

# Supplemental Information

## Why does 4-biphenyl carbonyl azide have ultra-short-lived excited states? An ultrafast UV–vis spectroscopic and computational study

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### 1. Figure list:

**Fig. S1** Transient absorption spectra of 4-BpCON<sub>3</sub> in cyclohexene after 308 nm excitation (a) and time course at 465 nm for transient absorption (TA) (b) respectively.

**Fig. S2** 2-D transient absorption spectra of 4-biphenyl oxadiazole in acetonitrile. The pump wavelength is 282 nm.

**Fig. S3** 2-D transient absorption spectra of 4-biphenyl isocyanate in acetonitrile. The pump wavelength is 308 nm.

**Fig. S4** 2-D transient absorption spectra of 4-biphenyl carboxamide in acetonitrile. The pump wavelength is 270 nm.

**Fig. S5** Comparison of UV-vis absorption of 4-biphenyl carbonyl azide (4-BpCON<sub>3</sub>) and its photolysis products: 4-biphenyl oxadiazole (BpOx), 4-biphenyl isocyanate (BpNCO), and 4-biphenyl carboxamide (BpCONH<sub>2</sub>) in acetonitrile.

**Fig. S6** Transient absorption spectra of 4-BpCON<sub>3</sub> in CHCl<sub>3</sub> after 308 nm excitation. (1.a) and (1.b) are TA and time course at 465 nm in CHCl<sub>3</sub> (Amylene as stabilizer). (2.a) and (2.b) are TA and time course at 460 nm in CHCl<sub>3</sub> (EtOH as stabilizer).

**Fig. S7** UV-vis absorption spectra of 4-biphenyl carbonyl azide (4-BpCON<sub>3</sub>) in different solvents: CHCl<sub>3</sub>, MeOH, acetonitrile (ACN), and cyclohexane (cHexane). The wavelength of maximum absorption is shown in parentheses.

**Fig. S8** 2-D transient absorption spectra of 4-BpCON<sub>3</sub> in MeOH. The pump wavelength is 308 nm (a) and 270 nm (b). A slow decay component is seen upon 270 nm excitation.

**Fig. S9** 2-D transient absorption spectra of 4-BpCON<sub>3</sub> in acetonitrile. The pump wavelength is 308 nm in (a) and 270 nm in (b) respectively. The transient absorption of 4-biphenyl oxadiazole (BpOx) appears in both cases.

**Fig. S10** Transient absorption spectra of fluorene-2-carbonyl azide (F2CON<sub>3</sub>) in acetonitrile after 308 nm excitation (a), time course at 435 nm for transient absorption (TA) (b), and time course for the stimulated emission (SE) at 370 nm (c) respectively.

**Fig. S11** UV-vis absorption spectra of 4-biphenyl carbonyl azide (4-BpCON<sub>3</sub>), fluorene-2-carbonyl azide (F2CON<sub>3</sub>), 2-naphthalene sulfonyl azide (2-NpSO<sub>2</sub>N<sub>3</sub>), and 2-naphthalene carbonyl azide (2-NpCON<sub>3</sub>) in acetonitrile.

**Fig. S12** Structural information of MeCON<sub>3</sub>, PhCON<sub>3</sub>, 4-BpCON<sub>3</sub>, and 2-NpCON<sub>3</sub> at their ground states (S<sub>0</sub> structure at PBE0/ def2-TZVP level) and relaxed excited states (S<sub>1</sub>) at cam-B3LYP/ def2-TZVP level. Units: bond length, Å ; NNN angle, degrees(°). The dipole moment ( $\mu$ ) in Debye and  $\Delta E$ , the relative energy against the ground state (S<sub>0</sub>) in Kcal/mol were calculate at CAM-B3LYP/ def2-TZVP level and are also given.

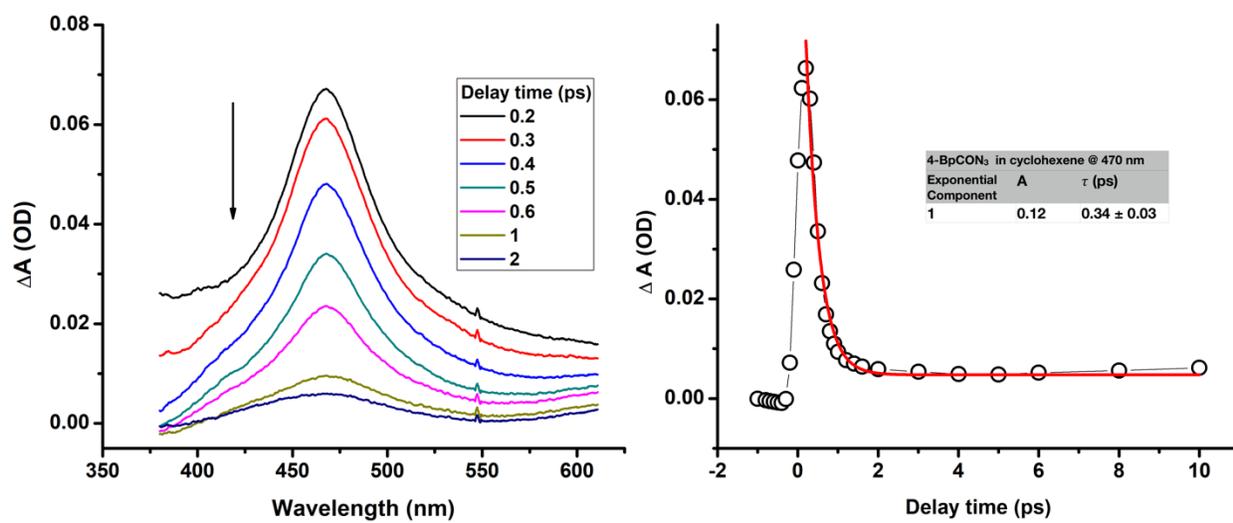
**Fig. S13** Comparison of potential energy surfaces of  $\pi(\text{Ar})\pi^*$  and  $n(\text{CO})\pi^*$  S<sub>1</sub> states of F2CON<sub>3</sub> and 2-NpCON<sub>3</sub>.

2. **Table S1.** TDDFT results of lowest electronic transitions of selected molecules.

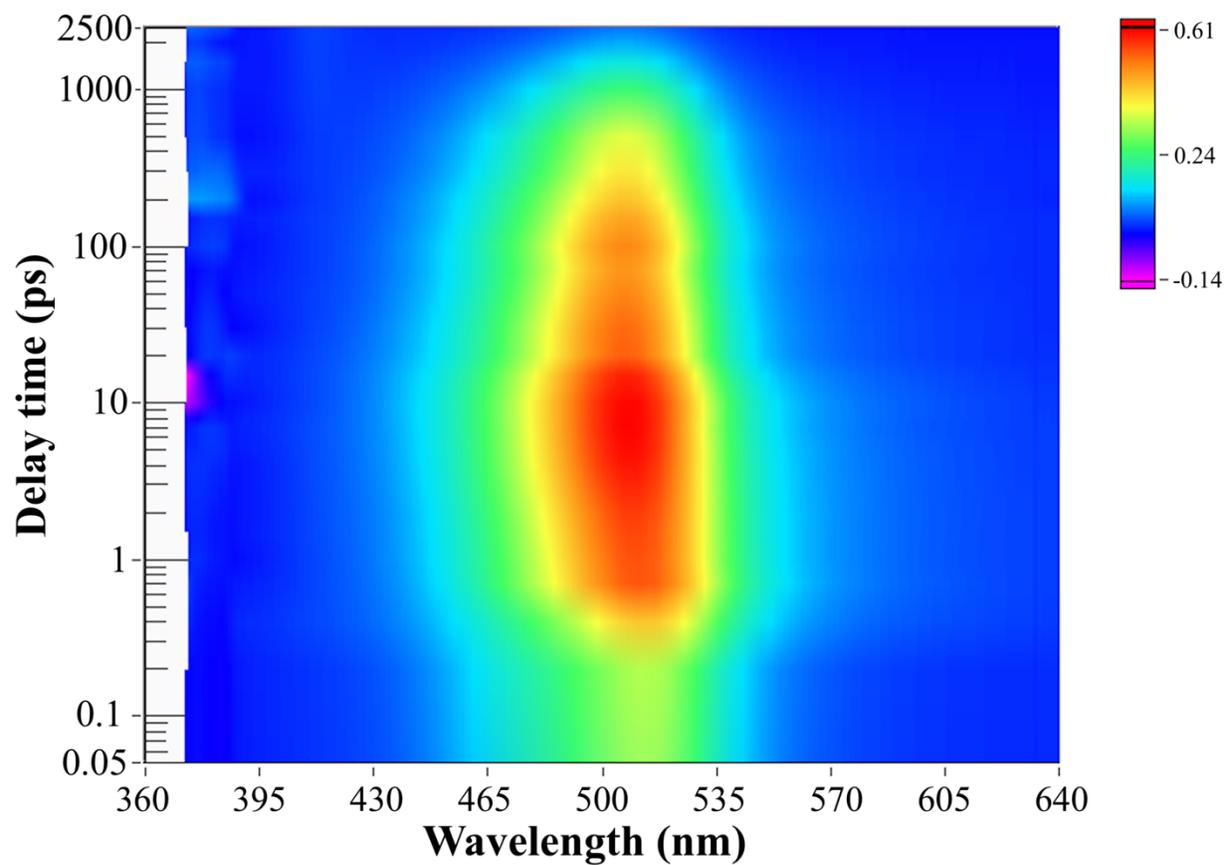
3. Synthesis of 2-biphenyl-5-methyl-1,3,4-oxadiazole (BpOx)

**Scheme S1.** Synthesis of 2-biphenyl-5-methyl-1,3,4-oxadiazole (BpOx)

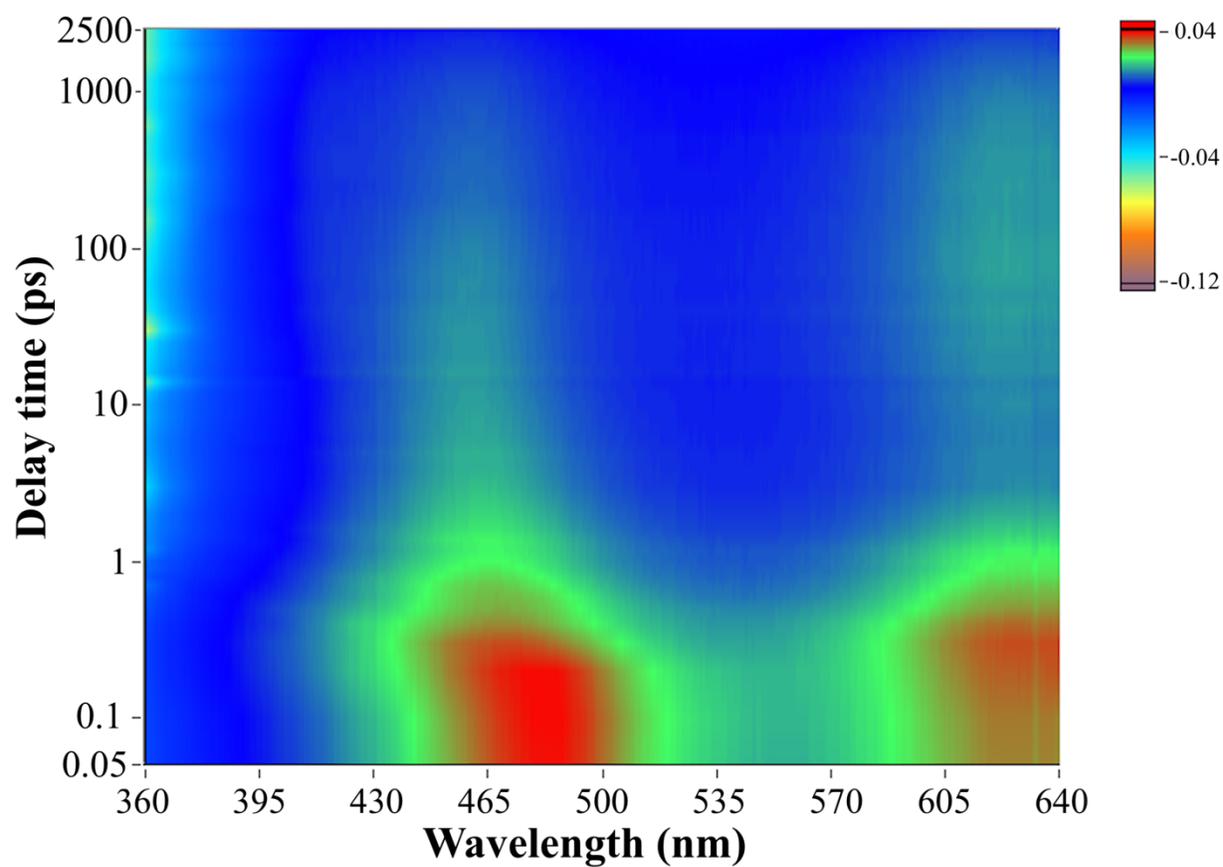
**Scheme S2.** Photolysis of carbonyl azides produces BpOx in acetonitrile (ACN).



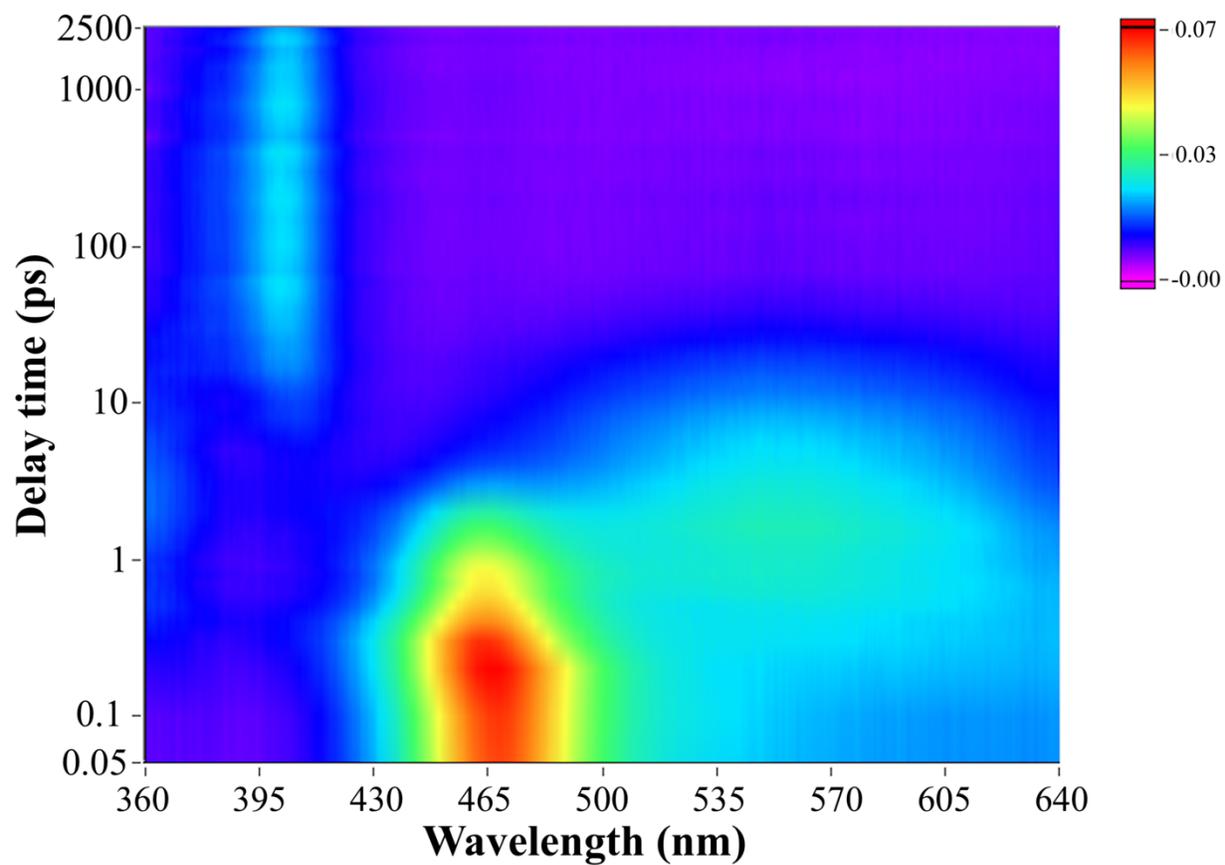
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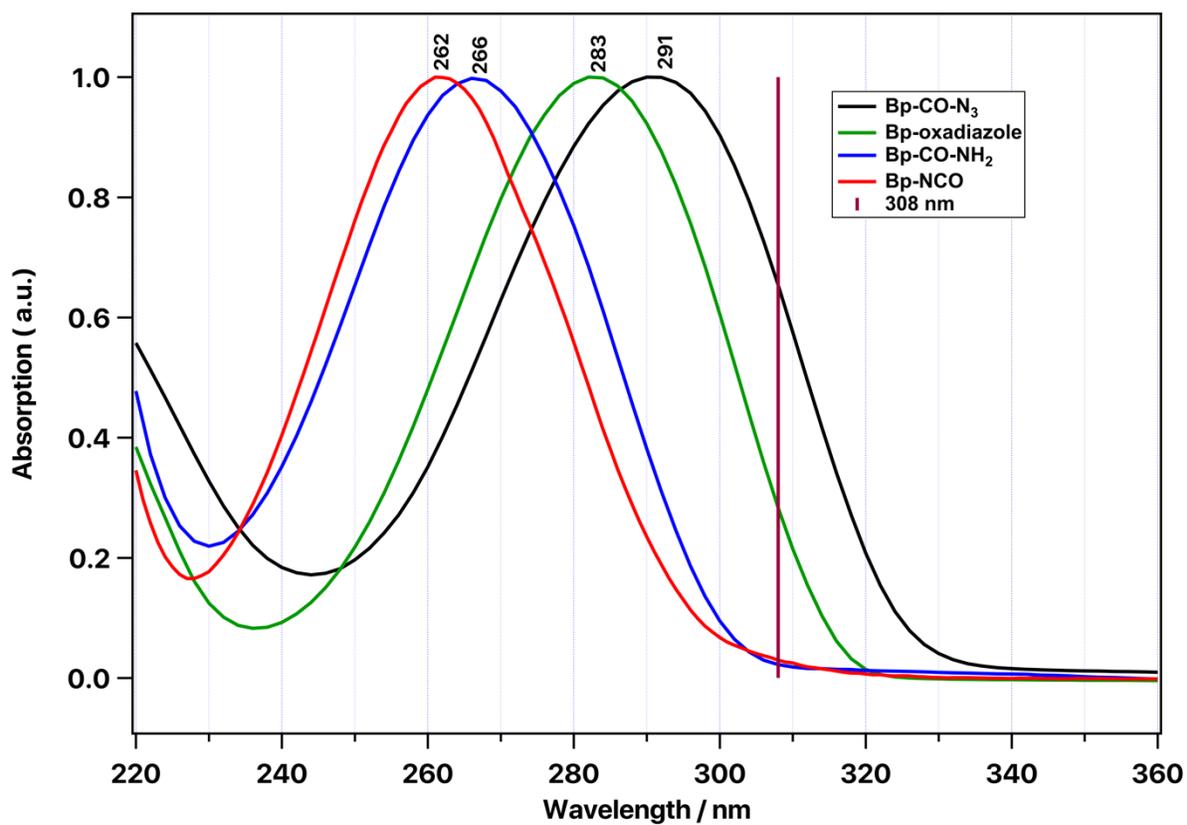
**Fig. S2** 2-D transient absorption spectra of 4-biphenyl oxadiazole in acetonitrile. The pump wavelength is 282 nm.



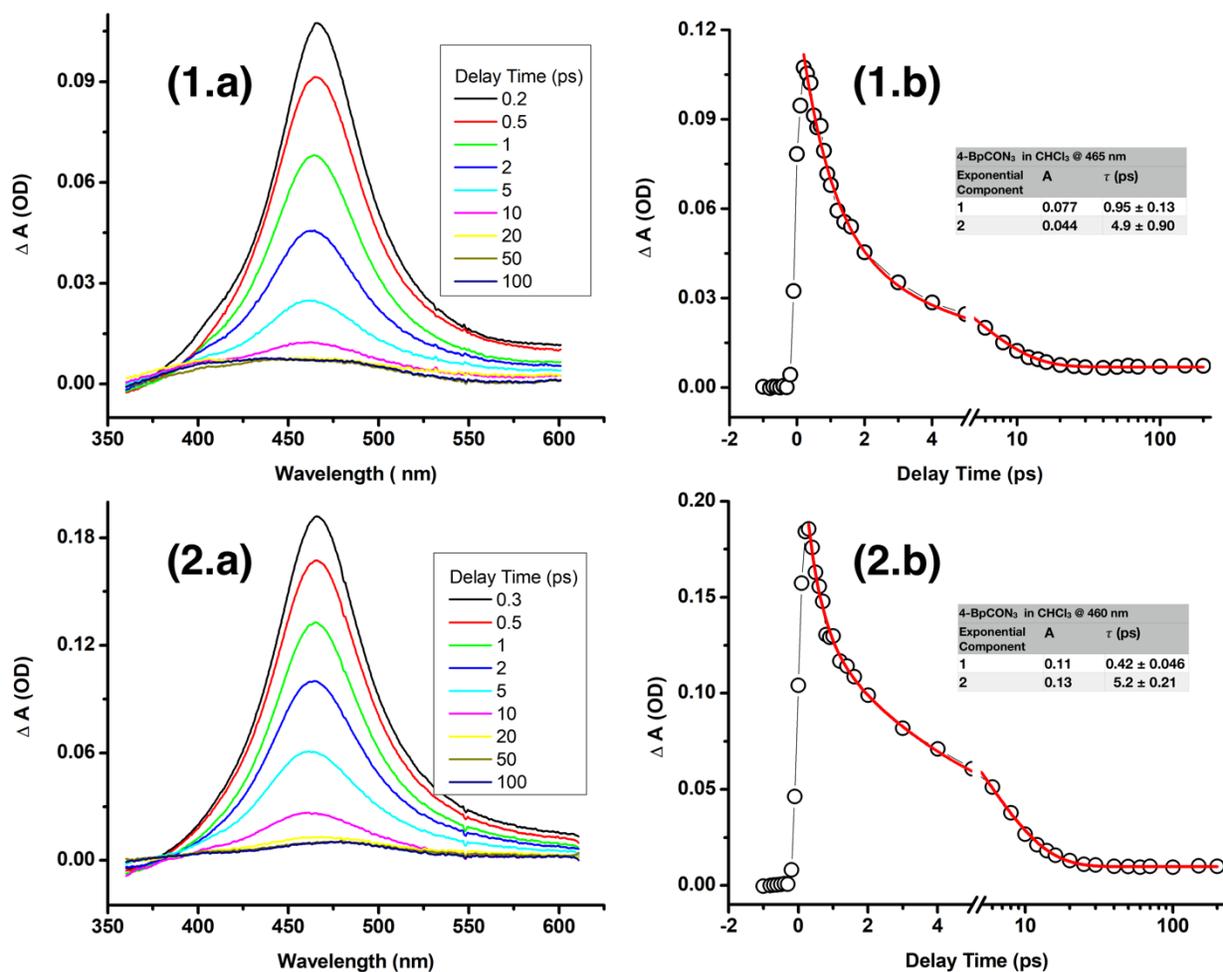
**Fig. S3** 2-D transient absorption spectra of 4-biphenyl isocyanate in acetonitrile. The pump wavelength is 308 nm.



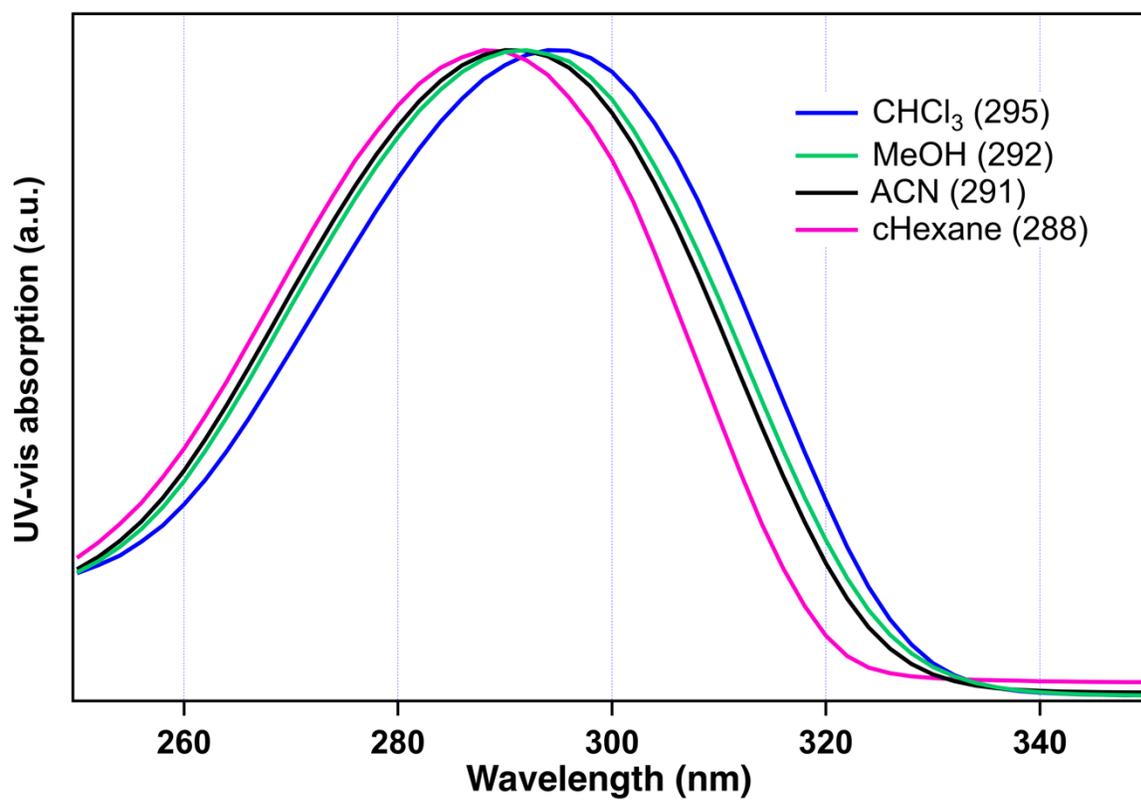
**Fig. S4** 2-D transient absorption spectra of 4-biphenyl carboxamide in acetonitrile. The pump wavelength is 270 nm.



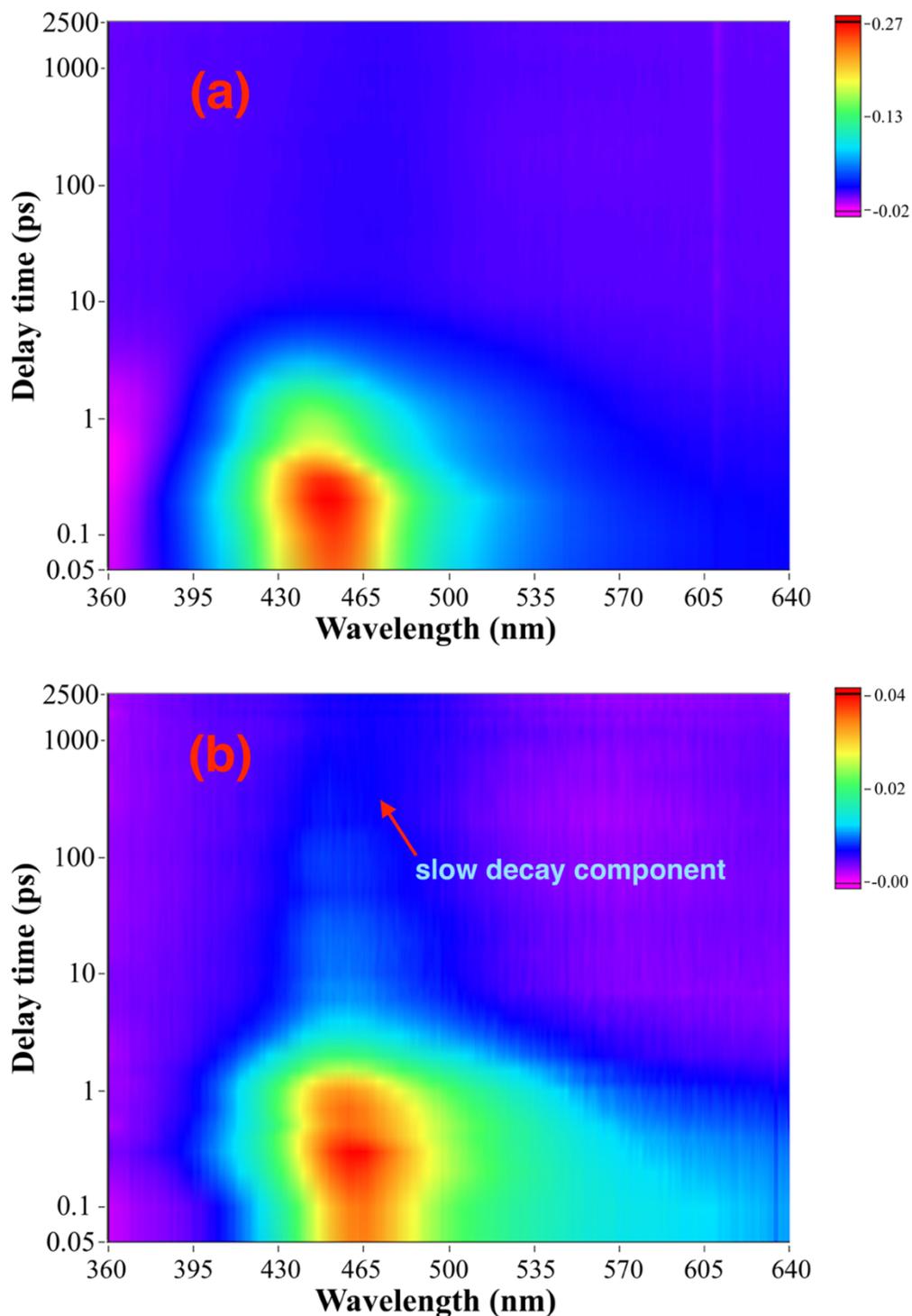
**Fig. S5** Comparison of UV-vis absorption of 4-biphenyl carbonyl azide (4-BpCON<sub>3</sub>) and its photolysis products: 4-biphenyl oxadiazole (BpOx), 4-biphenyl isocyanate (BpNCO), and 4-biphenyl carboxamide (BpCONH<sub>2</sub>) in acetonitrile.



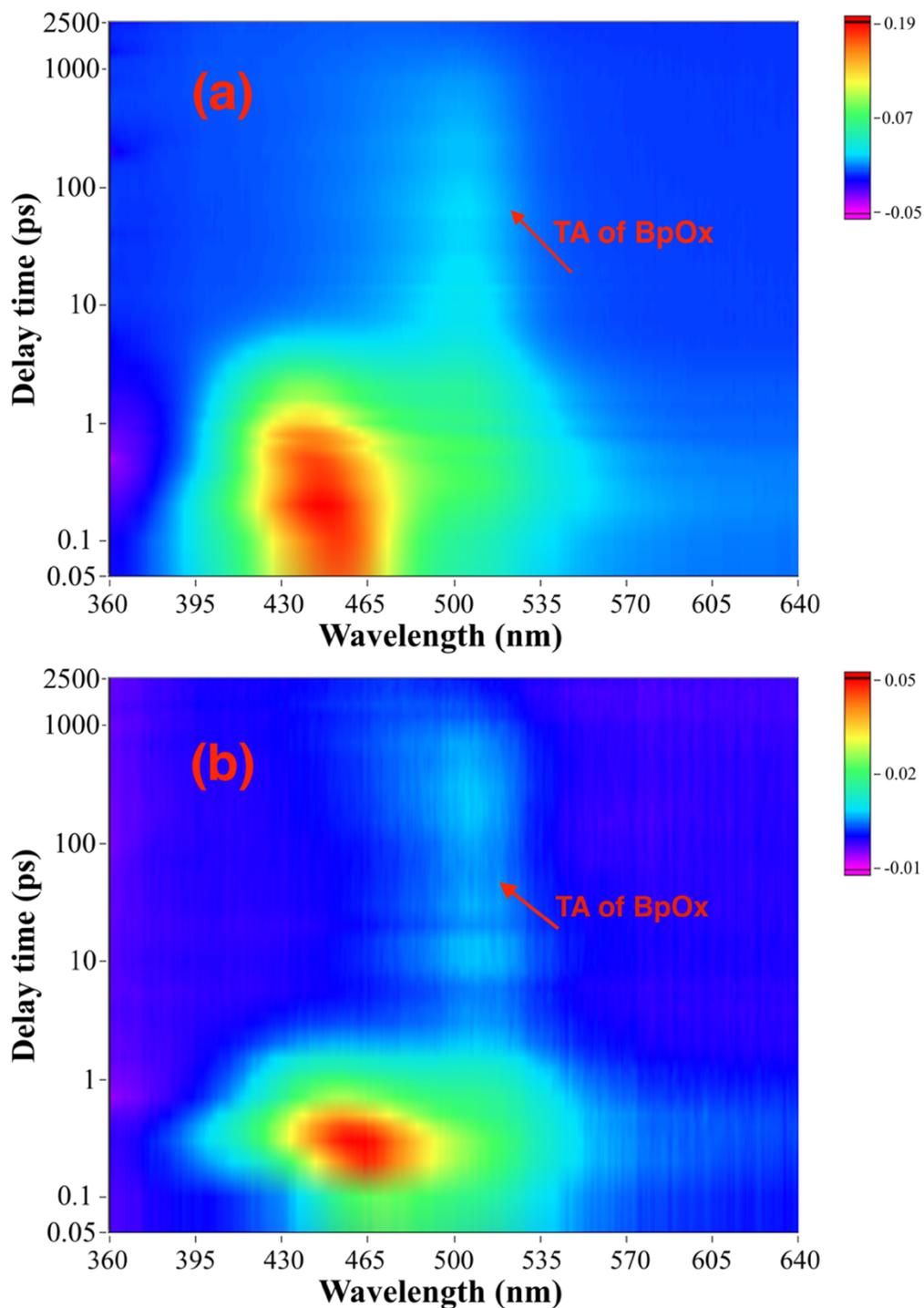
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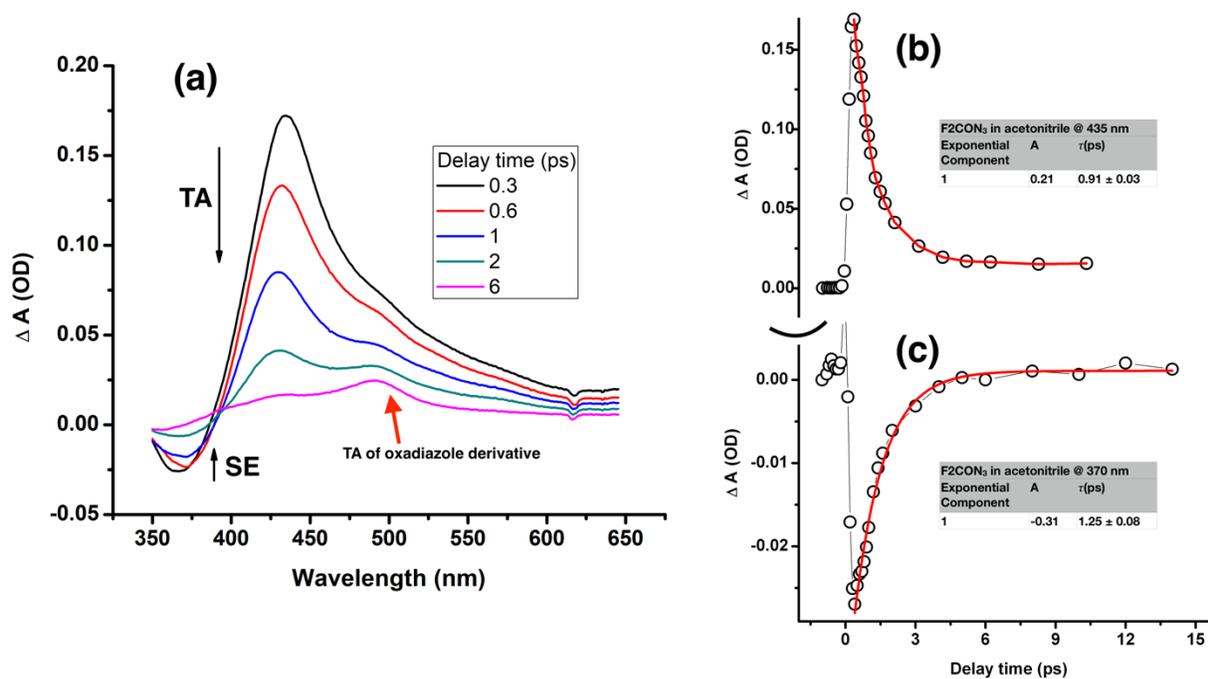
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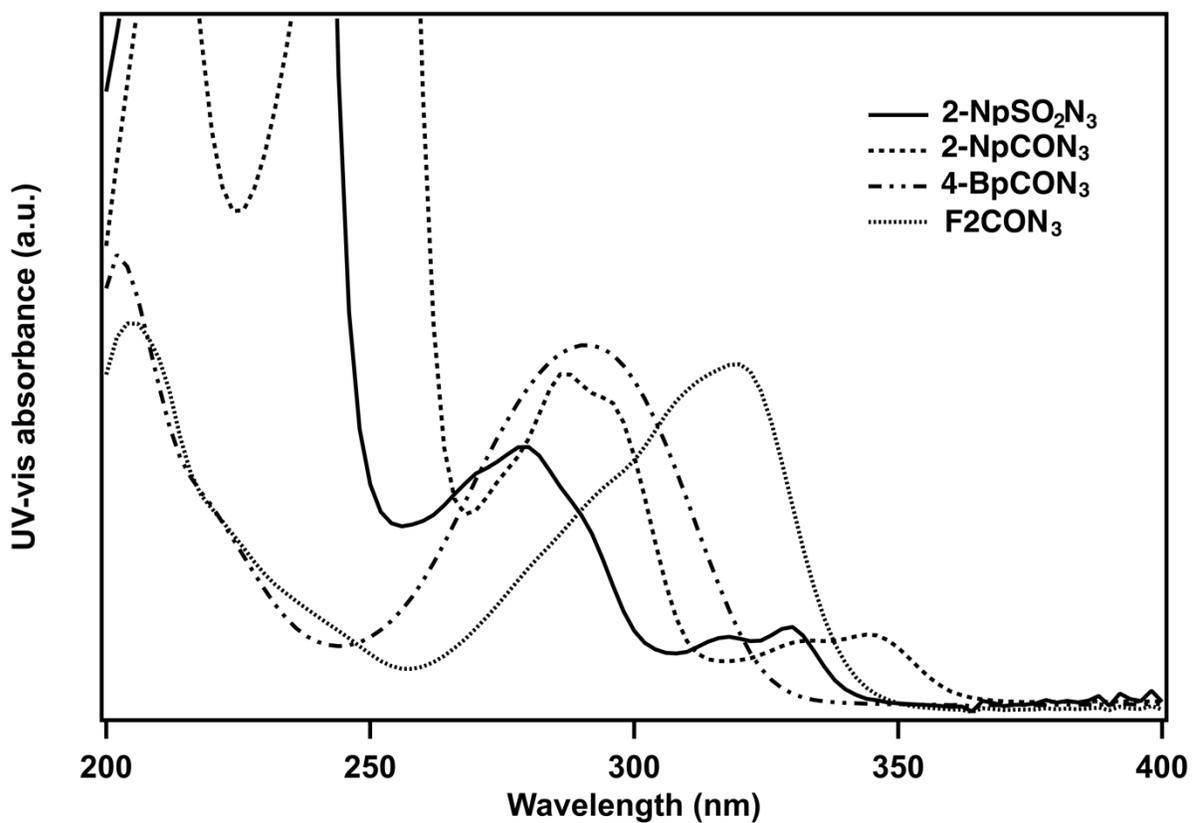
**Fig. S8** 2-D transient absorption spectra of 4-BpCON<sub>3</sub> in MeOH. The pump wavelength is 308 nm (a) and 270 nm (b). Red-shifted TA and a slow decaying component can be seen upon 270 nm excitation. Selected lifetimes upon 270 nm excitation:  $\tau$  (600 nm, decay) =  $1.2 \pm 0.05$  ps;  $\tau$  (450 nm, decay) =  $1.5 \pm 0.1$  ps. Maximum absorption is reached at 0.2 and 0.3 ps with 308 nm excitation and 270 nm excitation respectively.



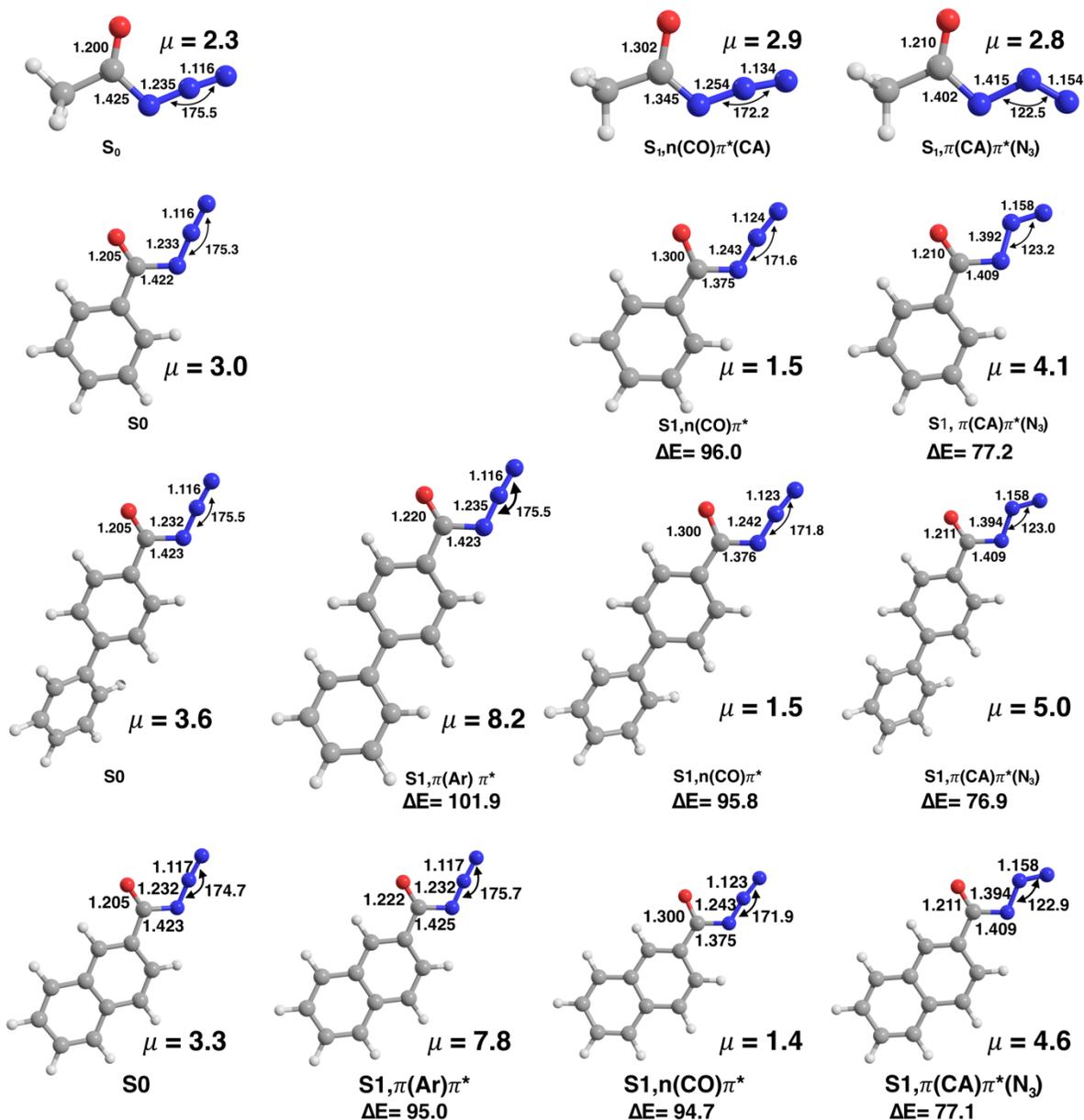
**Fig. S9** 2-D transient absorption spectra of 4-BpCON<sub>3</sub> in acetonitrile. The pump wavelength is 308 nm in (a) and 270 nm in (b) respectively. The transient absorption of 4-biphenyl oxadiazole (BpOx) appears in both cases. Red-shifted TA can be seen upon 270 nm excitation. Selected lifetimes upon 270 nm excitation:  $\tau$  (450 nm, decay) =  $0.83 \pm 0.03$  ps. Maximum absorption is reached at 0.3 and 0.5 ps with 308 nm excitation and 270 nm excitation respectively.



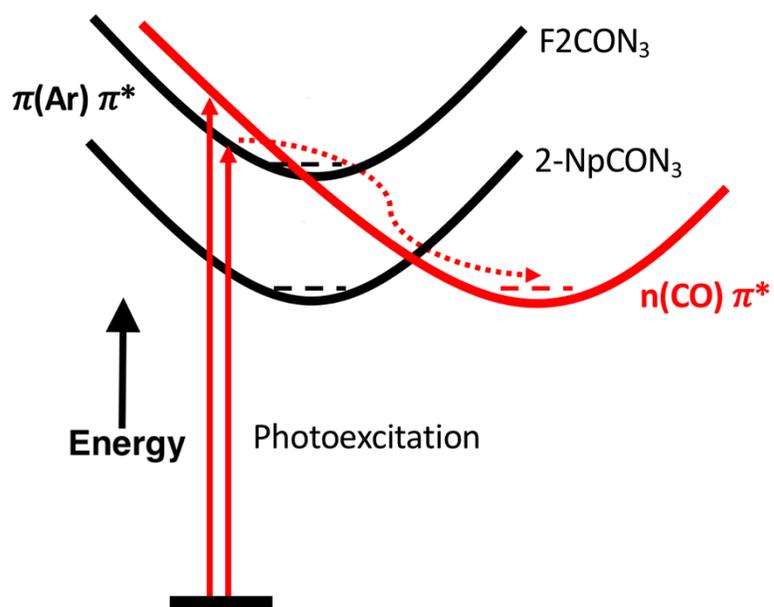
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**Fig. S12** Structural information of MeCON<sub>3</sub>, PhCON<sub>3</sub>, 4-BpCON<sub>3</sub>, and 2-NpCON<sub>3</sub> at their ground states (S<sub>0</sub> structure at PBE0/ def2-TZVP level) and relaxed excited states (S<sub>1</sub>) at cam-B3LYP/ def2-TZVP level. Units: bond length, Å ; NNN angle, degrees(°). The dipole moment (μ) in Debye and ΔE, the relative energy against the ground state (S<sub>0</sub>) in Kcal/mol were calculate at CAM-B3LYP/ def2-TZVP level and are also given.



**Fig. S13** Comparison of potential energy surfaces of  $\pi(\text{Ar})\pi^*$  and  $n(\text{CO})\pi^*$   $S_1$  states of  $\text{F}_2\text{CON}_3$  and  $2\text{-NpCON}_3$ .

**Table S1** TDDFT results of lowest electronic transitions of selected molecules.**1. Biphenyl**

Theory	S1		S2		S3	
	Energy (nm)	f	Energy (nm)	f	Energy (nm)	f
Experimental UV-vis peaks	247.5 <sup>a</sup>					
<b>B3LYP</b>	<b>251</b>	<b>0</b>	<b>247</b>	<b>0</b>	244	0.55
<b>BLYP</b>	<b>273</b>	<b>0</b>	<b>269</b>	<b>0</b>	<b>261</b>	<b>0.51</b>
<b>BP86</b>	<b>271</b>	<b>0</b>	<b>267</b>	<b>0</b>	<b>259</b>	<b>0.52</b>

<sup>a</sup>J.M. Dixon, M. Taniguchi and J. S. Lindsey, *Photochem. Photobiol.*, 2005, **81**, 212-213.

**2. Fluorene**

Theory	S1		S2		S3	
	Energy (nm)	f	Energy (nm)	f	Energy (nm)	f
Experimental observation	<b>296<sup>b</sup></b>					
<b>B3LYP</b>	<b>271</b>	<b>0.088</b>	<b>254</b>	<b>0.47</b>		
PBE0	264	0.11	249	0.46		
$\omega$ B97x	246	0.21	233	0.31		
LC-BLYP	249	0.17	237	0.0040		
TPSS	285	0.043	268	0.0085	262	0.50
<b>BLYP</b>	<b>294</b>	<b>0.039</b>	<b>276</b>	<b>0.0079</b>	<b>268</b>	<b>0.49</b>
<b>BP86</b>	<b>291</b>	<b>0.042</b>	<b>274</b>	<b>0.0085</b>	<b>266</b>	<b>0.50</b>
<b>PBE</b>	<b>291</b>	<b>0.042</b>	<b>274</b>	<b>0.0085</b>	<b>265</b>	<b>0.50</b>

<sup>b</sup>C. A. Pinkham and S. C. Wait Jr, *J. Mol. Spectrosc.*, 1968, **27**, 326–342.

Molecule / Theory	S1		S2		S3	
	Energy (nm)	f	Energy (nm)	f	Energy (nm)	f
<b>3. MeCON<sub>3</sub> / PBE0</b>	<b>258<sup>c</sup></b>	<b>0</b>	<b>237</b>	<b>0</b>		
4. PhCON <sub>3</sub> / PBE0	285 <sup>c</sup>	0	256	0.019		
5. 2-NpCON <sub>3</sub> / <b>B3LYP</b>	<b>322<sup>d</sup></b>	<b>0.062</b>	<b>299<sup>a</sup></b>	<b>0</b>	288	0.15
6. 2-NpCON <sub>3</sub> / PBE0	320 <sup>d</sup>	0.066	293 <sup>a</sup>	0	281	0.14
7. 4- BpNH <sub>2</sub> / PBE0	271 <sup>c</sup>	0.062	256	0.69		
8. 4- BpNCO / PBE0	257	0.44	254	0.49		

MeCON<sub>3</sub> = methyl carbonyl azide; PhCON<sub>3</sub> = benzoyl azide; 2-NpCON<sub>3</sub> = 2-naphthoyl azide; 4-BpNH<sub>2</sub> = 4-biphenyl amide; 4-BpNCO = 4-biphenyl isocyanate.

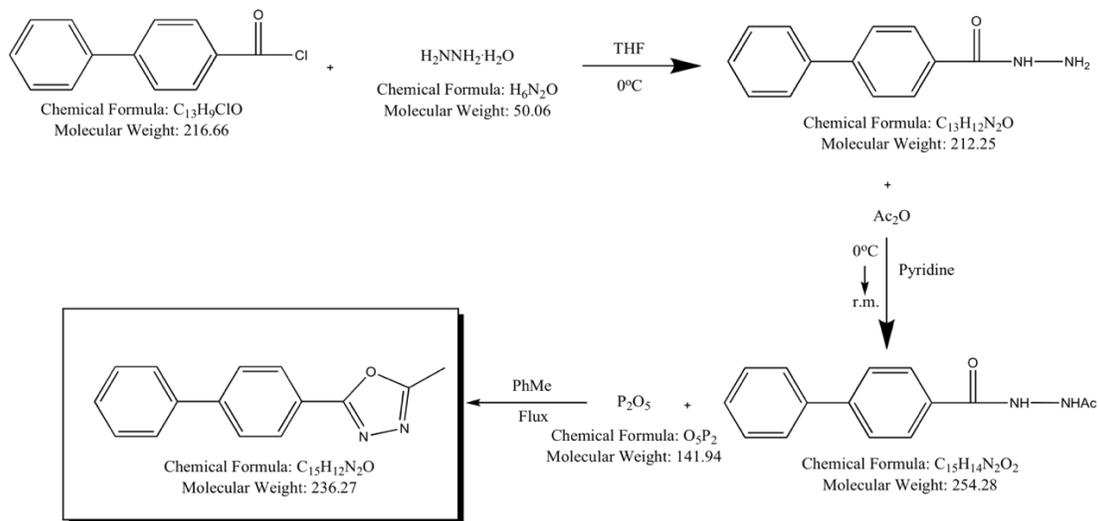
<sup>c</sup>n(CON<sub>3</sub>)-  $\pi^*$  transition;

<sup>d</sup> $\pi$ (naphthalene)- $\pi^*$  transition;

<sup>e</sup>n(amide)-  $\pi^*$  transition.

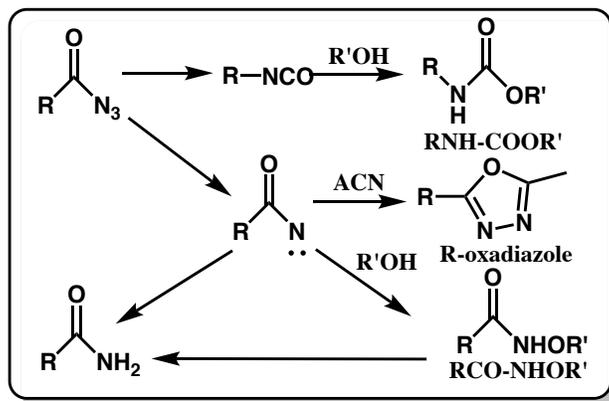
## Synthesis of 2-biphenyl-5-methyl-1,3,4-oxadiazole (BpOx)

The synthesis basically followed the processes (Scheme 1) described in two papers.<sup>1, 2</sup>



### Scheme S1. Synthesis of 2-biphenyl-5-methyl-1,3,4-oxadiazole (BpOx)

The final product was purified by column chromatography on silica gel (eluent: hexanes:AcOEt=1:3 (v:v)).  $^1H$ -NMR (250MHz,  $CDCl_3$ ):  $\delta$  2.62 (3H, s), 7.34-7.50 (3H, m), 7.59-7.65 (2H, m), 7.68-7.74 (2H, m), 8.05-8.11 (2H, m). IR (thin film): 1614, 1589, 1580\*, 1570, 1485\*, 1450, 1412, 1250, 1091, 958, 846, 773, 734  $cm^{-1}$ .



### Scheme S2. Photolysis of carbonyl azides produces BpOx in acetonitrile (ACN).

#### Reference:

- 1 D. H. Christensen, J. T. Nielsen and O. F. Nielsen, *J. Mol. Spectrosc.*, 1967, **24**, 225-234.
- 2 A. Kakefuda, *Biorg. Med. Chem.*, 2002, **10**, 1905-1912.