Supporting Information

Difference of the Micro-Dynamics Mechanism between Aromatic Nylon and Aliphatic Nylon during Water Absorption: Spectroscopic Evidence

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Theory of the scaling-MW2D spectroscopy based on auto-correlation

Q(v, T) is an $M \times N$ spectral intensity matrix. The v and T are spectral variable (e.g., wavenumber) and perturbation variable (e.g., temperature), respectively.

$$Q(v,T) = \begin{pmatrix} y(v,T_{I}) \\ \vdots \\ y(v,T_{j}) \\ \vdots \\ y(v,T_{M}) \end{pmatrix}$$
(1)

The reference spectrum and dynamic spectrum in the *j*th submatrix of Q(v, T):

$$\overline{y}(v) = \frac{1}{2m+1} \sum_{J=j-m}^{j+m} y(v, T_J)$$
(2)

$$\widetilde{y}(v,T_{J}) = y(v,T_{J}) - \overline{y}(v)$$
(3)

where J corresponds to the index of rows. A mean-centered *j*th submatrix of Q(v, T) is obtained.

$$q_{j}(v,T) = \begin{pmatrix} \widetilde{y}(v,T_{j-m}) \\ \vdots \\ \widetilde{y}(v,T_{j}) \\ \vdots \\ \widetilde{y}(v,T_{j+m}) \end{pmatrix}$$
(4)

The $q_j(v, T)$ has 2m+1 rows which is called the window size. The index range of the perturbation variable *I* of $q_j(v, T)$ is from *j*-*m* to *j*+*m*.

The generalized synchronous 2D correlation spectra is calculated from $q_j(v, T)$:

$$\Phi_{j}(v_{1},v_{2}) = \frac{1}{2m} \sum_{J=j-m}^{j+m} \widetilde{y}(v_{1},T_{J}) \cdot \widetilde{y}(v_{2},T_{J})$$
(5)

For each window, the standard deviations of spectral intensities at v_1 and v_2 are defined as follows.

$$\sigma(v_1) = \sqrt{\Phi_j(v_1, v_1)} \tag{6}$$

$$\sigma(v_2) = \sqrt{\Phi_j(v_2, v_2)}$$
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(7)

The correlation coefficient $\rho(v_1, v_2)$ is defined:

$$\rho(v_1, v_2) = \Phi_i(v_1, v_2) / [\sigma(v_1) \cdot \sigma(v_2)]$$
(8)

The scaled forms of the synchronous correlation spectrum in each window are defined:

$$\Phi_{j}(v_{1},v_{2})^{(Scaled)} = \Phi_{j}(v_{1},v_{2}) \cdot [\sigma(v_{1}) \cdot \sigma(v_{2})]^{-\alpha} \cdot |\rho(v_{1},v_{2})|^{\beta}$$
(9)

The constant α is the scaling factor, and β is a correlation enhance factor. The value of α is limited to 0–1.0.

For scaling-MW2D based on auto-correlation, each row of the correlation matrix of scaling-MW2D is directly extracted from a diagonal line of Φ_j ^(Scaled) matrix, namely $v_1 = v_2$. Thus, the auto-correlation scaling-MW2D can be described below.

$$\Phi_{j}(v_{1},v_{1})^{(Scaled)} = \Phi_{j}(v_{1},v_{1}) \cdot [\sigma(v_{1}) \cdot \sigma(v_{1})]^{-\alpha} \cdot |\rho(v_{1},v_{1})|^{\beta}$$
(10)

$$\rho(v_1, v_1) = \Phi_j(v_1, v_1) / [\sigma(v_1) \cdot \sigma(v_1)] = 1$$
(11)

$$\Phi_{j}(v_{1},v_{1})^{(Scaled)} = [\Phi_{j}(v_{1},v_{1})]^{1-\alpha}$$
(12)

The auto-correlation scaling-MW2D correlation spectrum is gained *via* sliding window position from j=1+m to *M-m* and repeating calculations of equation (2)-(12) at each window.



Figure S1. Time-dependent ATR FTIR spectra of the MXD6 film in the region of 1300–1120 cm⁻¹ collected during ordinary water (H₂O) absorption.



Figure S2. Spectral intensity changes extracted from the time-dependent ATR FTIR spectra of MXD6 during D₂O absorption from 2.0 min to 27.0 min.



Figure S3. WAXD spectra of MXD6 films before and after water absorption.



Figure S4. Spectral intensity changes extracted from the time-dependent ATR FTIR spectra of the PA6 film during D₂O absorption from 2.0 min to 27.0 min.



Figure S5. Synchronous (left) and asynchronous (right) generalized 2D correlation ATR FTIR spectra calculated from time-dependent ATR FTIR spectra of the PA6 film during D₂O absorption within 2.0–27.0 min in the regions of 1700–1490 cm⁻¹, 1700–1490 cm⁻¹ vs 1490–1400 cm⁻¹, and 1490–1400 cm⁻¹.

Cross correlation peaks	Sign in synchronous	Sign in asynchronous	Sequential		
(cm ⁻¹ , cm ⁻¹)	spectra	spectra	order		
(1632, 1537)	+	-	1632←1537		
(1632, 1463)	+	+	1632→1463		
(1632, 1421)	+	+	1632→1421		
(1632, 700)	+	+	1632→700		
(1537, 1463)	+	+	1537→1463		
(1537, 1421)	+	+	1537→1421		
(1537, 700)	+	+	1537→700		
(1463, 1421)	+	+	1463→1421		
(1463, 700)	+	+	1463→700		
(1421, 700)	+	+	1421→700		
(1632, 1204)	+	+	1632→1204		
(1537, 1204)	+	+	1537→1204		
(1463, 1204)	+	+	1463→1204		
(1421, 1204)	+	+	1421→1204		
(1204, 700)	+	-	1204←700		
$1537 \text{ cm}^{-1} \rightarrow 1632 \text{ cm}^{-1} \rightarrow 1463 \text{ cm}^{-1} \rightarrow 1421 \text{ cm}^{-1} \rightarrow 700 \text{ cm}^{-1} \rightarrow 1204 \text{ cm}^{-1}$					
δ (N-H, bonded) $\rightarrow v$ (C=O, bonded) $\rightarrow \delta$ (C-H, -CH ₂ -NH-) $\rightarrow \delta$ (C-H, -CH ₂ -CO-) $\rightarrow \delta$					
δ (=C-H, benzene ring) $\rightarrow \gamma$ (-CH ₂ - skeleton, crystalline)					

Table S1. Sequential orders of the process I (2.0–12.0 min) for MXD6 obtained from the synchronous and asynchronous 2D correlation ATR-FTIR spectra in **Figure 6**.

Table S2. Sequential orders of the process II (12.0–27.0 min) for MXD6 obtained from the synchronous and asynchronous 2D correlation ATR-FTIR spectra in **Figure 7.**

Cross correlation peaks	Sign in synchronous	Sign in asynchronous	Sequential		
(cm ⁻¹ , cm ⁻¹)	spectra	spectra	order		
(1632, 1463)	+	+	1632→1463		
(1632, 1421)	+	+	1632→1421		
(1632, 700)	+	+	1632→700		
(1463, 1421)	+	0	1463=1421		
(1463, 700)	+	-	1463←700		
(1421, 700)	+	-	1421←700		
(1632, 1204)	+	+	1632→1204		
(1463, 1204)	+	0	1463=1204		
(1421, 1204)	+	0	1421=1204		
(1204, 700)	+	-	1204←700		
$1632 \text{ cm}^{-1} \rightarrow 700 \text{ cm}^{-1} \rightarrow 1463 \text{ cm}^{-1} = 1421 \text{ cm}^{-1} = 1204 \text{ cm}^{-1}$					
$v(C=O, bonded) \rightarrow \delta(=C-H, benzene ring) \rightarrow \delta(C-H, -CH_2-NH-) = \delta(C-H, -CH_2-CO-) = \delta(C-H, -CH_2-CO-)$					
γ (-CH ₂ - skeleton, crystalline)					

Table S3. Sequential orders of PA6 (2.0–27.0 min) obtained from the synchronous and asynchronous 2D correlation ATR-FTIR spectra in **Figure S5.**

Cross correlation peaks (cm ⁻¹ , cm ⁻¹)	Sign in synchronous spectra	Sign in asynchronous spectra	Sequential order	
(1633, 1540)	+	-	1633←1540	
(1633, 1476)	+	+	1633→1476	
(1633, 1417)	+	+	1633→1417	
(1540, 1476)	+	+	1540→1476	
(1540, 1417)	+	+	1540→1417	
(1476, 1417)	+	-	1476←1417	
$1540 \text{ cm}^{-1} \rightarrow 1633 \text{ cm}^{-1} \rightarrow 1417 \text{ cm}^{-1} \rightarrow 1476 \text{ cm}^{-1}$				
δ (N-H, bonded) $\rightarrow \nu$ (C=O, bonded) $\rightarrow \delta$ (C-H, -CH ₂ -CO-) $\rightarrow \delta$ (C-H, -CH ₂ -NH-)				