

Structure and Dynamics of High-Temperature Strontium Aluminosilicate Melts

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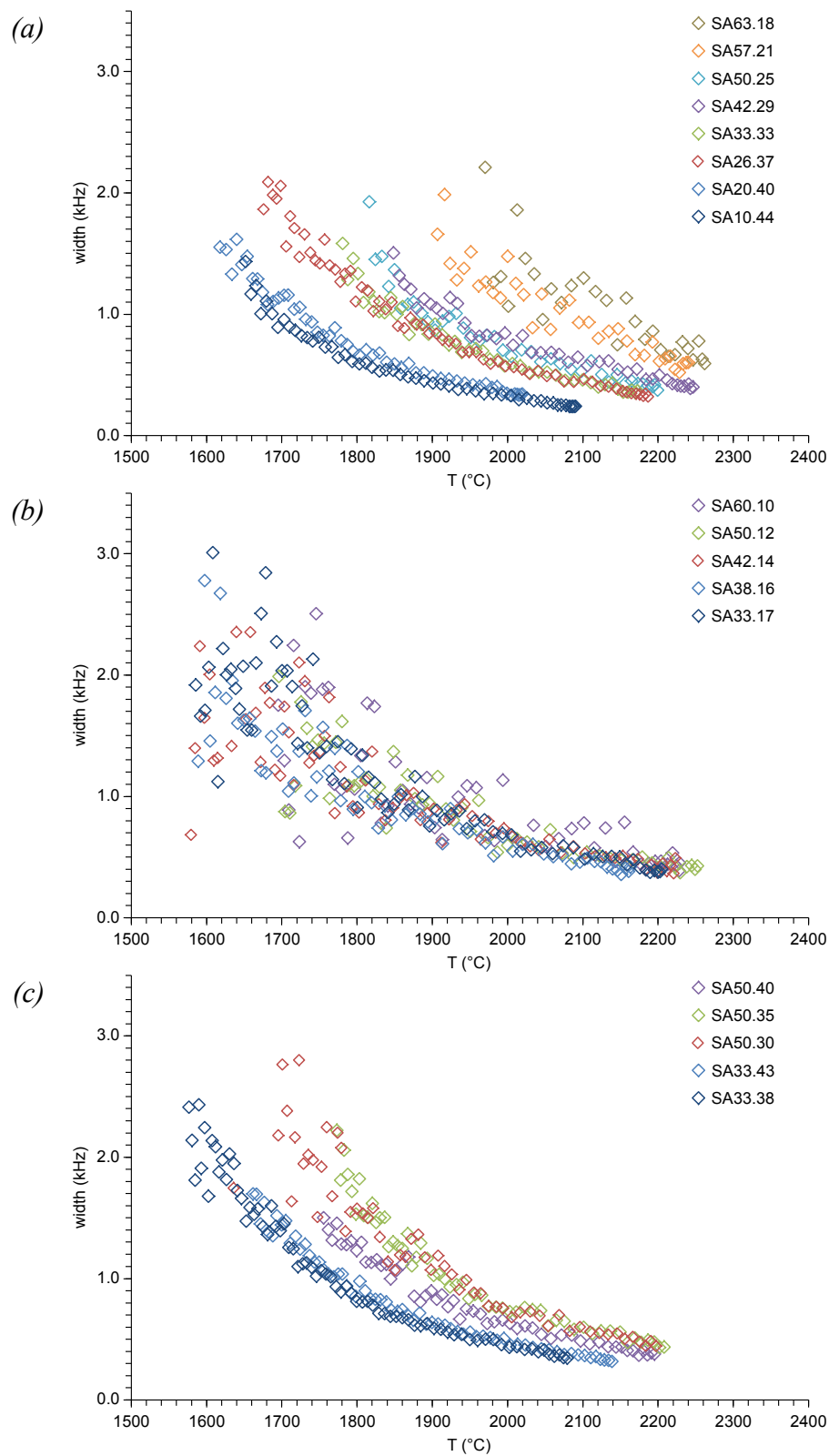


Figure S1. Evolution of the ^{27}Al NMR line width during free cooling of all compositions studied: (a) on the charge-compensation line ($R = 1$), (b) in the per-alkaline field ($R = 3$) and (c) in the per-aluminous region ($R < 1$).

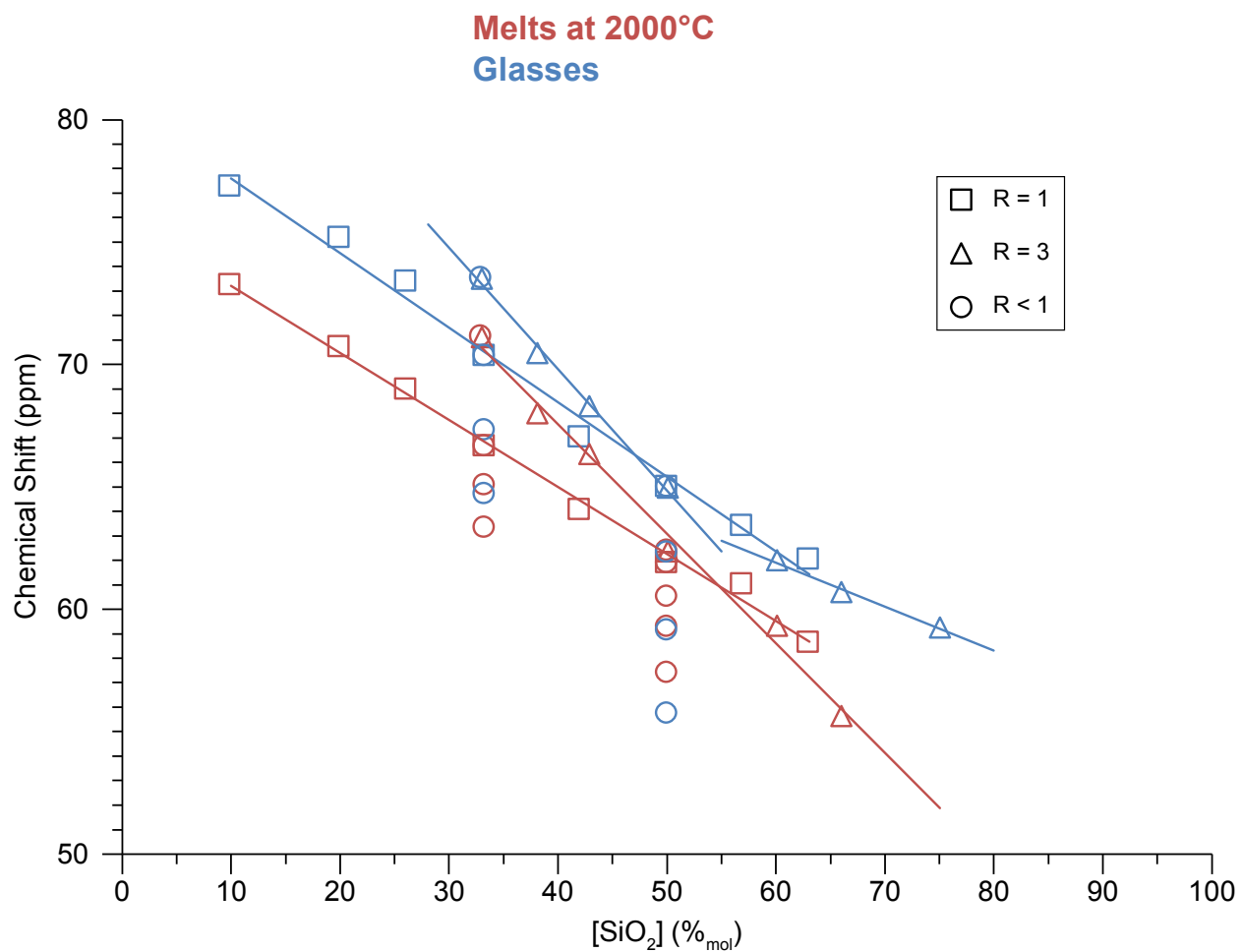


Figure S2. ²⁷Al chemical shifts at 2000°C $\delta_{2000^\circ\text{C}}$ (red) and “barycenter” of the glass spectra $\langle\delta_{\text{iso}}\rangle_{\text{glass}}$ (see text for details) (blue) for R = 1 (rectangles), R = 3 (triangles) and R < 1 (circles) compositions all given as a function SiO₂ content. Continuous lines are linear fit.

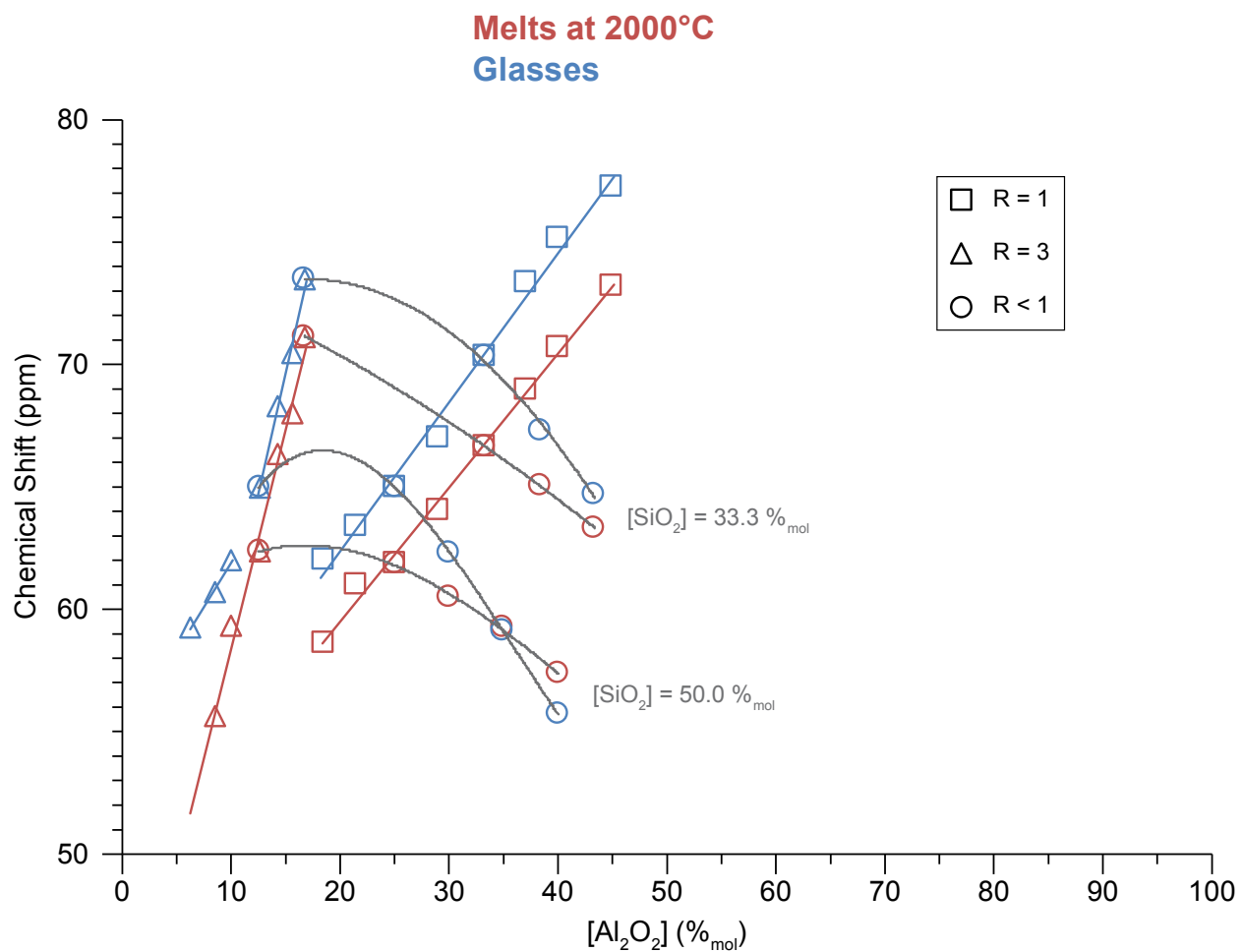


Figure S3. ²⁷Al chemical shifts at 2000°C $\delta_{2000^\circ\text{C}}$ (red) and “barycenter” of the glass spectra $\langle\delta_{\text{iso}}\rangle_{\text{glass}}$ (see text for details) (blue) for R = 1 (rectangles), R = 3 (triangles) and R < 1 (circles) compositions all given as a function SiO₂ content. Continuous lines are linear fit, grey lines are guides for the eyes for compositions at a given [SiO₂].

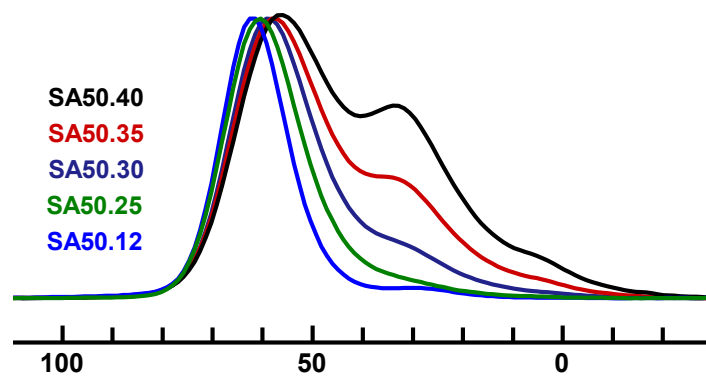


Figure S4. 27Al NMR MAS experiments (20.0 T) of peraluminous glasses.