

## Supporting information

# Flexible All-solid-state Ultrahigh-energy Asymmetric Supercapacitors Based on Tailored Morphology of NiCoO<sub>2</sub>/Ni(OH)<sub>2</sub>/Co(OH)<sub>2</sub> Electrodes

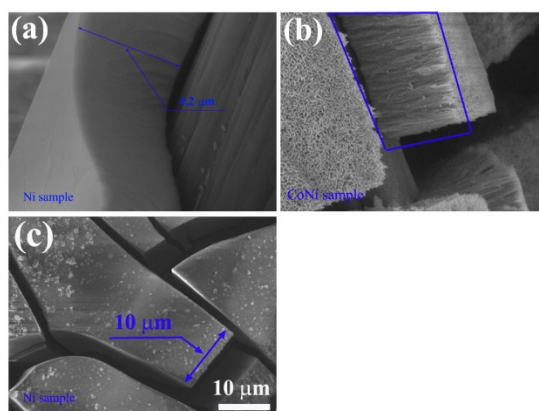
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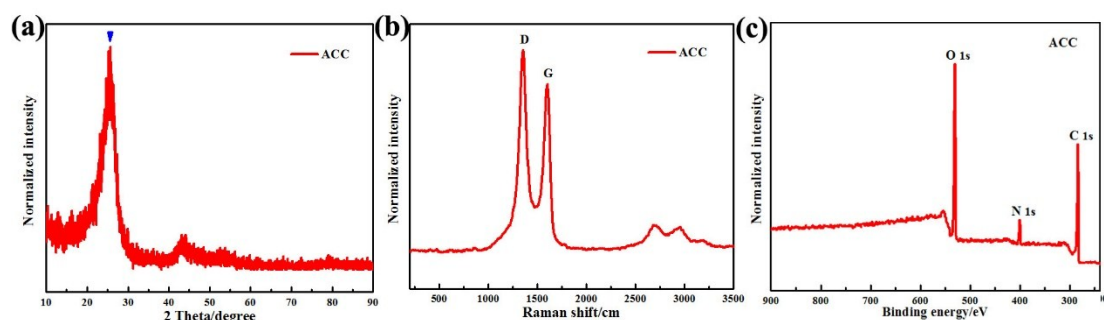
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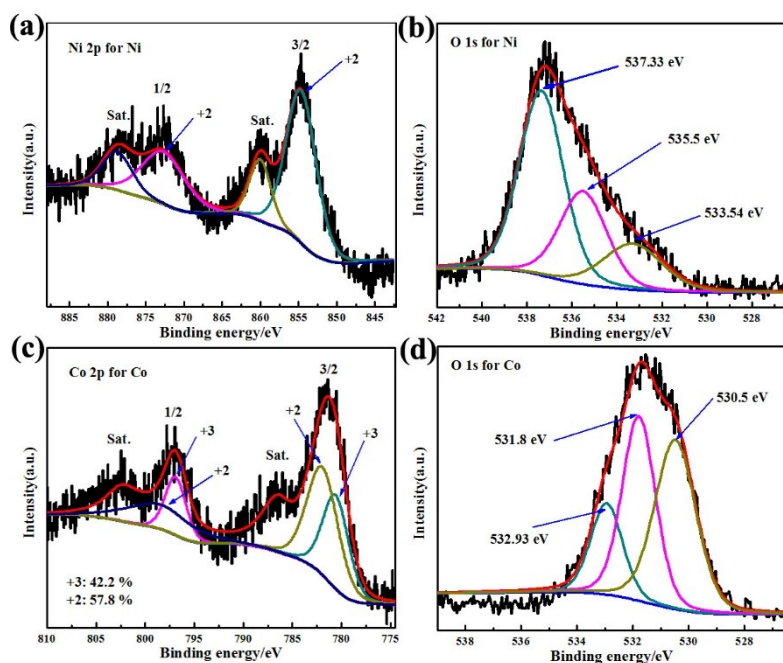
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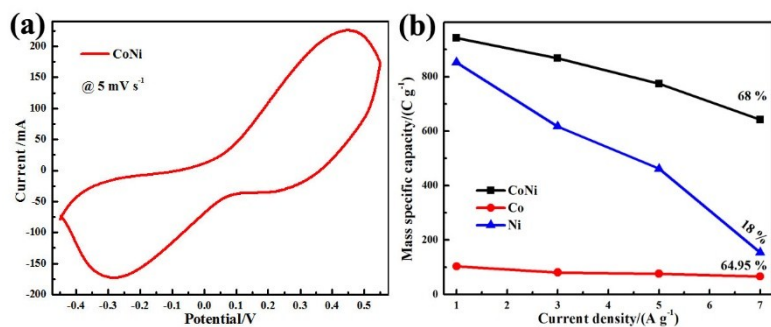
**Fig. S1** (a) High and (c) low-magnification SEM images for Ni sample. (c) Enlarged-view SEM image of CoNi sample.



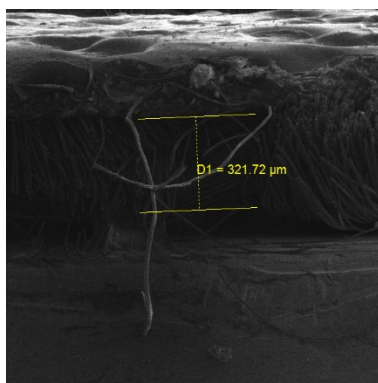
**Fig. S2** (a) XRD pattern, (b) Raman spectrum and (c) XPS curve of active carbon cloth (ACC).



**Fig. S3** (a) Ni 2p core-level spectrum and (b) O 1s core-level spectrum of sample Ni. (c) Co 2p core-level spectrum and (d) O 1s core-level spectrum of sample Co.



**Fig. S4** (a) CV curve of sample CoNi at  $5 \text{ mV s}^{-1}$ . (b) Mass capacitance as a function of current density of different samples. Among various electrodes, CoNi electrode shows the best rate ability of  $\sim 68\%$  compared to  $18\%$  of Ni electrode and  $\sim 65\%$  of Co electrode.



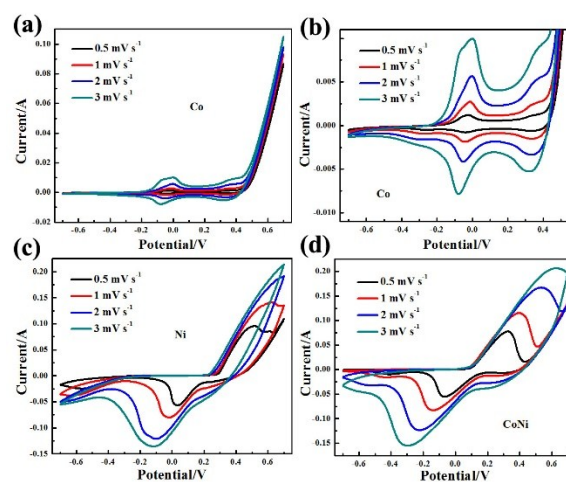
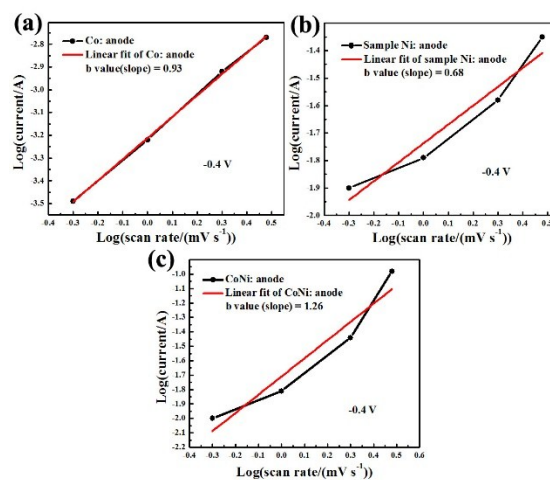
**Fig. S5** Thickness of the as-prepared electrodes.

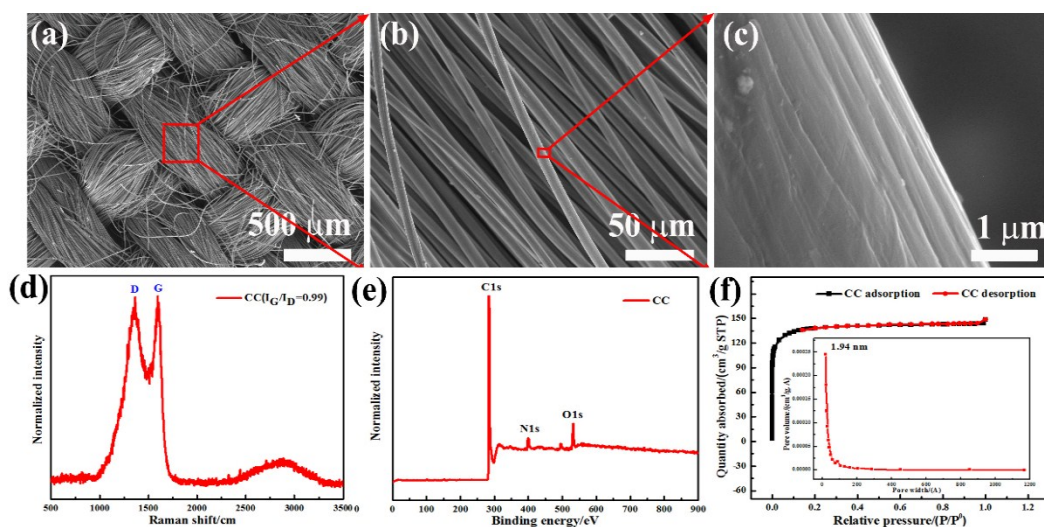
**Table S1** Comparison of volumetric capacitance with reported literatures

Electrode materials	Volumetric capacitance/ (F cm <sup>-3</sup> )	Reference
Mxene/CNT paper	435	1
Graphene/porous carbon	376	2
T <sub>3</sub> C <sub>2</sub> Tx paper	442	3
MoS <sub>2</sub> nanosheet	700	4
NiCoO <sub>2</sub> /Ni(OH) <sub>2</sub> /Co(OH) <sub>2</sub>	700	This work

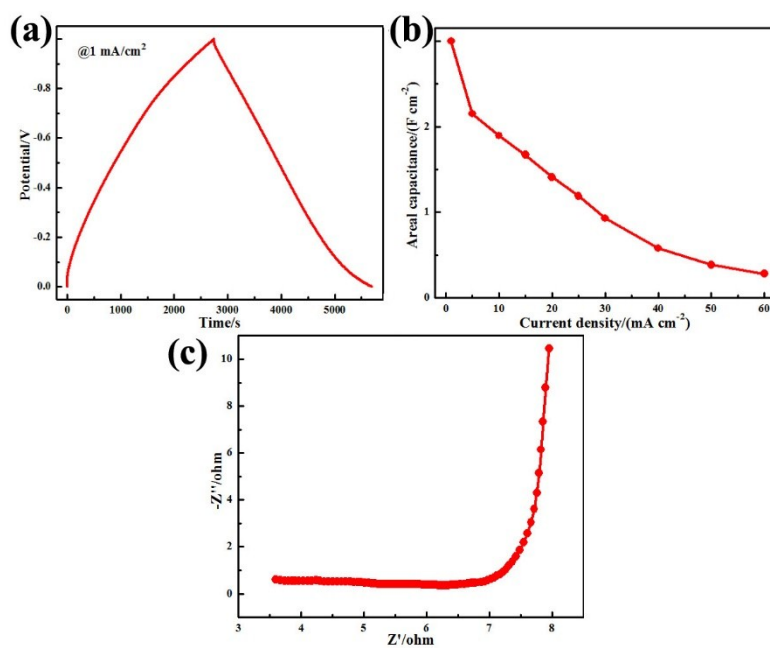
**Table S2** Resistances of different samples from EIS curves

	R <sub>SR</sub> /Ω	R <sub>CT</sub> /Ω	Angle at low frequency/°
Co sample	0.78	0.33	86
Ni sample	0.81	7.53	22
CoNi sample	1.02	0.32	66

**Fig. S6** CV curves of (a) and (b) sample Co, (c) sample Ni and (d) sample CoNi.**Fig.S7** The linear dependence of log current on log sweep rate (based on Equation (10)) for (a) sample Co, (b) sample Ni and (c) sample CoNi at -0.4 V during the correspondingly anodic CV processes.



**Fig. S6** (a-c) SEM images of carbon cloth with large surface area (CC). The corresponding (d) Raman spectrum, (e) XPS full pattern and (f) Nitrogen adsorption/desorption isotherm curve, inset: pore size distribution spectrum.



**Fig. S7** (a) GCD curve, (b) rate ability and (c) Nyquist plots of CC as negative electrode.

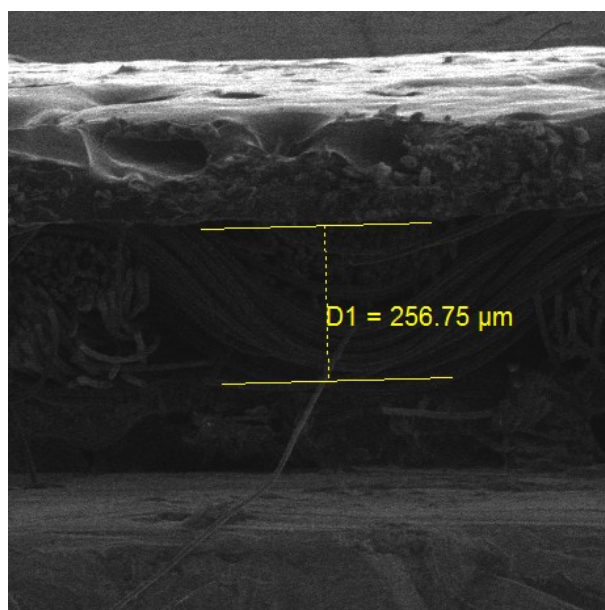


Figure S8. Thickness of CC negative electrode.

**Table S3** Volumetric energy density vs the reported value in literatures

Supercapacitor	Energy density/(mWh cm <sup>-3</sup> )	Reference
CoNi//CC using 6 M KOH	37.41	This work
CoNi//CC using KOH/PVA	11.98	This work
MnO <sub>2</sub> MN//Co <sub>3</sub> O <sub>4</sub>	27	5
MnO <sub>2</sub> /G//NF	0.27	6
MnO <sub>2</sub> //Fe <sub>2</sub> O <sub>3</sub>	0.41	7
H-TiO <sub>2</sub> @MnO <sub>2</sub> //H-TiO <sub>2</sub> @C	0.3	8
MnO <sub>2</sub> /G//VOS@C	0.87	9
MnO <sub>2</sub> @ZnO//rGO	0.234	10
ACC//ACC	9.4	11

## Reference

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