

## Supplementary information

### K modified P2- $\text{Na}_{0.7}\text{Mn}_{0.8}\text{Mg}_{0.2}\text{O}_2$ as a cathode material for sodium-ion batteries

Divya Sehrawat,<sup>a</sup> Soshan Cheong,<sup>b</sup> Aditya Rawal,<sup>b</sup> Alexey M. Glushenkov,<sup>c</sup> Helen E. A. Brand,<sup>d</sup> Bruce Cowie,<sup>d</sup> Elena Gonzalo,<sup>e</sup> Teófilo Rojo,<sup>e,f</sup> Pierre J. P. Naeyaert,<sup>g</sup> Chris D. Ling,<sup>g</sup> Maxim Avdeev<sup>h</sup> and Neeraj Sharma<sup>a,\*</sup>

<sup>a</sup>School of Chemistry, UNSW Sydney, Sydney, NSW 2052, Australia.

<sup>b</sup>Mark Wainwright Analytical Centre, UNSW Sydney, Sydney, NSW 2052, Australia.

<sup>c</sup>Department of Chemical Engineering, The University of Melbourne, Melbourne, VIC 3010, Australia.

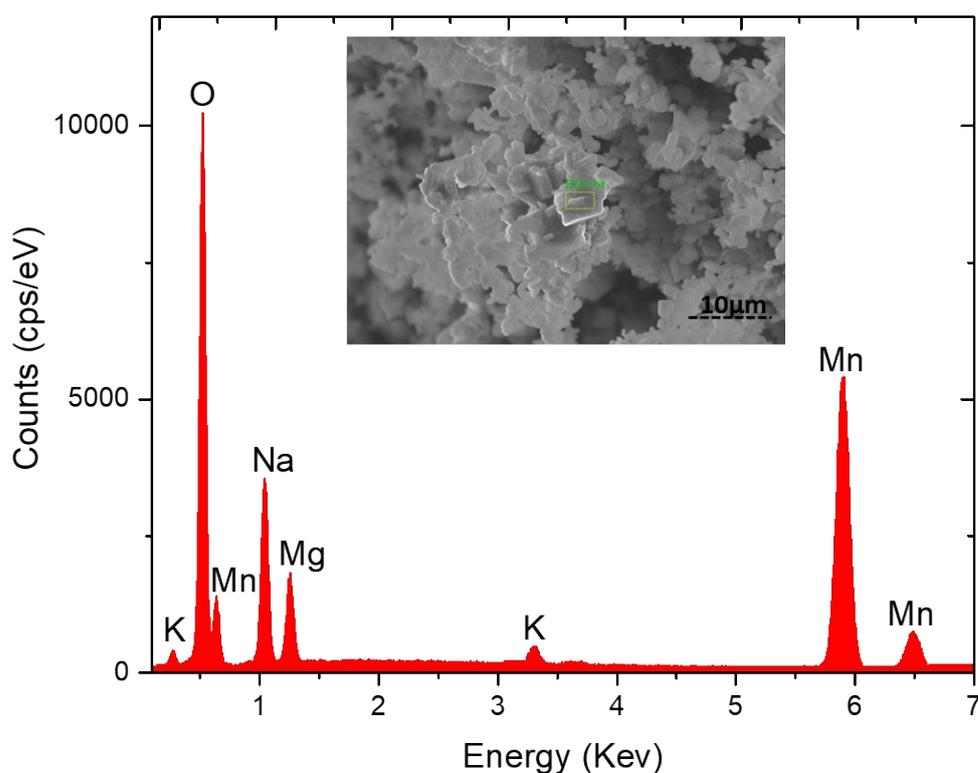
<sup>d</sup>Australian Synchrotron, 800 Blackburn Road, Clayton VIC 3168, Australia.

<sup>e</sup>CIENERGIGUNE, Parque Tecnológico de Álava, Albert Einstein 48, ED.CIC, 01510, Miñano, Spain

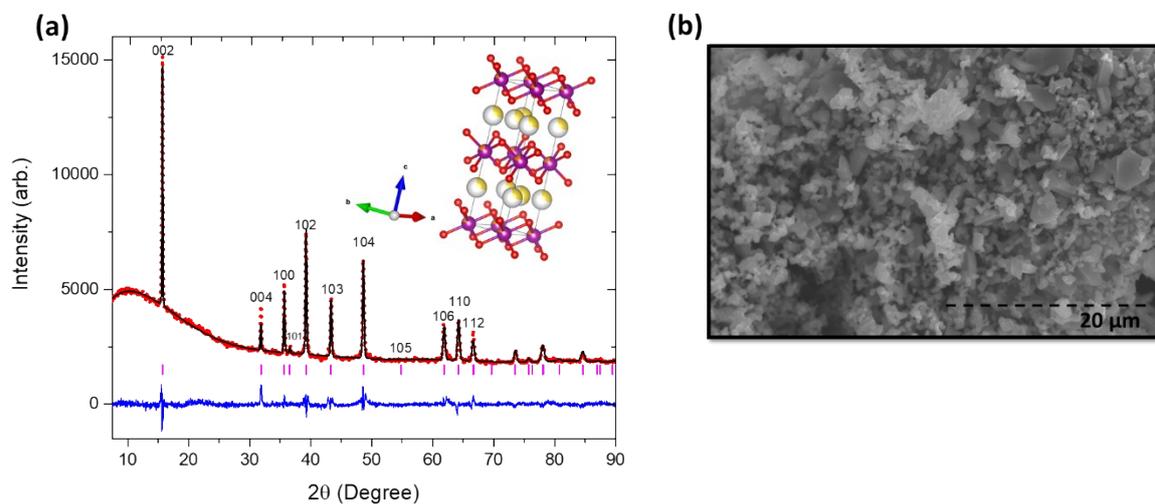
<sup>f</sup>Departamento de Química Inorgánica, Universidad del País Vasco UPV/EHU, P.O. Box. 644, 48080, Bilbao, Spain

<sup>g</sup>School of Chemistry, The University of Sydney, Sydney 2006, Australia

<sup>h</sup>Australian Nuclear Science and Technology Organisation, Locked Bag 2001, Kirrawee DC NSW 2232, Australia



**Fig. S1** EDS spectra on a selected area (rectangular region) of K-modified P2  $\text{Na}_{0.7}\text{Mn}_{0.8}\text{Mg}_{0.2}\text{O}_2$ . SEM image is shown, on which EDX spectra was measured.

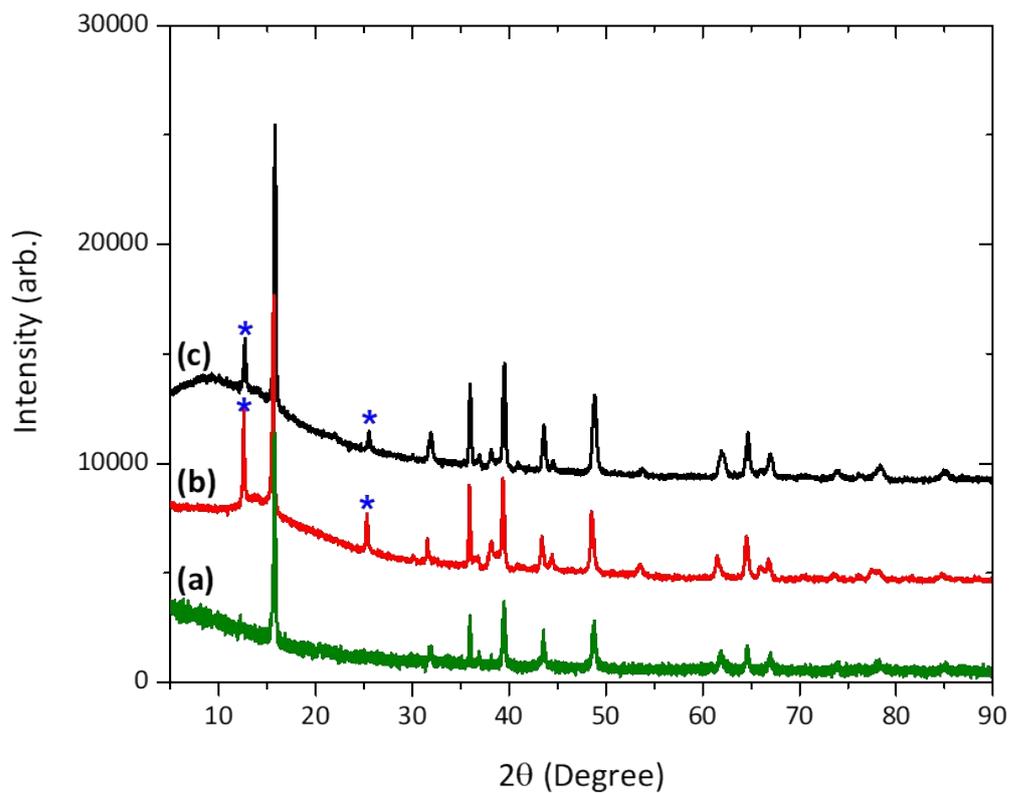


**Fig. S2 (a)** Rietveld refined fit of the  $\text{Na}_{0.7}\text{Mn}_{0.8}\text{Mg}_{0.2}\text{O}_2$  structural model to the XRD data. Data are shown as red dots, the calculated Rietveld model as a line through the data, and the difference between the data and the model as the line below the data. The crystal structure of the materials with Mn in purple, O in red, Mg in orange, Na indicated by the amount of shading in yellow. **(b)** SEM images of P2  $\text{Na}_{0.7}\text{Mn}_{0.8}\text{Mg}_{0.2}\text{O}_2$  at 20  $\mu\text{m}$  magnification.

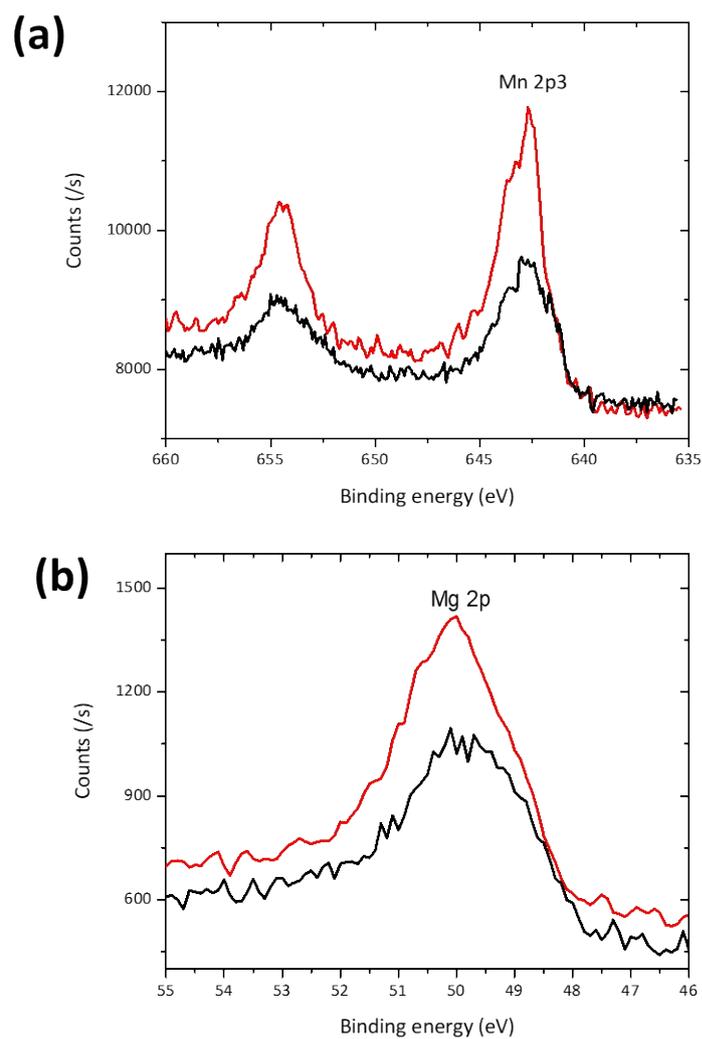
**Table S1.** Crystallographic details of P2  $\text{Na}_{0.7}\text{Mn}_{0.8}\text{Mg}_{0.2}\text{O}_2$ .

Name	$x (a)$	$y (b)$	$z (c)$	ADP ( $100 \text{ \AA}^2$ )	SOF#
Mn	0	0	0	1.2(1)	0.8*
Mg	0	0	0	1.2(1)	0.2*
O	0.3333	0.6666	0.0931(2)	3.1(2)	1
$\text{Na}_f$	0.3333	0.6666	0.75	1.5^	0.4
$\text{Na}_e$	0	0	0.25	6.8^	0.3

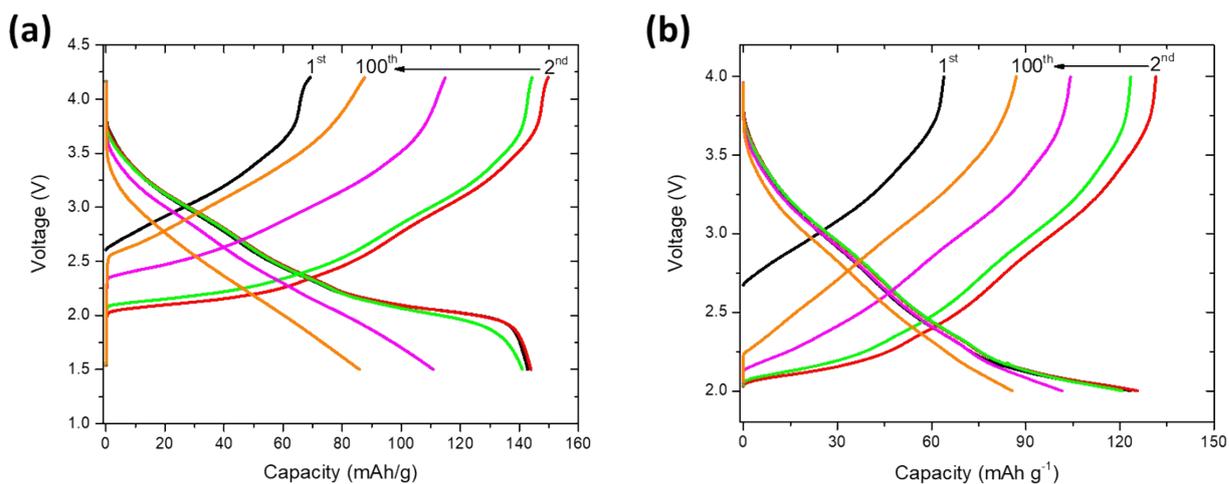
ADP = atomic displacement parameter, SOF = site occupation factors, # = fixed, ^ refined initially then fixed, \* constrained to be equal, nominal composition  $\text{Na}_{0.7}\text{Mn}_{0.8}\text{Mg}_{0.2}\text{O}_2$ , hexagonal  $P6_3/mmc$  symmetry with lattice parameters  $a = 2.8918(2) \text{ \AA}$  and  $c = 11.181(1) \text{ \AA}$ ,  $\chi^2 = 1.05$ ,  $wR_p = 3.58\%$ , 23 refinement parameters



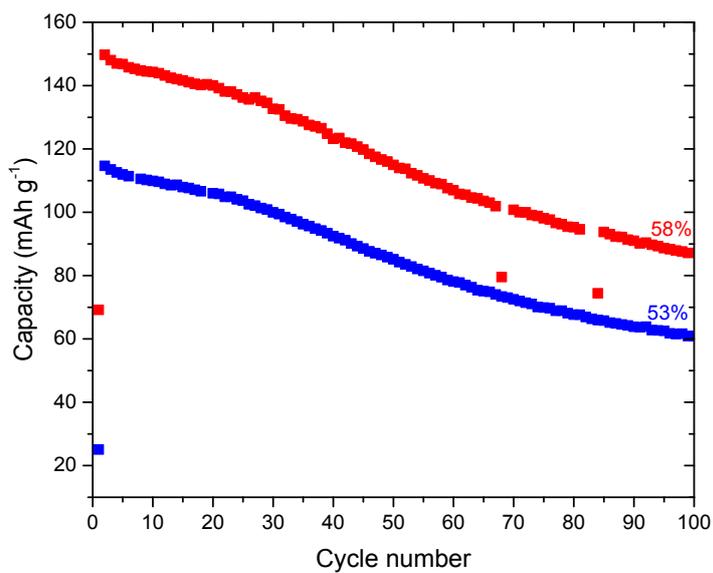
**Fig. S3** XRD pattern of **(a)** sample quenched at 900 °C after heating for 12 hours and stored in Ar-filled glove box, **(b)** sample left outside the glove box for 2 days after quenching and **(c)** sample slow cooled to room temperature, after heating at 900 °C for 12 hours. Blue asterisks around ~12.4 and ~25.3° correspond to the (002) and (004) reflections of the hydrated P2 phase.



**Fig. S4** XPS spectra of the  $\text{Na}_{0.7}\text{Mn}_{0.8}\text{Mg}_{0.2}\text{O}_2$  (black) and K modified  $\text{Na}_{0.7}\text{Mn}_{0.8}\text{Mg}_{0.2}\text{O}_2$  (red) electrode in the (a) Mn 2p<sup>3</sup> and (b) Mg 2p regions.



**Fig. S5** Charge-discharge curves of P2  $\text{Na}_{0.7}\text{Mn}_{0.8}\text{Mg}_{0.2}\text{O}_2$  **(a)** between 1.5-4.2 V and **(b)** between 2-4 V at a current density of  $15 \text{ mA g}^{-1}$ . Black shows 1<sup>st</sup>, red 2<sup>nd</sup>, green 10<sup>th</sup>, pink 50<sup>th</sup> and orange 100<sup>th</sup> cycle.



**Fig. S6** Comparison of capacity retention curve of P2  $\text{Na}_{0.7}\text{Mn}_{0.8}\text{Mg}_{0.2}\text{O}_2$  (red square) and K-modified  $\text{Na}_{0.7}\text{Mn}_{0.8}\text{Mg}_{0.2}\text{O}_2$  (blue square) between 1.5- 4.2 V at a current density of  $15 \text{ mA g}^{-1}$ .

**Table S2.** ICP results of K-modified P2 Na<sub>0.7</sub>Mn<sub>0.8</sub>Mg<sub>0.2</sub>O<sub>2</sub>.

Ratio	Expected	As found in synthesised powder	On separator after 1 <sup>st</sup> charge to 4.2 V	On separator after 1 <sup>st</sup> discharge to 1.5 V
K: Na	0.1428	0.10	0.066	0.048