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## **Supplementary Information for: Computational investigation of gas detection and selectivity on $\text{TiS}_3$ nanoflakes supported by experimental evidence**

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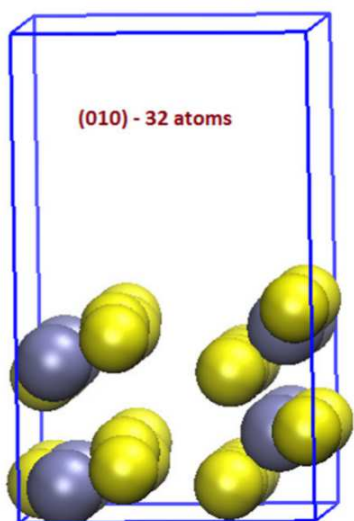
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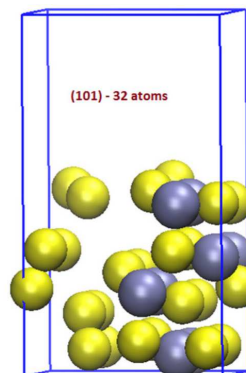
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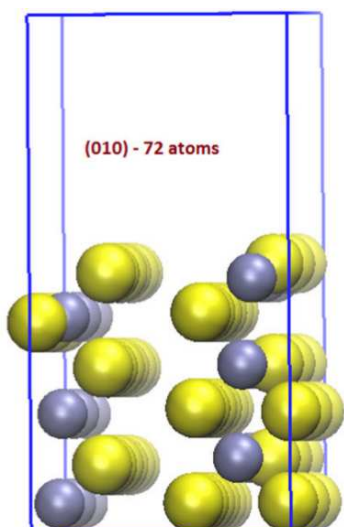
**Note:** In all figures S7 to S16, only the slab atoms of configuration No.1 are visible, that is, all Ti and S atoms that belong to the other configurations are hidden for clarity.



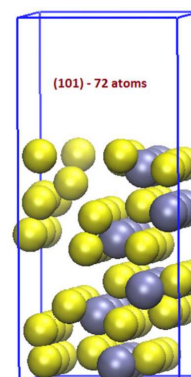
**Fig. S1** Computational slab model for surface (010) having 32 atoms.



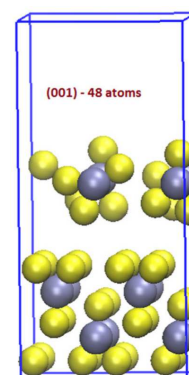
**Fig. S3** Computational slab model for surface (101) having 32 atoms.



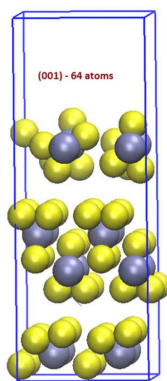
**Fig. S2** Computational slab model for surface (010) having 72 atoms.



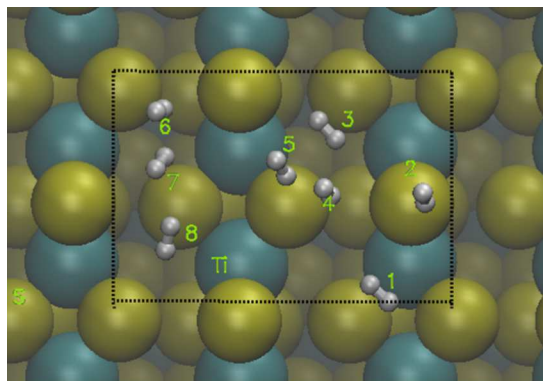
**Fig. S4** Computational slab model for surface (101) having 72 atoms.



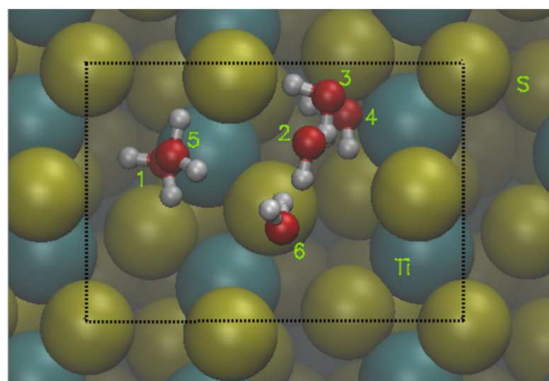
**Fig. S5** Computational slab model for surface (001) having 48 atoms.



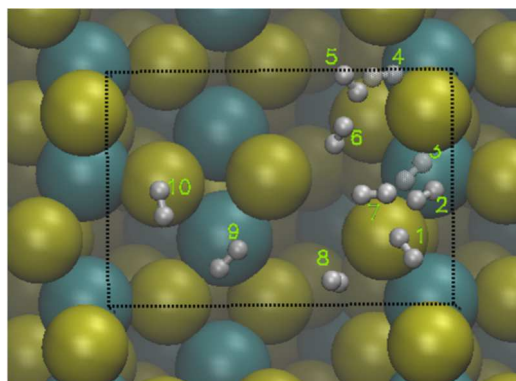
**Fig. S6** Computational slab model for surface (001) having 64 atoms.



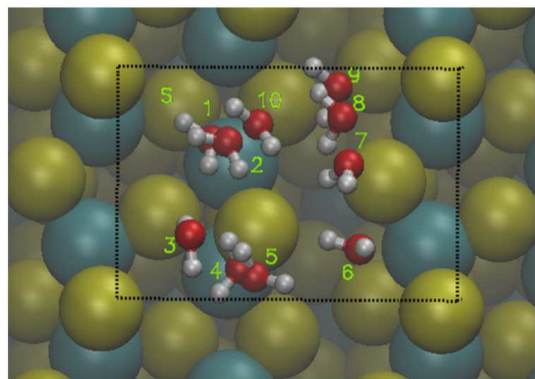
**Fig. S9** Optimized structure of adsorbed  $\text{H}_2$  molecule on defect free (001). Each number corresponds to a separate calculation.



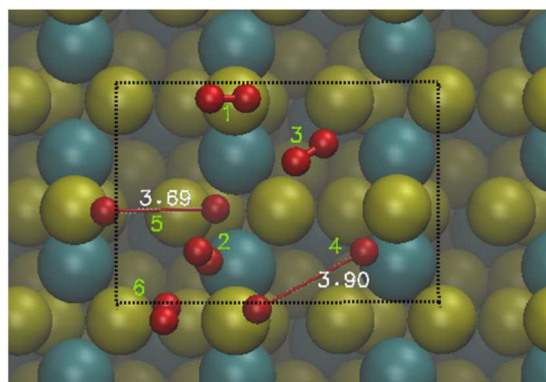
**Fig. S7** Optimized structure of adsorbed  $\text{H}_2\text{O}$  molecule on defect free (001). Each number corresponds to a separate calculation. Only one slab surface is visible.



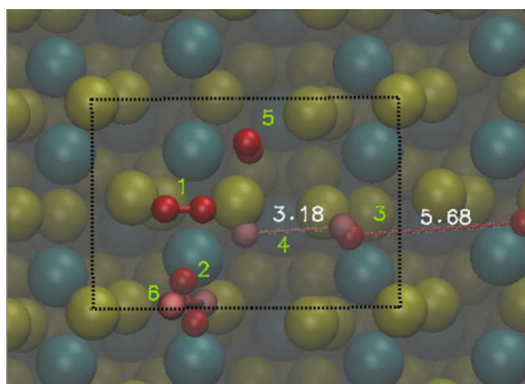
**Fig. S10** Optimized structure of adsorbed  $\text{H}_2$  molecule on double S-deficient (001). Each number corresponds to a separate calculation.



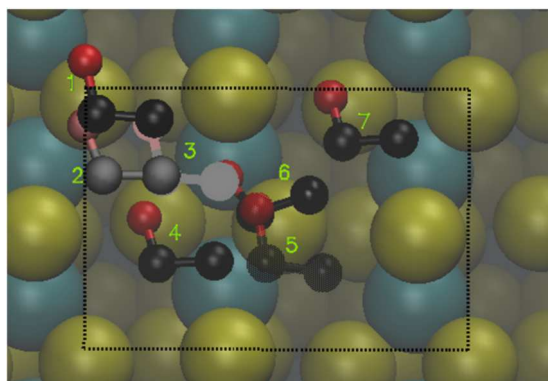
**Fig. S8** Optimized structure of adsorbed  $\text{H}_2\text{O}$  molecule on double S-deficient (001). Each number corresponds to a separate calculation.



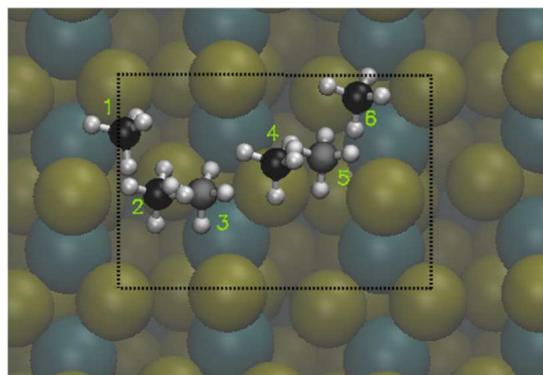
**Fig. S11** Optimized structure of adsorbed  $\text{O}_2$  molecule on defect free (001). Each number corresponds to a separate calculation. Those O-O pairs falling at more than 3 Å apart indicate dissociative adsorption.



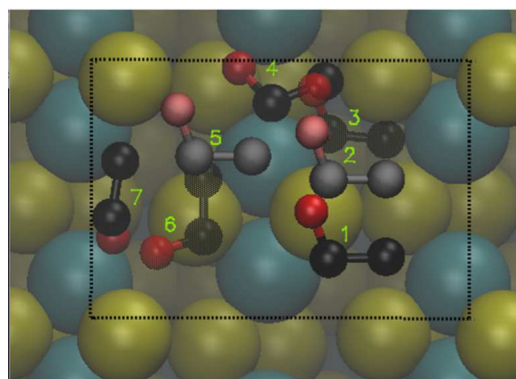
**Fig. S12** Optimized structure of adsorbed  $\text{O}_2$  molecule on double S-deficient (001). Each number corresponds to a separate calculation. Those O-O pairs falling at more than 3 Å apart indicate dissociative adsorption.



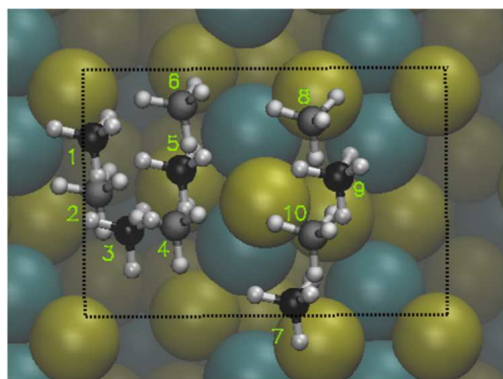
**Fig. S15** Optimized structure of adsorbed  $\text{C}_2\text{H}_5\text{OH}$  molecule on defect free (001). Each number corresponds to a separate calculation.



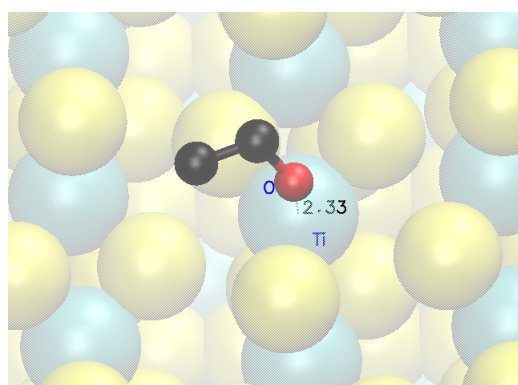
**Fig. S13** Optimized structure of adsorbed  $\text{CH}_4$  molecule on defect free (001). Each number corresponds to a separate calculation.



**Fig. S16** Optimized structure of adsorbed  $\text{C}_2\text{H}_5\text{OH}$  molecule on double S-deficient (001). Each number corresponds to a separate calculation.

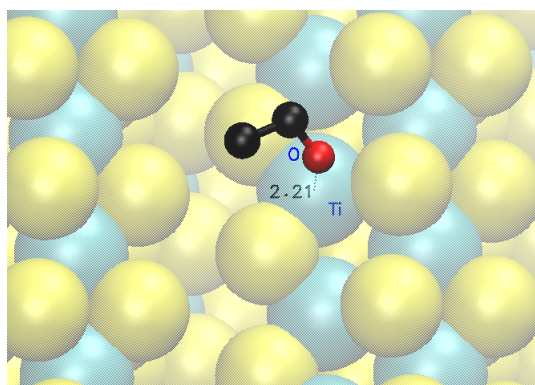


**Fig. S14** Optimized structure of adsorbed  $\text{CH}_4$  molecule on double S-deficient (001). Each number corresponds to a separate calculation.



**Fig. S17** Geometry of the ethanol molecule on the stoichiometric surface. This particular structure corresponds to the energy peak at about -1.1 eV.





**Fig. S18** Geometry of the ethanol molecule on the double S-deficient surface. This particular structure corresponds to the energy peak at about -1.1 eV.