Supplementary Information

Rapid Naked-Eye Luminescence Detection of Carbonate Ion through Acetonitrile Hydrolysis Induced Europium Complexes

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Fig. S1 PXRD patterns of the simulated and as-synthesized [Ln₂(Hhpip)₂(OAc)₆].



Fig. S2 PXRD patterns of the simulated and as-synthesized [Tb(hnpip)(OAc)₂]_n.



Fig. S3 Mass spectrum for the DMSO solution of [Eu₂(Hhpip)₂(OAc)₆].



Fig. S4 Relative emission intensity for $[Eu_2(Hhpip)_2(OAc)_6]$ at 542 nm as a function of carbonate concentration.



Fig. S5 Relative emission intensity for $[Eu_2(Hhpip)_2(OAc)_6]$ at 469 nm as a function of carbonate concentration.



Fig. S6 Relative emission intensity for $[Eu_2(pip)_2(OAc)_6]$ at 514 nm as a function of carbonate concentration.



Fig. S7 Relative emission intensity for $[Eu_2(pip)_2(OAc)_6]$ at 432 nm as a function of carbonate concentration.

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carbonate concentration.				
Complex	Wavelength	Standard	Slope (Sensitivity)	Limit of detection
	monitored /nm	deviation	$/mM^{-1}$	$/\mu M$
[Eu ₂ (Hhpip) ₂ (OAc) ₆]	469	0.01023	2.4952	12.3
	542	0.02389	9.142	7.8
[Eu ₂ (pip) ₂ (OAc) ₆]	432	0.06315	16.849	11.2

0.41437

148.062

8.4

514

Table S1 Limits of detection towards carbonate ion for the Eu(III) complexes at specific wavelengths based on linear fit statics of the relative luminescence intensity as a function of low carbonate concentration.