

Photochemistry of glyoxylate embedded in sodium chloride clusters, a laboratory model for tropospheric sea-salt aerosols

Nina K. Bersenkowitsch, Milan Ončák, Christian van der Linde, Andreas Herburger,

Martin K. Beyer

Institut für Ionenphysik und Angewandte Physik, Universität Innsbruck,

Technikerstraße 25, 6020 Innsbruck, Austria

Email: milan.oncak@uibk.ac.at; martin.beyer@uibk.ac.at

Electronic supplementary information (ESI)

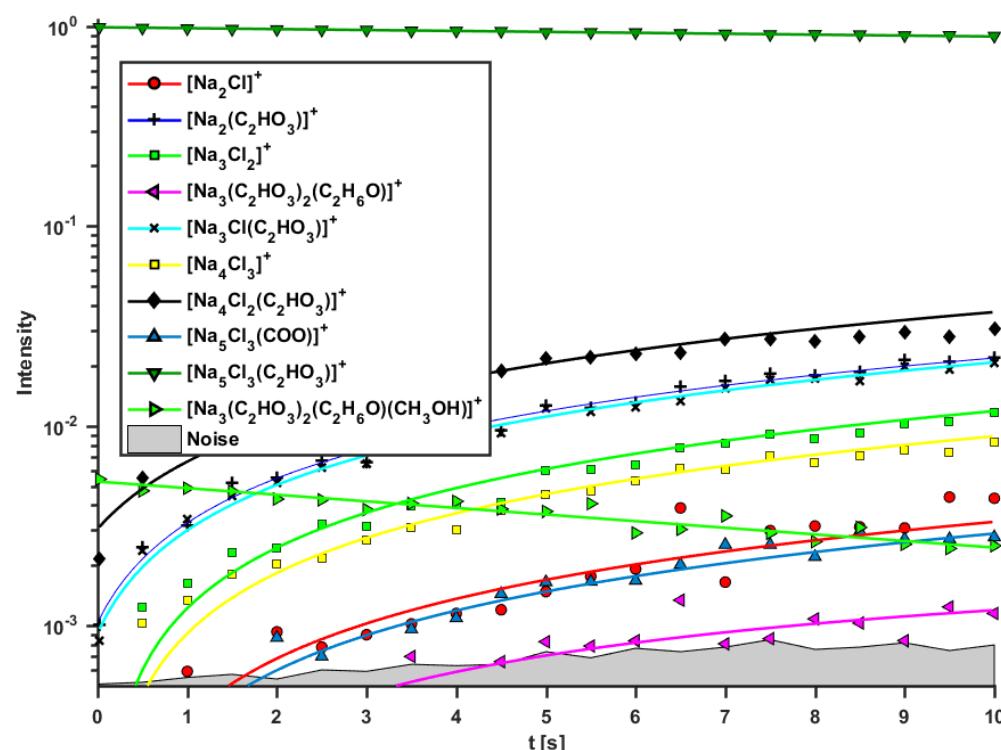


Figure S1: Kinetics of the cluster $[\text{Na}_5\text{Cl}_3(\text{C}_2\text{HO}_3)]^+$ at 335 nm and 20 accumulations. The linear decrease of the precursor ion with varying pulse numbers show that the fragmentation occurs due to single photon processes. Another peak, which was identified as $[\text{Na}_3(\text{C}_2\text{HO}_3)_2(\text{C}_2\text{H}_6\text{O})(\text{CH}_3\text{OH})]^+$, with very little intensity was directly situated beside the parent ion. Due to the very small mass difference of 0.17 m/z, it was not possible to eliminate this peak without exciting $[\text{Na}_5\text{Cl}_3(\text{C}_2\text{HO}_3)]^+$ in a way that enough signal remained. Therefore it contributes also to the first and fourth fragment listed in the legend. Due to the low intensity of this fragmenting cluster, the influence on the cross section is negligible.

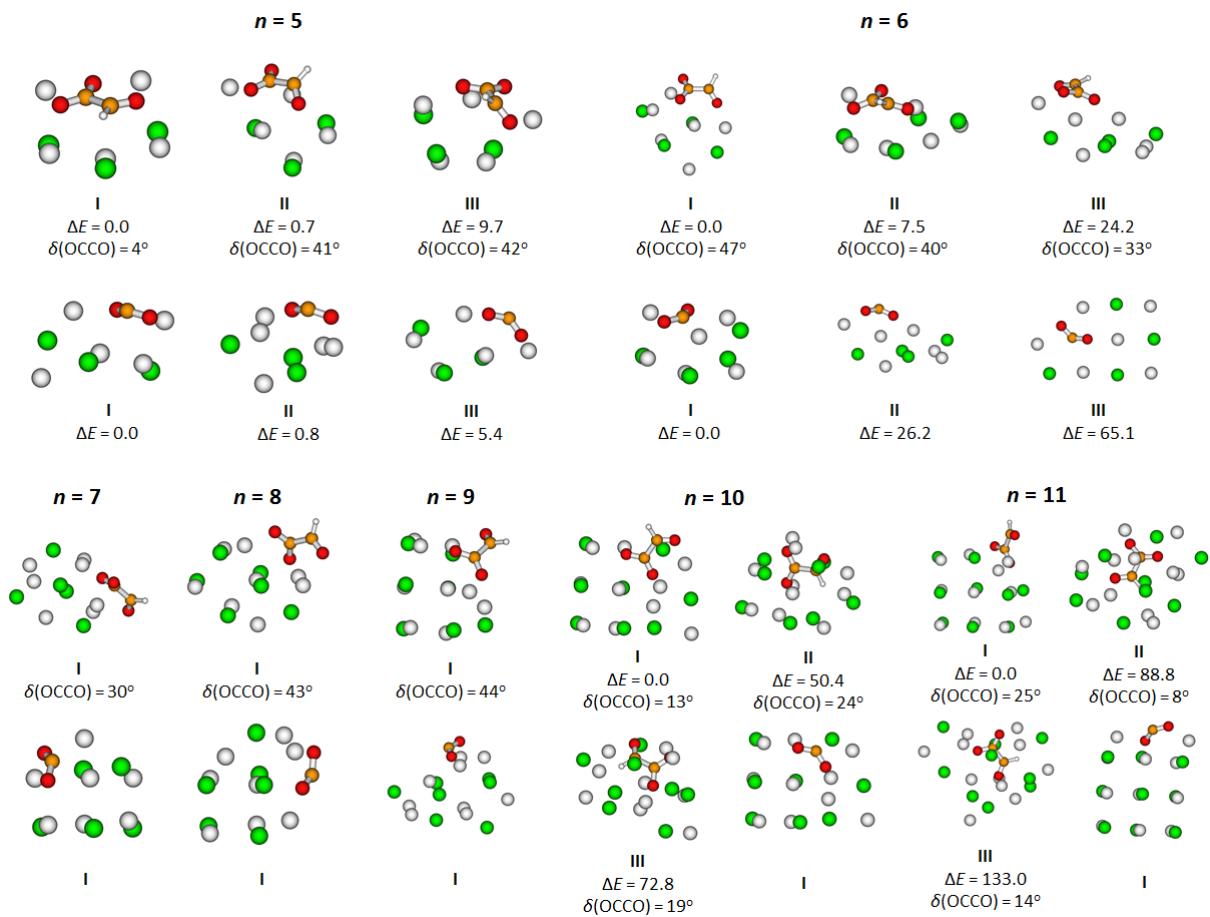


Figure 2: Structure, relative energy ΔE (in kJ mol^{-1}) and dihedral angles δ of the most stable $[\text{Na}_n\text{Cl}_{n-2}\text{C}_2\text{HO}_3]^+$ and $[\text{Na}_n\text{Cl}_{n-2}\text{CO}_2]^+$ isomers found, $n = 5–11$. For $n=10,11$, isomers with C_2HO_3^- inside the cluster are shown for comparison. Energy was calculated at the B3LYP+D2/def2TZVP//B3LYP+D2/6-31+g* level of theory. Zero point correction is included in all reported energies.

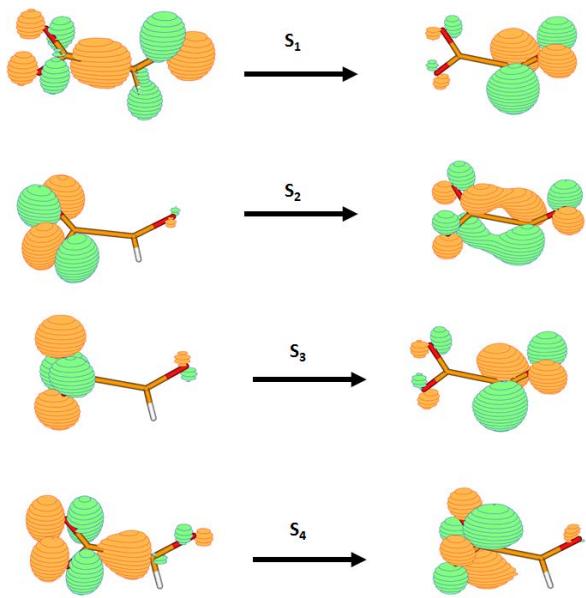


Figure S3: Natural transition orbitals (NTO) involved in the formation of excited states of the C₂HO₃⁻ ion, calculated at the TD-BHandHLYP/def2TZVP level of theory.

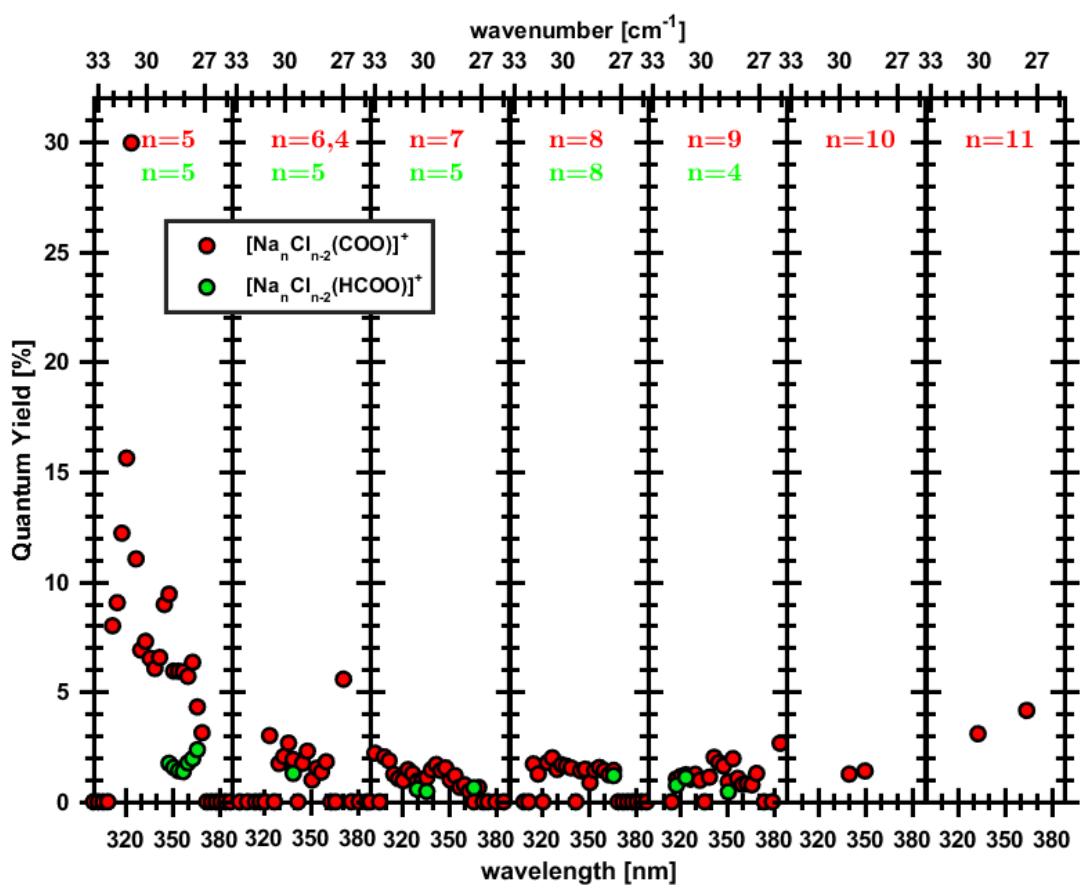


Figure S4: Quantum Yields for the fragments containing carbon dioxide anion radical (red line) and HCOO (green bullets).

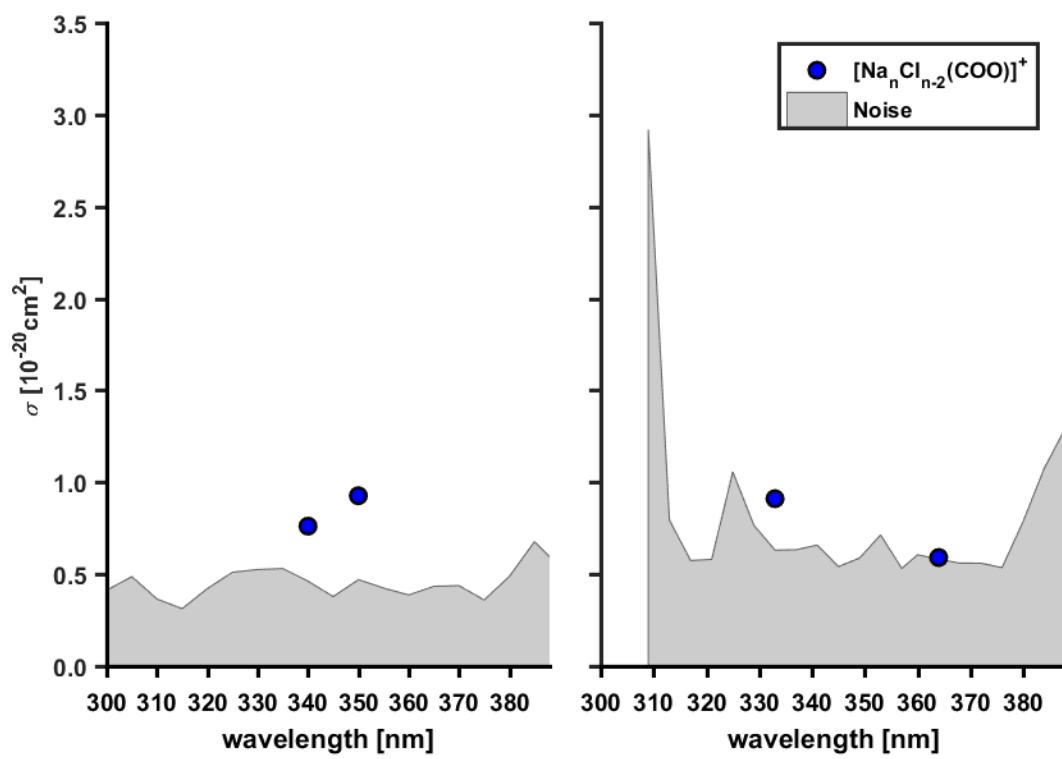


Figure S5: Quantum Yields for the fragments including carbon dioxide anion radical for the clusters $[\text{Na}_n \text{Cl}_{n-2}(\text{C}_2\text{HO}_3)]^+$, $n=10,11$. The products are formed hardly above the noise level.

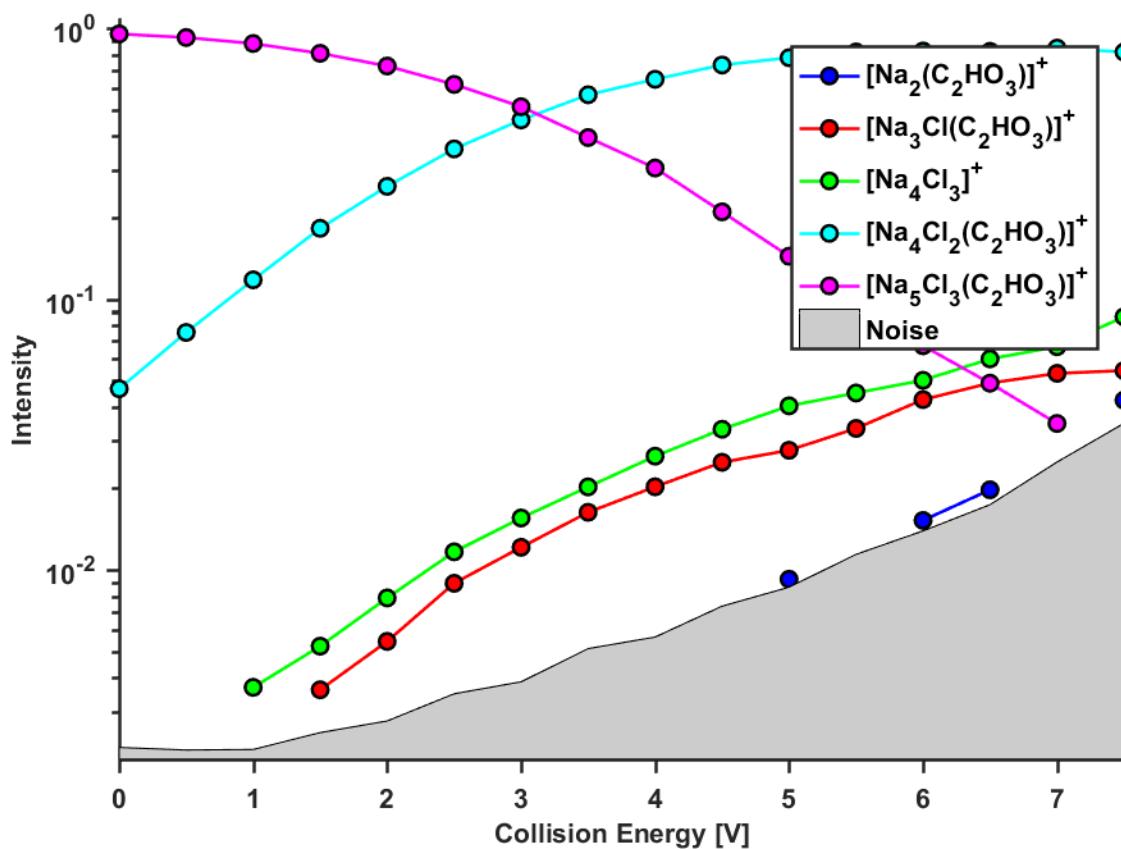


Figure S6: Collision induced dissociation of $Na_5Cl_3(C_2HO_3)^+$ in the hexapole collision cell. There is no evidence for C-C bond cleavage, since only stoichiometric fragments are observed.

Table S1: Relative energies ΔE (in kJ mol⁻¹) and dihedral angle δ (in degrees) of various isomers of $[\text{Na}_5\text{Cl}_3(\text{C}_2\text{HO}_3)]^+$ optimized at the B3LYP+D2/def2TZVP and B3LYP/def2TZVP levels of theory.

| isomer | B3LYP+D2/ def2TZVP | | B3LYP/def2TZVP | |
|--------|--------------------|----------|----------------|----------|
| | ΔE | δ | ΔE | δ |
| I | 0.0 | 6 | 12.4 | 14 |
| II | 1.1 | 45 | 16.7 | 89 |
| III | 10.1 | 44 | 12.4 | 14 |
| IV | 16.9 | 45 | 27.1 | 46 |
| V | 21.1 | 42 | 27.2 | 43 |
| VI | 23.3 | 6 | 0.0 | 4 |

Benchmarking of excited state calculations

Table S2 includes benchmarking of excitation energies of C_2HO_3^- , $[\text{Na}_2(\text{C}_2\text{HO}_3)]^+$, $[\text{Na}_3\text{Cl}(\text{C}_2\text{HO}_3)]^+$ and $[\text{Na}_5\text{Cl}_3(\text{C}_2\text{HO}_3)]^+$ ions. It can be seen that for C_2HO_3^- , results of the TD-BHandHLYP method are in much better agreement with the EOM-CCSD results than the ones of TD-CAMB3LYP. Oscillator strengths are comparable among all used methods, with the exception of S_2 and S_6 state whose oscillator strengths are underestimated at the TD-BHandHLYP level. There are only minor basis set effects.

For larger clusters, both DFT functionals predict low-lying states of charge-transfer character (the transfer takes place between Cl^- and C_2HO_3^- , i.e. forming doubly-charged $\text{C}_2\text{HO}_3^{2-}$). These states are not reproduced at the EOM-CCSD level and we therefore conclude that they are artificial.

Table S2: Excitation energies E (in eV) and oscillator strengths f of several investigated ions as calculated using various approaches in the structure optimized at the B3LYP+D2/def2TZVP level of theory.

| method | | EOMCCSD/def2TZVP | | EOMCCSD/aug-cc-pVDZ | | EOMCCSD/cc-pVDZ | | TD-BHandHLYP/def2TZVP | | TD-CAMB3LYP/def2TZVP | |
|---|-------|------------------|--------|---------------------|--------|-----------------|--------|-----------------------|--------|----------------------|--------|
| ion | State | E | f | E | f | E | f | E | f | E | f |
| C_2HO_3^- | S_1 | 3.88 | 0.0011 | - | - | 3.85 | 0.0013 | 3.91 | 0.0009 | 3.67 | 0.0009 |
| | S_2 | 5.66 | 0.0464 | - | - | 5.65 | 0.0472 | 5.73 | 0.0319 | 5.23 | 0.0422 |
| | S_3 | 5.98 | 0.0120 | - | - | 6.02 | 0.0113 | 5.87 | 0.0110 | 5.39 | 0.0085 |
| | S_4 | 6.14 | 0.0118 | - | - | 6.18 | 0.0121 | 6.12 | 0.0100 | 5.82 | 0.0099 |
| | S_5 | 6.71 | 0.0128 | - | - | 6.82 | 0.0144 | 6.74 | 0.0086 | 6.28 | 0.0082 |
| | S_6 | 7.00 | 0.0486 | - | - | 7.06 | 0.0422 | 7.15 | 0.0355 | 6.66 | 0.0217 |
| $[\text{Na}_2(\text{C}_2\text{HO}_3)]^+$ | S_1 | 3.81 | 0.0002 | 3.81 | 0.0004 | 3.83 | 0.0002 | 3.83 | 0.0002 | 3.61 | 0.0002 |
| | S_2 | 5.31 | 0.0000 | 5.28 | 0.0000 | 5.30 | 0.0000 | 5.31 | 0.0000 | 4.83 | 0.0000 |
| | S_3 | 6.20 | 0.0872 | 6.13 | 0.0845 | 6.17 | 0.0879 | 6.15 | 0.0645 | 5.71 | 0.0502 |
| | S_4 | 6.22 | 0.0000 | 6.21 | 0.0001 | 6.26 | 0.0001 | 6.25 | 0.0000 | 5.87 | 0.0000 |
| | S_5 | 7.10 | 0.0376 | 6.88 | 0.0364 | 6.79 | 0.0397 | 6.90 | 0.0384 | 6.44 | 0.0338 |
| | S_6 | 7.35 | 0.0032 | 7.15 | 0.0036 | 7.02 | 0.0032 | 7.08 | 0.0013 | 6.69 | 0.0201 |
| $[\text{Na}_3\text{Cl}(\text{C}_2\text{HO}_3)]^+$ | S_1 | 3.80 | 0.0001 | 3.79 | 0.0002 | 3.82 | 0.0001 | 3.80 | 0.0001 | 3.58 | 0.0001 |
| | S_2 | 5.43 | 0.0000 | 5.40 | 0.0000 | 5.45 | 0.0001 | 4.81 | 0.0025 | 4.63 | 0.0029 |
| | S_3 | 6.09 | 0.0004 | 6.08 | 0.0008 | 6.13 | 0.0367 | 5.07 | 0.0004 | 4.82 | 0.0003 |
| | S_4 | 6.32 | 0.0924 | 6.24 | 0.0703 | 6.15 | 0.0370 | 5.20 | 0.0000 | 5.00 | 0.0001 |
| | S_5 | 6.37 | 0.0718 | 6.26 | 0.0898 | 6.33 | 0.0963 | 5.47 | 0.0000 | 5.02 | 0.0000 |
| | S_6 | 6.54 | 0.0013 | 6.47 | 0.0010 | 6.47 | 0.0496 | 5.78 | 0.0338 | 5.58 | 0.0413 |
| $[\text{Na}_5\text{Cl}_3(\text{C}_2\text{HO}_3)]^+$ | S_1 | - | - | 3.88 | 0.0006 | 3.91 | 0.0003 | 3.92 | 0.0005 | 3.68 | 0.0006 |
| | S_2 | - | - | 5.25 | 0.0003 | 5.28 | 0.0004 | 5.29 | 0.0019 | 4.79 | 0.0010 |
| | S_3 | - | - | 6.00 | 0.0179 | 6.05 | 0.0164 | 5.42 | 0.0347 | 5.03 | 0.0310 |
| | S_4 | - | - | 6.12 | 0.0226 | 6.17 | 0.0258 | 5.52 | 0.0314 | 5.13 | 0.0340 |

Cartesian coordinates and energy (in Ångstrom and Hartree, respectively) of structures included in the manuscript

Structures optimized at the B3LYP+D2/def2TZVP level of theory

C2HO3-

E=-302.639808

C -0.595211 0.022156 0.027260
O -1.150479 -1.083327 -0.170202
O -1.043545 1.180884 0.021645
C 0.885883 -0.183715 0.367777
H 1.044748 -0.742778 1.326933
O 1.845426 0.116459 -0.313587

CHO-

E=-113.896773

C 0.079244 0.628114 0.000000
H -1.109415 1.050824 0.000000
O 0.079244 -0.602438 0.000000

CHO

E=-113.891303

C 0.061785 0.582494 0.000000
H -0.864996 1.220692 0.000000
O 0.061785 -0.589457 0.000000

CO2-

E=-188.636373

C 0.000000 0.341429 0.000000
O 1.144167 -0.128494 0.000000
O -1.144167 -0.127578 0.000000

CO2

E=-188.658348

C 0.000000 0.000000 -0.000052
O 0.000000 0.000000 -1.159716
O 0.000000 0.000000 1.159755

CO

E=-113.357481

C 0.000000 0.000000 -0.642728
O 0.000000 0.000000 0.482046

NaCl

E=-622.604860

Cl 0.000000 0.000000 0.932933
Na 0.000000 0.000000 -1.441805

Na2C2HO3+

E=-627.134431

C -0.987514 1.338271 0.000022
O -2.078647 0.817372 0.000046
C 0.319620 0.515759 0.000005
O 0.226401 -0.752148 -0.000010
O 1.370854 1.172224 0.000019
H -0.859477 2.431036 0.000095
Na -1.899561 -1.470479 -0.000021
Na 2.692104 -0.661775 -0.000043

Na2Cl2CO

E=-1358.653922

Na -0.653577 -0.002950 -0.002225
Cl 0.928423 -2.037954 -0.000217
Na 2.470220 0.004595 0.001787
Cl 0.918493 2.039695 -0.000217
O -3.186752 -0.003875 0.000391
C -4.314437 -0.002781 0.001515

Na₂Cl₂
E=1245.290151
Cl -2.034618 -0.000004 -0.000006
Na 0.000008 -1.560793 -0.000007
Na -0.000009 1.560792 -0.000007
Cl 2.034619 0.000004 0.000015

Na₂ClC₂HO₃
E=-1087.635966
C -2.470320 -0.762036 0.000167
C -1.806774 0.637570 0.000265
O -2.573930 1.589306 -0.000444
O -1.852719 -1.804793 0.000157
O -0.533026 0.622492 0.000563
Na 1.316926 1.705858 -0.000500
Na 0.504463 -1.375646 0.000492
H -3.574089 -0.748572 -0.000172
Cl 2.875222 -0.317236 -0.000267

Na₂ClHC₂O₃
E=-974.277987
C 2.284248 0.361080 0.000280
O 1.178593 0.984400 0.000041
O 2.414193 -0.875524 -0.000278
Na -0.832649 1.651022 -0.000204
Na 0.178663 -1.322787 -0.000041
Cl -2.262012 -0.448814 0.000107
H 3.200271 0.981753 0.001089

Na₂Cl+
E=-784.774210
Na 0.000000 0.000000 2.497394
Cl 0.000000 0.000000 0.000000
Na 0.000000 0.000000 -2.497394

Na₃Cl₂C₂HO₃
E=-1710.317310
C 2.346027 -2.036377 0.000640
O 3.171170 -1.154014 0.001351
C 0.820708 -1.783551 -0.000080
O 0.420519 -0.587398 0.000379
O 0.131066 -2.817503 -0.001155
Na -1.917695 -1.836348 -0.001228
Na -1.084167 1.414147 0.000377
Na 2.193513 0.977846 -0.000335
Cl 0.870489 3.112324 -0.000611
H 2.634905 -3.100485 0.000821
Cl -3.371990 0.204104 0.000862

Na₃Cl₂+
E=-1407.437993
Na 0.000000 0.000000 0.000000
Cl 0.000000 0.000000 2.571363
Na 0.000000 0.000000 -5.032304
Cl 0.000000 0.000000 -2.571363

Na 0.000000 0.000000 5.032304

Na3Cl3
E=-1867.958624
Cl -0.000004 1.833228 0.000225
Na -2.512589 1.170623 0.000023
Na 2.512581 1.170627 0.000063
Cl 2.604369 -1.331204 -0.000042
Cl -2.604361 -1.331211 -0.000005
Na 0.000002 -1.059778 -0.000359

Na3CIC2HO3+
E=-1249.802884
C 1.004786 0.405273 0.000320
O -0.119852 0.932486 0.000235
O 1.258839 -0.831170 0.000238
C 2.210412 1.375649 0.000255
O 3.351891 0.981929 0.000125
H 1.964609 2.448969 0.000118
Na 3.456173 -1.329973 0.000226
Na -1.076912 -1.454061 -0.000213
Cl -3.425501 -0.562828 -0.000624
Na -2.283744 1.671998 0.000191

Na3Cl2COOH
E=-1596.953677
C -0.215288 -1.951129 0.000479
O -0.649659 -1.554980 -1.116128
O -0.649824 -1.554333 1.116857
Na 0.563462 0.401087 1.549328
Na 0.561265 0.400169 -1.549691
Cl -1.146766 1.905161 -0.000025
Na -2.360026 -0.369253 0.000315
H 0.602458 -2.689497 0.000753
Cl 2.598144 0.125353 -0.000500

Na3CICOOH+
E=-1136.445487
C 1.848649 0.000047 -0.463519
O 1.447858 -1.115708 -0.037732
O 1.447792 1.115784 -0.037756
Na -0.420595 2.069783 -0.771861
Cl -1.826551 -0.000054 0.126210
Na -0.420474 -2.069802 -0.771865
Na 0.312503 0.000016 1.771243
H 2.608501 0.000058 -1.263246

Na4Cl2COOH+
E=-1759.133531
H -0.004882 -3.666818 -0.659599
C -0.003493 -2.666020 -0.197488
O 1.107909 -2.135754 0.050154
O -1.113434 -2.133268 0.051434
Na 2.923618 -1.049734 -0.391431
Na -0.000209 -0.187143 1.016560
Na -2.926444 -1.042925 -0.390726
Cl 2.035494 1.373897 0.119504
Na 0.003769 2.918064 -0.508510
Cl -2.031849 1.378704 0.118555

Na4Cl3+
E=-2030.139930

Na 0.000000 -2.658045 -0.703223
 Cl 0.000000 2.311568 0.123147
 Na -2.301934 1.329022 -0.703223
 Cl 2.001876 -1.155784 0.123147
 Na 2.301934 1.329022 -0.703223
 Cl -2.001876 -1.155784 0.123147
 Na 0.000000 0.000000 1.538714

Na5Cl3COH+
 E=-2306.436405
 Na 1.804949 -0.001219 1.696637
 Cl 1.835433 1.984273 -0.083299
 Na -3.566208 0.000502 0.007837
 Na -0.694557 1.805329 0.180652
 Cl -1.460737 0.001545 -1.695416
 Na 1.435289 0.000650 -1.849042
 Na -0.695690 -1.805319 0.176810
 Cl 1.834400 -1.984664 -0.084581
 C -1.891157 -0.001309 1.971932
 H -2.159977 -0.003228 3.059663
 O -0.646166 -0.000989 1.805369

Na5Cl3+
 E=-2192.449317
 Na -3.076154 -1.721797 -0.427679
 Cl -0.585654 -2.425862 0.289344
 Na 1.883027 -2.445352 -0.603496
 Cl 2.616732 0.000181 -0.003838
 Na 0.153585 -0.000012 1.173902
 Cl -0.586022 2.425713 0.289275
 Na 1.882757 2.445679 -0.603325
 Na -3.076482 1.721431 -0.427700

Na5Cl3COOH+
 E=-2381.812018
 Na -0.846472 -1.346430 0.845614
 Na 1.819165 -1.208998 -1.279435
 Cl 1.632542 -1.760890 1.327168
 Na 1.730318 0.904755 1.583265
 Cl 2.384464 1.381075 -0.973663
 Cl -1.111204 1.379261 1.275747
 Na -3.542378 0.568233 0.514312
 Na -0.144379 1.508004 -1.274497
 C -1.802211 -1.018771 -1.718650
 O -2.634627 -0.971859 -0.777963
 H -2.181850 -1.254439 -2.726787
 O -0.563162 -0.816230 -1.589590

NaC2H3O
 E=-464.951983
 C 0.369266 1.164589 0.000166
 O -0.820280 1.428360 -0.000010
 C 0.960577 -0.276495 -0.000135
 O 0.115241 -1.220343 0.000303
 O 2.186935 -0.298781 -0.000165
 Na -1.905116 -0.597713 -0.000142
 H 1.122041 1.972388 0.000359

NaCOOH
 E=-351.596854
 C 1.052620 -0.000504 -0.000021
 O 0.476876 1.120404 0.000040

O 0.475323 -1.120515 0.000040
H 2.160254 -0.001263 -0.000257
Na -1.463052 0.000471 -0.000024

NaCICO
E=-735.969427
Cl -2.379654 0.354644 0.000000
Na 0.000000 0.225456 0.000000
C 3.500455 -0.814776 0.000000
O 2.431424 -0.452539 0.000000

Na₅Cl₃C₂HO₃⁺, isomer I
E=-2495.184208
na -2.629105 1.683740 -0.776296
cl -2.547822 -0.869536 -1.330798
na -2.197781 -1.149735 1.320902
cl 0.357747 -1.916159 1.260453
na 0.031175 -1.583521 -1.461245
cl 2.397707 -0.301025 -1.720834
na 2.790581 -0.956108 0.812161
na 1.758607 2.091800 -0.878452
c -0.566334 1.469970 0.967856
o -0.364449 2.090168 -0.106092
o -1.674493 1.114468 1.420522
c 0.668471 1.074758 1.808538
o 1.803445 1.248034 1.412123
h 0.453149 0.636554 2.791497

Na₅Cl₃C₂HO₃⁺, isomer II
E=-2495.183787
C 0.785631 2.150014 0.876035
O -0.288684 2.168259 0.313626
C 1.933401 1.313084 0.306914
O 2.481981 0.550759 1.139416
O 2.146354 1.371499 -0.922213
Na 0.021561 0.735810 -1.779440
Na 1.048422 -1.127427 1.865899
Na -2.548567 1.366996 0.671160
Na 3.380971 -0.630043 -0.874092
Cl -2.560843 0.215013 -1.702707
Cl -1.522358 -0.687973 2.051053
Na -1.783559 -1.930998 -0.352630
Cl 0.948795 -1.803986 -0.820667
H 0.946371 2.647622 1.845277

Na₅Cl₃C₂HO₃⁺, isomer III
E=-2495.180377
C -1.381382 0.949179 -1.339501
O -1.920012 1.765902 -0.560258
C -2.122844 -0.346649 -1.668391
O -2.668223 -0.986859 -0.784952
H -2.049191 -0.735646 -2.695474
Na 1.855104 0.022175 -1.855178
Na -1.013478 -2.287406 0.489851
Na -3.028947 0.686061 1.059625
Na 0.296057 2.186613 0.393379
Cl 2.791199 1.430420 0.214341
Na 2.071535 -0.729814 1.515783
Cl 1.334352 -2.301036 -0.558599
Cl -0.620325 -0.046885 2.006134
O -0.223925 0.978964 -1.798185

Na5Cl3C2HO3+, isomer IV
E=-2495.177778
C -3.488551 1.029351 -1.020994
C -2.297836 0.090153 -1.243044
O -2.468150 -1.120887 -0.980676
O -1.223333 0.653801 -1.560664
O -4.143194 0.951500 -0.004699
H -3.671594 1.816844 -1.770457
Na -3.272535 -0.910591 1.218808
Cl -0.643931 -0.275559 1.695048
Na 2.070234 -0.396151 1.751415
Na -0.153161 -1.536896 -0.819776
Na -0.022100 1.824616 -0.019670
Cl 2.586077 1.889005 0.375000
Na 3.758935 0.265192 -1.313040
Cl 2.462103 -1.855626 -0.497819

Na5Cl3C2HO3+, isomer V
E=-2495.176177
C -2.786678 1.380307 -1.195351
C -1.709602 0.284950 -1.196684
O -0.525019 0.705279 -1.083279
O -3.768368 1.273931 -0.497643
O -2.093496 -0.896409 -1.207123
Na 1.675447 0.728102 -1.701543
Na -0.169056 -1.823517 -0.194093
Na -3.618124 -0.817450 0.656097
Cl 2.382105 -1.731594 -0.897464
Na 3.825800 -0.432133 0.864026
Cl -1.323820 -0.730763 2.026853
H -2.595455 2.286365 -1.794261
Cl 2.551604 1.848656 0.587331
Na 0.040487 1.389780 1.217931

Na5Cl3C2HO3+, isomer VI
E=-2495.175326
c 2.796060 0.605381 -0.300566
o 1.745173 -0.055051 -0.103403
o 2.941915 1.838799 -0.336360
c 4.101240 -0.201622 -0.513994
o 4.154233 -1.402053 -0.391783
h 4.989563 0.397398 -0.770844
na -0.564049 -0.040110 0.184743
cl -0.433964 -2.755367 0.758218
na 2.074991 -2.328361 0.169234
na 1.385833 3.278156 0.170522
cl -1.072248 2.646907 0.673077
na -3.468496 2.123916 -0.190589
cl -3.162208 -0.360060 -0.838353
na -2.859261 -2.843479 -0.131133

Na5Cl3CO2+, isomer I
E=-2381.165358
Na -3.098281 -1.232454 -0.587678
Cl -2.108035 0.940324 -1.623255
Na -1.699119 1.763567 0.902581
Cl 1.006481 1.385267 1.017385
Na 3.391340 1.317351 -0.210203
Cl 2.807753 -1.106252 -0.948417
Na 1.435164 -1.398382 1.324620
Na 0.278120 -0.194863 -1.259154
C -1.254919 -1.062842 1.408379

O -0.793952 -1.679133 0.424453
O -2.312964 -0.465757 1.588144

Na5Cl3CO2+, isomer II
E=-2381.164996
Cl 2.341221 1.728724 0.177688
Na -0.169633 2.083893 0.517688
Na 1.914492 -0.466898 1.623631
Na 1.626569 0.103106 -1.826661
Cl 1.658732 -2.168071 -0.407865
Cl -1.197391 0.565706 -1.586472
Na -0.859123 -1.821530 -0.079363
Na -3.645674 0.363453 -0.508673
C -1.724814 0.090167 1.651566
O -0.502043 -0.028327 1.781397
O -2.601412 -0.668094 1.216201

Na5Cl3CO2+, isomer III
E=-2381.163338
Na -3.293422 0.538089 -1.110594
Cl -2.361261 -1.719548 -0.156731
Cl -1.732604 2.067170 0.338457
Na -1.472557 -0.154040 1.859734
Na 0.200406 -1.888929 -0.786676
Cl 1.208543 -0.559747 1.484298
Na 0.776647 1.592275 -0.351283
Na 3.790910 -0.110488 0.872249
C 2.445755 -0.102026 -1.638384
O 3.034076 0.802629 -1.023019
O 1.260191 -0.243593 -1.953211

Na5Cl3CO2+, isomer IV
E=-2381.162867
Na -4.020715 -0.118440 -0.163417
Cl -1.928696 -0.931146 -1.531772
Na -0.805591 1.422830 -0.690541
Na -0.216311 -1.929785 0.278451
Na 1.652841 0.596894 1.711792
Cl 2.400425 -1.692039 0.498702
Na 3.182560 -0.074346 -1.411534
Cl 1.778103 1.988335 -0.596563
O -0.585001 0.121614 1.532054
C -1.802630 0.338306 1.655871
O -2.558997 1.115126 1.067477

Na5Cl3CO2+, isomer V
E=-2381.157699
Na -1.633939 -3.008609 -0.001013
Cl 0.936343 -2.701852 0.000940
Na 3.396136 -1.849381 0.000547
Cl 2.945757 0.710500 -0.001048
Na 0.259942 -0.013411 0.000227
Cl -0.504794 2.636060 0.000882
Na 2.050137 3.148969 -0.000220
Na -2.963937 1.809826 -0.000240
C -3.267871 -0.881667 -0.000145
O -2.025844 -0.694049 -0.001102
O -4.223993 -0.134869 0.000525

Structures optimized at the B3LYP+D2/6-31+g* level of theory

CHO

E=-113.845662

C 0.062303 0.589088 0.000000
H -0.872241 1.216161 0.000000
O 0.062303 -0.593836 0.000000

[Na5Cl3C2HO3]+, isomer I

E=-2494.926396

na 2.596868 1.727683 0.769220
cl 2.600191 -0.841068 1.318828
na 2.249578 -1.117445 -1.342055
cl -0.321365 -1.873673 -1.240742
na 0.007385 -1.530926 1.490592
cl -2.416361 -0.338880 1.715331
na -2.805345 -0.986358 -0.829133
na -1.812249 2.067352 0.857444
c 0.536071 1.426076 -0.954876
o 0.320503 2.007962 0.147510
o 1.662074 1.122648 -1.423426
c -0.691995 1.036622 -1.808472
o -1.838492 1.213111 -1.412960
h -0.477672 0.622218 -2.803726

[Na5Cl3C2HO3]+, isomer II

E=-2494.925716

C 0.713049 2.208671 0.648405
O -0.350216 2.192227 0.046843
C 1.885038 1.351276 0.155194
O 2.479662 0.695848 1.055783
O 0.082621 1.286347 -1.084585
Na -0.012711 0.545898 -1.861577
Na 1.132536 -0.929585 1.981940
Na -2.577616 1.428580 0.544815
Na 3.387617 -0.643387 -0.850259
Cl -2.592102 -0.023218 -1.670332
Cl -1.443593 -0.450599 2.109518
Na -1.752534 -2.012192 -0.113446
Cl 0.972584 -1.867172 -0.632030
H 0.837616 2.759299 1.596204

[Na5Cl3C2HO3]+, isomer III

E=-2494.921786

C 1.331676 -0.987948 -1.338964
O 1.878414 -1.807845 -0.555596
C 2.067653 0.313392 -1.669771
O 2.655896 0.934618 -0.787501
H 1.957575 0.731671 -2.684035
Na -1.884012 -0.035484 -1.862578
Na 1.092841 2.315049 0.473963
Na 3.029915 -0.734911 1.030200
Na -0.321777 -2.191579 0.432223
Cl -2.814600 -1.375998 0.267029
Na -2.048514 0.801598 1.533166
Cl -1.261466 2.291930 -0.599310
Cl 0.635409 0.071084 1.987055
O 0.163770 -1.022400 -1.790832

[Na5Cl3CO2]+, isomer I

E=-2380.948527

Na 3.113889 -1.266858 0.531452
 Cl 2.112674 0.881523 1.651787
 Na 1.707602 1.810833 -0.856600
 Cl -1.005544 1.400467 -0.966830
 Na -3.419919 1.327513 0.225088
 Cl -2.822657 -1.128227 0.910288
 Na -1.421373 -1.389587 -1.360514
 Na -0.280213 -0.247005 1.269348
 C 1.255531 -0.990048 -1.435820
 O 0.792716 -1.649981 -0.469413
 O 2.323652 -0.382212 -1.580682

[Na5Cl3CO2]+, isomer II
 E=-2380.947859
 Cl 2.358052 -1.708535 -0.178208
 Na -0.156885 -2.091402 -0.527509
 Na 1.907421 0.481585 -1.647600
 Na 1.633495 -0.101878 1.854532
 Cl 1.643995 2.168441 0.411973
 Cl -1.193166 -0.580529 1.578183
 Na -0.881126 1.821954 0.080319
 Na -3.657569 -0.366904 0.507130
 C -1.724109 -0.096099 -1.665641
 O -0.488495 0.017762 -1.764386
 O -2.599633 0.663525 -1.203720

[Na5Cl3CO2]+, isomer III
 E=-2380.946955
 Na -3.324741 0.547411 -1.115503
 Cl -2.362538 -1.706560 -0.151139
 Cl -1.727212 2.055153 0.336063
 Na -1.473250 -0.156631 1.892795
 Na 0.203466 -1.898306 -0.802459
 Cl 1.205910 -0.559408 1.475710
 Na 0.787994 1.596198 -0.375186
 Na 3.795544 -0.094560 0.886746
 C 2.457132 -0.113653 -1.644835
 O 3.041124 0.795299 -1.013137
 O 1.259295 -0.253983 -1.950875

[Na6Cl4C2HO3]+, isomer I
 E=-3117.566080
 Na 2.841097 1.003424 1.046286
 Cl 3.155176 -1.582063 0.535180
 Na 0.763059 -1.999769 1.564689
 Cl 0.150014 0.649564 1.783952
 Na -2.124300 2.109243 1.334616
 Cl -3.585772 0.170903 0.191054
 Na -3.281686 -2.355101 -0.391305
 Cl -0.622090 -2.448182 -0.774588
 Na 1.794079 -1.270473 -1.727140
 C 1.236240 1.888095 -1.274800
 Na -1.080693 0.270062 -0.775143
 C 0.182097 2.988963 -1.102203
 O 2.333186 2.203213 -0.771430
 O 0.876650 0.792889 -1.793291
 O -0.920476 2.747360 -0.622504
 H 0.493406 4.024927 -1.317406

[Na6Cl4C2HO3]+, isomer II
 E=-3117.563087
 C -0.529978 2.882134 -0.733845

O 0.643403 2.714775 -0.443307
 C -1.433783 1.703144 -1.119864
 O -0.946388 0.796065 -1.843527
 O -2.592125 1.735959 -0.625420
 Na 2.054611 2.089503 1.381572
 Na -2.768833 0.734821 1.428924
 Na 1.093627 0.219807 -0.943189
 Na -2.882039 -0.511505 -1.778834
 Cl -0.067595 0.402458 1.746008
 Cl 3.673022 0.629323 -0.071831
 Na -0.608950 -2.248670 1.120410
 Cl -3.167411 -1.736824 0.533864
 H -1.006425 3.875545 -0.661995
 Na 3.767468 -1.972753 -0.273244
 Cl 1.252287 -2.517594 -0.749792

[Na₆Cl₄C₂HO₃]⁺, isomer III
 E=-3117.556206
 Na -1.570054 1.892514 -0.866802
 C -2.131739 -0.673053 -1.541626
 Na -4.331256 -0.293561 0.289543
 Cl -2.586542 1.258239 1.589658
 Na -0.073962 0.667922 2.150035
 Cl 1.199869 -1.626651 1.600071
 Na 3.706894 -2.055806 1.053146
 Cl 3.956763 -0.469130 -1.039316
 Na 3.581820 2.094807 -0.701639
 Cl 1.041483 2.108897 0.012071
 Na 1.187448 -0.554238 -0.929867
 O -0.897605 -0.441002 -1.645482
 O -3.057948 0.176871 -1.576405
 C -2.512899 -2.128133 -1.216955
 H -1.873569 -2.927339 -1.631739
 O -3.440092 -2.379193 -0.467810

[Na₆Cl₄CO₂]⁺, isomer I
 E=-3003.593352
 C -0.452201 -1.402372 1.123711
 O -0.546803 -0.599248 2.068368
 O -1.296095 -2.004758 0.429722
 Na 1.153791 -2.319058 -0.740840
 Na -2.219520 -1.124348 -1.568445
 Na 1.636733 0.337105 2.106755
 Na -2.873057 -0.441304 1.975974
 Cl 0.224963 -0.165928 -2.190808
 Cl 3.157849 -1.108406 0.407555
 Na -0.897076 1.990549 -0.742116
 Cl -3.294559 0.878149 -0.237801
 Na 2.611404 0.961557 -1.200236
 Cl 1.318887 2.501877 0.558171

[Na₆Cl₄CO₂]⁺, isomer II
 E=-3003.583755
 Na -2.061854 -1.682726 -0.131847
 C -2.112863 -0.349937 2.246572
 Na -4.362078 0.933528 0.735802
 Cl -3.049358 0.465956 -1.488026
 Na -0.475434 1.024537 -1.892856
 Cl 1.098354 2.303918 -0.148574
 Na 3.652272 2.077335 0.310240
 Cl 3.744351 -0.497882 0.867338
 Na 2.996310 -2.194181 -0.966955

Cl 0.460932 -1.522239 -1.259405
Na 1.010350 -0.119476 1.168230
O -0.996136 -0.749751 1.871207
O -3.253961 -0.634668 1.823683

[Na6Cl4CO2]+, isomer III
E=-3003.570048
Na 4.938458 -0.973320 -0.437668
Cl 4.451643 1.361018 0.437157
Na 1.929157 1.880650 0.101423
Cl -0.593388 2.680844 0.010921
Na -3.150842 3.126900 -0.076890
Cl -3.964937 0.633752 -0.166373
Na -4.310840 -1.929894 0.121947
Cl -1.798272 -2.666947 0.224527
Na 0.779101 -3.007162 0.282940
C 2.191106 -0.844615 -0.500054
Na -1.234226 0.035988 -0.115559
O 2.799135 -1.934049 -0.396483
O 0.048202 -0.509006 -0.133984

[Na7Cl5C2HO3]+, isomer I
E=-3740.218175
C -4.007300 -0.150363 -0.239788
C -2.656046 -0.324351 -0.958457
O -2.386311 -1.482689 -1.370382
O -4.147449 0.654923 0.657869
O -1.902616 0.688415 -1.044498
Na -1.982064 1.345978 1.420722
Na -1.469571 -3.040055 0.036232
Na -0.674071 2.529637 -1.554217
Na -0.150902 -0.789673 -1.908766
Cl -1.226071 -1.212316 1.965297
Cl -0.182575 3.252010 0.915252
Na 1.407440 -1.618865 1.940859
Na 3.260361 -0.933542 -1.073493
Cl 3.425534 -0.003821 1.394778
Cl 1.199051 -2.733878 -0.632018
H -4.821890 -0.826702 -0.553822
Na 2.237797 2.221418 0.572849
Cl 1.688420 1.164237 -1.994879

[Na7Cl5CO2]+, isomer I
E=-3626.243084
C -1.738266 -1.787145 -1.309495
O -2.557945 -1.385756 -0.473969
O -0.812071 -2.621310 -1.312913
Na -2.666423 0.779488 -1.542020
Na -2.431167 -0.583921 1.736541
Na 0.651654 -1.305933 -2.547156
Na 0.543043 -3.079561 0.461877
Cl -2.699143 1.921576 0.830920
Cl 0.017502 1.323214 -2.134973
Na -0.286376 2.820480 0.282098
Na 2.039810 -0.143198 1.980118
Cl 2.173175 2.370086 1.147737
Cl -0.102260 -1.613684 2.541610
Na 2.633226 1.314370 -1.197033
Cl 2.497095 -1.356462 -0.548040

[Na8Cl6C2HO3]+, isomer I
E=-4362.866422

C 4.322437 -1.545678 0.992957
 O 4.965490 -0.532905 0.765100
 C 2.804516 -1.576539 0.758819
 O 2.144037 -0.558643 1.136443
 O 2.407305 -2.564814 0.115689
 Na 1.295815 0.900299 2.632618
 Na 1.252379 -2.433661 -1.782542
 Na 3.465196 1.053828 -0.179573
 Na 0.097001 0.129337 -0.059493
 Cl 1.452728 2.671574 0.588090
 Cl 1.891800 0.206684 -2.265352
 Na -3.284089 -0.526189 -1.844719
 Cl -4.659230 -0.504712 0.432846
 Na -2.651260 -1.970533 1.395535
 Na -3.132804 1.596637 1.067211
 Cl -1.289722 -2.166077 -0.972145
 Cl -1.250883 0.142682 2.423956
 H 4.815208 -2.491371 1.277972
 Na 0.568347 2.601449 -1.990743
 Cl -1.875753 1.744715 -1.357262

[Na8Cl6CO2]+, isomer I

E=-4248.900641
 C -2.218570 1.354793 1.988996
 O -1.992925 0.121098 1.948167
 O -2.945822 2.046664 1.256846
 Na -1.245488 -1.938195 2.438822
 Na -1.770028 3.066631 -0.405084
 Na -3.620404 -0.104523 0.122209
 Na -0.366185 -0.201118 0.031382
 Cl -2.174152 -2.438561 -0.093688
 Cl -2.168186 0.847411 -1.950251
 Na 2.449141 1.261751 -1.754498
 Cl 4.296766 0.437528 -0.018382
 Na 2.552139 1.155657 1.865724
 Na 3.003281 -1.892966 -0.077013
 Cl 0.784647 2.351997 0.141374
 Cl 1.328083 -1.298015 2.001061
 Na -1.339718 -1.683859 -2.581652
 Cl 1.258210 -1.180831 -2.057342

[Na9Cl7C2HO3]+, isomer I

E=-4985.503075
 C 2.102451 1.297842 -3.249169
 C 0.865511 0.453163 -2.920015
 O -0.230477 1.059749 -3.011086
 O 3.067886 1.303072 -2.506342
 O 1.050585 -0.729019 -2.505460
 Na 2.589643 0.018450 -0.569740
 Na 1.051561 -2.823803 -1.654853
 Na -1.408140 2.380604 -1.375910
 Na -1.413465 -0.655926 -1.878953
 Cl 0.030733 0.516191 0.308679
 Cl 2.399061 -2.460732 0.707086
 Na 1.170621 2.737188 1.575820
 Cl -1.185462 3.782289 0.931309
 Na -2.385662 1.402620 1.580043
 Cl -2.465307 -1.052735 2.735105
 Na -0.258551 -1.957370 1.525452
 Na -3.652301 -1.778484 0.475395
 Cl -3.524133 0.627724 -0.851239
 Cl -1.471295 -3.071383 -0.719557

H 2.046507 1.953265 -4.135894
Cl 3.472674 1.314483 1.686995
Na 3.914037 -1.112275 2.482900

[Na9Cl7CO2]+, isomer I
E=-4871.535761
C 0.387587 2.042084 2.307776
O 1.419310 2.703454 2.108536
O 0.226928 0.800610 2.411803
Na -1.640908 0.538110 0.830381
Na 0.265135 -1.383519 3.101631
Na 1.590760 3.544228 -0.022914
Na 2.674968 0.721687 1.690737
Cl -0.650682 2.509554 -1.010498
Cl -1.687630 -2.234312 1.586035
Na -3.082991 1.603494 -1.819560
Cl -1.742294 -0.836930 -1.752654
Na 0.733132 0.428721 -2.263174
Cl 1.748928 -2.047160 -2.584135
Na 0.150467 -2.551259 -0.484259
Na 3.670971 -1.345252 -1.021462
Cl 3.054291 1.359157 -0.881441
Cl 2.300053 -1.902399 1.343332
Cl -4.311025 0.864466 0.388185
Na -3.779111 -1.683065 -0.058597

[Na10Cl8C2HO3]+, isomer I
E=-5608.156315
C -1.345126 3.788885 -1.455676
O -2.539024 3.848015 -1.656315
C -0.592028 2.452279 -1.244977
O 0.597184 2.574065 -0.821743
O -1.181456 1.376323 -1.523428
Na -3.517786 1.928957 -0.790628
Na 2.916554 2.914278 -0.534992
Na -1.485792 -1.058956 -0.971212
Na 1.064167 0.275388 -1.441972
Cl -4.162293 -0.650615 -0.449220
Cl 0.320460 -0.402741 1.161405
Na 0.271623 2.310024 1.690439
Na -2.409866 -0.483227 1.851444
Cl -2.021028 -3.145679 0.978718
Cl -2.396276 2.216183 1.655488
Na 0.725890 -3.122289 0.616704
Cl 3.358789 -2.859875 0.895118
Na 3.202860 -0.197434 1.394867
Na 3.420151 -2.253120 -1.671001
Cl 0.673654 -2.307190 -2.087745
Cl 3.697869 0.459103 -1.355726
H -0.716764 4.696491 -1.412518
Cl 2.886859 2.440150 2.032631
Na -4.438761 -3.247204 -0.006652

[Na10Cl8C2HO3]+, isomer II
E=-5608.139436
C -0.950836 0.349862 -1.370047
O -1.880674 -0.069119 -2.037600
C 0.363951 -0.417569 -1.248603
O 0.982577 -0.202882 -0.178907
O 0.699132 -1.232709 -2.151303
Na -1.449292 -2.505874 -1.901451
Na -3.816595 0.348082 -0.623413

Na 0.131151 -0.382021 1.958948
 Na 2.499247 1.342675 0.377766
 Na 2.199296 -2.119636 0.026436
 Cl 4.328594 -0.549527 -0.990402
 Na 5.337139 -0.805368 1.453140
 Cl -2.633424 -0.150640 1.924662
 Cl -3.993748 -2.365709 -1.036452
 Cl 2.912897 -0.713138 2.386901
 Na -2.949032 -2.777824 1.345323
 Cl -0.338307 -3.017583 0.651212
 H -1.051121 1.239314 -0.732550
 Na 2.552521 0.083984 -2.946352
 Cl 0.354831 2.501936 1.766842
 Na -0.383315 3.599888 -0.660065
 Cl 1.571750 2.413022 -2.079288
 Cl -3.010067 3.001745 -0.367049
 Na -2.312826 2.503624 2.154004

[Na10Cl8C2HO3]+, isomer III
 E=-5608.129879
 C -0.791756 1.047759 -0.969886
 C 0.551305 0.325704 -1.073078
 O 1.095253 0.061773 0.026857
 O 0.969523 0.012580 -2.223493
 O -1.606262 1.031842 -1.883325
 H -1.040206 1.439942 0.023097
 Na -1.983289 0.553737 2.867065
 Na -2.317991 -2.856380 -0.134880
 Na -1.008761 -1.184502 -2.932704
 Cl -3.266550 2.188380 1.166101
 Na -0.953206 3.656614 1.091762
 Cl 0.797846 3.452114 -0.975217
 Cl 0.411928 1.920924 2.739761
 Na 2.424874 1.982146 0.719456
 Cl 4.131114 0.950349 -1.085265
 Na 2.693059 -1.443630 -0.927462
 Na 1.165436 -0.659143 2.195032
 Cl 2.984737 -2.463865 1.624186
 Na 0.791036 -3.784152 1.287186
 Cl -1.248153 -2.028887 2.236228
 Cl 0.156426 -3.081723 -1.252639
 Na -3.865788 1.011443 -1.190484
 Cl -3.569718 -1.514082 -1.990559
 Na 2.332369 1.930700 -2.701312

[Na10Cl8CO2]+, isomer I
 E=-5494.185759
 C 0.257184 2.321598 1.685020
 O -0.942303 2.423948 1.371513
 O 0.955406 1.336739 2.011941
 Na 2.965619 2.555470 1.296289
 Na -3.279815 2.524685 0.986050
 Na 1.688369 -0.894789 1.084416
 Na -1.126594 -0.031307 1.709831
 Cl 4.199374 0.202549 0.758253
 Cl -0.127716 0.004491 -0.936280
 Na -0.443522 2.782024 -1.130804
 Na 2.560355 0.392595 -1.621726
 Cl 2.628700 -2.392142 -1.187495
 Cl 2.220438 3.032016 -1.195929
 Na -0.111864 -2.808242 -1.050784
 Cl -2.724234 -2.856601 -1.539322

Na -2.958822 -0.157570 -1.472876
 Na -3.048902 -2.810603 1.076251
 Cl -0.354555 -2.598922 1.758656
 Cl -3.723291 -0.148289 1.282747
 Cl -3.032441 2.561107 -1.613686
 Na 5.017483 -2.160140 -0.125354

[Na11Cl9C2HO3]+, isomer I
 E=6230.809015
 Na -2.250871 1.011881 0.247600
 Cl 0.079683 0.095492 -1.153206
 Cl -1.287908 0.687851 2.794162
 Na -2.116039 -1.856698 2.443628
 Cl 0.273199 -2.851073 1.488574
 Na -0.340492 3.199054 2.645126
 Cl 2.048670 2.499176 1.762613
 Cl -3.840218 1.586685 -1.951079
 Na -2.376011 -0.550598 -2.352986
 Na -3.251311 4.073479 -1.623874
 C -2.852618 -2.421580 -0.362921
 Cl -1.303738 3.715349 0.126599
 Na 1.119813 -0.166019 1.575687
 Cl 3.688142 -0.862631 0.783581
 Na 4.340288 1.739861 0.514559
 Cl 3.440722 2.069412 -1.970986
 Na 1.116167 2.705399 -0.830867
 Na 2.819304 -0.553992 -1.864409
 Cl 2.305480 -3.206865 -2.088693
 Na -0.109728 -2.750989 -1.212570
 Na 2.812892 -3.381057 0.530124
 C -3.447045 -3.608340 0.425695
 O -2.410670 -2.706363 -1.513755
 O -2.828377 -1.294195 0.192248
 O -3.468194 -3.625171 1.640484
 H -3.816753 -4.455928 -0.177253

[Na11Cl9C2HO3]+, isomer II
 E=6230.776607
 C -0.263914 -0.317683 -0.489939
 O -1.459302 -0.056667 -0.526871
 C 0.730161 0.338408 -1.473349
 O 1.946133 0.154576 -1.209663
 O 0.248066 1.030271 -2.409540
 Na -1.371994 -0.631480 -3.153885
 Na -3.449286 1.622275 -0.879233
 Na 3.878877 1.405060 -1.686332
 Na 2.854878 -2.057146 -1.211063
 Na -3.648008 -1.427352 -0.029243
 Na 2.864716 0.196490 1.041433
 Na -1.913602 0.627204 1.882250
 Cl -4.475143 0.667550 1.366374
 Na 0.379215 3.176274 -1.569580
 Cl 2.583082 3.059790 0.039797
 Cl 5.006335 -0.591962 -0.501655
 Na 0.733261 3.359411 1.990278
 Cl -1.432764 3.101789 0.386436
 Cl -3.895086 -0.447976 -2.572837
 Cl 2.095087 -2.579423 1.390683
 Na 0.446643 -1.848501 3.368017
 Cl -1.786885 -2.298289 1.892174
 Cl 0.709941 0.775332 2.763371
 Cl 0.255088 -2.649741 -2.160959

Na -0.125653 -3.679031 0.283544
H 0.169702 -0.955233 0.292748

[Na11Cl9C2HO3]+, isomer III
E=-6230.759845
C -0.931809 -0.470608 -0.273661
C 0.196842 -1.446883 0.084271
O -0.159814 -2.579859 0.488136
O -0.700821 0.712533 -0.460480
O 1.366976 -1.013696 -0.090548
Na 2.825875 -1.228033 1.659915
Na -1.733800 -1.815691 2.302193
Na -1.076586 0.668199 -3.009060
Cl -0.329678 -2.011275 -2.880198
Na 2.184037 -2.390385 -1.725708
Cl 4.425604 -2.542754 -0.269609
Na 5.728903 -0.292335 -0.163831
Cl -2.962153 0.762892 2.072201
Na -0.628813 1.803048 3.083292
Cl 0.564653 -0.534283 3.136687
Na -2.279126 2.407345 -0.049500
Cl -0.019676 3.557793 1.056745
Na 1.040475 4.629146 -1.134298
Cl 0.932948 2.505790 -2.617701
Na 1.527500 1.383173 -0.118312
Cl 3.976255 1.228474 0.988955
H -1.955206 -0.871483 -0.337685
Na -4.595641 -0.354638 0.128232
Cl -3.678243 -2.806133 0.618551
Cl -3.593216 1.046999 -1.952447
Na -1.726106 -3.455550 -1.029920

[Na11Cl9CO2]+, isomer I
E=-6116.829379
Na -2.187086 0.179653 -0.012034
Cl 0.281985 0.013243 -1.111496
Cl -1.289679 0.147863 2.610566
Na -1.273528 -2.534683 2.431277
Cl 1.228636 -2.861621 1.653465
Na -0.900424 2.795533 2.378151
Cl 1.619677 2.771083 1.602923
C -5.566817 0.326405 -0.897971
Na -4.724589 -2.221623 -1.156521
Na -4.301179 2.669463 -1.280699
Cl -2.384045 -2.668403 -0.074399
Cl -1.990051 3.041614 -0.130821
Na 1.372077 -0.043996 1.514937
Cl 4.008357 -0.232658 0.607575
Na 4.011104 2.436431 0.307958
Cl 3.090953 2.544631 -2.170782
Na 0.684212 2.736244 -1.017879
Na 2.943741 -0.180250 -2.010171
Cl 2.706149 -2.901221 -2.123332
Na 0.299220 -2.739383 -0.966715
Na 3.636606 -2.879998 0.353153
O -4.316115 0.365065 -1.112491
O -6.356945 -0.599662 -0.793892