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## **Supplementary Information for**

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## Modeling of type IV and V sigmoidal adsorption isotherms

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<u>Figure A1:</u> Conversion of the exponent *m* of the Klotz equation (20) to the exponent *s* of the Sips isotherm (eqn. 12) from non-linear regression in the range of  $\theta$  = 0.2.... 0.8.



Figure A2: Conversion of the parameter *m* of the Klotz equation (20) to the adsorption constant  $K_{FG}$  and the exponent *n* of the Fowler-Guggenheim equation (4) from non-linear regression in the range of  $\theta$  = 0.2.... 0.8.



Figure A3: Conversion of the parameters *m* of the Klotz equation (20) with K = 5 to the energy term *E* / RT and the exponent *n* of the Dubinin-Asthakov equation (13) from non-linear regression in the range of  $\theta = 0.2....0.8$ .

 $m/n = 2.84 (1 - \exp(-0.60 m^{0.44})); RT/E = 0.618 (1 - \exp(-1.27 m^{0.30}))$ 



Figure A4: Conversion of the parameters *K* of the Klotz equation (20) with m = 10 to the energy term *E* / RT and the exponent *n* of the Dubinin-Asthakov equation (13) from non-linear regression in the range of  $\theta = 0.2...0.8$ .

 $n = 11.4 (1 - \exp(-0.14 \text{ K}^{0.73})); E / RT = 4.4 (1 - \exp(-0.16 K^{0.68}))$ 



<u>Figure A5</u>: Modeling of the adsorption of water on carbon no. 1 with the Klotz isotherm (eqn. (24)), green line: modeled with eqn. (31)



<u>Figure A6:</u> Modeling of the adsorption of water on carbon no. 4 with the Klotz isotherm (eqn. (24)), green line: modeled with eqn. (31).



Figure A7: Modeling of the adsorption of water on carbon no. 6 with the Klotz isotherm (eqn. (24)).



<u>Figure A8</u>: Modeling of the adsorption desorption hysteresis of water on carbon no. 7 with the Klotz isotherm (eqn. (24))



<u>Figure A9</u>: Modeling of the adsorption desorption hysteresis of water on carbon no. 8 with the Klotz isotherm (eqn. (24)), green line: desorption modeled with eqn. (29)



Figure A10: Modeling of the adsorption of water on carbon no. 9 with the Klotz isotherm (eqn. (24)).



<u>Figure A11:</u> Modeling of the adsorption of water on carbon no. 11 with the Klotz isotherm (eqn. (24)), green line: modeled with eqn. (31).



<u>Figure A12:</u> Modeling of the adsorption of water on carbon no. 12 with the Klotz isotherm (eqn. (24)), green line: modeled with eqn. (31).



Figure A13: Modeling of the adsorption of water on carbon no. 13 with the Klotz isotherm (eqn. (24)).



<u>Figure A14</u>: Modeling of the adsorption desorption hysteresis of water on carbon no. 16 with the Klotz isotherm (eqn. (24))



Figure A15: Modeling of the adsorption of water on carbon no. 17 with the Klotz isotherm (eqn. (24)).



Figure A16: Modeling of the adsorption of water on carbon no. 18 with the Klotz isotherm (eqn. (24)).



Figure A17: Modeling of the adsorption of water on carbon no. 20 with the Klotz isotherm (eqn. (24)).



Figure A18: Modeling of the adsorption of water on carbon no. 21 with the Klotz isotherm (eqn. (24)).



<u>Figure A19</u>: Modeling of the adsorption of water on carbon no. 24 with the Klotz isotherm (eqn. (24)), green line: modeled with eqn. (31)



<u>Figure A20:</u> Modeling of the adsorption of water on carbon no. 25 with the Klotz isotherm (eqn. (24)), green line: modeled with eqn. (31).; gray line modeling with DD (eqn. (30)).



Figure A21: Correlation between K and  $Q_{max}$  (all values of Table 1).



<u>Figure A22</u>: Modeling of the adsorption of water on an ferro aluminophosphate molecular sieve (No. 28 in Table 1) at 25°C. Blue: extended Klotz isotherm (eqn. (31)). Red: additive superposition of the original Klotz isotherm (eqn. (24)) with the part of the Sips-isotherm used in the fitting of the same underlying data by the "universal isotherm" [50].



<u>Figure A23</u>: Modeling of the adsorption of water on ferro aluminophosphate molecular sieve (No. 29 in Table 1) at 60°C by the extended Klotz isotherm (eqn. (31))



<u>Figure A24</u>: Modeling of the adsorption of water on aluminophosphate molecular sieve (No. 30 in Table 1) at 45°C by the Klotz isotherm (eqn. (24)).



<u>Figure A25</u>: Unsuccessful fitting of the adsorption of water on aluminophosphate molecular sieve AIPO<sub>4</sub>-5 (AFI-structure) [6] by the Klotz isotherm (eqn.(24)).