

Supplementary Information

for the article

Magnetostructural relationships in polymorphic ethylmalonate-containing copper(II) coordination polymers

Jorge Pasán,^{*a} Joaquín Sanchiz,^b Francesc Lloret,^c Miguel Julve^c and Catalina Ruiz-Pérez^a

^a. Laboratorio de Rayos X y Materiales Moleculares (MATMOL), Departamento de Física, Facultad de Ciencias, Universidad de La Laguna, Avda. Astrofísico Francisco Sánchez s/n, 38200, La Laguna, Spain. E-mail:
jpasang@ull.edu.es

^b. Departamento de Química, Facultad de Ciencias, Universidad de La Laguna, Avda. Astrofísico Francisco Sánchez s/n, 38200, La Laguna, Spain.

^c. Departamento de Química Inorgánica/Instituto de Ciencia Molecular (ICMol), Universitat de València, C/Catedrático José Beltrán 2, 46980 Paterna, València, Spain.

Table S1. Crystallographic data for complexes **1-3**

	1	2	3
Formula	C ₁₀ H ₂₂ O ₁₃ Cu ₂	C ₅ H ₈ O ₅ Cu	C ₅ H ₆ O ₆ Cu
Fw	477.35	211.64	225.64
Crystal system	Orthorhombic	Orthorhombic	Hexagonal
Space group	<i>C222</i> ₁	<i>C2cb</i>	<i>P6</i> ₅
<i>a</i> (Å)	9.9885(2)	9.770(3)	11.5965(8)
<i>b</i> (Å)	11.9939(2)	9.921(3)	-
<i>c</i> (Å)	14.1518(3)	14.477(3)	10.5437(6)
<i>V</i> (Å ³)	1695.40(6)	1403.2(7)	1227.94(14)
<i>Z</i>	4	8	6
<i>T</i> (K)	293(2)	293(2)	293(2)
ρ_{calc} (g·cm ⁻³)	1.870	1.985	1.831
μ (Mo K α , mm ⁻¹)	2.574	3.081	2.656
λ (Mo K α , Å)	0.71073	0.71073	0.71073
Index ranges	-14 < <i>h</i> < 14 -17 < <i>k</i> < 17 -20 < <i>l</i> < 20	-12 < <i>h</i> < 10 -10 < <i>k</i> < 12 -18 < <i>k</i> < 15	-15 < <i>h</i> < 13 -15 < <i>k</i> < 15 -13 < <i>l</i> < 9
Indep. Reflections (R_{int})	2801 (0.0286)	1408 (0.0653)	1916 (0.0440)
Obs. Reflections [$I > 2\sigma(I)$]	2706	1058	1813
Flack Parameter	0.005(13)	0.52(4)	0.03(4)
Parameters	135	108	120
Goodness of fit	1.053	1.030	1.159
<i>R</i> [$I > 2\sigma(I)$]	0.0213	0.0458	0.0509
<i>R_w</i> [$I > 2\sigma(I)$]	0.0489	0.0894	0.1333
<i>R</i> (all data)	0.0225	0.0759	0.0558
<i>R_w</i> (all data)	0.0494	0.0980	0.1356

Table S2. Selected bond lengths (\AA) and angles ($^\circ$) for complexes **1-3^a**

1			
Cu(1)–O(2)	2.0209(15)	Cu(2)–O(2w)	1.9727(17)
Cu(1)–O(4)	1.9477(14)	Cu(2)–O(3w)	1.9700(17)
Cu(1)–O(1w)	1.999(2)	Cu(2)–O(1)	2.3936(14)
O(2)–Cu(1)–O(4)	90.87(6)	O(2w)–Cu(2)–O(3w)	174.35(8)
O(2)–Cu(1)–O(2a)	115.07(10)	O(2w)–Cu(2)–O(1)	88.43(7)
O(2)–Cu(1)–O(1w)	122.46(5)	O(2w)–Cu(2)–O(3wb)	92.55(8)
O(2)–Cu(1)–O(4a)	92.35(7)	O(2w)–Cu(2)–O(1b)	91.04(7)
O(4)–Cu(1)–O(1w)	87.00(5)	O(2w)–Cu(2)–O(2wb)	84.65(11)
O(4)–Cu(1)–O(4a)	174.01(10)	O(3w)–Cu(2)–O(1)	96.54(7)
		O(3w)–Cu(2)–O(1b)	83.96(7)
2			
Cu(1)–O(2)	1.947(5)	O(2)–Cu(1)–O(1w)	96.5(3)
Cu(1)–O(4)	1.956(6)	O(4)–Cu(1)–O(1c)	86.3(3)
Cu(1)–O(1c)	1.963(6)	O(4)–Cu(1)–O(3d)	85.2(3)
Cu(1)–O(3d)	1.972(6)	O(4)–Cu(1)–O(1w)	94.0(3)
Cu(1)–O(1w)	2.264(7)	O(1c)–Cu(1)–O(3d)	92.7(3)
O(2)–Cu(1)–O(4)	89.8(3)	O(1c)–Cu(1)–O(1w)	102.0(3)
O(2)–Cu(1)–O(1c)	161.3(3)	O(3d)–Cu(1)–O(1w)	104.2(2)
O(2)–Cu(1)–O(3d)	85.2(3)		
3			
Cu(1)–O(1e)	1.977(5)	O(1e)–Cu(1)–O(1w)	92.1(3)
Cu(1)–O(2)	1.974(5)	O(2)–Cu(1)–O(3f)	89.2(2)
Cu(1)–O(3f)	1.948(5)	O(2)–Cu(1)–O(4)	90.9(2)
Cu(1)–O(4)	1.954(5)	O(2)–Cu(1)–O(1w)	105.7(3)
Cu(1)–O(1w)	2.313(8)	O(3f)–Cu(1)–O(4)	179.8(3)
O(1e)–Cu(1)–O(2)	161.9(2)	O(3f)–Cu(1)–O(1w)	95.5(2)
O(1e)–Cu(1)–O(3f)	92.1(2)	O(4)–Cu(1)–O(1w)	84.6(3)
O(1e)–Cu(1)–O(4)	87.7(2)		

^aSymmetry codes: (a) = $-x+1, y, -z+3/2$; (b) = $x, -y+1, -z+1$; (c) = $x, y-1/2, -z+1/2$; (d) = $x+1/2, y, -z+1/2$; (e) = $-y+1, x-y+1, z-1/3$; (f) = $y-1, -x+y, z+1/6$.

Table S3. Relevant hydrogen bonds for complexes **1-2^a**

1			
<i>H-bond</i>	<i>D···A / Å</i>	<i>H···A / Å</i>	<i>D–H···A / °</i>
O(1W)–H···O(1g)	2.681(2)	2.06(2)	168(3)
O(2W)–H···O(3h)	2.668(2)	1.93(3)	172(4)
O(2W)–H···O(2b)	2.664(2)	1.97(3)	161(3)
O(3W)–H···O(3i)	2.707(2)	2.09(3)	163(3)
O(3W)–H···O(4j)	2.789(2)	2.02(3)	170(3)
C(2)–H···O(1Wk)	3.484(2)	2.69(3)	132(2)
C(4)–H···O(2a)	3.334(2)	2.36(3)	161.5(18)
2			
<i>H-bond</i>	<i>D···A / Å</i>	<i>H···A / Å</i>	<i>D–H···A / °</i>
O(1W)–H···O(2c)	2.748(8)	1.96(8)	134(3)
O(1W)–H···O(1d)	2.762(7)	1.90(9)	142(6)
C(2)–H···O(1W)	3.278(9)	2.49(8)	138(3)

^a Symmetry codes: (a) = -x+2, *y*, -z+3/2; (b) = *x*, -y+1, -z+2; (c) = *x*, *y*-1/2, -z+3/2; (d) = *x*-1/2, *y*, -z+3/2; (g) = -x+3/2, *y*-1/2, -z+3/2; (h) = -x+3/2, *y*+1/2, -z+3/2; (i) = -x+1, *y*, -z+3/2; (j) = -x+3/2, -y+1/2, *z*+1/2; (k) = *x*-1/2, *y*+1/2, *z*.

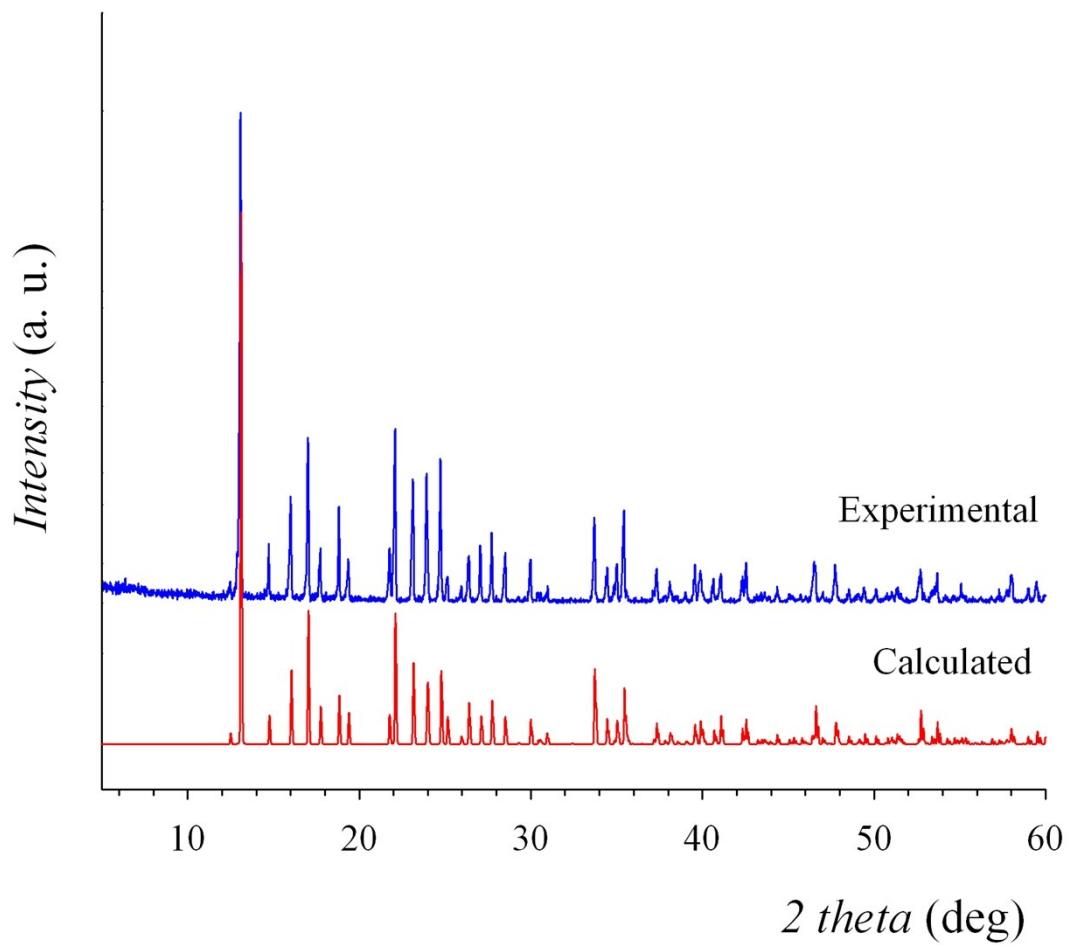


Figure S1. Powder X-ray diffraction pattern for compound **1**.

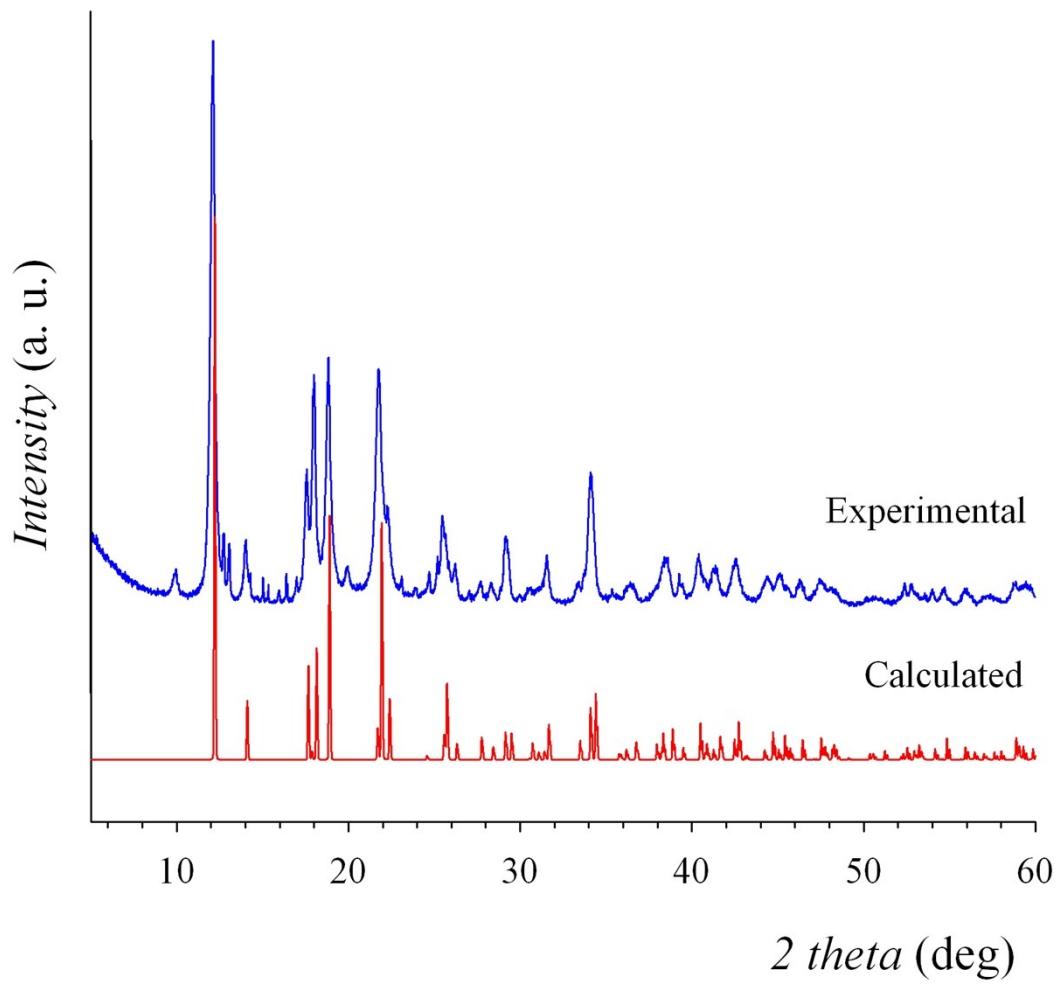


Figure S2. Powder X-ray diffraction pattern for compound 2.

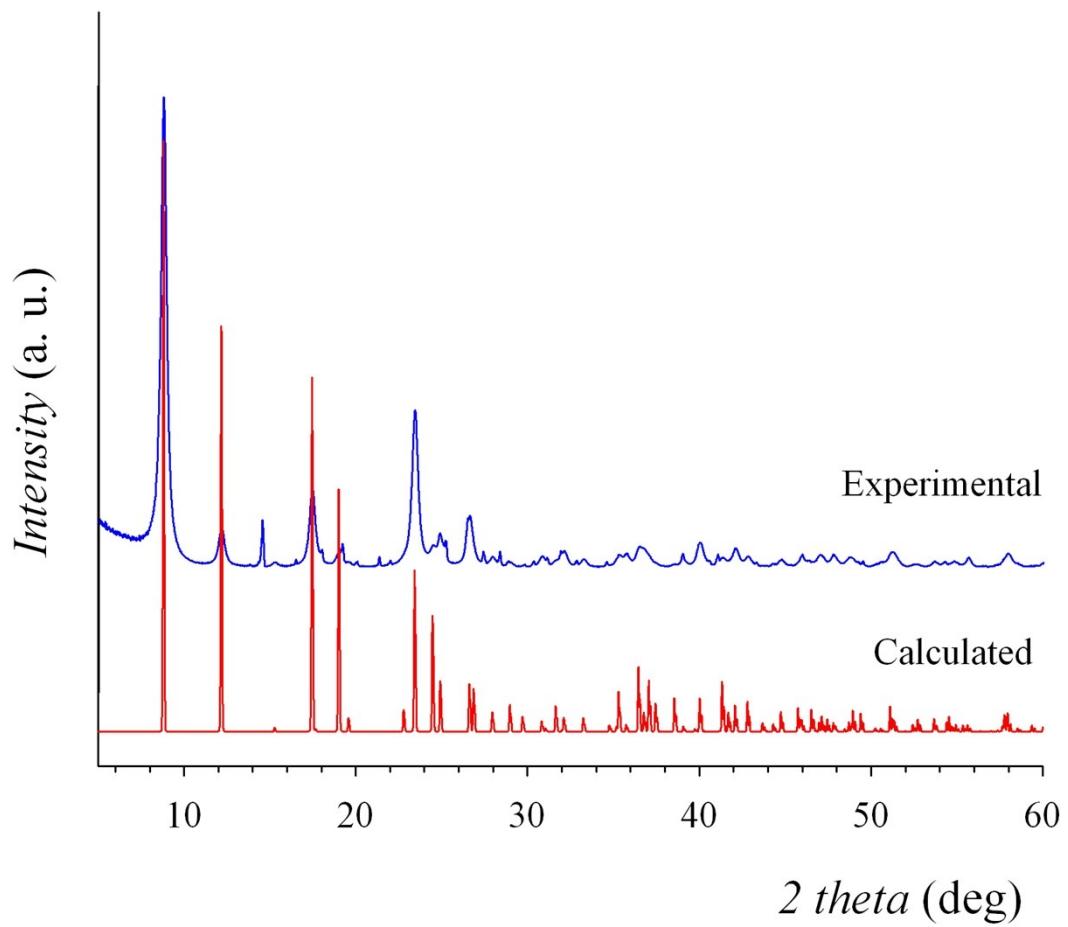


Figure S3. Powder X-ray diffraction pattern for compound 3.

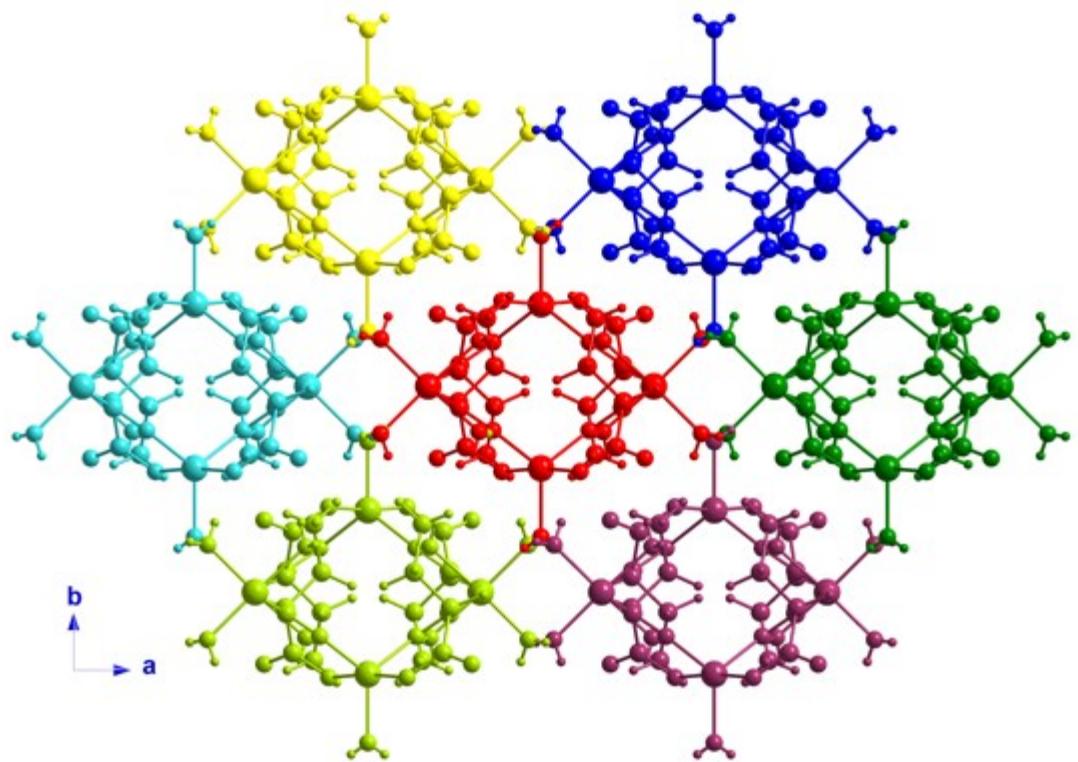


Figure S4. View along the c crystallographic axis of the crystal packing of the chain compound **1**. Each $\{[\text{Cu}(\text{H}_2\text{O})_4][\text{Cu}(\text{Etmal})_2(\text{H}_2\text{O})]\}_n$ chain is depicted in a different colour.

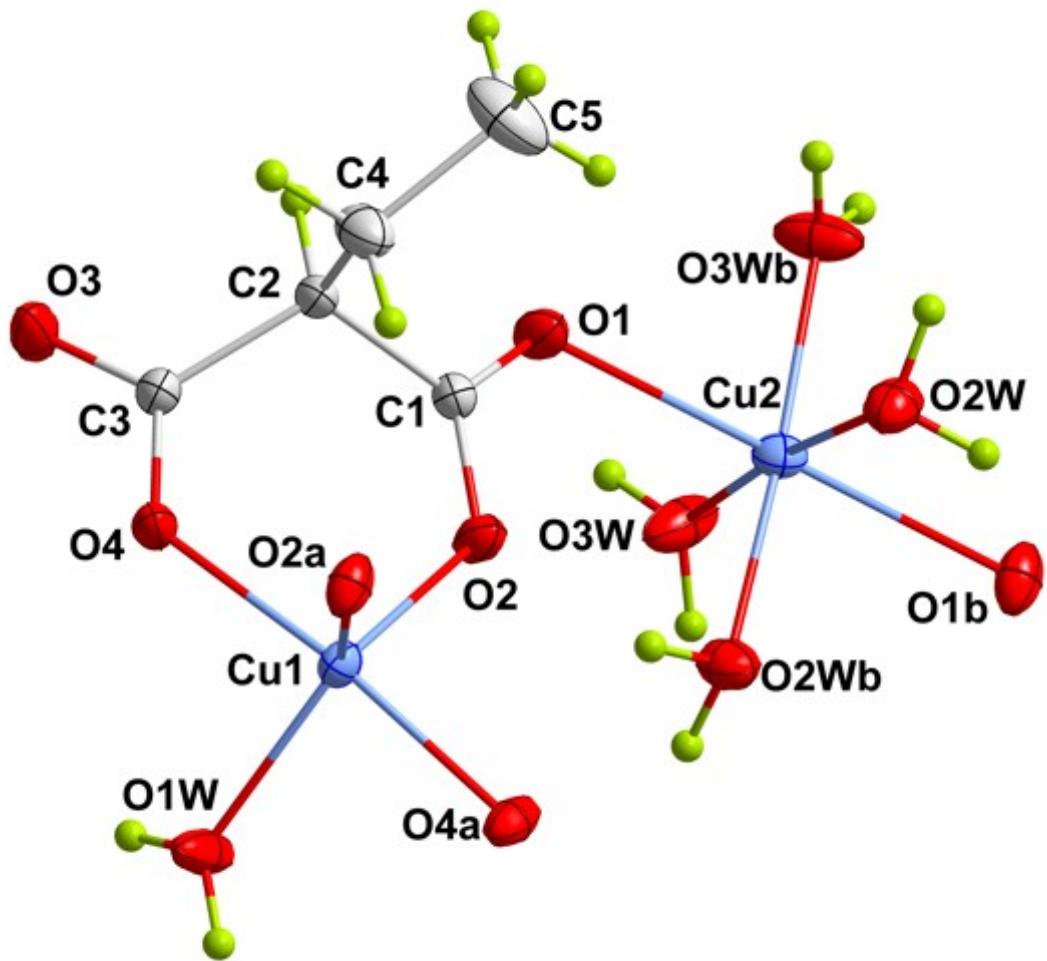


Figure S5. Fragment of the crystal structure of **1** with the atom numbering. The ellipsoids are presented in 50% probability.

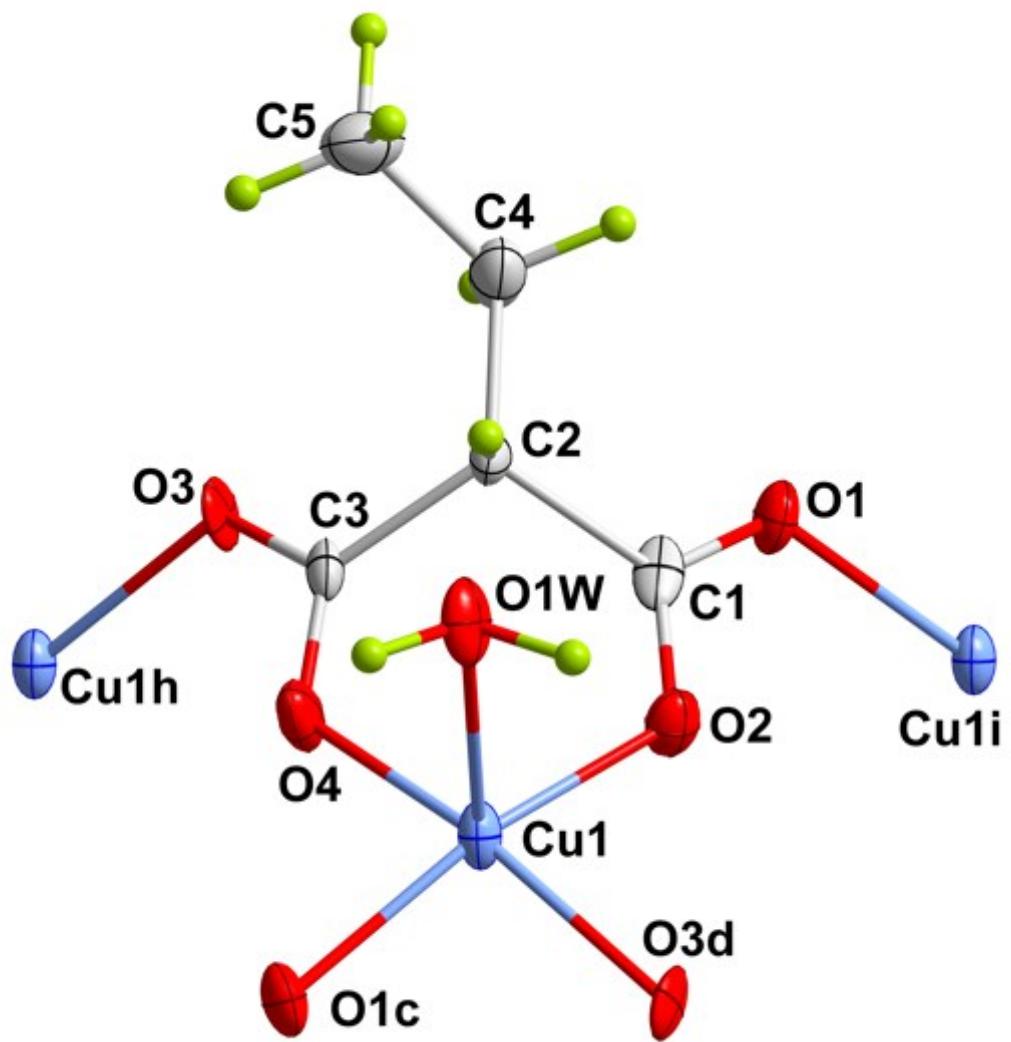


Figure S6. Fragment of the crystal structure of **2** with the atom numbering. Ellipsoids are represented in 50% probability.

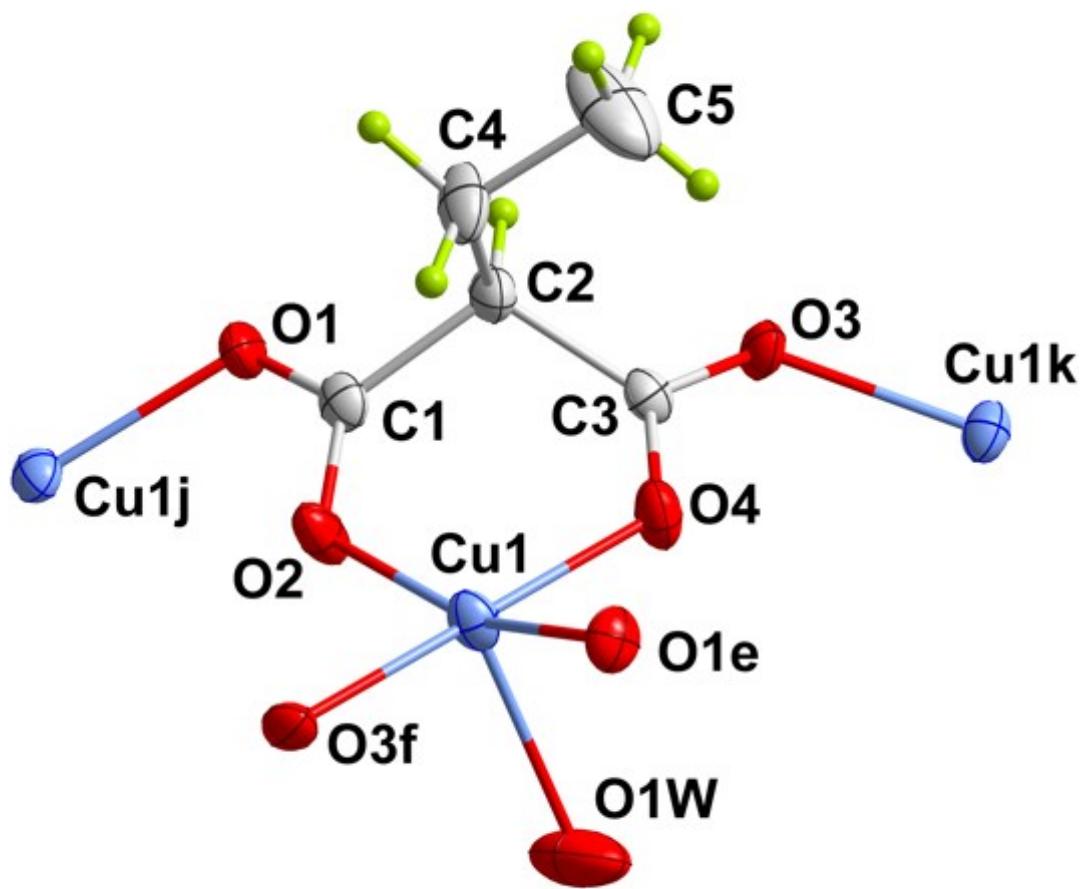
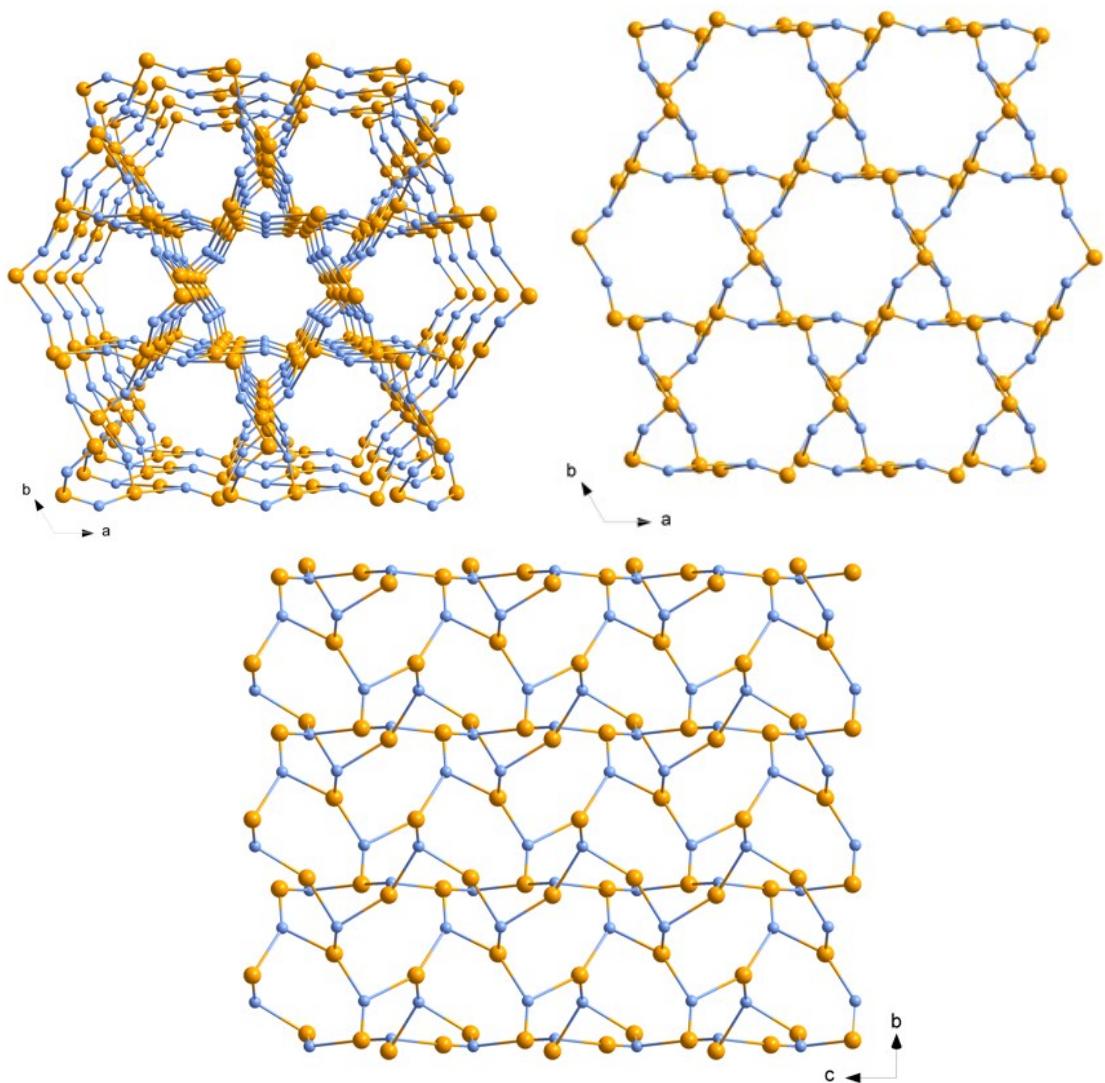


Figure S7. Fragment of the crystal structure of **3** with the atom numbering. Ellipsoids are represented in 50% probability.



Figures S8. Views of the topological representation of the (8,3)-etd structure of **3** along the *c* axis (top right), with central projection (top left) and along the *a* axis (bottom). The blue spheres represent the copper(II) ions and the orange ones the Etmal²⁻ ligands.