## **Supporting Information**

## The catalytic activity and mechanism of oxygen reduction

## reaction on P-doped MoS<sub>2</sub>

Xiaoming Zhang<sup>1,2</sup>, Shaodong Shi<sup>1</sup>, Tianwei Gu, Leyi Li<sup>1</sup>, Shansheng Yu<sup>1\*</sup>

<sup>1</sup>State Key Laboratory of Automotive Simulation and Control, Department of Materials Science, Jilin University, Changchun 130012, China

<sup>2</sup>Division of Fuel Cell & Battery, Dalian National Laboratory for Clean Energy, Dalian Institution of Chemical Physics, Chinese Academy of Sciences, Dalian 116023, China

**Keywords**: MoS<sub>2</sub>, single-atom dope, Phosphorus, DFT, ORR mechanism \* Corresponding author: Shansheng Yu, E-mail: yuss@jlu.edu.cn

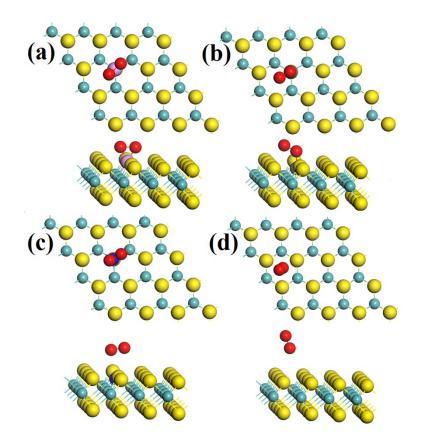


Figure S1 The different kinds of possible adsorption geometries of O<sub>2</sub>.

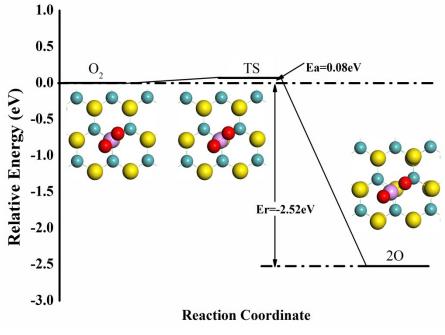


Figure S2 The dissociation step of the oxygen molecule.

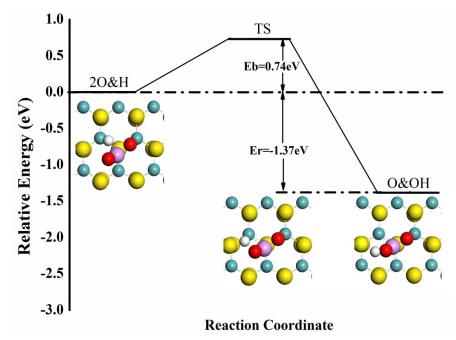
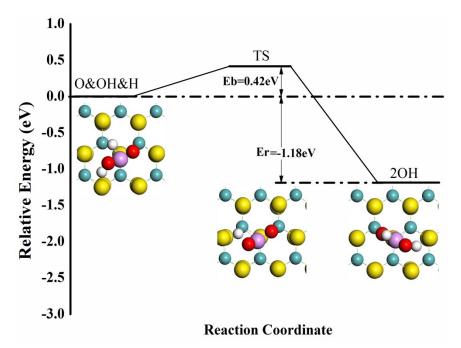
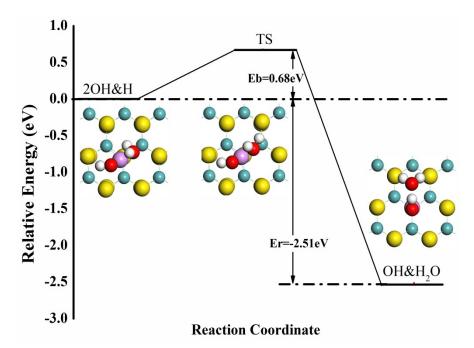


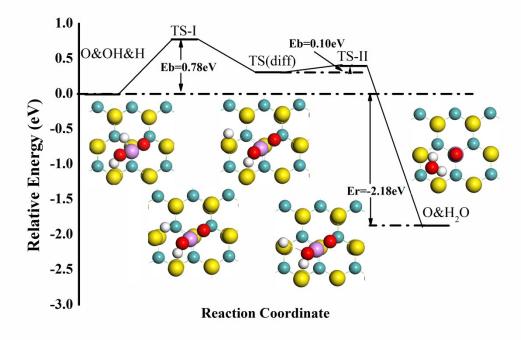
Figure S3 The first hydrogenation step of two atomic O.



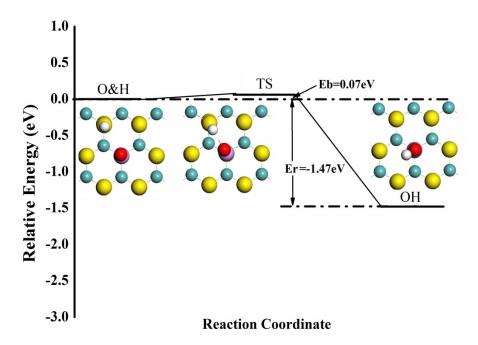
**Figure S4** The second hydrogenation step of two atomic O and the formation of two OH species.



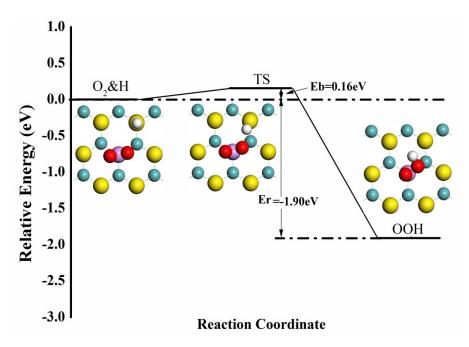
**Figure S5** The hydrogenation process of two OH species and the formation of the  $H_2O$  from the hydrogenation of the OH, and the release of the  $H_2O$ .



**Figure S6** The diffusion of the H atom from the P-Mo bridge site to the S atom near to P and the formation of the  $H_2O$  from the hydrogenation of the OH, and the release of the  $H_2O$ .



**Figure S7** The hydrogenation process of the adsorbed O and the formation of the OH species.



**Figure S8** The hydrogenation step of the oxygen molecule and the formation of the OOH species.

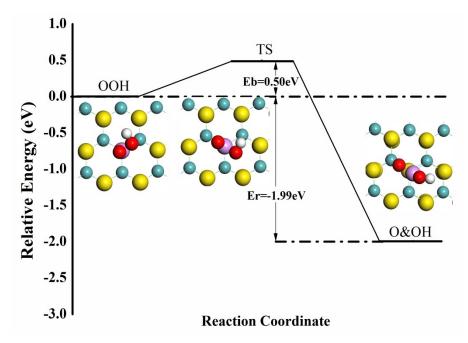


Figure S9 The dissociation of the OOH species into atomic O and OH.