Electronic Supplementary Information

Bulk crystal growth and characterization of bismuth ferritebased material Bi3FeO4(MoO4)2

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2. Figure S1. The experimental patterns of powder XRD of the residue in the platinum pan after DSC/TGA.

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crystal faces (100), (010) and (001), respectively, from which they are rhombic and regular.

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Crystal	BFM
Empirical formula	$Bi_3FeO_4(MoO_4)_2$
$M_{ m r}$	1066.67
Crystal system	Monoclinic
Space group	<u>C2</u> /c (No. 15)
Temperature (K)	<u>296</u>
<i>a</i> , <i>b</i> , <i>c</i> (Å)	<u>16.927(12)</u> , 11.672(12), 5.263(4)
β (°)	<u>107.160(10)</u>
$V(\text{\AA}^3)$	993.6(15)
Ζ	4
Dcalcd, g/cm ³	7.131
$\mu \text{ (mm}^{-1}\text{)}$	56.88
F (000)	1820
Crystal size (mm)	0.10 imes 0.08 imes 0.07
R _{int}	0.092
R indices $[I \ge 2\sigma(I)]^a$	$R_1 = 0.0482$, wR_2 = 0.1203
R indices (all data)	$R_1 = 0.0584$, wR_2 = 0.1254
Extinction coeff	0.0067(4)
GOF on F^2	1.018
CCDC	1879545

 Table S1. Crystal data and structure refinement for BFM.

$${}^{a}R_{1} = \sum ||F_{o}| - |F_{c}|| / \sum |F_{o}|, \ wR_{2} = \left\{ \sum [w(F_{o}^{2} - F_{c}^{2})^{2}] / \sum [w(F_{o}^{2})^{2}] \right\}^{1/2}$$



Figure S1. The experimental patterns of powder XRD of the residue in the platinum pan after DSC/TGA.



Figure S2. Photographs of the as-grown $Bi_3FeO_4(MoO_4)_2$ crystals through spontaneous

crystallization with MoO₃ flux.

Bi2—Bi1 ⁱ	3.649 (3)	Bi1—O6 ^{viii}	2.722 (12)
Bi2—O2 ⁱⁱ	2.235 (12)	Bi1—O5 ^{ix}	2.346 (14)
Bi2—O2	2.236 (14)	Bi1—O5 ^x	2.346 (14)
Bi2—O1 ⁱⁱⁱ	2.257 (14)	Mo1—O3	1.776 (13)
Bi2—O3 ⁱⁱⁱ	2.652 (12)	Mo1—O6	1.758 (13)
Bi2—O3	2.546 (14)	Mo1—O4	1.775 (14)
Bi2—O4 ^{iv}	2.683 (13)	Mo1—O5	1.825 (14)
Bi2—O5 ^v	2.519 (14)	Fe1—O2 ^{xi}	1.908 (13)
Bi1—O1 ^{vi}	2.202 (13)	Fe1—O2	1.908 (13)
Bi1—O1	2.202 (13)	Fe1—O1	1.920 (14)
Bi1—O6 ^{vii}	2.722 (12)	Fe1—O1 ^{xi}	1.920 (14)
O2—Bi2—Bi1 ⁱ	106.6 (4)	O1 ^{vi} —Bi1—O5 ^x	77.9 (5)
O2 ⁱⁱ —Bi2—Bi1 ⁱ	87.5 (4)	O6 ^{viii} —Bi1—Bi2 ⁱⁱ	88.2 (3)

Table S2. Bond Distances (Å) and Angles (deg.) for Bi₃FeO₄(MoO₄)₂ crystal.

O2 ⁱⁱ —Bi2—O2	77.3 (3)	O6 ^{vii} —Bi1—Bi2 ⁱⁱ	136.4 (3)
O2—Bi2—O1 ⁱⁱⁱ	80.9 (5)	O6 ^{vii} —Bi1—Bi2 ⁱ	88.2 (3)
O2 ⁱⁱ —Bi2—O1 ⁱⁱⁱ	104.1 (5)	O6 ^{viii} —Bi1—Bi2 ⁱ	136.4 (3)
O2 ⁱⁱ —Bi2—O3 ⁱⁱⁱ	137.9 (4)	O6 ^{vii} —Bi1—O6 ^{viii}	124.3 (6)
O2—Bi2—O3	80.9 (5)	O5 ^{ix} —Bi1—Bi2 ⁱⁱ	108.6 (3)
O2—Bi2—O3 ⁱⁱⁱ	76.3 (5)	O5 ^x —Bi1—Bi2 ⁱ	108.6 (3)
O2 ⁱⁱ —Bi2—O3	78.6 (5)	O5 ^{ix} —Bi1—Bi2 ⁱ	43.2 (3)
O2—Bi2—O4 ^{iv}	126.1 (4)	O5 ^x —Bi1—Bi2 ⁱⁱ	43.2 (3)
O2 ⁱⁱ —Bi2—O4 ^{iv}	152.9 (5)	O5 ^{ix} —Bi1—O6 ^{vii}	94.2 (4)
O2 ⁱⁱ —Bi2—O5 ^v	77.4 (4)	O5 ^x —Bi1—O6 ^{viii}	94.2 (4)
O2—Bi2—O5 ^v	138.1 (4)	O5 ^{ix} —Bi1—O6 ^{viii}	100.7 (4)
O1 ⁱⁱⁱ —Bi2—Bi1 ⁱ	34.6 (3)	O5 ^x —Bi1—O6 ^{vii}	100.7 (4)
O1 ⁱⁱⁱ —Bi2—O3	160.5 (5)	O5 ^x —Bi1—O5 ^{ix}	147.9 (7)
O1 ⁱⁱⁱ —Bi2—O3 ⁱⁱⁱ	103.4 (4)	O3—Mo1—O5	110.5 (6)
O1 ⁱⁱⁱ —Bi2—O4 ^{iv}	70.7 (4)	O6—Mo1—O3	112.6 (6)
O1 ⁱⁱⁱ —Bi2—O5 ^v	73.7 (5)	O6—Mo1—O4	107.2 (6)
O3—Bi2—Bi1 ⁱ	162.4 (3)	O6—Mo1—O5	106.5 (7)
O3 ⁱⁱⁱ —Bi2—Bi1 ⁱ	131.5 (3)	O4—Mo1—O3	106.4 (7)
O3—Bi2—O3 ⁱⁱⁱ	65.2 (3)	O4—Mo1—O5	113.8 (7)
O3 ⁱⁱⁱ —Bi2—O4 ^{iv}	67.5 (4)	O2 ^{xi} —Fe1—O2	123.3 (8)
O3—Bi2—O4 ^{iv}	115.4 (4)	O2 ^{xi} —Fe1—O1 ^{xi}	106.8 (6)
O4 ^{iv} —Bi2—Bi1 ⁱ	73.4 (3)	O2—Fe1—O1	106.8 (6)
O5 ^v —Bi2—Bi1 ⁱ	39.6 (3)	O2 ^{xi} —Fe1—O1	101.9 (5)
O5 ^v —Bi2—O3	125.3 (4)	O2—Fe1—O1 ^{xi}	101.9 (5)
O5v—Bi2—O3 ⁱⁱⁱ	141.5 (4)	O1—Fe1—O1 ^{xi}	117.2 (8)

O5v—Bi2—O4 ^{iv}	75.7 (4)	Bi2 ⁱⁱⁱ —O2—Bi2	113.0 (6)
Bi2 ⁱⁱ —Bi1—Bi2 ⁱ	84.42 (8)	Fe1—O2—Bi2	122.8 (6)
O1 ^{vi} —Bi1—Bi2 ⁱ	35.6 (4)	Fe1—O2—Bi2 ⁱⁱⁱ	123.0 (7)
O1—Bi1—Bi2 ⁱⁱ	35.6 (4)	Bi1—O1—Bi2 ⁱⁱ	109.8 (6)
O1—Bi1—Bi2 ⁱ	72.7 (4)	Fe1—O1—Bi2 ⁱⁱ	116.1 (6)
O1 ^{vi} —Bi1—Bi2 ⁱⁱ	72.7 (4)	Fe1—O1—Bi1	129.9 (7)
O1—Bi1—O1 ^{vi}	82.8 (7)	Bi2—O3—Bi2 ⁱⁱ	91.7 (4)
O1—Bi1—O6 ^{vii}	159.0 (4)	Mo1—O3—Bi2 ⁱⁱ	123.8 (7)
O1 ^{vi} —Bi1—O6 ^{viii}	159.0 (4)	Mo1—O3—Bi2	138.5 (6)
O1—Bi1—O6 ^{viii}	76.5 (4)	Mo1—O6—Bi1 ^{viii}	116.8 (6)
O1 ^{vi} —Bi1—O6 ^{vii}	76.5 (4)	Mo1—O4—Bi2 ^{xii}	127.8 (7)
O1—Bi1—O5 ^x	78.2 (5)	Bi1 ^x —O5—Bi2 ^{xiii}	97.1 (5)
O1—Bi1—O5 ^{ix}	77.9 (5)	Mo1—O5—Bi2 ^{xiii}	134.3 (7)
O1 ^{vi} —Bi1—O5 ^{ix}	78.2 (5)	Mo1—O5—Bi1 ^x	127.4 (7)
Symmetry codes: (i) $-x+1$, $-y+1$, $-z+2$; (ii) x , $-y+1$, $z+1/2$; (iii) x , $-y+1$, $z-1/2$; (iv) $-x+1/2$,			
y-1/2, -z+1/2; (v) $-x+1/2, y-1/2, -z+3/2;$ (vi) $-x+1, y, -z+5/2;$ (vii) $x+1/2, -y+3/2, z+3/2;$			
(viii) $-x+1/2$, $-y+3/2$, $-z+1$; (ix) $x+1/2$, $-y+3/2$, $z+1/2$; (x) $-x+1/2$, $-y+3/2$, $-z+2$; (xi) $-x+1$,			
y, -z+3/2; (xii) $-x+1/2, y+1/2, -z+1/2;$ (xiii) $-x+1/2, y+1/2, -z+3/2.$			



Figure S3. (a)-(c) Photographs of the indentation morphologies of the $Bi_3FeO_4(MoO_4)_2$ crystal faces (100), (010) and (001), respectively, from which they are rhombic and regular. $_{5}$

Raman spectra (cm ⁻¹)	Assignments
< 100 (vw)	O-Bi-O and O-Mo-O bond bending
124.49 (vs)	Bi-O bands and Bi and Mo translation and libration
200-600 (w)	Bi-O bonds, Mo-O bonds and translations of Fe atoms
761.12 (m)	Mo-O bonds and Mo-O-Bi bridges
868.67 (vs)	MoO ₄ asymmetric stretching
870.41 (m)	MoO ₄ symmetric stretching

Table S3. Observed Raman wavenumbers (cm⁻¹) and vibrational assignments for the $Bi_3FeO_4(MoO_4)_2$ crystal.¹⁻³

In the table, vs: very strong; m: medium; w: weak

References

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