Supporting Information for

Synthesis, Properties, and Formation Mechanism of Mn-doped Zn₂SiO₄ nanowires and Associated Heterostructures

Haiqing Liu,^a Dominic Moronta,^a Luyao Li,^a Shiyu Yue,^a and Stanislaus S. Wong^{a,*}

^aDepartment of Chemistry, State University of New York at Stony Brook, Stony Brook, NY 11794-3400, USA

*To whom correspondence should be addressed. <u>Phone:</u> 631-632-1703; 631-344-3178 <u>Email: Stanislaus.wong@stonybrook.edu; sswong@bnl.gov</u>



Figure S1. EDAX elemental mapping images of a series of as-synthesized Mn-doped Zn_2SiO_4 series with intended Mn dopant molar concentrations of 2%, 4%, 6%, 8%, and 10%, respectively. The scale bars in each row are 2 µm, 1 µm, 1 µm, 0.5 µm, and 1 µm, respectively.



Figure S2. XRD patterns of a series of as-synthesized pure Zn_2SiO_4 samples prepared with reaction time of 12 hours and 24 hours, respectively. Database standards are shown at the bottom as JCPDS #37-1485 and 05-0555, respectively, for Zn_2SiO_4 and $Zn_4Si_2O_7(OH)_2$ ·H₂O.



Figure S3. TEM images of as-synthesized pure Zn_2SiO_4 samples prepared with a reaction time

of (A) 12 hours and (B) 24 hours, respectively.



Figure S4. Large-scale SEM images of a series of optimized Mn-doped Zn_2SiO_4 samples with intended Mn dopant concentrations of (A) 3%, (B) 4%, (C) 6%, and (D) 8%, respectively.



Figure S5. High-magnification HRTEM images (A-D) in addition to SAED patterns (E-H) and their corresponding lower-magnification HRTEM images (I-L) associated with a series of optimized Mn-doped Zn_2SiO_4 samples with intended Mn dopant concentrations of (A, E, I) 3%, (B, F, J) 4%, (C, G, K) 6%, and (D, H, L) 8%, respectively.



Figure S6. Survey spectra of as-prepared Zn_2SiO_4 : Mn (0% undoped samples; black curve) and analogous Zn_2SiO_4 : Mn (4% doped samples; red curve) nanorods.



Figure S7. The Mn 2p XPS spectrum of as-prepared Zn₂SiO₄: Mn (4% dopant) nanorods is shown. Both the raw data (dashed black line) as well as the as-generated curve obtained after data fitting (solid red line) are highlighted.



Figure S8. Zn 2p XPS spectra of (A) as-prepared Zn₂SiO₄: Mn (0% undoped control sample)

and of corresponding (B) Zn₂SiO₄: Mn (4% dopant) nanorods



Figure S9. STEM image (top) as well as EDAX elemental mapping images (bottom) of the CdSe QD – Mn-doped (4%) Zn₂SiO₄ heterostructures.



Figure S10. UV-visible and corresponding photoluminescent spectra of as-synthesized CdSe quantum dots. The calculated average size of the QDs is 4.8 nm.



Figure S11. Representative TEM images of (A) MPA-capped CdSe QDs and of (B) MPA-capped CdSe QD – Mn-doped Zn_2SiO_4 0D-1D heterostructures.



Figure S12. PL emission spectra of not only MPA-capped CdSe QDs (black) but also MPAcapped CdSe QD - undoped Zn_2SiO_4 heterostructures (blue) upon 388 nm excitation.



Figure S13. UV-visible spectra of as-synthesized 6% Mn-doped Zn_2SiO_4 (black), of MPAcapped CdSe quantum dots (red), and of the resulting MPA-capped CdSe QD - 6% Mn-doped Zn_2SiO_4 heterostructures (blue), respectively.

Lifetime of CdSe QDs (excited at 388 nm)	
Sample	Observed lifetimes (ns) with corresponding standard deviation values
As-prepared CdSe QDs	24.8 (±0.6)
MPA-capped CdSe QDs	5.7 (±0.2)
MPA-capped CdSe QD - 4% Mn-doped Zn ₂ SiO ₄ heterostructures	3.2 (±0.3)

Table S1. Lifetimes of CdSe QD emission, measured at an excitation wavelength of 388 nm.