

Electronic Supplementary Information (ESI) for
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On the Structure of Superbasic $(\text{MgO})_n$ sites solvated in a Faujasite Zeolite

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S1. Experimental Section.

ATR-FTIR spectroscopy. Infrared spectra in Attenuated Total Reflection (ATR-IR) were collected on loose powder on a Bruker Alpha spectrophotometer (2 cm^{-1} resolution, average on 256 scans), equipped with an internal reflection element in diamond and placed in the glove box. The spectrum intensity was corrected for the effective thickness value for the different incident wavelengths.

S2. MgOHY synthesis: FTIR spectra at intermediate temperatures.

The ATR-IR spectrum of the HY zeolite impregnated with $\text{Mg}(\text{NO}_3)_2 \cdot 6 \text{ H}_2\text{O}$ after activation at 80°C in vacuum is reported in Figure S1 (dark blue line). It is evident from the comparison with the HY spectrum (black line in Figure S1) the presence of additional bands at about 1350 cm^{-1} associated with nitrate group (see Figure S2 for the spectrum of the pure salt). The comparison of these bands with the ones present in the bulk $\text{Mg}(\text{NO}_3)_2 \cdot 6 \text{ H}_2\text{O}$ activated at the same temperature (blue line in Figure S2) suggests a different environments around the NO_3^{2-} groups in the two cases. The thermal decomposition of $\text{Mg}(\text{NO}_3)_2 \cdot 6 \text{ H}_2\text{O}$ after the impregnation of HY was followed by IR spectroscopy (see Figure S1). The comparison with the spectra recorded for the pure $\text{Mg}(\text{NO}_3)_2 \cdot 6 \text{ H}_2\text{O}$ after a treatment at the same temperatures (Figure S2), evidences that the nitrate encapsulated in the HY zeolite is significantly less stable. In fact, the bands corresponding to the nitrate start to decrease in intensity already at 200°C, whereas for the bulk compound no decomposition is observed for $T < 400^\circ\text{C}$.

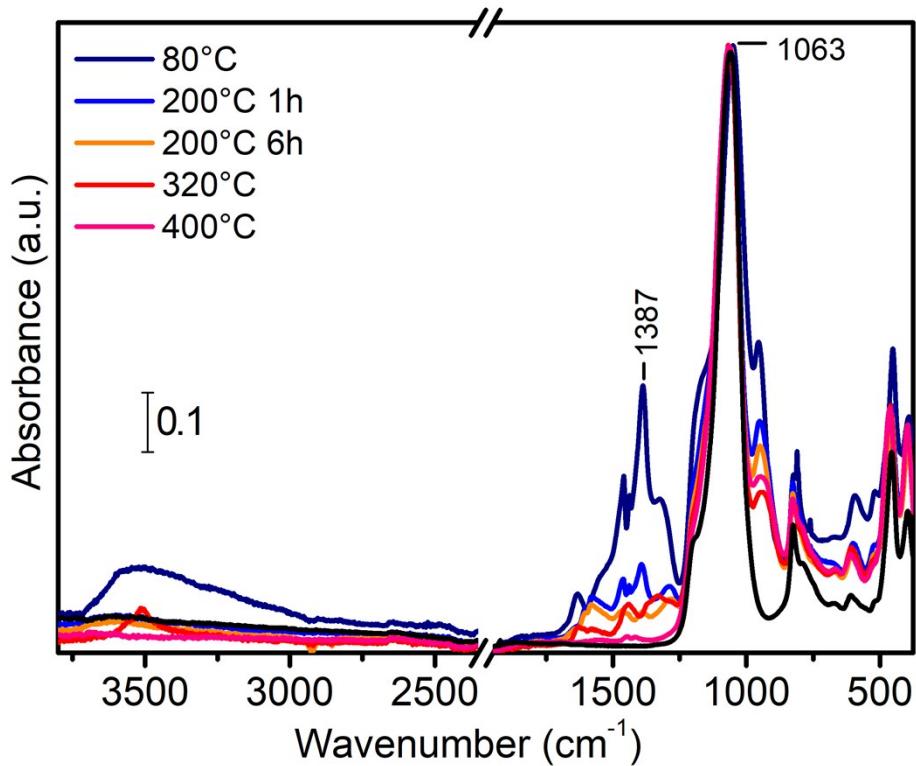


Figure S1. ATR-FTIR spectra of HY impregnated with $\text{Mg}(\text{NO}_3)_2 \cdot 6 \text{ H}_2\text{O}$ after activation in vacuum at increasing temperatures from 80°C to 400°C . At 200°C , the spectra obtained after 1 h (light blue line) and 6 h (orange line) are also reported. All the spectra have been recorded in a glove box and normalized at the intensity of the zeolite band at 1063 cm^{-1} . Spectrum of pristine HY activated at 400°C in vacuum is also reported (black curve).

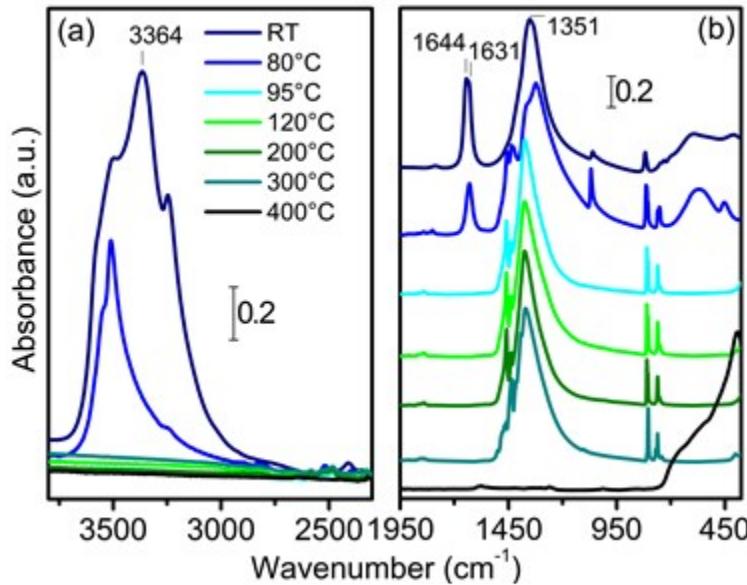


Figure S2. ATR-FTIR spectra of $\text{Mg}(\text{NO}_3)_2 \cdot 6 \text{ H}_2\text{O}$ after activation in vacuum at increasing temperatures from 25°C to 400°C recorded in a glove box. The spectra in the RT - 300°C range have been normalized at the intensity of the band at 1351 cm^{-1} , whereas the spectrum recorded after the 400°C treatment has been normalized with respect to the intensity at 400 cm^{-1} .

S3. MgOHY synthesis: SEM images

Low resolution SEM images of the samples show that no appreciable change in gross sample morphology or particle aggregation is observed upon MgO addition (b and c). Under identical synthesis conditions, MgO adopts a disordered but recognizably different morphology (a)

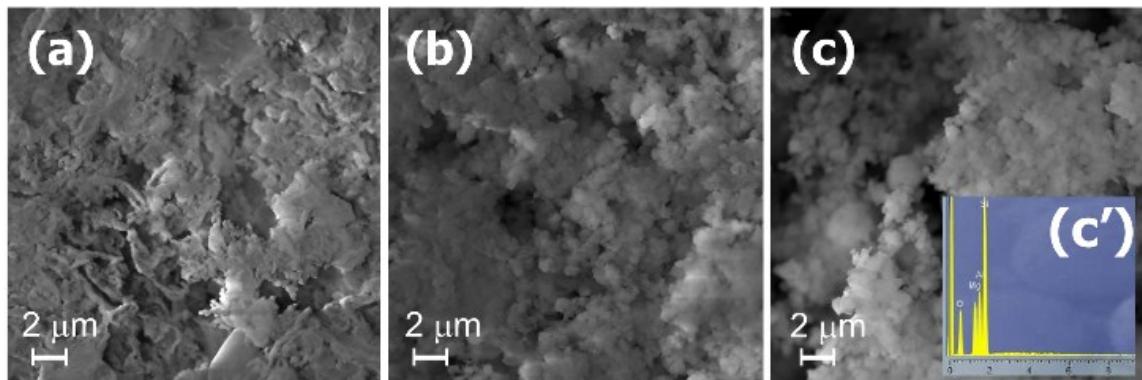


Figure S3. SEM images (secondary electrons signal) obtained for (a) MgO, (b) HY and (c) MgO/HY. In (c') one exemplificative EDS spectrum obtained for MgO/HY sample is reported.

S4. Additional XRPD data.

The patterns recorded for the reference materials (MgO , $\text{Mg}(\text{OH})_2$ and $\text{Mg}(\text{NO}_3)_2 \cdot 6 \text{ H}_2\text{O}$) are reported in Figure S4. Besides the MgO samples (synthesized as reported in the Experimental Section of the main text), the two other samples were used as received from Sigma-Aldrich.

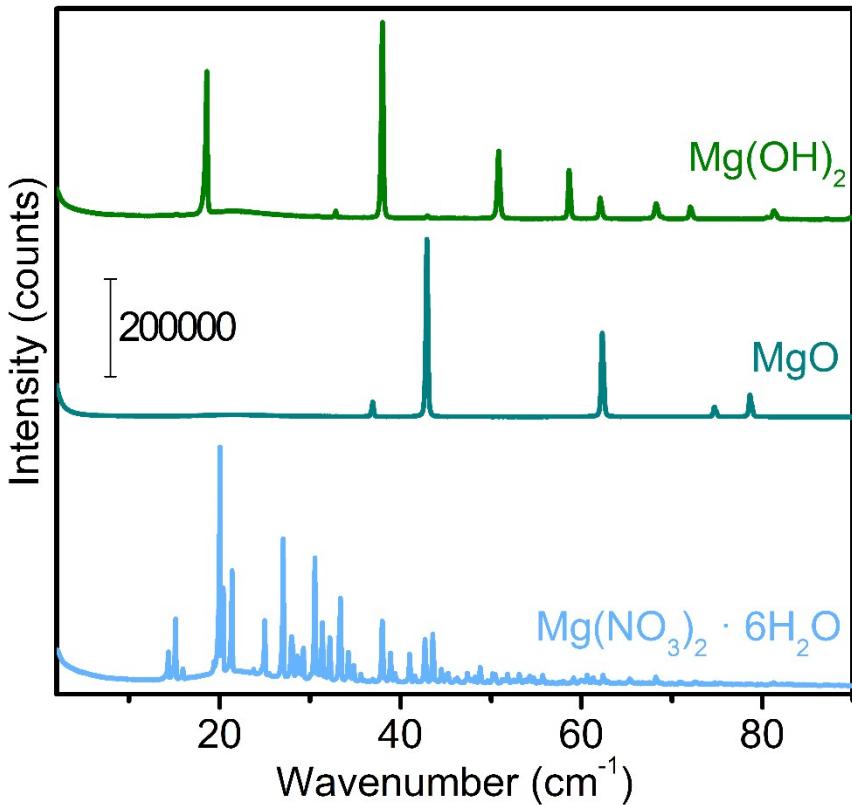


Figure S4. XRPD pattern of $\text{Mg}(\text{NO}_3)_2 \cdot 6 \text{H}_2\text{O}$, Mg(OH)_2 and MgO recorded in air.

In Figure S5 it is reported a comparison between the pattern recorded for HY zeolite before (grey curve) and after the synthetic procedure to insert the MgO functionalities (blue curve). The two patterns are almost indistinguishable. It is evident from the comparison with the curves reported in Figure S4 that in the blue pattern are absent clear peaks associated to Mg(OH)_2 or MgO , suggesting a subnanometric dimensions of these species in the zeolite channels.

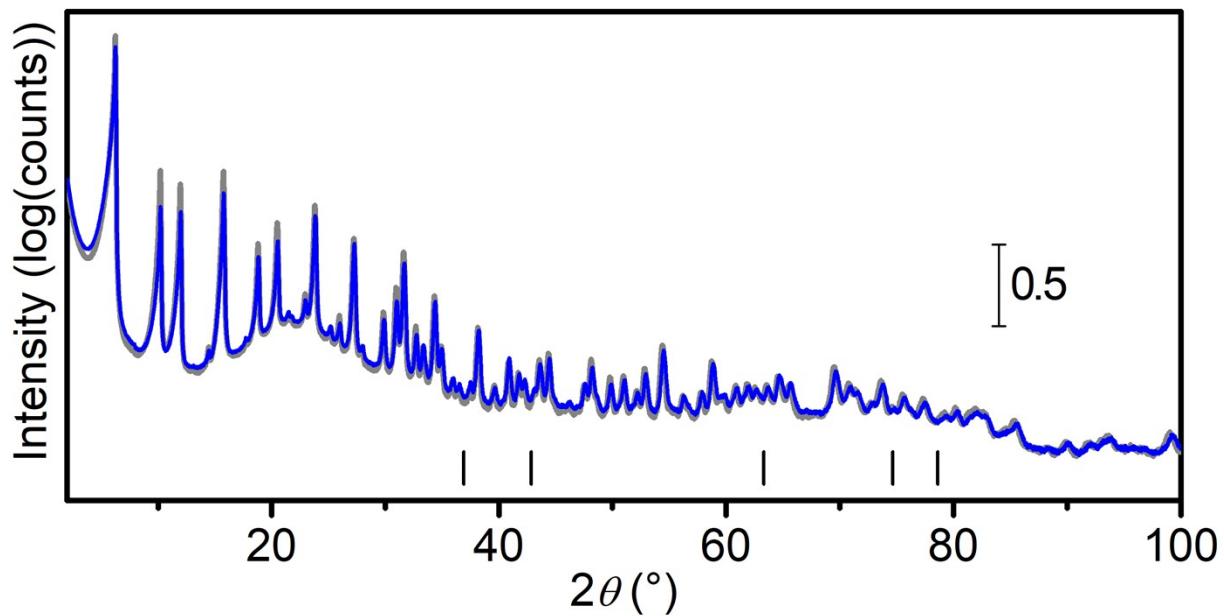


Figure S5. XRPD patterns of the HY (grey curve) and MgOHY (blue curve) samples in inert atmosphere after activation at 400°C overnight.

CIF file as obtained by Rietveld refinement of the MgOHY pattern.

The refined structures of the Y zeolite without and with MgO were deposited within the COD database (<http://www.crystallography.net/cod/>) with entry numbers 3000174 and 3000175.

S5. Modeling Results

The coordinates of the optimized structures displayed in Figs. 6, 7 and 8 are listed below (in CIF Format). Cut and paste to an ASCII text file with .cif extension for visualization with any molecular graphics program.

CIF file relative to the structure reported in Fig. 6a.

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CIF file relative to the structure reported in Fig. 6b.

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CIF file relative to the structure reported in Fig.7a

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CIF file relative to the structure reported in Fig.7b

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048	O	0.101319248589	0.372763992517	0.141197665831
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051	O	0.114374625223	0.155792025836	0.333701338801
052	O	0.106036371023	0.360212489936	0.856289854758
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A17	Al	0.107711858551	0.787750863999	0.604433678898
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A111	Al	0.567268927525	0.157889863450	0.774540860517
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Si4	Si	0.819657210565	0.038936394884	0.206296244211
Si5	Si	0.171183660939	0.394184508132	0.862870592880
Si6	Si	0.999879463656	0.860571428400	0.383210152548
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Si9	Si	0.115324541621	0.786088523625	0.956272617510
Si10	Si	0.750321413798	0.154358447321	0.601447156078
Si11	Si	0.933745651345	0.609943908635	0.134968273500
Si12	Si	0.571676253814	0.974368424463	0.776830136281
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Si14	Si	0.923565055114	0.155477011876	0.780847590136
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Si21	Si	0.933764901805	0.612771488097	0.774814057764
Si22	Si	0.759049028635	0.616154157324	0.126226217778
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Si29	Si	0.005258768369	0.864755794873	0.205113026560
Si30	Si	0.355016885127	0.220632164858	0.849756841929
Si31	Si	0.823889215248	0.400295902831	0.201627109634
Si32	Si	0.828664403476	0.038906479080	0.376311006501
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CIF file relative to the structure reported in Fig.7c

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H2 H 0.683700211592 0.271945807653 0.479070254897
H3 H 0.244585642018 0.719275064851 0.477192276610
H4 H 0.366900022124 0.009400518101 0.082653002274
H5 H 0.792032137059 0.808903167677 0.683479812776
H6 H 0.952928807164 0.737495213134 0.265090552973
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H12 H 0.867393451895 0.633829896783 0.967986145092
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O6 O 0.739983824103 0.216989133428 0.477804510993
O7 O 0.470269072615 0.991831274609 0.765419809569
O8 O 0.759507171053 0.999469026999 0.187686641268
O9 O 0.169236376630 0.012429582467 0.486038909798
O10 O 0.177063660425 0.494119900616 0.775525311534
O11 O 0.482487172159 0.791908842157 0.191092187334
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Si9	Si	0.129036176082	0.769765296036	0.947021640851
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Si15	Si	0.941339796943	0.778907125324	0.580996444350
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CIF file relative to the structure reported in Fig. 8a

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CIF file relative to the structure reported in Fig. 8b

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Si6	Si	0.017214952289	0.858152237964	0.372402249066
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CIF file relative to the structure reported in Fig. 8c

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H2 H 0.684058906527 0.272508172249 0.487617919411
H3 H 0.238479877275 0.717792766489 0.486713160862
H4 H 0.363829318491 0.001346328855 0.089502408086
H5 H 0.791357791354 0.813896751377 0.698599718627
H6 H 0.986998009700 0.722026946508 0.255984362978
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CIF file relative to the structure reported in Fig. 8d

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_atom_site_fract_z
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H2 H 0.682417858709 0.277074358697 0.477354369917
H3 H 0.244205998151 0.724091199471 0.475253688542
H4 H 0.364562909907 0.014273188469 0.081480611525
H5 H 0.790827244729 0.813542654086 0.681400462988
H6 H 0.949993684109 0.743285641012 0.264416828233
H7 H 0.142602485970 0.194374664392 0.264994133817
H8 H 0.958558870211 0.003763128377 0.596072294656
H9 H 0.577253619096 0.308216691182 0.072346082108
H10 H 0.425724599962 0.592691802047 0.061473506730
H11 H 0.214185607260 0.025492105786 0.708907488107
H12 H 0.867146340199 0.636779365125 0.966882562613

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O7	O	0.469939031812	0.996790133791	0.765087867662
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