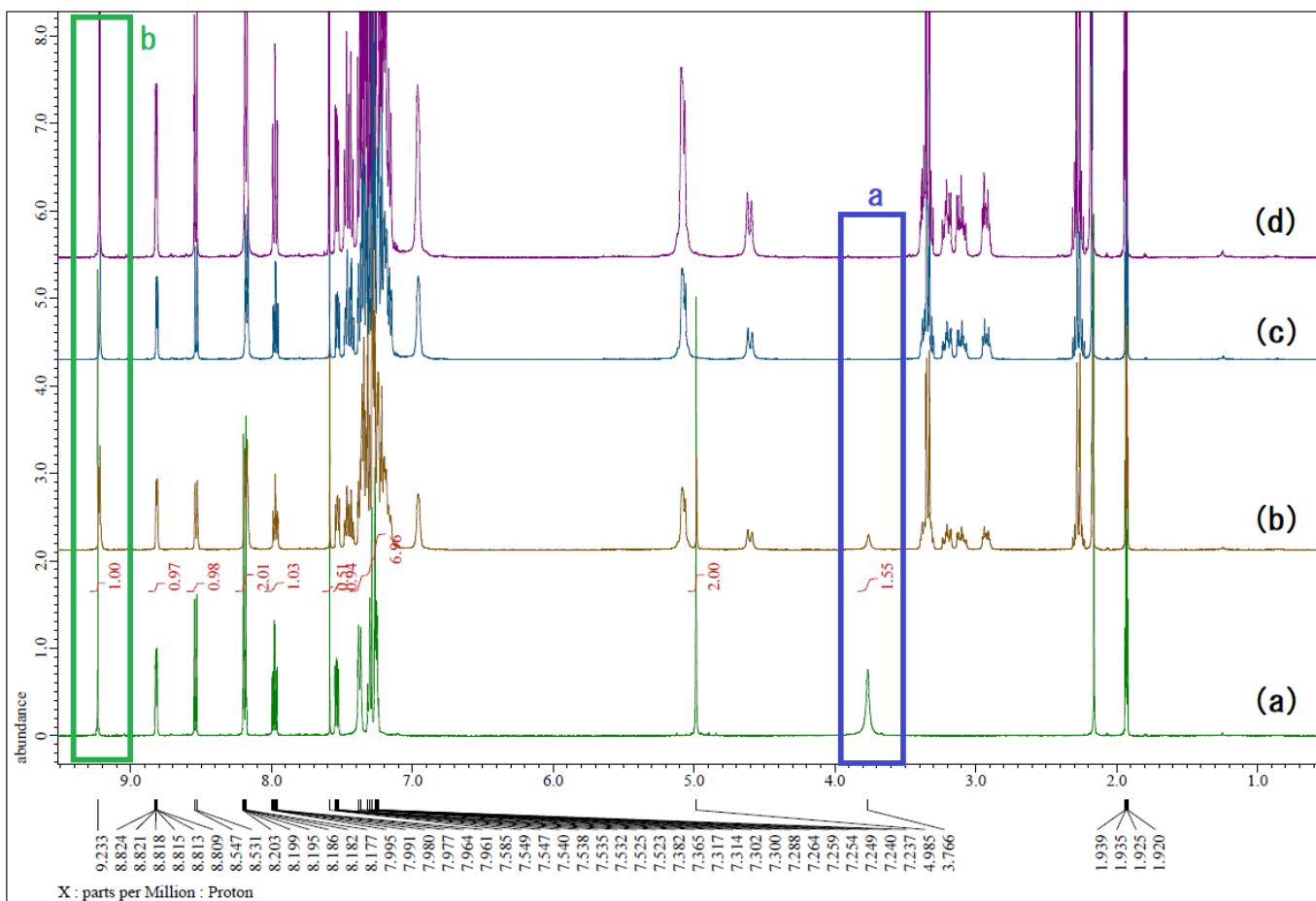


**Figure S1.** Comparison of NMR charts of the reaction (a) just after addition of **11** (0 h), and (b) 5 h after addition of **11** in CD<sub>3</sub>CN / D<sub>2</sub>O (1:1). Molecular ratio which is not identical to the reaction conditions is due to the insufficient solubility of the substrates in the aqueous solvent.

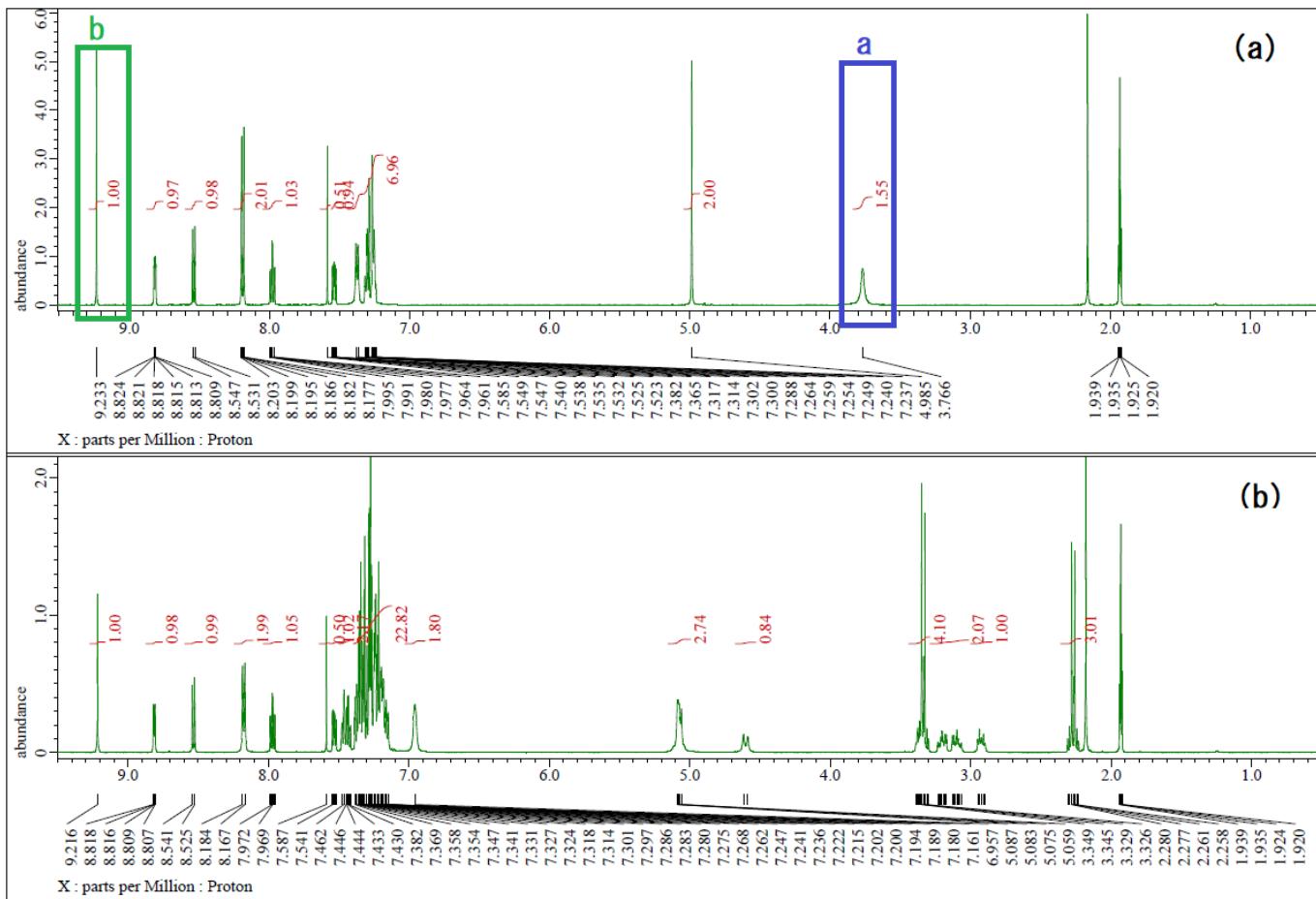
•Azido triazine **10** with dibenzocyclooctyne **12**



Azido triazine **10** (15.2 mg, 0.0375 mmol) was dissolved in acetonitrile- $d_3$  (0.75 mL, 0.05 M) in an NMR tube (Figures S2a and S3a). And then, dibenzocyclooctyne **12**<sup>6</sup> (15.3 mg, 0.075 mmol, 2.0 equiv) was added to the mixture (Figure S1b). The reaction was monitored the disappearance of the peaks of the protons on each click functional groups (protons **a** and **b**) by  $^1\text{H}$  NMR after 40 min (Figure S2c) and 7 h (Figures S2d, and S3b).

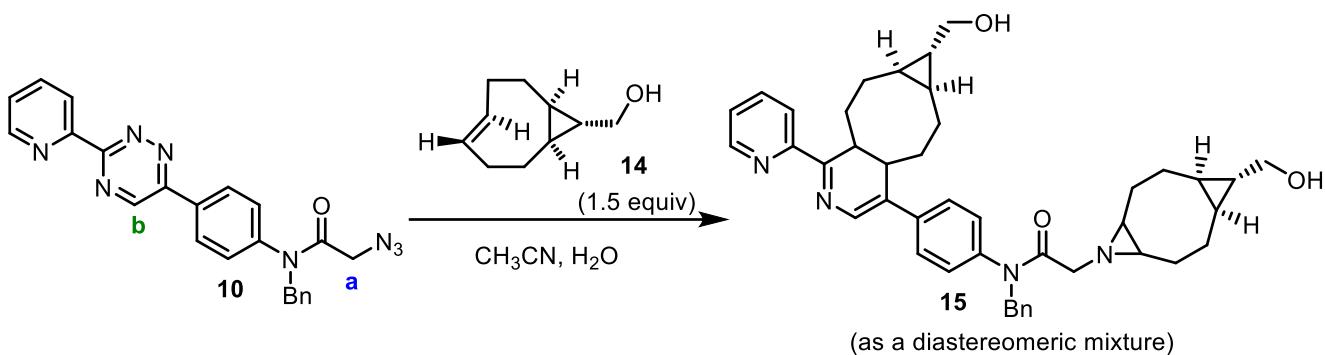


**Figure S2.**  $^1\text{H}$  NMR experiments (a) **10** in  $\text{CD}_3\text{CN}$ , (b) soon after addition of **11**, (c) after 40 min, and (d) after 7 h.



**Figure S3.** Comparison of NMR charts of the reaction (a) before addition of **12**, and (b) 7 h after addition of **12**.

•Azido triazine **10** with cyclopropane-fused *trans*-cyclooctene **14**



To a solution of azido triazine **10** (20.8 mg, 0.05 mmol) in acetonitrile/water (1.0 mL, 1/1 vol., 0.05 M) was added *trans*-cyclooctene **14**<sup>7</sup> (11.1 mg, 0.075 mmol, 1.5 equiv) at room temperature. After 30 min, concentration and purification by silica gel column chromatography (hexane/ethyl acetate = 1/1 to dichloromethane to dichloromethane/methanol = 20/1) followed by GPC to afford 12.5 mg of **15** (50% based on **14**) as stereomixture. Loss of triazine and azido structures were confirmed by  $^1\text{H}$  NMR (Figure S4), and IR spectra as well as HRMS.

Compound **15**: Pale yellow amorphous solid;  $R_f$  value 0.17 (dichloromethane/methanol = 15/1); IR (NaCl, neat)  $\nu_{\text{max}}$  2925, 2856, 1662, 1509  $\text{cm}^{-1}$ ; HRMS (ESI) calcd for  $\text{C}_{23}\text{H}_{18}\text{N}_8\text{O}$  [ $\text{M}+\text{H}]^+$  671.3961, found 671.3910.

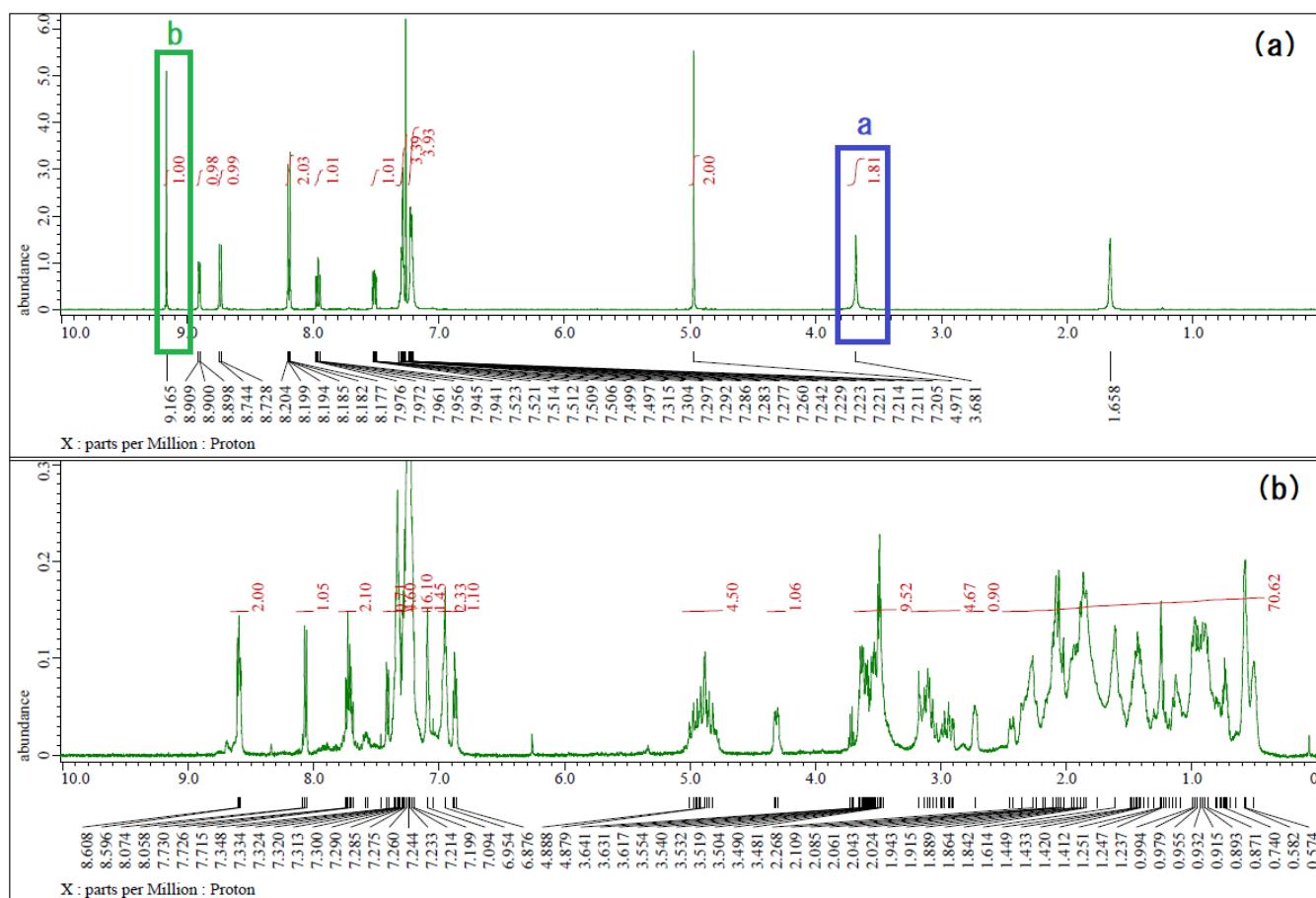
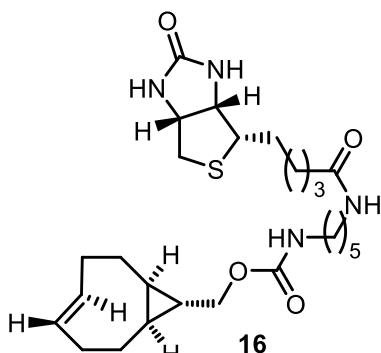


Figure S4. Comparison of NMR charts of (a) starting material **10**, and (b) addition product **15** as a diastereomeric mixture in  $\text{CDCl}_3$ .

## [5] One-pot sequential triple click conjugation reaction

### Biotin-conjugated *trans*-cyclooctene (16)



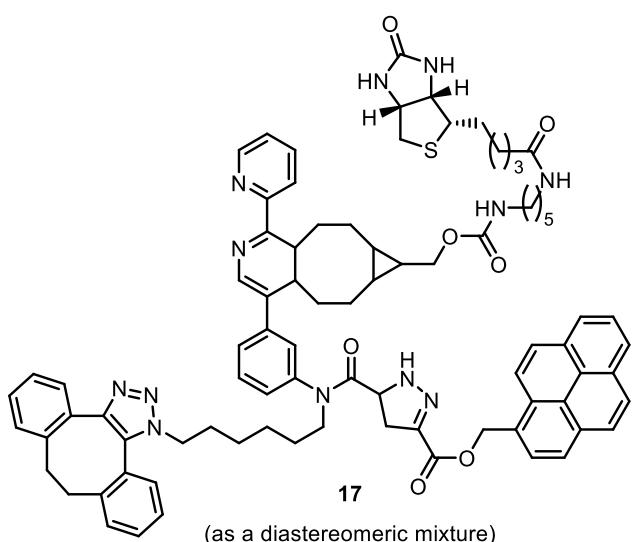
(as a diastereomeric mixture  
at *trans*-cyclooctene position)

To a solution of 5-((+)-biotinamido)pentylamine trifluoroacetic acid salt<sup>8</sup> (264 mg, 0.596 mmol) and racemic (*rel*-1*R*,8*S*,9*R*,4*E*)-bicyclo[6.1.0]non-4-ene-9-ylmethyl (4-nitrophenyl) carbonate<sup>7</sup> (227 mg, 0.716 mmol, 1.2 equiv) in DMF (60 mL, 0.01 M) was added triethylamine (0.25 mL, 1.79 mmol, 3.0 equiv) at room temperature. After 24 h, to the mixture was added triethylamine (0.25 mL, 1.79 mmol, 3.0 equiv) again. After 24 h, the resulting mixture was concentrated under reduced pressure. The residue was purified by silica gel column chromatography (dichloromethane elution to dichloromethane/methanol = 15/1) gave **16** (157 mg, 0.310 mmol, 52%). Although residual nitrophenol was inseparable even after GPC purification, which also

ended large loss of the product (down to ca. 5%), the obtained material was used in the next reaction without further purification.

White amorphous solid; R<sub>f</sub> value 0.17 (dichloromethane/methanol = 10/1); IR (NaCl, neat) ν<sub>max</sub> 3301, 2926, 2854, 1702, 1642, 1545, 1267 cm<sup>-1</sup>; <sup>1</sup>H NMR (500 MHz, CDCl<sub>3</sub>) δ 6.49–6.41 (br, 1H), 6.22 (br, 1H), 5.85 (ddd, 1H, J = 17, 8.0, 6.5 Hz), 5.62–5.55 (br, 1H), 5.12 (ddd, 1H, J = 14.0, 10.0, 3.5 Hz), 4.88 (br, 1H), 4.51 (dd, 1H, J = 7.0, 5.0 Hz), 4.31 (dd, 1H, J = 4.5, 2.0 Hz), 3.91 (d, 2H, J = 8.0 Hz), 3.22–3.20 (m, 2H), 3.15 (m, 4H), 2.90 (dd, 1H, J = 8.0, 5.0 Hz), 2.73 (d, 1H, J = 13.0 Hz), 2.35 (d, 1H, J = 14.0 Hz), 2.27–2.24 (m, 2H), 2.21–2.18 (m, 2H), 1.95–1.87 (m, 2H), 1.73–1.65 (m, 4H), 1.52–1.41 (m, 6H), 1.34 (m, 2H), 0.87–0.81 (m, 2H), 0.55–0.51 (m, 2H), 0.41 (m, 2H); <sup>13</sup>C NMR (126 MHz, CDCl<sub>3</sub>) δ 173.2, 163.9, 157.0, 138.3, 131.2, 70.5, 69.4 (assigned to diastereomer), 61.8, 60.2, 55.6, 40.5, 39.2, 38.7, 35.9, 33.8, 32.6, 29.7, 29.6, 29.0, 28.1, 28.0, 27.6, 25.6, 24.7, 23.8, 21.9, 20.9; HRMS (ESI) calcd for C<sub>22</sub>H<sub>23</sub>N<sub>11</sub>ONa [M+Na]<sup>+</sup> 529.2824, found 529.2821.

### Four-component coupling product (17)

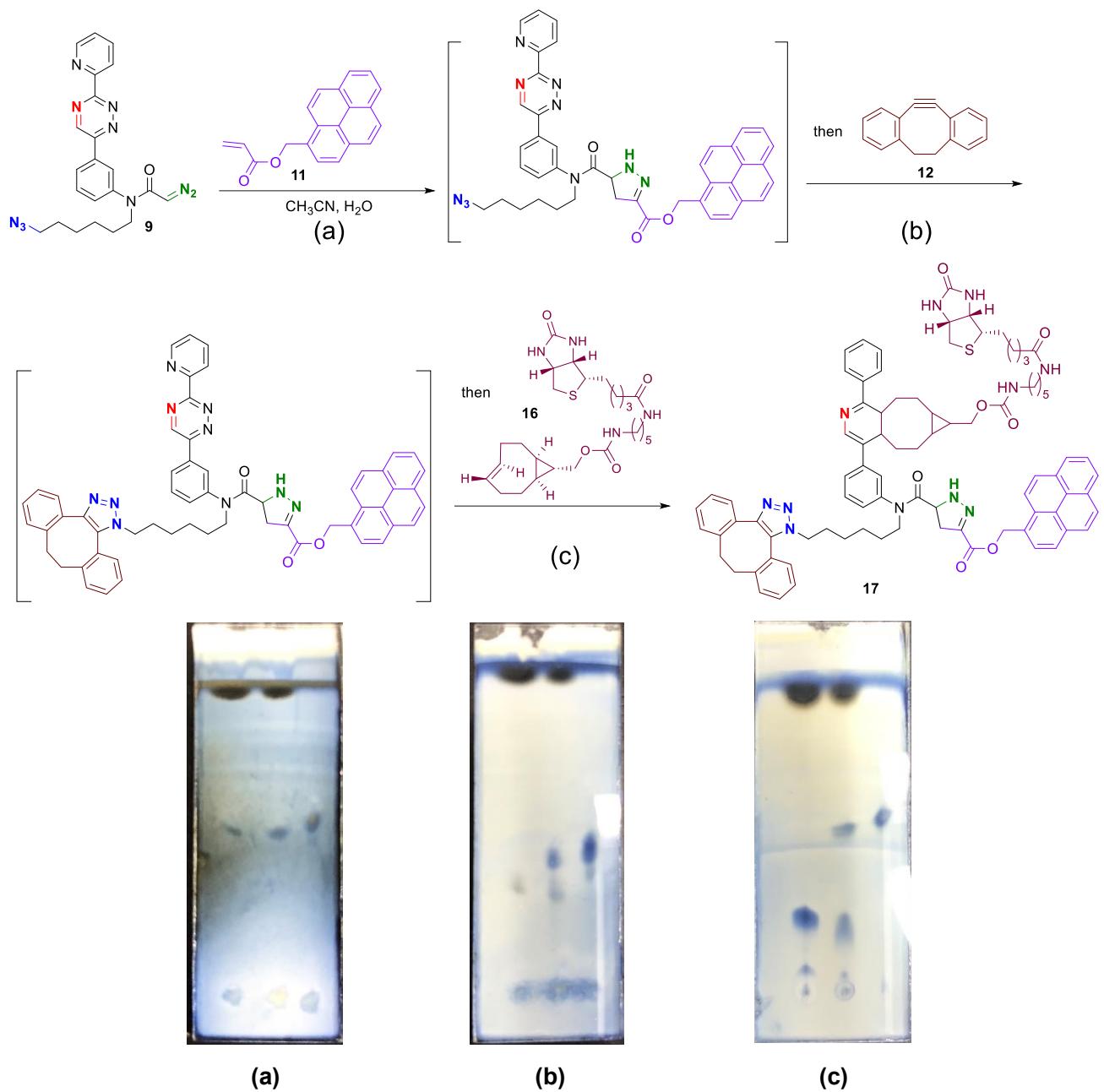


(as a diastereomeric mixture)

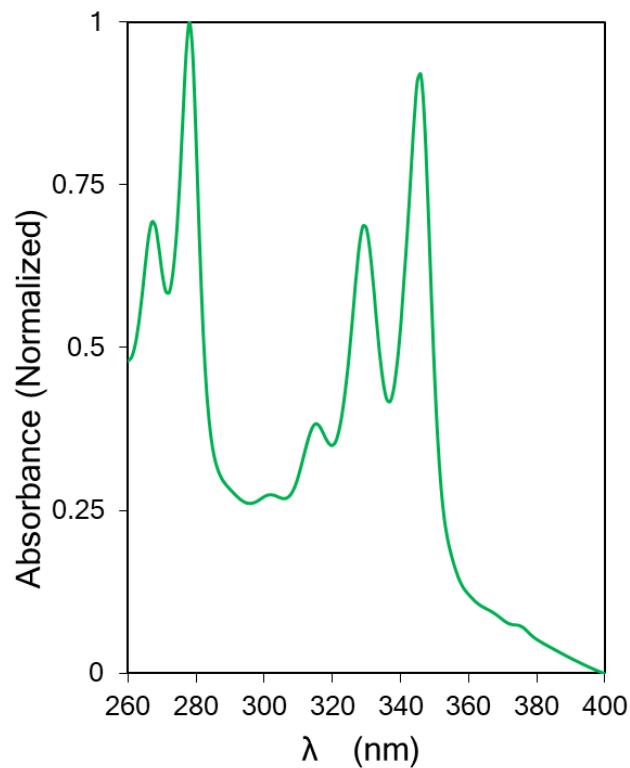
To a solution of azido-diazo-triazine **9** (16.4 mg, 0.0361 mmol) in acetonitrile-water (2.9 mL, 1:1, 0.013 M) was added 1-pyrenemethyl acrylate **11**<sup>5</sup> (52.4 mg, 0.181 mmol, 5.0 equiv) at room temperature. After 24 h, dibenzocyclooctyne **12**<sup>6</sup> (9.1 mg, 0.0433 mmol, 1.2 equiv) was added to the resulting mixture at same temperature. Then after 2 h, *trans*-alkene-biotin tag **16** (18.3 mg, 0.0361 mmol, 1.0 equiv) dissolved in acetonitrile-water (1.0 mL + rinsed with 0.5 mL×2, 1:1) at same temperature. After 1 h, the solvent was removed under reduced pressure followed by silica gel column chromatography (hexane/ethyl acetate = 1/1 to dichloromethane elution to dichloromethane/methanol = 20/1 to 15/1 to 10/1)

gave **17** (22.1 mg, 0.0157 mmol, 44%) as diastereomeric mixture.

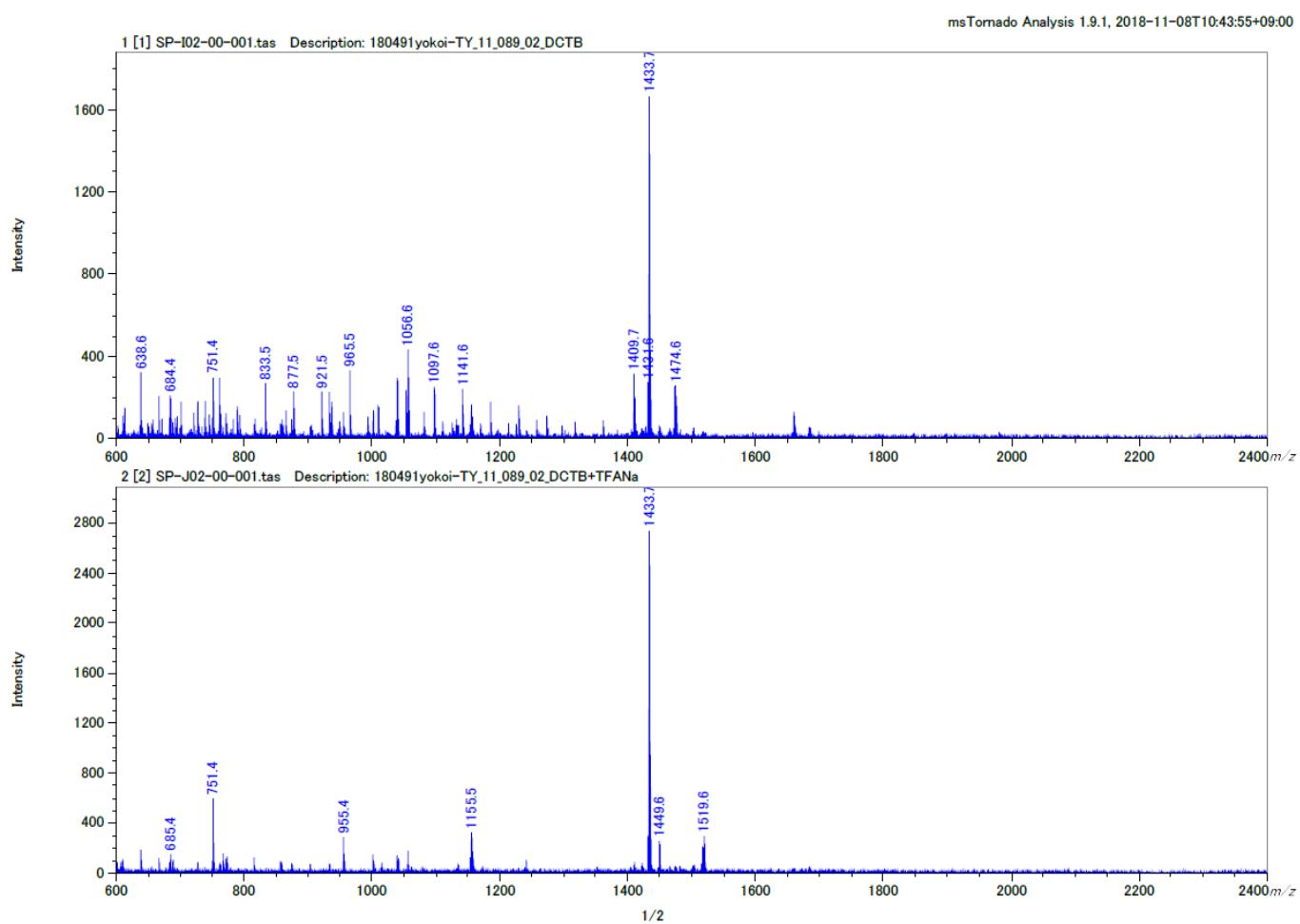
Yellow amorphous solid;  $R_f$  value 0.3 (dichloromethane/methanol = 10/1); IR (NaCl, neat)  $\nu_{\text{max}}$  3309, 2930, 1699, 1654, 1460, 1245  $\text{cm}^{-1}$ ; UV-vis ( $\text{CHCl}_3$ )  $\lambda_{\text{max}} = 267, 278, 302(\text{sh}), 315, 329, 346, 376(\text{sh})$ ;  $^1\text{H}$  NMR (500 MHz,  $\text{CDCl}_3$ ) See NMR spectra section; HRMS (MALDI-TOF) calcd for  $\text{C}_{84}\text{H}_{90}\text{N}_{12}\text{O}_7\text{SNa}$  [ $\text{M}+\text{Na}]^+$  1433.6668, found 1433.6659.



**Figure S5.** TLC analysis of the progress of four-component coupling reaction (left: reaction mixture, middle: reaction mixture + starting material scaffold 9; right: reference sample of 9) (a) 24 h after addition of acrylate 11 [(dichloromethane/methanol = 20/1) $\times$ 2], (b) 2 h after addition of dibenzocyclooctyne 12 [(dichloromethane/methanol = 20/1) $\times$ 2], (c) 1 h after addition of biotin-connected *trans*-cyclooctene 16 (dichloromethane/methanol = 10/1).



**Figure S6.** UV-vis spectrum of **17** (0.019  $\mu\text{M}$  in chloroform)



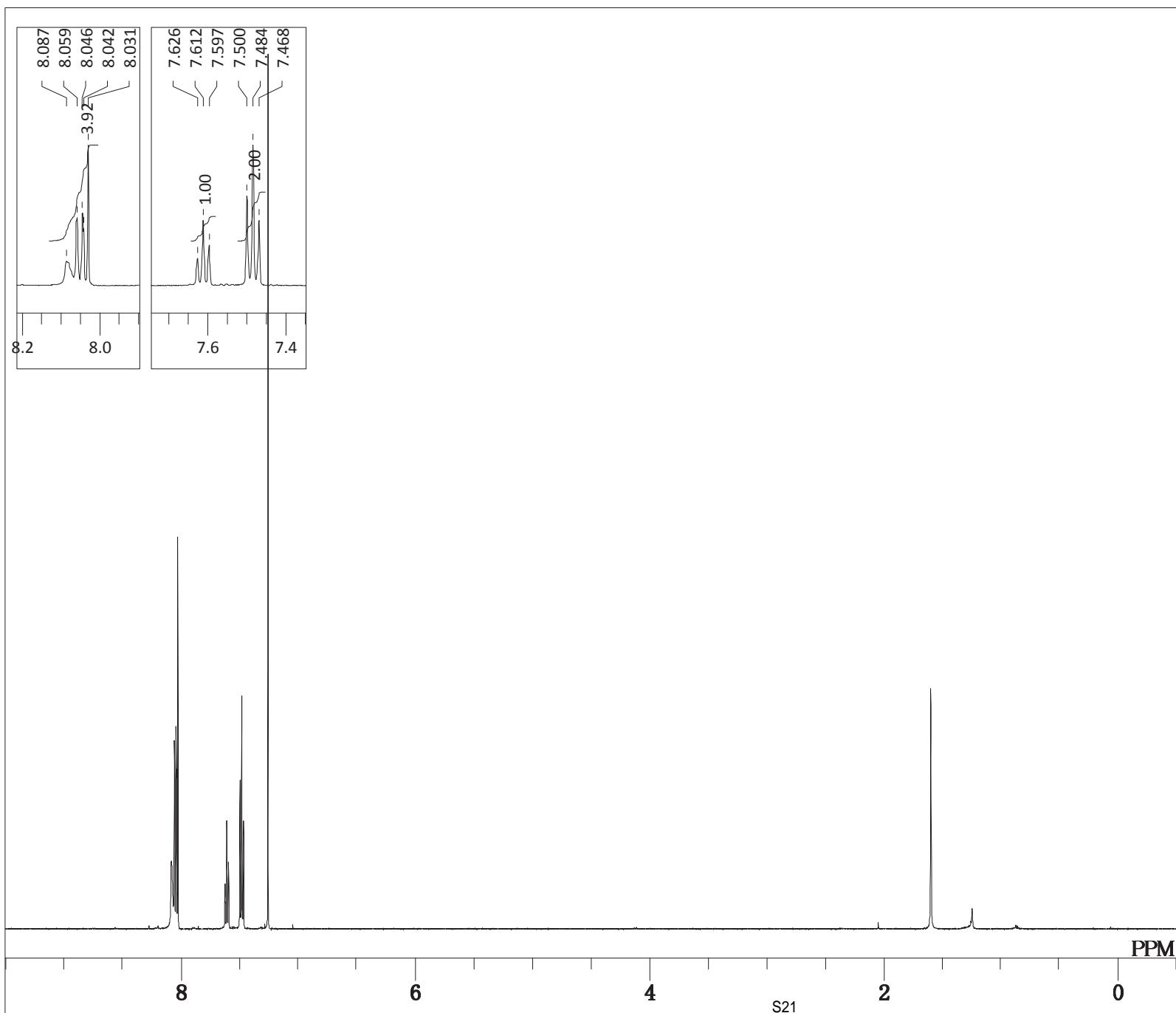
**Figure S7.** Mass spectra of **17** by MALDI-Spiral TOFMS: (top) with DCTB (*trans*-2-[3-(4-*tert*-butylphenyl)-2-methyl-2-propenylidene]malononitrile), (bottom) with DCTB + sodium trifluoroacetate.

## [6] References

- (1) R. E. Conrow and W. D. Dean, *Org. Proc. Res. Dev.*, 2008, **12**, 1285.
- (2) (a) H. Sun and S. G. DiMagno, *J. Am. Chem. Soc.*, 2005, **127**, 2050; (b) R. I. Hogrefe, A. P. McCaffrey, L. U. Borozdina, E. S. McCampbell, and M. M. Vaghefi, *Nucleic Acids Res.*, 1993, **21**, 4739.
- (3) T. Yokoi, H. Tanimoto, T. Ueda, T. Morimoto, and K. Kakiuchi, *J. Org. Chem.*, 2018, **83**, 12103.
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- (5) S. Amemori, K. Kokado, and K. Sada, *Angew. Chem. Int. Ed.* 2013, **52**, 4174.
- (6) (a) S. Yoshida, F. Karaki, K. Uchida, and T. Hosoya, *Chem. Commun.*, 2015, **51**, 8745; (b) C. S. McKay, J. Moran, and J. P. Pezacki, *Chem. Commun.*, 2010, **46**, 931.
- (7) M. Royzen, G. P. A. Yap, and J. M. Fox, *J. Am. Chem. Soc.*, 2008, **130**, 3760.
- (8) H. Xu, H. Sabit, G. L. Amidon, and H. D. H. Showalter, *Beilstein J. Org. Chem.* 2013, **9**, 89.

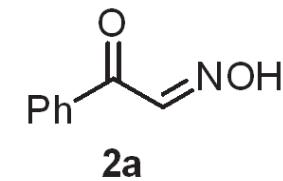
## [7] $^1\text{H}$ , and $^{13}\text{C}$ NMR Spectra

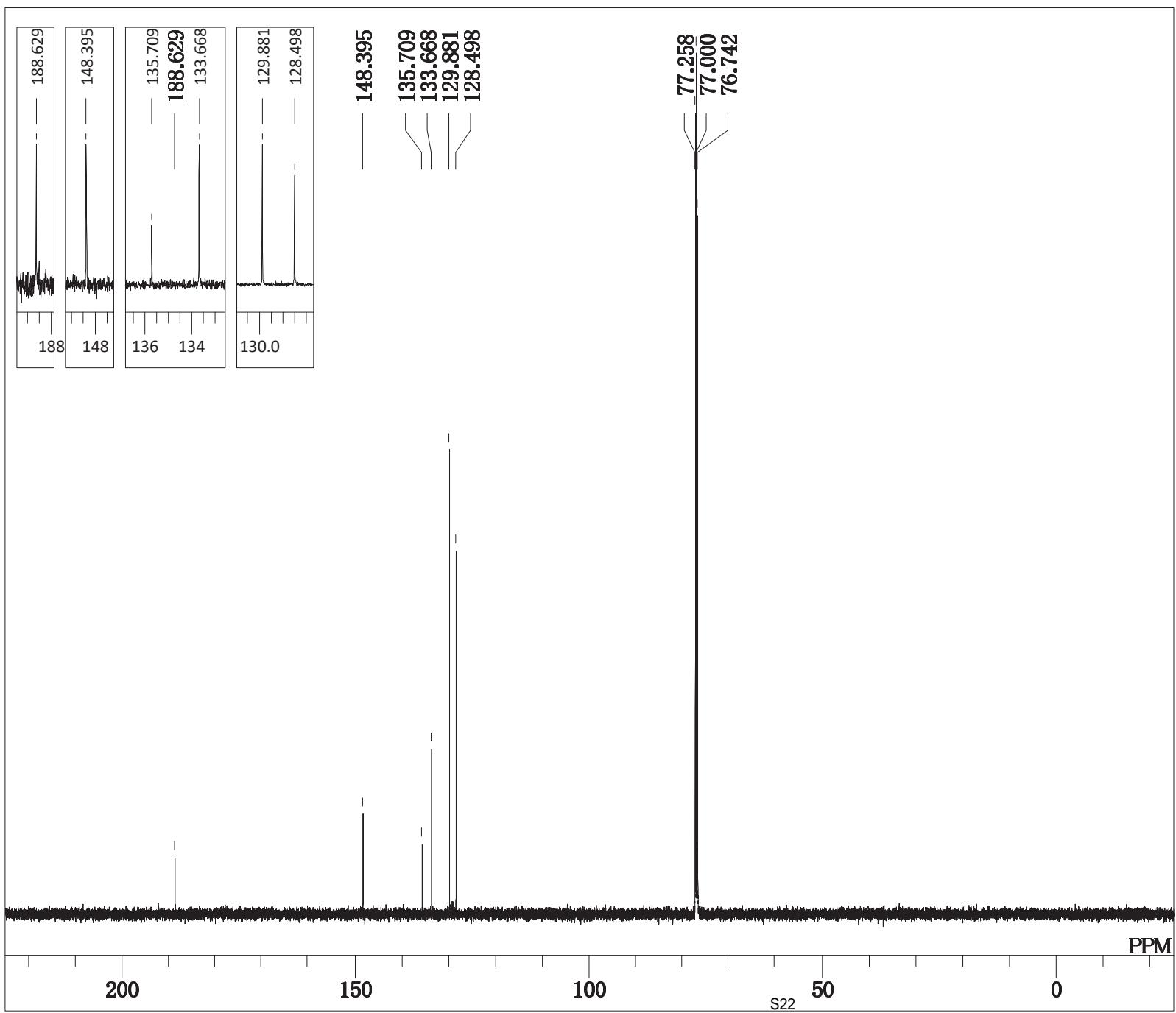
See next pages.



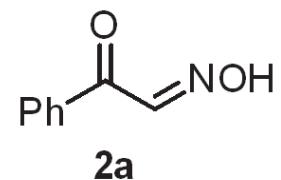
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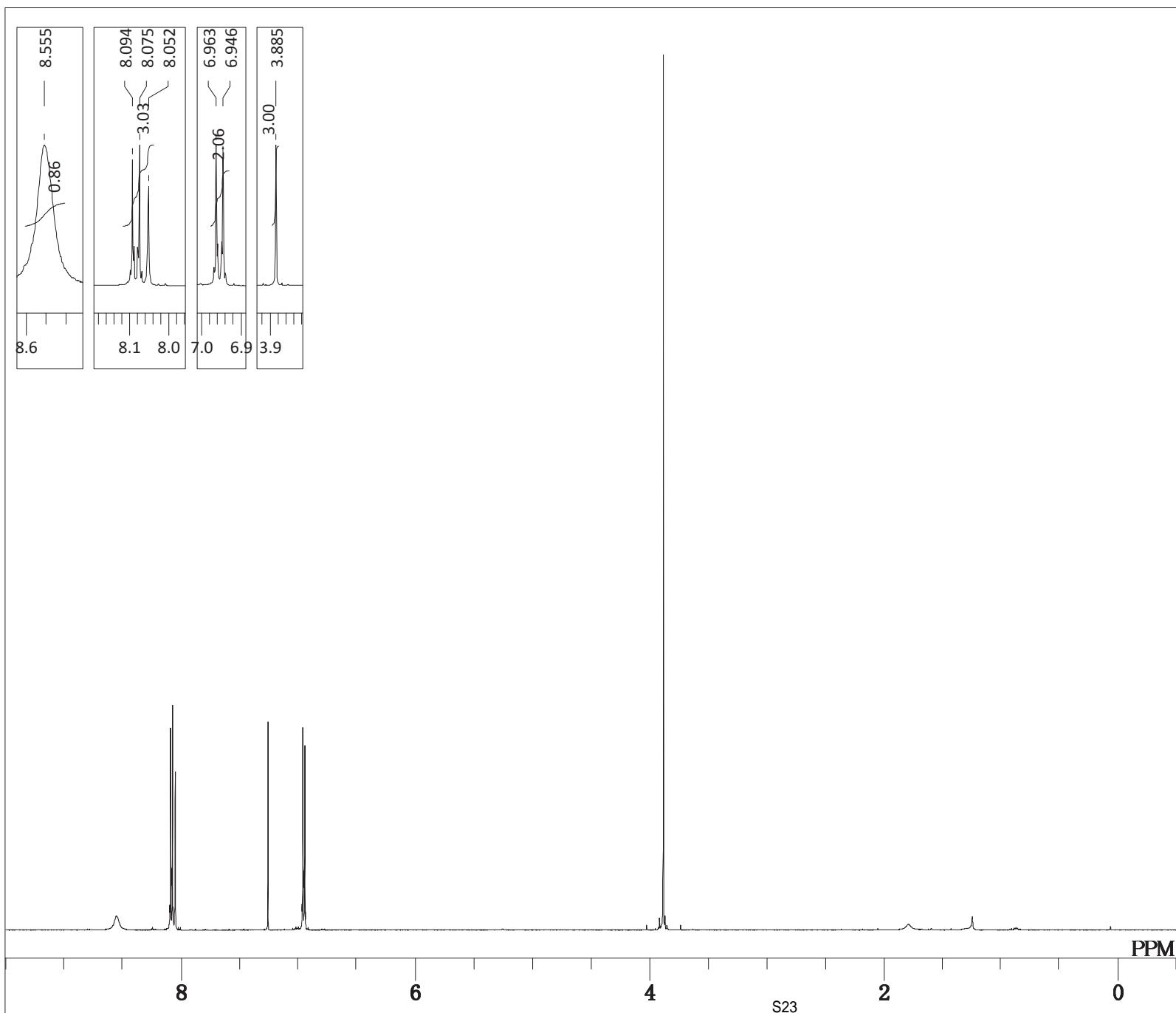
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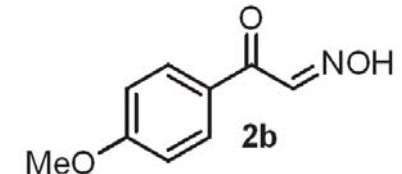
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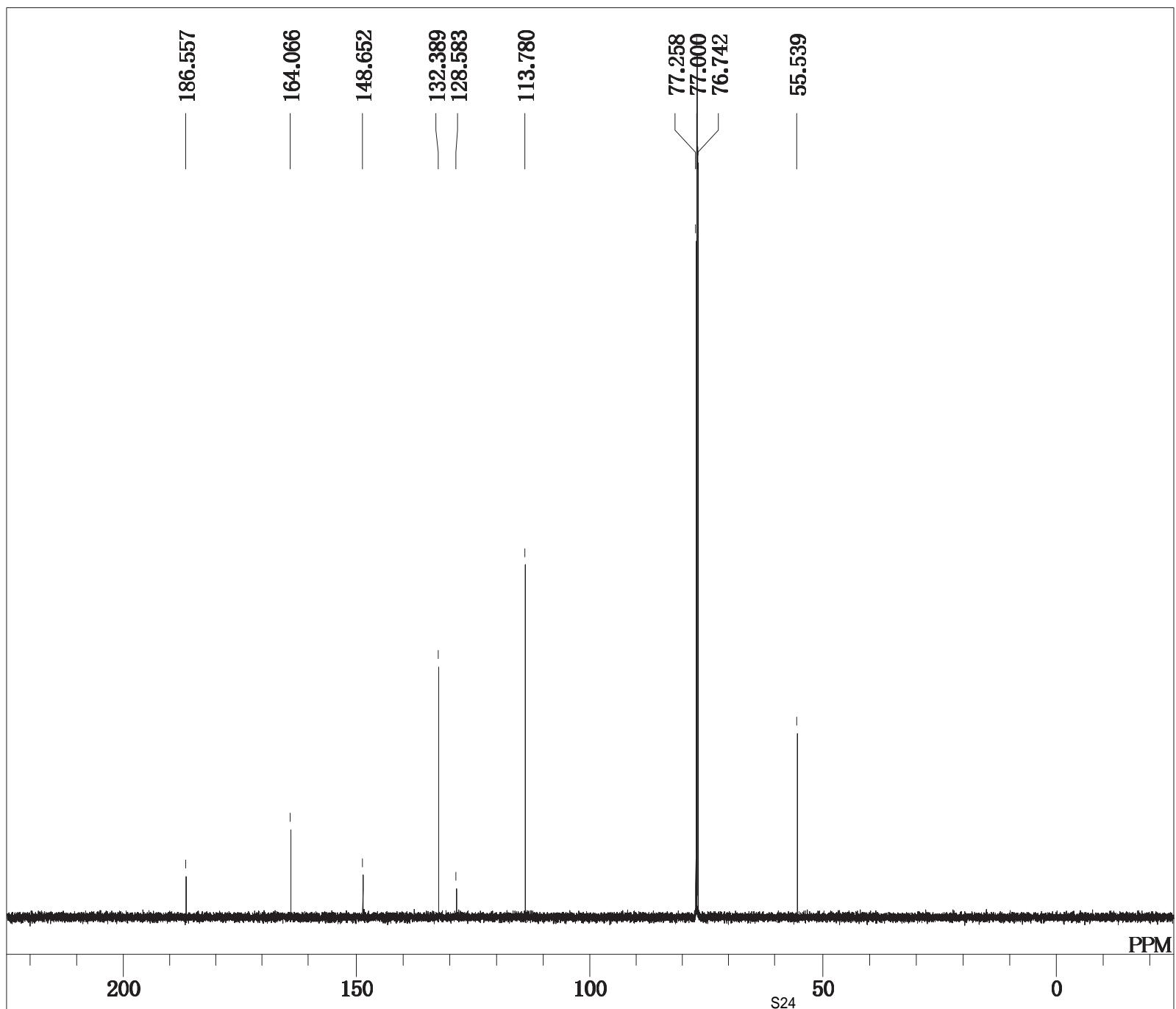




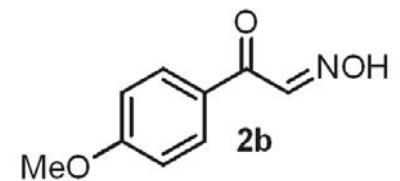
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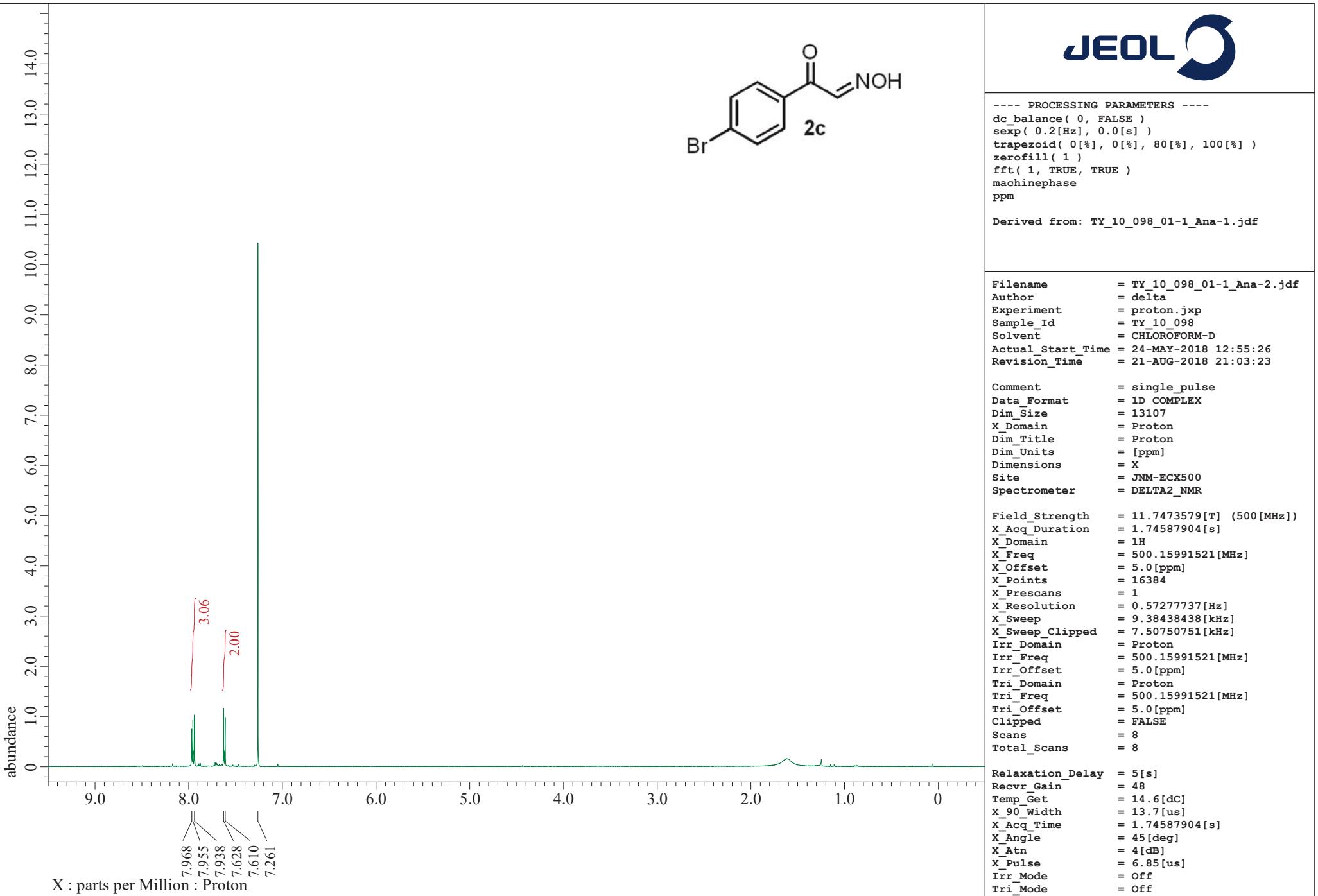
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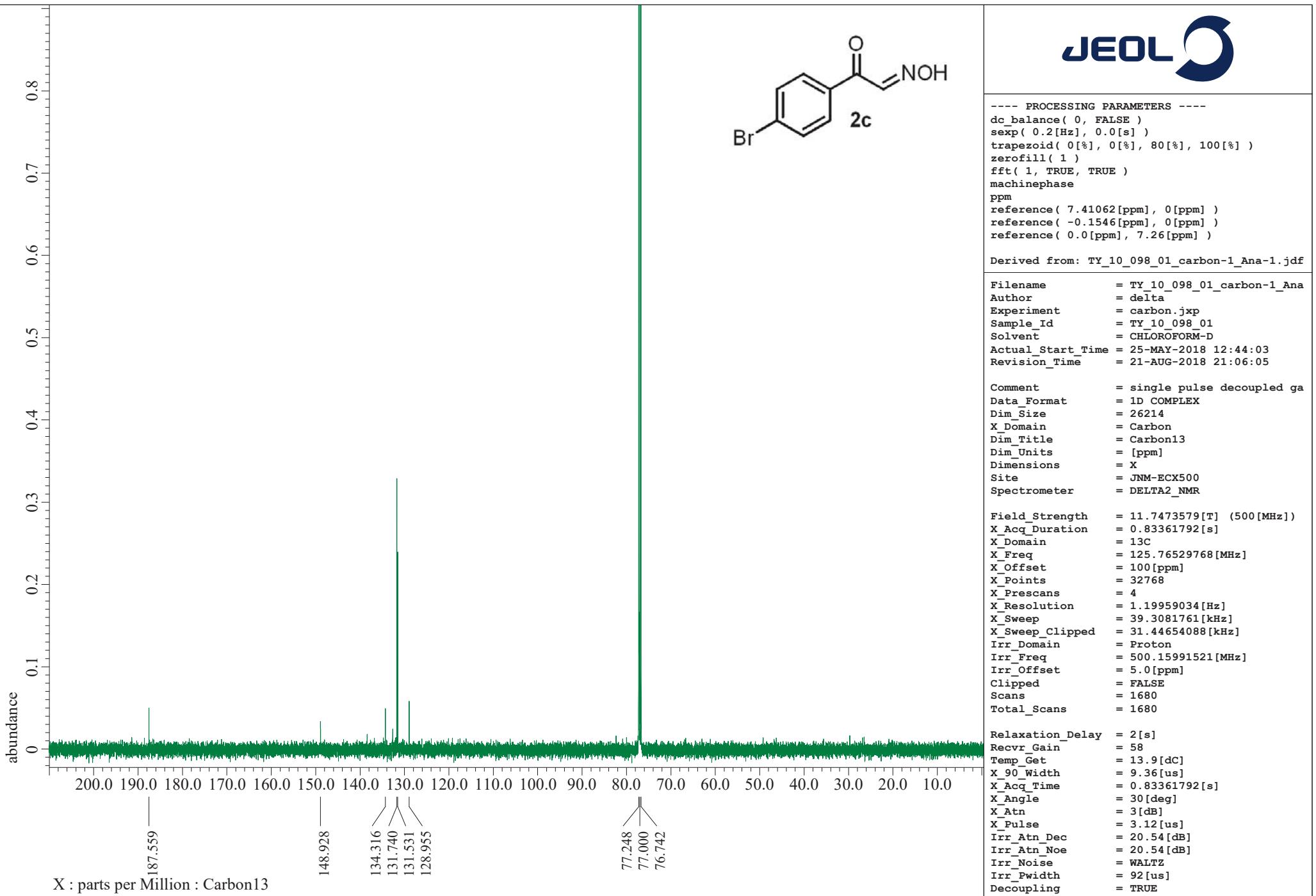


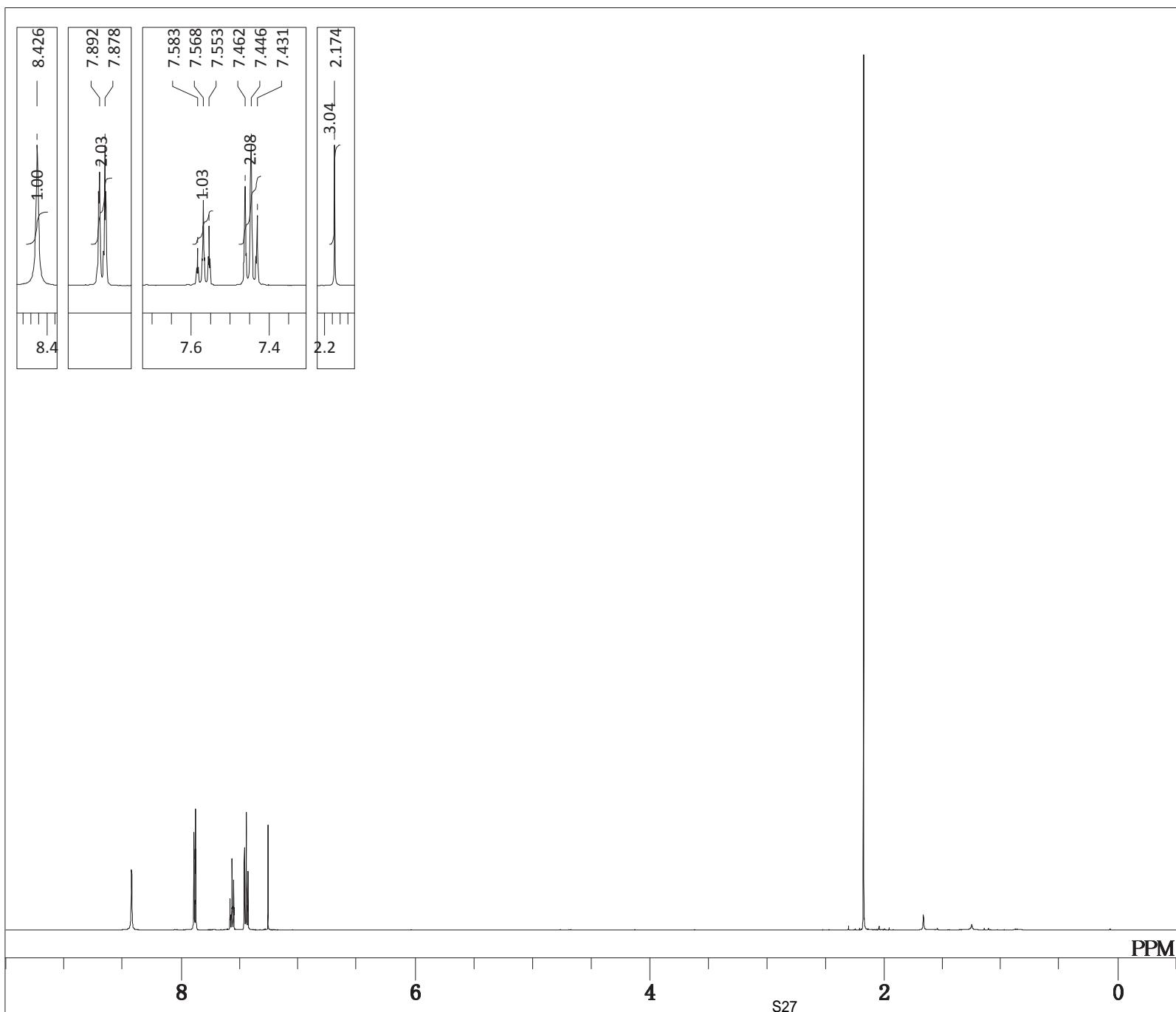


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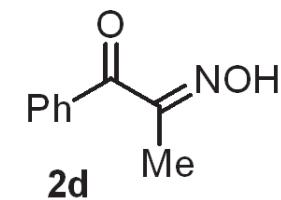


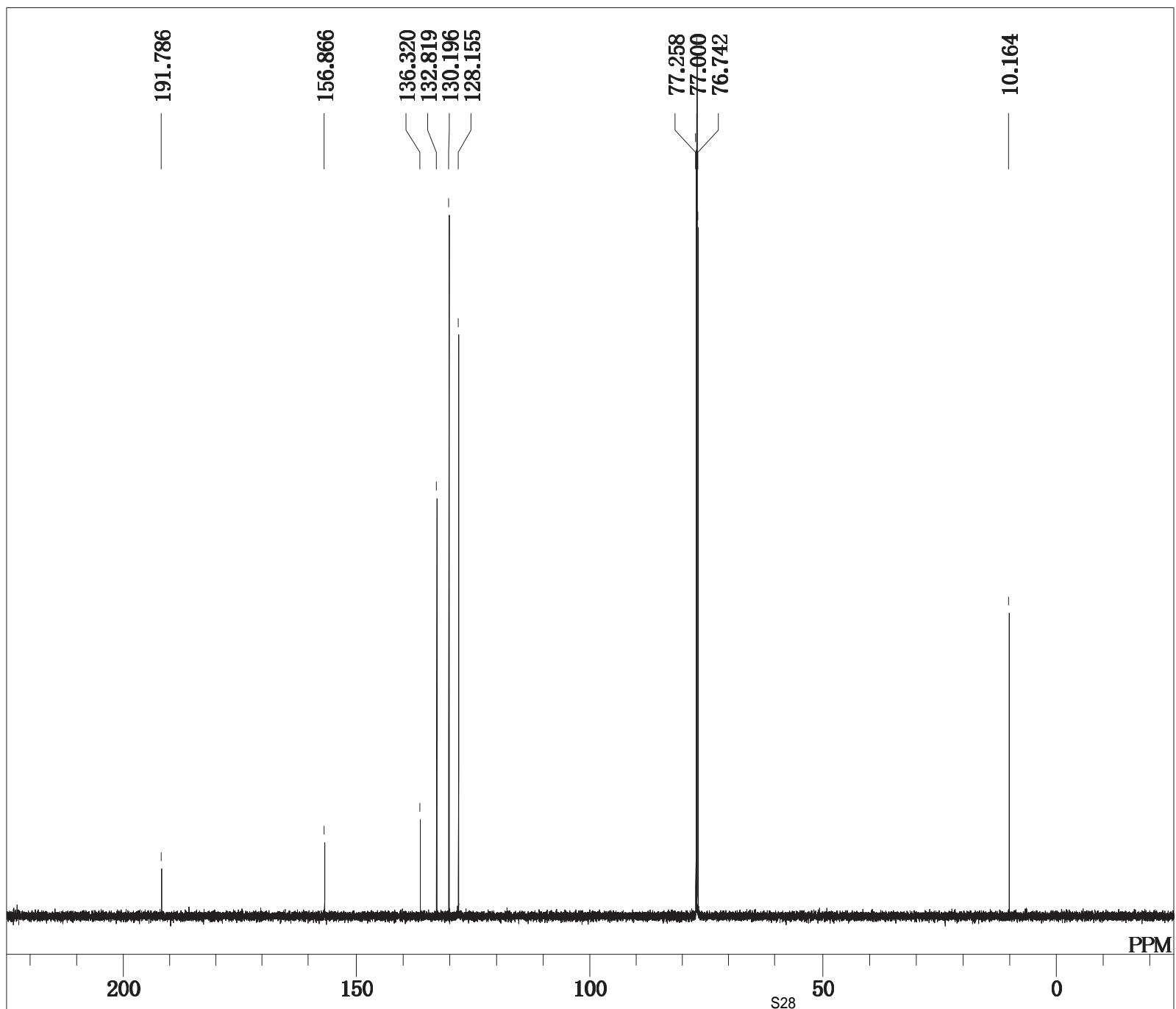




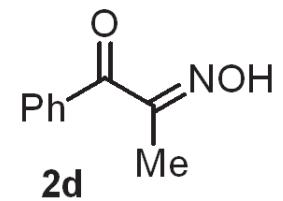


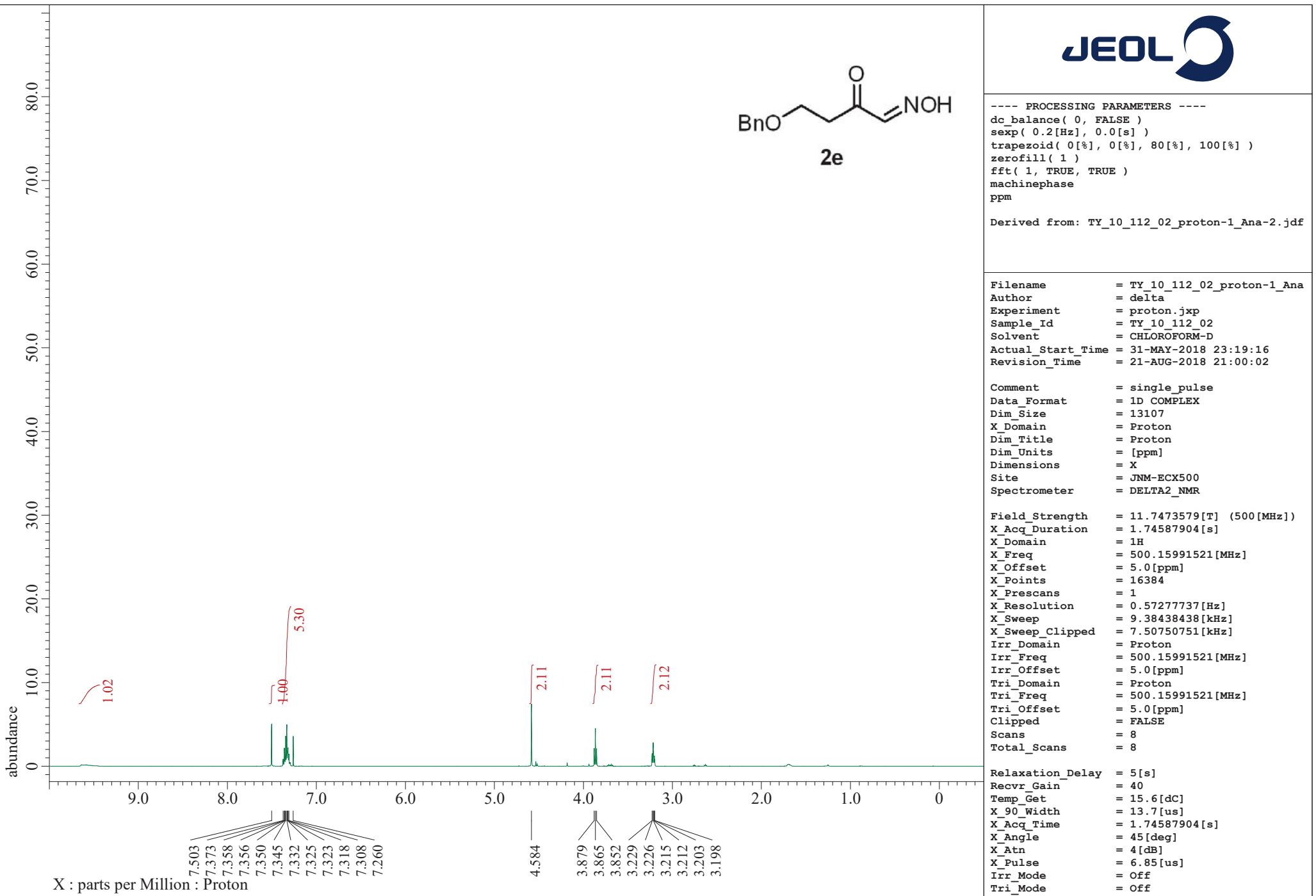
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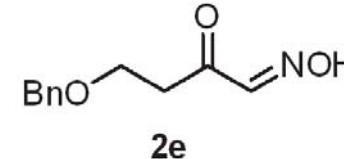
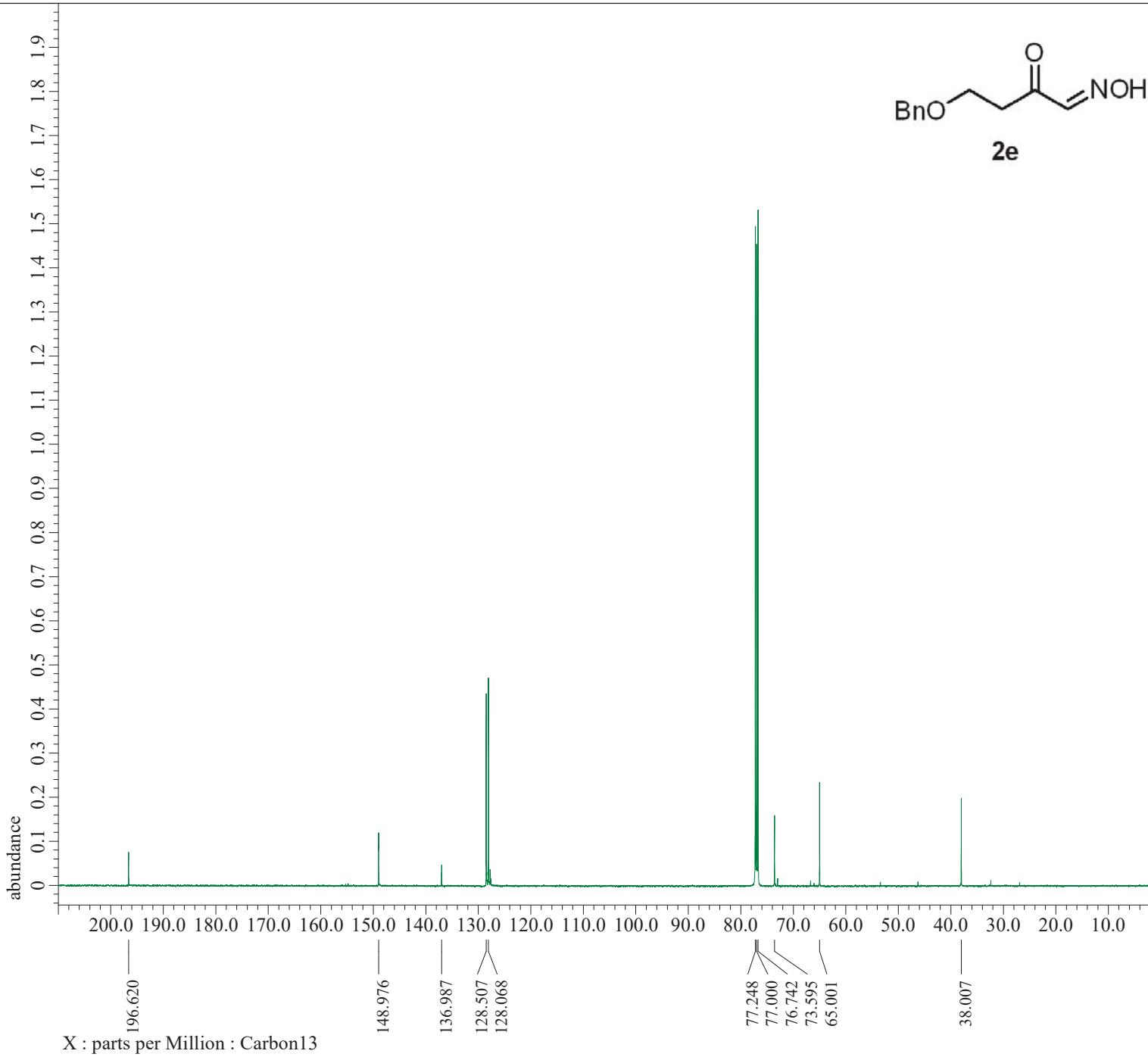




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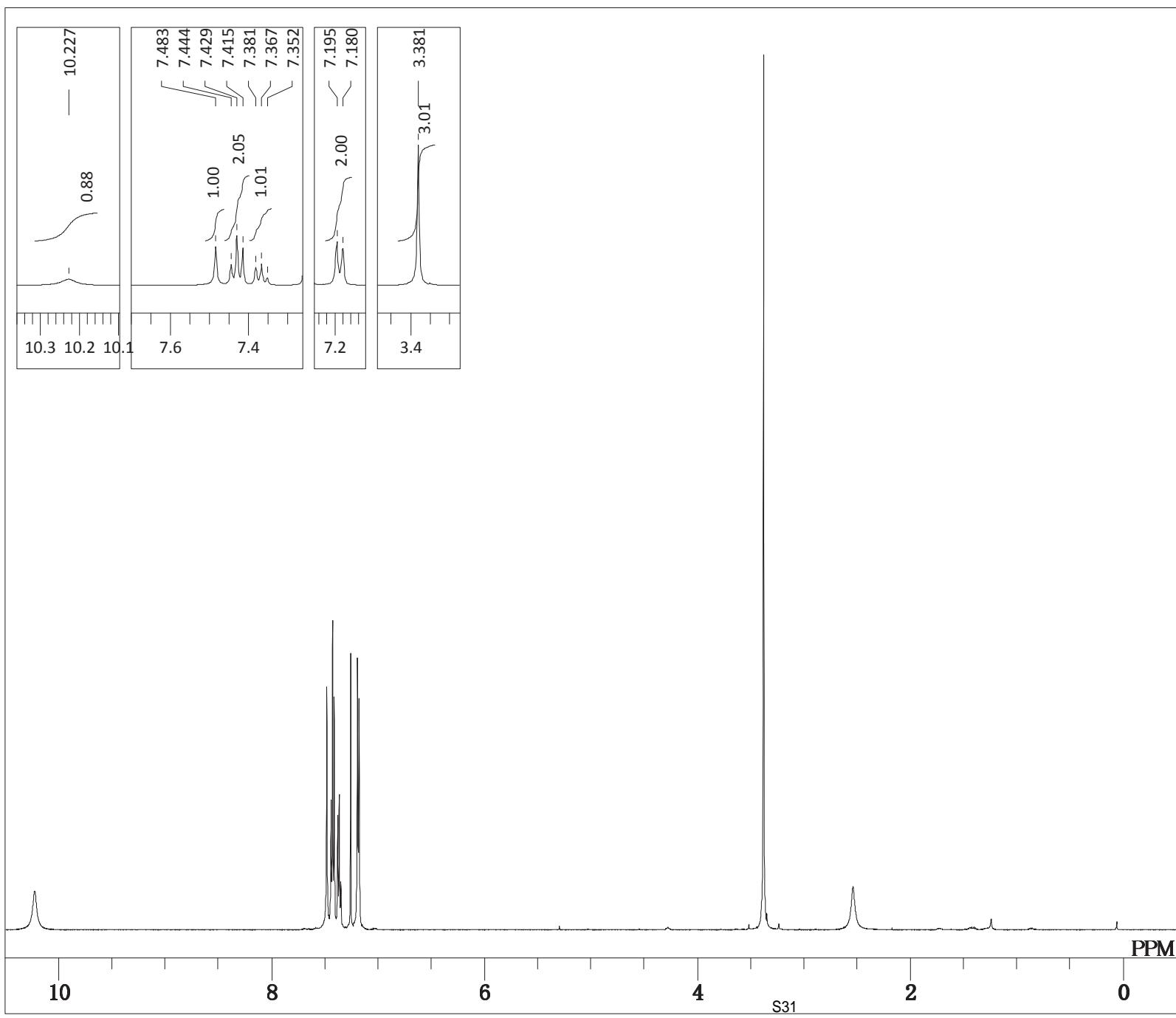




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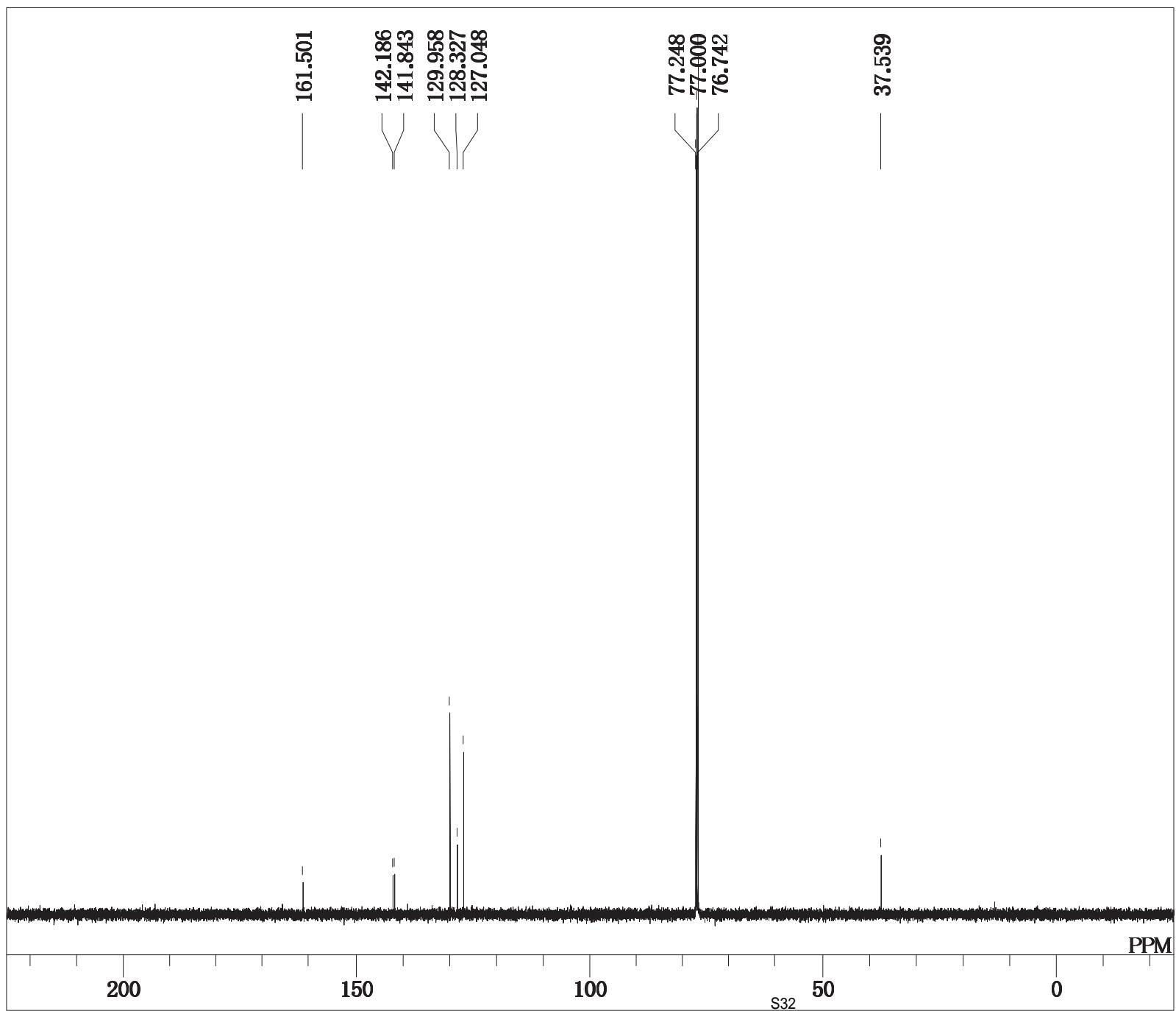
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X\_Domain = Carbon  
Dim\_Title = Carbon13  
Dim\_Units = [ppm]  
Dimensions = X  
Site = JNM-ECX500  
Spectrometer = DELTA2\_NMR  
Field\_Strength = 11.7473579[T] (500[MHz])  
X\_Acq\_Duration = 0.83361792[s]  
X\_Domain = 13C  
X\_Freq = 125.76529768[MHz]  
X\_Offset = 100[ppm]  
X\_Points = 32768  
X\_Prescans = 4  
X\_Resolution = 1.19959034[Hz]  
X\_Sweep = 39.3081761[kHz]  
X\_Sweep\_Clipped = 31.44654088[kHz]  
Irr\_Domain = Proton  
Irr\_Freq = 500.15991521[MHz]  
Irr\_Offset = 5.0[ppm]  
Clipped = FALSE  
Scans = 9620  
Total\_Scans = 9620  
Relaxation\_Delay = 2[s]  
Recv\_Gain = 58  
Temp\_Get = 16.8[dC]  
X\_90\_Width = 9.36[us]  
X\_Acq\_Time = 0.83361792[s]  
X\_Angle = 30[deg]  
X\_Atn = 3[dB]  
X\_Pulse = 3.12[us]  
Irr\_Atn\_Dec = 20.54[dB]  
Irr\_Atn\_Noe = 20.54[dB]  
Irr\_Noise = WALTZ  
Irr\_Pwidth = 92[us]  
Decoupling = TRUE

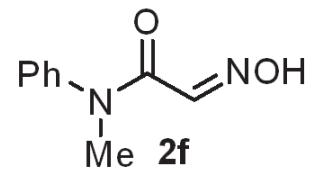


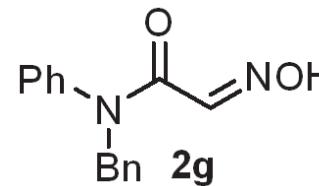
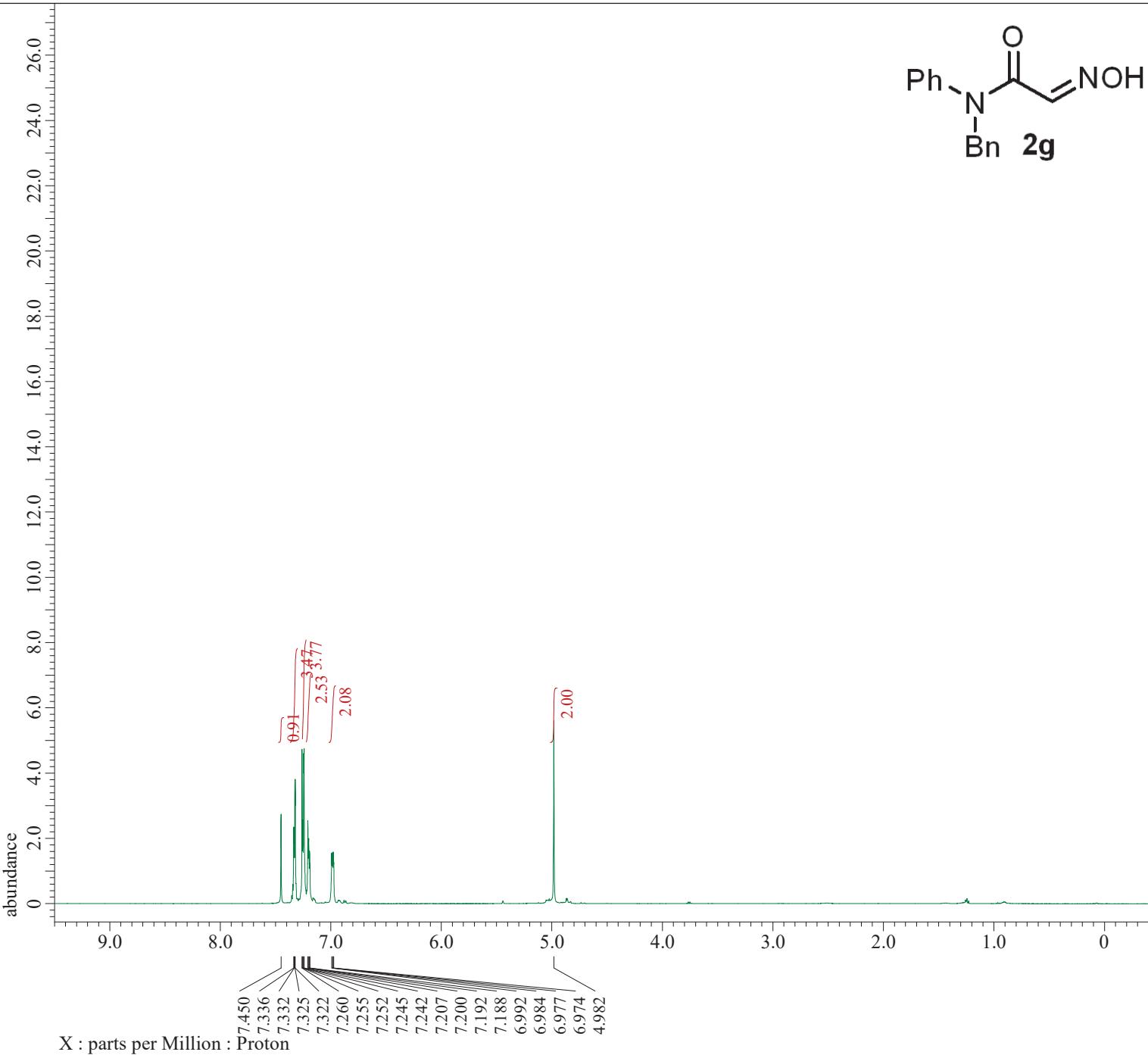
DFILE TU-02-037-2\_170614\_proton-1-  
 COMNT single\_pulse  
 DATIM 2017-06-14 19:32:21  
 OBNUC 1H  
 EXMOD proton.jxp  
 OBFRQ 500.16 MHz  
 OBSET 2.41 KHz  
 OBFIN 6.01 Hz  
 POINT 13107  
 FREQU 7507.51 Hz  
 SCANS 8  
 ACQTM 1.7459 sec  
 PD 5.0000 sec  
 PW1 6.85 usec  
 IRNUC 1H  
 CTEMP 13.7 c  
 SLVNT CDCL<sub>3</sub>  
 EXREF 7.26 ppm  
 BF 0.10 Hz  
 RGAIN 42





DFILE TU-02-037-2 170614\_carbon-1-  
COMNT single pulse decoupled gated NO  
DATIM 2017-06-14 19:33:45  
OBNUC 13C  
EXMOD carbon.jxp  
OBFRQ 125.77 MHz  
OBSET 7.87 KHz  
OBFIN 4.21 Hz  
POINT 26214  
FREQU 31446.54 Hz  
SCANS 512  
ACQTM 0.8336 sec  
PD 2.0000 sec  
PW1 3.12 usec  
IRNUC 1H  
CTEMP 13.7 c  
SLVNT CDCL<sub>3</sub>  
EXREF 77.00 ppm  
BF 0.10 Hz  
RGAIN 60

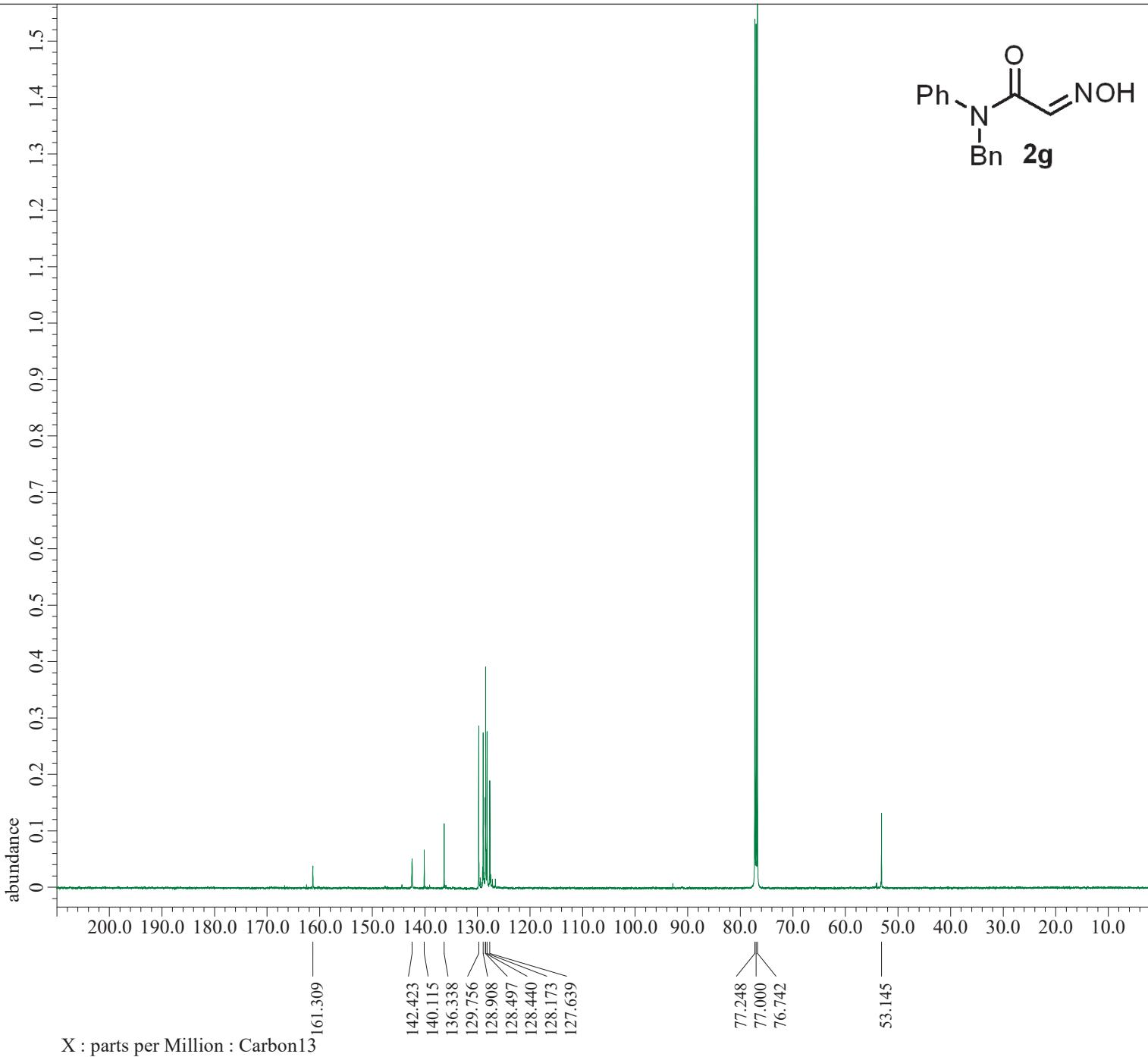




JEOL

---- PROCESSING PARAMETERS ----  
dc\_balance( 0, FALSE )  
sexp( 0.2[Hz], 0.0[s] )  
trapezoid( 0[%], 0[%], 80[%], 100[%] )  
zerofill( 1 )  
fft( 1, TRUE, TRUE )  
machinephase  
ppm  
Derived from: TY\_10\_110\_01\_proton-1\_Ana-1.jdf

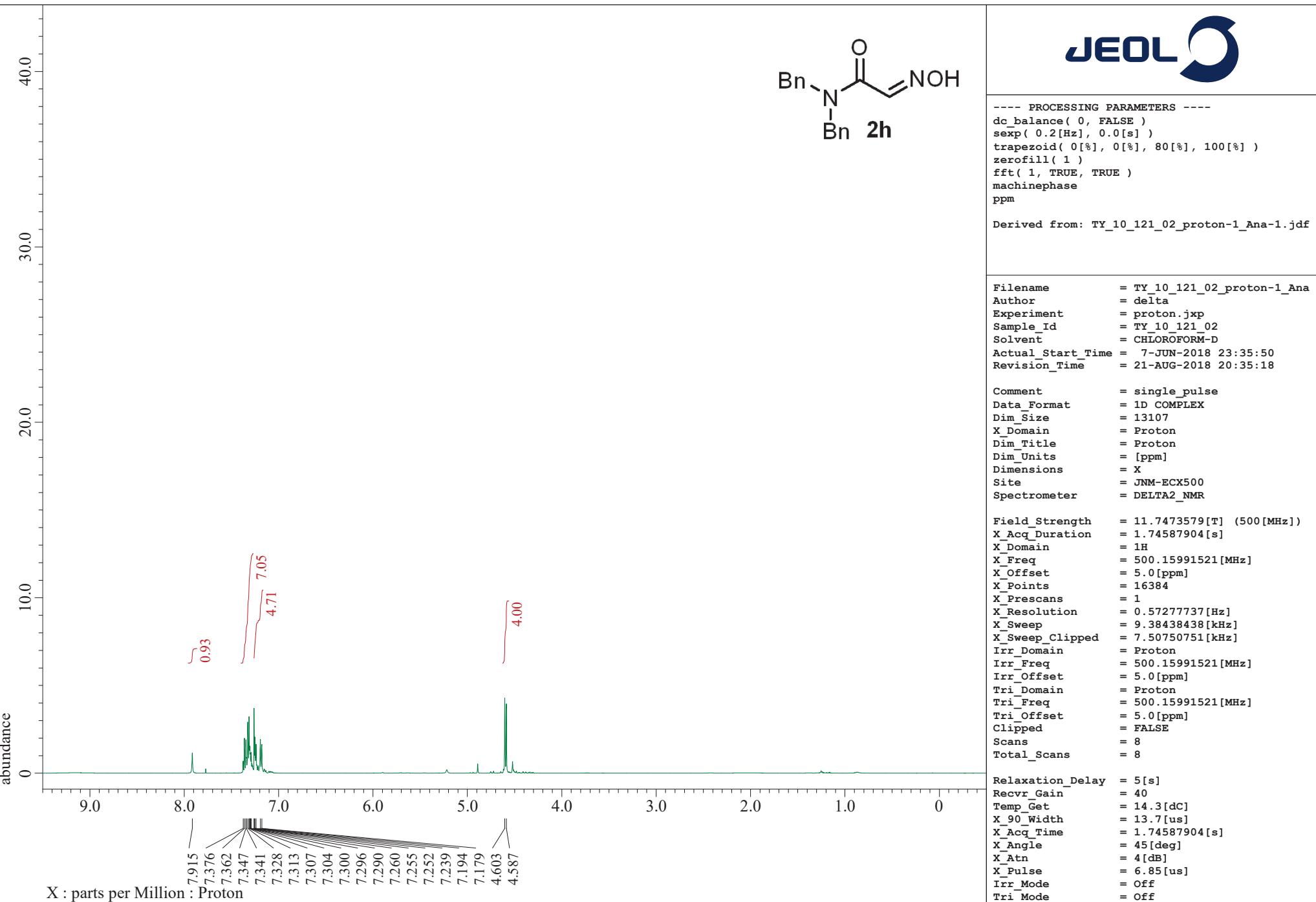
Filename = TY\_10\_110\_01\_proton-1\_Ana  
Author = delta  
Experiment = proton.jxp  
Sample\_Id = TY\_10\_110\_01  
Solvent = CHLOROFORM-D  
Actual\_Start\_Time = 30-MAY-2018 23:13:45  
Revision\_Time = 15-NOV-2018 12:21:05  
Comment = single\_pulse  
Data\_Format = 1D COMPLEX  
Dim\_Size = 13107  
X\_Domain = Proton  
Dim\_Title = Proton  
Dim\_Units = [ppm]  
Dimensions = X  
Site = JNM-ECX500  
Spectrometer = DELTA2\_NMR  
Field\_Strength = 11.7473579[T] (500[MHz])  
X\_Acq\_Duration = 1.74587904[s]  
X\_Domain = 1H  
X\_Freq = 500.15991521[MHz]  
X\_Offset = 5.0[ppm]  
X\_Points = 16384  
X\_Prescans = 1  
X\_Resolution = 0.57277737[Hz]  
X\_Sweep = 9.38438438[kHz]  
X\_Sweep\_Clipped = 7.50750751[kHz]  
Irr\_Domain = Proton  
Irr\_Freq = 500.15991521[MHz]  
Irr\_Offset = 5.0[ppm]  
Tri\_Domain = Proton  
Tri\_Freq = 500.15991521[MHz]  
Tri\_Offset = 5.0[ppm]  
Clipped = FALSE  
Scans = 8  
Total\_Scans = 8  
Relaxation\_Delay = 5[s]  
Recvrv\_Gain = 40  
Temp\_Get = 14.7[dC]  
X\_90\_Width = 13.7[us]  
X\_Acq\_Time = 1.74587904[s]  
X\_Angle = 45[deg]  
X\_Atn = 4[dB]  
X\_Pulse = 6.85[us]  
Irr\_Mode = Off  
Tri\_Mode = Off

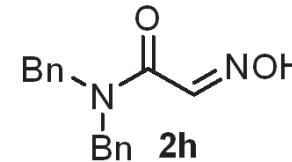
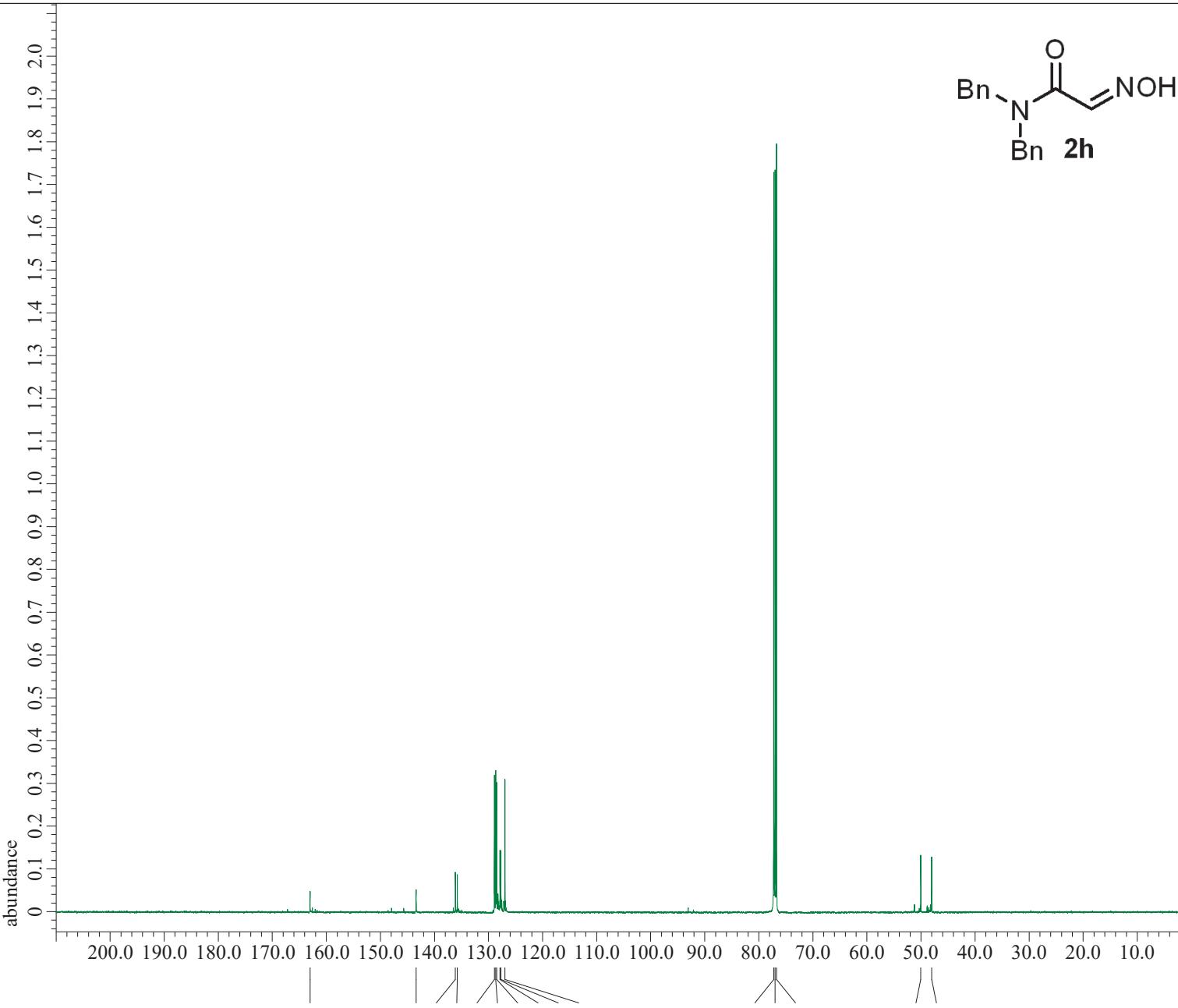


JEOL

---- PROCESSING PARAMETERS ----  
dc\_balance( 0, FALSE )  
sexp( 2.0[Hz], 0.0[s] )  
trapezoid( 0[%], 0[%], 80[%], 100[%] )  
zerofill( 1 )  
fft( 1, TRUE, TRUE )  
machinephase  
ppm  
Derived from: TY\_10\_110\_01\_carbon-1-1.jdf

Filename = TY\_10\_110\_01\_carbon-1-2.j  
Author = delta  
Experiment = carbon.jxp  
Sample\_Id = TY\_10\_110\_01  
Solvent = CHLOROFORM-D  
Actual\_Start\_Time = 30-MAY-2018 23:15:32  
Revision\_Time = 21-AUG-2018 20:53:27  
Comment = single pulse decoupled ga  
Data\_Format = 1D COMPLEX  
Dim\_Size = 26214  
X\_Domain = Carbon  
Dim\_Title = Carbon13  
Dim\_Units = [ppm]  
Dimensions = X  
Site = JNM-ECX500  
Spectrometer = DELTA2\_NMR  
Field\_Strength = 11.7473579[T] (500[MHz])  
X\_Acq\_Duration = 0.83361792[s]  
X\_Offset = 13C  
X\_Freq = 125.76529768[MHz]  
X\_Points = 100[ppm]  
X\_Rescans = 32768  
X\_Prescans = 4  
X\_Resolution = 1.19959034[Hz]  
X\_Sweep = 39.3081761[kHz]  
X\_Sweep\_Clipped = 31.44654088[kHz]  
Irr\_Domain = Proton  
Irr\_Freq = 500.15991521[MHz]  
Irr\_Offset = 5.0[ppm]  
Clipped = FALSE  
Scans = 9850  
Total\_Scans = 9850  
Relaxation\_Delay = 2[s]  
Recv\_Gain = 58  
Temp\_Get = 16[dC]  
X\_90\_Width = 9.36[us]  
X\_Acq\_Time = 0.83361792[s]  
X\_Angle = 30[deg]  
X\_Atn = 3[dB]  
X\_Pulse = 3.12[us]  
Irr\_Atn\_Dec = 20.54[dB]  
Irr\_Atn\_Noe = 20.54[dB]  
Irr\_Noise = WALTZ  
Irr\_Pwidth = 92[us]  
Decoupling = TRUE

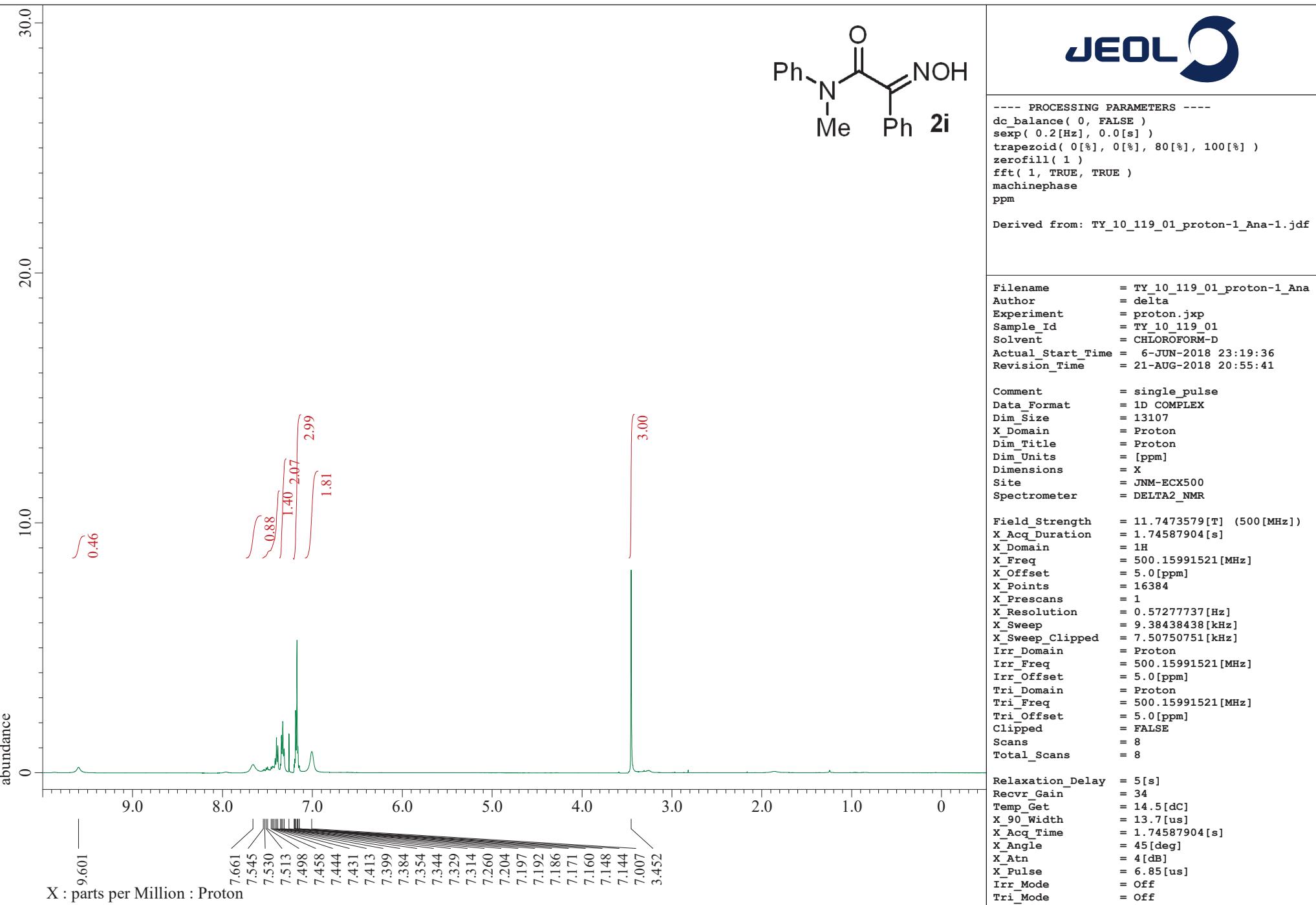


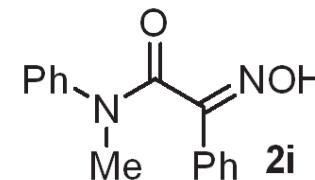
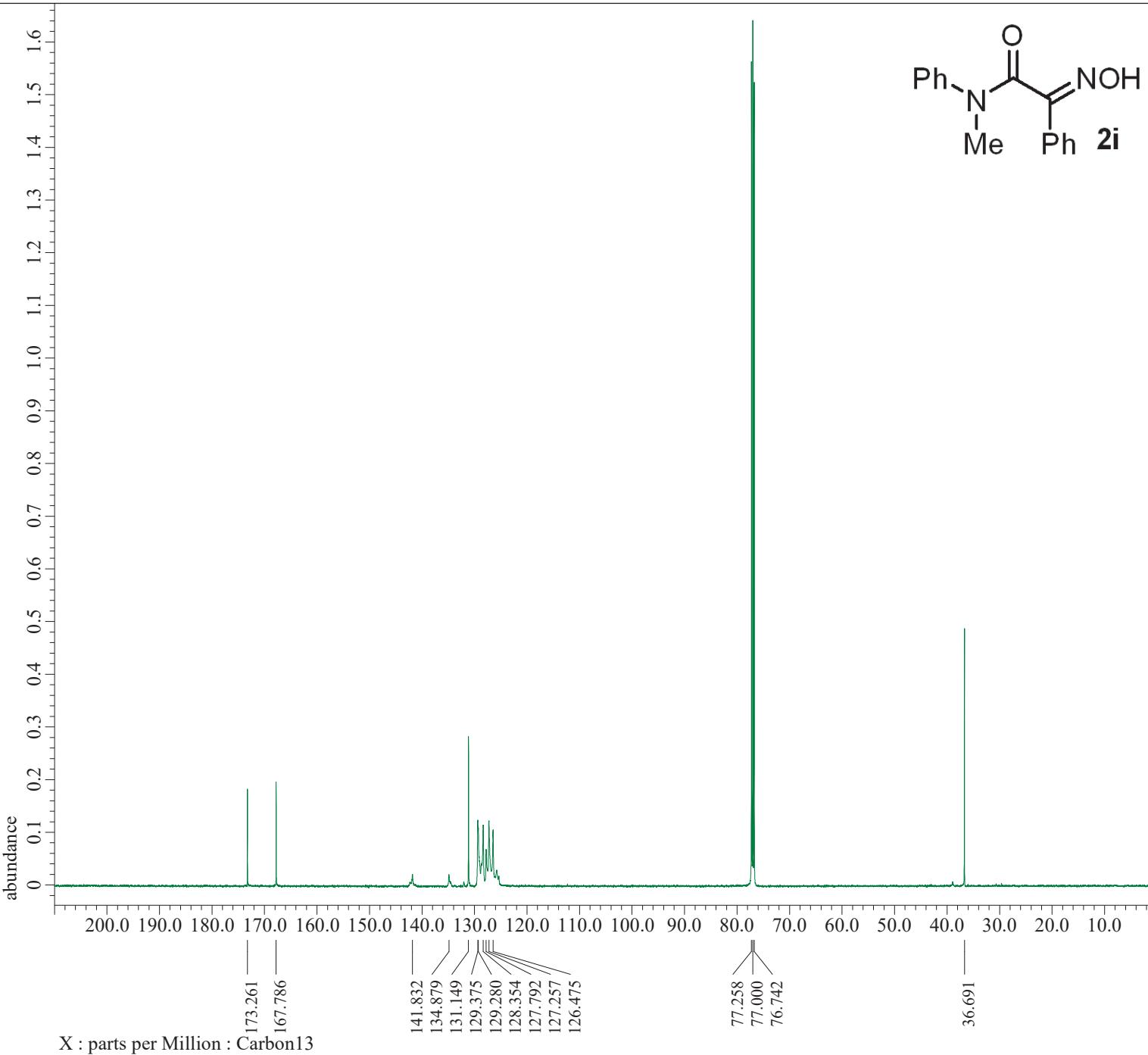


**JEOL**

---- PROCESSING PARAMETERS ----  
dc\_balance( 0, FALSE )  
sexp( 2.0[Hz], 0.0[s] )  
trapezoid( 0[%], 0[%], 80[%], 100[%] )  
zerofill( 1 )  
fft( 1, TRUE, TRUE )  
machinephase  
ppm  
Derived from: TY\_10\_121\_02\_carbon-1\_Ana-1.jdf

Filename = TY\_10\_121\_02\_carbon-1\_Ana  
Author = delta  
Experiment = carbon.jxp  
Sample\_Id = TY\_10\_121\_02  
Solvent = CHLOROFORM-D  
Actual\_Start\_Time = 7-JUN-2018 23:37:34  
Revision\_Time = 21-AUG-2018 20:36:38  
Comment = single pulse decoupled ga  
Data\_Format = 1D COMPLEX  
Dim\_Size = 26214  
X\_Domain = Carbon  
Dim\_Title = Carbon13  
Dim\_Units = [ppm]  
Dimensions = X  
Site = JNM-ECX500  
Spectrometer = DELTA2\_NMR  
Field\_Strength = 11.7473579[T] (500[MHz])  
X\_Acq\_Duration = 0.83361792[s]  
X\_Domain = 13C  
X\_Freq = 125.76529768[MHz]  
X\_Offset = 100[ppm]  
X\_Points = 32768  
X\_Prescans = 4  
X\_Resolution = 1.19959034[Hz]  
X\_Sweep = 39.3081761[kHz]  
X\_Sweep\_Clipped = 31.44654088[kHz]  
Irr\_Domain = Proton  
Irr\_Freq = 500.15991521[MHz]  
Irr\_Offset = 5.0[ppm]  
Clipped = TRUE  
Scans = 9800  
Total\_Scans = 9800  
Relaxation\_Delay = 2[s]  
Recv\_Gain = 58  
Temp\_Get = 14.7[dC]  
X\_90\_Width = 9.36[us]  
X\_Acq\_Time = 0.83361792[s]  
X\_Angle = 30[deg]  
X\_Atn = 3[dB]  
X\_Pulse = 3.12[us]  
Irr\_Atn\_Dec = 20.54[dB]  
Irr\_Atn\_Noe = 20.54[dB]  
Irr\_Noise = WALTZ  
Irr\_Pwidth = 92[us]  
Decoupling = TRUE

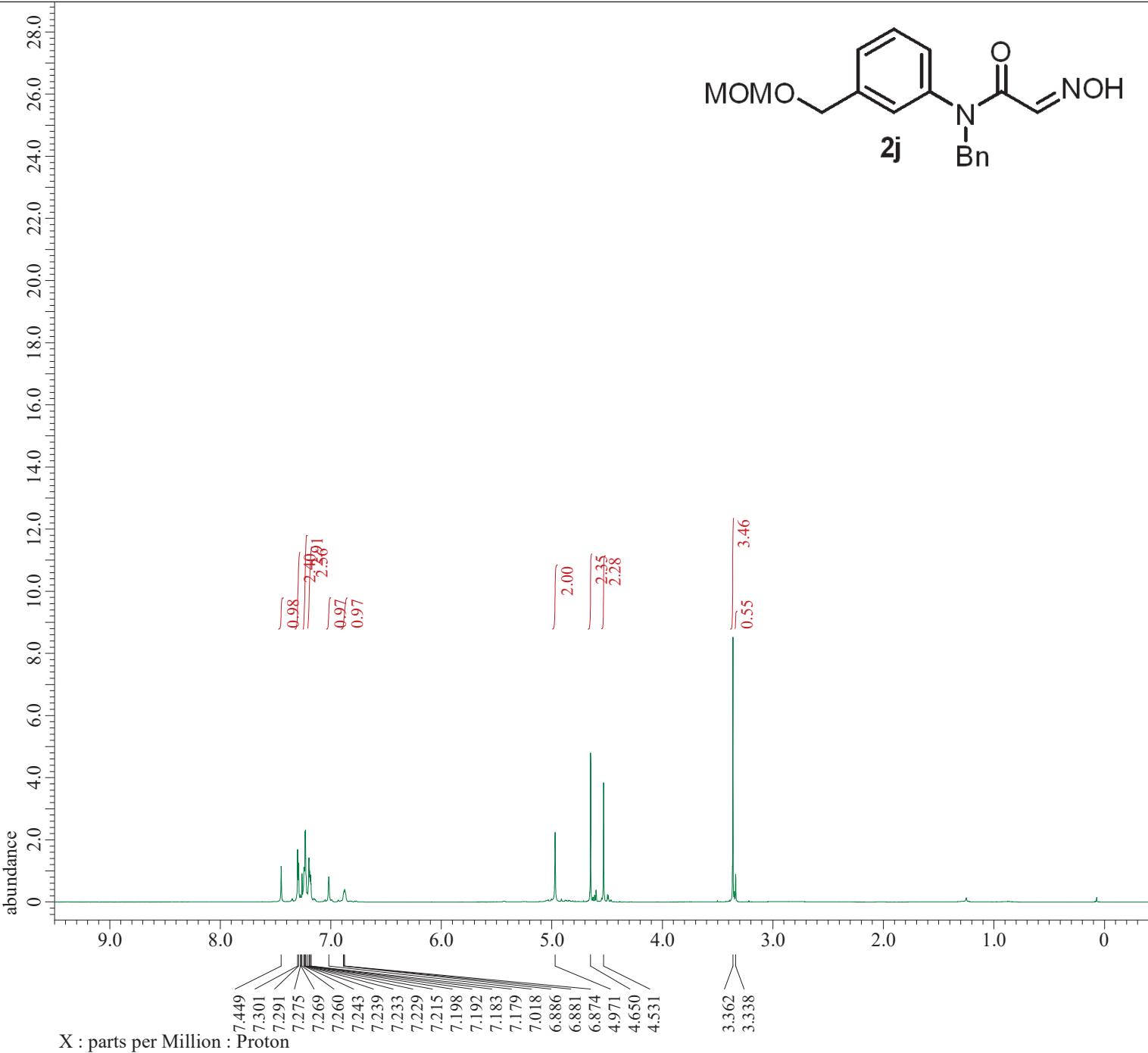




**JEOL**

---- PROCESSING PARAMETERS ----  
dc\_balance( 0, FALSE )  
sexp( 2.0[Hz], 0.0[s] )  
trapezoid( 0[%], 0[%], 80[%], 100[%] )  
zerofill( 1 )  
fft( 1, TRUE, TRUE )  
machinephase  
ppm  
Derived from: TY\_10\_119\_01\_carbon-1\_Ana-1.jdf

Filename = TY\_10\_119\_01\_carbon-1\_Ana  
Author = delta  
Experiment = carbon.jxp  
Sample\_Id = TY\_10\_119\_01  
Solvent = CHLOROFORM-D  
Actual\_Start\_Time = 6-JUN-2018 23:21:21  
Revision\_Time = 21-AUG-2018 20:57:01  
Comment = single pulse decoupled ga  
Data\_Format = 1D COMPLEX  
Dim\_Size = 26214  
X\_Domain = Carbon  
Dim\_Title = Carbon13  
Dim\_Units = [ppm]  
Dimensions = X  
Site = JNM-ECX500  
Spectrometer = DELTA2\_NMR  
Field\_Strength = 11.7473579[T] (500[MHz])  
X\_Acq\_Duration = 0.83361792[s]  
X\_Domain = 13C  
X\_Freq = 125.76529768[MHz]  
X\_Offset = 100[ppm]  
X\_Points = 32768  
X\_Prescans = 4  
X\_Resolution = 1.19959034[Hz]  
X\_Sweep = 39.3081761[kHz]  
X\_Sweep\_Clipped = 31.44654088[kHz]  
Irr\_Domain = Proton  
Irr\_Freq = 500.15991521[MHz]  
Irr\_Offset = 5.0[ppm]  
Clipped = FALSE  
Scans = 9800  
Total\_Scans = 9800  
Relaxation\_Delay = 2[s]  
Recv\_Gain = 58  
Temp\_Get = 15.7[dC]  
X\_90\_Width = 9.36[us]  
X\_Acq\_Time = 0.83361792[s]  
X\_Angle = 30[deg]  
X\_Atn = 3[dB]  
X\_Pulse = 3.12[us]  
Irr\_Atn\_Dec = 20.54[dB]  
Irr\_Atn\_Noe = 20.54[dB]  
Irr\_Noise = WALTZ  
Irr\_Pwidth = 92[us]  
Decoupling = TRUE



JEOL

----- PROCESSING PARAMETERS -----

```

dc_balance( 0, FALSE )
sexp( 0.2[Hz], 0.0[s] )
trapezoid( 0[%], 0[%], 80[%], 100[%] )
zerofill( 1 )
fft( 1, TRUE, TRUE )
machinephase
ppm

```

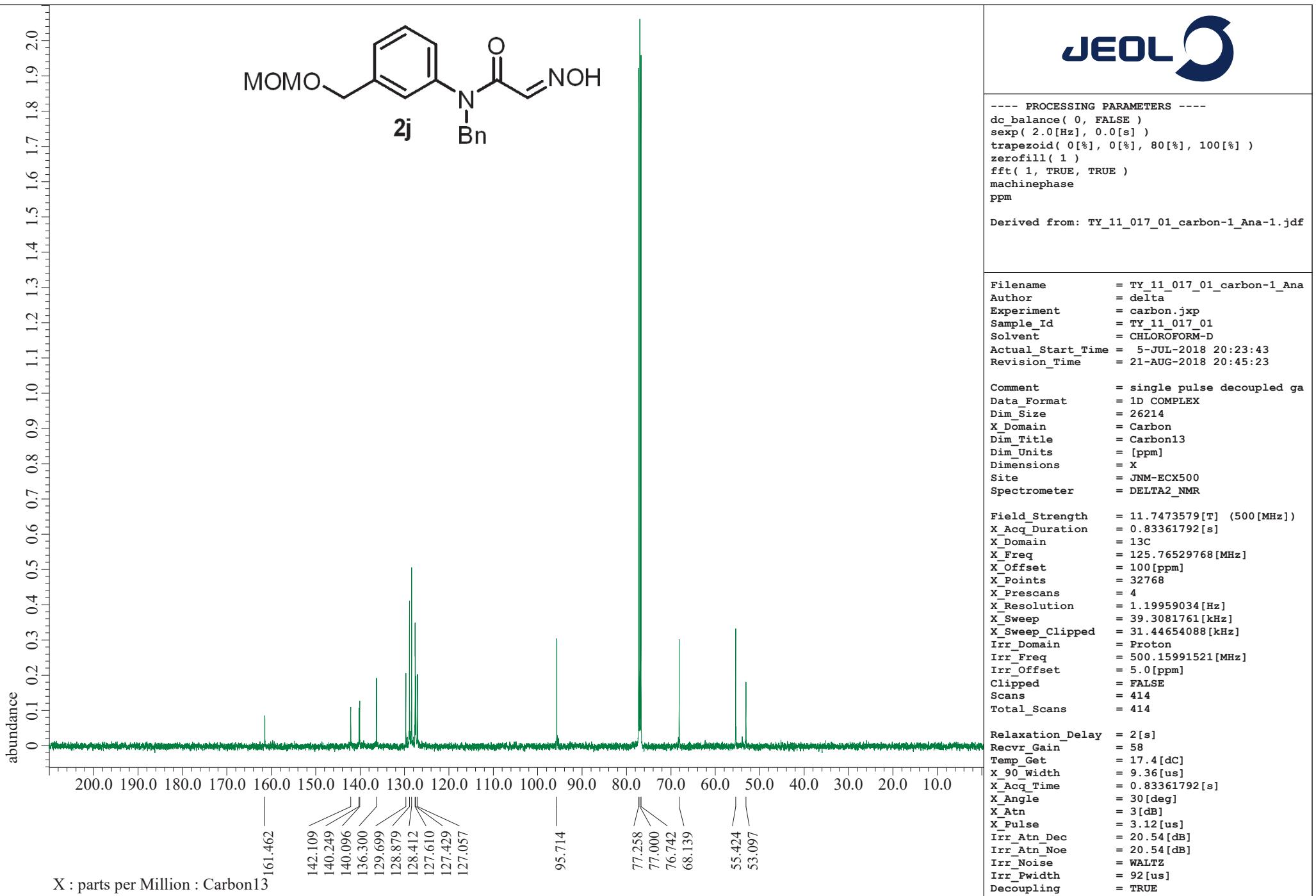
Derived from: TY\_11\_017\_01\_proton-1\_Ana-3.jdf

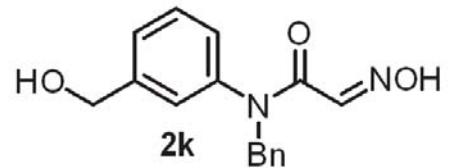
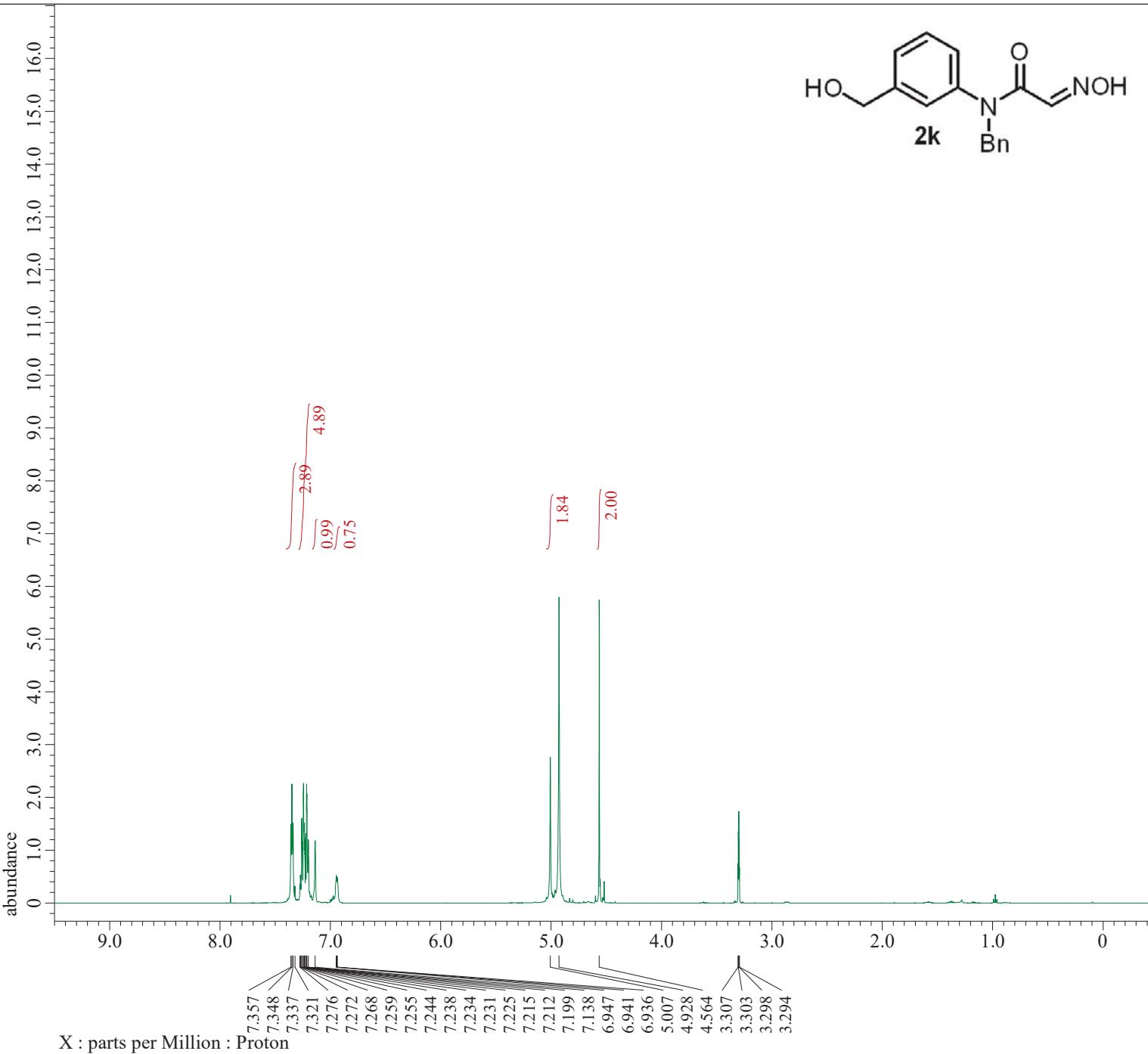
Filename = TY\_11\_017\_01\_proton-1\_Ana
Author = delta
Experiment = proton.jxp
Sample\_Id = TY\_11\_017\_01
Solvent = CHLOROFORM-D
Actual\_Start\_Time = 5-JUL-2018 20:21:22
Revision\_Time = 21-AUG-2018 20:44:54

Comment = single\_pulse
Data\_Format = 1D COMPLEX
Dim\_Size = 13107
X\_Domain = Proton
Dim\_Title = Proton
Dim\_Units = [ppm]
Dimensions = X
Site = JNM-ECX500
Spectrometer = DELTA2\_NMR

Field\_Strength = 11.7473579[T] (500[MHz])
X\_Acq\_Duration = 1.74587904[s]
X\_Domain = 1H
X\_Freq = 500.15991521[MHz]
X\_Offset = 5.0[ppm]
X\_Points = 16384
X\_Prescans = 1
X\_Resolution = 0.57277737[Hz]
X\_Sweep = 9.38438438[kHz]
X\_Sweep\_Clipped = 7.50750751[kHz]
Irr\_Domain = Proton
Irr\_Freq = 500.15991521[MHz]
Irr\_Offset = 5.0[ppm]
Tri\_Domain = Proton
Tri\_Freq = 500.15991521[MHz]
Tri\_Offset = 5.0[ppm]
Clipped = FALSE
Scans = 8
Total\_Scans = 8

Relaxation\_Delay = 5[s]
Recvr\_Gain = 30
Temp\_Get = 17.1[dC]
X\_90\_Width = 13.7[us]
X\_Acq\_Time = 1.74587904[s]
X\_Angle = 45[deg]
X\_Atn = 4[dB]
X\_Pulse = 6.85[us]
Irr\_Mode = Off
Tri\_Mode = Off

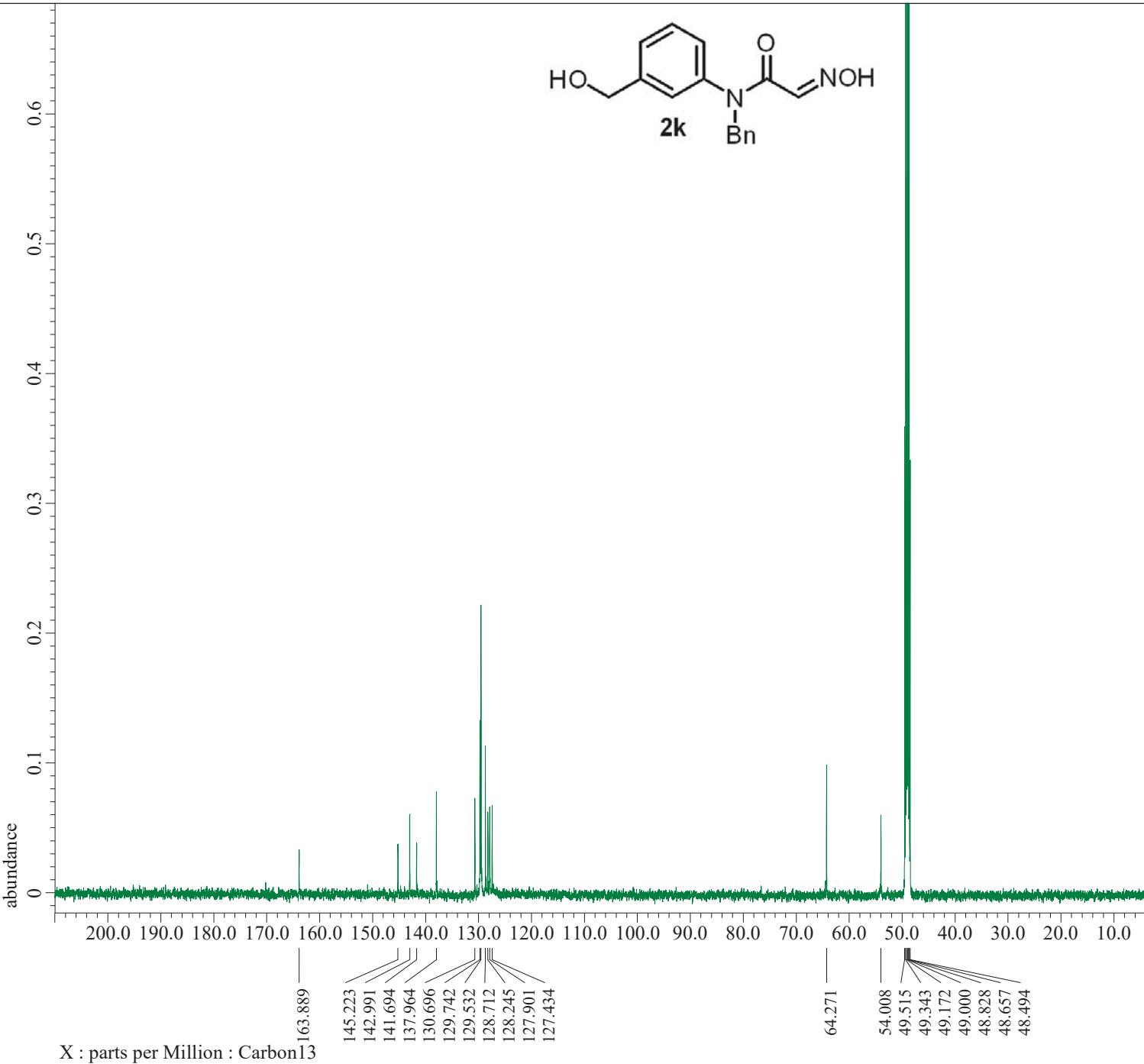
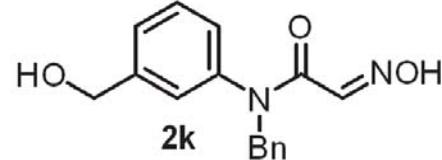




**JEOL**

---- PROCESSING PARAMETERS ----  
dc\_balance( 0, FALSE )  
sexp( 0.2[Hz], 0.0[s] )  
trapezoid( 0[%], 0[%], 80[%], 100[%] )  
zerofill( 1 )  
fft( 1, TRUE, TRUE )  
machinephase  
ppm  
Derived from: TY\_10\_146\_02\_proton-1\_Ana-1.jdf

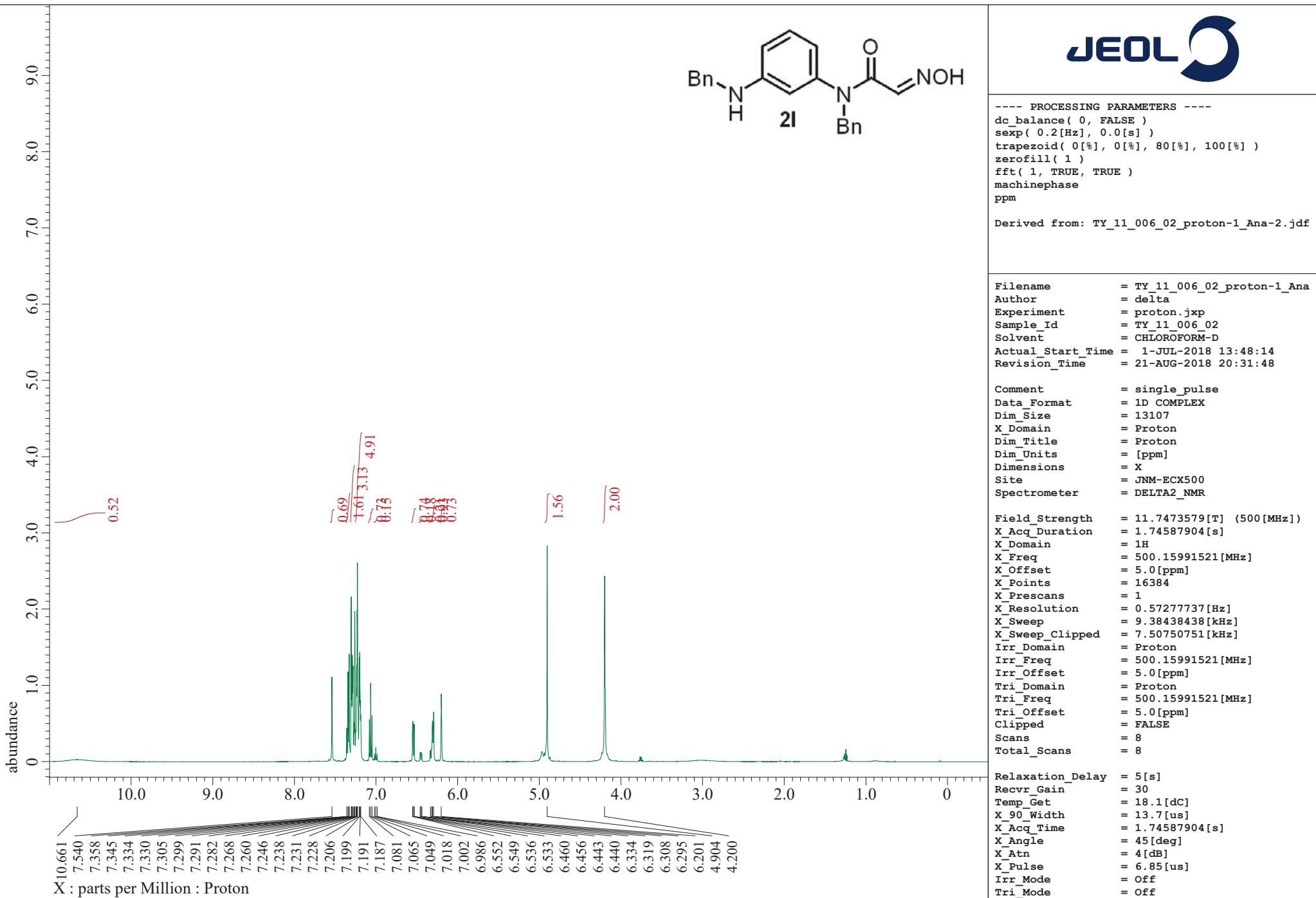
Filename = TY\_10\_146\_02\_proton-1\_Ana  
Author = delta  
Experiment = proton.jxp  
Sample\_Id = TY\_10\_146\_02  
Solvent = METHANOL-D4  
Actual\_Start\_Time = 23-JUN-2018 15:03:09  
Revision\_Time = 21-AUG-2018 20:39:36  
Comment = single\_pulse  
Data\_Format = 1D COMPLEX  
Dim\_Size = 13107  
X\_Domain = Proton  
Dim\_Title = Proton  
Dim\_Units = [ppm]  
Dimensions = X  
Site = JNM-ECX500  
Spectrometer = DELTA2\_NMR  
Field\_Strength = 11.7473579[T] (500[MHz])  
X\_Acq\_Duration = 1.74587904[s]  
X\_Domain = 1H  
X\_Freq = 500.15991521[MHz]  
X\_Offset = 5.0[ppm]  
X\_Points = 16384  
X\_Prescans = 1  
X\_Resolution = 0.57277737[Hz]  
X\_Sweep = 9.38438438[kHz]  
X\_Sweep\_Clipped = 7.50750751[kHz]  
Irr\_Domain = Proton  
Irr\_Freq = 500.15991521[MHz]  
Irr\_Offset = 5.0[ppm]  
Tri\_Domain = Proton  
Tri\_Freq = 500.15991521[MHz]  
Tri\_Offset = 5.0[ppm]  
Clipped = FALSE  
Scans = 8  
Total\_Scans = 8  
Relaxation\_Delay = 5[s]  
Recvr\_Gain = 36  
Temp\_Get = 15.1[dC]  
X\_90\_Width = 13.7[us]  
X\_Acq\_Time = 1.74587904[s]  
X\_Angle = 45[deg]  
X\_Atn = 4[dB]  
X\_Pulse = 6.85[us]  
Irr\_Mode = Off  
Tri\_Mode = Off

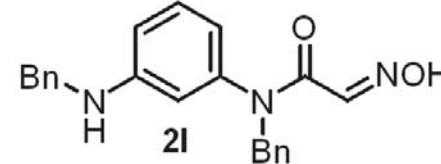
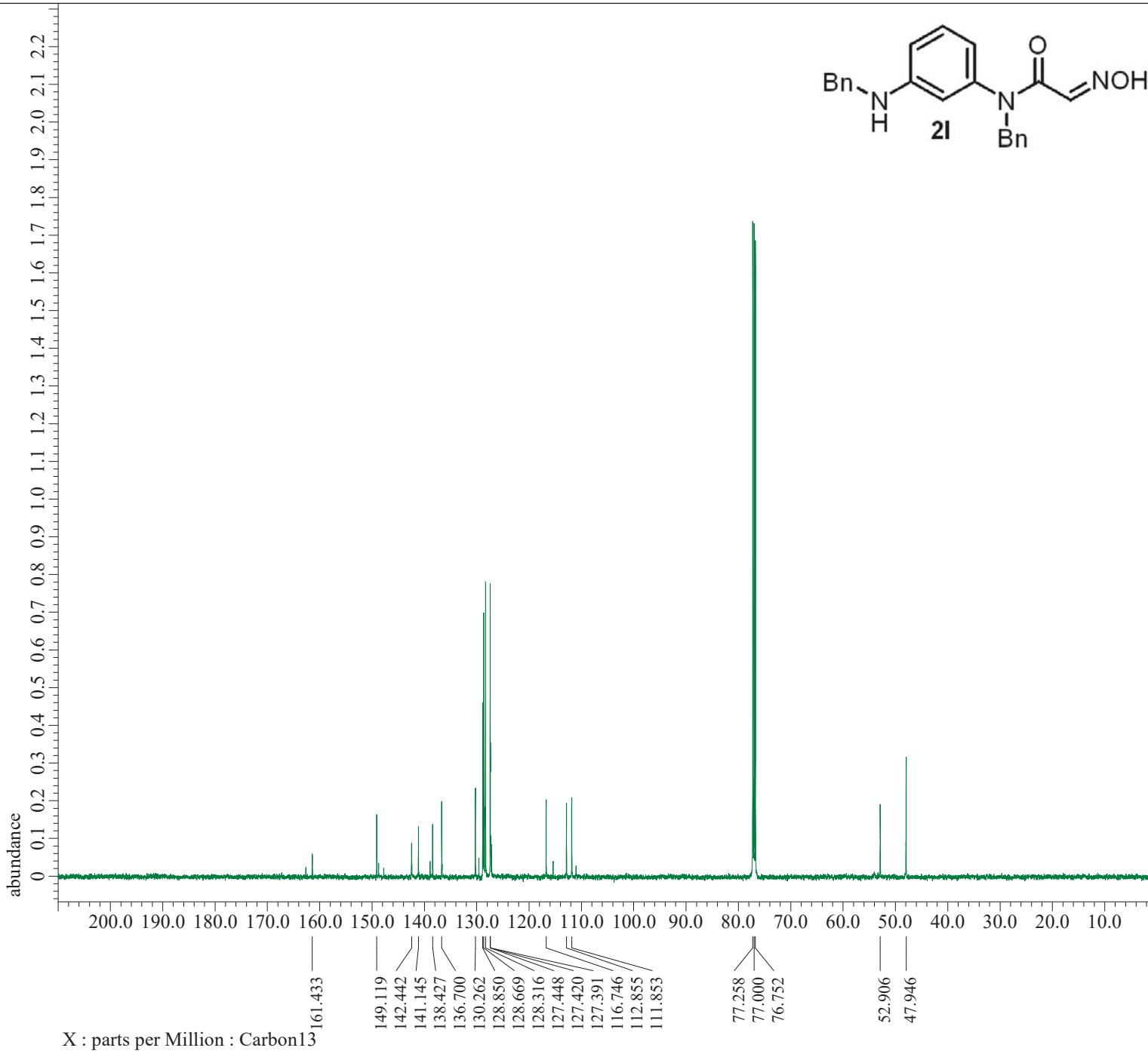


**JEOL**

---- PROCESSING PARAMETERS ----  
dc\_balance( 0, FALSE )  
sexp( 2.0[Hz], 0.0[s] )  
trapezoid( 0[%], 0[%], 80[%], 100[%] )  
zerofill( 1 )  
fft( 1, TRUE, TRUE )  
machinephase  
ppm  
Derived from: TY\_10\_146\_02\_carbon-1\_Ana-1.jdf

Filename = TY\_10\_146\_02\_carbon-1\_Ana  
Author = delta  
Experiment = carbon.jxp  
Sample\_Id = TY\_10\_146\_02  
Solvent = METHANOL-D4  
Actual\_Start\_Time = 23-JUN-2018 15:04:48  
Revision\_Time = 21-AUG-2018 20:40:34  
Comment = single pulse decoupled ga  
Data\_Format = 1D COMPLEX  
Dim\_Size = 26214  
X\_Domain = Carbon  
Dim\_Title = Carbon13  
Dim\_Units = [ppm]  
Dimensions = X  
Site = JNM-ECX500  
Spectrometer = DELTA2\_NMR  
Field\_Strength = 11.7473579[T] (500[MHz])  
X\_Acq\_Duration = 0.83361792[s]  
X\_Domain = 13C  
X\_Freq = 125.76529768[MHz]  
X\_Offset = 100[ppm]  
X\_Points = 32768  
X\_Prescans = 4  
X\_Resolution = 1.19959034[Hz]  
X\_Sweep = 39.3081761[kHz]  
X\_Sweep\_Clipped = 31.44654088[kHz]  
Irr\_Domain = Proton  
Irr\_Freq = 500.15991521[MHz]  
Irr\_Offset = 5.0[ppm]  
Clipped = FALSE  
Scans = 1720  
Total\_Scans = 1720  
Relaxation\_Delay = 2[s]  
Recv\_Gain = 58  
Temp\_Get = 15.1[dC]  
X\_90\_Width = 9.36[us]  
X\_Acq\_Time = 0.83361792[s]  
X\_Angle = 30[deg]  
X\_Atn = 3[dB]  
X\_Pulse = 3.12[us]  
Irr\_Atn\_Dec = 20.54[dB]  
Irr\_Atn\_Noe = 20.54[dB]  
Irr\_Noise = WALTZ  
Irr\_Pwidth = 92[us]  
Decoupling = TRUE

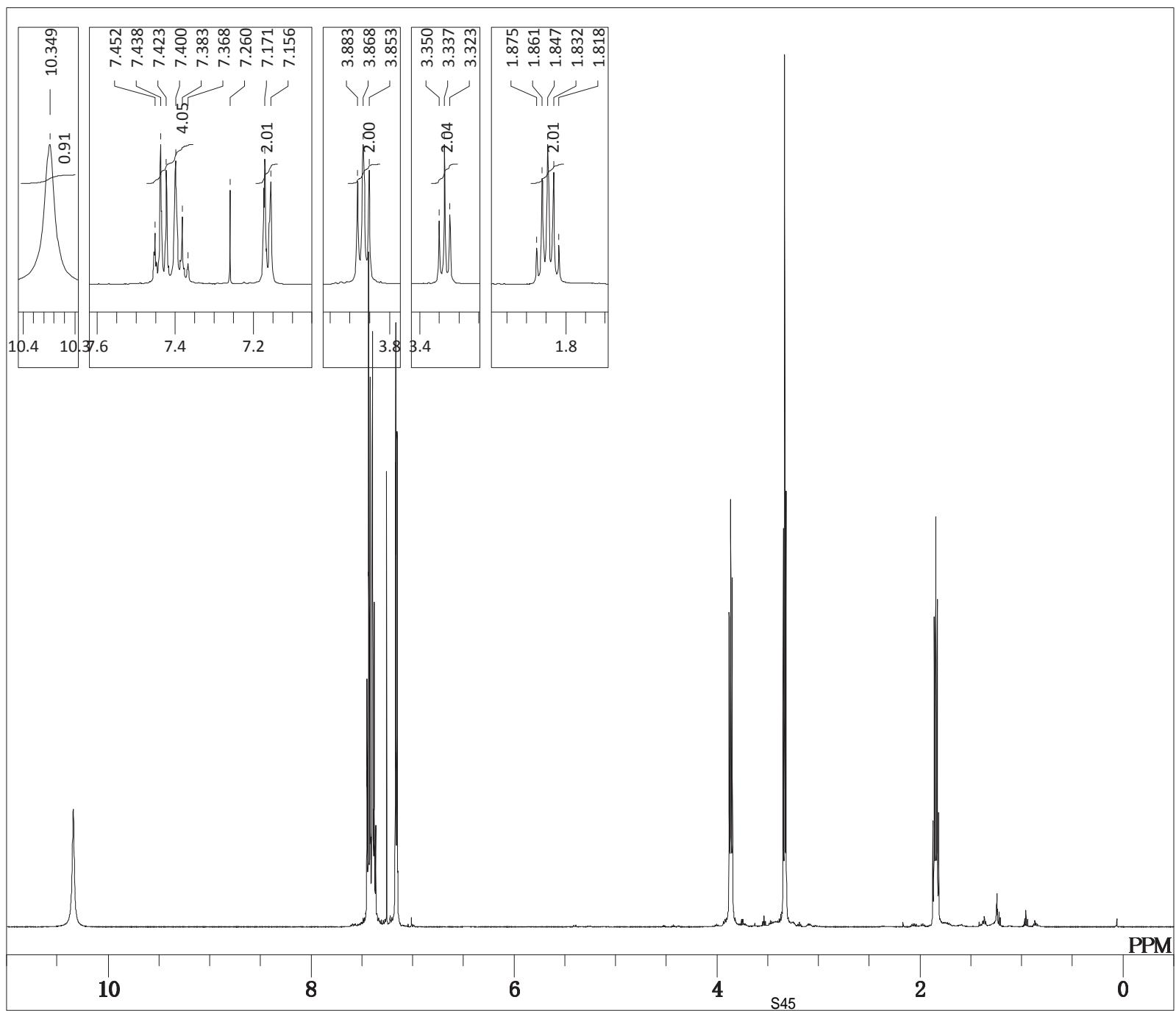




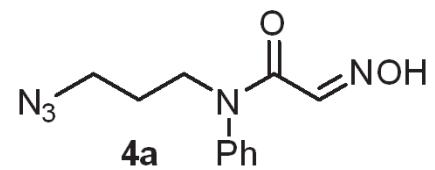
**JEOL**

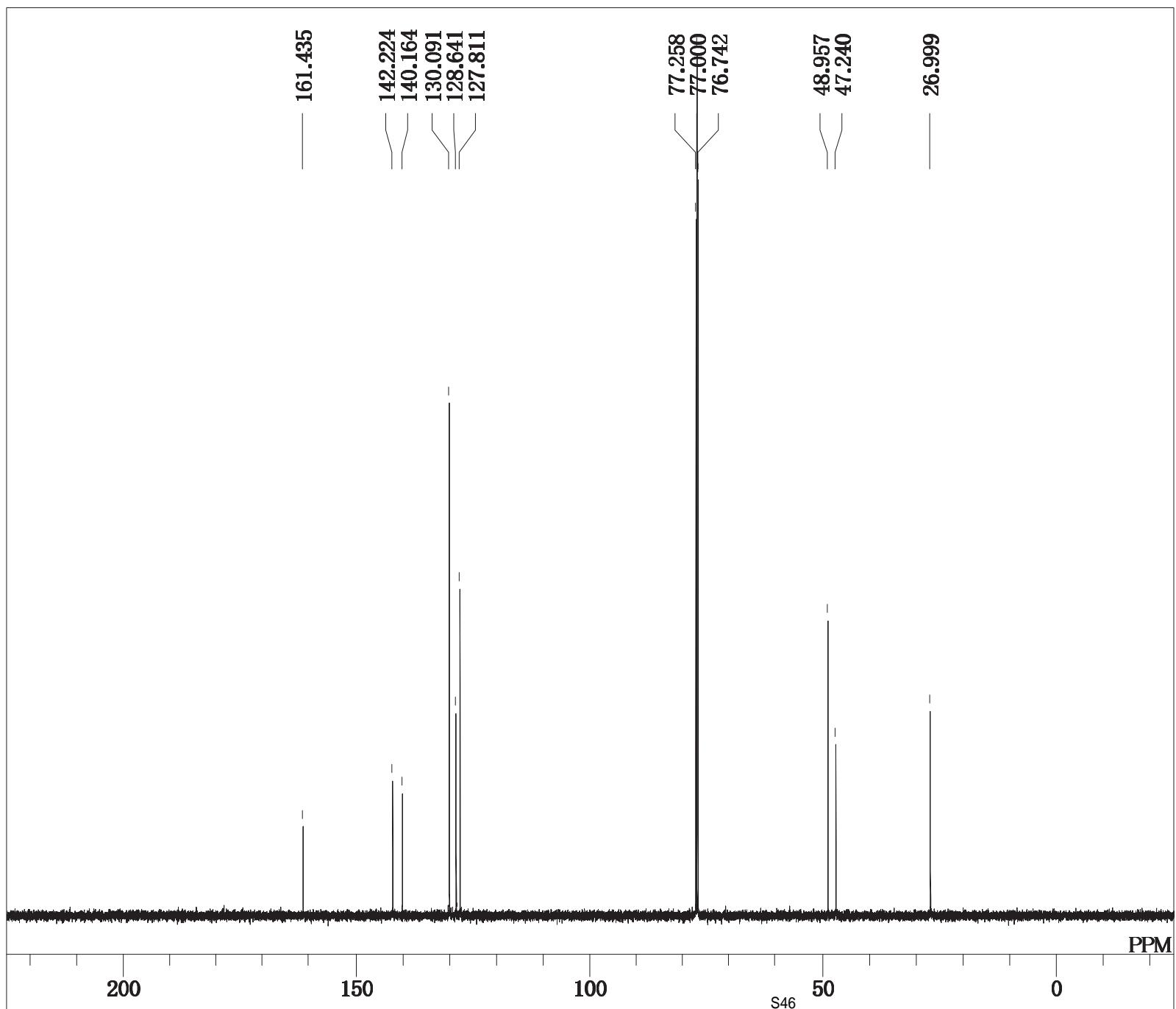
---- PROCESSING PARAMETERS ----  
dc\_balance( 0, FALSE )  
sexp( 2.0[Hz], 0.0[s] )  
trapezoid( 0[%], 0[%], 80[%], 100[%] )  
zerofill( 1 )  
fft( 1, TRUE, TRUE )  
machinephase  
ppm  
Derived from: TY\_11\_006\_02\_carbon-1\_Ana-1.jdf

Filename = TY\_11\_006\_02\_carbon-1\_Ana  
Author = delta  
Experiment = carbon.jxp  
Sample\_Id = TY\_11\_006\_02  
Solvent = CHLOROFORM-D  
Actual\_Start\_Time = 1-JUL-2018 13:49:58  
Revision\_Time = 21-AUG-2018 20:32:34  
Comment = single pulse decoupled ga  
Data\_Format = 1D COMPLEX  
Dim\_Size = 26214  
X\_Domain = Carbon  
Dim\_Title = Carbon13  
Dim\_Units = [ppm]  
Dimensions = X  
Site = JNM-ECX500  
Spectrometer = DELTA2\_NMR  
Field\_Strength = 11.7473579[T] (500[MHz])  
X\_Acq\_Duration = 0.83361792[s]  
X\_Domain = 13C  
X\_Freq = 125.76529768[MHz]  
X\_Offset = 100[ppm]  
X\_Points = 32768  
X\_Prescans = 4  
X\_Resolution = 1.19959034[Hz]  
X\_Sweep = 39.3081761[kHz]  
X\_Sweep\_Clipped = 31.44654088[kHz]  
Irr\_Domain = Proton  
Irr\_Freq = 500.15991521[MHz]  
Irr\_Offset = 5.0[ppm]  
Clipped = FALSE  
Scans = 900  
Total\_Scans = 900  
Relaxation\_Delay = 2[s]  
Recv\_Gain = 58  
Temp\_Get = 18.6[dC]  
X\_90\_Width = 9.36[us]  
X\_Acq\_Time = 0.83361792[s]  
X\_Angle = 30[deg]  
X\_Atn = 3[dB]  
X\_Pulse = 3.12[us]  
Irr\_Atn\_Dec = 20.54[dB]  
Irr\_Atn\_Noe = 20.54[dB]  
Irr\_Noise = WALTZ  
Irr\_Pwidth = 92[us]  
Decoupling = TRUE

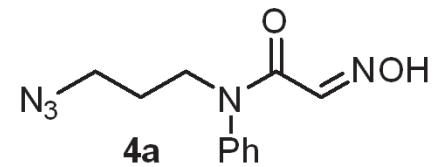


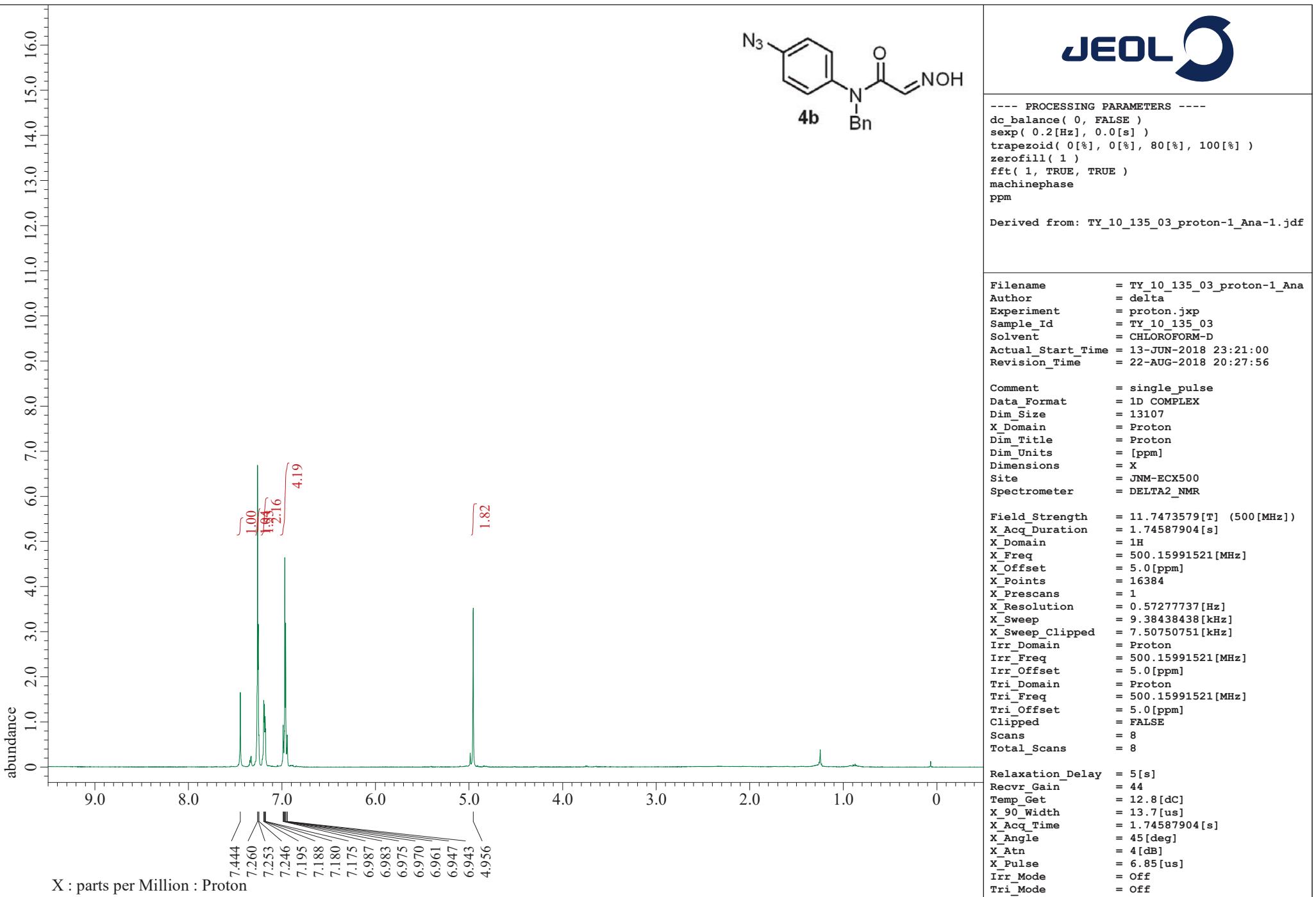
DFILE TU-01-147-2 170126\_proton-1-  
 COMNT single\_pulse  
 DATIM 2017-01-26 13:50:37  
 OBNUC 1H  
 EXMOD proton.jxp  
 OBFRQ 500.16 MHz  
 OBSET 2.41 KHz  
 OBFIN 6.01 Hz  
 POINT 13107  
 FREQU 7507.51 Hz  
 SCANS 8  
 ACQTM 1.7459 sec  
 PD 5.0000 sec  
 PW1 6.85 usec  
 IRNUC 1H  
 CTEMP 17.7 c  
 SLVNT CDCL<sub>3</sub>  
 EXREF 7.26 ppm  
 BF 0.10 Hz  
 RGAIN 34

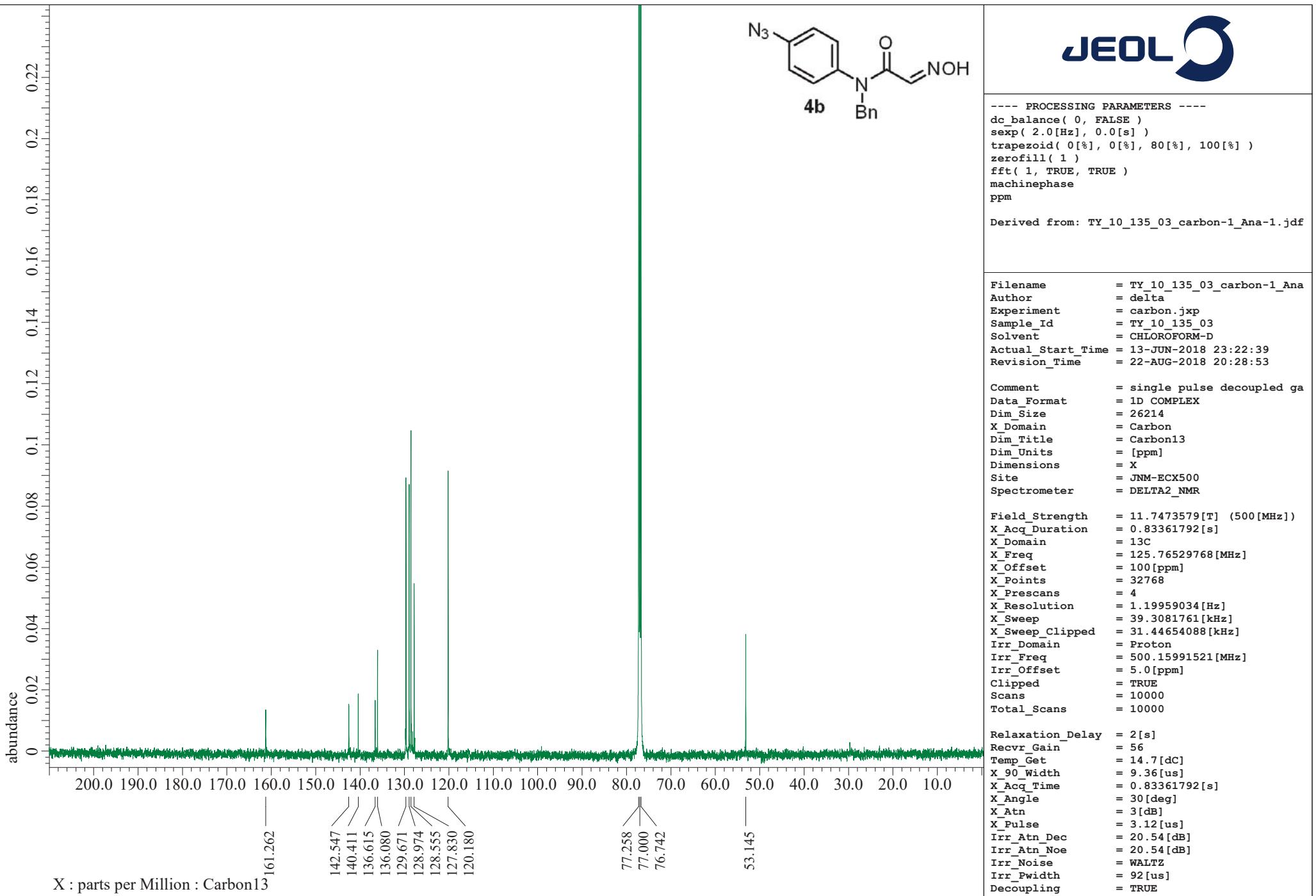


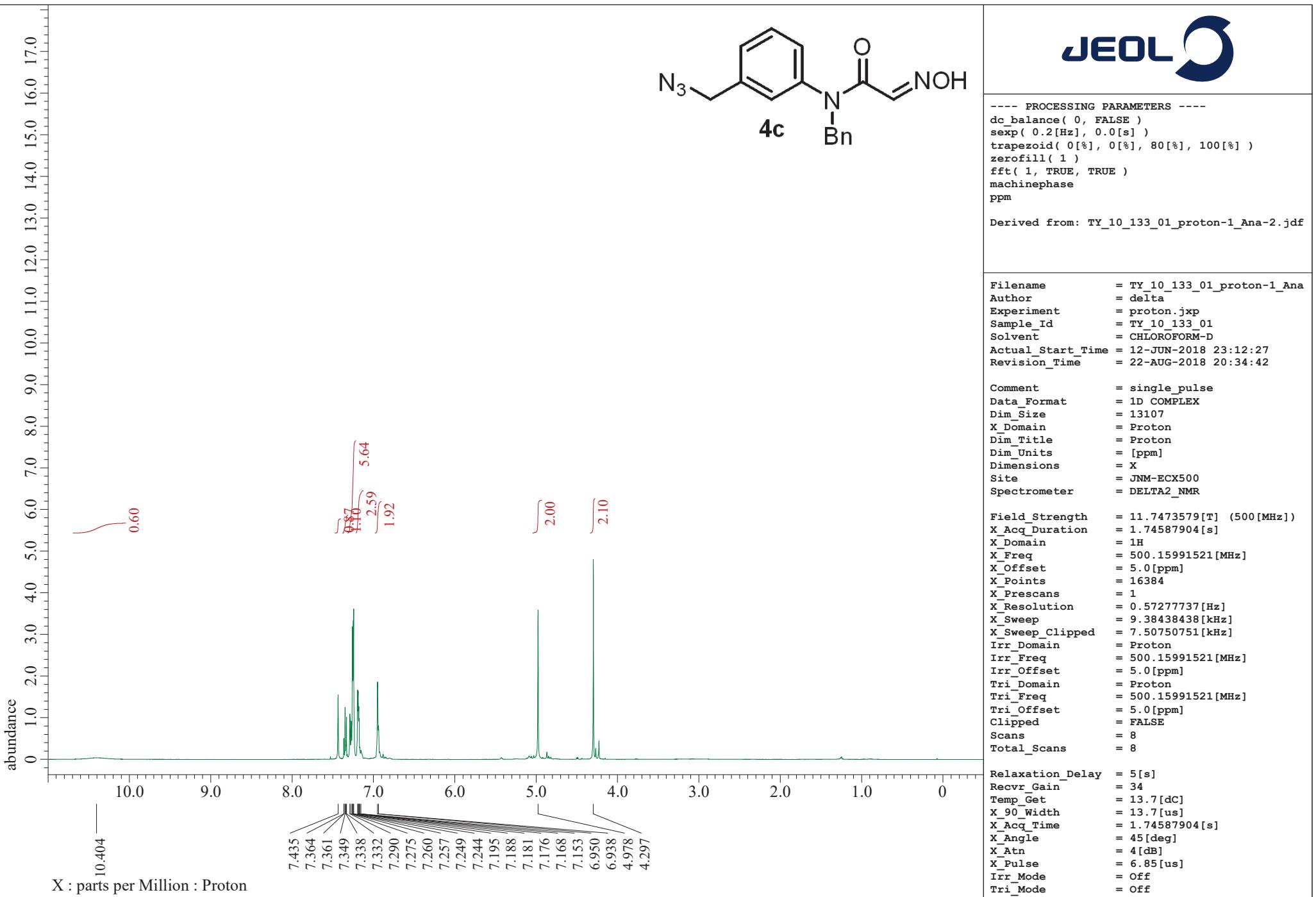


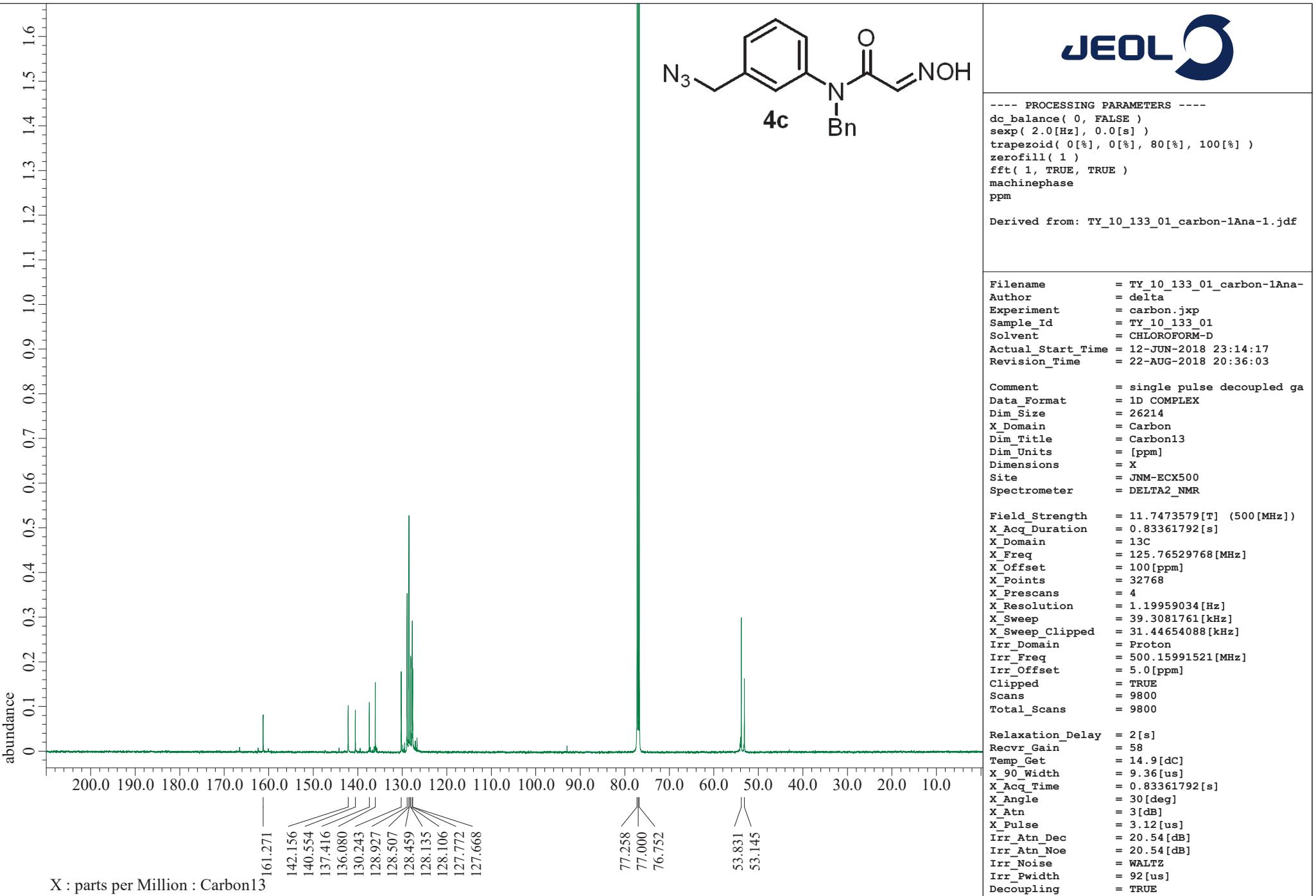
DFILE TU-01-147-2 170126\_carbon-1-  
COMNT single pulse decoupled gated NO  
DATIM 2017-01-26 13:52:28  
OBNUC 13C  
EXMOD carbon.jxp  
OBFRQ 125.77 MHz  
OBSET 7.87 KHz  
OBFIN 4.21 Hz  
POINT 26214  
FREQU 31446.54 Hz  
SCANS 512  
ACQTM 0.8336 sec  
PD 2.0000 sec  
PW1 3.12 usec  
IRNUC 1H  
CTEMP 18.2 c  
SLVNT CDCL3  
EXREF 77.00 ppm  
BF 0.10 Hz  
RGAIN 60

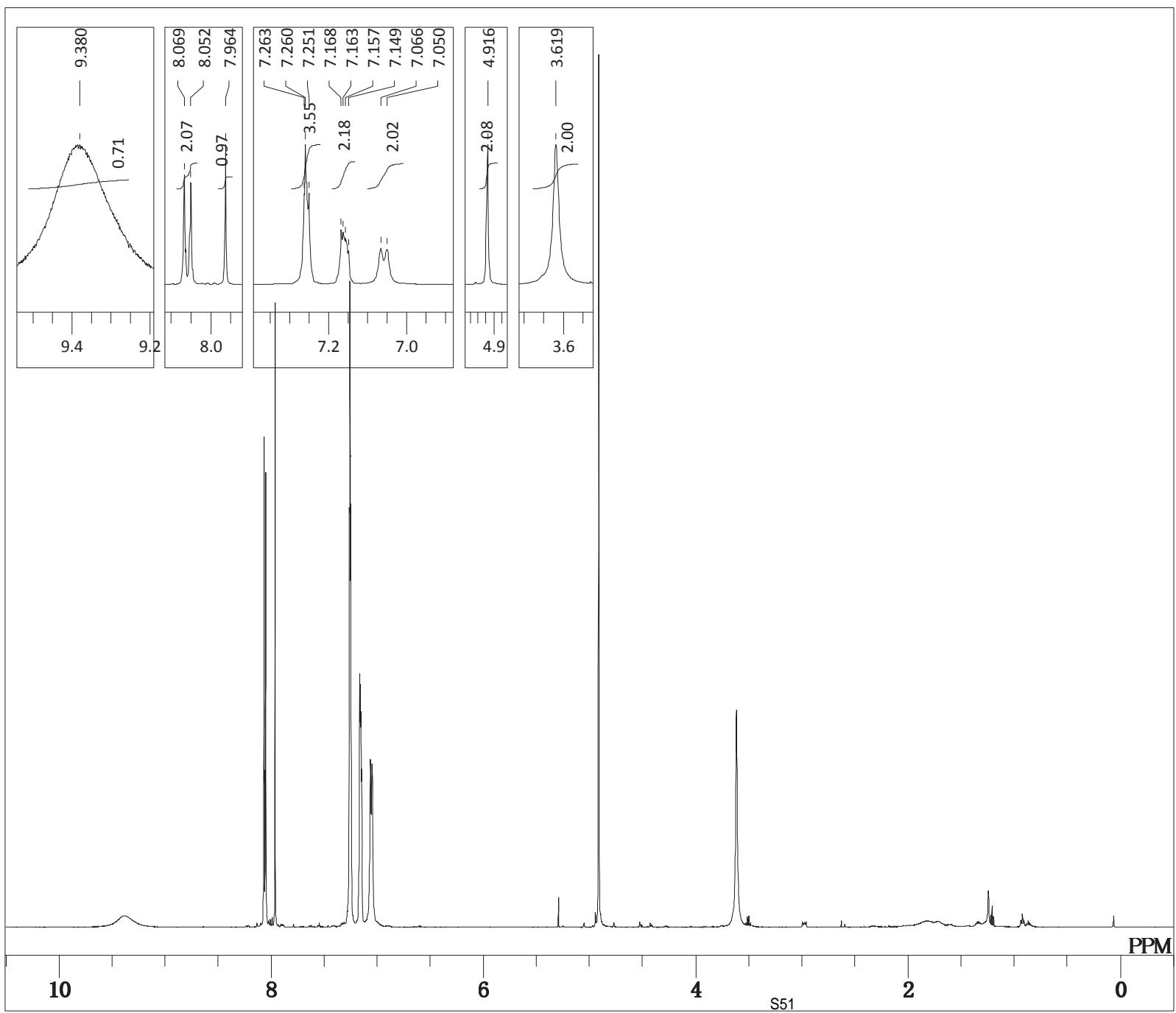




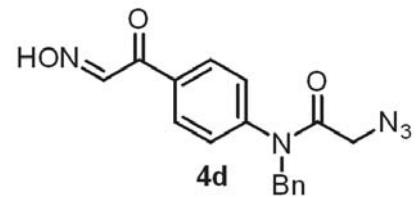


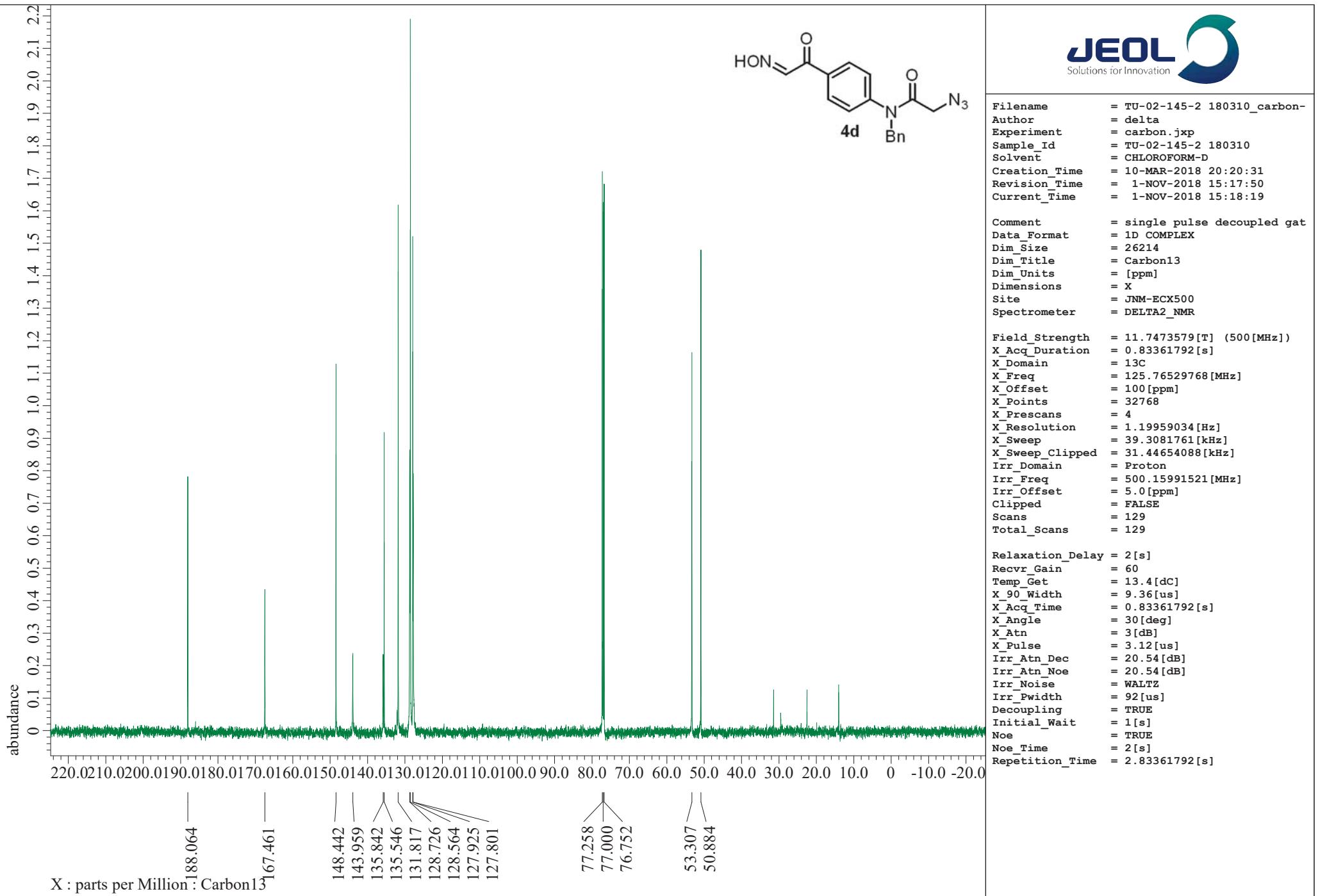


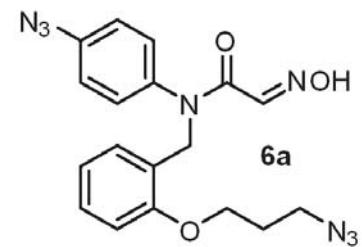
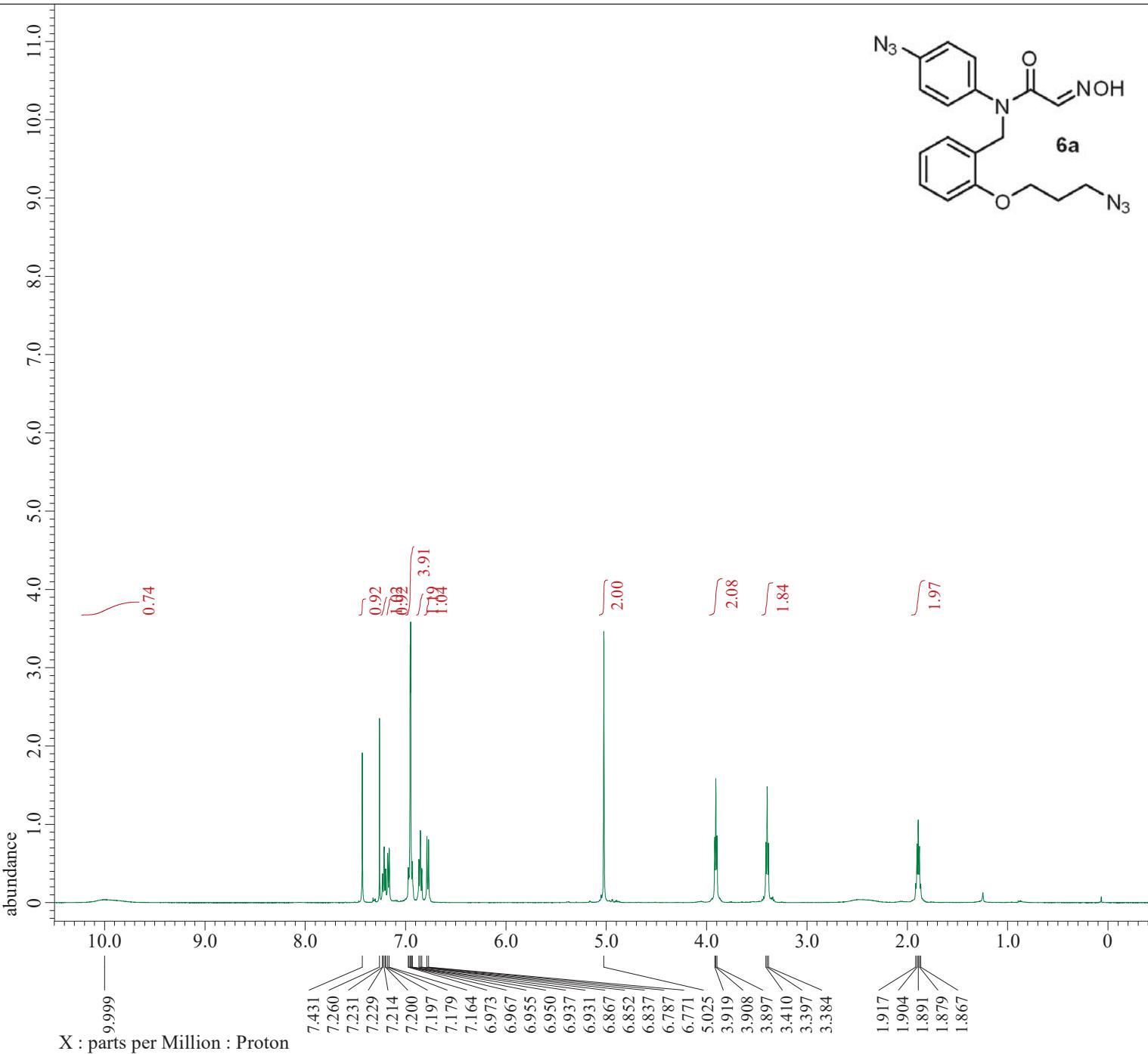




DFILE TU-02-045-2\_170707\_proton-1-  
 COMNT single\_pulse  
 DATIM 2017-07-07 13:16:27  
 OBNUC 1H  
 EXMOD proton.jxp  
 OBFRQ 500.16 MHz  
 OBSET 2.41 KHz  
 OBFIN 6.01 Hz  
 POINT 13107  
 FREQU 7507.51 Hz  
 SCANS 8  
 ACQTM 1.7459 sec  
 PD 5.0000 sec  
 PW1 6.85 usec  
 IRNUC 1H  
 CTEMP 17.3 c  
 SLVNT CDCL<sub>3</sub>  
 EXREF 7.26 ppm  
 BF 0.10 Hz  
 RGAIN 38



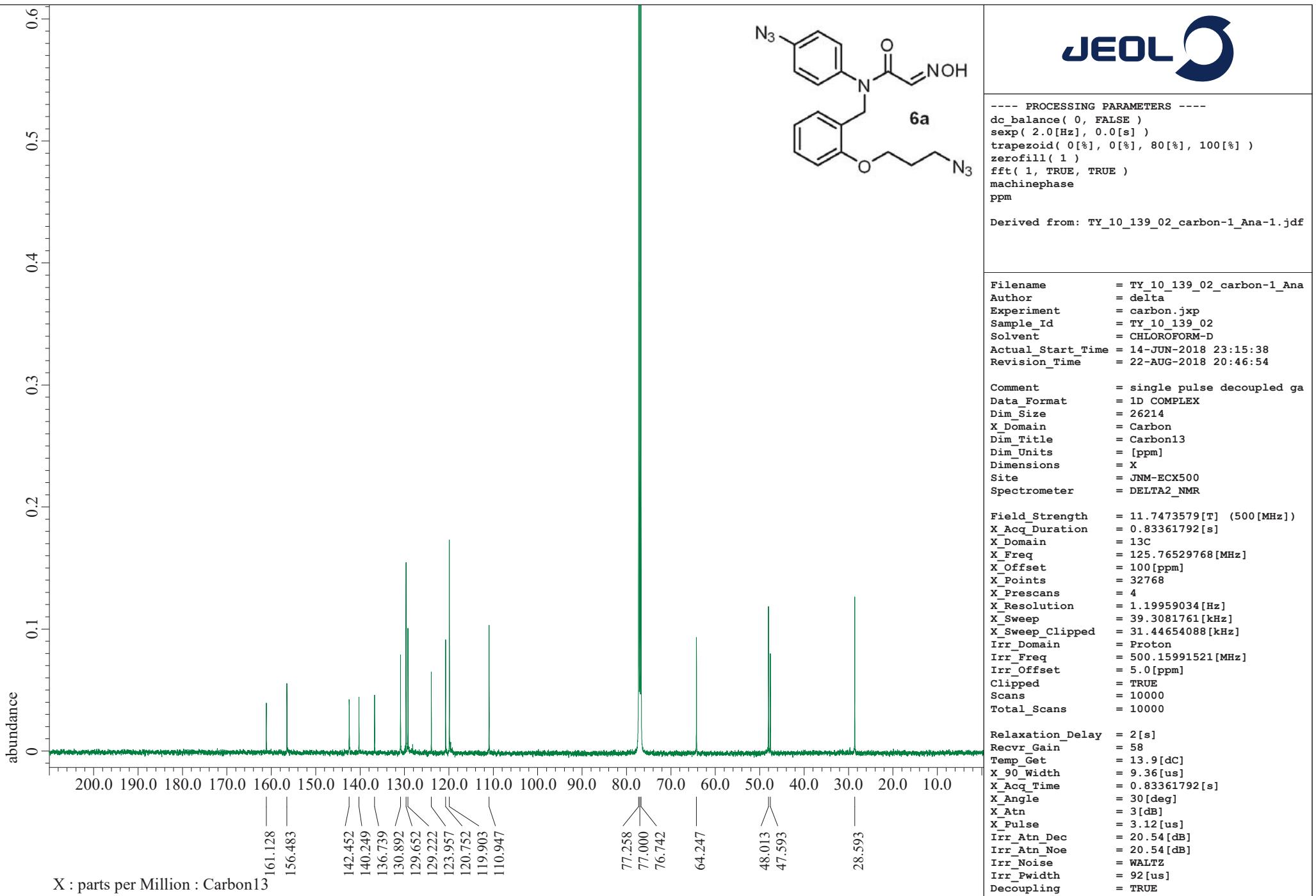


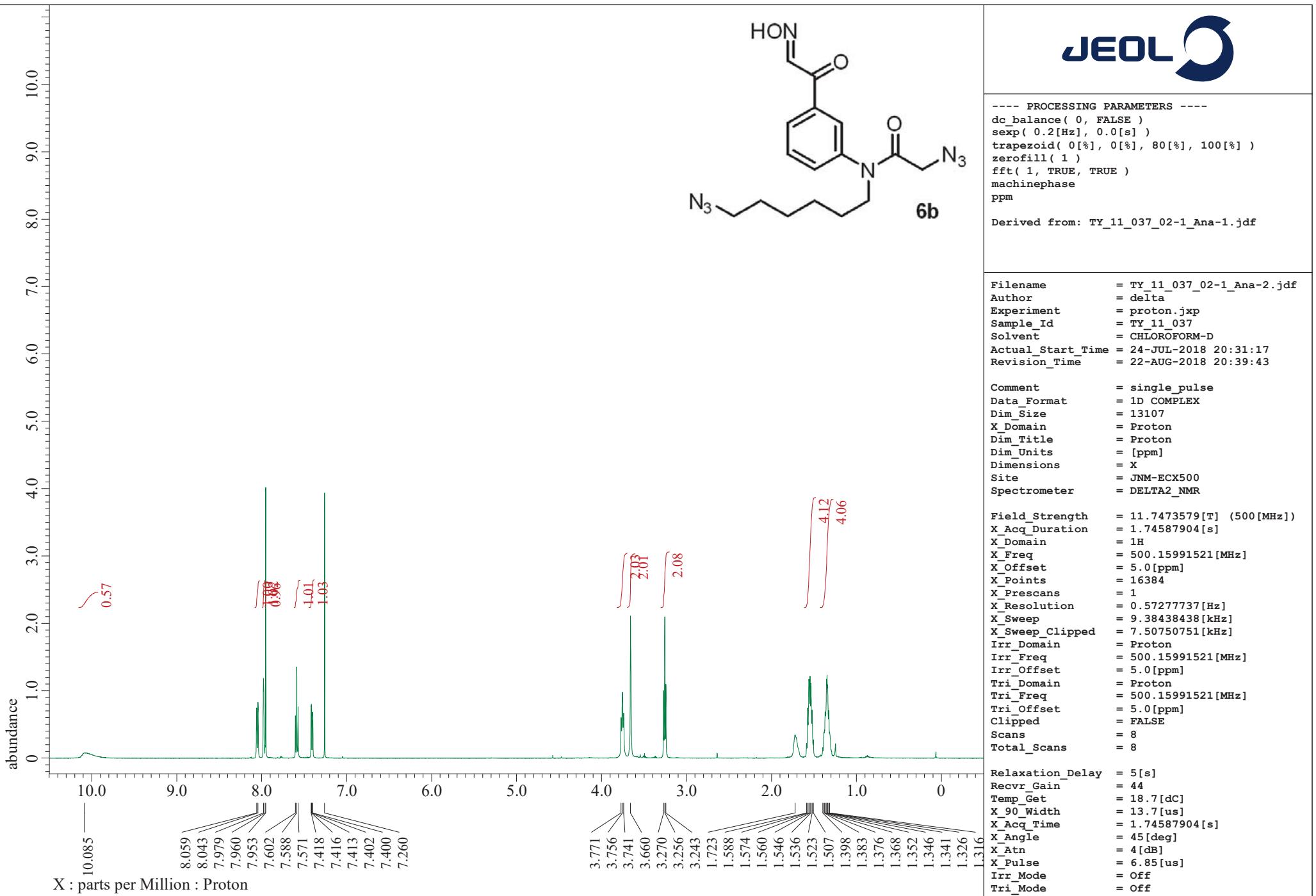


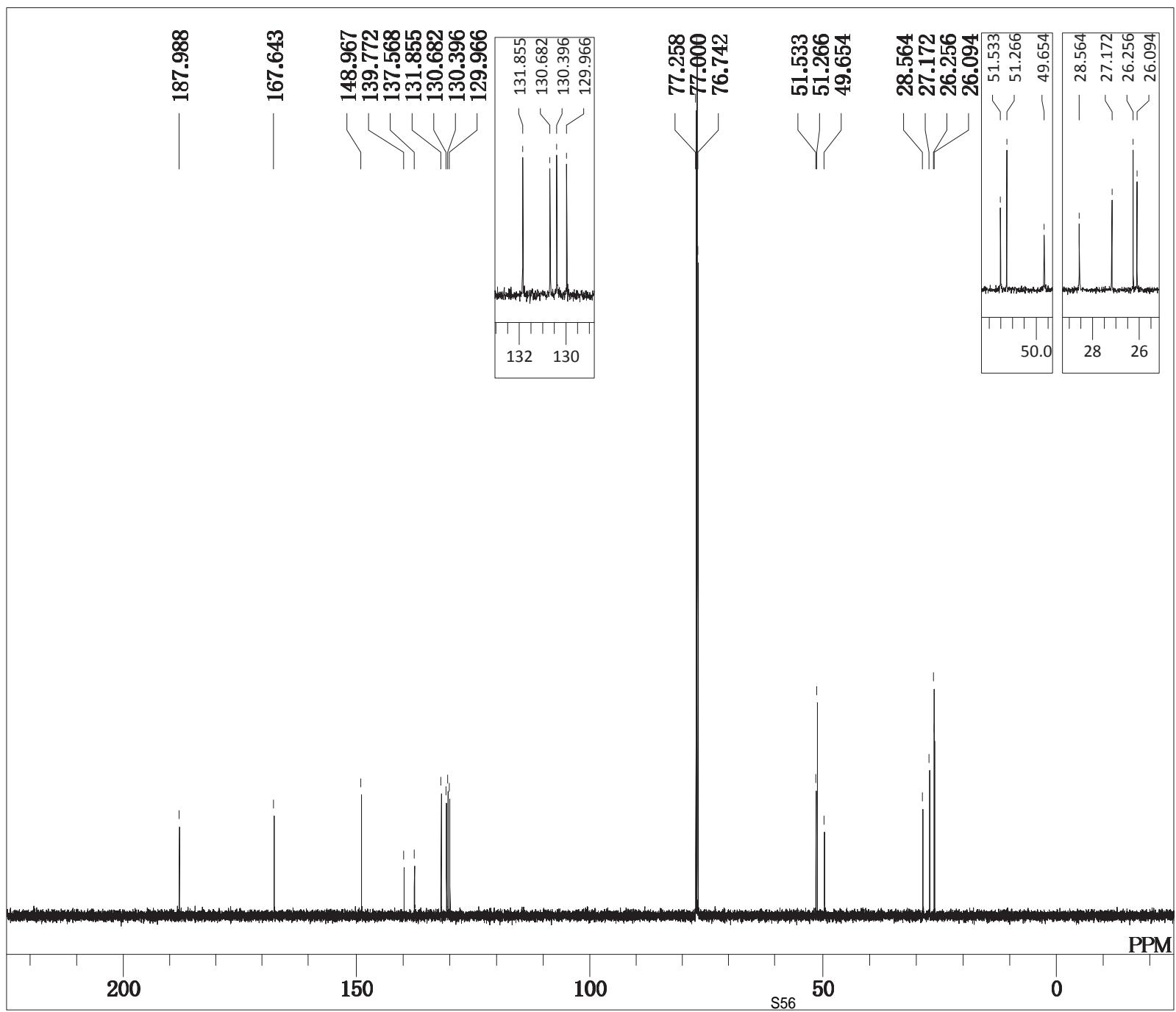
JEOL

---- PROCESSING PARAMETERS ----  
dc\_balance( 0, FALSE )  
sexp( 0.2[Hz], 0.0[s] )  
trapezoid( 0[%], 0[%], 80[%], 100[%] )  
zerofill( 1 )  
fft( 1, TRUE, TRUE )  
machinephase  
ppm  
Derived from: TY\_10\_139\_02\_proton-1\_Ana-3.jdf

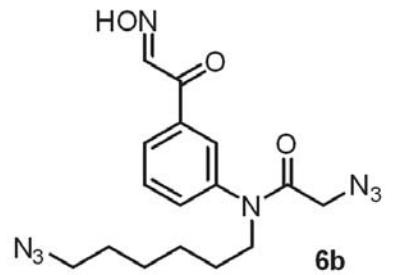
Filename = TY\_10\_139\_02\_proton-1\_Ana  
Author = delta  
Experiment = proton.jxp  
Sample\_Id = TY\_10\_139\_02  
Solvent = CHLOROFORM-D  
Actual\_Start\_Time = 14-JUN-2018 23:13:53  
Revision\_Time = 22-AUG-2018 20:45:57  
Comment = single\_pulse  
Data\_Format = 1D COMPLEX  
Dim\_Size = 13107  
X\_Domain = Proton  
Dim\_Title = Proton  
Dim\_Units = [ppm]  
Dimensions = X  
Site = JNM-ECX500  
Spectrometer = DELTA2\_NMR  
Field\_Strength = 11.7473579[T] (500[MHz])  
X\_Acq\_Duration = 1.74587904[s]  
X\_Domain = 1H  
X\_Freq = 500.15991521[MHz]  
X\_Offset = 5.0[ppm]  
X\_Points = 16384  
X\_Prescans = 1  
X\_Resolution = 0.57277737[Hz]  
X\_Sweep = 9.38438438[kHz]  
X\_Sweep\_Clipped = 7.50750751[kHz]  
Irr\_Domain = Proton  
Irr\_Freq = 500.15991521[MHz]  
Irr\_Offset = 5.0[ppm]  
Tri\_Domain = Proton  
Tri\_Freq = 500.15991521[MHz]  
Tri\_Offset = 5.0[ppm]  
Clipped = FALSE  
Scans = 8  
Total\_Scans = 8  
Relaxation\_Delay = 5[s]  
Recvrv\_Gain = 40  
Temp\_Get = 13.2[dC]  
X\_90\_Width = 13.7[us]  
X\_Acq\_Time = 1.74587904[s]  
X\_Angle = 45[deg]  
X\_Atn = 4[dB]  
X\_Pulse = 6.85[us]  
Irr\_Mode = Off  
Tri\_Mode = Off

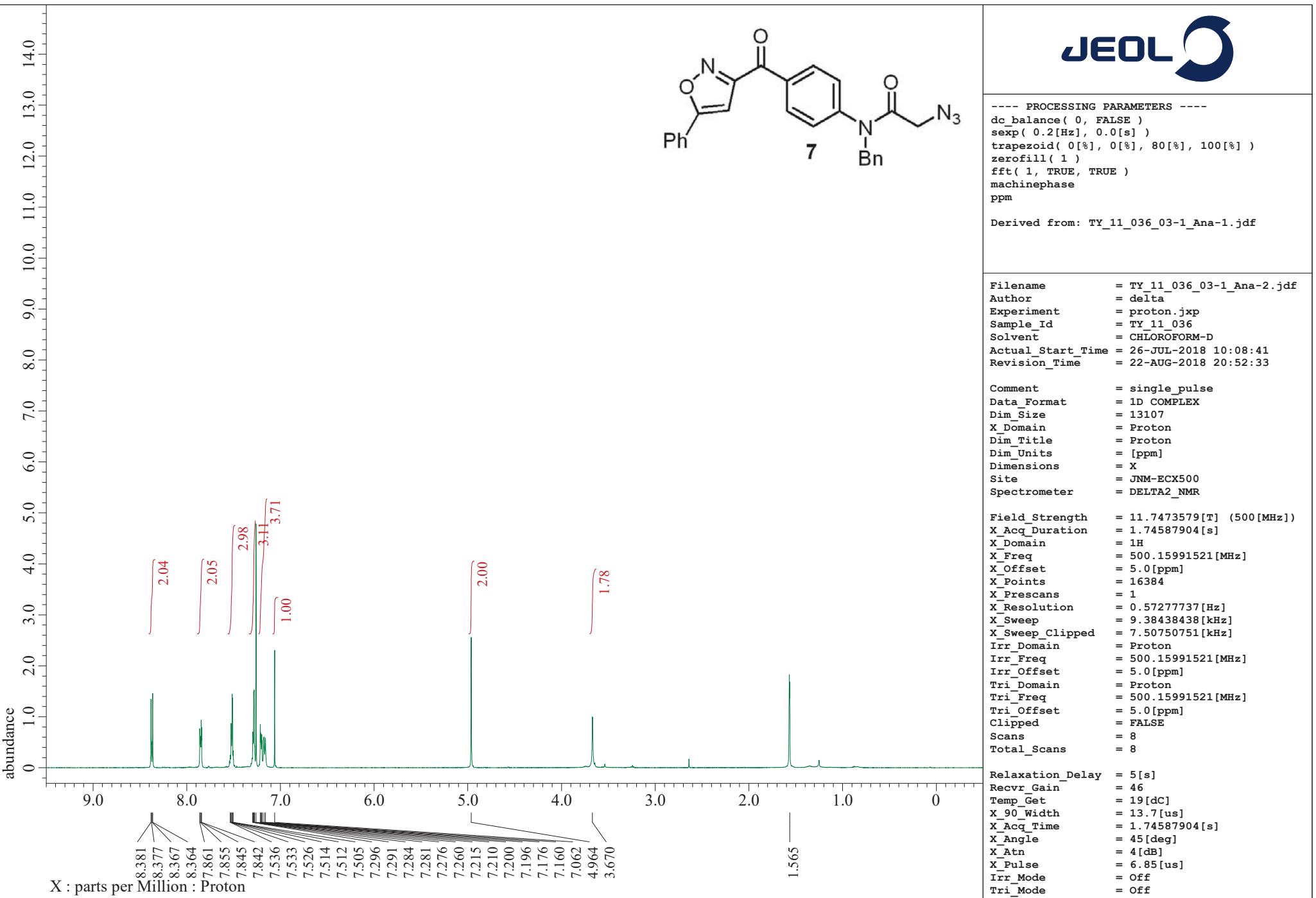


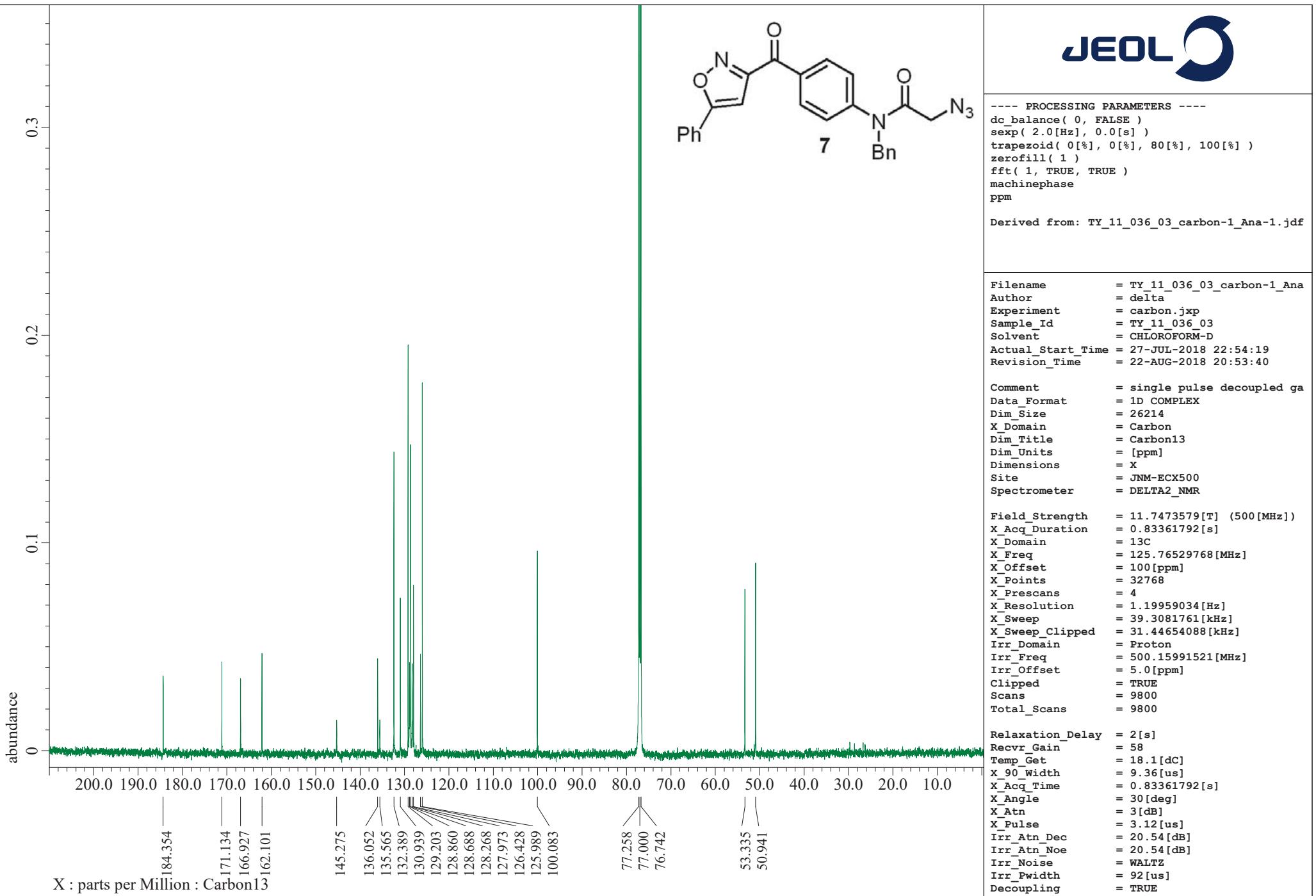


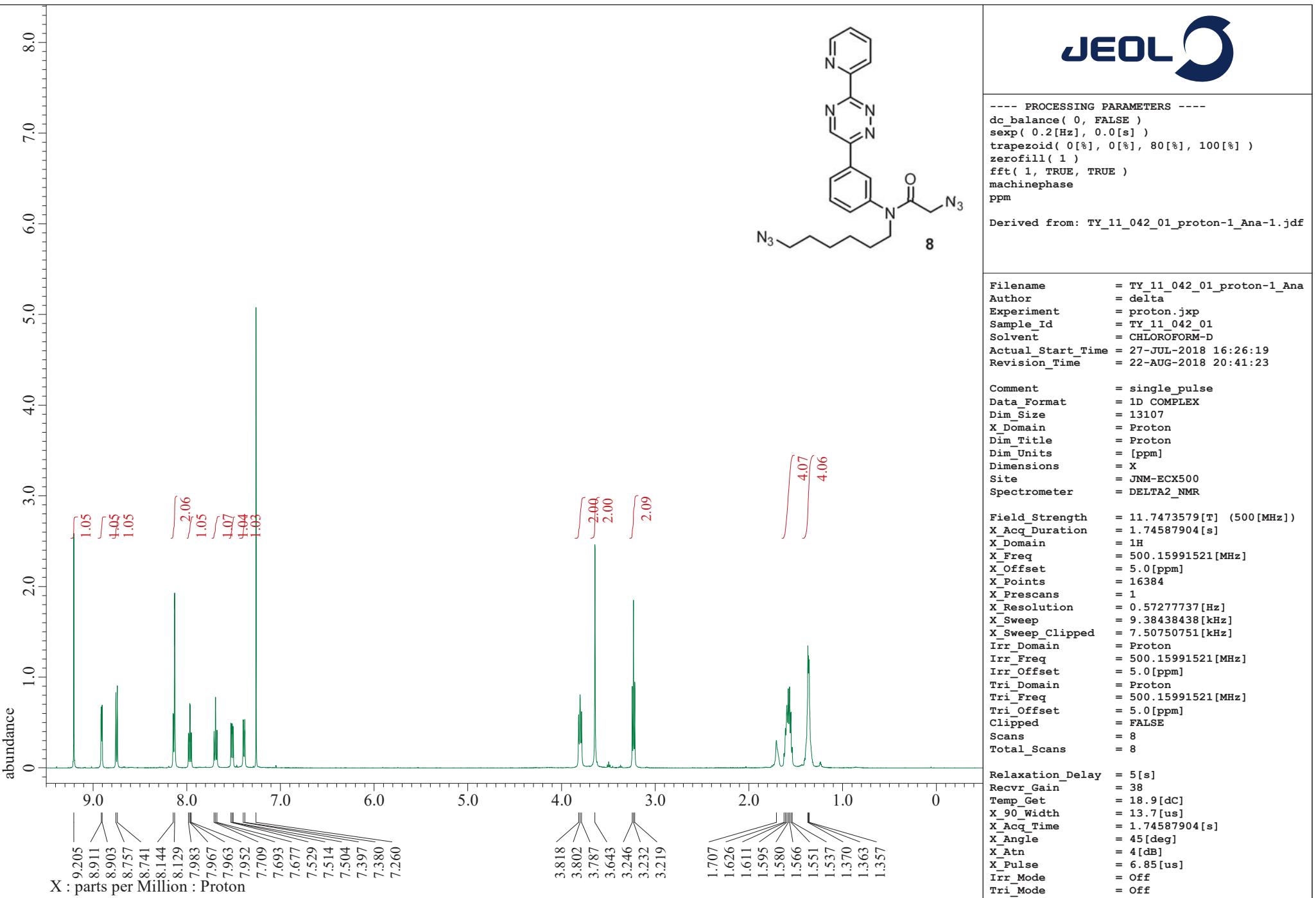


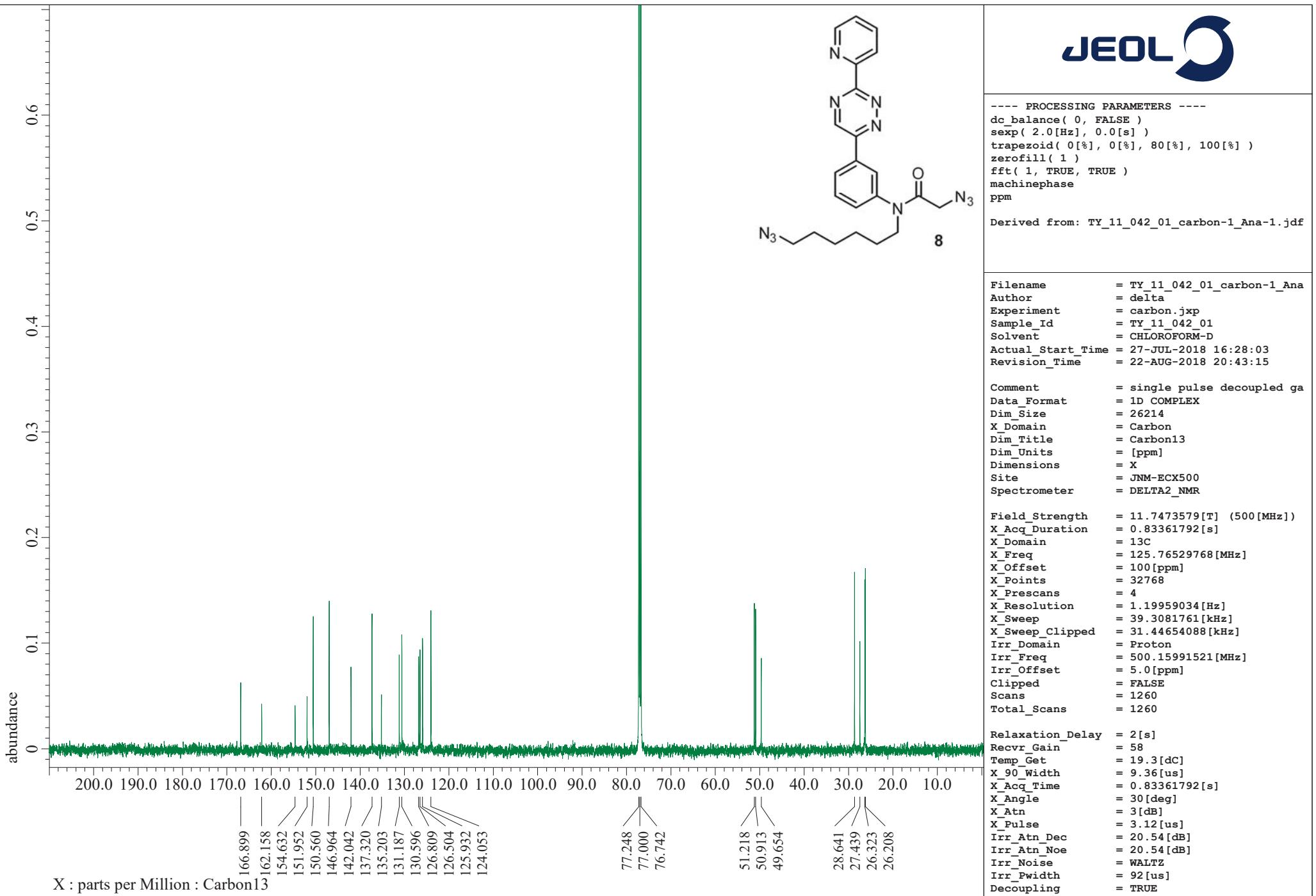
DFILE TU-02-092-5 180308\_carbon-1-  
 COMNT single pulse decoupled gated NO  
 DATIM 2018-03-08 12:12:47  
 OBNUC 13C  
 EXMOD carbon.jxp  
 OBFRQ 125.77 MHz  
 OBSET 7.87 KHz  
 OBFIN 4.21 Hz  
 POINT 32767  
 FREQU 39308.18 Hz  
 SCANS 366  
 ACQTM 0.8336 sec  
 PD 2.0000 sec  
 PW1 3.12 usec  
 1H  
 IRNUC 15.3 c  
 CTEMP CDCL3  
 SLVNT 77.00 ppm  
 EXREF 0.10 Hz  
 BF 60  
 RGAIN

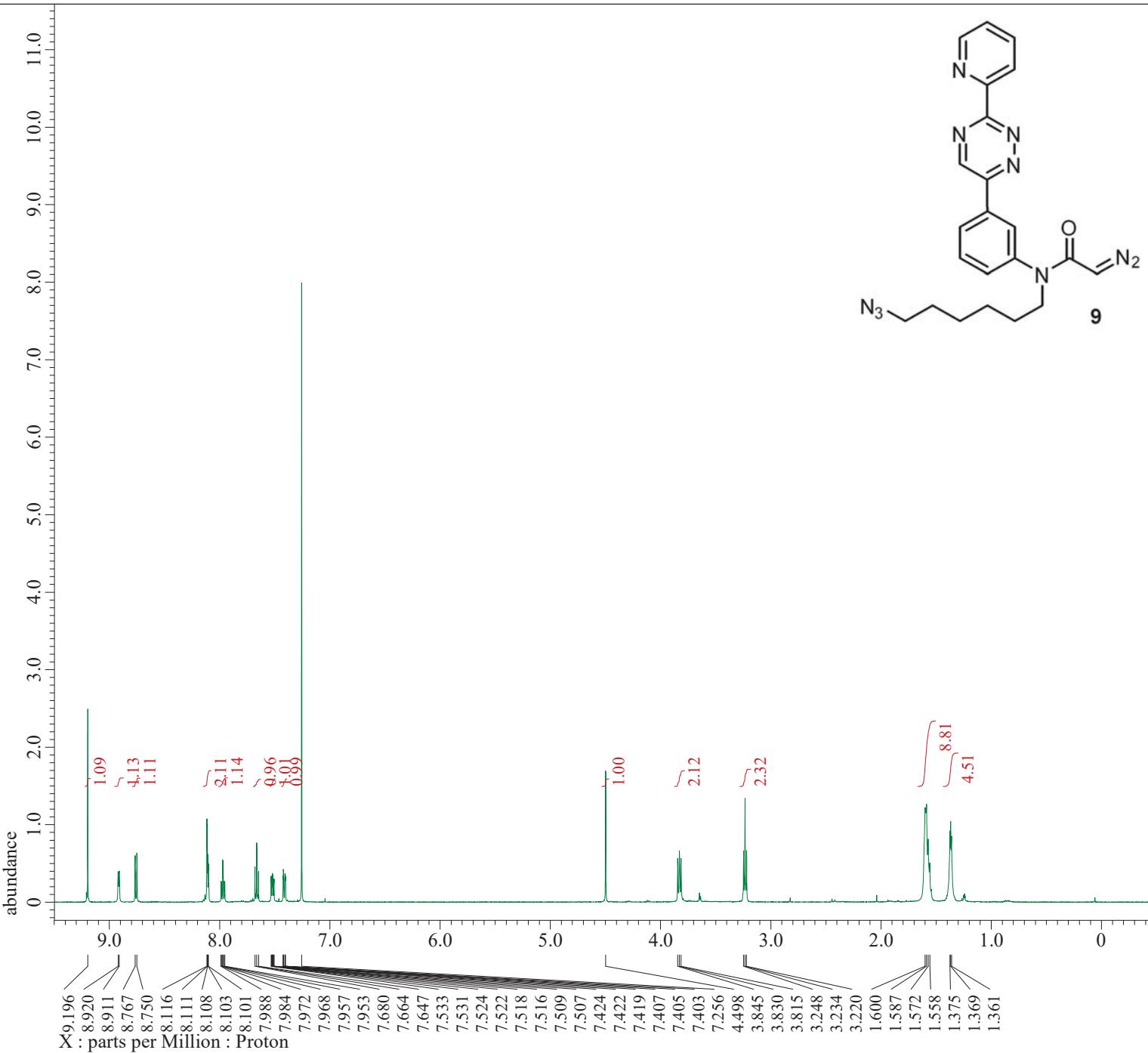








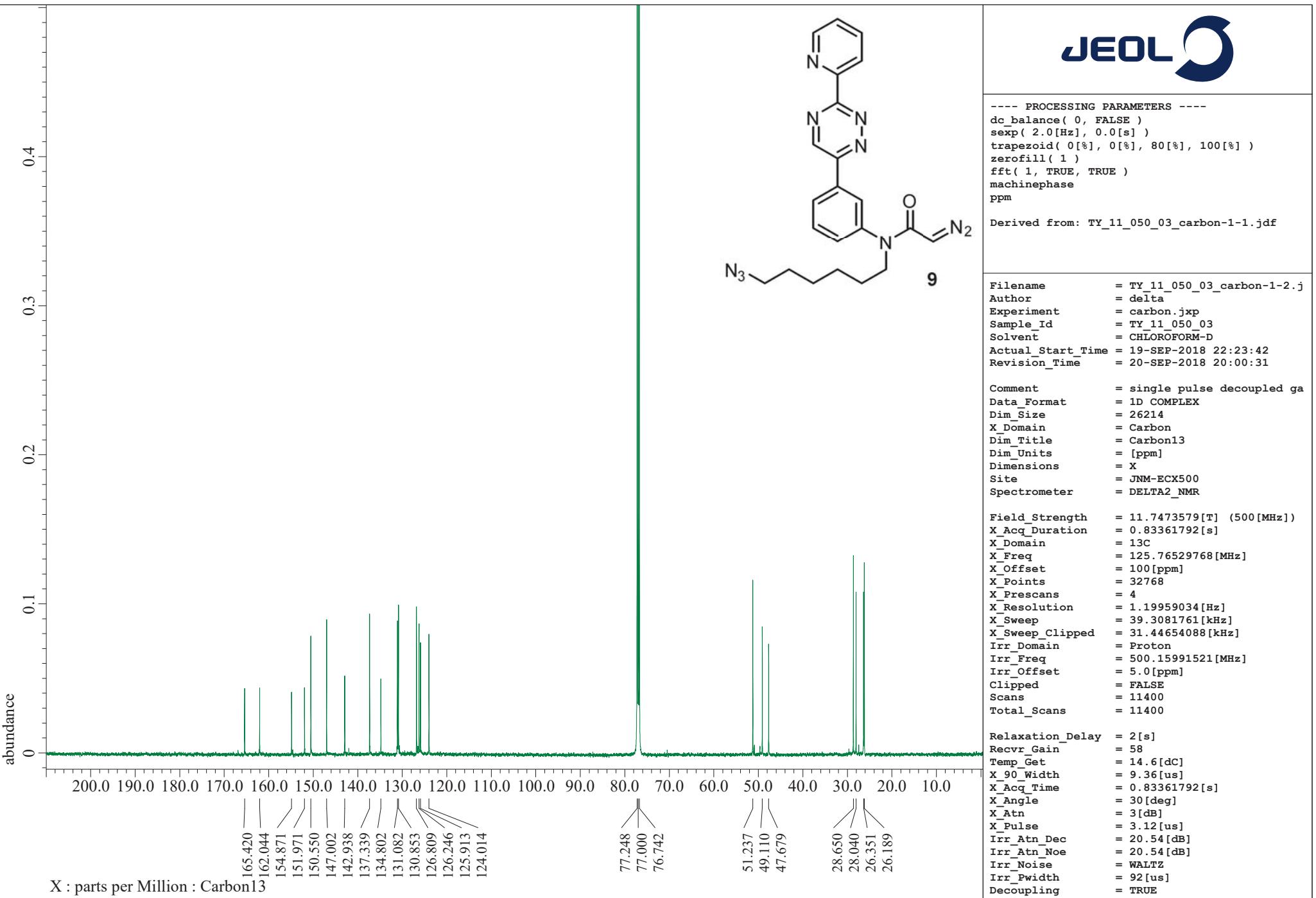


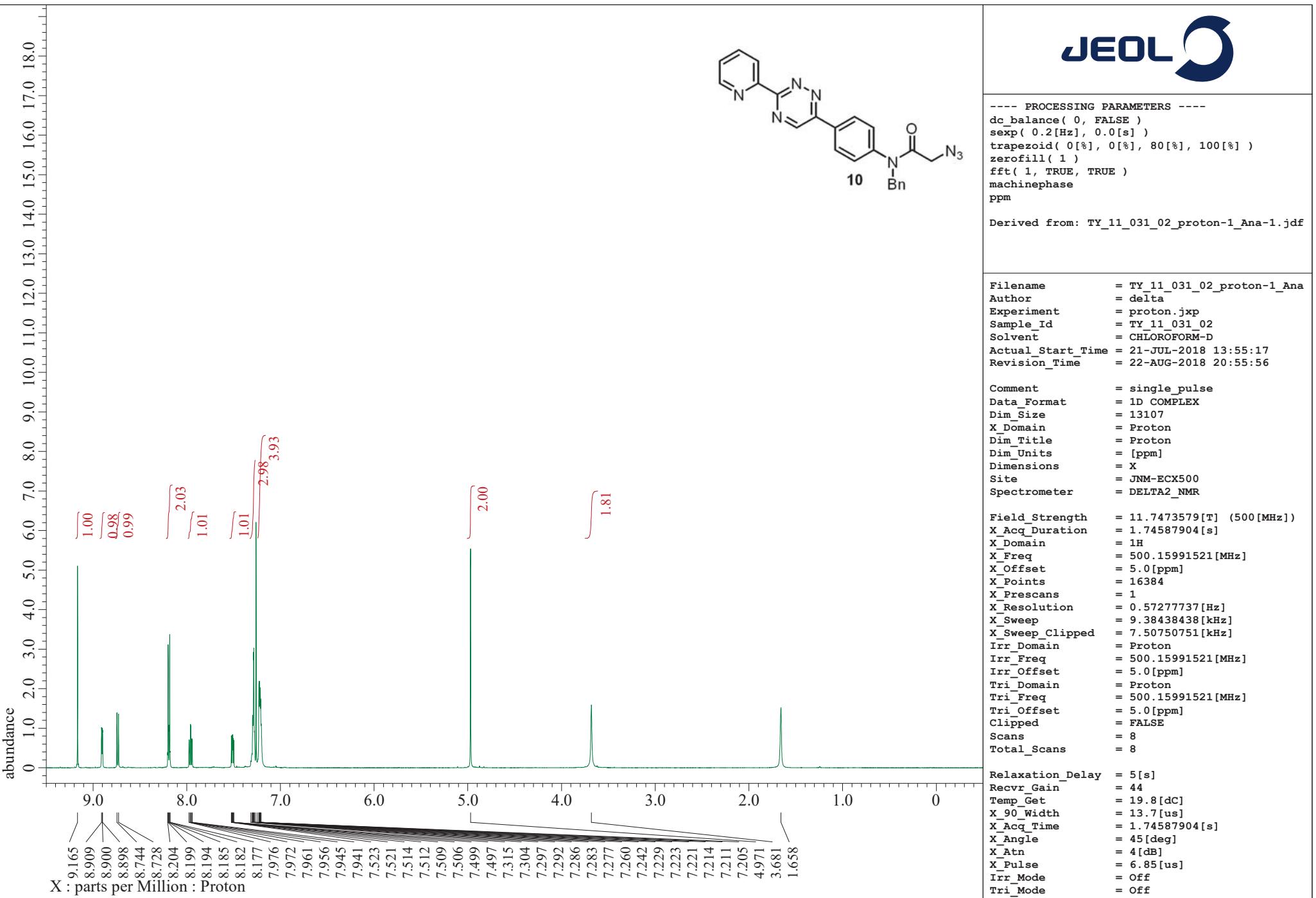


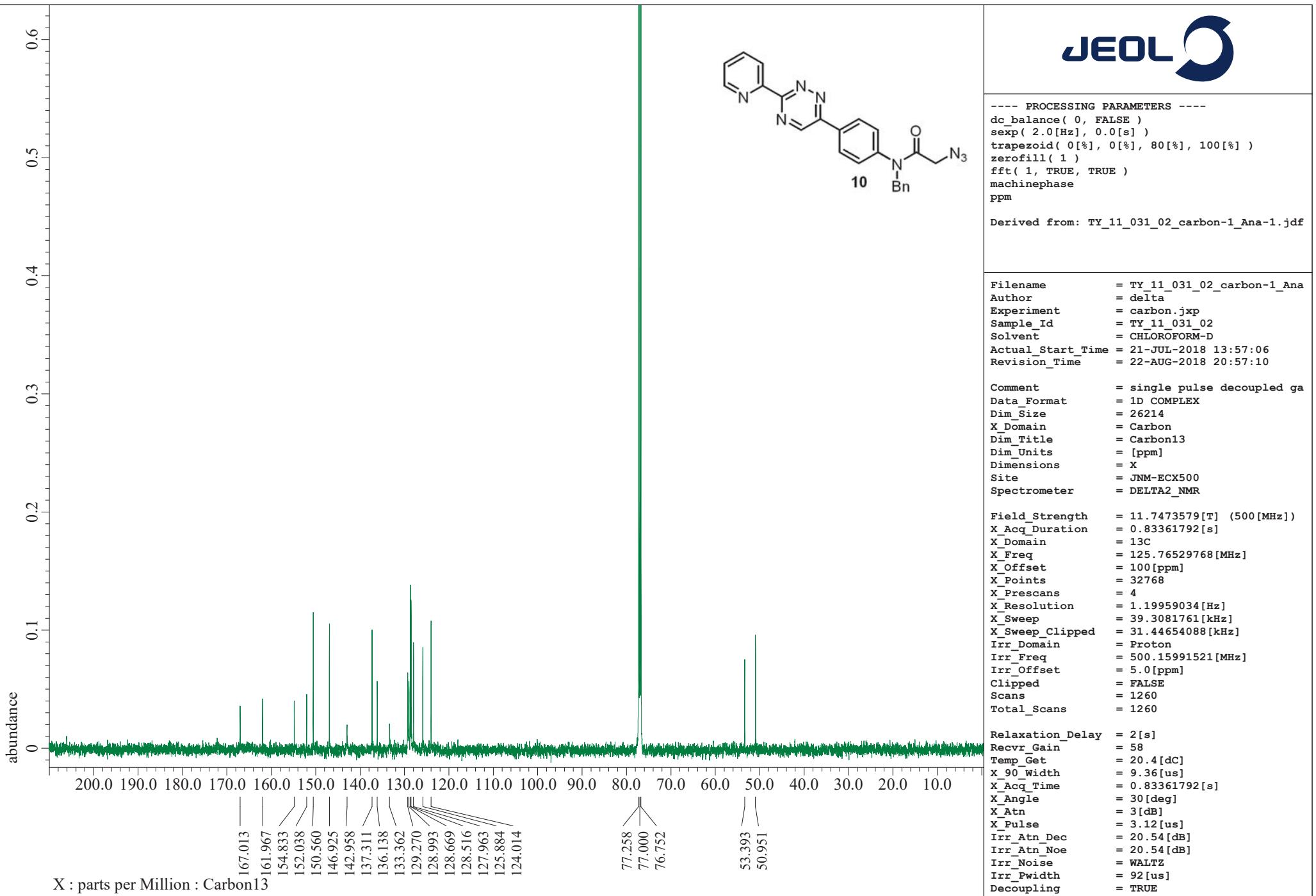
**JEOL**

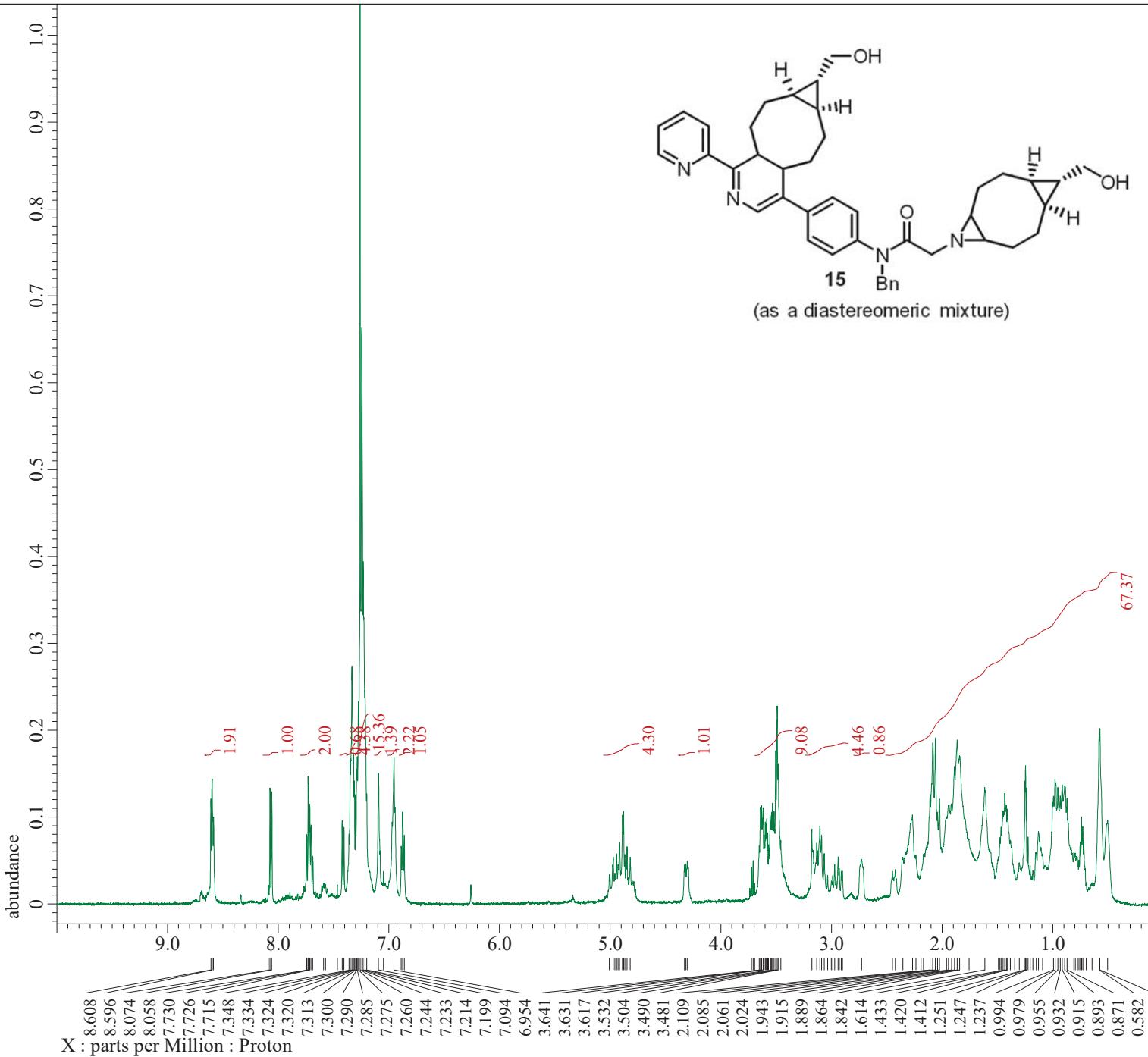
---- PROCESSING PARAMETERS ----  
dc\_balance( 0, FALSE )  
sexp( 0.2[Hz], 0.0[s] )  
trapezoid( 0[%], 0[%], 80[%], 100[%] )  
zerofill( 1 )  
fft( 1, TRUE, TRUE )  
machinephase  
ppm  
Derived from: TY\_11\_050\_03-1-1.jdf

Filename = TY\_11\_050\_03-1-2.jdf  
Author = delta  
Experiment = proton.jxp  
Sample\_Id = TY\_11\_050  
Solvent = CHLOROFORM-D  
Actual\_Start\_Time = 7-AUG-2018 10:46:32  
Revision\_Time = 20-SEP-2018 20:10:57  
Comment = single\_pulse  
Data\_Format = 1D COMPLEX  
Dim\_Size = 13107  
X\_Domain = Proton  
Dim\_Title = Proton  
Dim\_Units = [ppm]  
Dimensions = X  
Site = JNM-ECX500  
Spectrometer = DELTA2\_NMR  
Field\_Strength = 11.7473579[T] (500[MHz])  
X\_Acq\_Duration = 1.74587904[s]  
X\_Domain = 1H  
X\_Freq = 500.15991521[MHz]  
X\_Offset = 5.0[ppm]  
X\_Points = 16384  
X\_Prescans = 1  
X\_Resolution = 0.57277737[Hz]  
X\_Sweep = 9.38438438[kHz]  
X\_Sweep\_Clipped = 7.50750751[kHz]  
Irr\_Domain = Proton  
Irr\_Freq = 500.15991521[MHz]  
Irr\_Offset = 5.0[ppm]  
Tri\_Domain = Proton  
Tri\_Freq = 500.15991521[MHz]  
Tri\_Offset = 5.0[ppm]  
Clipped = FALSE  
Scans = 8  
Total\_Scans = 8  
Relaxation\_Delay = 5[s]  
Recvr\_Gain = 46  
Temp\_Get = 17.9[dC]  
X\_90\_Width = 13.7[us]  
X\_Acq\_Time = 1.74587904[s]  
X\_Angle = 45[deg]  
X\_Atn = 4[dB]  
X\_Pulse = 6.85[us]  
Irr\_Mode = Off  
Tri\_Mode = Off





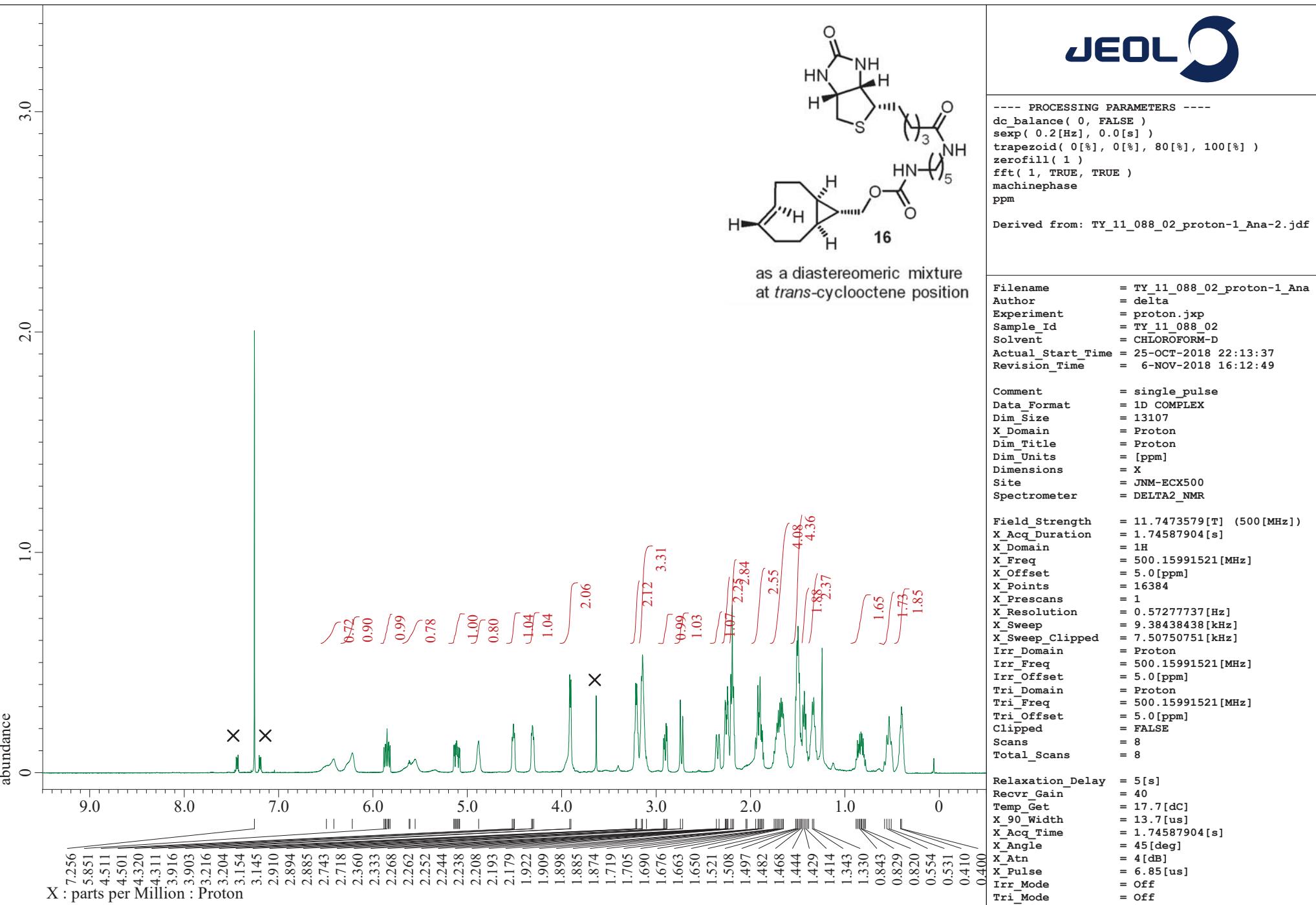


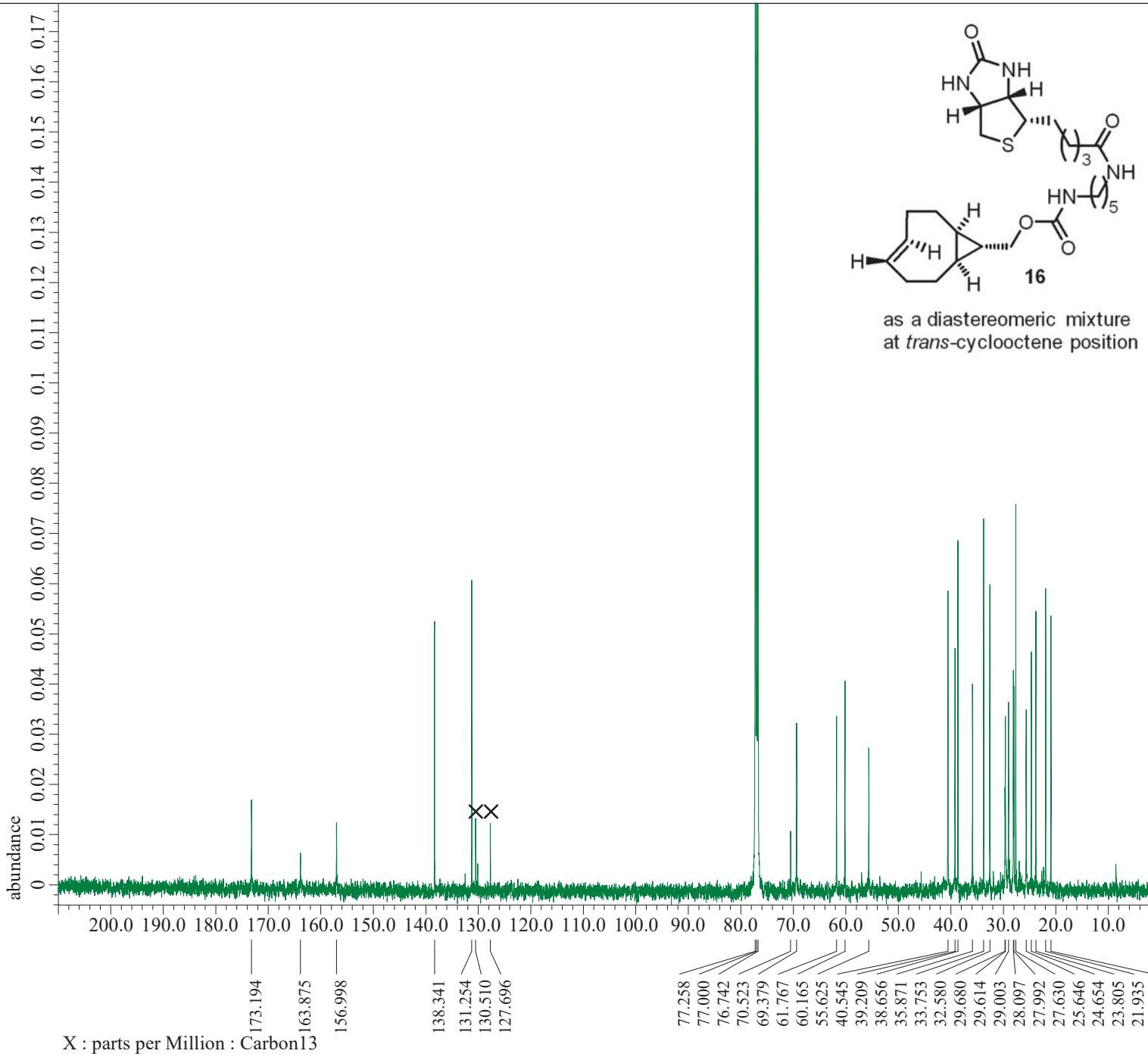


JEOL

---- PROCESSING PARAMETERS ----  
dc\_balance( 0, FALSE )  
sexp( 0.2[Hz], 0.0[s] )  
trapezoid( 0[%], 0[%], 80[%], 100[%] )  
zerofill( 1 )  
fft( 1, TRUE, TRUE )  
machinephase  
ppm  
Derived from: TY\_11\_053\_03-1\_Ana-1.jdf

Filename = TY\_11\_053\_03-1\_Ana-2.jdf  
Author = delta  
Experiment = proton.jxp  
Sample\_Id = TY\_11\_053  
Solvent = CHLOROFORM-D  
Actual\_Start\_Time = 7-AUG-2018 19:05:39  
Revision\_Time = 1-NOV-2018 11:10:26  
Comment = single\_pulse  
Data\_Format = 1D COMPLEX  
Dim\_Size = 13107  
X\_Domain = Proton  
Dim\_Title = Proton  
Dim\_Units = [ppm]  
Dimensions = X  
Site = JNM-ECX500  
Spectrometer = DELTA2\_NMR  
Field\_Strength = 11.7473579[T] (500[MHz])  
X\_Acq\_Duration = 1.74587904[s]  
X\_Domain = 1H  
X\_Freq = 500.15991521[MHz]  
X\_Offset = 5.0[ppm]  
X\_Points = 16384  
X\_Prescans = 1  
X\_Resolution = 0.57277737[Hz]  
X\_Sweep = 9.38438438[kHz]  
X\_Sweep\_Clipped = 7.50750751[kHz]  
Irr\_Domain = Proton  
Irr\_Freq = 500.15991521[MHz]  
Irr\_Offset = 5.0[ppm]  
Tri\_Domain = Proton  
Tri\_Freq = 500.15991521[MHz]  
Tri\_Offset = 5.0[ppm]  
Clipped = FALSE  
Scans = 8  
Total\_Scans = 8  
Relaxation\_Delay = 5[s]  
Recvr\_Gain = 42  
Temp\_Get = 18.4[dC]  
X\_90\_Width = 13.7[us]  
X\_Acq\_Time = 1.74587904[s]  
X\_Angle = 45[deg]  
X\_Atn = 4[dB]  
X\_Pulse = 6.85[us]  
Irr\_Mode = Off  
Tri\_Mode = Off

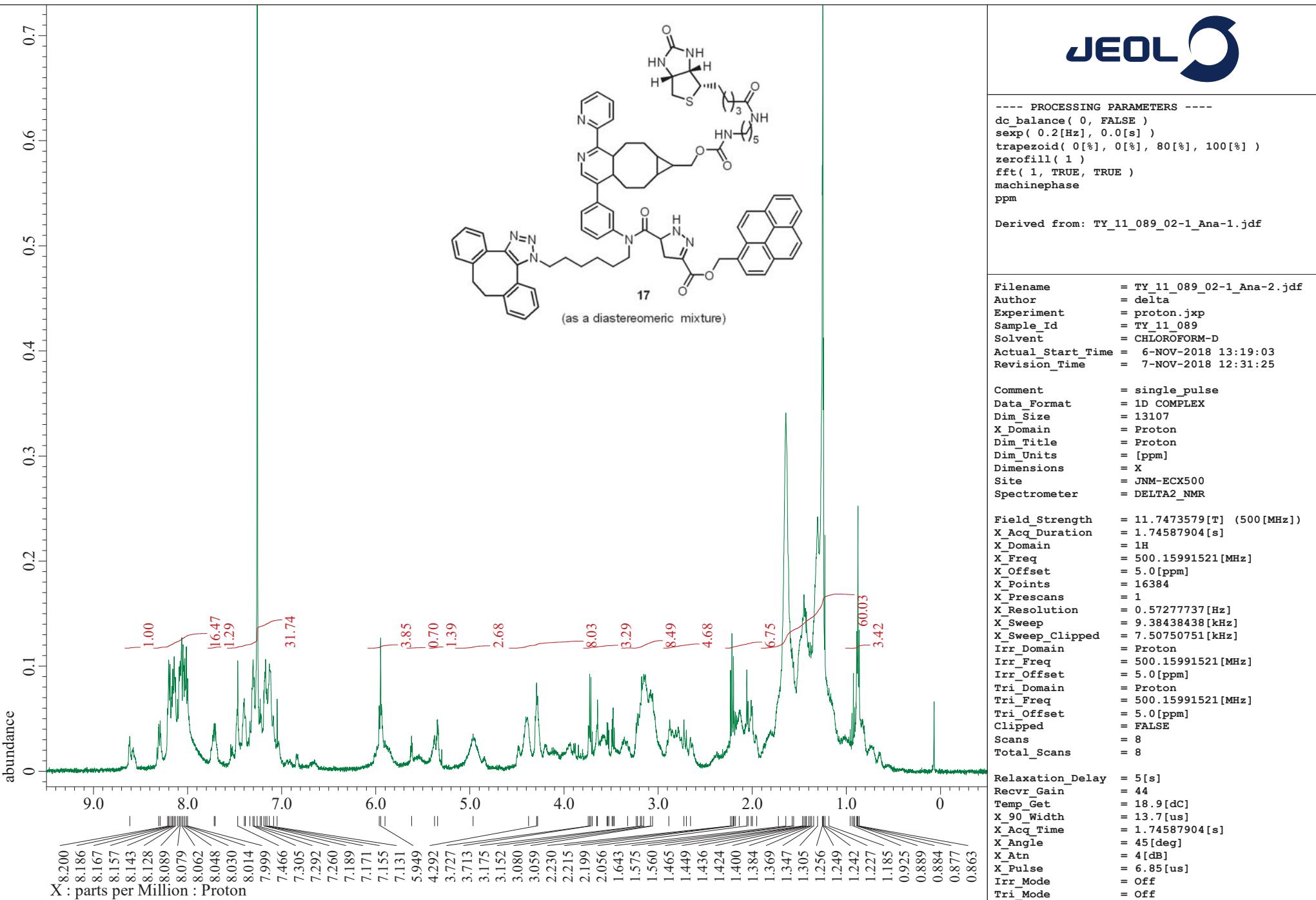




JEOL

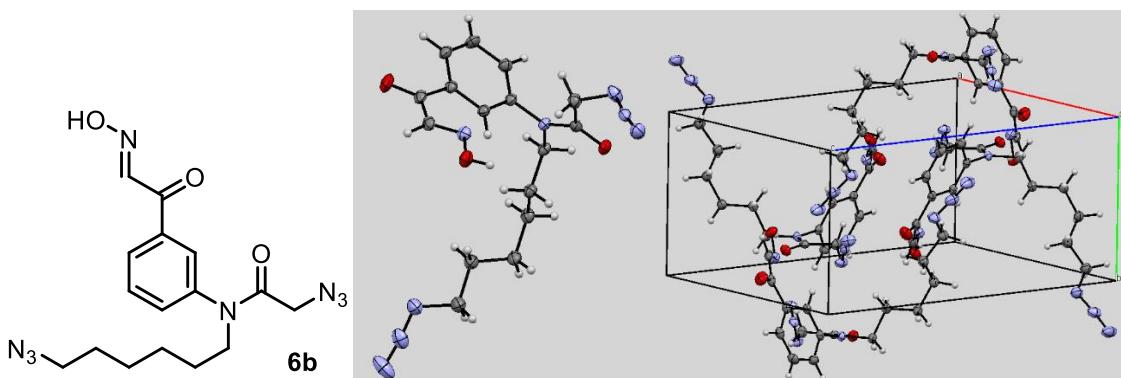
---- PROCESSING PARAMETERS ----  
dc\_balance( 0, FALSE )  
sexp( 2.0[Hz], 0.0[s] )  
trapezoid( 0[%], 0[%], 80[%], 100[%] )  
zerofill( 1 )  
fft( 1, TRUE, TRUE )  
machinephase  
ppm  
Derived from: TY\_11\_088\_02\_carbon-1\_Ana-2.jdf

Filename = TY\_11\_088\_02\_carbon-1\_Ana  
Author = delta  
Experiment = carbon.jxp  
Sample\_Id = TY\_11\_088\_02  
Solvent = CHLOROFORM-D  
Actual\_Start\_Time = 25-OCT-2018 22:15:06  
Revision\_Time = 6-NOV-2018 14:54:50  
Comment = single pulse decoupled ga  
Data\_Format = 1D COMPLEX  
Dim\_Size = 26214  
X\_Domain = Carbon  
Dim\_Title = Carbon13  
Dim\_Units = [ppm]  
Dimensions = X  
Site = JNM-ECX500  
Spectrometer = DELTA2\_NMR  
Field\_Strength = 11.7473579[T] (500[MHz])  
X\_Acq\_Duration = 0.83361792[s]  
X\_Domain = 13C  
X\_Freq = 125.76529768[MHz]  
X\_Offset = 100[ppm]  
X\_Points = 32768  
X\_Prescans = 4  
X\_Resolution = 1.19959034[Hz]  
X\_Sweep = 39.3081761[kHz]  
X\_Sweep\_Clipped = 31.44654088[kHz]  
Irr\_Domain = Proton  
Irr\_Freq = 500.15991521[MHz]  
Irr\_Offset = 5.0[ppm]  
Clipped = FALSE  
Scans = 10780  
Total\_Scans = 10780  
Relaxation\_Delay = 2[s]  
Recv\_Gain = 60  
Temp\_Get = 20.2[dC]  
X\_90\_Width = 9.36[us]  
X\_Acq\_Time = 0.83361792[s]  
X\_Angle = 30[deg]  
X\_Atn = 3[dB]  
X\_Pulse = 3.12[us]  
Irr\_Atn\_Dec = 20.54[dB]  
Irr\_Atn\_Noe = 20.54[dB]  
Irr\_Noise = WALTZ  
Irr\_Pwidth = 92[us]  
Decoupling = TRUE



**[8] X-ray Crystallographic Analysis Information**  
 (ORTEP thermal ellipsoids at 50% probability)

**(1) 6b (CCDC No. 1879115)**



Empirical formula	C <sub>16</sub> H <sub>20</sub> N <sub>8</sub> O <sub>3</sub>
Formula weight	372.39
Temperature	-150.0 °C
Wavelength	MoKα ( $\lambda = 0.71075 \text{ \AA}$ )
Crystal system	monoclinic
Space group	P2 <sub>1</sub> /c (#14)
Unit cell dimensions	$a = 13.5247(4) \text{ \AA}$ $b = 7.64772(17) \text{ \AA}$ $\beta = 97.938(7)^\circ$ $c = 17.7480(5) \text{ \AA}$
Volume	$V = 1818.14(9) \text{ \AA}^3$
Z	4
Density (calculated)	1.360 g/cm <sup>3</sup>
$2\theta_{\max}$	50.7 °
Absorption coefficient $\mu$ for Mo-Kα	0.992 cm <sup>-1</sup>
$F(000)$	784.00
Crystal size	0.140 × 0.030 × 0.030 mm
No. of reflection collected	Total: 24512 Unique: 3334 ( $R_{\text{int}} = 0.0332$ )
Transmission factor	min: 0.836, max: 0.997
Refinement method	Full-matrix least-squares on $F^2$
Goodness-of-fit on $F^2$	1.046
Final R indices [ $I > 2\sigma(I)$ ]	$R_1 = 0.0359$
R indices (all data)	$R_1 = 0.0454$ $wR_2 = 0.0822$
Largest diff. peak and hole	-0.17 and 0.21 e·Å <sup>-3</sup>