## Theoretical Study of Radiative and Nonradiative Decay Rates for Cu(I) Complexes with Double Heteroleptic Ligands

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**Fig. S1.** Absorption(a) and emission(b) spectra obtained by TDDFT method with different functionals for **3**, together with experimentally (unit: nm).



**Fig. S2.** Dependence of  $E(S_1)$  of **3** on the HF% in TD-DFT functionals.



Fig. S3. *Dushin* program to deal with changes in the electronic state process of reorganization energy diagram.



Fig. S4. Super imposed structures of  $S_0$  (green) and  $T_1$  (red) states of the studied complexes.



Fig. S5. Spin density plot for the lowest  $T_1$  state for the complexes.







**Fig. S6.** The transition dipole moment vector of the contribution from N^N ligand, P^P ligand and the whole complex is described by the blue, orange and purple arrows, respectively.





**Fig. S7.** Calculated reorganization energies and the displacement vectors for the normal modes with the largest reorganization energies over 100 cm<sup>-1</sup> at high-frequency (inset).







**Fig. S8.** Calculated reorganization energies and the displacement vectors for the normal modes with the reorganization energies over 200 cm<sup>-1</sup> at low-frequency(inset).

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	B3LYP	B3PW91	PBE0	M06	M06-2X	Expt.
Cu-N <sub>1</sub>	2.2870	2.2620	2.2196	2.2216	2.2582	2.2275
Cu-N <sub>2</sub>	2.2388	2.2220	2.2349	2.2116	2.2056	2.1372
Cu-P <sub>1</sub>	2.4094	2.3828	2.3649	2.3297	2.3700	2.2851
Cu-P <sub>2</sub>	2.3932	2.3693	2.3327	2.2768	2.4497	2.2319
N <sub>1</sub> -Cu-P <sub>1</sub>	109.25	109.13	104.93	102.35	103.24	101.95
N <sub>1</sub> -Cu-P2	110.44	110.39	114.13	114.15	116.13	112.77
N <sub>2</sub> -Cu-P <sub>1</sub>	106.02	105.68	100.60	97.84	94.15	98.24
N <sub>2</sub> -Cu-P <sub>2</sub>	114.56	114.41	116.80	121.28	129.71	120.26
N <sub>1</sub> -Cu-N <sub>2</sub>	80.37	80.80	81.11	81.35	81.33	82.58
P <sub>1</sub> -Cu-P <sub>2</sub>	126.61	126.88	128.64	128.85	122.97	130.05
DHA <sub>1</sub>	1.15	1.33	0.90	2.34	0.74	0.62

**Table S1** Partial optimized geometric structural parameters of **3** in the ground state by B3LYP, B3PW91, PBE0, M06 and M06-2X functionals, together with the observed experimentally.

**Table S2** The maximum absorption spectra  $\lambda_{max(nm)}$  (oscillator strengths), emission spectra and experimental values of **3** calculated by different functionals.

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	B3LYP	B3PW91	PBE0	M06	M06-2X	Expt
	$\lambda_{ m max}/f$	$\lambda_{ m max}/f$	$\lambda_{ m max}/f$	$\lambda_{ m max}/f$	$\lambda_{\max}/f$	$\lambda_{max}$
$S_0 \rightarrow S_1$	458.2/ 0.0892	463.3/ 0.0930	437.0/ 0.0975	433.7/ 0.0964	339.8/ 0.1490	421
$T_1 \rightarrow S_0$	684.6	698.4	748.8	719.2	426.9	631

**Table S3** Calculated  $E(S_1)$ /ev using various functionals and LANL2DZ/6–31G(d) basis set in

				DCM s	olution for	<b>: 3</b> .				
	BLYP	MPWLYP1M	TPSSH	B3LYP*	B3LYP	PBE0	M06	MPW1B95	BMK	M062X
HF%	0	5	10	15	20	25	27	31	42	54
$E(\mathbf{S}_1)$	1.9841	2.1605	2.3735	2.4763	2.7063	2.8500	2.9104	3.0813	3.4524	3.7458

	PB	E0	M	)6
	$\lambda^{Abs}_{max}/Expt$	$\lambda^{Em}_{max}/Expt$	$\lambda^{Abs}_{max}/Expt$	$\lambda^{Em}_{max}\!/\!Expt$
1	$430.2/420^{a}$	834.2/646ª	429.2 /420ª	851.5/646 <sup>a</sup>
2	437.6/425ª	811.9/648 <sup>a</sup>	$435.1/425^{a}$	$742.8/648^{a}$
3	437.0/421ª	$705.1/625^{a}$	433.7/421ª	719.2/625ª

M06 for all, together with experimental values (unit: nm).

Table S4 Absorption and emission values obtained by TD-DFT method with PBE0 and

Table S5 Negative value of the HOMO (- $\epsilon_{HOMO}$ ) and LUMO (- $\epsilon_{LUMO}$ ) energies, HOMO-LUMO

gaps calculated by DFT, and the lowest singlet energies  $(E_{S1})$  calculated by TDDFT in eV for

	-E <sub>HOMO</sub>	-E <sub>LUMO</sub>	$\Delta E_{ ext{H-L}}$	$E_{S1}$
1	6.44	2.76	3.68	2.88
2	6.40	2.76	3.64	2.83
3	6.41	2.81	3.60	2.84

these complexes.

Table S6 Molecular orbital compositions of Cu at the optimized S<sub>0</sub> geometries in solvent DCM.

	МО	Energy(eV)	Cu contribution (%)
1	LUMO	-2.76	$2.1(0.90d_{xz}+0.42d_z^2)$
	НОМО	-6.44	$34.0(19.77 d_{xz}+4.25 d_z^2+0.86 d_x^2 - y^2)$
	HOMO-1	-6.79	$42.2(4.78  \text{d}_{xz} + 0.95  \text{d}_{yz} + 2.85  \text{d}_{xy} + 3.03  \text{d}_{z}^{2} + 21.45  \text{d}_{x}^{2}_{-y}^{2})$
	HOMO-2	-7.01	50.7 (38.61 $d_{yz}$ +6.31 $d_{xy}$ +0.77 $d_{xz}$ +2.09 $d_x^2 {}_{-y}^2$ )
	HOMO-3	-7.08	$27.4(24.64d_{yz}+0.16d_{xz}+0.25d_{z}^{2}+0.77d_{x}^{2}_{-y}^{2})$
	HOMO-4	-7.29	$2.3(0.45d_{xz}+0.20d_{yz})$
2	LUMO	-2.76	2.0 (0.53d <sub>xz</sub> )
	НОМО	-6.40	$33.3(11.07 d_{xz} + 10.15 d_{z}^{2} + 2.52 d_{x}^{2} g^{2})$
	HOMO-1	-6.80	$41.8(5.38 d_{yz}+2.64 d_{xy}+9.63 d_{xz}+13.21 d_{z}^{2}+1.86 d_{x-y}^{2^{2}})$
	НОМО-2	-6.93	11.7 (7.54 $d_{xy}$ +2.04 $d_{xz}$ + 1.37 $d_z^2$ )
	НОМО-3	-7.03	$51.5(3.91  \text{d}_{yz} + 40.23  \text{d}_{xy} + 3.49 \text{d}_{xz} + 0.74  \text{d}_{z}^{2} + 0.60 \text{d}_{x - y}^{2^{2}})$
	HOMO-4	-7.14	$21.2(0.66 \text{ d}_{yz}+13.87 \text{ d}_{xy}+1.69 \text{ d}_{xz}+1.06 \text{ d}_{z}^{2}+0.65 \text{ d}_{x}^{2}_{-y}^{2})$

3	LUMO+1	-1.66	0.2(0.09d <sub>yz</sub> )
	LUMO	-2.81	1.4(0.61d <sub>xz</sub> )
	НОМО	-6.41	$31.9(10.68d_{xz}+10.37 d_z^2+1.48d_x^2)$
	HOMO-1	-6.67	58.2 $(18.35d_z^2+9.85 d_{xz}+3.89d_x^2)^2$
	HOMO-2	-6.95	$5.0(2.95 d_{yz} + 1.83 d_{yz})$
	HOMO-3	-7.04	$3.4(1.84d_{x}^{2}-y^{2})$
	HOMO-4	-7.23	$66.4(4.92  \text{d}_{yz} + 1.15 \text{d}_{xz} + 58.03 \text{d}_{xy} + 0.05 \text{d}_{z}^{2})$

Table S7 Molecular orbital compositions of Cu at the optimized  $S_0$  geometries in gas phase.

МО	Energy(eV)	Cu contribution (%)
LUMO	-4.66	$2.1(0.96d_{xz}+0.44d_{z}^{2})$
НОМО	-8.21	$32.6(18.55 d_{xz}+3.80 d_z^2+0.62 d_{x-y}^2^2)$
HOMO-1	-8.58	$39.39(3.43  d_{xz} + 2.34  d_{xy} + 3.05  d_{z}^{2} + 20.58  d_{x}^{2} _{-y}^{2})$
HOMO-2	-8.86	$55.6 \ (43.22 d_{yz} + 6.57 d_{xy} + 1.29 d_{xz} + 1.47 d_{x - y}^{2})$
HOMO-3	-8.94	21.0(18.89d <sub>yz</sub> )
HOMO-4	-9.09	1.54(0.77d <sub>xz</sub> )
LUMO	-4.47	2.0 (0.54d <sub>xz</sub> )
НОМО	-8.07	$33.5(10.63  \text{d}_{xz} + 1.27 \text{d}_{yz} + 2.46 \text{d}_{z}^{2} + 9.80 \text{d}_{x}^{2} \text{-}_{y}^{2})$
HOMO-1	-8.47	$40.5(5.25d_{yz}+2.26d_{xy}+9.54d_{xz}+12.49d_{z}^{2}+1.80d_{x-y}^{2}^{2})$
HOMO-2	-8.61	$9.63(5.94d_{xy}+1.69d_{xz}+1.29d_{z}^{2})$
НОМО-3	-8.73	$46.73(3.37d_{yz} + 36.81d_{xy} + 3.06d_{xz})$
HOMO-4	-8.82	$28.7(19.97d_{xy}+2.39d_{xz}+1.36d_{z}^{2})$
LUMO+1	-3.36	$0.21(0.09d_{yz})$
LUMO	-4.51	$1.4(0.34d_z^2)$
НОМО	-8.07	$31.3(10.53d_{xz}+0.63d_{xy}+9.72d_z^2+1.51d_x^2_{-y}^2)$
HOMO-1	-8.35	$41.1(17.98d_z^2+9.18 d_{xz}+1.23d_{yz}+3.65d_{x-y}^{2-2})$
НОМО-2	-8.63	4.7(2.70d <sub>xy</sub> +1.80 d <sub>yz</sub> )
HOMO-3	-8.72	$2.9(0.13d_{xy}+0.12d_{x-y}^{2}^{2})$
HOMO-4	-8.92	$64.9(4.70 \text{ d}_{yz}+1.05 \text{ d}_{xz}+56.97 \text{ d}_{xy})$
	MO         LUMO         HOMO-1         HOMO-2         HOMO-3         HOMO-4         LUMO         HOMO-4         LUMO         HOMO-4         LUMO         HOMO-1         HOMO-1         HOMO-3         HOMO-4         LUMO         HOMO-4         HOMO-4         LUMO         HOMO-4         LUMO         HOMO-4         LUMO         HOMO-4         LUMO         HOMO-3         HOMO-1         HOMO-1         HOMO-3         HOMO-1         HOMO-3         HOMO-4	MOEnergy(eV)LUMO-4.66HOMO-8.21HOMO-1-8.58HOMO-2-8.86HOMO-3-8.94HOMO-4-9.09LUMO-4.47HOMO<1-8.07HOMO-1-8.47HOMO-2-8.61HOMO-3-8.73HOMO-4-8.82LUMO-4.51HOMO-1-3.36LUMO-4.51HOMO<1-8.07HOMO-1-8.35HOMO-1-8.35HOMO-2-8.63HOMO-3-8.72HOMO-4-8.72HOMO-4-8.92

	МО	Energy(eV)	Cu contribution (%)
1	LUMO	-3.10	$3.8(0.96 d_z^2 + 0.87 d_{yz})$
	НОМО	-5.80	36.1(19.78 d <sub>xy</sub> +9. 85d <sub>xz</sub> )
	HOMO-1	-6.76	$41.3(1.23d_{z}^{2}+2.18d_{xy}+7.27 d_{xz}+2.20d_{yz}+10.91d_{x}^{2}{_{-y}^{2}})$
	HOMO-2	-6.82	53.5 $(15.60d_z^2 + 14.37 d_{xz} + 9.80 d_x^2 - y^2)$
	HOMO-5	-7.41	$9.5(1.79d_z^2+2.80d_{xz}+2.18d_{xy}+2.20d_{yz})$
2	LUMO	-3.12	3.7(2.37 d <sub>xz</sub> )
	НОМО	-5.83	$36.7(25.39 d_{xy} + 5.55 d_{xz})$
	HOMO-1	-6.68	$36.0(7.64d_{z}^{2}+1.35d_{xy}+0.85~d_{xz}+17.23d_{yz})$
	HOMO-2	-6.76	58.4(33.26 d <sub>xy</sub> +19.69 d <sub>xz</sub> )
	HOMO-5	-7.42	$0.2(0.06 d_{x^{2}-y}^{2}+0.04 d_{xz})$
3	LUMO+1	-1.82	0.3(0.14d <sub>yz</sub> )
	LUMO	-3.09	$2.3(1.48d_{xz})$
	НОМО	-6.16	$30.9 (15.39 d_{xz} + 6.64 d_{xy} + 2.68 d_z^2)$
	HOMO-1	-6.63	$39.3(11.58d_{yz}+1.37d_{xz}+16.75d_{z}^{2}+2.21d_{x-y}^{2})$
	HOMO-2	-6.81	$5.6 (0.05 d_z^2)$
	HOMO-5	-7.33	$0.1(3.17d_{yz}+3.11d_z^2)$

**Table S8** Molecular orbital compositions of Cu at the optimized  $T_1$  geometries in solvent DCM.

**Table S9** Frontier molecular orbital energies(eV) and compositions (%) of different fragments in the ground state for the complexes.

МО	E/eV	MC	compositio	Assign		
		Cu	N^N	P^P		
1						
LUMO+5	-0.95	5.5	4.3	90.2	$\pi^*(P^P)$	
LUMO+4	-1.09	3.3	1.8	95.0	<b>π*</b> (P^P)	
LUMO+3	-1.15	3.2	2.6	94.2	<b>π*</b> (P^P)	
LUMO+2	-1.42	1.6	94.5	3.9	$\pi^{*}(N^{N})$	
LUMO+1	-1.71	0.3	96.6	3.1	$\pi^{*}(N^{N})$	
LUMO	-2.76	2.1	94.2	3.7	$\pi^{*}(N^{N})$	
HOMO	-6.44	34.0	5.6	60.4	$d(Cu)+\pi(P^P)$	

HOMO-1	-6.79	42.2	11.7	46.1	$d(Cu)+\pi(N^N+P^P)$
HOMO-2	-7.01	50.7	43.8	5.6	$d(Cu)+\pi(N^N)$
HOMO-3	-7.08	27.4	67.2	5.4	$d(Cu)+\pi(N^N)$
HOMO-4	-7.29	2.3	72.	25.7	$\pi(N^N+P^P)$
HOMO-5	-7.42	0.5	91.1	8.5	π(N^N)
2					
LUMO+5	-0.951	5.5	5.1	89.4	$\pi^*(P^P)$
LUMO+4	-1.05	1.9	14.8	83.3	$\pi^*(N^N)+\pi^*(P^P)$
LUMO+3	-1.17	2.1	6.0	92.0	<b>π*</b> ( <b>P</b> ^ <b>P</b> )
LUMO+2	-1.44	1.8	94.0	4.2	$\pi^*(N^N)$
LUMO+1	-1.86	0.5	97.7	1.7	π*( N^N)
LUMO	-2.76	2.0	95.5	2.5	$\pi^*(N^N)$
НОМО	-6.40	34.1	9.7	56.1	$d(Cu)+\pi(P^P)$
HOMO-1	-6.79	41.8	16.3	41.9	$d(Cu)+\pi(N^N+P^P)$
HOMO-2	-6.93	11.7	83.9	4.3	$d(Cu)+\pi(N^N)$
HOMO-3	-7.03	51.5	44.7	3.8	$d(Cu)+\pi(N^N)$
HOMO-4	-7.14	21.2	66.6	12.2	$d(Cu)+\pi(N^N+P^P)$
HOMO-5	-7.38	0.4	95.1	4.5	$\pi(N^N)$
3					
LUMO+5	-0.95	5.8	3.4	90.9	$\pi^*(P^P)$
LUMO+4	-1.05	2.0	2.9	95.1	$\pi^*(P^P)$
LUMO+3	-1.15	1.4	2.9	95.7	$\pi^*(P^P)$
LUMO+2	-1.30	1.9	92.5	5.6	<b>π*</b> ( N^N)
LUMO+1	-1.66	0.2	98.6	1.2	$\pi^*(N^N)$
LUMO	-2.81	1.4	96.2	2.4	$\pi^*(N^N)$
НОМО	-6.41	31.9	10.5	57.6	$d(Cu)+\pi(N^N+P^P)$
HOMO-1	-6.67	42.3	14.1	43.6	$d(Cu)+\pi(N^N+P^P)$
HOMO-2	-6.95	5.0	92.9	2.1	$\pi(N^N)$
HOMO-3	-7.04	3.4	84.0	12.6	$\pi$ (N^N+P^P)
HOMO-4	-7.23	66.4	27.2	6.3	$d(Cu)+\pi(N^N)$
HOMO-5	-7.27	4.6	92.5	2.9	π(N^N)

MO	E/eV	MC	compositio	Assign	
		Cu	N^N	P^P	
1					
LUMO+5	-0.92	5.0	4.1	90.8	$\pi^*(P^P)$
LUMO+4	-1.02	1.6	1.5	97.0	$\pi^*(P^P)$
LUMO+3	-1.12	5.5	3.9	90.7	$\pi^*(P^P)$
LUMO+2	-1.45	1.5	91.8	6.8	$\pi^*(N^N)$
LUMO+1	-1.78	0.6	96.3	3.1	$\pi^*(N^N)$
LUMO	-3.10	3.8	90.9	5.3	$\pi^*(N^N)$
HOMO	-5.80	36.1	14.4	49.5	$d(Cu)+\pi(N^N+P^P)$
HOMO-1	-6.76	41.3	15.7	43.0	$d(Cu)+\pi(N^N+P^P)$
HOMO-2	-6.83	53.5	25.7	20.8	$d(Cu)+\pi(N^N+P^P)$
HOMO-3	-7.00	9.0	78.6	12.4	$\pi(N^N+P^P)$
HOMO-4	-7.29	7.2	63.3	29.5	$\pi(N^N+P^P)$
HOMO-5	-7.41	9.5	47.4	43.1	$\pi(N^N+P^P)$
2					
LUMO+5	-0.96	3.6	13.9	82.5	<b>π*</b> ( N^N)+ <b>π*</b> (P^P)
LUMO+4	-0.99	2.6	20.3	77.1	<b>π*</b> ( N^N)+ <b>π*</b> (P^P)
LUMO+3	-1.10	5.9	6.5	87.6	<b>π*</b> ( P^P)
LUMO+2	-1.51	1.6	94.0	4.4	<b>π*</b> ( N^N)
LUMO+1	-1.91	0.5	97.8	1.7	<b>π*</b> ( N^N)
LUMO	-3.12	3.7	92.2	4.1	$\pi^*(N^N)$
HOMO	-5.83	36.7	16.5	46.8	$d(Cu)+\pi(N^N+P^P)$
HOMO-1	-6.68	36.0	17.3	46.7	$d(Cu)+\pi(N^N+P^P)$
HOMO-2	-6.76	58.4	28.3	13.3	$d(Cu)+\pi(N^N+P^P)$
HOMO-3	-6.88	7.8	83.5	8.7	π(N^N)
HOMO-4	-7.13	7.2	75.7	17.1	$\pi(N^N+P^P)$
HOMO-5	-7.42	0.2	95.8	4.1	$\pi(P^{P})$
3					
LUMO+5	-0.94	2.5	2.7	94.9	$\pi^*(P^P)$
LUMO+4	-1.02	1.8	3.9	94.3	$\pi^*(P^P)$
LUMO+3	-1.04	3.3	4.1	92.6	$\pi^*(P^P)$

**Table S10** Frontier molecular orbital energies (eV) and compositions (%) of different fragments in the triplet state for the complexes.

LUMO+2	-1.33	2.7	91.6	5.8	<b>π*</b> ( N^N)
LUMO+1	-1.82	0.3	98.9	0.8	π*(N^N)
LUMO	-3.09	2.3	94.5	3.2	$\pi^*(N^N)$
HOMO	-6.16	30.9	15.1	54.0	$d(Cu)+\pi(N^N+P^P)$
HOMO-1	-6.63	39.3	13.5	47.2	$d(Cu)+\pi(N^N+P^P)$
HOMO-2	-6.81	5.6	85.3	9.0	$\pi(N^N)$
HOMO-3	-7.01	11.3	82.0	6.7	$d(Cu)+\pi(N^N)$
HOMO-4	-7.05	65.7	26.3	8.0	$d(Cu)+\pi(N^N)$
HOMO-5	-7.33	0.1	99.5	0.5	$\pi(N^N)$

**Table S11** SOC matrix elements  $< T_1^{\alpha}|H_{SOC}|S_m>(cm^{-1})$ , the transition dipole moment  $M_{S_{m,j}}$  of **1** calculated at the  $T_1^{opt}$  optimized geometry with DCM included. The radiative decay rate constant  $(k_r/s^{-1})$  is also given.

$E(\mathbf{T}_1)=1$	1988cm <sup>-1</sup>								
S <sub>m</sub>	$M_{S_{m,j}}$	M <sub>Sm,j</sub>	M <sub>Sm,j</sub>	$< T_1^x  H_{SOC} S_m>$		$< T_1^{y}  H_{SOC} S_m>$		$< T_1^{z}  H_{SOC} S_m>$	
				Re	Im	Re	Im	Im	
$\mathbf{S}_1$	0.0008	-0.6330	-0.4067	-8.0561	-30.8048	-8.0561	30.8048	8.8035	
$S_2$	0.0773	0.5922	0.5199	127.6256	-8.6501	127.6256	8.6501	-55.9092	
$S_3$	0.1912	-0.5392	-0.0193	57.2251	-58.3964	57. 2251	58.3964	166.6064	
$\mathbf{S}_4$	-0.6639	-0.2856	0.1479	0.8508	-8.9057	0.8508	8.9057	-3.7694	
$S_5$	-0.6324	-0.3100	0.3045	10.0211	-88.8470	10.0211	88.8470	-104.9409	
$S_6$	0.0419	0.4374	0.0835	4.5139	9.8495	4.5139	-9.8495	15.8400	
$S_7$	-0.7361	0.0936	0.4877	89.0185	174.5850	89.0185	-174.5850	426.6527	
$S_8$	0.1658	0.2371	-0.0366	76.1690	-11.8938	76.1690	11.8938	-0.8565	
$S_9$	-0.1707	-0.2073	0.1563	0.6071	-52.4358	0.6071	52.4358	4.0796	
$S_{10}$	1.2138	0.9098	-0.7204	-16.4388	-9.8422	-16.4388	9.8422	6.4022	
$k_m^x$				598	5.70				
$k_m^{y}$		5985.70							
$k_m^{z}$								2568.82	
k <sub>r</sub>		4846.74							

a  $M_{S_{m,j}}$  is the *j*-axis  $(j \in x, y, z)$  ojection of the  $S_m \rightarrow S_0$  transition dipole moment.

b the superscript  $\alpha$  denotes the spin sub-level (x, y, or z) of the T<sub>1</sub> excited state.

$$c k_r(T_1 \to S_0) = \frac{1}{3} \sum_{\alpha} k_r^{\alpha}$$

d ( $T_1^{\alpha}$ ) is the energy shift (cm<sup>-1</sup>) of the spin sub-level  $\alpha$  ( $\alpha = x, y, \text{ or } z$ ) due to SOC with the Sm excited state.  $E(T_1)$  is taken as the reference point.

**Table S12** SOC matrix elements  $< T_1^{\alpha}|H_{SOC}|S_m>(cm^{-1})$ , the transition dipole moment  $M_{S_{m,j}}$  of **2** calculated at the  $T_1^{opt}$  optimized geometry with DCM included. The radiative decay rate constant

 $(k_r/s^{-1})$  is also given.

 $E(T_1) = 12371 \text{ cm}^{-1}$ 

$S_{m}$	$M_{S_{m,j}}$	$M_{S_{m,j}}$	$M_{S_{m,j}}$	$< T_1^x  H_{SOC} S_m>$		$< T_1^{y}  H_{SOC} S_m>$		$< T_1^{z}  H_{SOC} S_m>$	
				Re	Im	Re	Im	Im	
$\mathbf{S}_1$	-1.0561	0.0438	0.0486	32.0066	4.1489	32.0066	-4.1489	2.7563	
$S_2$	1.2269	0.0339	0.0725	15.8802	-136.1803	15.8802	136.1803	41.6169	
$S_3$	0.5371	-0.1055	-0.3296	105.6034	21.3574	105.6034	-21.3574	-81.2957	
$\mathbf{S}_4$	0.1742	-0.1292	-0.2380	-2.4138	2.1208	-2.4138	-2.1208	21.2989	
$S_5$	0.0610	-0.7402	0.0191	-39.1828	-79.6429	-39.1828	79.6429	-174.1596	
$S_6$	-1.2169	-0.0655	-0.0502	-2.4743	-1.1103	-2.4743	1.1103	-2.4205	
$S_7$	0.4284	-1.2015	0.2444	42.5899	42.7375	42.5899	-42.7375	514.7130	
$S_8$	0.1043	1.0373	-0.2440	37.2695	-22.2640	37.2695	22.2640	9.3032	
<b>S</b> <sub>9</sub>	-0.0062	1.3071	-0.2028	-9.1107	-20.0546	-9.1107	20.0546	60.8538	
$S_{10}$	0.1307	1.2340	-0.1321	-0.4287	-10.5200	-0.4287	10.5200	-9.6280	
$k_m^x$	4908.30								
$k_m^{y}$	4908.30								
$k_m^{z}$								9514.32	
$k_r$		6443.64							

$E(T_1) = 14182 \text{ cm}^{-1}$								
S <sub>m</sub>	M <sub>Sm,j</sub>	M <sub>Sm,j</sub>	M <sub>Sm,j</sub>	$< T_1^x  H_{SOC} S_m>$		$< T_1^{y}  H_{SOC} S_m>$		$< T_1^z  H_{SOC} S_m>$
				Re	Im	Re	Im	Im
$\mathbf{S}_1$	-1.5978	0.0092	-0.1665	88.6776	-49.0399	88.6776	49.0399	-7.9374
$S_2$	0.3655	-0.1326	-0.3594	12.2639	114.9570	12.2639	-144.9570	-52.7289
$S_3$	0.4827	-0.0437	0.0686	114.9070	-61.8399	114.9070	61.8399	18.5976
$\mathbf{S}_4$	0.7062	0.4484	-0.0807	65.3256	-24.8984	65.3256	24.8984	-198.1577
$S_5$	1.0978	-0.4620	0.1104	12.0904	-165.7141	12.0904	165.7141	-365.5163
$S_6$	0.0497	2.1554	-0.4749	-2.4853	-5.6882	-2.4853	5.6882	8.3283
$S_7$	0.3610	-1.4467	0.3576	-2.1243	-4.1736	-2.1243	4.1736	-1.5673
$S_8$	0.0026	-0.0263	0.0119	-12.9428	14.8570	-12.9428	-14.8570	-23.4307
$S_9$	0.0550	-0.4595	0.0008	-14.0689	14.1006	-14.0689	-14.1006	11.3669
$\mathbf{S}_{10}$	0.0697	-0.2017	0.1314	-9.9454	-19.9932	-9.9454	19.9932	50.9251
$k_m^{x}$				656	5.84			
$k_m^{y}$						656	5.84	
$k_m^{z}$								35154.51
$k_r$						1609	5.39	

**Table S13** SOC matrix elements  $< T_1^{\alpha}|H_{SOC}|S_m>(cm^{-1})$ , the transition dipole moment  $M_{S_{m,j}}$  of **3** calculated at the  $T_1^{opt}$  optimized geometry with DCM included. The radiative decay rate constant

 $(k_r/s^{-1})$  is also given.