

Construction of d¹⁰ metal coordination polymers based on *in situ* formed 3,5-di(1H-1,2,4-triazol-1-yl)benzoic acid from different precursors: influence of *in situ* hydrolysis reactions on assembly process

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Supporting information

Table S1 Selected bond lengths [Å] and angles[°] for the compounds 1-7.^a

Compound 1			
Zn(1)-O(1)	1.9415(15)	Zn(1)-N(6)#1	1.9975(17)
Zn(1)-O(3)	1.9445(16)	Zn(1)-N(3)#2	2.0372(17)
O(1)-Zn(1)-O(3)	106.47(7)	O(1)-Zn(1)-N(3)#2	96.18(7)
O(1)-Zn(1)-N(6)#1	120.24(8)	O(3)-Zn(1)-N(3)#2	112.13(7)
O(3)-Zn(1)-N(6)#1	115.48(7)	N(6)#1-Zn(1)-N(3)#2	104.67(7)
Compound 2			
Cd(1)-O(2)	2.282(3)	Cd(1)-N(6)#3	2.395(3)
Cd(1)-O(2)#1	2.282(3)	Cd(1)-N(1)#4	2.444(3)
Cd(1)-N(6)#2	2.395(3)	Cd(1)-N(1)#5	2.444(3)
O(2)-Cd(1)-O(2)#1	180	N(6)#2-Cd(1)-N(1)#4	88.54(11)
O(2)-Cd(1)-N(6)#2	89.16(14)	N(6)#3-Cd(1)-N(1)#4	91.46(11)
O(2)#1-Cd(1)-N(6)#2	90.84(14)	O(2)-Cd(1)-N(1)#5	103.22(12)
O(2)-Cd(1)-N(6)#3	90.84(14)	O(2)#1-Cd(1)-N(1)#5	76.78(12)
O(2)#1-Cd(1)-N(6)#3	89.16(14)	N(6)#2-Cd(1)-N(1)#5	91.46(11)
N(6)#2-Cd(1)-N(6)#3	180	N(6)#3-Cd(1)-N(1)#5	88.54(11)
O(2)-Cd(1)-N(1)#4	76.78(12)	N(1)#4-Cd(1)-N(1)#5	180
O(2)#1-Cd(1)-N(1)#4	103.22(12)		
Compound 3			
Cd(1)-O(3)#1	2.282(4)	Cd(1)-O(4)#3	2.320(4)
Cd(1)-O(2)#2	2.288(4)	Cd(1)-N(12)	2.323(4)
Cd(1)-N(6)#3	2.315(4)	Cd(1)-O(1)	2.327(4)
O(3)#1-Cd(1)-O(2)#2	179.81(19)	N(6)#3-Cd(1)-N(12)	179.89(19)
O(3)#1-Cd(1)-N(6)#3	82.45(15)	O(4)#3-Cd(1)-N(12)	84.82(16)
O(2)#2-Cd(1)-N(6)#3	97.50(15)	O(3)#1-Cd(1)-O(1)	99.22(15)
O(3)#1-Cd(1)-O(4)#3	80.95(15)	O(2)#2-Cd(1)-O(1)	80.59(15)
O(2)#2-Cd(1)-O(4)#3	99.24(15)	N(6)#3-Cd(1)-O(1)	85.18(15)
N(6)#3-Cd(1)-O(4)#3	95.12(16)	O(4)#3-Cd(1)-O(1)	179.67(16)
O(3)#1-Cd(1)-N(12)	97.45(16)	N(12)-Cd(1)-O(1)	94.88(16)
O(2)#2-Cd(1)-N(12)	82.60(16)		
Compound 4			
Cd(1)-N(9)	2.3246(19)	Cd(1)-N(6)#3	2.364(2)
Cd(1)-N(3)	2.328(2)	Cd(1)-O(4)#1	2.4600(18)
Cd(1)-O(3)#1	2.3473(18)	Cd(1)-O(2)#2	2.582(2)
Cd(1)-O(1)#2	2.3526(18)	N(9)-Cd(1)-O(4)#1	90.32(7)
N(9)-Cd(1)-N(3)	176.91(7)	N(3)-Cd(1)-O(4)#1	92.44(8)
N(9)-Cd(1)-O(3)#1	97.17(7)	O(3)#1-Cd(1)-O(4)#1	54.26(6)

N(3)-Cd(1)-O(3)#1	85.58(8)	O(1)#2-Cd(1)-O(4)#1	159.19(7)
N(9)-Cd(1)-O(1)#2	81.64(7)	N(6)#3-Cd(1)-O(4)#1	80.57(7)
N(3)-Cd(1)-O(1)#2	95.29(8)	N(9)-Cd(1)-O(2)#2	94.27(7)
O(3)#1-Cd(1)-O(1)#2	145.60(7)	N(3)-Cd(1)-O(2)#2	84.13(8)
N(9)-Cd(1)-N(6)#3	89.35(8)	O(3)#1-Cd(1)-O(2)#2	93.68(6)
N(3)-Cd(1)-N(6)#3	89.72(8)	O(1)#2-Cd(1)-O(2)#2	52.43(7)
O(3)#1-Cd(1)-N(6)#3	134.22(7)	N(6)#3-Cd(1)-O(2)#2	131.10(7)
O(1)#2-Cd(1)-N(6)#3	80.17(7)	O(4)#1-Cd(1)-O(2)#2	147.95(6)

Compound 5

Zn(1)-O(2)#1	1.9630(14)	Zn(1)-N(6)	2.0324(15)
Zn(1)-N(3)#2	2.0309(15)	Zn(1)-Cl(1)	2.1948(6)
O(2)#1-Zn(1)-N(3)#2	104.37(7)	O(2)#1-Zn(1)-Cl(1)	127.03(5)
O(2)#1-Zn(1)-N(6)	94.06(6)	N(3)#2-Zn(1)-Cl(1)	106.37(5)
N(3)#2-Zn(1)-N(6)	108.02(7)	N(6)-Zn(1)-Cl(1)	115.42(5)

Compound 6

Zn(1)-O(2)#1	1.924(2)	Zn(1)-O(3)	2.003(3)
Zn(1)-N(6)#2	1.988(3)	Zn(1)-N(1)	2.032(3)
O(2)#1-Zn(1)-N(6)#2	122.07(12)	O(2)#1-Zn(1)-N(1)	95.00(11)
O(2)#1-Zn(1)-O(3)	104.20(12)	N(6)#2-Zn(1)-N(1)	112.19(12)
N(6)#2-Zn(1)-O(3)	111.32(11)	O(3)-Zn(1)-N(1)	110.77(12)

Compound 7

Cd(1)-N(1)	2.251(16)	Cd(1)-O(3)	2.34(2)
Cd(1)-N(6)#1	2.283(16)	Cd(1)-O(1W)	2.35(2)
Cd(1)-O(2)#2	2.322(14)	Cd(1)-O(1)#2	2.458(15)
N(1)-Cd(1)-N(6)#1	119.0(6)	O(2)#2-Cd(1)-O(1W)	87.8(8)
N(1)-Cd(1)-O(2)#2	140.7(6)	O(3)-Cd(1)-O(1W)	175.7(8)
N(6)#1-Cd(1)-O(2)#2	99.0(6)	N(1)-Cd(1)-O(1)#2	86.7(5)
N(1)-Cd(1)-O(3)	99.1(7)	N(6)#1-Cd(1)-O(1)#2	153.5(5)
N(6)#1-Cd(1)-O(3)	90.9(7)	O(2)#2-Cd(1)-O(1)#2	54.6(5)
O(2)#2-Cd(1)-O(3)	89.2(7)	O(3)-Cd(1)-O(1)#2	91.4(6)
N(1)-Cd(1)-O(1W)	85.2(8)	O(1W)-Cd(1)-O(1)#2	89.4(8)
N(6)#1-Cd(1)-O(1W)	86.5(8)		

^aSymmetry transformations used to generate equivalent atoms : #1 $x + 1, y, z$, #2 $x, y - 1, z$, #3 $x - 1, y, z$, #4 $x, y + 1, z$ for compound 1; #1 $-x + 1, -y + 1, -z$, #2 $-x + 2, y - 1/2, -z + 1/2$, #3 $x - 1, -y + 3/2, z - 1/2$, #4 $x, y, z - 1$, #5 $-x + 1, -y + 1, -z + 1$, #6 $-x + 2, y + 1/2, -z + 1/2$, #7 $x, y, z + 1$ for compound 2; #1 $x, y, z + 1$, #2 $x - 1, y, z$, #3 $x - 1, y, z + 1$, #4 $x + 1, y, z - 1$, #5 $x, y, z - 1$, #6 $x + 1, y, z$ for compound 3; #1 $-x + 3/2, y - 1/2, -z + 1/2$, #2 $-x + 1, y + 1, -z + 1/2$, #3 $x, -y - 1, z - 1/2$, #4 $-x + 3/2, y + 1/2, -z + 1/2$, #5 $-x + 1, y - 1, -z + 1/2$, #6 $x, -y - 1, z + 1/2$ for compound 4; #1 $-x + 1, -y + 1, -z + 1$, #2 $-x + 2, y + 1/2, -z + 3/2$, #3 $-x + 2, y - 1/2, -z + 3/2$ for compound 5; #1 $x - 1, y, z$, #2 $x - 1/2, -y + 3/2, z + 1/2$, #3 $x + 1/2, -y + 3/2, z - 1/2$, #4 $x + 1, y, z$ for compound 6; #1 $x + 1, y, z + 1$, #2 $-x - 1, y + 1/2, -z + 1/2$, #3 $-x - 1, y - 1/2, -z + 1/2$, #4 $x - 1, y, z - 1$ for compound 7;

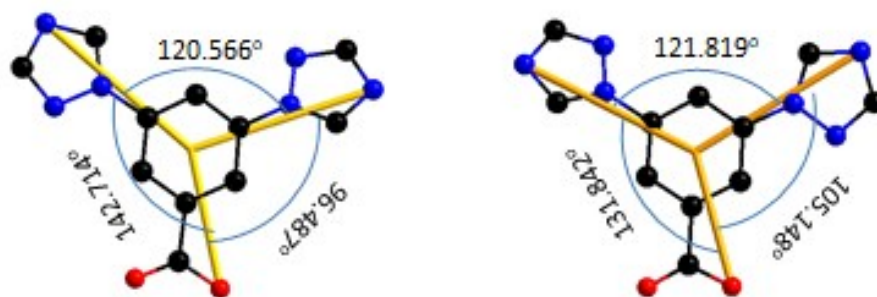


Figure S1. The angles between the center of benzene ring and the three terminal coordination atoms in compounds **5** and **6**.

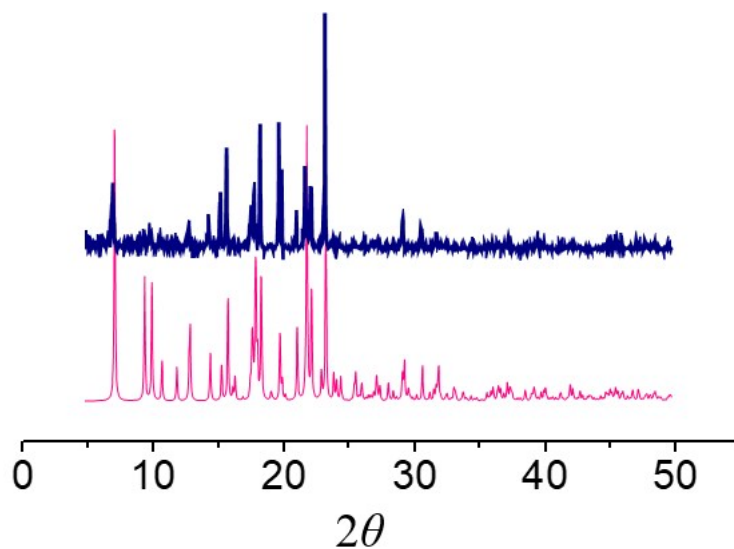


Figure S2. The simulated X-ray powder diffraction patterns (lower) and the measured one (upper) of compound **1**.

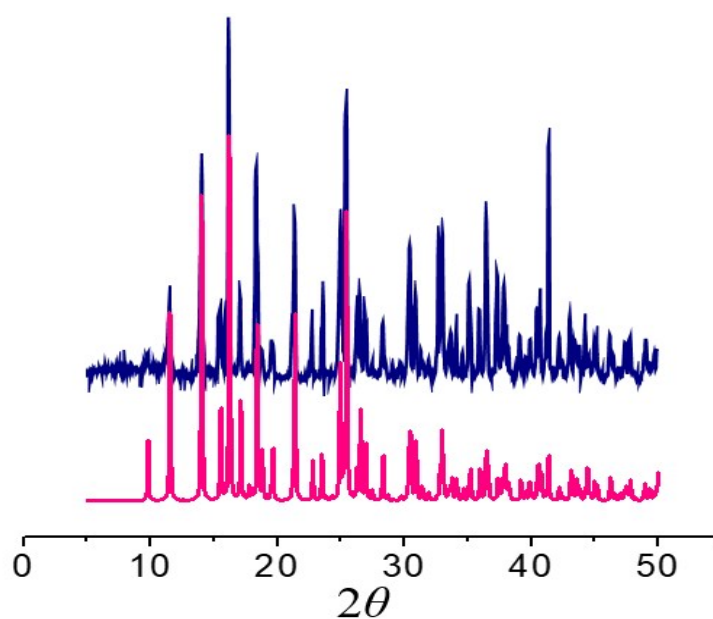


Figure S3. The simulated X-ray powder diffraction patterns (lower) and the measured one (upper) of compound **2**.

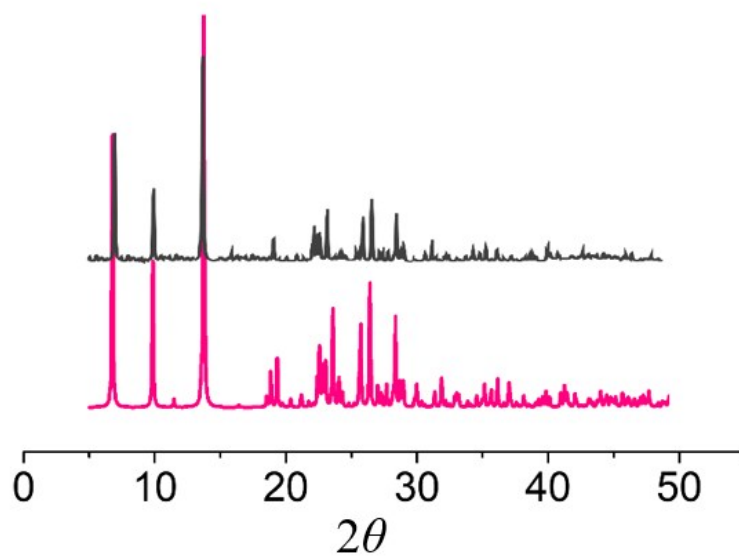


Figure S4. The simulated X-ray powder diffraction patterns (lower) and the measured one (upper) of compound 3.

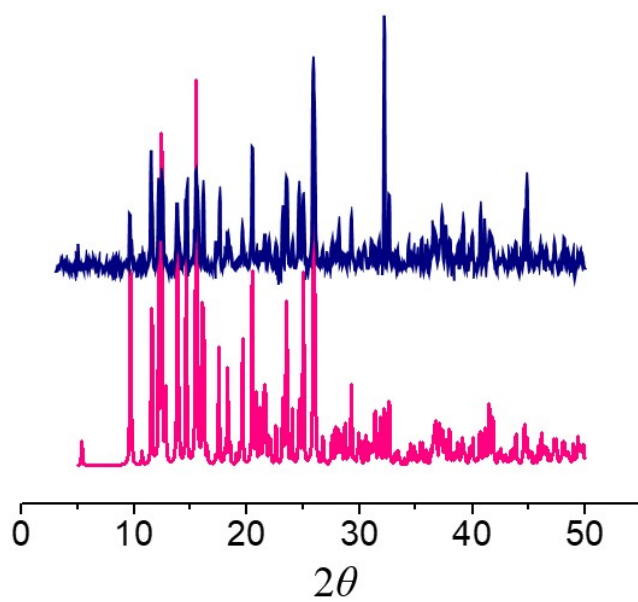


Figure S5. The simulated X-ray powder diffraction patterns (lower) and the measured one (upper) of compound 4.

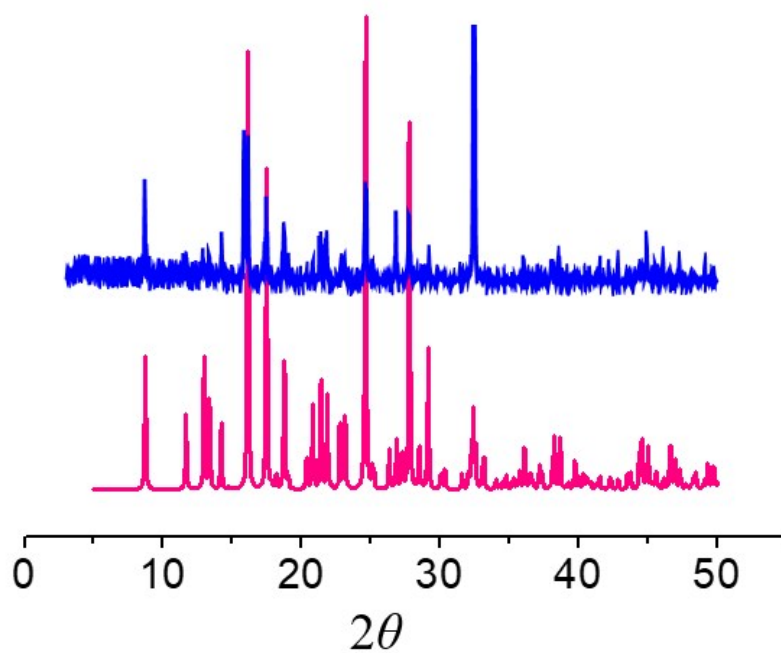


Figure S6. The simulated X-ray powder diffraction patterns (lower) and the measured one (upper) of compound **5**.

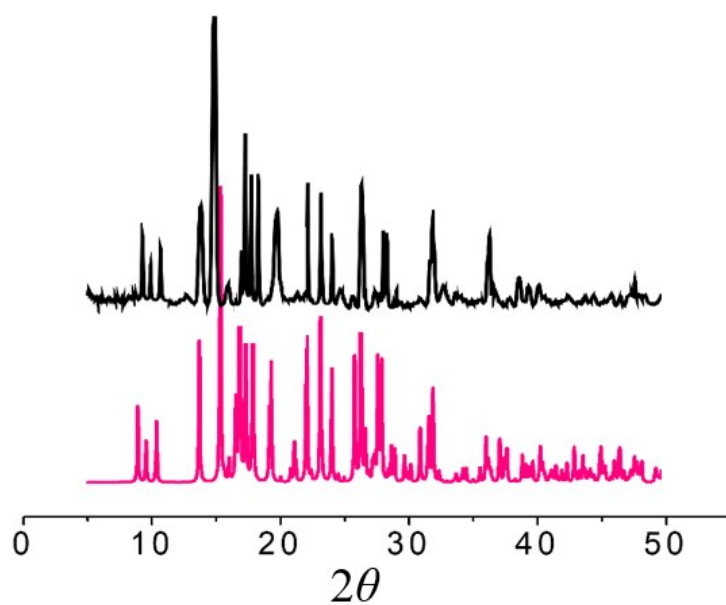


Figure S7. The simulated X-ray powder diffraction patterns (lower) and the measured one (upper) of compound **6**.

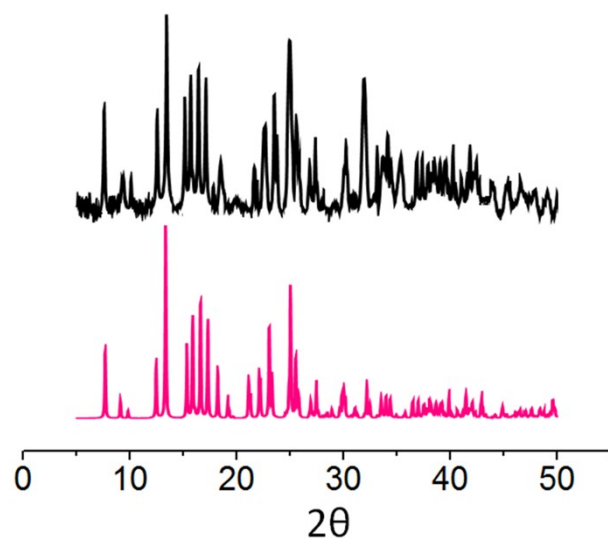


Figure S8. The simulated X-ray powder diffraction patterns (lower) and the measured one (upper) of compound 7.

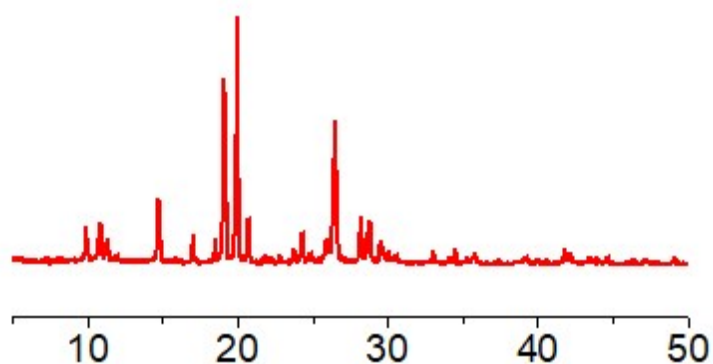


Figure S9. The measured XRD of product obtained by reaction of $\text{Zn}(\text{NO}_3)_2 \cdot 6\text{H}_2\text{O}$ with HDTBA.

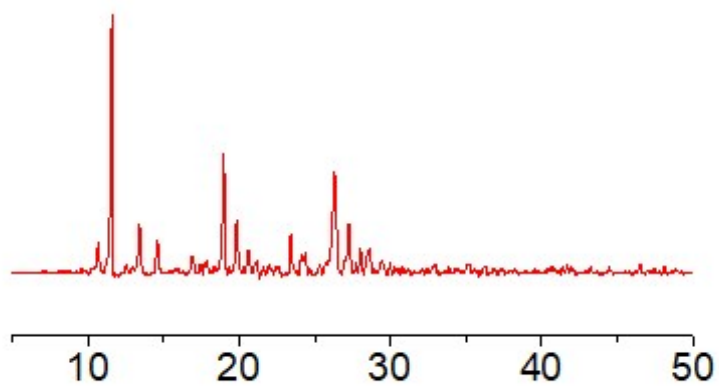


Figure S10. The measured XRD of product obtained by reaction of $\text{ZnCl}_2 \cdot 2\text{H}_2\text{O}$ with HDTBA.

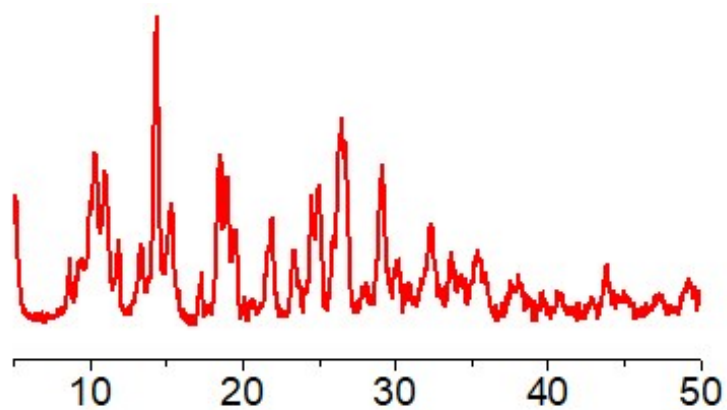


Figure S11. The measured PXRD of product obtained by reaction of $\text{Cd}(\text{NO}_3)_2 \cdot 4\text{H}_2\text{O}$ with HDTBA.

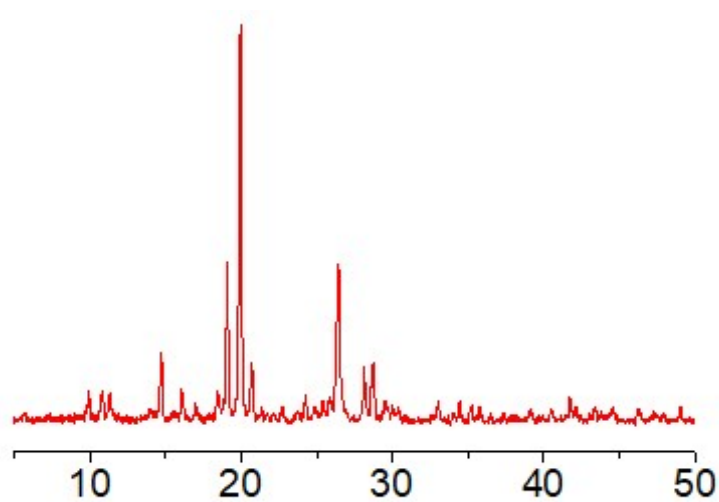


Figure S12. The measured PXRD of product obtained by reaction of $\text{Cd}(\text{NO}_3)_2 \cdot 4\text{H}_2\text{O}$ with HDTBA at the present of $\text{NH}_3 \cdot \text{H}_2\text{O}$.

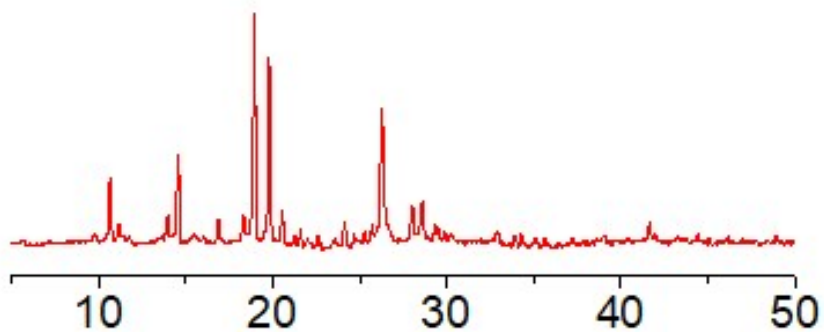


Figure S13. The measured PXRD of product obtained by reaction of $\text{Cd}(\text{ClO}_4)_2 \cdot 6\text{H}_2\text{O}$ with HDTBA.

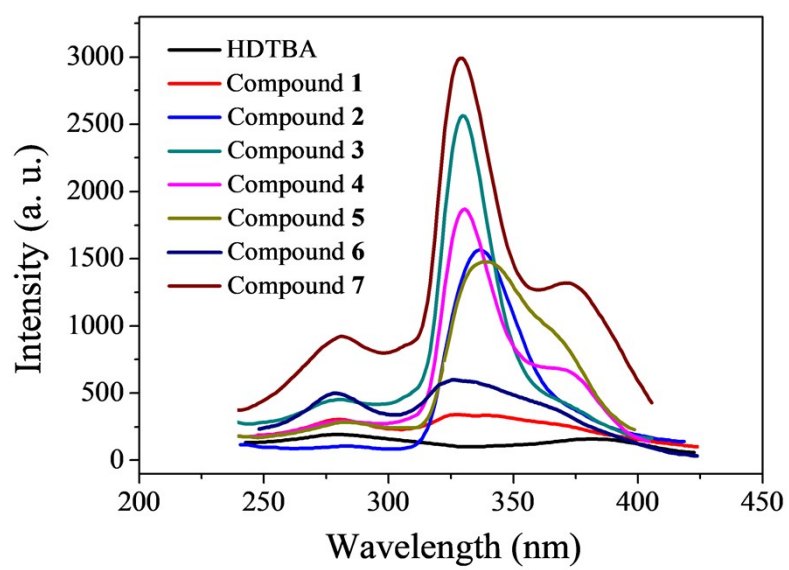


Figure S14. The solid-state excitation spectra for compounds 1-7 at room temperature.