

Supporting Information

SO₂ Absorption in EmimCl-TEG Deep Eutectic Solvents

*Dezhong Yang**^a, *Shaoze Zhang*^{b,c}, *De-en Jiang**^c, *Sheng Dai**^{d,e}

^a School of Science, China University of Geosciences, Haidian District, Beijing 100083, China

^b Key Laboratory for Advanced Materials and School of Chemistry & Molecular Engineering, East China University of Science and Technology, Shanghai 200237, China.

^c Department of Chemistry, University of California, Riverside, California 92521, United States

^d Chemical Sciences Division, Oak Ridge National Laboratory, Oak Ridge, Tennessee 37831, United States

^e Department of Chemistry, University of Tennessee, Knoxville, Tennessee 37996, United States.

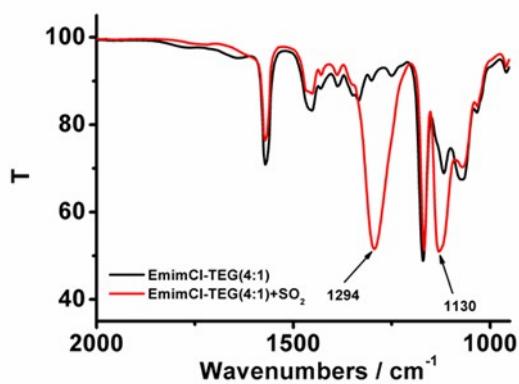


Fig. S1 FTIR spectra of EmimCl-TEG(4:1) before and after SO₂ absorption.

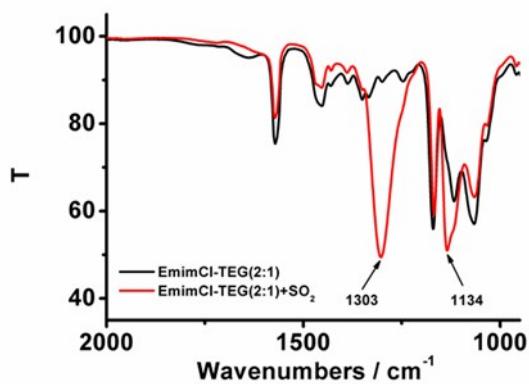


Fig. S2 FTIR spectra of EmimCl-TEG(2:1) before and after SO₂ absorption.

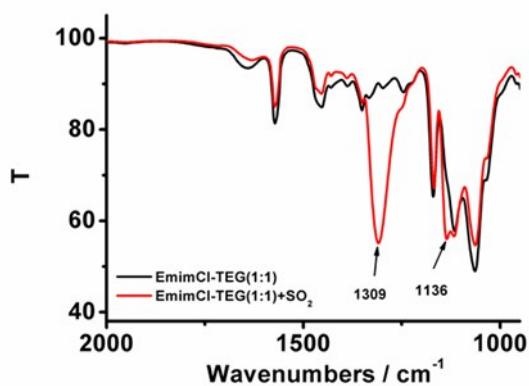


Fig. S3 FTIR spectra of EmimCl-TEG(1:1) before and after SO₂ absorption.

Table S1 The FTIR wavenumbers of EmimCl, TEG and EmimCl-TEG DESs

	O-H	C _{4,5} -H	C ₂ -H	C-O-C	C-OH
EmimCl	—	3139	3038	—	—
EmimCl-TEG(6:1)	3254	3142	3049	1119	1080
EmimCl-TEG(4:1)	3264	3144	3058	1118	1069
EmimCl-TEG(2:1)	3282	3146	3070	1117	1066
EmimCl-TEG(1:1)	3321	3151	3094	1116	1064
TEG	3395	—	—	1116	1059