



**Journal Name**

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Supporting Information

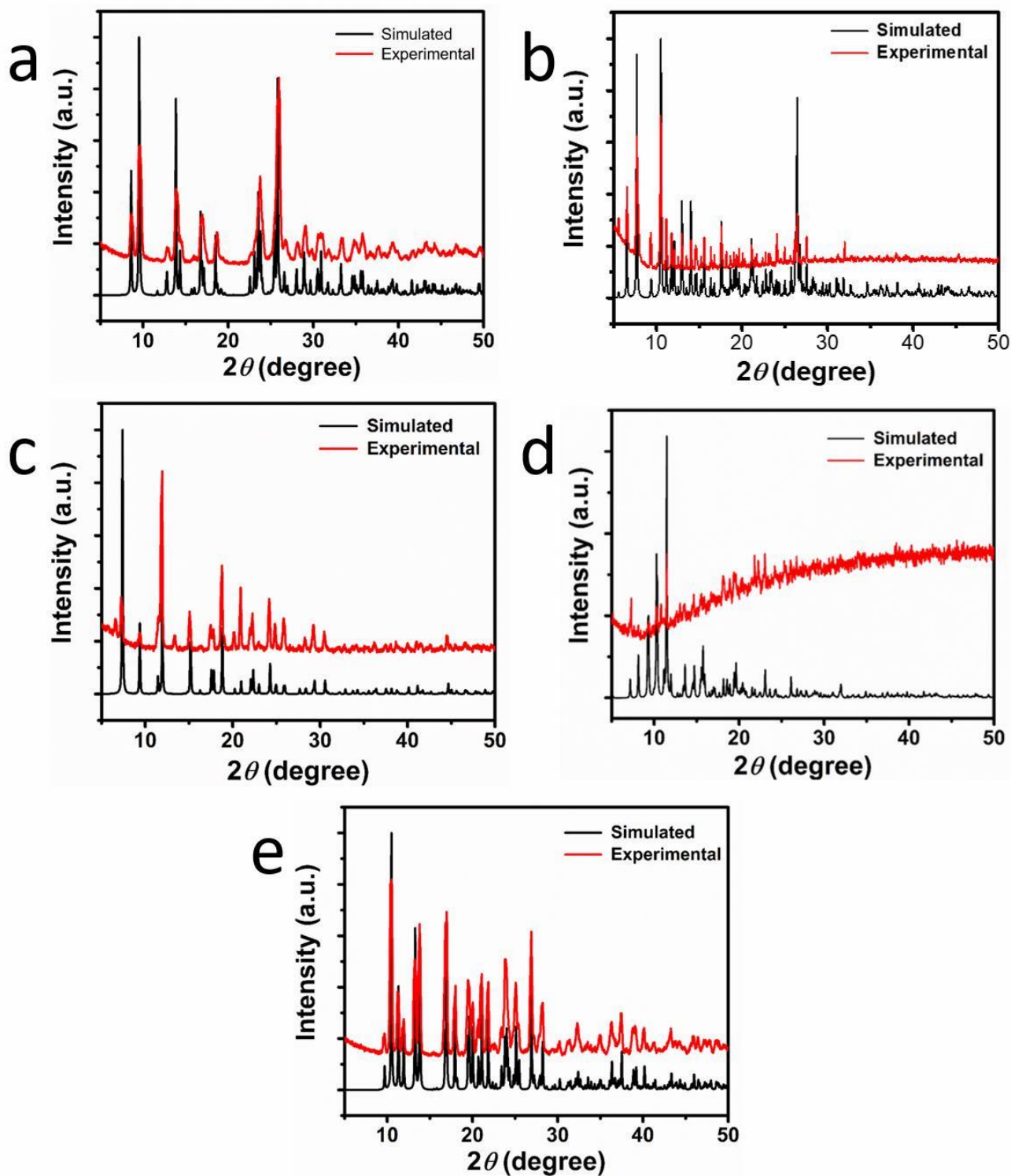
**Hybrid materials based on transition Metal-BTC-Benzimidazole: Solvent assisted crystallographic and structural switching**

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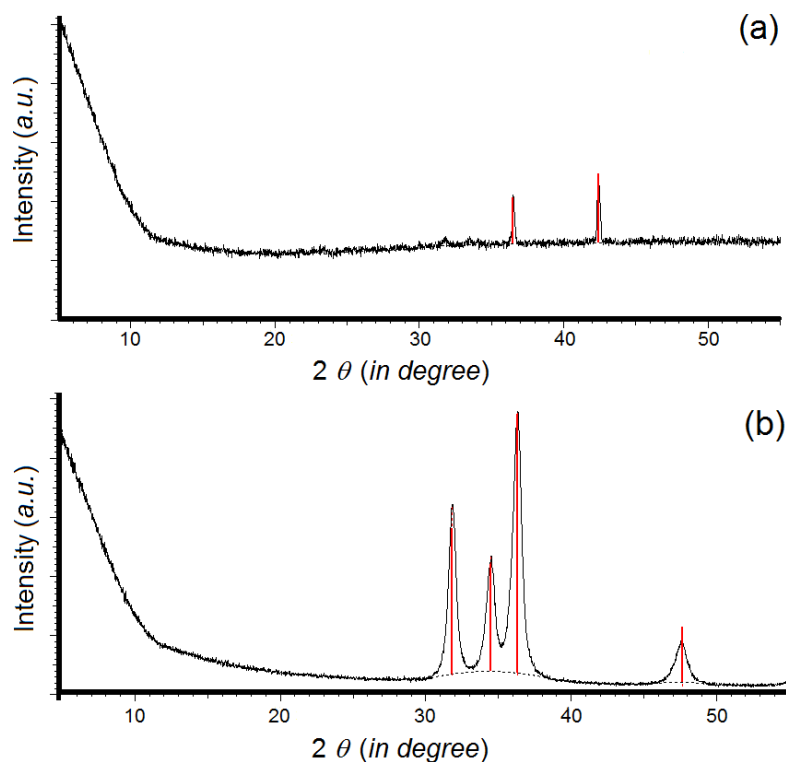
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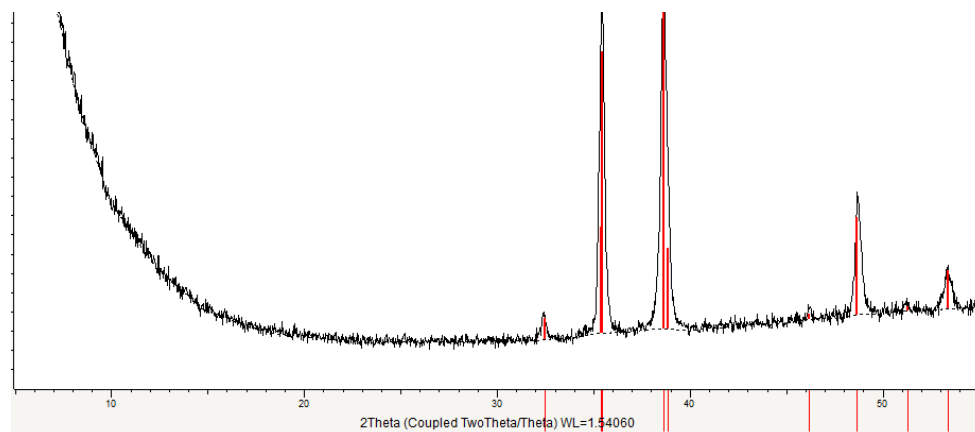
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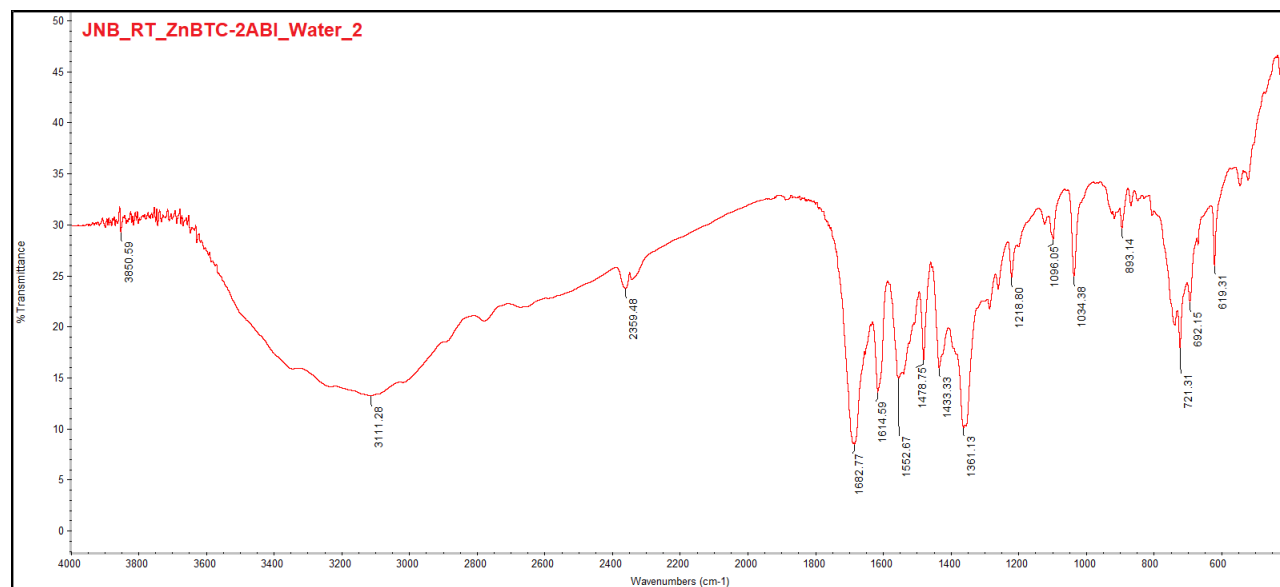
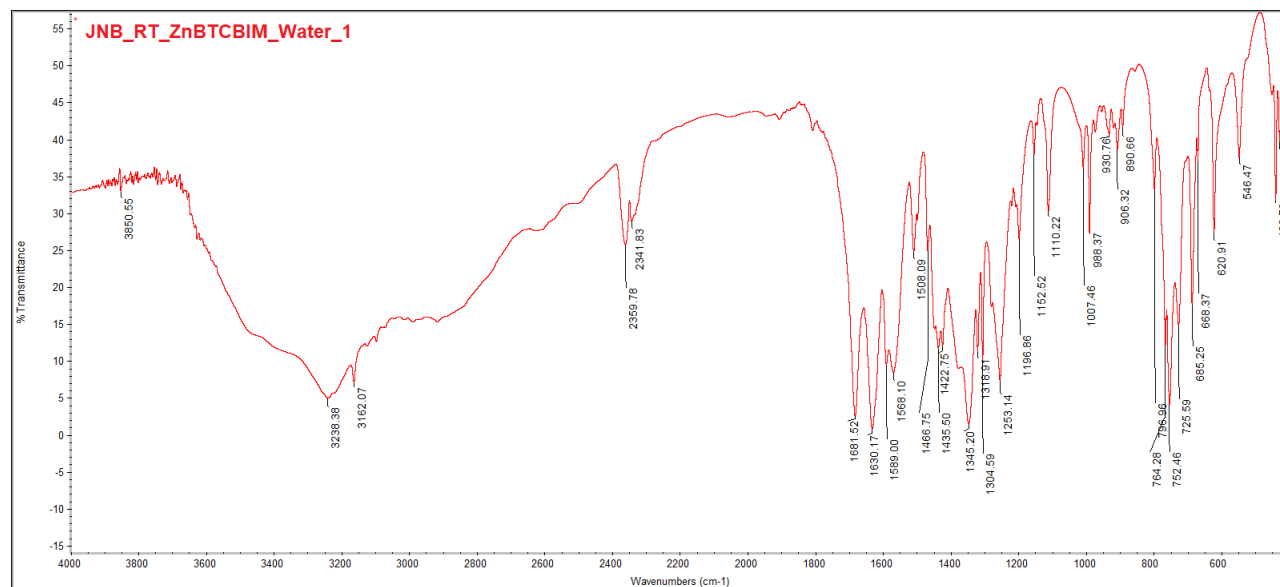
**Figure S1.** PXRD pattern for (a)  $[\text{Zn}_2(\text{H}_2\text{O})_2(\text{HBTC})_2(\text{Blm})_2](\text{H}_2\text{O})_3$ , **1**, (b)  $[\text{2-HABl}]_2[\text{Zn}_2(\text{BTC})_2(\text{2-ABI})(\text{H}_2\text{O})](\text{H}_2\text{O})_5$ , **2**, (c)  $[\text{Cu}_3(\text{BTC})_2(\text{Blm})_6][\text{solvent}]$ , **3**, (d)  $[\text{Co}_4(\text{BTC})_3(\text{Blm})_6][\text{solvent}]$ , **4** and (e)  $[(\text{CH}_3\text{CH}_2)_2\text{NH}_2][\text{Zn}(\text{BTC})(\text{Blm})](\text{H}_2\text{O})$ , **5**.

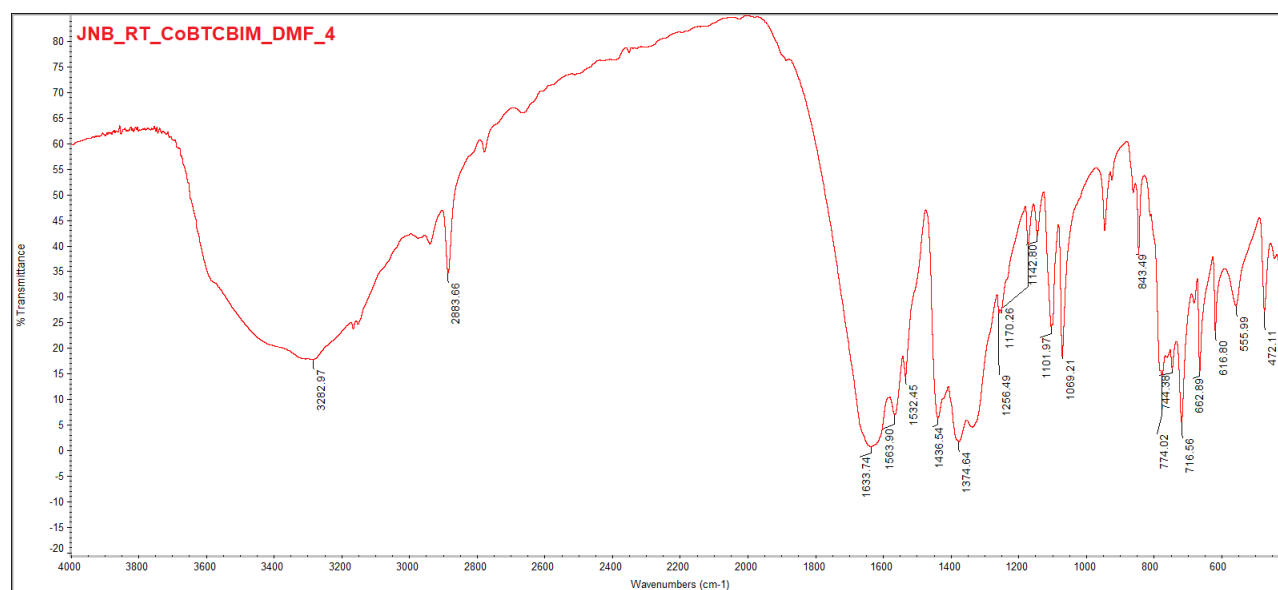
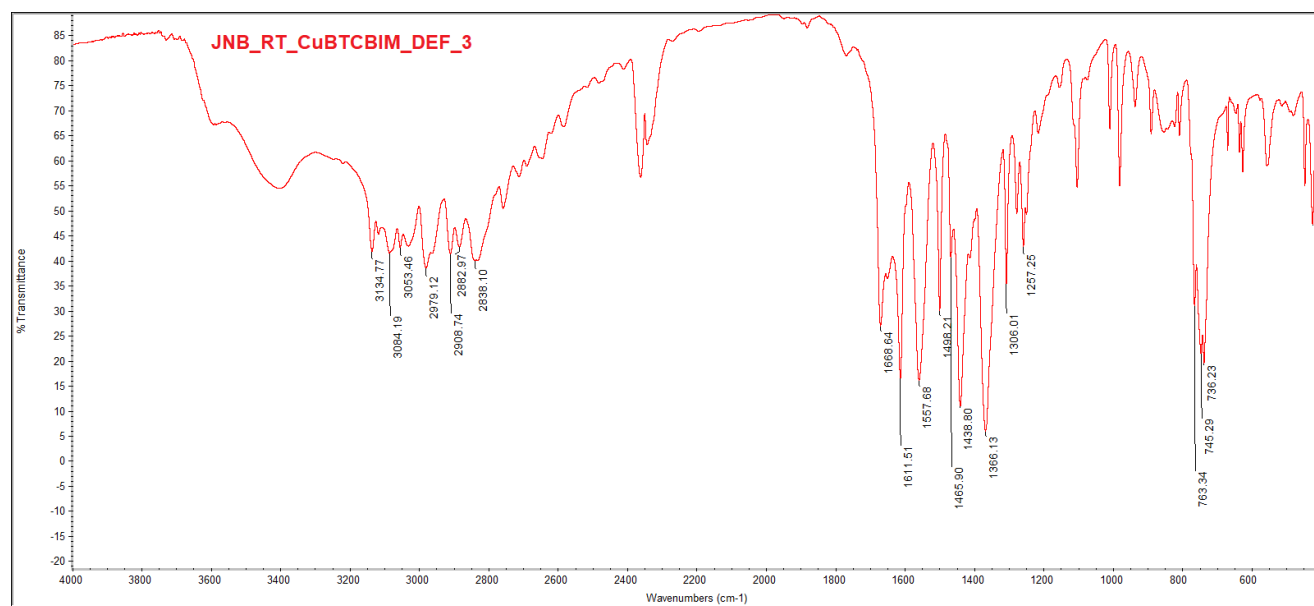


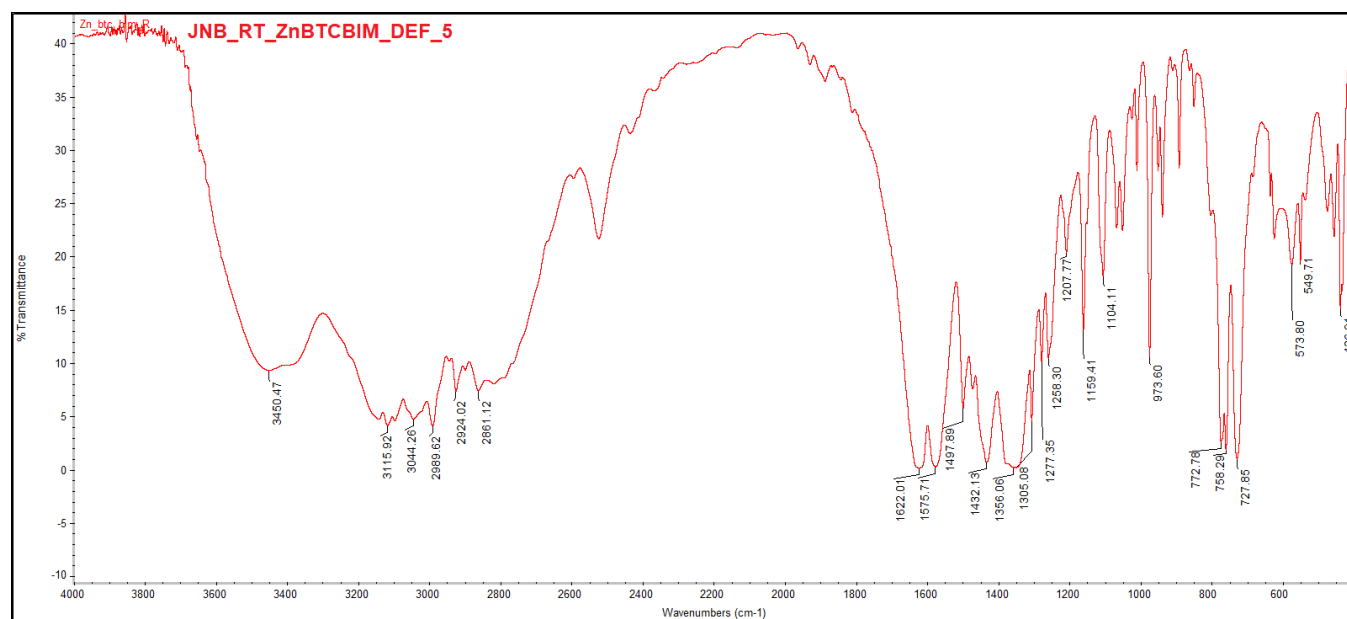
**Figure S2.** Post calcination PXRD analysis of (a)  $[\text{Co}_4(\text{BTC})_3(\text{BIm})_6][\text{solvent}]$ , **4** showing formation of CoO (PDF-00-043-1004) as final product and (b) Post calcination PXRD analysis of  $[\text{Zn}_2(\text{H}_2\text{O})_2(\text{HBTC})_2(\text{BIm})_2](\text{H}_2\text{O})_3$ , **1**,  $[2\text{-HABI}]_2[\text{Zn}_2(\text{BTC})_2(2\text{-ABI})(\text{H}_2\text{O})].(\text{H}_2\text{O})_5$ , **2**,  $[(\text{CH}_3\text{CH}_2)_2\text{NH}_2][\text{Zn}(\text{BTC})(\text{BIm})].(\text{H}_2\text{O})$ , **5**, whereas the ZnO (PDF-01-075-0576) as calcined product.



**Figure S3.** Post calcination PXRD analysis of  $[\text{Cu}_3(\text{BTC})_2(\text{BIm})_6][\text{solvent}]$ , **3**, showing formation of CuO (COD-9016326).







**Figure S4.** FTIR Spectra diagram of 1-5.

**Table S1.** Selected bond length tables of **1-5**

[Zn <sub>2</sub> (H <sub>2</sub> O) <sub>2</sub> (HBTC) <sub>2</sub> (BIm) <sub>2</sub> ](H <sub>2</sub> O) <sub>3</sub> , <b>1</b>					
Zn1-O1	1.9581(9)	Zn1-O4 <sup>a</sup>	1.9722(8)	Zn1-O1W	2.0313(10)
Zn1-N1	1.9693(10)				
[2-HABI] <sub>2</sub> [Zn <sub>2</sub> (BTC) <sub>2</sub> (2-ABI)(H <sub>2</sub> O)].(H <sub>2</sub> O) <sub>5</sub> , <b>2</b>					
Zn1-O1	1.9506(18)	Zn1-O6 <sup>b</sup>	1.9675(19)	Zn1-O7	1.9770(17)
Zn1-N1	1.975(2)	Zn2 – O3	1.9766(18)	Zn2-O9 <sup>c</sup>	1.9845(18)
Zn2-O12 <sup>d</sup>	2.0195(18)	Zn2-O1W	2.023(2)		
[Cu <sub>3</sub> (BTC) <sub>2</sub> (BIm) <sub>6</sub> ][solvent], <b>3</b>					
Cu1-O1 <sup>e</sup>	1.9702(9)	Cu1-O1	1.9702(9)	Cu1-N1	1.9851(14)
Cu1-N1 <sup>e</sup>	1.9851(14)				
[Co <sub>4</sub> (BTC) <sub>3</sub> (BIm) <sub>6</sub> ][solvent], <b>4</b>					
Co1-Co1 <sup>f</sup>	2.8713(8)	Co1-O3	2.024(2)	Co1-O4 <sup>f</sup>	2.058(2)
Co1-O7	2.048(2)	Co1-O8 <sup>f</sup>	2.039(2)	Co1-N5	2.025(3)
Co2-O1	1.971(2)	Co2-O5 <sup>g</sup>	2.405(3)	Co2-O6 <sup>g</sup>	2.019(2)
Co2-N1	2.037(3)	Co2-N3	2.014(3)		
[(CH <sub>3</sub> CH <sub>2</sub> ) <sub>2</sub> NH <sub>2</sub> ][Zn(BTC)(BIm)].(H <sub>2</sub> O), <b>5</b>					
Zn1-O1	1.951(2)	Zn1-O3 <sup>h</sup>	1.980(2)	Zn1-O6 <sup>i</sup>	1.986(2)
Zn1-N1	2.027(3)				

Symmetry transformations used to generate equivalent atoms:

a)  $x+1/2, -y+3/2, z+1/2$ , b)  $-x+1, -y+1, -z+1$  c)  $x, y, z-1$  d)  $x-1, y, z-1$  e)  $1-x, 1-y, 1-z$ ; f)  $-x+1, -y+1, -z+1$  g)  $-x+1/2, y-1/2, -z+1/2$  h)  $x-1/2, -y+1/2, z$  i)  $-x+1, -y+1, z-1/2$ .

**Table S2.** Classical Hydrogen bonding table for complexes **1-5**.<sup>#</sup>

D—H···A		D—H	H···A	D—A	∠D—H···A
<b>[Zn<sub>2</sub>(H<sub>2</sub>O)<sub>2</sub>(HBTC)<sub>2</sub>(BIm)<sub>2</sub>](H<sub>2</sub>O)<sub>3</sub>, <b>1</b></b>					
O6—H6A···O3W	1/2+x,-1/2+y,z	0.82	1.80(2)	2.6057(17)	170
N2—H2···O5	-1/2+x,1/2-y,-1/2+z	0.86	2.03	2.8233(14)	152
O1W—H1W1···O4	1-x,y,1/2-z	0.82(3)	1.93(2)	2.7254(13)	163
O1W—H2W1···O2	1/2-x,3/2-y,-z	0.81(3)	1.85(3)	2.6459(15)	169
O2W—H1W2···O3	-1/2+x,-1/2+y,z	0.84(2)	1.98(3)	2.8064(13)	170
O3W—H1W3···O2	1/2-x,3/2-y,-z	0.78(2)	2.16(2)	2.8517(18)	148
O3W—H2W3···O2W		0.77(3)	2.15(2)	2.8876(16)	163
<b>[2-HABI]<sub>2</sub>[Zn<sub>2</sub>(BTC)<sub>2</sub>(2-ABI)(H<sub>2</sub>O)].(H<sub>2</sub>O)<sub>5</sub>, <b>2</b></b>					
O1W—H1WA···O4W	x,y,-1+z	0.88	2.05	2.669(8)	127
N2—H2···O5W	-1+x,y,z	0.86	1.91	2.749(5)	163
O1W—H1WB···O12	-x,-y,1-z	0.88	1.87	2.734(3)	168
O6W—H6WA		0.85			
N3—H3A···O7		0.86	2.12	2.893(4)	149
N3—H3B···O4W	-1+x,y,z	0.86	2.10	2.934(7)	163
O6W—H6WB		0.85			
N4—H4A···O2	1+x,y,z	0.86	1.84	2.683(4)	167
N5—H5···O3	1-x,-y,1-z	0.86	2.14	2.915(3)	149
N5—H5···O10	1-x,-y,-z	0.86	2.14	2.915(3)	149
O5W—H5WA		0.85			



O5W—H5WB···O1	$x, y, -1+z$	0.85	2.02	2.858(4)	167
O5W—H5WB···O6	$1-x, 1-y, 1-z$	0.85	2.58	3.056(4)	116
N6—H6A···O10	$1-x, -y, 1-z$	0.86	1.99	2.760(3)	148
N6—H6B···O6W	$1+x, y, z$	0.86	2.09	2.875(6)	153
N7—H7···O4	$2-x, 1-y, 1-z$	0.86	1.93	2.774(4)	167
O2W—H2WA···O3W	$x, y, -1+z$	0.85	1.95	2.757(10)	159
N8—H8···O5	$x, y, -1+z$	0.86	2.02	2.766(4)	144
O2W—H2WB···O9	$x, y, -1+z$	0.85	2.15	2.983(5)	167
O4W—H4WA···O8	$x, y, -1+z$	0.85	2.32	2.701(6)	107
N9—H9A···O5	$x, y, -1+z$	0.86	2.39	3.067(4)	136
N9—H9B···O2W	$2-x, 1-y, -z$	0.86	1.99	2.850(6)	176
O4W—H4WB···O8	$x, y, -1+z$	0.85	2.31	2.701(6)	108
O3W—H3WA···O5	$1-x, 1-y, 1-z$	0.85	2.47	3.316(9)	177
O3W—H3WB		0.85			
<b>[Cu<sub>3</sub>(BTC)<sub>2</sub>(BIm)<sub>6</sub>][solvent], 3</b>					
N2—H2···O2	$y, -x+y, -z$	0.86	1.88	2.729(5)	170
<b>[Co<sub>4</sub>(BTC)<sub>3</sub>(BIm)<sub>6</sub>][solvent], 4</b>					
N2—H2A···O9	$1/2+x, 1/2-y, 1/2+z$	0.86	1.93	2.762(5)	161
N4—H4A···O2	$1/2-x, -1/2-y, -z$	0.86	1.83	2.687(5)	175
N6—H6A···O9	$x, -1+y, z$	0.86	2.04	2.851(5)	158
<b>[(CH<sub>3</sub>CH<sub>2</sub>)<sub>2</sub>NH<sub>2</sub>][Zn(BTC)(BIm)].(H<sub>2</sub>O), 5</b>					
N1A—H1···O2	$1/2+x, 1/2-y, z$	0.89	1.84	2.726(7)	170

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N1A—H2···O1W	$1/2-x, 1/2+y, 1/2+z$	0.89	1.90	2.785(9)	171	
N2—H2A···O4	$1/2-x, 1/2+y, -1/2+z$	0.86	1.89	2.700(4)	157	
O1W—H1W···O3		0.87	2.10	2.904(6)	155	
O1W—H2W···O5	$-x, -y, -1/2+z$	0.87	1.90	2.754(6)	168	

#Where 'D' is a donor and 'A' is acceptor, the bond lengths are in (Å) and angles are in (°).