

## **Supplementary Information for:**

# **Providing evidence on the requirements to achieve supramolecular materials based on metal-nucleobase entities**

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## S1. Tables of most relevant coordination bonds.

**Table S1.** Most relevant coordination bonds ( $\text{\AA}$ ) for compound **1**.<sup>[a]</sup>

Compound 1	
Bonds	
Cu1–N3	2.025(4)
Cu1–N9 <sup>i</sup>	2.020(4)
Cu1–Cl1	2.402(2)
Cu1···Cu1 <sup>i</sup>	3.035(2)

[a]Symmetry codes: (i)  $-x, -y, z$ .

**Table S2.** Most relevant coordination bonds ( $\text{\AA}$ ) for compound **2**.

Compound 2			
Bonds			
Co1–N3	2.115(2)	Co2–N9	2.105(2)
Co1–O1w	2.122(2)	Co2–Cl2	2.511(1)
Co1–O2w	2.154 (2)	Co2–Cl3	2.474(1)
Co1–Cl1	2.436(1)	Co1···Co2	3.476(2)
Co1–Cl2	2.411(1)		
Co1–Cl3	2.482(1)		

**Table S3.** Most relevant coordination bonds ( $\text{\AA}$ ) for compound **3**.<sup>[a]</sup>

Compound 3			
Bonds			
Cu1–N29	1.987(4)	Cu1–Cl2A	2.242(3)
Cu1–N19	1.991(5)	Cu1–Cl1A	2.350(3)
Cu1–O1w	2.134(5)	Cu1–Cl2B <sup>i</sup>	2.260(4)
		Cu1···Cu1 <sup>i</sup>	3.658(4)

[a]Symmetry codes: (i)  $-x+1, -y, -z+1$ .

**Table S4.** Most relevant coordination bonds ( $\text{\AA}$ ) for compound **4**.

Compound <b>4</b>	
Bonds	
Co1–O1w	2.091(1)
Co1–N7	2.149(2)

**Table S5.** Most relevant coordination bonds ( $\text{\AA}$ ) for compound **5**.<sup>[a]</sup>

Compound <b>5</b>			
Bonds			
Cu1–O2 <sup>i</sup>	1.952(4)	Cu2–O6 <sup>ii</sup>	1.962(4)
Cu1–O1	1.955(4)	Cu2–O7	1.982(4)
Cu1–O3	1.967(4)	Cu2–N11	2.211(5)
Cu1–O4 <sup>i</sup>	1.977(4)	Cu1···Cu1 <sup>i</sup>	2.643(1)
Cu1–N17	2.211(5)	Cu1···Cu2	7.643(1)
Cu2–O5	1.957(4)	Cu2···Cu2 <sup>ii</sup>	2.662(1)
Cu2–O8 <sup>ii</sup>	1.959(4)		

[a] Symmetry codes: (i)  $-x+1, -y, -z$ ; (ii)  $-x+1, -y+1, -z+1$ .

**Table S6.** Most relevant coordination bonds ( $\text{\AA}$ ) for compound **6**.<sup>[a]</sup>

Compound <b>6</b>			
Bonds			
Co1–O2w	2.086(1)	Co1–O2w <sup>i</sup>	2.086(1)
Co1–O1w	2.122(1)	Co1–O1w <sup>i</sup>	2.122(1)
Co1–N9	2.128(1)	Co1–N9 <sup>i</sup>	2.128(1)

[a] Symmetry code: (i)  $-x+2, -y+1, -z+1$ .

## S2. Tables of non-covalent interactions.

**Table S7.** Structural parameters ( $\text{\AA}$ ,  $^\circ$ ) for non-covalent interactions in compound **2**.<sup>[a]</sup>

D–H $\cdots$ A <sup>[b]</sup>	H $\cdots$ A	D $\cdots$ A	D–H $\cdots$ A
N7–H7 $\cdots$ O3w	1.92	2.731(3)	157.1
N6–H6A $\cdots$ Cl3 <sup>i</sup>	2.70	3.435(2)	144.6
N6–H6B $\cdots$ O3w	2.23	3.054(3)	161.6
O1w–H11 $\cdots$ N1 <sup>ii</sup>	1.93	2.785(3)	176.4
O2w–H21 $\cdots$ Cl1 <sup>iii</sup>	2.20	3.055(2)	168.8
O1w–H12 $\cdots$ O2w <sup>iv</sup>	2.08	2.859(2)	153.6
O3w–H31 $\cdots$ O1w <sup>v</sup>	2.30	3.071(2)	141.4
O3w–H32 $\cdots$ Cl1 <sup>vi</sup>	2.24	3.1437(19)	163.5

[a] Symmetry code: (i) x, y–1,z; (ii)–x+2, –y–1, –z+3; (iii) –x+1, –y, –z+3 (iv)–x+2, –y, –z+3; (v) –x+3, –y–1, –z+2; (vi) –x+2, –y–1, –z+2. [b] **D**: donor; **A**: acceptor.

**Table S8.** Structural parameters ( $\text{\AA}$ ,  $^\circ$ ) for non-covalent interactions in compound **3**.<sup>[a]</sup>

D–H···A <sup>[b]</sup>	H···A	D···A	D–H···A
O1w–H2w···Cl1B <sup>i</sup>	1.92	2.786(10)	153.9
N13–H13···O1w	1.86	2.576(6)	139.4
N23–H23···O1w	1.93	2.604(6)	133.9
N13–H13···Cl1B <sup>i</sup>	2.66	3.389(6)	143.8
N23–H23···Cl1B <sup>i</sup>	2.68	3.393(6)	141.8
N11–H11···O26 <sup>ii</sup>	1.90	2.723(6)	161.1
N12–H12A···Cl1A <sup>i</sup>	2.60	3.421(6)	161.1
N12–H12A···Cl1B <sup>i</sup>	2.37	3.182(6)	158.4
N12–H12B···O26 <sup>ii</sup>	2.42	3.105(6)	137.5
N17–H17···Cl4 <sup>iii</sup>	2.27	3.118(5)	170.0
N21–H21···Cl3 <sup>iv</sup>	2.31	3.123(5)	156.8
N22–H22A···Cl1A <sup>i</sup>	2.62	3.429(7)	156.4
N22–H22A···Cl1B <sup>i</sup>	2.36	3.172(7)	156.7
N22–H22B···Cl3 <sup>iv</sup>	2.52	3.274(6)	147.5
N27–H27···O16 <sup>v</sup>	1.90	2.754(6)	174.7

[a] Symmerty code: (i)  $-x+1, -y+1, -z+1$ ; (ii)  $x, y, z+1$ ; (iii)  $x+1, y, z+1$ ; (iv)  $x, y+1, z$ ; (v)  $x, y, z-1$ . [b] D: donor; A: acceptor.

**Table S9.** Structural parameters ( $\text{\AA}$ ,  $^\circ$ ) for non-covalent interactions in compound **4**.<sup>[a]</sup>

Hydrogen bonding interactions. <sup>[a]</sup>			
D–H···A <sup>[b]</sup>	H···A	D···A	D–H···A
N6–H6A···N1 <sup>i</sup>	2.12	2.974(3)	175.6
N6–H6B···O1w	2.31	2.980(2)	135.3
N6–H6B···O1w <sup>II</sup>	2.31	2.980(2)	135.3
O1w–H1A···C11 <sup>III</sup>	2.13	3.073(1)	167.2
O1w–H1B···O2w	1.93	2.771(2)	171.2
O2w–H2A···N3 <sup>IV</sup>	2.06	2.920(2)	156.6
O2w–H2B···C11	2.35	3.209(2)	171.4
$\pi$ - $\pi$ interactions <sup>c</sup>			
Ring···Ring <sup>[d]</sup>	Angle	DC	$\alpha$
Ring(1)···Ring(1) <sup>v</sup>	0	3.9521(8)	31.62
			3.3654(1)
			2.072

[a] Symmetry: (i)  $-x+1, -y+1, -z+1$ ; (ii)  $x, -y+1, z$ ; (iii)  $-x+2, -y+1, -z+1$ ; (iv)  $-x+3/2, -y+3/2, -z+3/2$ ; (v)  $-x+3/2, y-1/2, -z+3/2$ . [b] **D**: donor; **A**: acceptor. [c] Angle: Dihedral Angle between Planes I and J ( $^\circ$ ), DC: Distance between ring centroids ( $\text{\AA}$ ),  $\alpha$ : Angle  $Cg(I) \rightarrow Cg(J)$  vector and normal to plane I ( $^\circ$ ), DZ: Perpendicular distance of  $Cg(I)$  on ring J ( $\text{\AA}$ ), DXY: Slippage. [d] Ring 1: N1, C2, N3, C4, N9, C6.

**Table S10.** Structural parameters ( $\text{\AA}$ ,  $^\circ$ ) for non-covalent interactions in compound **5**.<sup>[a]</sup>

Hydrogen bonding interactions. <sup>[a]</sup>			
D–H···A <sup>[b]</sup>	H···A	D···A	D–H···A
N16–H16A···O7	1.96	2.793(6)	162.3
N16–H16B···O3	1.94	2.801(6)	178.1
O9–H9···O5 <sup>i</sup>	2.04	2.835(6)	161.8

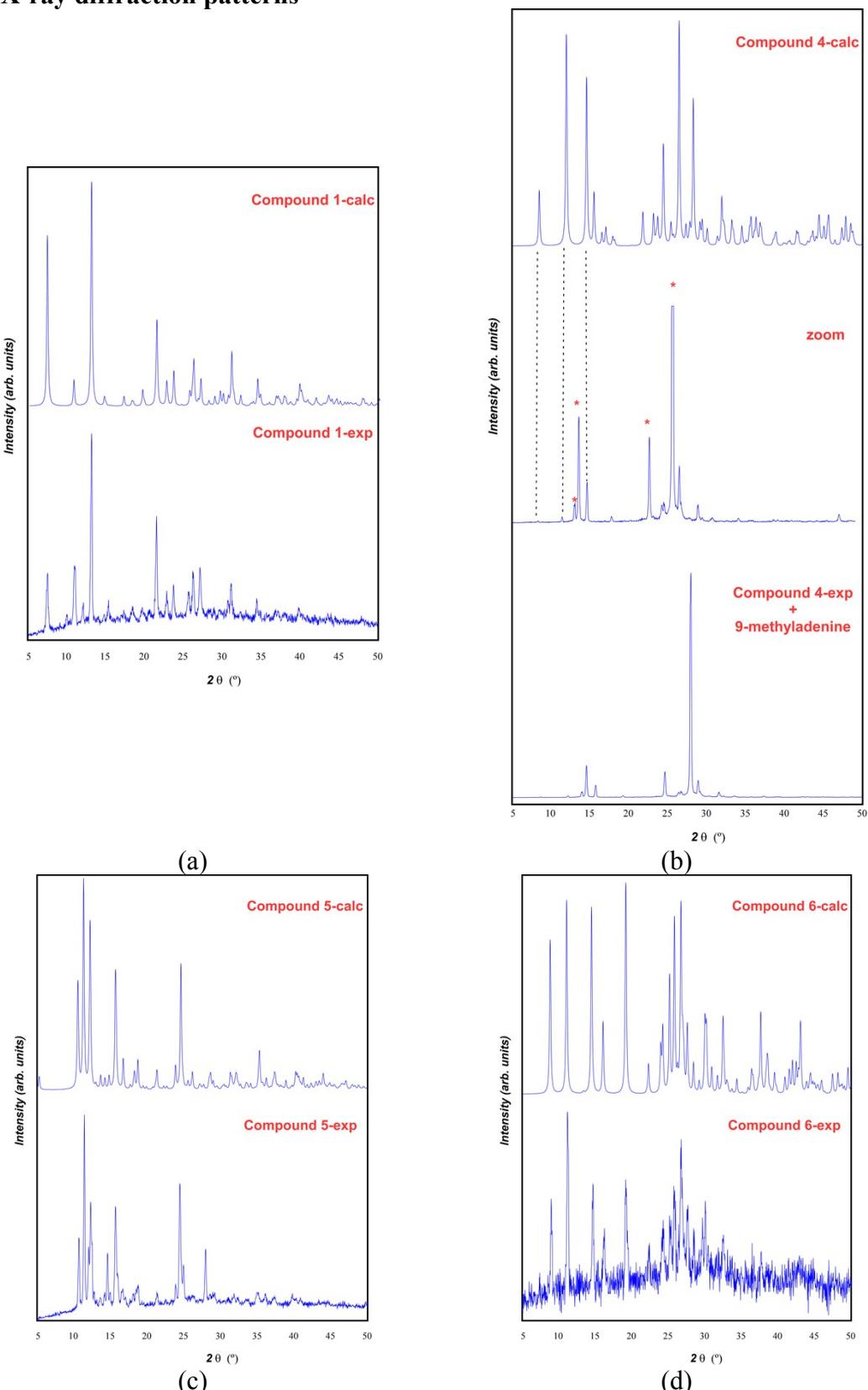
[a] Symmetry code: (i)  $-x+1, -y+1, -z+1$ . [b] **D**: donor; **A**: acceptor.

**Table S11.** Structural parameters ( $\text{\AA}$ ,  $^\circ$ ) for non-covalent interactions in compound **6**.<sup>[a]</sup>

Hydrogen bonding interactions. <sup>[a]</sup>					
D–H···A <sup>[b]</sup>	H···A	D···A	D–H···A		
O1w–H11···N7 <sup>i</sup>	2.24	3.020(2)	149.8		
O1w–H12···N7 <sup>ii</sup>	2.11	2.902(2)	151.1		
O2w–H21···N3 <sup>iii</sup>	1.94	2.737(2)	153.4		
O2w–H22···O4w <sup>iv</sup>	1.81	2.709(2)	169.4		
O3w–H31···O1w	2.19	2.979(2)	153.7		
O4w–H41···N1	1.90	2.827(2)	164.0		
O4w–H42···O3w <sup>v</sup>	1.82	2.673(2)	175.5		
$\pi$ – $\pi$ stacking interactions. <sup>[c]</sup>					
Ring···Ring <sup>[d]</sup>	Angle	DC	$\alpha$	DZ	DXY
Ring1–Ring1 <sup>vi</sup>	2	3.6908(9)	19.22	3.4987(6)	1.1752
Ring1–Ring1 <sup>vii</sup>	2	3.6908(9)	18.56	3.4851(6)	1.2149
Ring1–Ring2 <sup>vi</sup>	1.79(8)	3.6505(8)	18.26	3.4958(6)	1.1434
Ring2–Ring1 <sup>vii</sup>	1.79(8)	3.6506(8)	16.74	3.4668(6)	1.1438
Ring2–Ring2 <sup>vi</sup>	1	3.7471(8)	22.24	3.4857(6)	1.3750
Ring2–Ring2 <sup>vii</sup>	1	3.7472(8)	21.53	3.4684(6)	1.4183

[a] Symmetry: (i)  $-x+2, y-1/2, -z+3/2$ ; (ii)  $x, -y+3/2, z+1/2$ ; (iii)  $-x+2, -y+1, -z+1$ ; (iv)  $x+1, -y+3/2, z+1/2$ ; (v)  $-x+1, y+1/2, -z+3/2$ ; (vi)  $x, 3/2-y, -1/2+z$ ; (vii)  $x, 3/2-y, 1/2+z$ ; [b] D: donor; A: acceptor. [c] Angle: Dihedral Angle between Planes I and J ( $^\circ$ ), DC: Distance between ring centroids ( $\text{\AA}$ ),  $\alpha$ : Angle Cg(I)–>Cg(J) vector and normal to plane I ( $^\circ$ ), DZ: Perpendicular distance of Cg(I) on ring J ( $\text{\AA}$ ), DXY: Slippage. [d] Ring1: N7, C5, C4, N9, C8; Ring 2: N1, C2, N3, C4, C5, C6.

### S3. X-ray diffraction patterns



**Figure S1.** Experimental and calculated X-ray diffraction patterns of: (a) compound **1**, (b) compound **4**, (c) compound **5** and (d) compound **6**. Note that compound **4** appears cocrystallized with 9-methyladenine, so a zoom has been included in order to show more clearly the peaks corresponding to compound **4**. \* corresponds to the peaks assigned to 9-methyladenine, and some discontinuous lines have been included which correspond to compound **4**.

#### S4. Thermogravimetric data

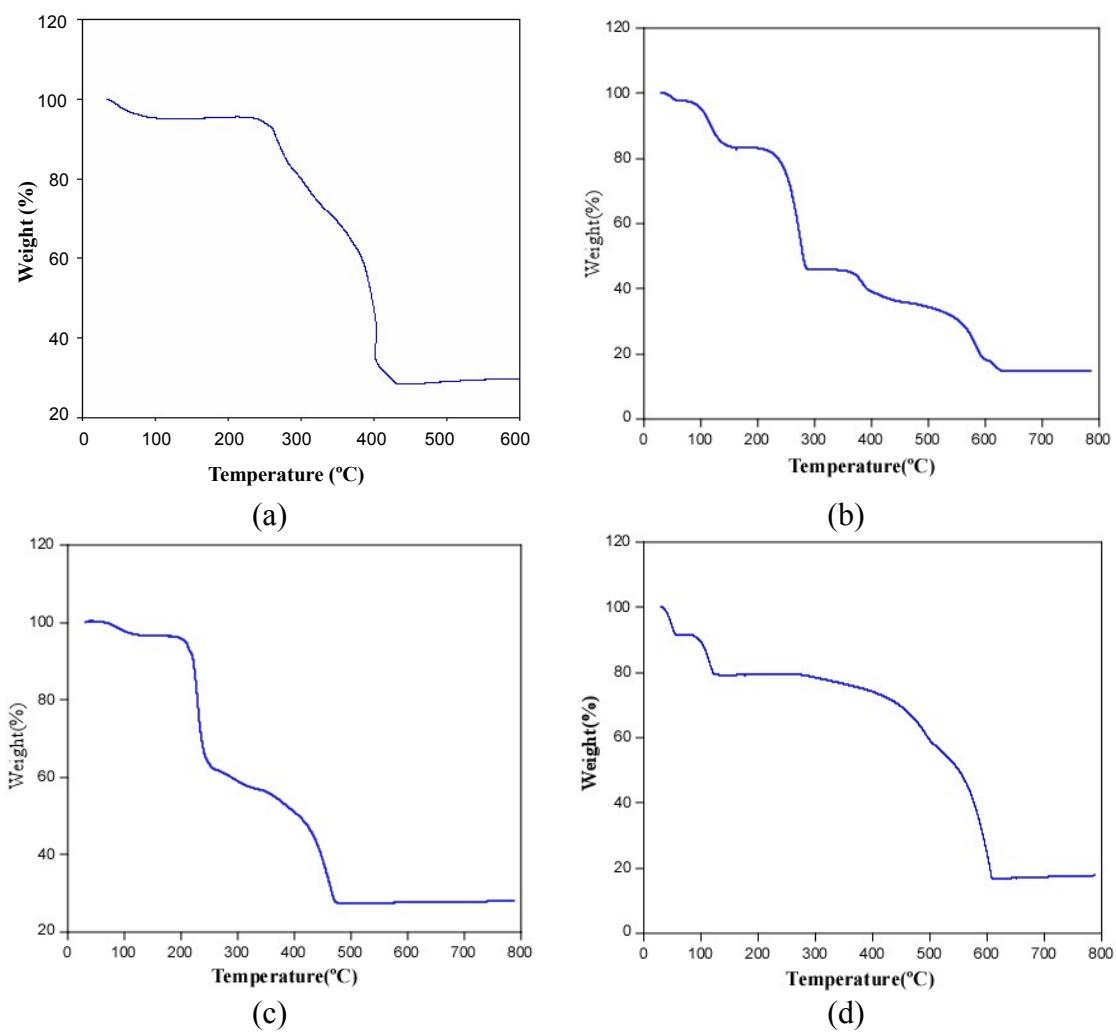


Figure S2. Thermogravimetric data of: (a) compound **1**, (b) compound **4**, (c) compound **5** and (d) compound **6**.