Supporting Information

Guest-host interactions of nanoconfined anti-cancer drug in metal-organic framework exposed by Terahertz dynamics

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1. Materials synthesis

1.1. Mechanochemistry of HKUST-1 and encapsulation of 5-FU

The HKUST-1 MOF material was synthesized *via* a manual grinding process. $Cu(NO_3)_3 \cdot 3H_2O$ [copper(II) nitrate trihydrate] (3 mmol) and H₃BTC [benzene-1,3,5-tricarboxylic acid] (2 mmol) were placed in agate mortar and manually ground for 10 min. The product was washed by centrifugation (8000 rpm for 10 min) with methanol to remove unreacted components. HKUST-1 crystals were then dried at room temperature and activated at 100 °C.

Two guest@MOF composite samples were synthesized as follows. For the (one-pot) *in situ* encapsulation of 5-FU, analogous procedure was followed with the addition of 3.5 mmol of 5-FU during the grinding process, producing the 5-FU@HKUST-1_IN particles. *Ex situ* encapsulation of 5-FU was performed by immersion of pre-activated HKUST-1 in a saturated methanolic drug solution under continuous stirring for 48 h. The drug-loaded 5-FU@HKUST-1_EX particles were separated by centrifugation. Pristine HKUST-1 and drug-loaded samples were washed by centrifugation (8000 rpm for 10 min) with methanol and then activated at 90 °C for 12 h in a vacuum oven.

| | Sample | Synthesis method | Drug incorporation strategy |
|---|-----------------|------------------|-----------------------------|
| 1 | HKUST-1 | Manual grinding | - |
| 2 | 5-FU@HKUST-1_IN | Manual grinding | in situ encapsulation |
| 3 | 5-FU@HKUST-1_EX | Manual grinding | ex situ encapsulation |

Table S1: Description of samples used in this study.

2. Materials characterization

2.1. Inelastic neutron scattering

Inelastic neutron scattering (INS) measurements were performed using the TOSCA¹ spectrometer at the ISIS Pulsed Neutron and Muon Source, Rutherford Appleton Laboratory (Chilton, UK). The high-resolution ($\Delta E/E \sim 1.25\%$) and broadband (0-4000 cm⁻¹) spectra of each sample (~1 g) were acquired at ~10 K.

TOSCA is an indirect geometry time-of-flight spectrometer where a pulsed, polychromatic beam of neutrons collide with the sample at a distance of ~17 m from the source. The neutrons scattered from the sample were Bragg reflected by a pyrolytic graphite analyser, while higher-order reflections beyond (002) were blocked by a cooled (T < 30 K) Beryllium filter in order to define the final energy. Neutrons with final energy of ~32 cm⁻¹ were passed towards the detector array composed by thirteen ³He tubes with effective length of 250 mm. Five banks were located in forward direction (scattering angle ~45°) and five in backwards direction (~135°). The use of a low final energy translated into a direct relationship between energy transfer ($E_{\rm T}$, cm⁻¹) and momentum transfer (Q, Å⁻¹) such that $E_{\rm T} \approx 16Q^2$. Energy transfer and spectral intensity, i.e. $S(Q, \omega)$, were then obtained using the Mantid software.² Each sample was wrapped in 4 cm × 4.6 cm aluminium sachet and placed into a 2.0 mm spaced flat aluminium cell, which was sealed with indium wire. Sample preparation and cell loading into the cell took place in a glovebox to avoid moisture uptake by the sample. To reduce the effect of the Debye-Waller factor on the experimental spectral intensity and allow comparison with the theoretical spectra, the sample cell was cooled to ~ 10 K by a closed cycle refrigerator (CCR). The INS spectra were collected under vacuum over a duration of 4-6 hours.

The neutron guide upgrade of the TOSCA spectrometer, completed in 2017, has increased the neutron flux at the sample position by as much as 82 times. This upgrade improves the performance through faster measurements and by reducing the required sample mass.³

2.2. X-ray diffraction

The powder samples were analyzed by X-ray diffraction (XRD) using the Rigaku MiniFlex diffractometer with a Cu K_{α} source (1.541 Å). Samples were weighed before measurement and diffraction data was collected from 3° to 40°, using a 0.02° step size and 1.0° min⁻¹ step speed. The patterns were then normalized with respect to the most intense peak and used for phase identification.

2.3. Thermogravimetric analysis

Thermogravimetric analysis (TGA) was performed using TGA-Q50 (TA instruments). Approximately 4 mg of each sample was placed in a platinum pan (maximum volume 50 μ L) and heated from 50 °C to 500 °C with a heating rate of 10 °C min⁻¹. The measurements were performed under a dry nitrogen flow of 40 mL min⁻¹.

2.4. Attenuated total reflectance Fourier transform infrared spectroscopy

Attenuated total reflectance Fourier transform infrared spectroscopy (ATR-FTIR) spectra were acquired at room temperature with a Nicolet iS10 FTIR spectrometer with an ATR sample holder. The spectra were collected in the range of 750-2000 cm⁻¹ with a resolution of 0.5 cm⁻¹ and normalized in respect to the most intense vibrational peak to facilitate comparison across the different samples under study.

2.5. Scanning electron microscopy and atomic force microscopy

Analyses of the morphology and particle size determination were carried out by scanning electron microscopy (SEM) and atomic force microscopy (AFM). SEM images were obtained using Carl Zeiss EVO LS15 at 15 keV under high vacuum. Atomic force microscopy (AFM) was performed using the Veeco Dimension 3100 AFM equipped with an in-line optical zoom microscope with colour CCD camera for precise placement of probe onto the sample. The microscope was operated under the tapping mode, equipped with a Tap300G silicon probe with resonance frequency of 30 kHz, spring constant of 40 Nm⁻¹, and tip radius < 10 nm.

2.6. Specific surface area measurements

The Brunauer-Emmett-Teller (BET) specific surface area of samples was determined from nitrogen adsorption-desorption isotherms at 77 K, measured with Quantachrome Nova 1200. The isotherms were obtained using a Ø9 mm sample cell containing 60-100 mg of samples under study. The outgassing temperature was 90 °C during sample activation under vacuum.

2.7. Ab initio density functional theory calculations

Density functional theory (DFT) calculations to generate the INS spectra of HKUST-1 were obtained from the work published by Ryder and Tan.⁴ The vibrational calculation was carried out at the B3LYP level of theory using the periodic CRYSTAL14 code. We used the DFT

output file of this study to generate the INS spectrum using the Mantid software² through the AbINS extension.⁵ During the spectrum generation a total cross section was considered with a quantum order events number of 1.

2.8. Calculation of area under INS curve

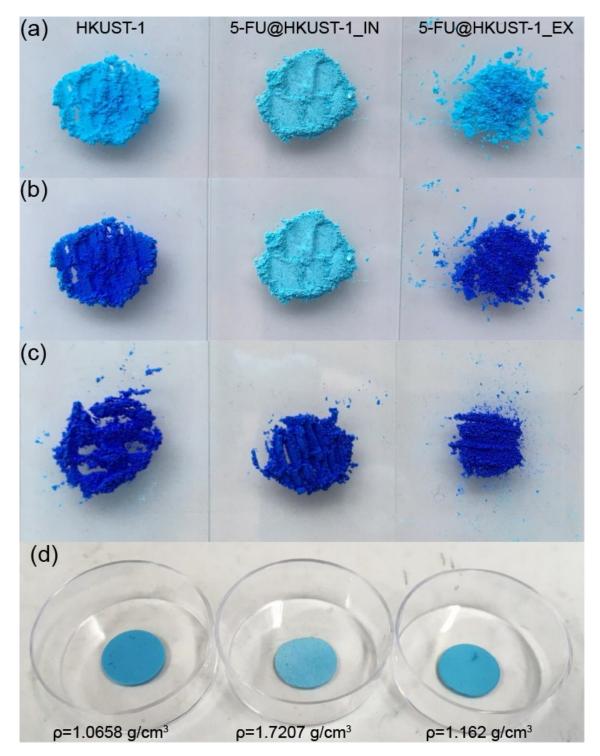
The Integrate Gadget in OriginPro was used to perform the numerical integration on the INS spectra to determine the area under the curve of specific vibrational modes. The range of data was selected to include the peaks of the vibrational modes of interest, using the *x*-axis as the baseline.

2.9. Release of 5-FU drug molecules (guests) from HKUST-1 host

To perform the release of the guest molecules from the host framework, 20 mg of the drug-loaded composite samples were immersed in abundant methanol and constantly stirred at room temperature for a period of 72 hours. The material was recovered and washed by centrifugation twice with methanol (8000 rpm for 10 min) to guarantee complete removal of guest molecules present in the final composite sample.

2.10. Fabrication of pellets

Pellets were prepared on a standard hydrostatic lab press with a die of 13 mm and under a constant force of 1 ton. The nominal density of each pellet was determined by weighing each one of them and dividing this quantity by its nominal volume. The dimensions of the pellets were determine using a micrometer.



3. Recovery of the pristine crystalline structure

Fig. S1 (a) HKUST-1 (left), 5-FU@HKUST-1_IN (center) and 5-FU@HKUST-1_EX (right) before sample activation. (b) Colour change observed after samples activation at 90 °C under vacuum. HKUST-1 framework presents a characteristic color switch from light to dark blue upon heating, linked to the change in coordination of the copper site. 5-FU@HKUST-1_IN did not present the expected colour change, due to the coordination of the guest molecule to the copper metal site.⁶ (c) Colour change presented by all samples upon guest release. (d) Pellets produced to estimate nominal density of samples, whose values considerably differ as a result of differences in porosity.

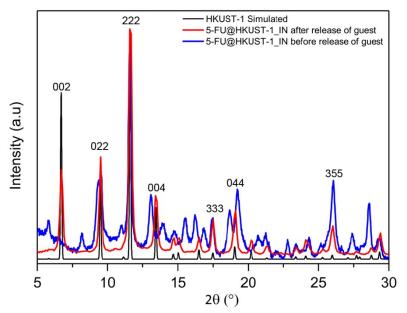


Fig. S2 Diffraction patterns of 5-FU@HKUST-1_IN before and after the release of the 5-FU (guest) drug molecules. Recovery of cubic symmetry is clear when comparison is made between the diffraction pattern of the host framework obtained after guest release and simulated pattern of HKUST-1. Labeling of Bragg peaks accompany the HKUST-1simulated pattern for reference.

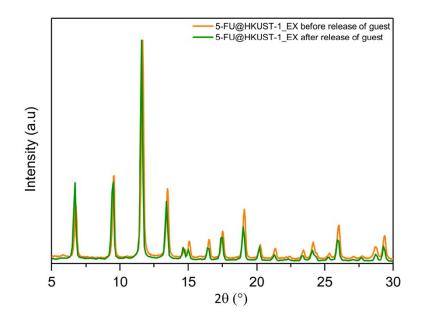


Fig. S3 Diffraction patterns of 5-FU@HKUST-1_EX before and after the release of the guest drug molecule. No apparent structural changes detected upon the release of 5-FU.

4. Confirmed guest release from host structure

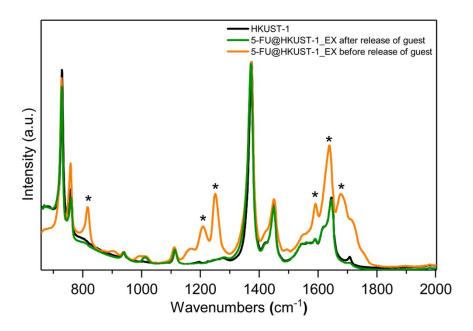


Fig. S4 ATR-FTIR spectra of 5-FU@HKUST-1_EX before and after the release of the guest drug molecule confirming the release of 5-FU. Asterisks mark the position of the 5-FU peaks.

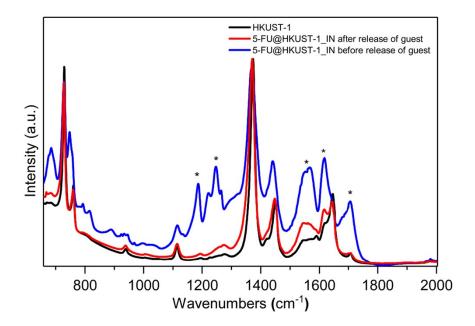


Fig. S5 ATR-FTIR spectra of 5-FU@HKUST-1_IN before and after the release of the guest drug molecule. The disappearance of the main 5-FU peaks shows the successful release of the drug molecule after sample was immersed in methanol and stirred for 72 hours. Asterisks mark the position of the 5-FU peaks.

5. Nitrogen adsorption/desorption isotherms

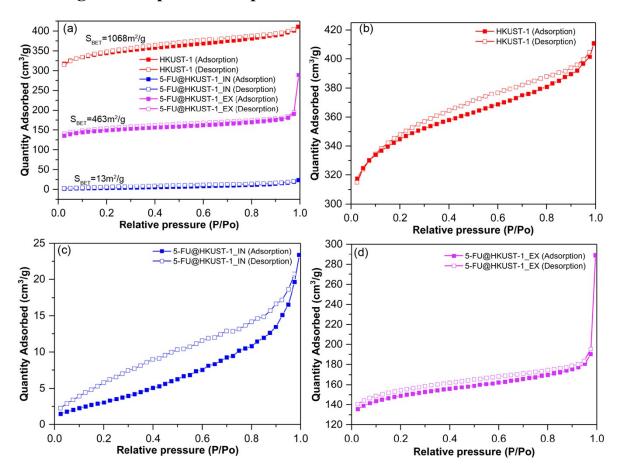


Fig. S6 a) Nitrogen adsorption and desorption isotherms of HKUST-1 and drug-loaded counterparts. Individual isotherms of b) HKUST-1, c) 5-FU@HKUST-1_IN and d) 5-FU@HKUST-1_EX. The HKUST-1 samples were activated at 90 °C under high vacuum for 12 hours prior to the N₂ adsorption measurements at 77K.

| | Method | Grinding time | BET surface area (m ² /g) | Reference |
|-----------------|--|------------------|--------------------------------------|--------------------------------------|
| HKUST-1 | Manual grinding | 10 min | 1068 | This work |
| 5-FU@HKUST-1_IN | Manual grinding | 10 min | 13 | This work |
| 5-FU@HKUST-1_EX | Manual grinding | 10 min | 463 | This work |
| HKUST-1 | Ball milling: neat grinding | 30 min | 1188 | Li, Y., et al. ⁷ |
| | Grinding jar: neat grinding | 20 min | 1119 | Schlesinger, M., et al ⁸ |
| | Grinding jar: assisted by solid solvent (NaCl) | 20 min | 1281 | Yang, J., et al. ⁹ |
| | Extrusion: liquid assisted extrusion | - | 1738 | Crawford, D., et al. ¹⁰ |
| | Ball mill: liquid assisted grinding | 25 min | 758-1713 | Klimakow, M., et al. ¹¹ |
| | Kitchen grinder: liquid assisted grinding | 5-60 min | 707 | Samal, M., et al. ¹² |
| | Solvothermal | - | 300-1500 | Xiao, J., et al. 13 |
| | Solution-based | - | 227 | Li, Y., et al. ¹⁴ |
| | Solution-based | - | 1564 | Sofi, F. A., et al. ¹⁵ |
| | Solvothermal | - | 629 | Lestari, W. W., et al. ¹⁶ |
| | Electrochemical | - | 324 | Lestari, W. W., et al. ¹⁶ |

Table S2 Reported BET surface areas of HKUST-1 materials synthesised by mechanochemistry

6. Thermogravimetric analysis

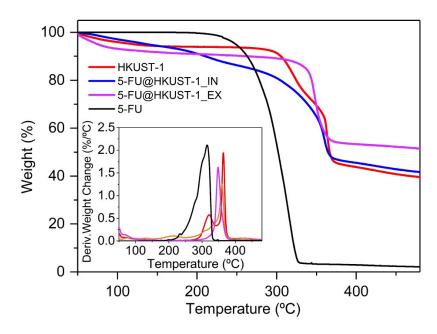


Fig. S7 TGA plots of HKUST-1 samples showing different material decomposition behavior as a function of temperature. Inset shows the derivative weight change with respect to temperature.

7. SEM and AFM images

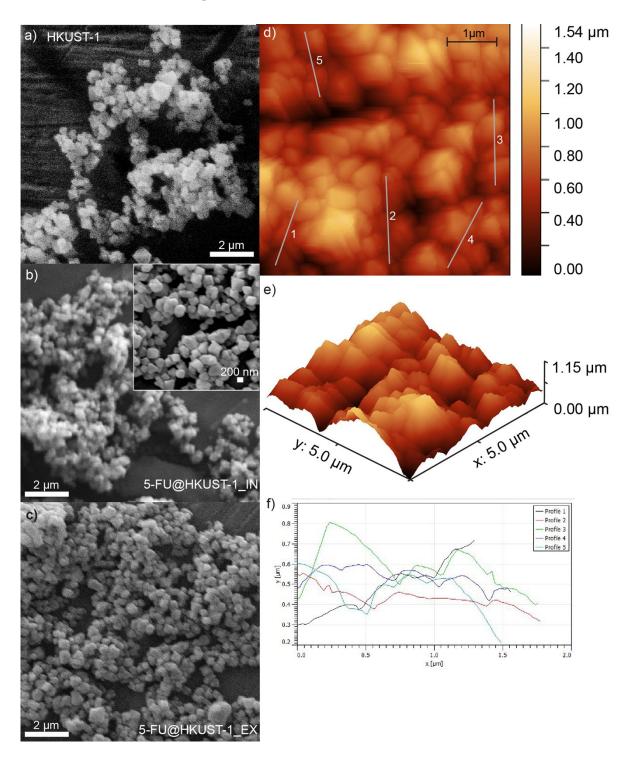
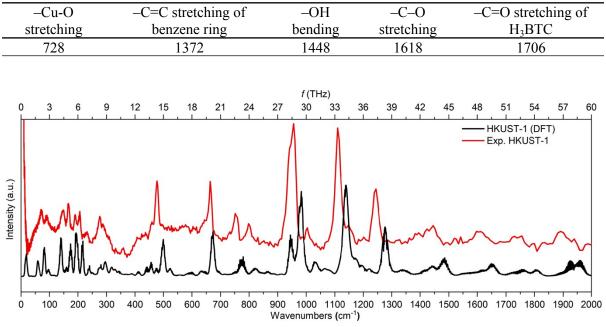


Fig. S8 SEM images of as synthesized (a) pristine HKUST-1, (b) 5-FU@HKUST-1_IN, and (c) 5-FU@HKUST-1_EX. The crystals show a uniform particle distribution, similar to the nanoHKUST-1 crystals fabricated *via* solution-based methods.¹⁴ (d-e) AFM images of pristine HKUST-1 crystals for morphology characterization, accompanied by 3D representation of the surface topography. (f) Height profiles from traces marked in (d) for the calculation of crystal size of HKUST-1, which was found to be at 432 ± 83 nm (averaged of 100 measurements).



8. ATR-FTIR and INS data



Fig. S9 Theoretical and experimental spectra of HKUST-1 for comparison. Theoretical spectrum is displayed in black and experimental spectrum is shown in red.

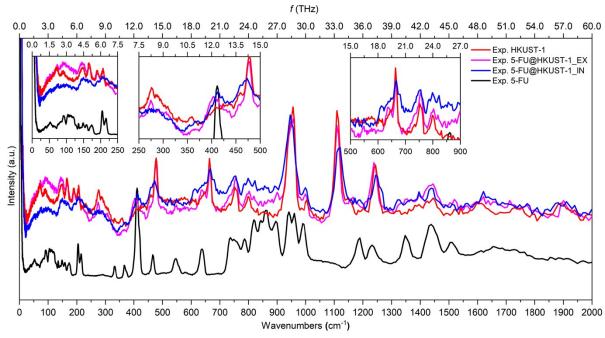


Fig. S10 Full INS spectra of HKUST-1 samples up to ~60 THz (2000 cm⁻¹). The experimental spectrum of 5-FU was scaled down by a factor of 5 to facilitate comparison with experimental data of HKUST-1.

| Table S4 Description of vibrational and rotational modes of HKUS1-1 | | | | | |
|---|---------------------|---------------------|---|--|--|
| Mode | Theo. | Exp. | Descriptions based on Ryder <i>et al.</i> ⁴ | | |
| no. | (cm ⁻¹) | (cm ⁻¹) | | | |
| 0 | 19 | (hidden) | Strong paddle wheel deformation and translational motion | | |
| 1 | 60 | 70 | Paddle wheel deformation and translational motion with | | |
| | | | organic linker rotation | | |
| 2 | 83 | 91 | Paddle wheel deformation and rotation with organic linker | | |
| | | | rocking | | |
| 3 | 97 | (hidden) | Organic linker trampoline-like motion | | |
| 4 | 140 | 148 | Paddle wheel rocking with organic linker rocking | | |
| 5 | 160 | (hidden) | Paddle wheel deformation | | |
| 6 | 174 | 166 | Paddle wheel translational motion | | |
| 7 | 193 | 190 | Asymmetric paddle wheel deformation (Cu-Cu buckling) | | |
| | | | with linker rocking | | |
| | | | Asymmetric paddle wheel deformation | | |
| | | | (O-Cu-O bending) | | |
| 8 | 217 | 207 | Paddle wheel rotation with strong linker rocking | | |
| 9 | 241 | 232 | Paddle wheel deformation (strong Cu-Cu buckling) with | | |
| | | | linker rocking | | |
| 10 | 280 | 260 | Asymmetric paddle wheel deformation (O-Cu-O bending | | |
| 11 | 296 | 276 | and Cu-Cu buckling) | | |
| 12 | 318 | 288 | | | |
| 13 | 330 | 332 | Paddle wheel deformation with linker rocking | | |
| 14 | 345 | 359 | | | |
| 15 | 411 | 408 | | | |
| 16 | 442 | 425 | In Plane aromatic ring Rotation with Asymmetric Cu-O | | |
| 17 | 457 | 438 | stretching | | |
| 18 | 477 | 460 | In Plane aromatic ring deformation with partially symmetric | | |
| | | | Cu-O stretching | | |
| 19 | 499 | 478 | Out of Plane aromatic ring deformation | | |
| | | | | | |

Table S4 Description of vibrational and rotational modes of HKUST-1

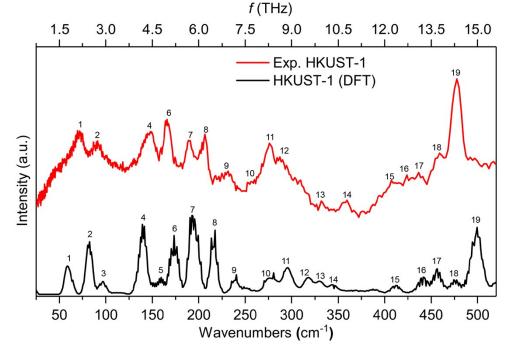


Fig. S11 Theoretical and experimental INS spectra of HKUST-1 with vibrational modes marked as #1-19 for easy correlation between theoretical and experimental data.

| | Wavenumber (cm ⁻¹) | Sample | Area under the graph $(S \cdot \omega)$ | FWHM (cm ⁻¹) |
|-------------|-----------------------------------|-----------------|---|-----------------------------|
| | (cm) | | (5.67) | |
| | | HKUST-1 | 504.8 | - |
| | 0-214 | 5-FU@HKUST-1_IN | 402.6 | - |
| | | 5-FU@HKUST-1_EX | 477.9 | - |
| | 155-175 | HKUST-1 | 15.9 | 30.1 |
| Mode #6 | | 5-FU@HKUST-1_IN | 11.7 | 88.8 |
| | | 5-FU@HKUST-1_EX | 15.2 | 38.5 |
| | 182-198 | HKUST-1 | 11.9 | 44.7 |
| Mode #7 | | 5-FU@HKUST-1_IN | 9.1 | 139.3 |
| | | 5-FU@HKUST-1_EX | 10.9 | 39.05 |
| | 198-217 | HKUST-1 | 13.6 | 24.8 |
| Mode #8 | | 5-FU@HKUST-1_IN | 11.9 | 97.7 |
| | | 5-FU@HKUST-1_EX | 12.9 | 30.3 |
| | | HKUST-1 | 17.8 | 66.5 |
| Mode #9 | 220-250 | 5-FU@HKUST-1_IN | 16.5 | 83.5 |
| | | 5-FU@HKUST-1_EX | 15.2 | 68.6 |
| | 269-282 | HKUST-1 | 9.5 | 23.2 |
| Mode #11 | | 5-FU@HKUST-1_IN | 6.6 | - |
| | | 5-FU@HKUST-1_EX | 8.1 | 30.8 |
| Organic | | HKUST-1 | 21.8 | 25.3 |
| linker | 465-489 | 5-FU@HKUST-1_IN | 17.9 | 44.9 |
| deformation | | 5-FU@HKUST-1_EX | 19.1 | 24.5 |

Table S5 Values of area under the inelastic curves and full width at half maximum (FWHM) values of the vibrational peaks of HKUST-1

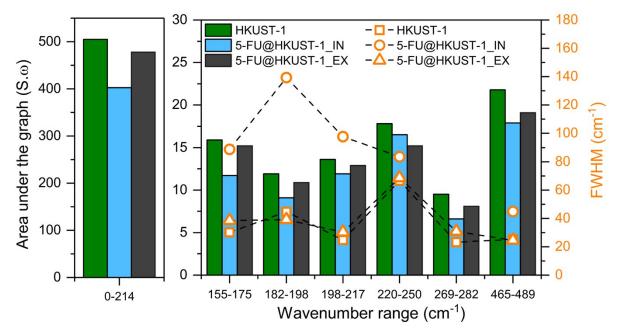


Fig. S12 Bar plot summarizing data presented in Table S5. The plot demonstrates the decrease in the area and the broadening of specific peaks related to copper (II) paddle-wheel vibrations in 5-FU@HKUST-1_IN sample.

References

- 1. S. F. Parker, F. Fernandez-Alonso, A. J. Ramirez-Cuesta, J. Tomkinson, S. Rudic, R. S. Pinna, G. Gorini and J. Fernández Castañon, *J. Phys. Conf. Ser.*, 2014, **554**.
- O. Arnold, J. C. Bilheux, J. M. Borreguero, A. Buts, S. I. Campbell, L. Chapon, M. Doucet, N. Draper, R. Ferraz Leal, M. A. Gigg, V. E. Lynch, A. Markvardsen, D. J. Mikkelson, R. L. Mikkelson, R. Miller, K. Palmen, P. Parker, G. Passos, T. G. Perring, P. F. Peterson, S. Ren, M. A. Reuter, A. T. Savici, J. W. Taylor, R. J. Taylor, R. Tolchenov, W. Zhou and J. Zikovsky, *Nucl. Instrum. Meth. A*, 2014, **764**, 156-166.
- 3. R. S. Pinna, S. Rudić, S. F. Parker, J. Armstrong, M. Zanetti, G. Škoro, S. P. Waller, D. Zacek, C. A. Smith, M. J. Capstick, D. J. McPhail, D. E. Pooley, G. D. Howells, G. Gorini and F. Fernandez-Alonso, *Nucl. Instrum. Meth. A*, 2018, **896**, 68-74.
- 4. M. R. Ryder, B. Civalleri, G. Cinque and J.-C. Tan, *CrystEngComm*, 2016, **18**, 4303-4312.
- 5. S. F. P. K. Dymkowski, F. Fernandez-Alonso, *Phys. B Cond. Mat.*, 2018, DOI: 10.1016/j.physb.2018.02.034.
- A. A. Talin, A. Centrone, A. C. Ford, M. E. Foster, V. Stavila, P. Haney, R. A. Kinney, V. Szalai, F. El Gabaly, H. P. Yoon, F. Leonard and M. D. Allendorf, *Science*, 2014, 343, 66-69.
- 7. B. Yuan, X. Q. Yin, X. Q. Liu, X. Y. Li and L. B. Sun, *ACS Appl. Mater. Interfaces*, 2016, **8**, 16457-16464.
- 8. M. Schlesinger, S. Schulze, M. Hietschold and M. Mehring, *Microporous Mesoporous Mat.*, 2010, **132**, 121-127.
- 9. J. Yang, X. Feng, G. Lu, Y. Li, C. Mao, Z. Wen and W. Yuan, *Dalton Trans.*, 2018, **47**, 5065-5071.
- 10. D. Crawford, J. Casaban, R. Haydon, N. Giri, T. McNally and S. L. James, *Chem. Sci.*, 2015, **6**, 1645-1649.
- 11. M. Klimakow, P. Klobes, A. F. Thünemann, K. Rademann and F. Emmerling, *Chem. Mater.*, 2010, **22**, 5216-5221.
- 12. M. Samal, J. Panda, B. P. Biswal and R. Sahu, *CrystEngComm*, 2018, **20**, 2486-2490.
- 13. J. Xiao, Y. Zhu, S. Huddleston, P. Li, B. Xiao, O. K. Farha and G. A. Ameer, *ACS Nano*, 2018, **12**, 1023-1032.
- 14. Y. Li, X. Li, Q. Guan, C. Zhang, T. Xu, Y. Dong, X. Bai and W. Zhang, *Int. J. Nanomedicine*, 2017, **12**, 1465-1474.
- 15. F. A. Sofi, K. Majid and O. Mehraj, *J. Alloy Compd.*, 2018, **737**, 798-808.
- 16. W. W. Lestari, M. Adreane, C. Purnawan, H. Fansuri, N. Widiastuti and S. B. Rahardjo, *IOP Conference Series: Materials Science and Engineering*, 2016, **107**.