Direct measurement of Ni incorporation into Fe₃O₄(001)

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Supplementary Information

Normal incidence X-ray standing wave (NIXSW) XP spectra analysis

One complication in the NIXSW XP spectra data analysis is that the Ni 2p and Fe 3s photoemission peaks are energetically almost coincident. To remove the influence of the bulk peak a template background was constructed. First XP spectra were measured from a clean SCV reconstructed surface over the binding energy range corresponding to the Ni 2p peak, at the same photon energies as the Ni NIXSW measurement. These spectra were summed to produce an integrated spectrum that was then fitted with a Shirley background¹. The XP Ni 2p spectra subsequently recorded for each NIXSW measurement were then fitted utilising this Shirley background-subtracted integrated spectrum (IS) and the Shirley background (SB) itself, in addition to the peaks arising from the presence of the Ni. Both the IS and SB components were scaled by separate linear corrections whose gradient and offset were parameters in the fitting of the XP Ni 2p spectra. In part, this fitting procedure is depicted in Figures S1 and S2.

This convoluted analysis, to remove the contribution from the Fe 3*s*, was due to a subtle difference observed in the shape of the background as a function of the photon energy (relative to the Bragg energy), that we attribute to differing absorption profiles of the elastic and inelastic contributions in the background.



Figure S1 – Depicted is the Ni 2p core level with no background subtracted. The large background peak is the bulk Fe 3s core level peak. The Shirley background for this Fe 3s peak, which was subtracted from the clean background template to give the IS and SB templates as discussed in the text, is depicted in cyan. The Nisub, Niad and satellite fitted peaks are depicted in blue, orange and red respectively. Figure S2 gives an enlarged view of this spectrum in an energy range encompassing only the Ni 2p3/2 doublet.





Figure S2 – An enlarged view of Figure S1, only showing the Ni $2p_{3/2}$ doublet. No background has been subtracted from this spectrum. The Nisub, Niad and satellite fitted peaks are depicted in blue, orange and red respectively.

Soft X-ray photoelectron spectroscopy (SXPS)

Ni $2p_{3/2}$ SXP spectra (hv = 1.1 keV), without background subtraction of the Fe 3s SXP spectra, are shown in Figure S1 for Ni deposited onto Fe₃O₄(001) with the substrate held at room temperature and after annealing to 425 and 875 K. Ni $2p_{3/2}$ SXP spectra for Ni deposited onto Fe₃O₄(001) with the substrate held at 150 K, with and without background subtraction, are shown in Figures S2 and S3 (respectively). Figure S2 also shows the Ni $2p_{3/2}$ SXP spectra for Ni deposited onto Fe₃O₄(001) with the substrate held at room temperature, for comparison.



Figure S3 – Raw Ni $2p_{3/2}$ SXP spectra (1.1 keV incident photon energy) for as deposited Ni (blue) and deposited Ni annealed to 425 K and 875 K (red and black respectively). No background has been subtracted from these spectra and they have not been normalised.



Figure S4 – Ni $2p_{3/2}$ SXP spectra for Ni deposited at 300 K and 150 K. The species present at 150 K is assigned to a subsurface species, as confirmed by the NIXSW results, which exists in conjunction with an adatom species at

300 K. The sole occupation of the subsurface sites by the Ni at 150 K is possibly due to dissociative water adsorption and the subsequent loss of the SCV surface reconstruction.



Figure S5 – Raw soft Ni $2p_{3/2}$ XP spectrum for the Ni metal deposited at 150 K. No background has been subtracted from this spectrum.

Density functional theory (DFT)

Computational details

All the theoretical calculations were performed using the Vienna *ab initio* Simulation Package (VASP)^{2,3} using the Projector Augmented Wave (PAW) approach^{4,5} with a basis set cut-off energy of 550 eV. The hybrid functional HSE⁶ utilised the standard mixing factor 25%, and screening length (0.207^{-1} Å⁻¹). The k-mesh of 2 x 2 x 1 was optimised such that it delivers total energy with an accuracy of better than 1 meV, while reducing the k-mesh by a factor of 2 at PBE level leads to a change in the total energy of only 1 meV. In all cases, structures were relaxed until forces were smaller than 0.02 eV/Å.

The surface calculations utilised an asymmetric surface slab (i.e. a slab with a relaxed surface on only a single side), resulting in a significantly cheaper calculation, with 9 fixed layers and 4 relaxed layers with the subsurface cation vacancy reconstruction⁷. Due to large size of the unit-cell (~100 atoms) the adoption of a symmetric setup would be computationally prohibitive for HSE-type calculations. The vacuum gap (separation between adjacent supercells perpendicular to the surface) was set to 14 Å. The resulting relaxed structure for a Ag adatom on Fe₃O₄(001) and a clean Fe₃O₄(001) surface are shown in Tables S1 and S2 (respectively).

layer	atom	х	У	Z	Δz
adatom	Ag	1.055	5.261	5.084	-
	01	-0.302	6.604	4.175	-0.00
	0 2	0.142	2.032	4.220	-0.03
	O 3	1.969	0.045	4.191	-0.06
	O 4	2.399	3.904	4.175	-0.00
	O 5	4.181	2.115	4.087	-0.09
1st Oct	O 6	4.250	6.174	4.191	-0.06
layer	07	6.238	4.348	4.220	-0.03
	O 8	6.321	-0.025	4.087	-0.09
	Fe 1	-0.118	0.145	3.963	-0.25
	Fe 2	2.212	1.939	3.973	-0.24
	Fe 3	4.350	4.088	3.963	-0.25
	Fe 4	6.144	6.417	3.973	-0.24
1st Tet	Fe 1	7.453	3.248	3.282	+0.12
layer	Fe 2	5.258	1.053	3.006	-0.16
(Interstitial)	Fe 3	3.057	7.263	3.225	
	01	0.096	-0.114	2.095	+0.03
	02	0.137	4.342	2.190	+0.05
	O 3	1.989	2.188	2.096	-0.05
	O 4	1.996	6.202	2.021	-0.05
2nd Oct	O 5	4.091	4.301	2.095	-0.05
layer	O 6	4.103	-0.102	2.020	-0.05
	07	6.393	6.194	2.096	-0.05
	O 8	6.419	2.214	2.041	-0.03
	Fe 1	-0.083	6.342	2.085	-0.02
	Fe 2	2.136	4.122	2.085	-0.02
2nd Tet	Fe 1	1.026	1.049	1.033	-0.02
layer	Fe 2	5.254	5.231	1.033	-0.02
	01	-0.038	2.141	-0.038	
	02	0.038	6.270	0.038	
	03	2.065	4.243	0.038	
	04	2.141	-0.038	-0.038	
	O 5	4.167	6.346	-0.038	
3rd Oct	O 6	4.243	2.065	0.038	
layer	07	6.270	0.038	0.038	
	08	6.346	4.167	-0.038	
	Fe 1	0.000	4.205	0.000	
	Fe 2	2.103	6.308	0.000	
	Fe 3	4.205	0.000	0.000	
	Fe 4	6.308	2.103	0.000	
3rd Tet	Fe 1	3.154	3.154	-1.051	
layer	Fe 2	7.359	7.359	-1.051	

Table S1 – Coordinates for the first three octahedrally (Oct) and tetrahedrally (Tet) coordinated layers of the SCV reconstructed surface with an adsorbed Ag adatom, in Å. z = 0 is defined as the 3rd bulk octahedrally coordinated layer. The difference from bulk z positions are shown in the final column (Δz).

layer	atom	х	У	Z	Δz
	01	-0.248	6.561	4.068	-0.11
	0 2	0.133	2.029	4.166	-0.09
	O 3	1.978	0.079	4.168	-0.08
	O 4	2.355	3.958	4.068	-0.11
	05	4.177	2.137	4.091	-0.08
1st Oct	O 6	4.285	6.183	4.168	-0.08
layer	07	6.234	4.338	4.166	-0.09
	O 8	6.343	-0.029	4.091	-0.08
	Fe 1	-0.086	0.116	3.975	-0.24
	Fe 2	2.196	1.992	3.974	-0.24
	Fe 3	4.322	4.119	3.975	-0.24
	Fe 4	6.197	6.401	3.974	-0.24
1st Tet	Fe 1	5.258	1.053	3.032	-0.13
layer	Fe 2	7.457	3.251	3.254	+0.09
(Interstitial)	Fe 3	3.060	7.265	3.255	-
	01	0.093	4.298	1.996	-0.07
	02	0.109	-0.102	2.076	-0.07
	O 3	2.003	2.208	2.074	-0.07
	04	2.012	6.217	1.997	-0.07
2nd Oct	05	4.103	4.314	2.076	-0.07
layer	O 6	4.107	-0.098	2.014	-0.05
	07	6.408	2.202	2.016	-0.05
	08	6.413	6.209	2.074	-0.07
	Fe 1	-0.052	6.364	2.167	+0.06
	Fe 2	2.159	4.154	2.167	+0.06
2nd Tet	Fe 1	1.051	1.056	1.056	0.00
layer	Fe 2	5.261	5.257	1.056	0.00
3rd Oct layer	01	-0.038	2.141	-0.038	
	02	0.038	6.270	0.038	
	O 3	2.065	4.243	0.038	
	04	2.141	-0.038	-0.038	
	05	4.167	6.346	-0.038	
	O 6	4.243	2.065	0.038	
	07	6.270	0.038	0.038	
	08	6.346	4.167	-0.038	
	Fe 1	0.000	4.205	0.000	
	Fe 2	2.103	6.308	0.000	
	Fe 3	4.205	0.000	0.000	
	Fe 4	6.308	2.103	0.000	
3rd Tet	Fe 1	3.154	3.154	-1.051	
layer	Fe 2	7.359	7.359	-1.051	

Table S2 – Coordinates for the first three octahedrally (Oct) and tetrahedrally (Tet) coordinated layers of the SCV reconstructed surface, in Å. z = 0 is defined as the 3rd bulk octahedrally coordinated layer. The difference from bulk z positions are shown in the final column (Δz).

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