#### SUPPLEMENTARY INFORMATION

# Boron-based binary Be<sub>6</sub>B<sub>10</sub><sup>2–</sup> cluster: three-layered aromatic sandwich, electronic transmutation, and dynamic structural fluxionality

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#### Supplementary Information – Part I

- **Table S1.** Calculated orbital compositions for canonical molecular orbitals (CMOs) of the global-minimum (GM) structure of  $Be_6B_{10}^{2-}$  (1,  $C_{2v}$ , <sup>1</sup>A<sub>1</sub>) cluster. Main components are marked in red color.
- Figure S1. Optimized geometries at PBE0/6-311+G(d) level of (a) salt complex cluster Be<sub>6</sub>B<sub>10</sub>Na<sub>2</sub>, a local-minimum (LM) structure, and (b) re-optimized GM of Be<sub>6</sub>B<sub>11</sub><sup>-</sup> cluster (ref 11). Bond distances are in Å. B atoms are in red color, Be in yellow, and Na in purple.
- **Figure S2.** An alternative AdNDP scheme for delocalized  $\pi$  and  $\sigma$  frameworks of Be<sub>6</sub>B<sub>10</sub><sup>2-</sup> (1) cluster, in which Be centers are excluded. The occupation numbers (ONs) are only slightly reduced with respect to those in the 16c-2e  $\pi/\sigma$  AdNDP scheme (see Fig. 7).
- **Figure S3.** AdNDP bonding pattern for  $D_{10h}$  C<sub>10</sub> cluster.

- **Figure S4.** Canonical molecular orbitals (CMOs) of the TS structure of  $Be_6B_{10}^{2-}$  cluster (see Fig. 2(c)).
- **Figure S5.** AdNDP bonding pattern for the TS structure of  $Be_6B_{10}^{2-}$  cluster (Fig. 2(c)). ONs are shown.
- **Figure S6.** Displacement vectors of (a) selected vibrational frequencies of GM  $Be_6B_{10}^{2-}$  (1) cluster and (b) those of two TS structures (**TS1** and **TS2**).

### Supplementary Information – Part II

Three short movies extracted from the molecular dynamics simulations for  $Be_6B_{10}^{2-}$  cluster at temperatures of 300, 600, and 1500 K.

**Table S1.**Calculated orbital compositions for canonical molecular orbitals (CMOs) of the<br/>global-minimum (GM) structure of  $Be_6B_{10}^{2-}$  (1,  $C_{2v}$ ,  $^1A_1$ ). Main components are<br/>marked in red color.

СМО	B 2s/2p (%)	Be 2s/2p (%)	СМО	B 2s/2p (%)	Be 2s/2p (%)
HOMO	89.0	8.2	НОМО-2	65.3	26.5
НОМО-1	85.7	12.0	НОМО-3	70.2	28.6
НОМО-6	93.3	5.7	НОМО-10	76.9	22.1
<b>Е</b> НОМО-7	91.1	8.0	НОМО-9	75.4	23.4
НОМО-13	78.7	20.6	НОМО-14	60.0	38.1
НОМО-5	39.3	58.9	НОМО-4	32.8	66.1

Figure S1. Optimized geometries at PBE0/6-311+G(d) level of (a) salt complex cluster Be<sub>6</sub>B<sub>10</sub>Na<sub>2</sub>, a local-minimum (LM) structure, and (b) re-optimized GM of Be<sub>6</sub>B<sub>11</sub><sup>-</sup> cluster (ref 11). Bond distances are in Å. B atoms are in red color, Be in yellow, and Na in purple.





**Figure S2.** An alternative AdNDP scheme for delocalized  $\pi$  and  $\sigma$  frameworks of Be<sub>6</sub>B<sub>10</sub><sup>2-</sup> (1) cluster, in which Be centers are excluded. The occupation numbers (ONs) are only slightly reduced with respect to those in the 16c-2e  $\pi/\sigma$  AdNDP scheme (see Fig. 7).



## **Figure S3.** AdNDP bonding pattern for $D_{10h}$ C<sub>10</sub> cluster.













5×10c–2e σ bonds ON = 2.00 |e|



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5×10c–2e π bonds ON = 2.00 |e|



**Figure S4.** Canonical molecular orbitals (CMOs) of the TS structure of  $Be_6B_{10}^{2-}$  cluster (see Fig. 2(c)).

(a)				
HOMO–21 (a₁)	HOMO-20 (b <sub>2</sub> )	HOMO–19 (a <sub>1</sub> )	HOMO-17 (b <sub>2</sub> )	HOMO–18 (a₁)
HOMO-15 (b <sub>2</sub> )	HOMO-16 (a <sub>1</sub> )	HOMO-12 (b <sub>2</sub> )	HOMO–11 (a <sub>1</sub> )	HOMO-8 (a <sub>2</sub> )
(b)				
HOMO-13 (a <sub>1</sub> )	HOMO-7 (b <sub>2</sub> )	HOMO-6 (a <sub>1</sub> )	HOMO-1 (b <sub>2</sub> )	HOMO (a <sub>1</sub> )
HOMO-14 (b <sub>1</sub> )	HOMO-10 (a <sub>2</sub> )	HOMO-9 (b <sub>1</sub> )	HOMO-3 (a <sub>2</sub> )	HOMO-2 (b <sub>1</sub> )
	(d)	(e)		

- HOMO-5 (a<sub>1</sub>) HOMO-4 (b<sub>1</sub>)

**Figure S5.** AdNDP bonding pattern for the TS structure of  $Be_6B_{10}^{2-}$  cluster (see Fig. 2(c)). ONs are shown.



**Figure S6.** Displacement vectors of (a) selected vibrational frequencies of GM  $Be_6B_{10}^{2-}$  (1) cluster and (b) those of two TS structures (**TS1** and **TS2**).

