## **Supporting Information**

## Probing the $\pi \rightarrow \pi^*$ photoisomerization mechanism of *trans*-azobenzene by multi-state *ab initio* on-the-fly trajectory dynamics simulation

Chao Xu,<sup>*a*</sup> Le Yu,<sup>*b*</sup> Feng Long Gu<sup>\**a*</sup> and Chaoyuan Zhu<sup>\**acd*</sup>

<sup>*a*</sup>Key Laboratory of Theoretical Chemistry of Environment, Ministry of Education, School of Chemistry & Environment of South China Normal University, Guangzhou 51006, P. R. China. E-mail: gu@scnu.edu.cn

<sup>b</sup>Key Laboratory of Synthetic and Natural Functional Molecule Chemistry of Ministry of Education, College of Chemistry & Materials Science and Shaanxi Key Laboratory of Physico-Inorganic Chemistry, Northwest University, Xi'an 710069, China.

<sup>c</sup>Department of Applied Chemistry and Institute of Molecular Science, National Chiao-Tung University, Hsinchu 30010, Taiwan. E-mail: cyzhu@mail.nctu.edu.tw

<sup>d</sup>Center for Emergent Functional Matter Science, National Chiao Tung University, Hsinchu 30010, Taiwan



**Fig. S1** Typical resonance trajectory starts from Franck-Condon region on  $S_2$  state of *trans*-azobenzene and ends in Franck-Condon region of  $S_1$  state. Upper panel is for evolution of potential energy profile and lower panel is for evolution of two NNC bond angles, CNNC and four NNCC dihedral angles.



Fig. S2 Typical resonance trajectory starts from Franck-Condon region on  $S_2$  state of *trans*-azobenzene and ends in *rot*  $S_2$  state. Upper panel is for evolution of potential energy profile and lower panel is for evolution of two NNC bond angles, CNNC and four NNCC dihedral angles.



**Fig. S3** The *trans*-1 case (starting from  $S_3$  state of *trans*-azobenzene) with upper panels for evolution of potential energy profiles and lower panels for evolution of two NNC bond angles, CNNC and four NNCC dihedral angles. The left (right) panels represent one typical fast (slow) nonreactive trajectory.



Fig. S4 The *trans*-2 case (starting from  $S_3$  state of *trans*-azobenzene) with upper panels for evolution of potential energy profiles and lower panels for evolution of two NNC bond angles, CNNC and four NNCC dihedral angles. The left (right) panels represent one typical reactive (nonreactive) trajectory.



Fig. S5 The *trans*-3 case (starting from  $S_3$  state of *trans*-azobenzene) with upper panels for evolution of potential energy profiles and lower panels for evolution of two NNC bond angles, CNNC and four NNCC dihedral angles. The left (right) panels represent one typical reactive (nonreactive) trajectory.



**Fig. S6** The *trans*-4 case (starting from  $S_3$  state of *trans*-azobenzene) with upper panels for evolution of potential energy profiles and lower panels for evolution of two NNC bond angles, CNNC and four NNCC dihedral angles. The left (right) panels represent one typical reactive (nonreactive) trajectory.



**Fig. S7** The *trans-5* case (starting from  $S_3$  state of *trans*-azobenzene) with upper panels for evolution of potential energy profiles and lower panels for evolution of two NNC bond angles, CNNC and four NNCC dihedral angles. The left (right) panels represent one typical reactive (nonreactive) trajectory.



**Fig. S8** Typical resonance trajectory starts from Franck-Condon region on  $S_3$  state of *trans*-azobenzene and ends in Franck-Condon region of  $S_1$  state. Upper panel is for evolution of potential energy profile and lower panel is for evolution of two NNC bond angles, CNNC and four NNCC dihedral angles.



**Fig. S9** Typical resonance trajectory starts from Franck-Condon region on  $S_3$  state of *trans*-azobenzene and ends in *rot*  $S_2$  state. Upper panel is for evolution of potential energy profile and lower panel is for evolution of two NNC bond angles, CNNC and four NNCC dihedral angles.