

Six novel coordination polymers based on 5-(1H-tetrazol-5-yl)isophthalic acid ligand: structures, luminescence, and magnetic properties

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Table S1. Selected bond lengths [Å] and angles [°] for CPs **1-6**.

CP 1			
Mn1-O6	2.089(2)	Mn1-O5A	2.143(2)
Mn1-O3B	2.152(2)	Mn1-O1	2.184(2)
Mn1-N4C	2.392(3)	Mn1-N1	2.401(3)
Mn2-O4D	2.060(2)	Mn2-O1	2.105(2)
Mn2-O2B	2.148(2)	Mn2-O1E	2.158(2)
Mn2-N2	2.235(3)		
O6-Mn1-O5A	87.19(11)	O6-Mn1-O3B	88.61(11)
O5A-Mn1-O3B	175.17(9)	O6-Mn1-O1	179.13(10)
O5A-Mn1-O1	93.31(9)	O3B-Mn1-O1	90.86(9)
O6-Mn1-N4C	94.82(10)	O5A-Mn1-N4C	90.39(10)
O3B-Mn1-N4C	87.60(10)	O1-Mn1-N4C	84.47(8)
O6-Mn1-N1	92.59(10)	O5A-Mn1-N1	95.63(9)
O3B-Mn1-N1	86.92(9)	O1-Mn1-N1	88.08(8)
N4C-Mn1-N1	170.68(9)	O4D-Mn2-O1	147.71(11)
O4D-Mn2-O2B	86.74(10)	O1-Mn2-O2B	93.41(9)
O4D-Mn2-O1E	93.12(9)	O1-Mn2-O1E	82.43(9)
O2B-Mn2-O1E	171.98(9)	O4D-Mn2-N2	121.94(12)
O1-Mn2-N2	90.19(9)	O2B-Mn2-N2	95.91(10)

O1E-Mn2-N2 90.98(9)

A $-x+1,-y+1,-z+1$; B $x+1/2,-y+3/2,z-1/2$; C $x+1,y,z$; D $x,y,z-1$; E $-x+1,-y+1,-z$.

CP 2

Cu1-O1	1.936(4)	Cu1-O2	1.943(4)
Cu1-O5A	1.944(4)	Cu1-N2	1.980(4)
Cu1-O4B	2.236(4)	Cu2-O3B	2.111(4)
Cu2-O6A	1.921(4)	Cu2-O1C	1.945(4)
Cu2-O1	2.041(4)	Cu2-N3C	2.048(4)
O1-Cu1-O2	178.26(18)	O1-Cu1-O5A	93.44(16)
O2-Cu1-O5A	87.71(17)	O1-Cu1-N2	88.48(17)
O2-Cu1-N2	90.10(18)	O5A-Cu1-N2	166.3(2)
O1-Cu1-O4B	86.54(15)	O2-Cu1-O4B	94.43(19)
O5A-Cu1-O4B	104.76(17)	N2-Cu1-O4B	88.94(17)
N3C-Cu2-O3B	107.21(17)	O6A-Cu2-O1C	172.85(17)
O6A-Cu2-O1	93.13(16)	O1C-Cu2-O1	81.87(17)
O6A-Cu2-N3C	91.63(16)	O1C-Cu2-N3C	89.02(16)
O1-Cu2-N3C	140.31(17)	O6A-Cu2-O3B	89.93(17)
O1C-Cu2-O3B	96.68(16)	O1-Cu2-O3B	112.17(15)

A $x+1,y,z+1$; B $x+1,-y+3/2,z+1/2$; C $-x+2,-y+1,-z+1$.

CP 3

Co1-O3A	1.993(2)	Co1-O1	2.029(2)
Co1-O6	2.064(2)	Co1-O5	2.259(3)
Co1-N4B	2.269(3)	Co1-N1C	2.347(3)
Co2-O6D	2.031(2)	Co2-O4E	2.044(3)
Co2-O6	2.053(2)	Co2-N3B	2.104(3)
Co2-N2F	2.122(3)	Co2-O7	2.219(3)
O3A-Co1-O1	95.89(10)	O3A-Co1-O6	88.83(10)
O1-Co1-O6	175.26(9)	O3A-Co1-O5	168.11(11)
O1-Co1-O5	93.49(10)	O6-Co1-O5	81.89(10)

O3A-Co1-N4B	100.99(11)	O1-Co1-N4B	91.88(10)
O6-Co1-N4B	86.73(9)	O5-Co1-N4B	85.95(13)
O3A-Co1-N1C	84.98(11)	O1-Co1-N1C	89.41(10)
O6-Co1-N1C	91.48(9)	O5-Co1-N1C	87.83(12)
N4B-Co1-N1C	173.72(11)	O6D-Co2-O4E	90.86(10)
O6D-Co2-O6	165.59(13)	O4E-Co2-O6	103.33(10)
O6D-Co2-N3B	93.18(10)	O4E-Co2-N3B	90.03(11)
O6-Co2-N3B	84.43(10)	O6D-Co2-N2F	89.16(10)
O4E-Co2-N2F	89.82(11)	O6-Co2-N2F	93.33(10)
N3B-Co2-N2F	177.66(11)	O6D-Co2-O7	83.60(11)
O4E-Co2-O7	174.45(11)	O6-Co2-O7	82.22(10)
N3B-Co2-O7	90.59(11)	N2F-Co2-O7	89.78(11)

A $-x+1, y-1/2, -z+3/2$; B $x, -y+3/2, z-1/2$; C $x, -y+3/2, z+1/2$; D $x, -y+1/2, z-1/2$; E $-x+1, -y+1, -z+1$; F $x, y-1, z$.

CP 4

Zn1-O1	1.912(2)	Zn1-O3A	1.935(2)
Zn1-O5B	1.957(2)	Zn1-N1	2.050(2)
Zn2-O1	1.937(2)	Zn2-N5	2.099(3)
Zn2-N6	2.126(3)	Zn2-N4B	2.214(2)
Zn2-N2	2.254(2)		
O1-Zn1-O3A	126.66(9)	O1-Zn1-O5B	111.18(10)
O3A-Zn1-O5B	96.42(9)	O1-Zn1-N1	94.79(9)
N5-Zn2-N6	77.89(10)	O1-Zn2-N4B	103.45(9)
N5-Zn2-N4B	98.45(10)	N6-Zn2-N4B	89.18(9)
O1-Zn2-N2	86.58(9)	N5-Zn2-N2	94.64(10)
N6-Zn2-N2	172.38(9)	N4B-Zn2-N2	90.42(9)

A $-x+2, -y, -z+1$; B $x, -y+1/2, z-1/2$.

CP 5

Zn1-O5	1.9051(16)	Zn1-O1	1.9734(14)
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Zn1-O4A	1.9916(14)	Zn1-N1B	2.0392(18)
Zn2-O5	1.9306(15)	Zn2-N4C	2.0474(18)
Zn2-N5	2.0941(18)	Zn2-O3A	2.1306(15)
Zn2-N6	2.1883(18)		
O5-Zn1-O1	122.57(7)	O5-Zn1-O4A	108.06(7)
O1-Zn1-O4A	104.64(6)	O5-Zn1-N1B	106.83(7)
O1-Zn1-N1B	111.59(7)	O4A-Zn1-N1B	100.98(7)
O5-Zn2-N4C	106.27(7)	O5-Zn2-N5	134.18(7)
N4C-Zn2-N5	119.46(7)	O5-Zn2-O3A	91.40(6)
N4C-Zn2-O3A	96.41(7)	N5-Zn2-O3A	86.48(7)
O5-Zn2-N6	96.16(7)	N4C-Zn2-N6	95.82(7)
N5-Z2-N6	77.49(7)	O3A-Zn2-N6	163.21(7)

A $x,y-1,z$; B $-x,-y+1,-z+1$; C $-x+1,-y+1,-z+1$.

CP 6

Zn1-O5	1.9430(18)	Zn1-O3A	1.9656(18)
Zn1-O2	1.985(2)	Zn1-N4B	2.040(2)
Zn1-O1	2.554(2)	Zn2-N5	2.054(2)
Zn2-O5	1.8858(18)	Zn2-O4C	1.9685(18)
Zn2-N1D	2.029(2)		
O5-Zn1-O3A	105.07(8)	O5-Zn1-O2	103.15(8)
O3A-Zn1-O2	116.10(9)	O5-Zn1-N4B	101.86(8)
O3A-Zn1-N4B	105.87(8)	O2-Zn1-N4B	122.42(9)
O5-Zn1-O1	155.15(7)	O3A-Zn1-O1	97.42(8)
O2-Zn1-O1	56.39(8)	N4B-Zn1-O1	81.63(8)
N1D-Zn2-N5	108.04(8)	O5-Zn2-O4C	122.11(8)
O5-Zn2-N1D	107.88(8)	O4C-Zn2-N1D	105.79(9)
O5-Zn2-N5	116.47(8)	O4C-Zn2-N5	95.27(8)

A $-x+1,-y+2,-z+1$; B $-x+1,-y+1,-z+1$; C $x-1,y-1,z-1$; D $-x,-y+1,-z+1$.

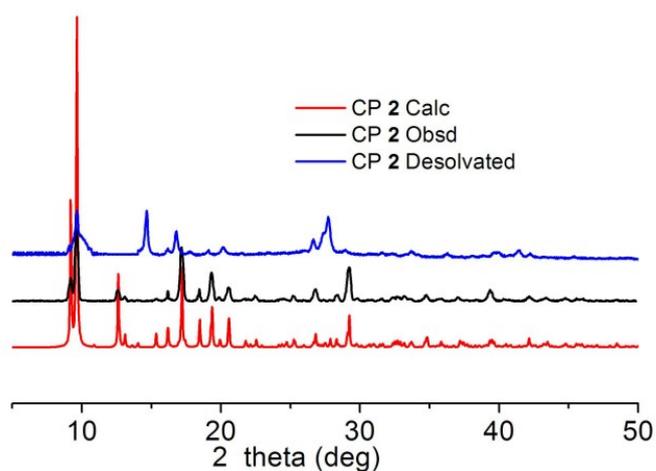


Fig. S1. PXRD patterns for CP 2 (simulated: red; experimental: black; desolvated sample: blue).

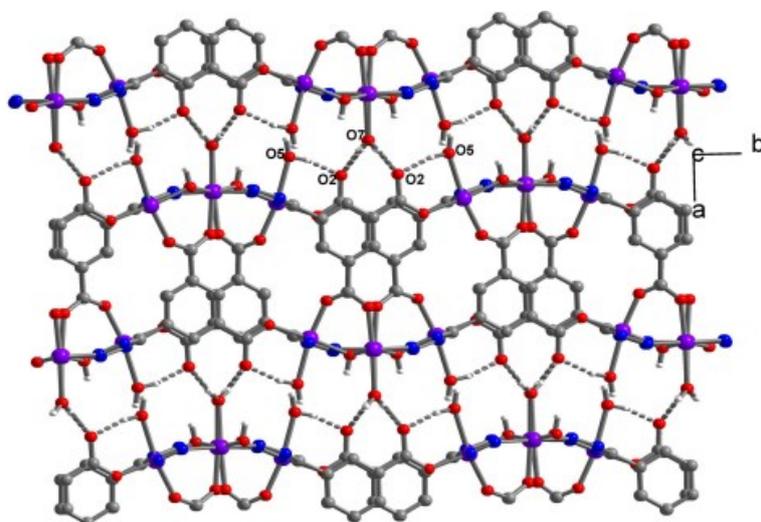


Fig. S2. The 3D architecture with hydrogen bonds between the layers in CP 3. ($O5-H \cdots O2 = 1.768 \text{ \AA}$, $O5-H \cdots O2 = 166^\circ$, $O5 \cdots O2 = 2.593 \text{ \AA}$; $O7-H \cdots O2 [-x+2, -y+1, -z+1] = 2.006 \text{ \AA}$, $O7-H \cdots O2 = 177^\circ$, $O7 \cdots O2 = 2.699 \text{ \AA}$; $O7-H \cdots O2 [-x+2, y-1/2, -z+3/2] = 1.841 \text{ \AA}$, $O7-H \cdots O2 = 176^\circ$, $O7 \cdots O2 = 2.711 \text{ \AA}$).

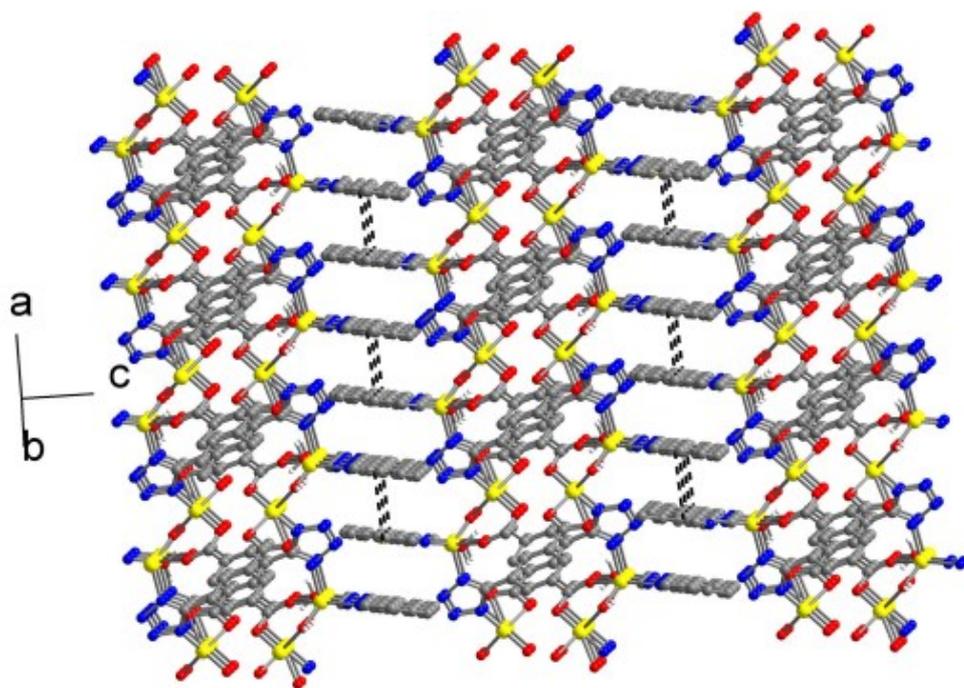


Fig. S3. The 3D architecture by π - π stacking in CP 5.

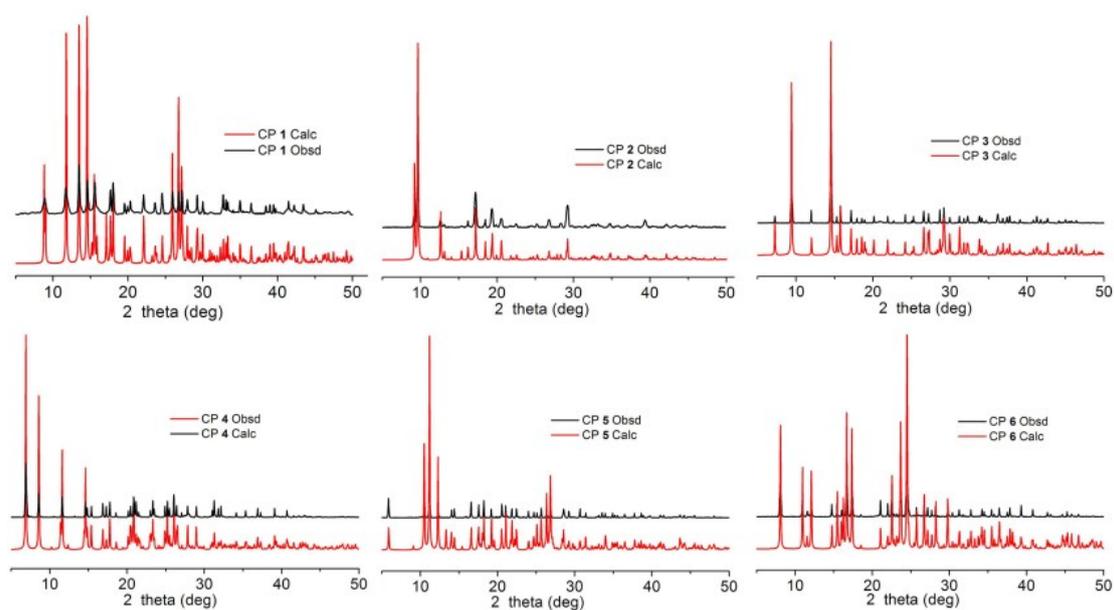


Fig. S4. PXR D patterns for CPs 1-6 (simulated: red; experimental: black)

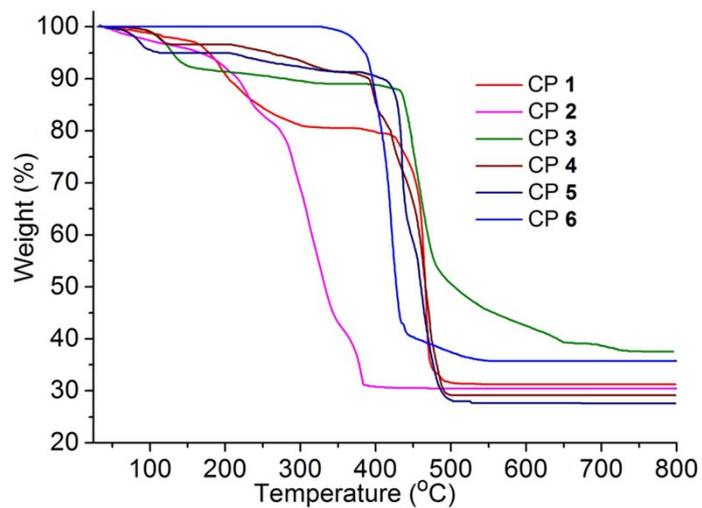


Fig. S5. TGA curves of CPs 1-6.

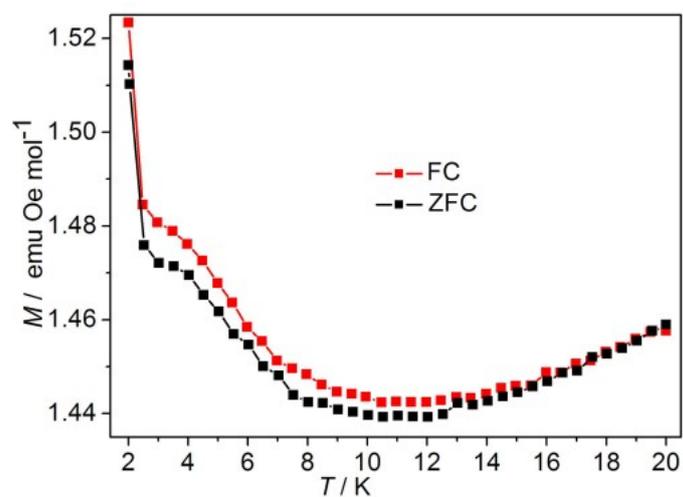


Fig. S6. The FCM and ZFCM curves at 20 Oe for CP 1.

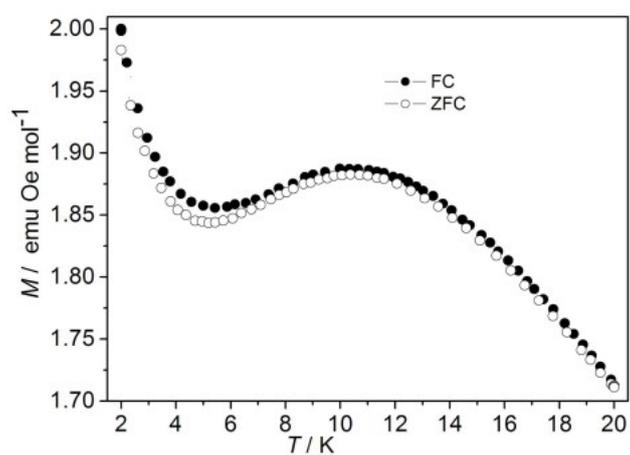


Fig. S7. The FCM and ZFCM curves at 20 Oe for CP 3.

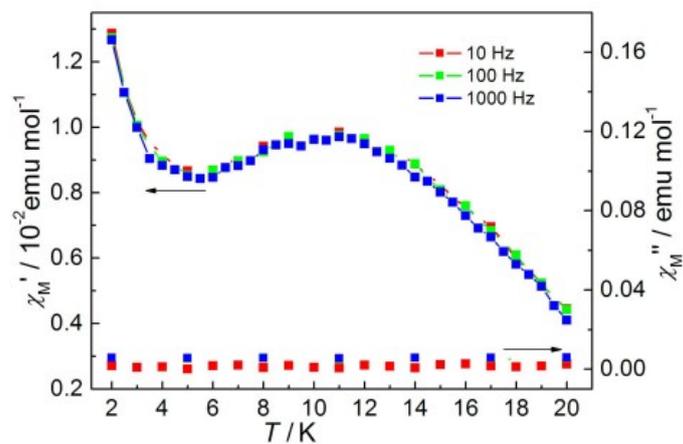


Fig. S8. The AC curves obtained at 3 Oe field for CP 3.

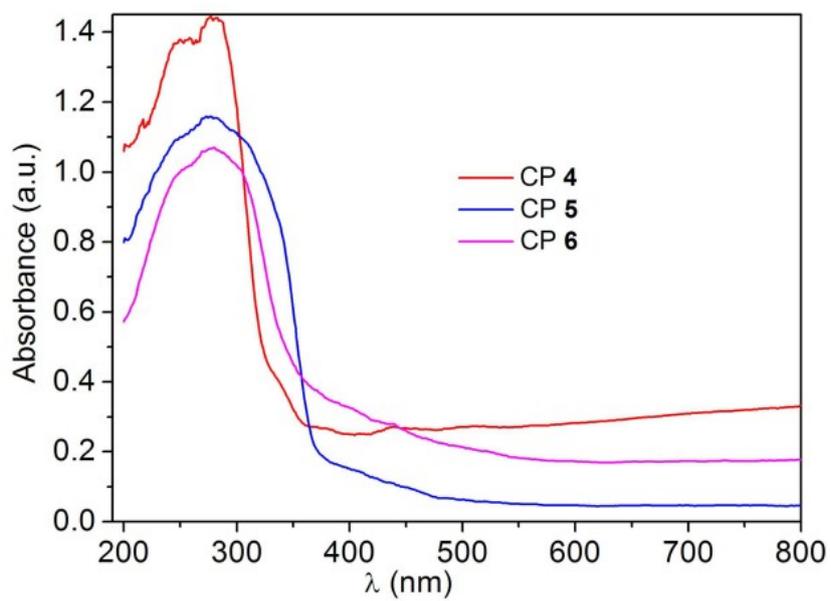


Fig. S9. UV-vis spectra of CPs 4-6.

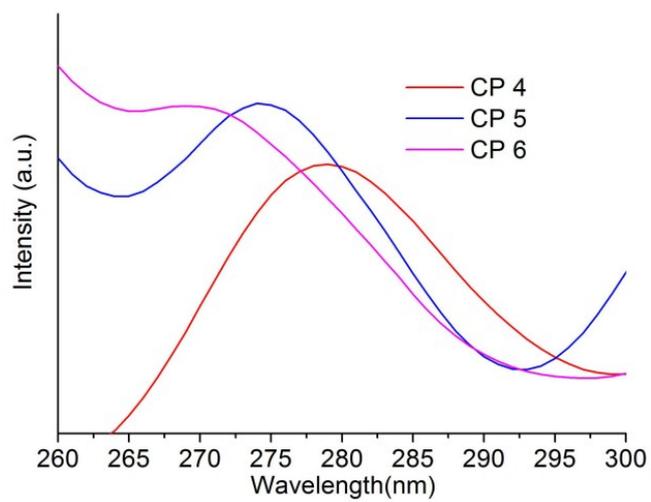


Fig. S10. Excitation spectra of CPs **4-6** at room temperature in the solid state.

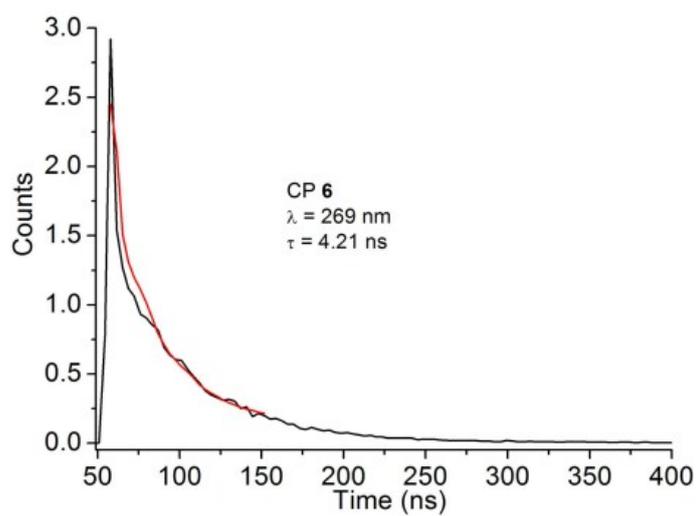
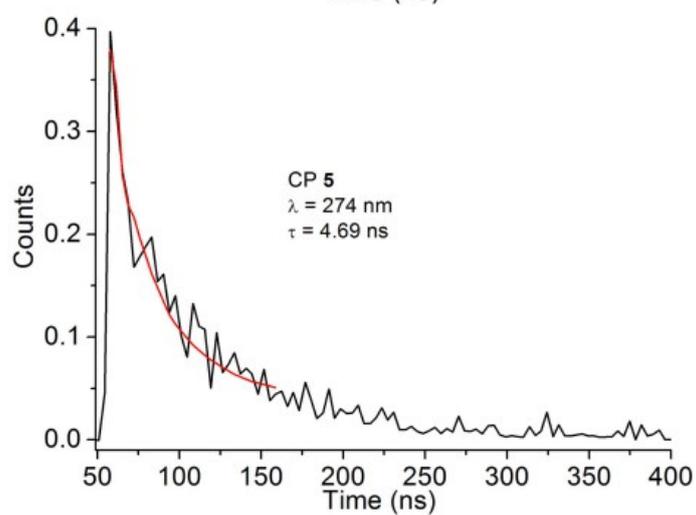
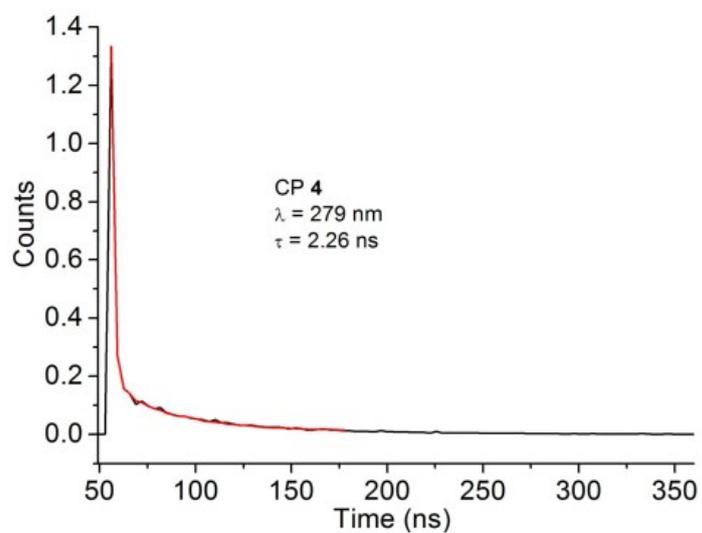


Fig. S11. Decay curves and fitting curves of CPs 4-6