## **Electronic Supplementary Information (ESI)**

## Dimethylamine Substituted Bisbenzocoumarins: Solvatochromic,

### Mechanochromic and Acidochromic properties

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# Table 1 Crystal data and structure refinement for DB-C2.

Empirical formula	$C_{36}H_{32}N_2O_8$
Formula weight	620.63
Temperature/K	100.00(10)
Crystal system	monoclinic
Space group	C2/c
a/Å	29.9695(15)
b/Å	8.1566(5)
c/Å	12.1516(7)
α/°	90
β/°	90.466(5)
γ/°	90
Volume/Å <sup>3</sup>	2970.3(3)
Z	4
$\rho_{calc}g/cm^3$	1.388
µ/mm <sup>-1</sup>	0.813
F(000)	1304.0
Crystal size/mm <sup>3</sup>	$0.14 \times 0.13 \times 0.12$
Radiation	$CuK\alpha (\lambda = 1.54184)$
$2\Theta$ range for data collection/°	5.898 to 146.762
Index ranges	$-37 \le h \le 28, -9 \le k \le 10, -14 \le l \le 13$
Reflections collected	5681
Independent reflections	2896 [ $R_{int} = 0.0489$ , $R_{sigma} = 0.0520$ ]
Data/restraints/parameters	2896/0/215
Goodness-of-fit on F <sup>2</sup>	1.023
Final R indexes [I>= $2\sigma$ (I)]	$R_1 = 0.0743, wR_2 = 0.2221$
Final R indexes [all data]	$R_1 = 0.0909, wR_2 = 0.2329$
Largest diff. peak/hole / e Å <sup>-3</sup>	0.46/-0.40



Figure S1 FL emission spectra in DCM with increasing concentrations. (A) Spectra of **DB-C2**, (B) Spectra of **DB-C6**, (C) Spectra of **DB-C12**, concentration range:  $1 \mu M - 3 mM$ ,  $\lambda_{ex} = 415 nm$ .



**Figure S2** The control <sup>1</sup>H NMR experiment. (a) The <sup>1</sup>H NMR spectra of **DB-C2** in CDCl<sub>3</sub>. (b) The <sup>1</sup>H NMR spectra of **DB-C2** with trifluoroacetic acid (10 eq.) in CDCl<sub>3</sub>.



Figure S4 <sup>13</sup>C NMR of DB-C2 in CD<sub>2</sub>Cl<sub>2</sub>.



Figure S6  $^{13}$ C NMR of 2 in D<sub>2</sub>O.





Figure S8 <sup>13</sup>C NMR of DB-C6 in CDCl<sub>3</sub>.



Figure S10 <sup>13</sup>C NMR of DB-C12 in CDCl<sub>3</sub>