

Supporting information:

A novel carbazole derivative containing fluorobenzene unit: Aggregation-induced fluorescence emission, polymorphism, mechanochromism and non-reversible thermo-stimulus fluorescence

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Contents:

Scheme S1 Synthetic routes of compound **3**.

Table S1 Crystal data and details of collection and refinement of **3Y** and **3O**.

Fig. S1 ¹H NMR of compound **3Y** (CDCl₃, 400 MHz)

Fig. S2 ¹H NMR of compound **3Y** and **3O** (CDCl₃, 400 MHz)

Fig. S3 ¹³C NMR of compound **3Y** (CDCl₃, 400 MHz)

Fig. S4 The absorption spectra for **3Y** and **3O** in the solid state.

Fig. S5 Lifetime decay curves of **3Y** (a) and **3O** (b).

Fig. S6 As-experimental and as-simulated XRD patterns for powder **3Y**, **3O** and crystals **3Y**, **3O**.

Fig. S7 Highest occupied molecular orbital (HOMO), lowest unoccupied molecular orbital (LUMO) diagrams and the energy gap (ΔE) of **3Y** and **3O**

Fig. S8 Molecular stacks in the crystals of **3O**, (a) structure A, (b) structure B, (c) **3O** viewed from the a-axis.

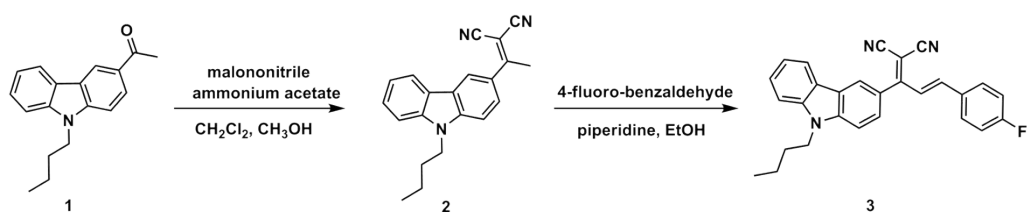
Fig. S9 The molecules of **3O** (structure A and B)

Fig. S10 Molecular stacks in the crystals of **3O**, (a) structure A, (b) structure B, and **3O** (c) viewed from the a-axis.

Fig. S11 Molecular stacks in the crystals of **3O** viewed from the b-axis (a) and c-axis (b).

Fig. S12 Cycle behaviors of the maximum emission peaks for **3Y**.

Fig. S13 The PXRD diagram of **3Y**, heated **3Y** and **3O**.



Scheme S1 Synthetic routes of compound **3**

Table S1 Crystal data and details of collection and refinement of **3Y** and **3O**.

Identification code	3Y	3O
CCDC	1511130	1576277
Empirical formula	C ₂₈ H ₂₂ FN ₃	C ₂₈ H ₂₂ FN ₃
Formula weight	419.49	419.48
Temperature (K)	296(2)	296(2)
Wavelength (Å)	0.71073	0.71073
Crystal system, space group	Orthorhombic, <i>P</i> 2 ₁ 2 ₁ 2 ₁	Monoclinic, <i>P</i> 2 ₁ / <i>c</i>
<i>a</i> (Å)	8.7593(18)	12.758(5)
<i>b</i> (Å)	13.386(4)	14.223(5)
<i>c</i> (Å)	19.541(4)	12.808(5)
α (deg)	90	90
β (deg)	90	93.522(4)
γ (deg)	90	90
Volume (Å ³)	2291.2(9)	2319.8(14)
Z, Calculated density (Mg/m ³)	4, 1.216	4, 1.201
Absorption coefficient (mm ⁻¹)	0.078	0.077
F(000)	880	880
Crystal size (mm)	0.19 × 0.18 × 0.17	0.19 × 0.18 × 0.17
Theta range for data collection (deg)	1.84 to 25.00 .	1.599 to 24.999
Limiting indices	-10 ≤ <i>h</i> ≤ 10, -15 ≤ <i>k</i> ≤ 15, -21 ≤ <i>l</i> ≤ 23	-15 ≤ <i>h</i> ≤ 14, -16 ≤ <i>k</i> ≤ 16, -15 ≤ <i>l</i> ≤ 14
Reflections collected / unique	16322 / 4029	16199 / 4073
<i>R</i> (int)	0.0199	0.0521
Data / restraints / parameters	4029 / 0 / 291	4073 / 23 / 290
Goodness-of-fit on <i>F</i> ²	1.049	1.064
Final <i>R</i> indices [<i>I</i> > 2σ(<i>I</i>)]	<i>R</i> ₁ = 0.0395, <i>wR</i> ₂ = 0.1048	<i>R</i> ₁ = 0.0879, <i>wR</i> ₂ = 0.2700
<i>R</i> indices (all data)	<i>R</i> ₁ = 0.0458, <i>wR</i> ₂ = 0.1106	<i>R</i> ₁ = 0.1019, <i>wR</i> ₂ = 0.2873

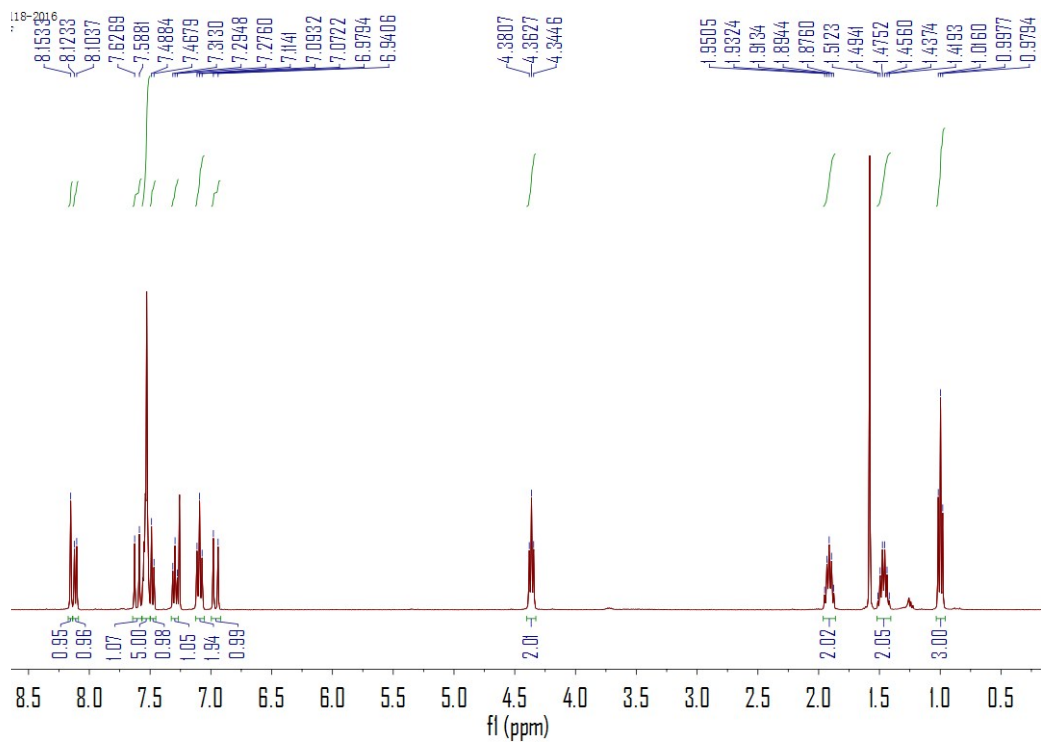


Fig. S1 ^1H NMR of compound **3** (CDCl_3 , 400 MHz)

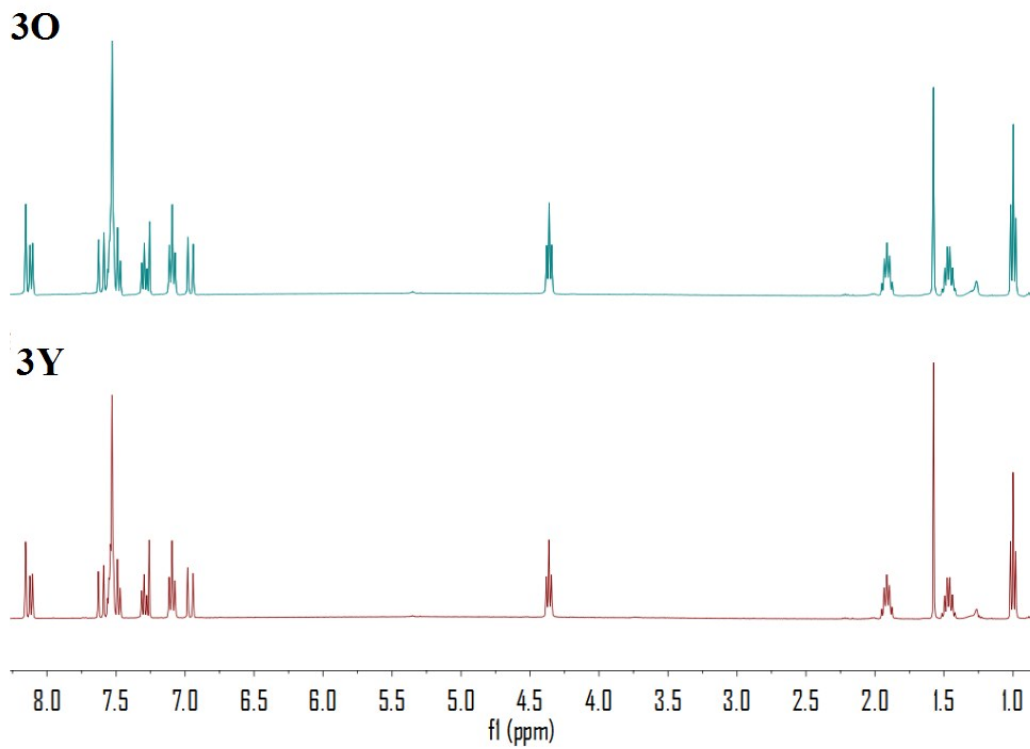


Fig. S2 ^1H NMR of compound **3Y** and **3O** (CDCl_3 , 400 MHz)

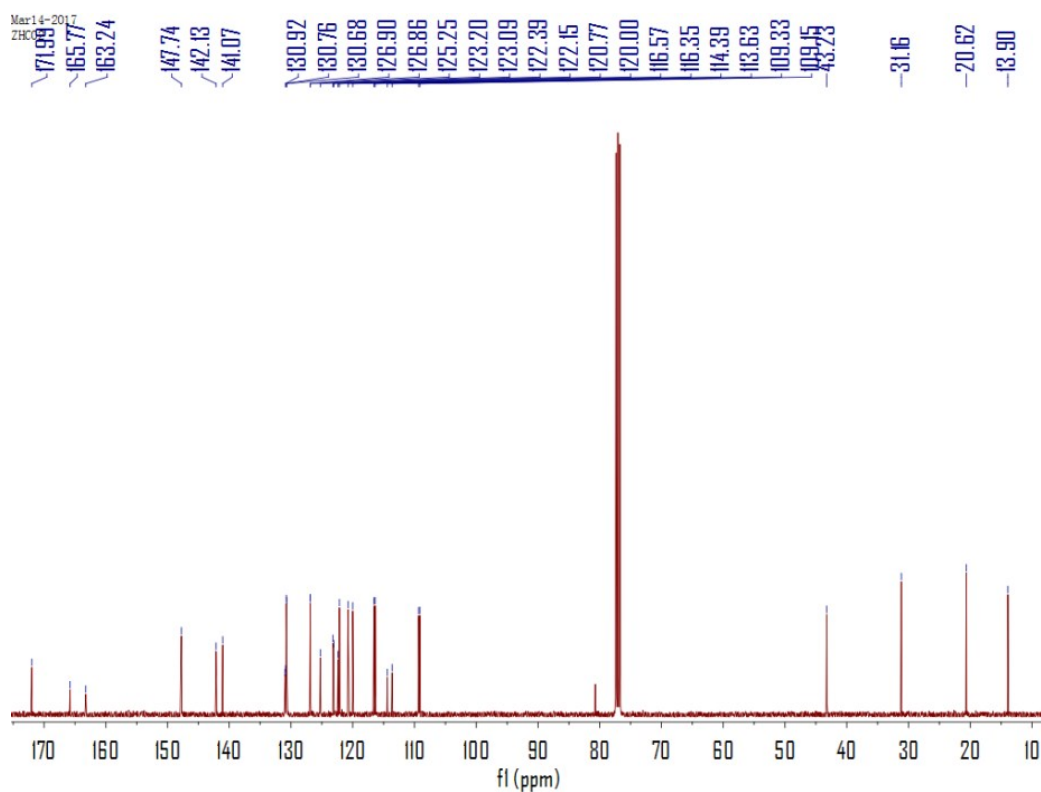


Fig. S3 ^{13}C NMR of compound **3Y** (CDCl_3 , 400 MHz).

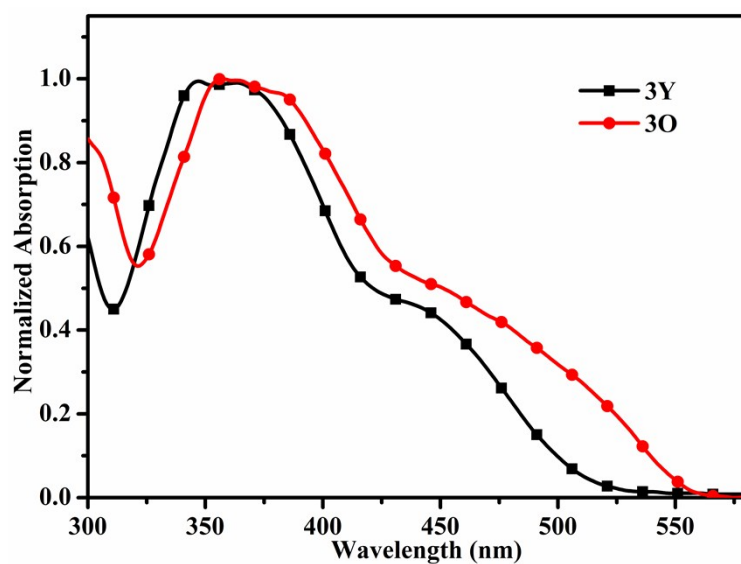


Fig. S4 The absorption spectra for **3Y** and **3O** in the solid state.

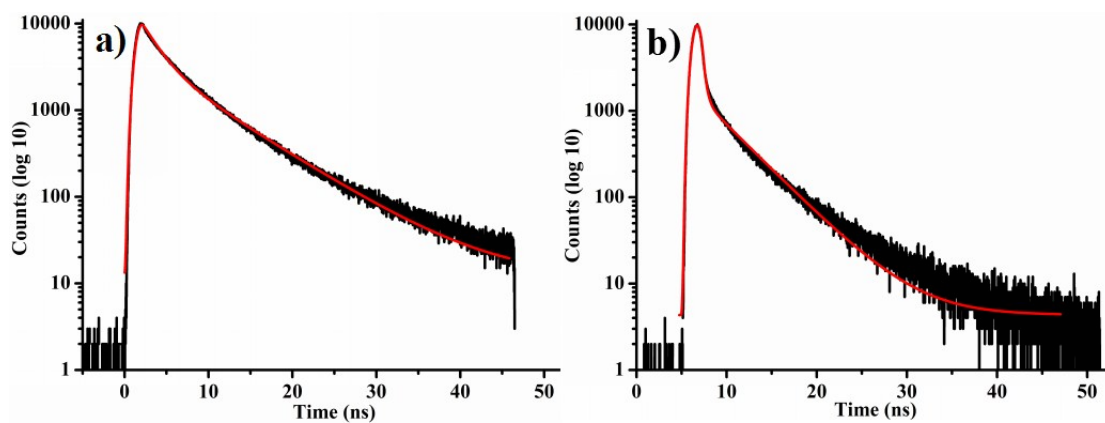


Fig. S5 Lifetime decay curves of **3Y** (a) and **3O** (b).

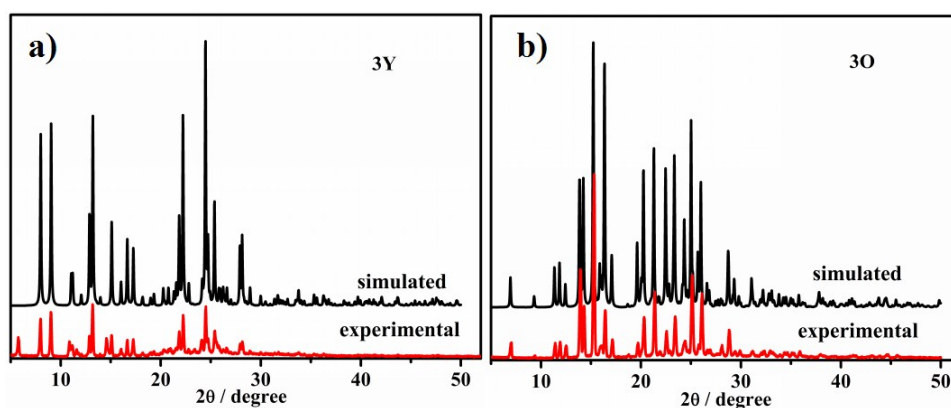


Fig. S6 As-experimental and as-simulated XRD patterns for powder **3Y**, **3O** and crystals **3Y**, **3O**.

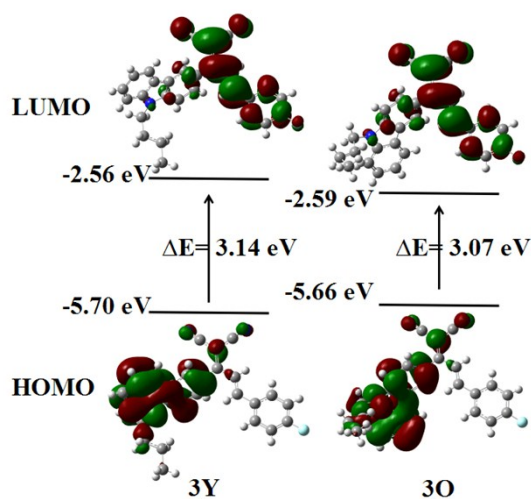


Fig S7 Highest occupied molecular orbital (HOMO), lowest unoccupied molecular orbital (LUMO) diagrams and the energy gap (ΔE) of **3Y** and **3O**

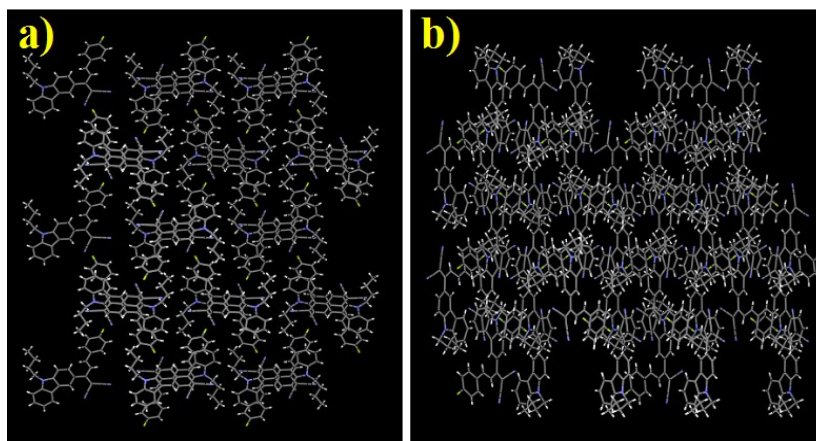


Fig. S8 Molecular stacks in the crystals of **3Y** viewed from the a-axis (a) and c-axis (b).

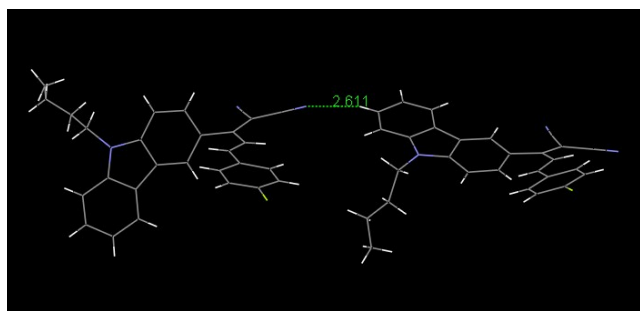


Fig. S9 The molecules of **3O** (structure A and B)

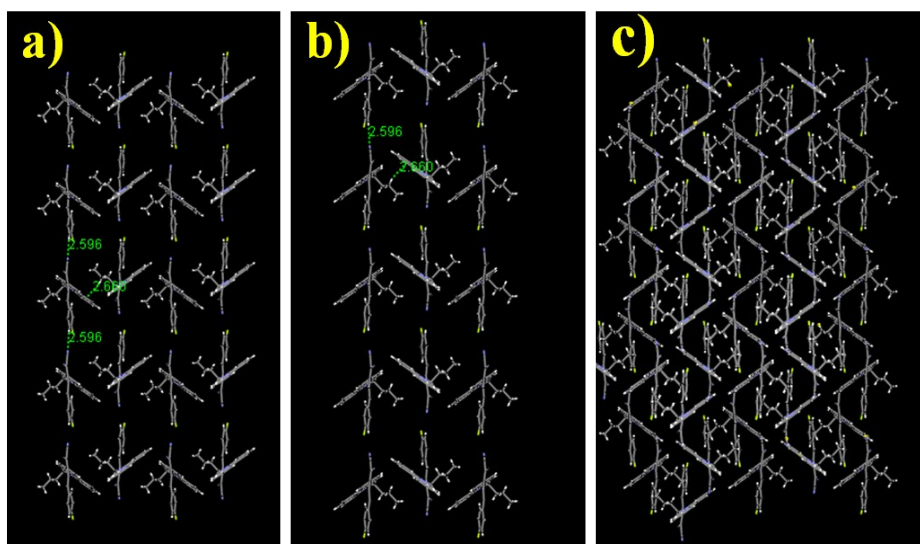


Fig. S10 Molecular stacks in the crystals of **3O**, (a) structure A, (b) structure B, and **3O** (c) viewed from the a-axis.

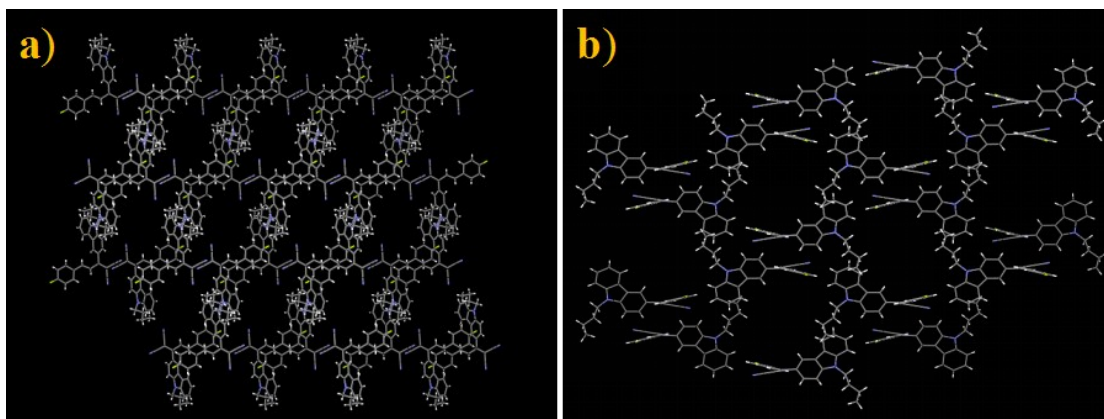


Fig. S11 Molecular stacks in the crystals of **3O** viewed from the b-axis (a) and c-axis (b).

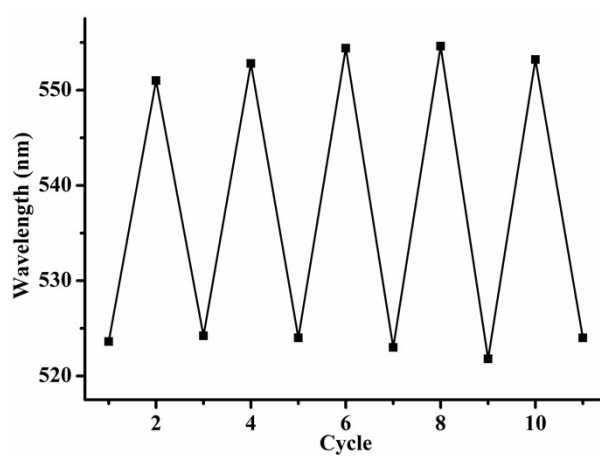


Fig. S12 Cycle behaviors of the maximum emission peaks for **3Y**.

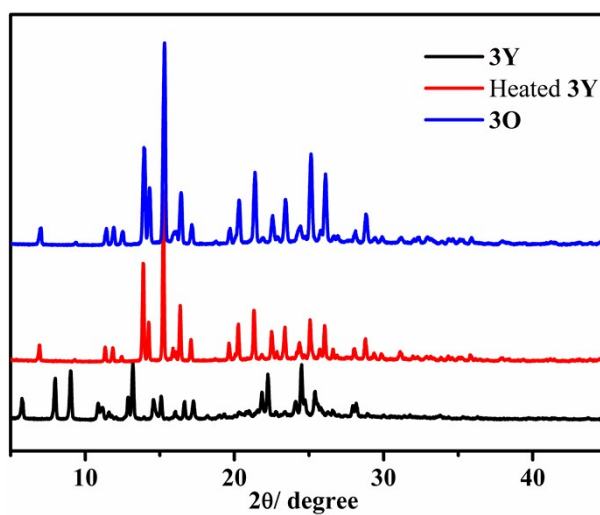


Fig. S13 The PXRD diagram of **3Y**, heated **3Y** and **3O**.