

**Supplementary Information:**

**Experimental and Computational Studies of Criegee Intermediate Reactions with NH<sub>3</sub> and CH<sub>3</sub>NH<sub>2</sub>**

Rabi Chhantyal-Pun,<sup>1\*</sup> Robin J. Shannon,<sup>1,7</sup> David P. Tew,<sup>2</sup> Rebecca L. Caravan,<sup>3</sup> Marta Duchi,<sup>1</sup> Callum Wong,<sup>1</sup> Aidan Ingham,<sup>1</sup> Charlotte Feldman,<sup>1</sup> Max R. McGillen,<sup>1</sup> M. Anwar H. Khan,<sup>1</sup> Ivan O. Antonov,<sup>3</sup> Brandon Rotavera,<sup>4,5</sup> Krupa Ramasesha,<sup>3</sup> David L. Osborn,<sup>3</sup> Craig A. Taatjes,<sup>3</sup> Carl J. Percival,<sup>6</sup> Dudley E. Shallcross,<sup>1</sup> Andrew J. Orr-Ewing<sup>1</sup>

<sup>1</sup>*School of Chemistry, University of Bristol, Cantock's Close, Bristol BS8 1TS, UK*

<sup>2</sup>*Max Planck Institute for Solid State Research, Heisenbergstraße 1, 70569 Stuttgart, Germany*

<sup>3</sup>*Combustion Research Facility, Mailstop 9055, Sandia National Laboratories, Livermore, California, 94551, USA*

<sup>4</sup>*University of Georgia, College of Engineering, Athens, GA, USA*

<sup>5</sup>*University of Georgia, Department of Chemistry, Athens, GA, USA*

<sup>6</sup>*Jet Propulsion Laboratory, California Institute of Technology, 4800 Oak Grove Drive, Pasadena, CA 91109, USA*

<sup>7</sup>*Mechanical Engineering, Stanford University, Stanford, CA 94305, USA*

**Table of Contents**

Figure S1

Figure S2

Figure S3

Figure S4

Figure S5

Figure S6

Table S1

Table S2

Table S3

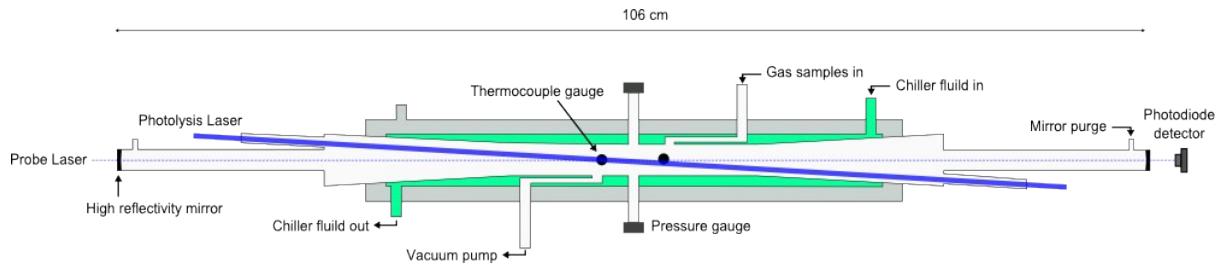
Table S4

Table S5

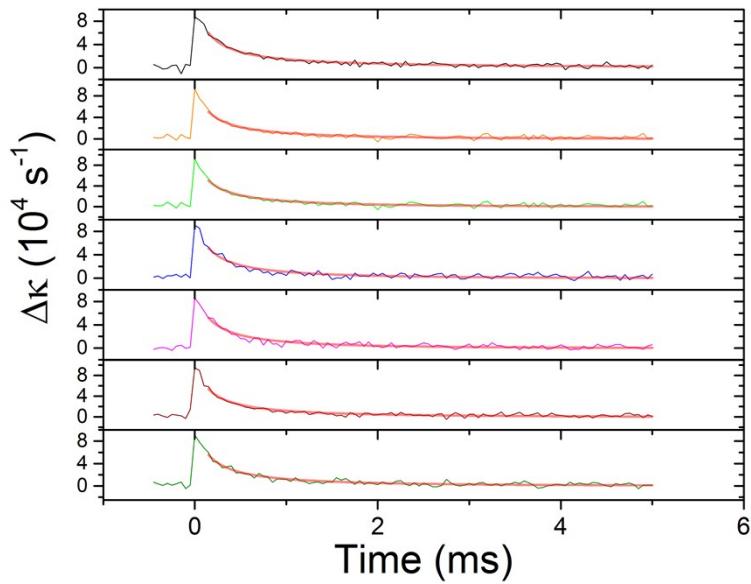
Table S6

Example of MESMER input for CH<sub>2</sub>OO + NH<sub>3</sub> reaction

Example of MEMSER input for CH<sub>2</sub>OO + CH<sub>3</sub>NH<sub>2</sub> reaction



*Figure S1:* Schematic diagram of the variable temperature flow reactor used for measuring reaction of Criegee intermediates with  $\text{NH}_3$  and  $\text{CH}_3\text{NH}_2$ . The bright green layer shows the jacket containing chiller fluid. The outer grey layer is a jacket containing air to insulate the reactor and chiller fluid from ambient temperature. The paths of the photolysis and probe CRDS lasers are shown by solid and dashed lines respectively.



*Figure S2:* Bimolecular reaction of  $(\text{CH}_3)_2\text{COO}$  with  $\text{NH}_3$ . The plot shows  $(\text{CH}_3)_2\text{COO}$  decay traces obtained in the presence of various concentrations of  $\text{NH}_3$ , shown in Figure S3, at 10 Torr total pressure and 293 K. All the traces were corrected for depletion signal of the  $(\text{CH}_3)_2\text{Cl}_2$  precursor. The red traces show fits to equation E1 in the main text. The traces are shown separately for clarity.

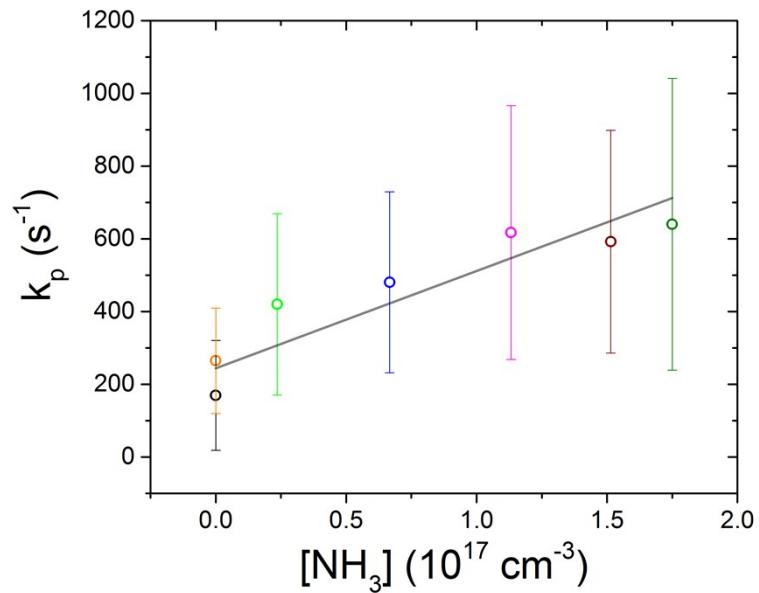


Figure S3: Bimolecular fit for the  $(CH_3)_2COO + NH_3$  reaction. The  $k_p$  values were obtained from the fits shown in Figure S2. The gradient of the linear fit gives the second-order rate coefficient for the reaction.

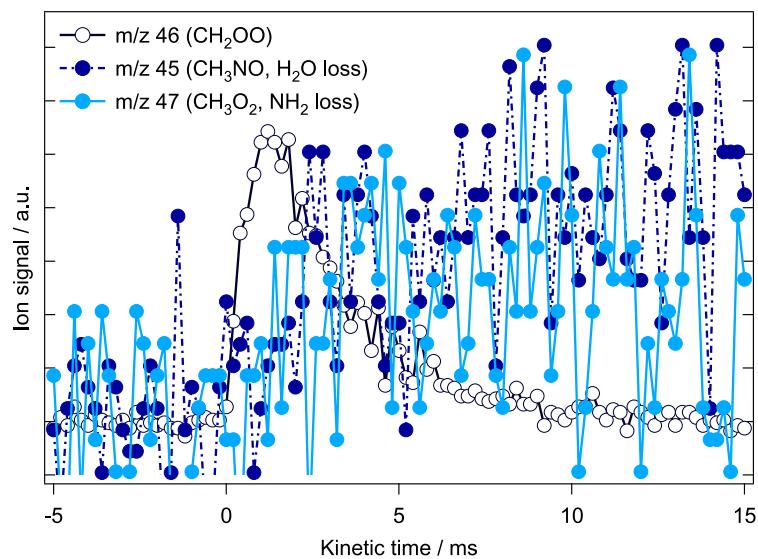


Figure S4: Kinetic profiles of the  $CH_2OO$  Criegee Intermediate (black open circles) in the presence of  $NH_3$ , and proposed daughter ion species from the ionization of predicted functionalized hydroperoxide products. Kinetic profiles were obtained from data recorded at a fixed photoionization energy of 10.5 eV.

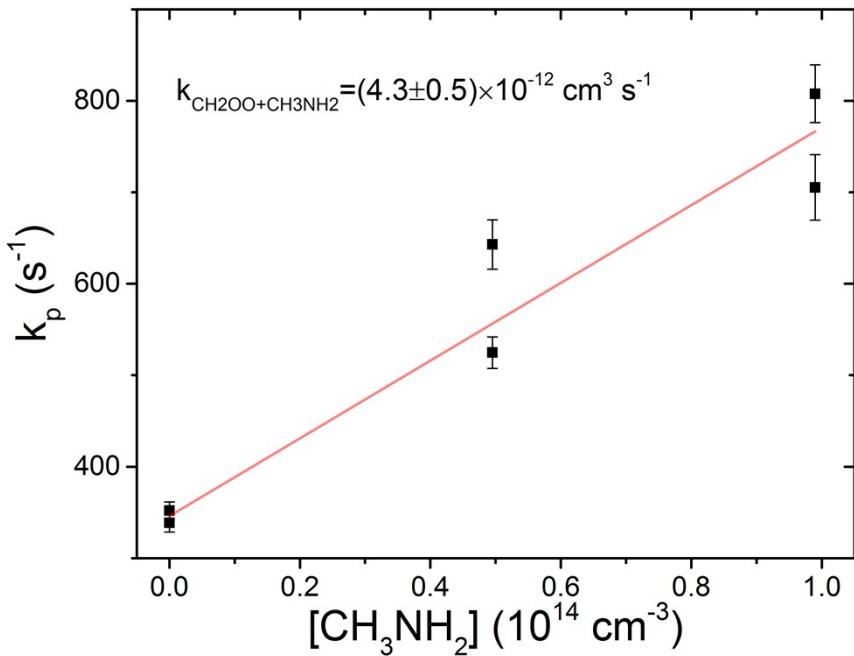


Figure S5: Bimolecular plot obtained from single exponential fits to Criegee Intermediate decays measured at a photoionization energy of 10.5 eV in the presence and absence of  $\text{CH}_3\text{NH}_2$ , yielding a bimolecular rate coefficient of  $(4.3 \pm 0.5) \times 10^{-12} \text{ cm}^3 \text{ s}^{-1}$  ( $1\sigma$  error from the weighted linear fit).

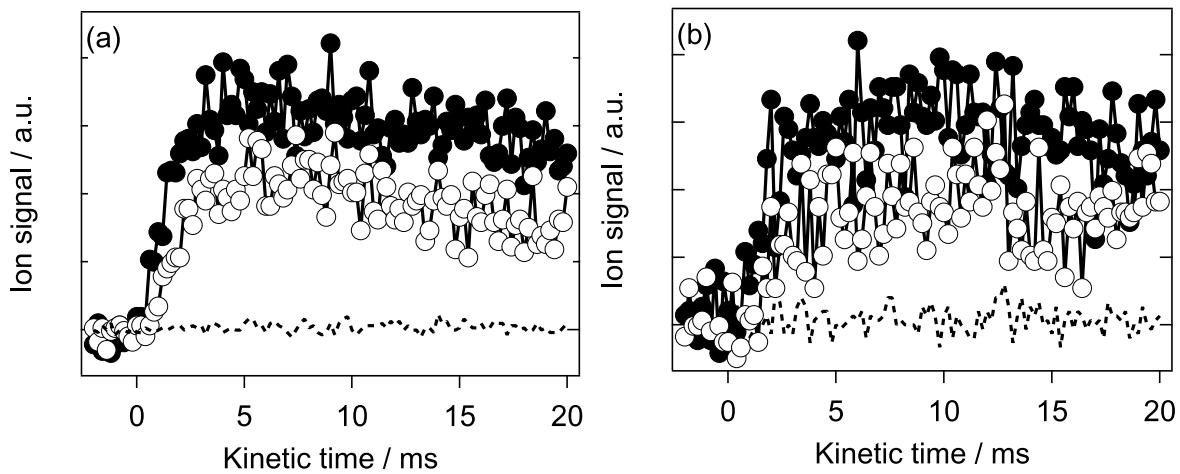


Figure S6: Temporal profiles of (a)  $m/z$  44, and (b)  $m/z$  59 species, from MPIMS measurements for the reaction of  $\text{CH}_2\text{OO}$  with  $\text{CH}_3\text{NH}_2$ . In the plots,  $[\text{CH}_3\text{NH}_2] = 0$  (black dashed line),  $4.95 \times 10^{13} \text{ molecule cm}^{-3}$  (black open circles) and  $9.90 \times 10^{13} \text{ molecule cm}^{-3}$  (black closed circles).

*Table S1: Rate coefficients  $k(p, T)$  for the  $\text{CH}_2\text{OO} + \text{NH}_3$  reaction at various pressures and temperatures. The rate coefficient error shown are  $2\sigma$  values obtained from the weighted linear fit. The temperature error shown is the difference between the inlet and outlet temperature values as shown in Figure S1.*

| Pressure (Torr) | Temperature (K) | $k$ ( $10^{-13} \text{ cm}^3\text{s}^{-1}$ ) |
|-----------------|-----------------|--|
| 10              | $247.6 \pm 1.8$ | $11.5 \pm 0.7$                               |
| 10              | $253.9 \pm 1.4$ | $10.7 \pm 1.1$                               |
| 10              | $258.1 \pm 1.3$ | $10.0 \pm 0.4$                               |
| 10              | $260.6 \pm 1.1$ | $10.1 \pm 0.9$                               |
| 10              | $267.0 \pm 0.9$ | $10.6 \pm 0.9$                               |
| 10              | $276.0 \pm 0.6$ | $8.2 \pm 0.8$                                |
| 10              | $284.5 \pm 0.3$ | $8.0 \pm 0.6$                                |
| 10              | $289.5 \pm 0.1$ | $7.9 \pm 0.4$                                |
| 10              | $293.2 \pm 0.0$ | $8.4 \pm 0.9$                                |
| 10              | $294.4 \pm 0.0$ | $9.2 \pm 0.4$                                |
| 10              | $293.8 \pm 0.1$ | $7.7 \pm 0.6$                                |
| 10              | $294.5 \pm 0.0$ | $8.4 \pm 0.3$                                |
| 10              | $302.1 \pm 0.2$ | $8.1 \pm 0.5$                                |
| 10              | $302.1 \pm 0.2$ | $7.8 \pm 0.2$                                |
| 10              | $302.2 \pm 0.2$ | $8.1 \pm 0.7$                                |
| 10              | $311.3 \pm 0.4$ | $8.4 \pm 0.8$                                |
| 10              | $311.4 \pm 0.4$ | $7.1 \pm 0.6$                                |
| 10              | $320.4 \pm 0.6$ | $7.5 \pm 0.2$                                |
| 10              | $320.2 \pm 0.5$ | $6.8 \pm 0.6$                                |
| 20              | 293             | $8.7 \pm 0.3$                                |
| 45              | 293             | $7.8 \pm 2.0$                                |
| 52              | 293             | $8.6 \pm 0.8$                                |
| 60              | 293             | $7.5 \pm 0.7$                                |
| 70              | 293             | $8.6 \pm 0.8$                                |
| 90              | 293             | $9.1 \pm 1.0$                                |

*Table S2: Rate coefficients  $k(p,T)$  for the  $CH_2OO + CH_3NH_2$  reaction at various pressures and temperatures. The rate coefficient error shown are  $2\sigma$  values obtained from the weighted linear fit. The temperature error shown is the difference between the inlet and outlet temperature values as shown in Figure S1.*

| Pressure (Torr) | Temperature (K) | $k (10^{-12} \text{ cm}^3 \text{s}^{-1})$ |
|-----------------|-----------------|---|
| 10              | $263.0 \pm 1.0$ | $6.64 \pm 0.66$                           |
| 10              | $267.4 \pm 0.8$ | $6.61 \pm 0.60$                           |
| 10              | $275.8 \pm 0.5$ | $6.35 \pm 1.01$                           |
| 10              | $288.5 \pm 0.2$ | $5.52 \pm 0.34$                           |
| 10              | $297.5 \pm 0.1$ | $4.41 \pm 0.70$                           |
| 10              | $311.2 \pm 0.4$ | $3.60 \pm 0.64$                           |
| 50              | $258.2 \pm 1.0$ | $7.40 \pm 0.80$                           |
| 50              | $266.8 \pm 0.8$ | $5.96 \pm 0.61$                           |
| 50              | $275.6 \pm 0.5$ | $5.13 \pm 1.71$                           |
| 50              | $284.4 \pm 0.2$ | $5.34 \pm 1.28$                           |
| 50              | $293.6 \pm 0.0$ | $4.74 \pm 0.29$                           |
| 50              | $302.2 \pm 0.2$ | $4.75 \pm 0.30$                           |
| 50              | $311.3 \pm 0.4$ | $2.80 \pm 1.32$                           |
| 10              | 293             | $5.73 \pm 0.52$                           |
| 20              | 293             | $5.69 \pm 0.28$                           |
| 30              | 293             | $5.85 \pm 0.69$                           |
| 60              | 293             | $5.38 \pm 0.47$                           |
| 100             | 293             | $5.38 \pm 0.33$                           |

*Table S3: Stationary point energies for key structures involved in the CH<sub>2</sub>OO + NH<sub>3</sub> reaction computed at various levels of theory. All the energies and frequencies are computed for CCSD(T)(F12\*)/cc-pVDZ-F12 optimised structures. Values are quoted in kJ mol<sup>-1</sup> relative to the reactants.*

|          | HF+CABS |        |        | dCCSD(F12*)(T) |     |     | ZPVE |      |      | B3LYP | CCSD(T) | Best   |
|----------|---------|--------|--------|----------------|-----|-----|------|------|------|-------|---------|--------|
|          | DZ      | TZ     | QZ     | DZ             | TZ  | QZ  | DZ   | TZ   | QZ   |       |         |        |
| Complex  | -32.3   | -32.2  | -32.1  | 2.4            | 2.4 | 2.4 | 2.1  | 2.3  | 2.4  | 6.7   | 7.7     | -19.6  |
| TS       | -21.3   | -21.0  | -20.9  | 1.9            | 2.0 | 2.1 | 2.7  | 3.0  | 3.2  | 11.2  | 12.8    | -2.7   |
| Products | -222.9  | -222.7 | -222.6 | 0.3            | 0.9 | 1.3 | 12.9 | 14.0 | 14.5 | 20.6  | 22.2    | -184.6 |

*Table S4: Stationary point energies for key structures involved in the CH<sub>2</sub>OO + CH<sub>3</sub>NH<sub>2</sub> reaction computed at various levels of theory. All the energies and frequencies are computed at CCSD(T)(F12\*)/cc-pVDZ-F12 optimised structures. Values are quoted in kJ mol<sup>-1</sup> relative to the reactants.*

|          | HF+CABS |        |      | dCCSD(F12*)(T) |      |      | ZPVE |      | B3LYP | CCSD(T) | Best |
|----------|---------|--------|------|----------------|------|------|------|------|-------|---------|------|
|          | DZ      | TZ     |      | DZ             | TZ   |      | DZ   | TZ   |       |         |      |
| Complex  | -32.3   | -32.1  | -0.4 | -0.6           | 1.6  | 1.7  | 6.8  | 6.6  |       | -24.4   |      |
| TS       | -31.0   | -30.7  | -0.6 | -0.7           | 1.8  | 1.9  | 8.4  | 8.8  |       | -20.7   |      |
| Products | -229.8  | -229.5 | -7.1 | -6.5           | 11.4 | 12.3 | 17.1 | 17.8 |       | -205.9  |      |

*Table S5: Comparison between theoretical rate coefficients from the full master equation treatment  $k_{ME}$  and a steady state treatment  $k_{CAN}$  based on canonical rate coefficients from MESMER for the  $CH_2OO + NH_3$  reaction. Also included are the individual rate coefficients  $k_4$ ,  $k_{-4}$  and  $k_5$ .  $k_4$  is formulated here as a pseudo first order rate coefficient with the master equation excess reactant concentration of  $1 \times 10^{15}$  molecule cm $^{-3}$ .*

| Temperature (K) | $k_{ME}$ (10 $^{-13}$ cm $^3$ s $^{-1}$ ) | $k_{CAN}$ (10 $^{-14}$ cm $^3$ s $^{-1}$ ) | $k_4$ (10 $^{-10}$ cm $^3$ s $^{-1}$ ) | $k_{-4}$ (10 $^{11}$ s $^{-1}$ ) | $k_5$ (10 $^8$ s $^{-1}$ ) |
|-----------------|---|--|--|----------------------------------|----------------------------|
| 247.6           | 10.3                                      | 11.6                                       | 1.01                                   | 1.09                             | 1.26                       |
| 253.9           | 9.97                                      | 11.1                                       | 1.01                                   | 1.38                             | 1.52                       |
| 258.1           | 9.76                                      | 10.8                                       | 1.01                                   | 1.59                             | 1.71                       |
| 260.6           | 9.63                                      | 10.6                                       | 1.01                                   | 1.73                             | 1.83                       |
| 267.0           | 9.36                                      | 10.2                                       | 1.01                                   | 2.13                             | 2.17                       |
| 276.0           | 9.00                                      | 9.74                                       | 1.00                                   | 2.80                             | 2.72                       |
| 284.5           | 8.70                                      | 9.33                                       | 1.00                                   | 3.56                             | 3.31                       |
| 289.5           | 8.53                                      | 9.11                                       | 1.00                                   | 4.07                             | 3.69                       |
| 293.2           | 8.41                                      | 8.96                                       | 1.00                                   | 4.48                             | 4.00                       |
| 293.8           | 8.40                                      | 8.94                                       | 1.00                                   | 4.55                             | 4.05                       |
| 294.4           | 8.38                                      | 8.91                                       | 1.00                                   | 4.62                             | 4.10                       |
| 294.5           | 8.38                                      | 8.91                                       | 1.00                                   | 4.63                             | 4.11                       |
| 302.1           | 8.18                                      | 8.63                                       | 1.00                                   | 5.58                             | 4.79                       |
| 302.1           | 8.18                                      | 8.63                                       | 1.00                                   | 5.58                             | 4.79                       |
| 302.2           | 8.17                                      | 8.63                                       | 1.00                                   | 5.59                             | 4.80                       |
| 311.3           | 7.94                                      | 8.33                                       | 1.00                                   | 6.88                             | 5.71                       |
| 311.4           | 7.93                                      | 8.33                                       | 1.00                                   | 6.90                             | 5.72                       |
| 320.4           | 7.74                                      | 8.08                                       | 1.00                                   | 8.36                             | 6.73                       |

*Table S6: Comparison between theoretical rate coefficients from the full master equation treatment  $k_{ME}$  and a steady state treatment  $k_{CAN}$  based on canonical rate coefficients from MESMER for the  $CH_2OO + CH_3NH_2$  reaction. Also included are the individual rate coefficients  $k_4$ ,  $k_{-4}$  and  $k_5$ .  $k_4$  is formulated here as a pseudo first order rate coefficient with the master equation excess reactant concentration of  $1 \times 10^{15}$  molecule cm $^{-3}$ .*

| Temperature (K) | $k_{ME}$ ( $10^{-12}$ cm $^3$ s $^{-1}$ ) | $k_{CAN}$ ( $10^{-10}$ cm $^3$ s $^{-1}$ ) | $k_4$ ( $10^{-10}$ cm $^3$ s $^{-1}$ ) | $k_{-4}$ ( $10^{11}$ s $^{-1}$ ) | $k_5$ ( $10^{11}$ s $^{-1}$ ) |
|-----------------|---|--|--|----------------------------------|-------------------------------|
| 258.2           | 6.85                                      | 14.9                                       | 1.01                                   | 2.04                             | 3.03                          |
| 263.0           | 6.58                                      | 12.5                                       | 1.01                                   | 2.47                             | 3.06                          |
| 266.8           | 6.37                                      | 10.9                                       | 1.01                                   | 2.86                             | 3.09                          |
| 267.4           | 6.34                                      | 10.7                                       | 1.01                                   | 2.92                             | 3.09                          |
| 275.6           | 5.93                                      | 8.06                                       | 1.01                                   | 3.93                             | 3.15                          |
| 275.8           | 5.91                                      | 8.01                                       | 1.01                                   | 3.96                             | 3.15                          |
| 284.4           | 5.51                                      | 6.10                                       | 1.01                                   | 5.29                             | 3.21                          |
| 288.5           | 5.33                                      | 5.39                                       | 1.01                                   | 6.03                             | 3.23                          |
| 293.6           | 5.12                                      | 4.64                                       | 1.01                                   | 7.06                             | 3.26                          |
| 297.5           | 4.97                                      | 4.16                                       | 1.01                                   | 7.94                             | 3.28                          |
| 302.2           | 4.79                                      | 3.66                                       | 1.01                                   | 9.09                             | 3.31                          |
| 311.2           | 4.47                                      | 2.90                                       | 1.01                                   | 11.6                             | 3.36                          |
| 311.3           | 4.46                                      | 2.89                                       | 1.01                                   | 11.7                             | 3.36                          |

### Example MESMER input for NH<sub>3</sub> + CH<sub>2</sub>OO reaction

```
<?xml version="1.0" encoding="utf-8" ?>
<?xmlstylesheet type='text/xsl' href='../../mesmer2.xsl' media='other'?>
<?xmlstylesheet type='text/xsl' href='../../mesmer1.xsl' media='screen'?>
<me:mesmer xmlns="http://www.xml-cml.org/schema"
    xmlns:me="http://www.chem.leeds.ac.uk/mesmer"
    xmlns:xsi="http://www.w3.org/2001/XMLSchema-instance">
    <me:title>NH3 + Criegee</me:title>
    <moleculeList>
        <molecule id="NH3" xmlns="http://www.xml-cml.org/schema">
            <atomArray>
                <atom id="a1" elementType="N" x3="-0.000004" y3="0.000000" z3="0.067775"/>
                <atom id="a2" elementType="H" x3="0.937101" y3="0.000000" z3="-0.313925"/>
                <atom id="a3" elementType="H" x3="-0.468520" y3="-0.811567" z3="-0.313949"/>
                <atom id="a4" elementType="H" x3="-0.468520" y3="0.811567" z3="-0.313949"/>
            </atomArray>
            <bondArray>
                <bond atomRefs2="a3 a1" order="1"/>
                <bond atomRefs2="a4 a1" order="1"/>
                <bond atomRefs2="a2 a1" order="1"/>
            </bondArray>
            <propertyList>
                <property title="Energy" dictRef="me:ZPE">
                    <scalar units="kJ/mol" >0 </scalar>
                </property>
                <property title="Vibrational Frequencies" dictRef="me:vibFreqs">
                    <array units="cm-1">1060.28 1681.07 1681.40 3486.91 3617.69 3617.71 </array>
                </property>
                <property title="Symmetry Number" dictRef="me:symmetryNumber">
                    <scalar>3 </scalar>
                </property>
            </propertyList>
        </molecule>
    </moleculeList>
</me:mesmer>
```

```

<me:DOSCMETHOD xsi:type="QMRotors"/>
</molecule>
<molecule id="Criegee" xmlns="http://www.xml-cml.org/schema">
  <atomArray>
    <atom id="a1" elementType="C" spinMultiplicity="2" x3="0.000000" y3="0.244135" z3="-1.127237"/>
    <atom id="a2" elementType="O" x3="0.000000" y3="-0.446241" z3="-0.066751"/>
    <atom id="a3" elementType="O" spinMultiplicity="2" x3="0.000000" y3="0.200216" z3="1.107455"/>
    <atom id="a4" elementType="H" x3="0.000000" y3="-0.327047" z3="-2.042958"/>
    <atom id="a5" elementType="H" x3="0.000000" y3="1.323082" z3="-1.043933"/>
  </atomArray>
  <bondArray>
    <bond atomRefs2="a4 a1" order="1"/>
    <bond atomRefs2="a1 a5" order="1"/>
    <bond atomRefs2="a1 a2" order="1"/>
    <bond atomRefs2="a2 a3" order="1"/>
  </bondArray>
  <propertyList>
    <property title="File Format">
      <scalar>mpo</scalar>
    </property>
    <property title="Energy" dictRef="me:ZPE">
      <scalar units="kJ/mol" >0 </scalar>
    </property>
    <property title="SpinMultiplicity" dictRef="me:spinMultiplicity">
      <scalar>1.00 </scalar>
    </property>
    <property title="Vibrational Frequencies" dictRef="me:vibFreqs">
      <array units="cm-1">537.23 657.15 893.88 934.34 1246.76 1340.72 1504.35 3144.13 3305.63
    </array>
    </property>
    <property title="Symmetry Number" dictRef="me:symmetryNumber">
      <scalar>1 </scalar>
    </property>
  </propertyList>

```

```

</property>
</propertyList>

<me:DOSCMETHOD xsi:type="QMRotors"/>
</molecule>

<molecule id="complex" spinMultiplicity="3" xmlns="http://www.xml-cml.org/schema">

  <atomArray>

    <atom id="a1" elementType="C" spinMultiplicity="2" x3="0.250399" y3="-1.127306" z3="-0.581988"/>
    <atom id="a2" elementType="O" x3="-0.375030" y3="-0.118727" z3="-0.988379"/>
    <atom id="a3" elementType="O" spinMultiplicity="2" x3="0.188624" y3="1.101404" z3="-0.737050"/>
    <atom id="a4" elementType="N" x3="-0.049145" y3="-0.006960" z3="2.054251"/>
    <atom id="a5" elementType="H" x3="1.203536" y3="-0.997512" z3="-0.090815"/>
    <atom id="a6" elementType="H" x3="-0.236086" y3="-2.074387" z3="-0.767930"/>
    <atom id="a7" elementType="H" x3="0.027183" y3="0.744284" z3="1.371369"/>
    <atom id="a8" elementType="H" x3="-0.962896" y3="0.073863" z3="2.481565"/>
    <atom id="a9" elementType="H" x3="0.626245" y3="0.185494" z3="2.782749"/>

  </atomArray>

  <bondArray>

    <bond atomRefs2="a2 a3" order="1"/>
    <bond atomRefs2="a2 a1" order="1"/>
    <bond atomRefs2="a6 a1" order="1"/>
    <bond atomRefs2="a1 a5" order="1"/>
    <bond atomRefs2="a7 a4" order="1"/>
    <bond atomRefs2="a4 a8" order="1"/>
    <bond atomRefs2="a4 a9" order="1"/>

  </bondArray>

<propertyList>

  <property title="Energy" dictRef="me:ZPE">
    <scalar units="kJ/mol" >-19.6 </scalar>
  </property>

  <property title="SpinMultiplicity" dictRef="me:spinMultiplicity">
    <scalar>1 </scalar>
  </property>


```

```

</property>

<property title="Vibrational Frequencies" dictRef="me:vibFreqs">
    <array units="cm-1">93.20 154.09 164.88 204.38 215.72 405.11 524.16 666.50 858.68 967.80
1115.94 1243.75 1390.90 1538.72 1674.78 1698.01 3153.88 3308.56 3438.44 3570.44 3614.52
</array>
</property>

<property title="Symmetry Number" dictRef="me:symmetryNumber">
    <scalar>1 </scalar>
</property>

<property dictRef="me:epsilon">
    <scalar>216.11</scalar>
</property>

<property dictRef="me:sigma">
    <scalar>4.6</scalar>
</property>
</propertyList>

<me:energyTransferModel xsi:type="me:ExponentialDown">
    <me:deltaEDown units="cm-1">200.0</me:deltaEDown>
</me:energyTransferModel>
<me:DOSCMETHOD xsi:type="QMRotors"/>
</molecule>

<molecule id="ts">
    <atomArray>
        <atom id="a1" elementType="C" spinMultiplicity="2" x3="0.305252" y3="-1.014577"
z3="0.048045"/>
        <atom id="a2" elementType="N" x3="-0.073590" y3="0.356817" z3="1.660095"/>
        <atom id="a3" elementType="H" x3="-0.019021" y3="1.032415" z3="0.884696"/>
        <atom id="a4" elementType="O" spinMultiplicity="2" x3="0.171771" y3="0.894371" z3="-
1.054695"/>
        <atom id="a5" elementType="O" x3="-0.385896" y3="-0.400282" z3="-0.804257"/>
        <atom id="a6" elementType="H" x3="-0.100201" y3="-1.958675" z3="0.391887"/>
        <atom id="a7" elementType="H" x3="1.347313" y3="-0.753338" z3="0.166000"/>
        <atom id="a8" elementType="H" x3="-1.013988" y3="0.346355" z3="2.030651"/>
        <atom id="a9" elementType="H" x3="0.569906" y3="0.622010" z3="2.392774"/>
    </atomArray>
</molecule>

```

```

</atomArray>

<bondArray>

  <bond atomRefs2="a4 a5" order="1"/>
  <bond atomRefs2="a5 a1" order="1"/>
  <bond atomRefs2="a1 a7" order="1"/>
  <bond atomRefs2="a1 a6" order="1"/>
  <bond atomRefs2="a3 a2" order="1"/>
  <bond atomRefs2="a2 a8" order="1"/>
  <bond atomRefs2="a2 a9" order="1"/>

</bondArray>

<propertyList>

  <property title="Energy" dictRef="me:ZPE">
    <scalar units="kJ/mol" upper="4" lower="-5" stepsize="0.1">-2.1</scalar>
  </property>

  <property title="SpinMultiplicity" dictRef="me:spinMultiplicity">
    <scalar>1</scalar>
  </property>

  <property title="Vibrational Frequencies" dictRef="me:vibFreqs">
    <array units="cm-1">231.91 261.56 330.05 503.13 534.11 668.93 798.12 852.24 1101.91
    1159.49 1224.85 1407.21 1571.15 1645.69 1698.68 3139.17 3259.69 3272.15 3556.91 3638.80
  </array>
  </property>

  <property title="Symmetry Number" dictRef="me:symmetryNumber">
    <scalar> 1 </scalar>
  </property>

  <property title="ImaginaryFrequency" dictRef="me:imFreqs">
    <scalar units="cm-1">268.10 </scalar>
  </property>
</propertyList>

<me:DOSCMETHOD xsi:type="QMRotors"/>

</molecule>

<molecule id="Prod" xmlns="http://www.xml-cml.org/schema">

```

```

<atomArray>
    <atom id="a1" elementType="C" x3="-0.598860" y3="0.315619" z3="-0.624270"/>
    <atom id="a2" elementType="N" x3="0.546101" y3="-0.141447" z3="-1.380468"/>
    <atom id="a3" elementType="O" x3="-0.622512" y3="-0.301953" z3="0.644405"/>
    <atom id="a4" elementType="O" x3="0.569093" y3="0.137825" z3="1.331021"/>
    <atom id="a5" elementType="H" x3="-0.538633" y3="1.398965" z3="-0.507127"/>
    <atom id="a6" elementType="H" x3="-1.564530" y3="0.034244" z3="-1.050987"/>
    <atom id="a7" elementType="H" x3="1.246239" y3="-0.224428" z3="0.738007"/>
    <atom id="a8" elementType="H" x3="0.345165" y3="-0.971067" z3="-1.919111"/>
    <atom id="a9" elementType="H" x3="0.907127" y3="0.572122" z3="-1.994929"/>
</atomArray>

<bondArray>
    <bond atomRefs2="a9 a2" order="1"/>
    <bond atomRefs2="a8 a2" order="1"/>
    <bond atomRefs2="a2 a1" order="1"/>
    <bond atomRefs2="a6 a1" order="1"/>
    <bond atomRefs2="a1 a5" order="1"/>
    <bond atomRefs2="a1 a3" order="1"/>
    <bond atomRefs2="a3 a4" order="1"/>
    <bond atomRefs2="a7 a4" order="1"/>
</bondArray>

<propertyList>
    <property title="Energy" dictRef="me:ZPE">
        <scalar units="kJ/mol" >-184.6</scalar>
    </property>
    <property title="Vibrational Frequencies" dictRef="me:vibFreqs">
        <array units="cm-1">202.96 236.24 388.16 488.72 613.08 788.44 877.18 972.42 1049.49
        1103.50 1273.75 1350.79 1426.39 1432.20 1504.44 1665.62 3046.70 3104.87 3552.90 3646.77
        3697.16 </array>
    </property>
    <property dictRef="me:epsilon">
        <scalar>216.11</scalar>
    </property>

```

```
<property dictRef="me:sigma">
  <scalar>4.6</scalar>
</property>
</propertyList>
<me:energyTransferModel xsi:type="me:ExponentialDown">
  <me:deltaEDown units="cm-1">200.0</me:deltaEDown>
</me:energyTransferModel>
<me:DOSCMETHOD xsi:type="QMRotors"/>
</molecule>
</moleculeList>
<reactionList>
<reaction id="R_1">
  <reactant>
    <molecule ref="NH3" role="excessReactant" />
  </reactant>
  <reactant>
    <molecule ref="Criegee" role="deficientReactant" />
  </reactant>
  <product>
    <molecule ref="complex" role="modelled" />
  </product>
<me:MCRCMethod xsi:type="MesmerLT">
  <me:preExponential>1.00e-10</me:preExponential>
  <me:activationEnergy units="cm-1" >0</me:activationEnergy>
  <me:nInfinity>0.0</me:nInfinity>
</me:MCRCMethod>
<me:excessReactantConc>1.0E15</me:excessReactantConc>
</reaction>
<reaction id="R_2">
  <reactant>
    <molecule ref="complex" role="modelled" />
  </reactant>
```

```

<product>
  <molecule ref="Prod" role="sink" />
</product>
<me:transitionState>
  <molecule ref="ts" role="transitionState" />
</me:transitionState>
<me:MCRCMethod name="SimpleRRKM"/>
<me:tunneling>Eckart</me:tunneling>
</reaction>

</reactionList>
<me:conditions>
  <me:bathGas>N2</me:bathGas>
  <me:PTs>
    <!-- <me:PTpair units="Torr" P="10" T="294.5"><me:experimentalEigenvalue EigenvalueID="1" error="10">91</me:experimentalEigenvalue></me:PTpair>
    <me:PTpair units="Torr" P="70" T="294.5"><me:experimentalEigenvalue EigenvalueID="1" error="8">86</me:experimentalEigenvalue></me:PTpair>
    <me:PTpair units="Torr" P="60" T="294.5"><me:experimentalEigenvalue EigenvalueID="1" error="7">75</me:experimentalEigenvalue></me:PTpair>
    <me:PTpair units="Torr" P="52" T="294.5"><me:experimentalEigenvalue EigenvalueID="1" error="8">86</me:experimentalEigenvalue></me:PTpair>
    <me:PTpair units="Torr" P="45" T="294.5"><me:experimentalEigenvalue EigenvalueID="1" error="20">78</me:experimentalEigenvalue></me:PTpair>
    <me:PTpair units="Torr" P="20" T="294.5"><me:experimentalEigenvalue EigenvalueID="1" error="3">87</me:experimentalEigenvalue></me:PTpair>-->
    <me:PTpair units="Torr" P="10" T="247.6"><me:experimentalEigenvalue EigenvalueID="1" error="7.09">115</me:experimentalEigenvalue></me:PTpair>
    <!--<me:PTpair units="Torr" P="10" T="253.9"><me:experimentalEigenvalue EigenvalueID="1" error="11.3">107</me:experimentalEigenvalue></me:PTpair>
    <me:PTpair units="Torr" P="10" T="258.1"><me:experimentalEigenvalue EigenvalueID="1" error="4.1">100</me:experimentalEigenvalue></me:PTpair>
    <me:PTpair units="Torr" P="10" T="260.6"><me:experimentalEigenvalue EigenvalueID="1" error="8.6">101</me:experimentalEigenvalue></me:PTpair>
    <me:PTpair units="Torr" P="10" T="267.0"><me:experimentalEigenvalue EigenvalueID="1" error="9.0">106</me:experimentalEigenvalue></me:PTpair>

```

```

<me:PTpair units="Torr" P="10" T="276"><me:experimentalEigenvalue EigenvalueID="1"
error="7.61">81.6</me:experimentalEigenvalue></me:PTpair>

<me:PTpair units="Torr" P="10" T="284.5"><me:experimentalEigenvalue EigenvalueID="1"
error="6.22">79.9</me:experimentalEigenvalue></me:PTpair>

<me:PTpair units="Torr" P="10" T="289.5"><me:experimentalEigenvalue EigenvalueID="1"
error="3.52">79.3</me:experimentalEigenvalue></me:PTpair>

<me:PTpair units="Torr" P="10" T="293.2"><me:experimentalEigenvalue EigenvalueID="1"
error="8.9">83.6</me:experimentalEigenvalue></me:PTpair>

<me:PTpair units="Torr" P="10" T="294.4"><me:experimentalEigenvalue EigenvalueID="1"
error="4.06">92.4</me:experimentalEigenvalue></me:PTpair>

<me:PTpair units="Torr" P="10" T="293.8"><me:experimentalEigenvalue EigenvalueID="1"
error="5.78">77.4</me:experimentalEigenvalue></me:PTpair>

<me:PTpair units="Torr" P="10" T="294.5"><me:experimentalEigenvalue EigenvalueID="1"
error="2.69">84.1</me:experimentalEigenvalue></me:PTpair>

<me:PTpair units="Torr" P="10" T="302.1"><me:experimentalEigenvalue EigenvalueID="1"
error="4.87">80.7</me:experimentalEigenvalue></me:PTpair>

<me:PTpair units="Torr" P="10" T="302.1"><me:experimentalEigenvalue EigenvalueID="1"
error="1.49">78.3</me:experimentalEigenvalue></me:PTpair>

<me:PTpair units="Torr" P="10" T="302.2"><me:experimentalEigenvalue EigenvalueID="1"
error="7.89">80.7</me:experimentalEigenvalue></me:PTpair>

<me:PTpair units="Torr" P="10" T="311.3"><me:experimentalEigenvalue EigenvalueID="1"
error="7.89">83.7</me:experimentalEigenvalue></me:PTpair>

<me:PTpair units="Torr" P="10" T="311.4"><me:experimentalEigenvalue EigenvalueID="1"
error="5.58">71.3</me:experimentalEigenvalue></me:PTpair>

<me:PTpair units="Torr" P="10" T="320.4"><me:experimentalEigenvalue EigenvalueID="1"
error="9.0">75.0</me:experimentalEigenvalue></me:PTpair>-->

<me:PTpair units="Torr" P="10" T="320.2"><me:experimentalEigenvalue EigenvalueID="1"
error="6.38">68.4</me:experimentalEigenvalue></me:PTpair>

</me:PTs>

</me:conditions>

<me:modelParameters>

<me:grainSize units="cm-1">10</me:grainSize>

<me:energyAboveTheTopHill>30.</me:energyAboveTheTopHill>

</me:modelParameters>

<me:control>

<!--<me:calcMethod xsi:type="me:marquardt">
<me:MarquardtIterations>10</me:MarquardtIterations>

```

```
<me:MarquardtTolerance>0.1</me:MarquardtTolerance>
<me:MarquardtDerivDelta>1.e-02</me:MarquardtDerivDelta>
</me:calcMethod>-->
<me:printSpeciesProfile/>
<me:testRateConstants/>
<me:printGrainedSpeciesProfile/>
<me:eigenvalues>3</me:eigenvalues>
</me:control>
</me:mesmer>
```

### Example MESMER input for CH<sub>2</sub>OO + CH<sub>3</sub>NH<sub>2</sub>

```
<?xml version="1.0" encoding="utf-8" ?>
<?xmlstylesheet type='text/xsl' href='../../mesmer2.xsl' media='other'?>
<?xmlstylesheet type='text/xsl' href='../../mesmer1.xsl' media='screen'?>
<me:mesmer xmlns="http://www.xml-cml.org/schema"
  xmlns:me="http://www.chem.leeds.ac.uk/mesmer"
  xmlns:xsi="http://www.w3.org/2001/XMLSchema-instance">
  <me:title>NH2CH3 + Criegee</me:title>
  <moleculeList>
    <molecule id="NH2CH3" xmlns="http://www.xml-cml.org/schema">
      <atomArray>
        <atom id="a1" elementType="N" x3="0.000000" y3="-0.071919" z3="-0.721253"/>
        <atom id="a2" elementType="C" x3="0.000000" y3="0.013616" z3="0.740309"/>
        <atom id="a3" elementType="H" x3="0.000000" y3="1.033003" z3="1.141405"/>
        <atom id="a4" elementType="H" x3="-0.809210" y3="0.404508" z3="-1.097147"/>
        <atom id="a5" elementType="H" x3="0.809210" y3="0.404508" z3="-1.097147"/>
        <atom id="a6" elementType="H" x3="0.878164" y3="-0.502428" z3="1.126941"/>
        <atom id="a7" elementType="H" x3="-0.878164" y3="-0.502428" z3="1.126941"/>
      </atomArray>
      <bondArray>
        <bond atomRefs2="a4 a1" order="1"/>
        <bond atomRefs2="a5 a1" order="1"/>
        <bond atomRefs2="a1 a2" order="1"/>
        <bond atomRefs2="a2 a6" order="1"/>
        <bond atomRefs2="a2 a7" order="1"/>
        <bond atomRefs2="a2 a3" order="1"/>
      </bondArray>
      <propertyList>
        <property title="Energy" dictRef="me:ZPE">
          <scalar units="kJ/mol" >0 </scalar>
        </property>
```

```

<property title="Vibrational Frequencies" dictRef="me:vibFreqs">
    <array units="cm-1">299.68 855.21 978.12 1072.12 1184.05 1360.35 1467.84 1514.18
1532.85 1672.06 3006.00 3088.21 3125.41 3516.44 3598.61 </array>
</property>
<property title="Symmetry Number" dictRef="me:symmetryNumber">
    <scalar>3 </scalar>
</property>
</propertyList>
<me:DOSCMETHOD xsi:type="QMROtors"/>
</molecule>
<molecule id="Criegee" spinMultiplicity="3" xmlns="http://www.xml-cml.org/schema">
    <atomArray>
        <atom id="a1" elementType="C" spinMultiplicity="2" x3="0.000000" y3="0.244135" z3="-1.127237"/>
        <atom id="a2" elementType="O" x3="0.000000" y3="-0.446241" z3="-0.066751"/>
        <atom id="a3" elementType="O" spinMultiplicity="2" x3="0.000000" y3="0.200216" z3="1.107455"/>
        <atom id="a4" elementType="H" x3="0.000000" y3="-0.327047" z3="-2.042958"/>
        <atom id="a5" elementType="H" x3="0.000000" y3="1.323082" z3="-1.043933"/>
    </atomArray>
    <bondArray>
        <bond atomRefs2="a4 a1" order="1"/>
        <bond atomRefs2="a1 a5" order="1"/>
        <bond atomRefs2="a1 a2" order="1"/>
        <bond atomRefs2="a2 a3" order="1"/>
    </bondArray>
    <propertyList>
        <property title="File Format">
            <scalar>mpo</scalar>
        </property>
        <property title="Energy" dictRef="me:ZPE">
            <scalar units="kJ/mol" >0 </scalar>
        </property>
        <property title="SpinMultiplicity" dictRef="me:spinMultiplicity">

```

```

<scalar>1.00 </scalar>
</property>
<property title="Vibrational Frequencies" dictRef="me:vibFreqs">
    <array units="cm-1">537.23 657.15 893.88 934.34 1246.76 1340.72 1504.35 3144.13
3305.63 </array>
</property>
<property title="Symmetry Number" dictRef="me:symmetryNumber">
    <scalar>1 </scalar>
</property>
</propertyList>
<me:DOSCMETHOD xsi:type="QMRotors"/>
</molecule>
<molecule id="complex" spinMultiplicity="3" xmlns="http://www.xml-cml.org/schema">
    <atomArray>
        <atom id="a1" elementType="C" spinMultiplicity="2" x3="0.149150" y3="-1.146497" z3="-
1.005727"/>
        <atom id="a2" elementType="O" x3="-0.512399" y3="-0.105892" z3="-1.227519"/>
        <atom id="a3" elementType="O" spinMultiplicity="2" x3="0.209596" y3="1.062706" z3="-
1.312140"/>
        <atom id="a4" elementType="N" x3="0.689074" y3="-0.094722" z3="1.405583"/>
        <atom id="a5" elementType="H" x3="1.227292" y3="-1.094129" z3="-0.978777"/>
        <atom id="a6" elementType="H" x3="-0.433318" y3="-2.050174" z3="-0.890964"/>
        <atom id="a7" elementType="H" x3="0.865898" y3="0.667027" z3="0.755116"/>
        <atom id="a8" elementType="H" x3="1.454542" y3="-0.110317" z3="2.066177"/>
        <atom id="a9" elementType="C" x3="-0.578106" y3="0.149459" z3="2.098288"/>
        <atom id="a10" elementType="H" x3="-0.783538" y3="-0.672071" z3="2.784485"/>
        <atom id="a11" elementType="H" x3="-0.606730" y3="1.086227" z3="2.662858"/>
        <atom id="a12" elementType="H" x3="-1.381652" y3="0.182951" z3="1.362163"/>
    </atomArray>
    <bondArray>
        <bond atomRefs2="a3 a2" order="1"/>
        <bond atomRefs2="a2 a1" order="1"/>
        <bond atomRefs2="a1 a5" order="1"/>
        <bond atomRefs2="a1 a6" order="1"/>
    
```

```

<bond atomRefs2="a7 a4" order="1"/>
<bond atomRefs2="a12 a9" order="1"/>
<bond atomRefs2="a4 a8" order="1"/>
<bond atomRefs2="a4 a9" order="1"/>
<bond atomRefs2="a9 a11" order="1"/>
<bond atomRefs2="a9 a10" order="1"/>
</bondArray>
<propertyList>
  <property title="Energy" dictRef="me:ZPE">
    <scalar units="kJ/mol" >-24.4 </scalar>
  </property>
  <property title="SpinMultiplicity" dictRef="me:spinMultiplicity">
    <scalar>1 </scalar>
  </property>
  <property title="Vibrational Frequencies" dictRef="me:vibFreqs">
    <array units="cm-1">62.51 73.02 112.67 145.51 170.33 259.24 435.26 518.95 675.92
842.27 922.84 981.86 1006.80 1072.15 1193.07 1237.60 1372.72 1396.67 1466.08 1511.60 1534.17
1546.37 1670.77 3018.45 3087.30 3120.96 3153.19 3306.21 3461.44 3578.81 </array>
  </property>
  <property title="Symmetry Number" dictRef="me:symmetryNumber">
    <scalar>3 </scalar>
  </property>
  <property dictRef="me:epsilon">
    <scalar>216.11</scalar>
  </property>
  <property dictRef="me:sigma">
    <scalar>4.6</scalar>
  </property>
</propertyList>
<me:energyTransferModel xsi:type="me:ExponentialDown">
  <me:deltaEDown units="cm-1">200.0</me:deltaEDown>
</me:energyTransferModel>
<me:DOSCMETHOD xsi:type="QMRotors"/>

```

```

</molecule>

<molecule id="ts">

  <atomArray>

    <atom id="a1" elementType="C" spinMultiplicity="2" x3="0.175668" y3="-1.127368" z3="-0.724912"/>
    <atom id="a2" elementType="N" x3="0.676143" y3="-0.008928" z3="1.251126"/>
    <atom id="a3" elementType="H" x3="0.799159" y3="0.742915" z3="0.571608"/>
    <atom id="a4" elementType="O" spinMultiplicity="2" x3="0.187169" y3="0.976665" z3="-1.436450"/>
    <atom id="a5" elementType="O" x3="-0.547285" y3="-0.190499" z3="-1.136676"/>
    <atom id="a6" elementType="H" x3="1.242831" y3="-1.081158" z3="-0.883594"/>
    <atom id="a7" elementType="H" x3="-0.346623" y3="-2.017418" z3="-0.397454"/>
    <atom id="a8" elementType="H" x3="1.498671" y3="-0.055064" z3="1.836637"/>
    <atom id="a9" elementType="C" x3="-0.535157" y3="0.223997" z3="2.037087"/>
    <atom id="a10" elementType="H" x3="-0.671478" y3="-0.586493" z3="2.752419"/>
    <atom id="a11" elementType="H" x3="-1.391022" y3="0.234812" z3="1.361508"/>
    <atom id="a12" elementType="H" x3="-0.527391" y3="1.172290" z3="2.580586"/>

  </atomArray>

  <bondArray>

    <bond atomRefs2="a4 a5" order="1"/>
    <bond atomRefs2="a5 a1" order="1"/>
    <bond atomRefs2="a6 a1" order="1"/>
    <bond atomRefs2="a1 a7" order="1"/>
    <bond atomRefs2="a3 a2" order="1"/>
    <bond atomRefs2="a2 a8" order="1"/>
    <bond atomRefs2="a2 a9" order="1"/>
    <bond atomRefs2="a11 a9" order="1"/>
    <bond atomRefs2="a9 a12" order="1"/>
    <bond atomRefs2="a9 a10" order="1"/>

  </bondArray>

  <propertyList>

    <property title="Energy" dictRef="me:ZPE">
      <scalar units="kJ/mol" upper="-16" lower="-24" stepsize="0.1">-22.4</scalar>
    
```

```

</property>

<property title="SpinMultiplicity" dictRef="me:spinMultiplicity">
    <scalar>1</scalar>
</property>

<property title="Vibrational Frequencies" dictRef="me:vibFreqs">
    <array units="cm-1">84.58 137.59 160.34 207.13 383.09 505.75 547.00 751.77 810.20
945.15 1029.10 1047.81 1081.63 1194.73 1226.66 1369.63 1407.11 1466.77 1511.56 1534.72
1567.86 1671.30 3028.48 3095.00 3126.14 3146.19 3289.59 3402.66 3582.76 </array>
</property>

<property title="Symmetry Number" dictRef="me:symmetryNumber">
    <scalar> 3 </scalar>
</property>

<property title="ImaginaryFrequency" dictRef="me:imFreqs">
    <scalar units="cm-1">151.41 </scalar>
</property>
</propertyList>

<me:DOSCMETHOD xsi:type="QMROtors"/>
</molecule>

<molecule id="Prod" xmlns="http://www.xml-cml.org/schema">

    <atomArray>
        <atom id="a1" elementType="C" x3="0.326883" y3="-0.857870" z3="-0.016248"/>
        <atom id="a2" elementType="N" x3="0.580908" y3="0.194868" z3="0.936006"/>
        <atom id="a3" elementType="O" x3="-0.569028" y3="-0.405500" z3="-1.008532"/>
        <atom id="a4" elementType="O" x3="0.110717" y3="0.660588" z3="-1.709198"/>
        <atom id="a5" elementType="H" x3="1.269445" y3="-1.165155" z3="-0.473908"/>
        <atom id="a6" elementType="H" x3="-0.193976" y3="-1.723571" z3="0.411080"/>
        <atom id="a7" elementType="H" x3="0.303897" y3="1.251480" z3="-0.962448"/>
        <atom id="a8" elementType="C" x3="-0.451883" y3="0.326366" z3="1.959579"/>
        <atom id="a9" elementType="H" x3="1.498292" y3="0.092734" z3="1.344294"/>
        <atom id="a10" elementType="H" x3="-0.212315" y3="1.161486" z3="2.615580"/>
        <atom id="a11" elementType="H" x3="-1.403041" y3="0.538671" z3="1.473380"/>
        <atom id="a12" elementType="H" x3="-0.570330" y3="-0.579104" z3="2.566987"/>
    </atomArray>

```

```

<bondArray>
  <bond atomRefs2="a4 a3" order="1"/>
  <bond atomRefs2="a4 a7" order="1"/>
  <bond atomRefs2="a3 a1" order="1"/>
  <bond atomRefs2="a5 a1" order="1"/>
  <bond atomRefs2="a1 a6" order="1"/>
  <bond atomRefs2="a1 a2" order="1"/>
  <bond atomRefs2="a2 a9" order="1"/>
  <bond atomRefs2="a2 a8" order="1"/>
  <bond atomRefs2="a11 a8" order="1"/>
  <bond atomRefs2="a8 a12" order="1"/>
  <bond atomRefs2="a8 a10" order="1"/>
</bondArray>
<propertyList>
  <property title="Energy" dictRef="me:ZPE">
    <scalar units="kJ/mol" >-205.9</scalar>
  </property>
  <property title="Vibrational Frequencies" dictRef="me:vibFreqs">
    <array units="cm-1">101.31 164.96 207.79 315.55 451.99 479.03 613.25 754.87 873.78
938.75 1050.02 1070.42 1154.26 1173.22 1230.95 1317.07 1408.78 1436.24 1461.22 1486.56
1498.62 1513.25 1531.48 2999.32 3010.65 3088.81 3099.11 3140.38 3580.64 3670.70 </array>
  </property>
  <property dictRef="me:epsilon">
    <scalar>216.11</scalar>
  </property>
  <property dictRef="me:sigma">
    <scalar>4.6</scalar>
  </property>
  <property title="Symmetry Number" dictRef="me:symmetryNumber">
    <scalar> 3 </scalar>
  </property>
</propertyList>
<me:energyTransferModel xsi:type="me:ExponentialDown">

```

```
<me:deltaEDown units="cm-1">200.0</me:deltaEDown>
</me:energyTransferModel>
<me:DOSCMETHOD xsi:type="QMRotors"/>
</molecule>
</moleculeList>
<reactionList>

<reaction id="R_1">
  <reactant>
    <molecule ref="NH2CH3" role="excessReactant" />
  </reactant>
  <reactant>
    <molecule ref="Criegee" role="deficientReactant" />
  </reactant>
  <product>
    <molecule ref="complex" role="modelled" />
  </product>
  <me:MCRCMethod xsi:type="MesmerILT">
    <me:preExponential>1.00e-10</me:preExponential>
    <me:activationEnergy units="cm-1" >0</me:activationEnergy>
    <me:nInfinity>0.0</me:nInfinity>
  </me:MCRCMethod>
  <me:excessReactantConc>1.0E15</me:excessReactantConc>
</reaction>
<reaction id="R_2">
  <reactant>
    <molecule ref="complex" role="modelled" />
  </reactant>
  <product>
    <molecule ref="Prod" role="sink" />
  </product>
  <me:transitionState>
```

```

<molecule ref="ts" role="transitionState" />
</me:transitionState>
<me:MCRCMethod name="SimpleRRKM"/>
</reaction>

</reactionList>

<me:conditions>
<me:bathGas>N2</me:bathGas>
<me:PTs>
    <!-- <me:PTpair units="Torr" P="10" T="263"><me:experimentalEigenvalue EigenvalueID="1"
error="660">6640</me:experimentalEigenvalue></me:PTpair>
        <me:PTpair units="Torr" P="10" T="267.4"><me:experimentalEigenvalue EigenvalueID="1"
error="600">6610</me:experimentalEigenvalue></me:PTpair>

        <me:PTpair units="Torr" P="10" T="275.8"><me:experimentalEigenvalue EigenvalueID="1"
error="1010">6350</me:experimentalEigenvalue></me:PTpair>
        <me:PTpair units="Torr" P="10" T="288.5"><me:experimentalEigenvalue EigenvalueID="1"
error="340">5520</me:experimentalEigenvalue></me:PTpair>
        <me:PTpair units="Torr" P="10" T="297.5"><me:experimentalEigenvalue EigenvalueID="1"
error="700">4410</me:experimentalEigenvalue></me:PTpair>-->
        <me:PTpair units="Torr" P="10" T="311.2"><me:experimentalEigenvalue EigenvalueID="1"
error="640">3600</me:experimentalEigenvalue></me:PTpair>
        <me:PTpair units="Torr" P="50" T="258.2"><me:experimentalEigenvalue EigenvalueID="1"
error="800">7400</me:experimentalEigenvalue></me:PTpair>
        <!--<me:PTpair units="Torr" P="50" T="266.8"><me:experimentalEigenvalue
EigenvalueID="1" error="610">5960</me:experimentalEigenvalue></me:PTpair>
        <me:PTpair units="Torr" P="50" T="275.6"><me:experimentalEigenvalue EigenvalueID="1"
error="1710">5130</me:experimentalEigenvalue></me:PTpair>
        <me:PTpair units="Torr" P="50" T="284.4"><me:experimentalEigenvalue EigenvalueID="1"
error="1280">5340</me:experimentalEigenvalue></me:PTpair>
        <me:PTpair units="Torr" P="50" T="293.6"><me:experimentalEigenvalue EigenvalueID="1"
error="290">4740</me:experimentalEigenvalue></me:PTpair>
        <me:PTpair units="Torr" P="50" T="302.2"><me:experimentalEigenvalue EigenvalueID="1"
error="300">4750</me:experimentalEigenvalue></me:PTpair>

```

```

<me:PTpair units="Torr" P="50" T="311.3"><me:experimentalEigenvalue EigenvalueID="1"
error="1320">2800</me:experimentalEigenvalue></me:PTpair>

<me:PTpair units="Torr" P="10" T="293"><me:experimentalEigenvalue EigenvalueID="1"
error="520">5730</me:experimentalEigenvalue></me:PTpair>

<me:PTpair units="Torr" P="20" T="293"><me:experimentalEigenvalue EigenvalueID="1"
error="280">5690</me:experimentalEigenvalue></me:PTpair>

<me:PTpair units="Torr" P="30" T="293"><me:experimentalEigenvalue EigenvalueID="1"
error="690">5850</me:experimentalEigenvalue></me:PTpair>

<me:PTpair units="Torr" P="60" T="293"><me:experimentalEigenvalue EigenvalueID="1"
error="470">5380</me:experimentalEigenvalue></me:PTpair>

<me:PTpair units="Torr" P="10" T="293"><me:experimentalEigenvalue EigenvalueID="1"
error="330">5380</me:experimentalEigenvalue></me:PTpair>-->

</me:PTs>

</me:conditions>

<me:modelParameters>

<me:grainSize units="cm-1">25</me:grainSize>
<me:energyAboveTheTopHill>10.</me:energyAboveTheTopHill>

</me:modelParameters>

<me:control>

<!-- <me:calcMethod xsi:type="me:marquardt">
<me:MarquardtIterations>10</me:MarquardtIterations>
<me:MarquardtTolerance>0.1</me:MarquardtTolerance>
<me:MarquardtDerivDelta>1.e-02</me:MarquardtDerivDelta>
</me:calcMethod>-->
<me:printSpeciesProfile/>
<me:testRateConstants/>
<me:printGrainedSpeciesProfile/>
<me:eigenvalues>3</me:eigenvalues>
</me:control>

</me:mesmer>
```