

Supporting information

Re-entrant supramolecular interactions in inverse melting α -cyclodextrin·4-methylpyridine·water mixtures: an NMR study

Reut Shapira, Yael S. Balazs, Shifi Kababya, Rachel Edrei and Yoav Eichen*a

*Schulich Faculty of Chemistry, Technion – Israel Institute of Technology, Technion City 3200008 Haifa
Israel. E-mail: yoav@ch.technion.ac.il*

Supporting Figures and tables

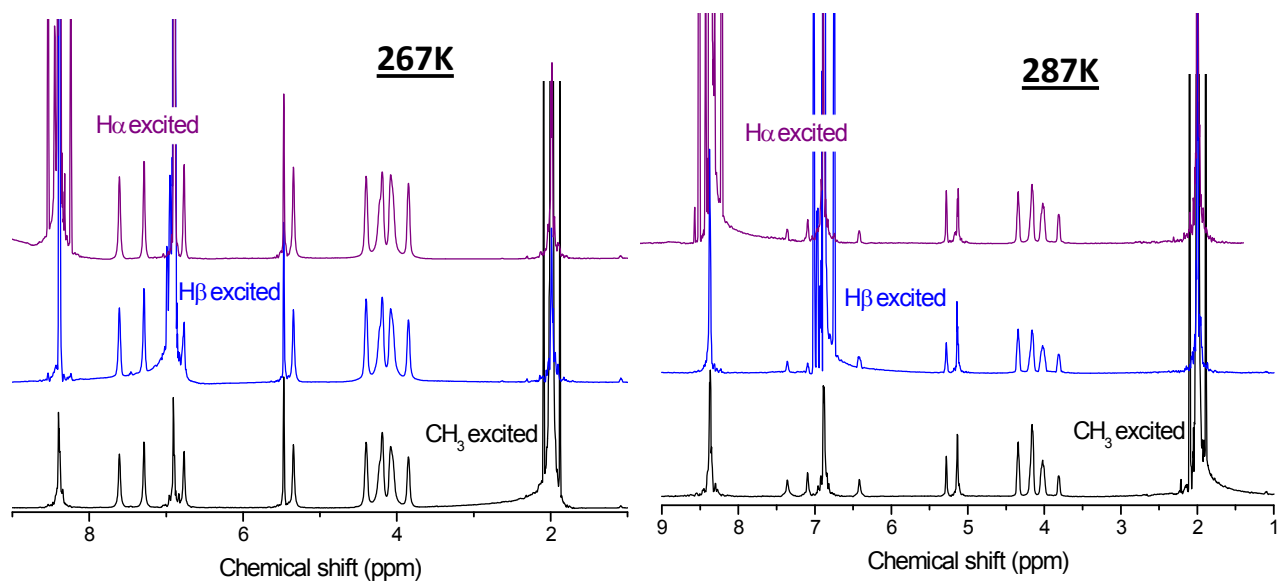


Figure S1: 1D-NOE spectra of α CD-4MP- H_2O mixture at 267K and 287K (projected from 2D-NOESY, excited protons are signed).

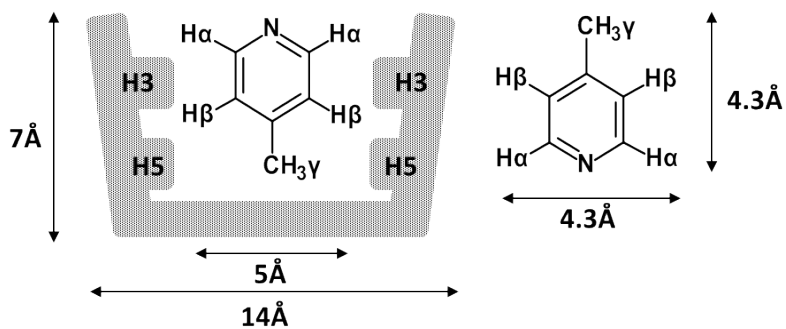


Figure S2: Two possible orientations for α CD-4MP host-guest complex; dimensions from literature¹.

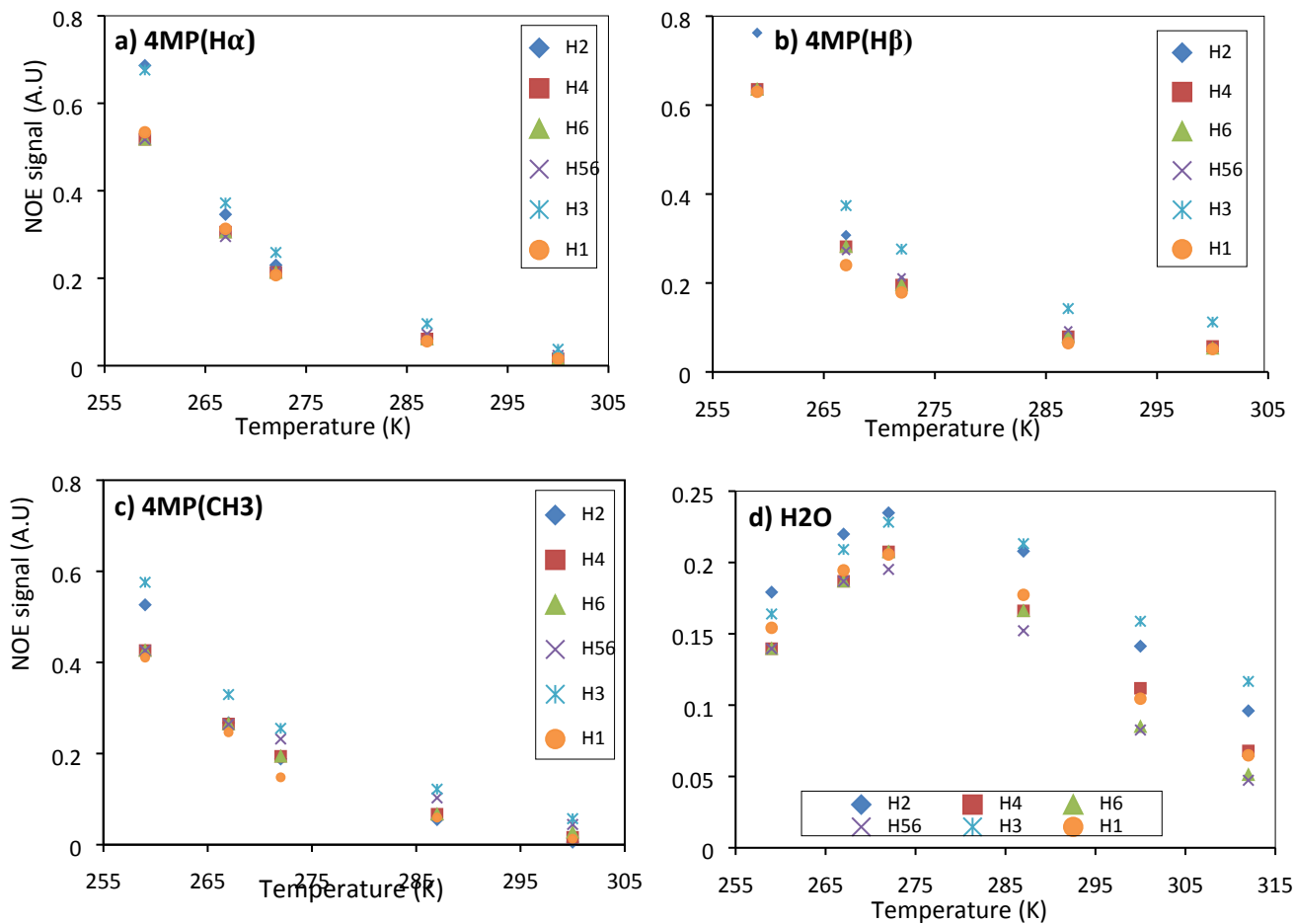


Figure S3: Relative integration of NOE signals vs. temperature of (a) 4MP(H α), (b) 4MP(H β), (c) 4MP(CH $_3$) and (d) H $_2$ O with α CD(H $_1$ -H $_6$) for host-guest investigation in α CD-4MP-H $_2$ O mixture.

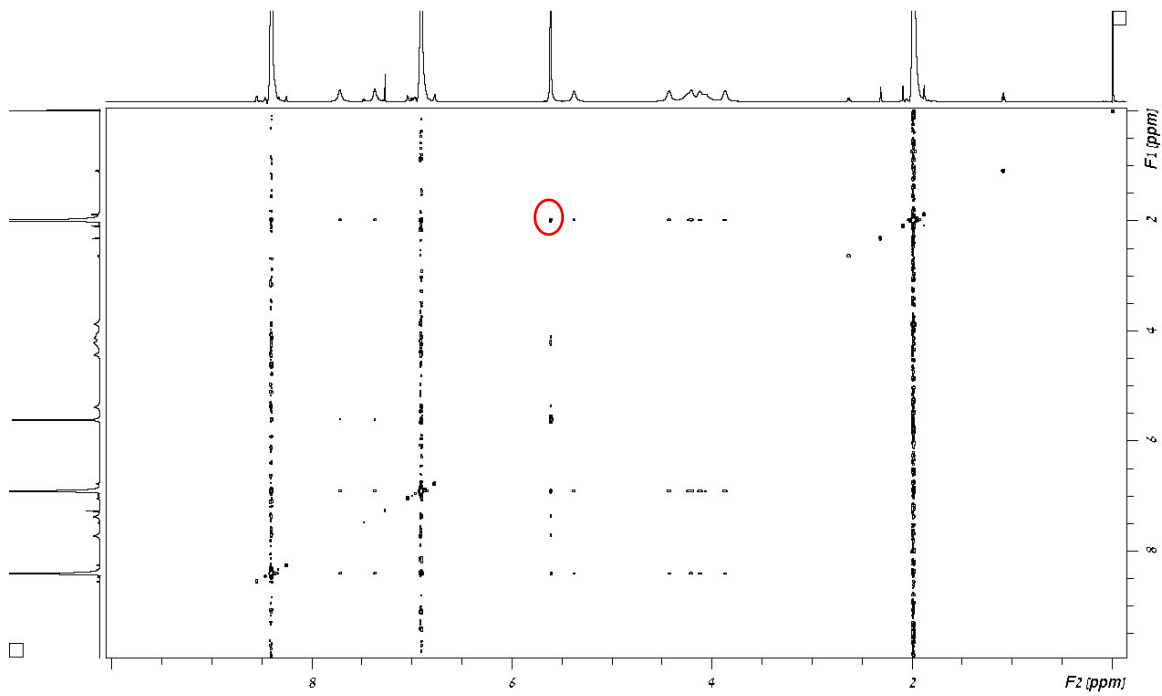


Figure S4: 2D-NOESY spectra of α CD·4MP·H₂O mixture at 259K. 4MP(CH₃)-H₂O cross peak is signed.

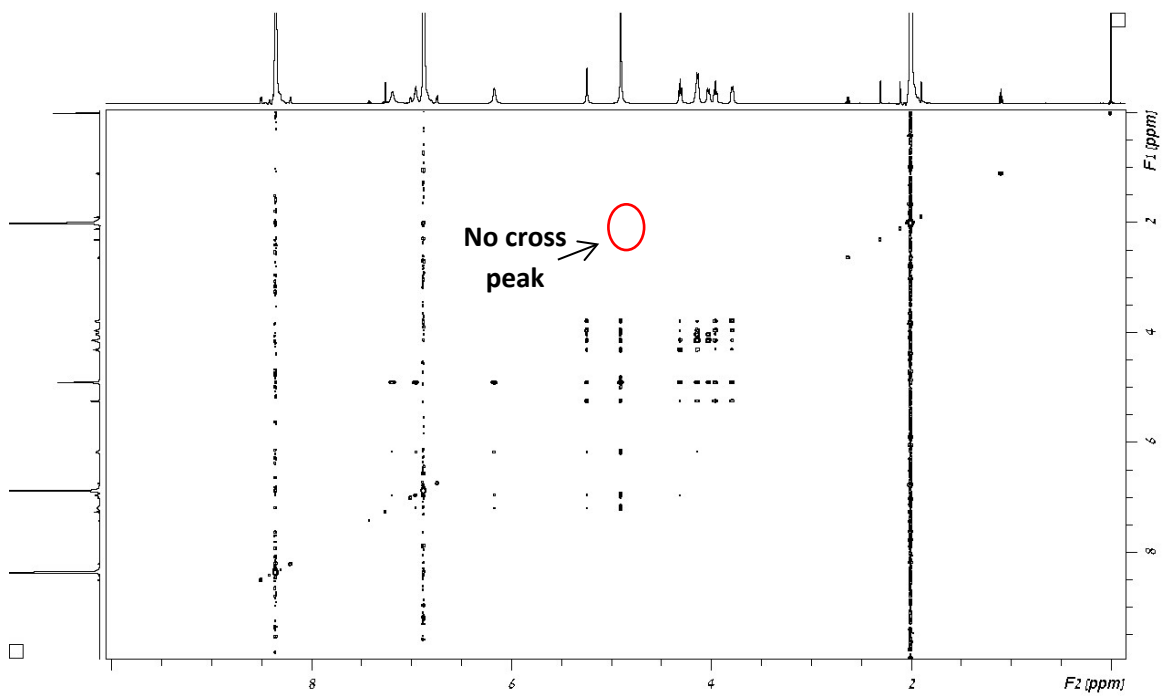


Figure S5: 2D-NOESY spectra of α CD·4MP·H₂O mixture at 300K, with no 4MP(CH₃)-H₂O cross peak.

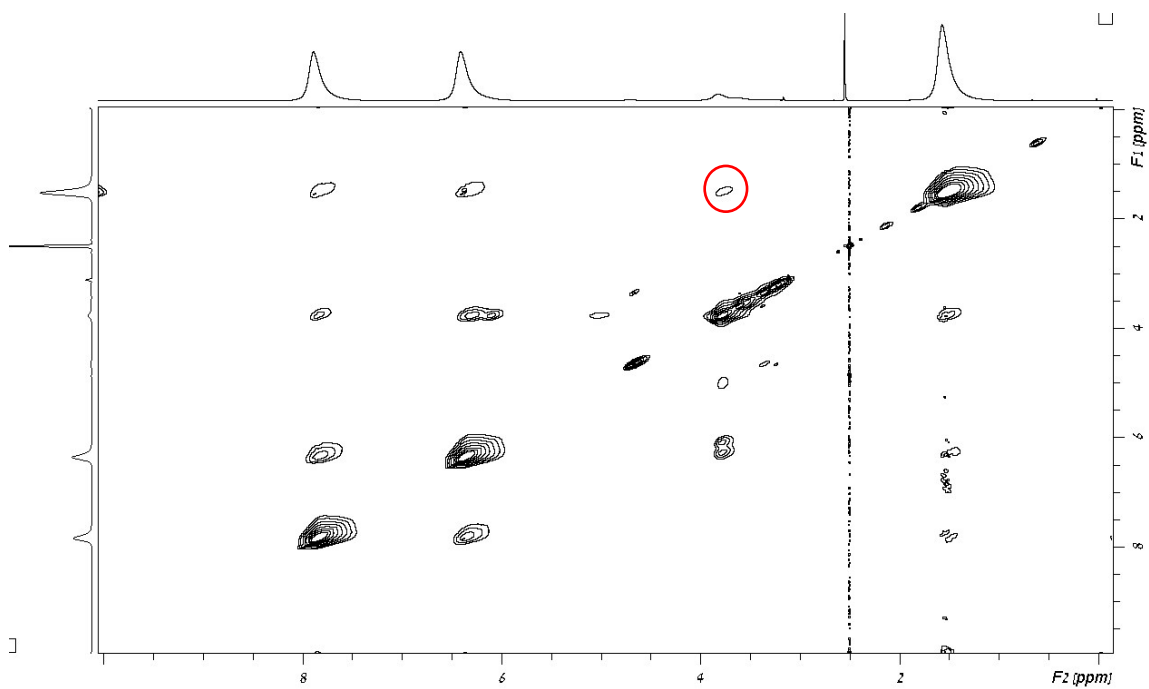


Figure S6: 2D-NOESY spectra of α CD-4MP-H₂O mixture at 335K. 4MP(CH₃)-H₂O cross peak is signed.

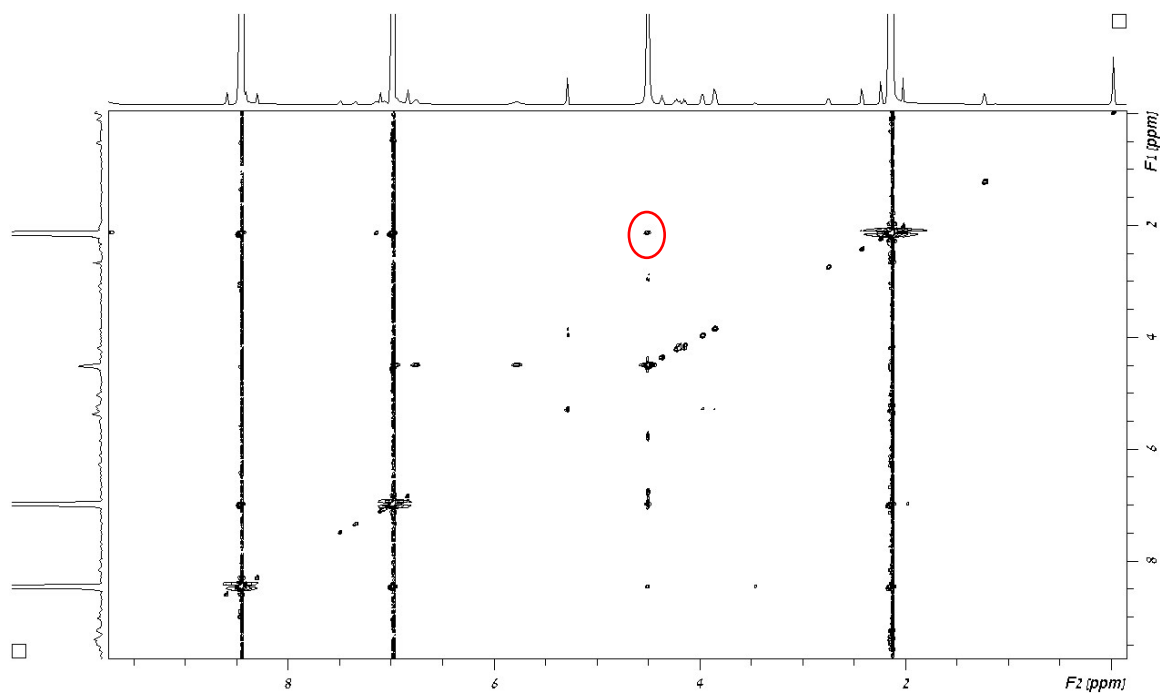


Figure S7: HRMAS 2D-NOESY spectra of α CD-4MP-H₂O mixture at 335K. 4MP(CH₃)-H₂O cross peak is signed.

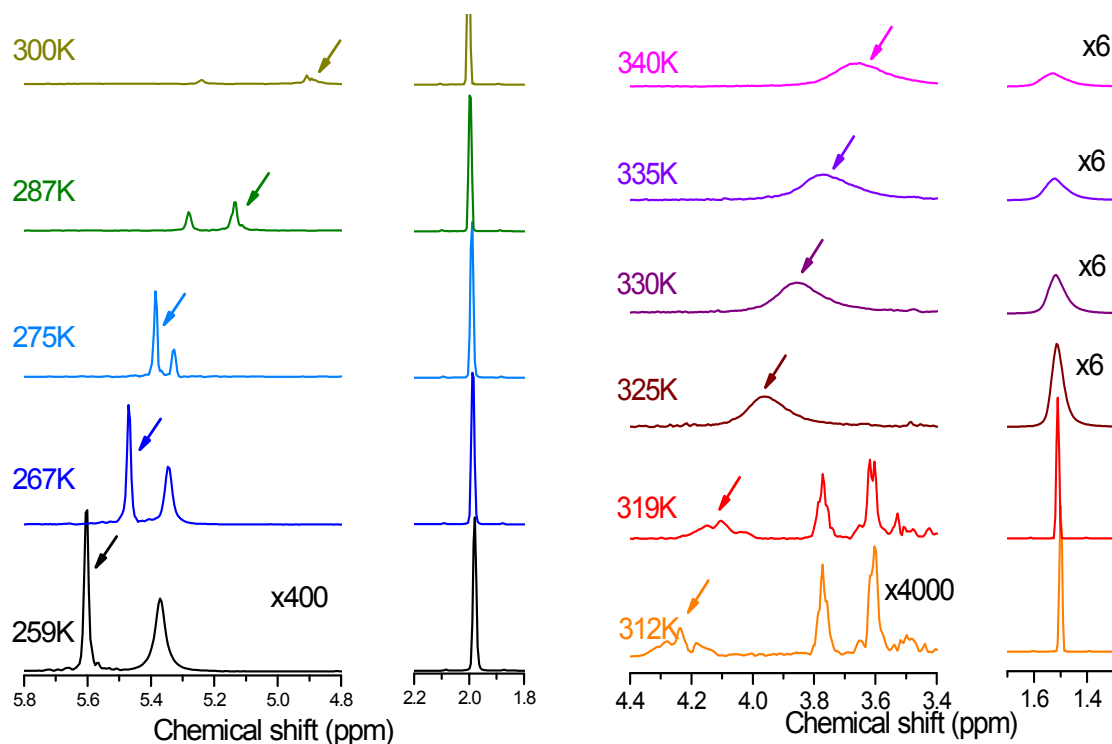
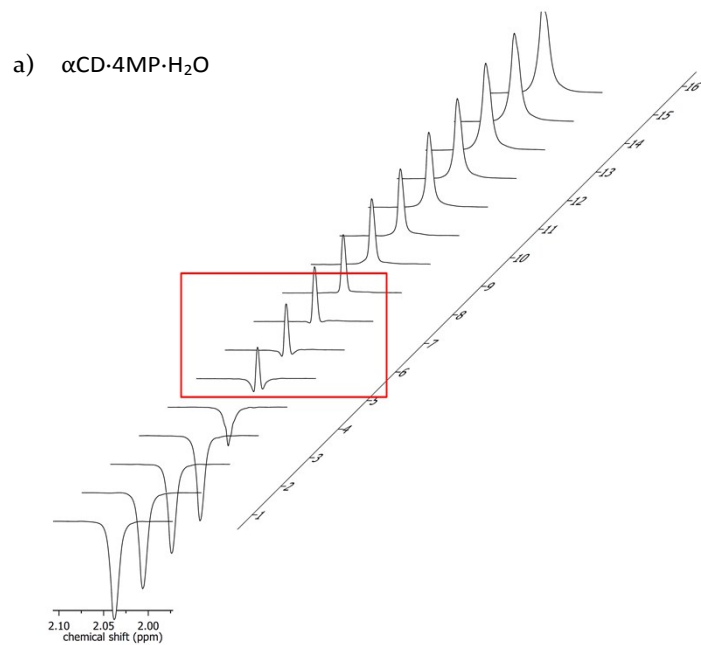


Figure S8: 1D-NOE spectra focused on $4MP(CH_3) \rightarrow H_2O$ cross peak in $\alpha CD \cdot 4MP \cdot H_2O$ mixture at different temperatures (projected from 2D-NOESY spectra, $4MP(CH_3)$ is excited). $4MP(CH_3)$ peak is at 1.5-2.0 ppm and its cross peak with water is sigthed with an arrow. These cross peak were integrated and their normalized values at different temperature are shown in Figure 3 in the main article.

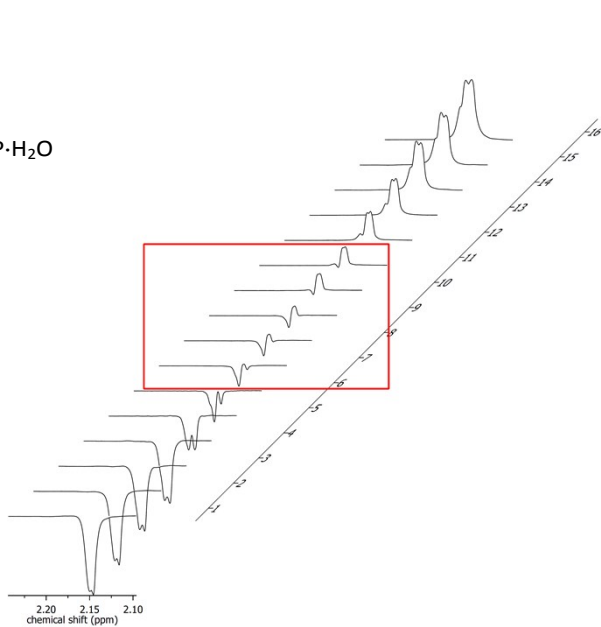
Table S1: Spin lattice relaxation time (T_1) values (in seconds) for the protons in $\alpha CD \cdot 4MP \cdot H_2O$ mixture.

Temperature	$4MP(CH_3)$	DMSO	H_2	H_4	H_6	$H_{5,6}$	H_3	H_2O	H_1	$4MP(H_\beta)$	$4MP(H_\alpha)$
293K	3.22 (0.14) ^a	2.3	1.03	1.04	1.03	1.03	1.03	1.03	1.02	4.84	4.43
312K	4.19 (0.15) ^a	3.6	1.03	1.03	1.02	1.03	1.03	1.03	1.03	5.85	5.94

^a Long and (short) relaxation time due to bi-exponential fitting.



b) 6:1 4MP-H₂O



c) 4MP (with residual water)

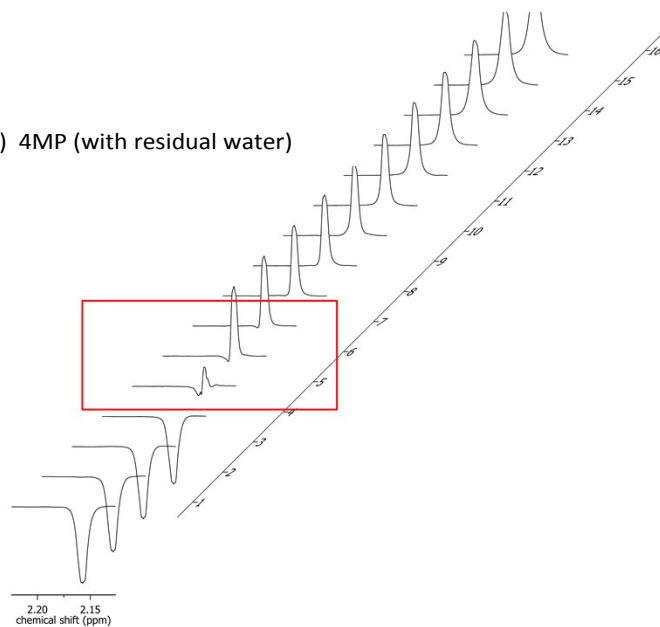


Figure S9: Inverse-recovery measurements of 4MP(CH₃) in a) α CD-4MP-H₂O, b) 6:1 4MP-H₂O and c) 4MP at 278±3K. Several populations with different relaxation times are clearly shown and signed.

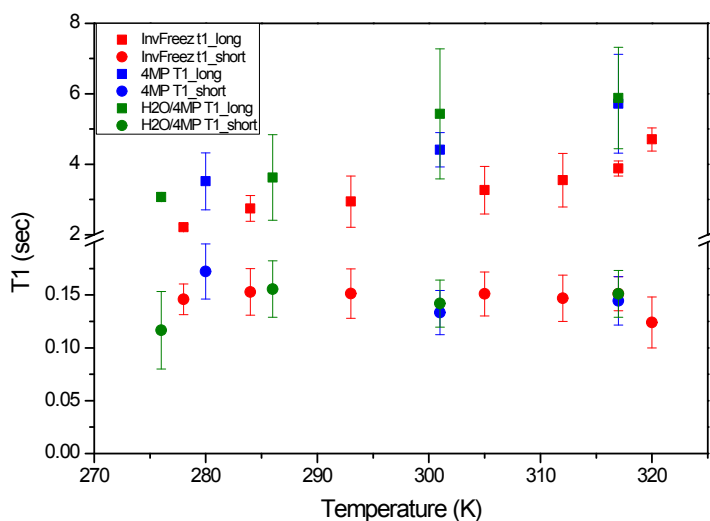


Figure S10: Temperature dependence of 4MP(CH₃) long (square) and short (circle) T₁ values of αCD·4MP·H₂O (red), 6:1 4MP·H₂O (green) and 4MP (blue). Error bars represent errors in fitting bi-exponential curves to data.

T₁ analysis of a peak with several populations was done by fitting the inverse recovery integration points

to $Mz = \sum_{i=1}^n M_{0,i} (1 - 2\exp(-t/T_{1,i}))$. We found that bi-exponential fitting

$Mz = M_{0,long} [1 - 2\exp(-t/T_{1,long})] + M_{0,short} [1 - 2\exp(-t/T_{1,short})]$ is sufficient, meaning that the 4MP(CH₃) populations could be divided to two main groups: with long and short relaxation times.

SI REFERENCES

ⁱ F. B. T. Pessine, A. Calderini and G. L. Alexandrino, in Book: *Magnetic Resonance Spectroscopy*, ed. D-H. Kim, InTech, Shanghai, 2012, 12, p. 237.