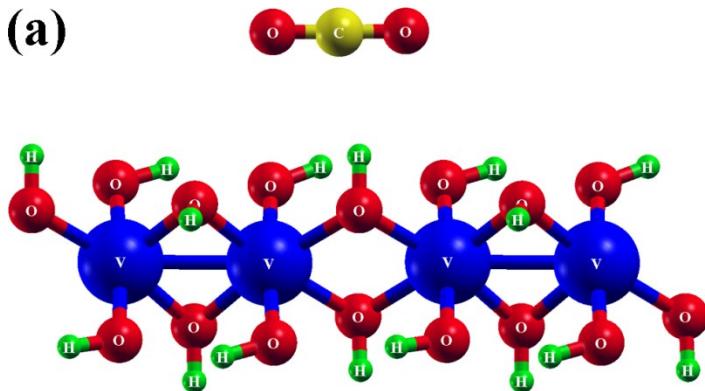


Electronic Supplementary Information (ESI)

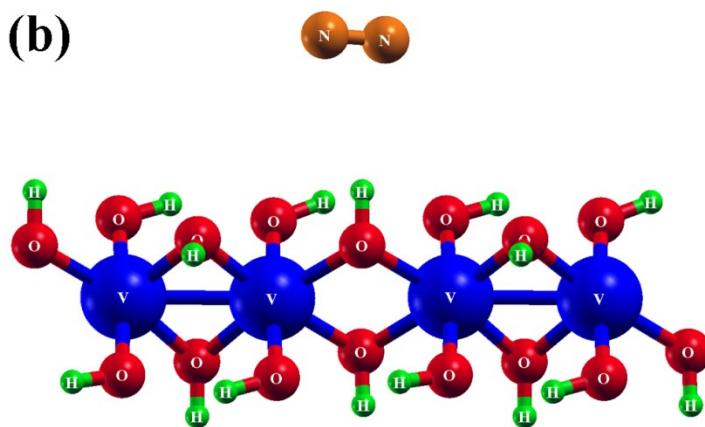
Highly infrared sensitive VO₂ nanowire as nano-optical device

Prabal Dev Bhuyan ^{a,d}, Sanjeev K. Gupta ^{a,*}, Ashok Kumar ^b, Yogesh Sonvane ^c and
P. N. Gajjar^d

(a)



(b)



(c)

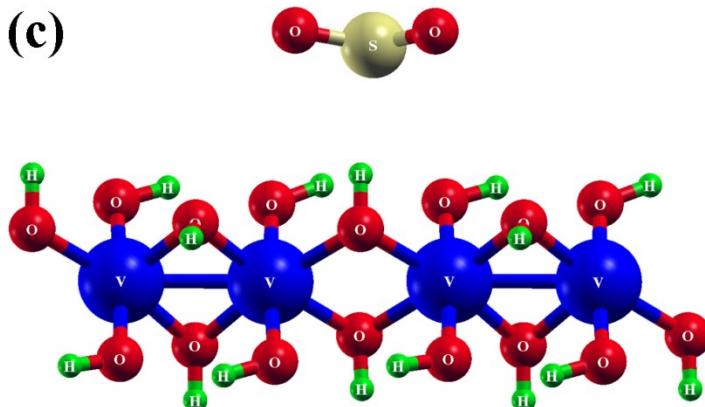


Figure S1: The optimized monoclinic structure of (a) CO₂ gas molecule adsorbed on 1D VO₂, (b) N₂ gas molecule adsorbed on VO₂ nanowire and (c) SO₂ gas molecule adsorbed on VO₂ nanowire structure.

Table S1: Structural parameters of 1D VO₂ rutile and monoclinic structure nanowires calculated within GGA and GGA+U method. The experimental values [18] of V-V bond length of bulk VO₂ is also given for comparison.

Structure	Lattice constant, a (Å)	V-V bond length for VO ₂ NW(Å)	V-V bond length in bulk VO ₂		
			GGA	GGA+U	Expt [1]
VO ₂ (R)	5.70	2.86	3.01	2.81	2.85
VO ₂ (M)	5.73	2.60	2.47	2.56	2.66
		3.13	3.12	3.16	3.12

Table S2: Table for polarization and magnetic moment for VO₂(R) and VO₂(M)

Structure	Polarization	Magnetic moment (Bohr magneton)
VO ₂ (M)	0	0.30
VO ₂ (R)	1	0.41

Table S3: Different parameters like bond length (b) of gas molecules, bond angle (θ), change in bond length (Δb), change in bond angle ($\Delta\theta$), change in charge transfer (ΔQ) and adsorption energy (E_{ad}) of different gas molecules adsorbed on monoclinic VO₂ nanowire structure.

Gas Molecules	Free molecules		Adsorption of gas molecules on VO ₂ (M) nanowire			
	b (Å)	θ (°)	Δb (Å)	$\Delta\theta$ (°)	ΔQ	E_{ad} (eV)
SO ₂	1.47	119	0.01	0.7	0.4877	-0.30
N ₂	1.10	-	0.01	-	0.1582	-0.06
CO ₂	1.16	180	0.01	0.84	0.2721	-0.12

References

1. J. Zhang, H. He, Y. Xie and B. J. Pan, Chem. Phys., 2013, **138(11)**, 114705.