

Elucidating the reactivity of methoxyphenol positional isomers towards hydrogen-transfer reactions by ATR-IR spectroscopy of the liquid-solid interface of RANEY® Ni

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Supporting information

- 1 – H-transfer at different initial substrate concentration
- 2 – Arrhenius equation
- 3 – ATR-IR spectroscopy bands assignment by DFT calculation
- 4 – Eyring–Polanyi equation

1 – H-transfer at different initial substrate concentration

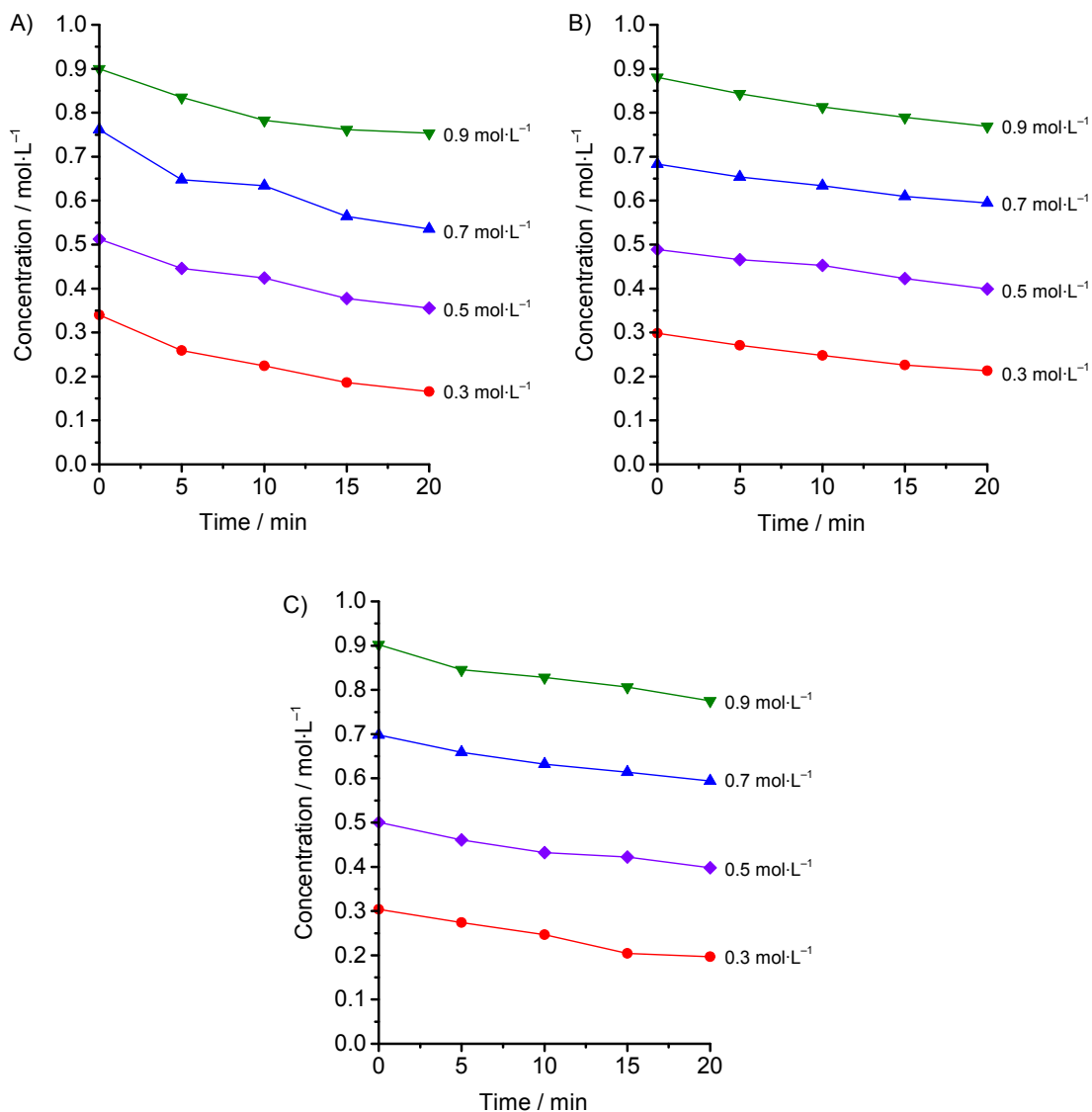


Figure S1. Evolution of the reactant concentration at 80 °C as a function of time using different initial substrate concentrations for A) 2-methoxyphenol, B) 3-methoxyphenol and C) 4-methoxyphenol in the presence of RANEY® Ni/2-PrOH.

2 – Arrhenius equation

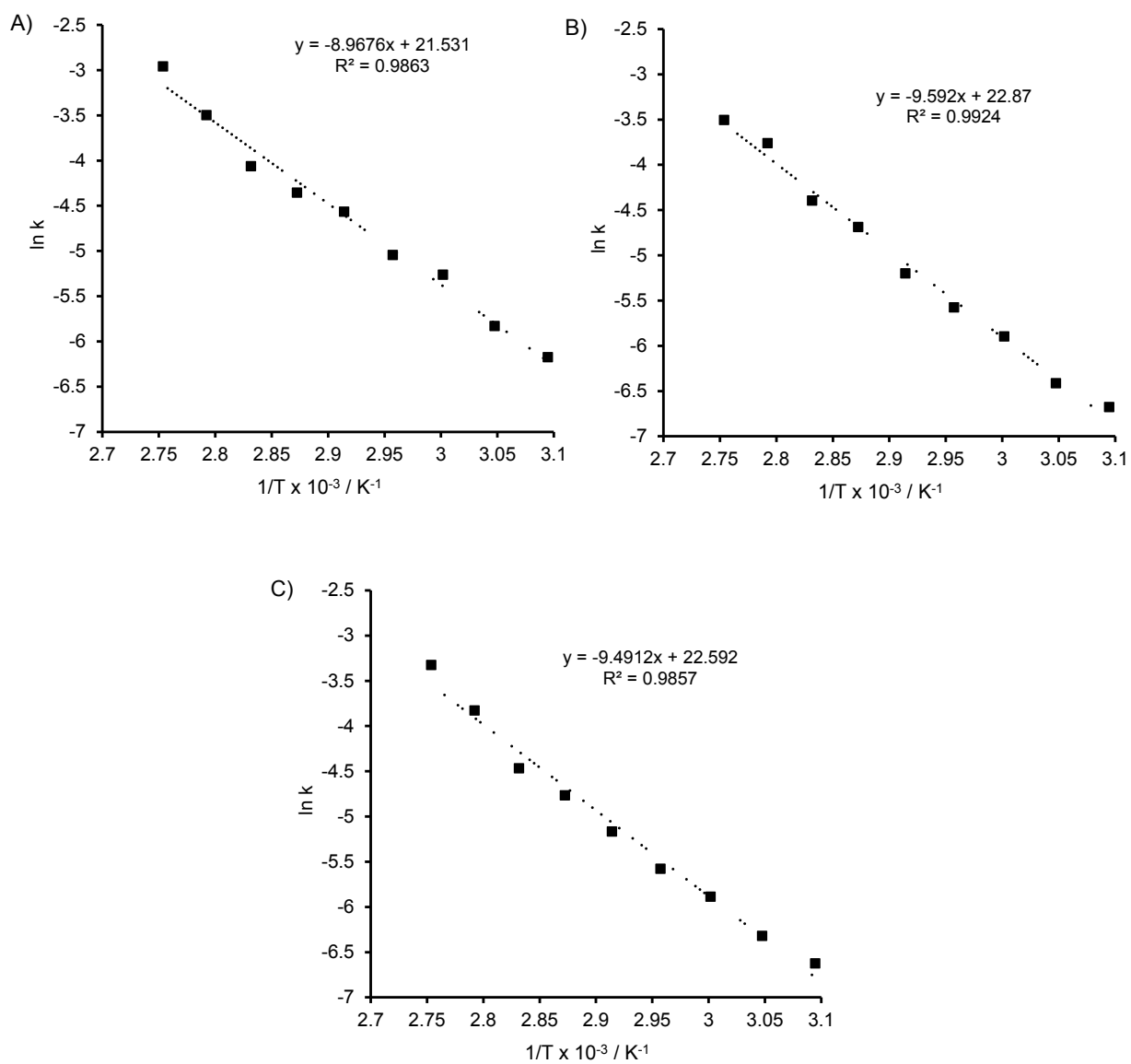


Figure S2. Arrhenius plots for A) 2-methoxyphenol, B) 3-methoxyphenol and C) 4-methoxyphenol in the presence of RANEY[®] Ni/2-PrOH.

3 - ATR-IR spectroscopy bands assignment by DFT calculation

Table S1. ATR-IR spectroscopy bands assignment by DFT calculation for phenol.

Phenol			
exp	B3LYP [def2-SV(P)]		assignment ^c
ν	ω^a	A^b	
3617	3447	50	$\nu(\text{OH})$
1607	1607	54	$\nu(\text{C-C})_{\text{arom}}$
1597	1590	32	$\nu(\text{C-C})_{\text{arom}}$
1499	1490	51	$\nu(\text{C-C})_{\text{arom}}, \delta(\text{C-H})$
1474	1464	38	$\nu(\text{C-C})_{\text{arom}}, \delta(\text{C-H})$
1340	1345	58	$\delta(\text{O-H}), \delta(\text{C-H})_{\text{arom}}$
1256	1266	87	$\nu(\text{C-O}), \delta(\text{C-H})_{\text{arom}}$
1174	1189	126	$\nu(\text{C-C})_{\text{arom}}, \delta(\text{O-H}), \delta(\text{C-H})_{\text{arom}}$

^a The scaling factor for frequencies was 0.972.

^b Signals below an intensity of 20 were not considered.

^c Assignment in agreement with reference ⁴².

Table S2. ATR-IR spectroscopy bands assignment by DFT calculation for 2-methoxyphenol.

2-methoxyphenol			
B3LYP			
exp	[def2-SV(P)]		assignment
ν	ω^a	A ^b	
3562	3374	43	$\nu(\text{OH})$
<i>overlap with solvent</i>	2975	20	$\nu(\text{C-H})_{\text{aliphatic}}$
	2901	37	$\nu(\text{C-H})_{\text{aliphatic}}$
	2838	55	$\nu(\text{C-H})_{\text{aliphatic}}$
	1600	1594	38
1502	1502	182	$\nu(\text{C-O})_{\text{arom}}, \nu(\text{C-C})_{\text{arom}}, \delta(\text{C-H})_{\text{aliphatic}}, \delta(\text{C-H})_{\text{arom}}$
1472	1466	27	$\nu(\text{C-C})_{\text{arom}}, \delta(\text{C-H})_{\text{aliphatic}}, \delta(\text{C-H})_{\text{arom}}, \delta(\text{O-H})$
1456	1460	38	$\delta(\text{C-H})_{\text{aliphatic}}, \delta(\text{C-H})_{\text{arom}}$
	1382	77	$\delta(\text{O-H}), \nu(\text{C-C})_{\text{arom}}, \delta(\text{C-H})_{\text{arom}}$
	1304	22	$\delta(\text{O-H}), \nu(\text{C-C})_{\text{arom}}, \delta(\text{C-H})_{\text{arom}}$
1260	1271	197	$\nu(\text{C-OH})_{\text{arom}}, \delta(\text{C-H})_{\text{arom}}, \nu(\text{C-C})_{\text{arom}}$
1223	1238	169	$\nu(\text{C-OMe})_{\text{arom}}, \nu(\text{C-C})_{\text{arom}}, \delta(\text{C-H})_{\text{arom}}$
1205	1211	54	$\delta(\text{O-H}), \delta(\text{C-H})_{\text{aliphatic}}, \delta(\text{C-H})_{\text{arom}}$
1175	1181	24	$\delta(\text{O-H}), \delta(\text{C-H})_{\text{aliphatic}}, \delta(\text{C-H})_{\text{arom}}$
1111	1103	38	$\delta(\text{C-H})_{\text{arom}}$
1043	1044	43	$\nu(\text{C-O})_{\text{aliphatic}}, \delta(\text{C-H})_{\text{arom}}$
1027	1019	21	$\delta(\text{C-H})_{\text{arom}}$

^a The scaling factor for frequencies was 0.973.^b Signals below an intensity of 20 were not considered.

Table S3. ATR-IR spectroscopy bands assignment by DFT calculation for 3-methoxyphenol.

3-methoxyphenol			
B3LYP			assignment
exp	[def2-SV(P)]		
ν	ω^a	A ^b	
3618	3456	53	$\nu(\text{OH})$
<i>overlap with solvent</i>	2971	24	$\nu(\text{C-H})_{\text{aliph}}^{\text{at}}$
	2895	40	$\nu(\text{C-H})_{\text{aliph}}^{\text{at}}$
	2833	53	$\nu(\text{C-H})_{\text{aliph}}^{\text{at}}$
	1615	1615	150
1596	1589	136	$\nu(\text{C-C})_{\text{arom}}, \nu(\text{C-O})_{\text{arom}}$
1493	1492	97	$\nu(\text{C-C})_{\text{arom}}, \nu(\text{C-O})_{\text{arom}}$
1470	1459	57	$\delta(\text{C-H})_{\text{aliph}}^{\text{at}}$
1437	1442	33	$\delta(\text{C-H})_{\text{aliph}}^{\text{at}}, \delta(\text{C-H})_{\text{arom}}$
1328	1338	87	$\nu(\text{C-C})_{\text{arom}}, \nu(\text{C-O})_{\text{arom}}$
1302	1309	46	$\delta(\text{C-H})_{\text{arom}}, \delta(\text{O-H})_{\text{arom}}$
1288	1294	102	$\nu(\text{C-O})_{\text{arom}}, \delta(\text{O-H}), \delta(\text{C-H})_{\text{arom}}$
	1219	78	$\nu(\text{C-O})_{\text{arom}}, \delta(\text{O-H}), \delta(\text{C-H})_{\text{arom}}$
1198	1198	87	$\delta(\text{O-H}), \delta(\text{C-H})_{\text{aliph}}^{\text{at}}$
1170	1171	125	$\nu(\text{C-O})_{\text{arom}}, \delta(\text{C-H})_{\text{arom}}, \delta(\text{C-O})_{\text{aliph}}^{\text{at}}$
1150	1141	29	$\delta(\text{C-H})_{\text{arom}}$
1045	1048	38	$\nu(\text{C-O})_{\text{aliph}}^{\text{at}}, \delta(\text{C-H})_{\text{arom}}$

^a The scaling factor for frequencies was 0.973.^b Signals below an intensity of 20 were not considered.

Table S4. ATR-IR spectroscopy bands assignment by DFT calculation for 4-methoxyphenol.

4-methoxyphenol			
exp	B3LYP [def2-SV(P)]		assignment
ν	ω^a	A^b	
3622	3459	54	$\nu(\text{OH})$
<i>overlap with solvent</i>	2968	25	$\nu(\text{C-H})_{\text{aliphatic}}$
	2882	47	$\nu(\text{C-H})_{\text{aliphatic}}$
	2825	63	$\nu(\text{C-H})_{\text{aliphatic}}$
1511	1511	229	$\nu(\text{C-C})_{\text{arom}}, \nu(\text{C-O})_{\text{arom}}, \delta(\text{C-H})_{\text{arom}}, \delta(\text{C-H})_{\text{aliphatic}}$
1488	1460	70	$\delta(\text{C-H})_{\text{aliphatic}}$
1465	1435	51	$\nu(\text{C-C})_{\text{arom}}, \delta(\text{O-H}), \delta(\text{C-H})_{\text{arom}}, \delta(\text{C-H})_{\text{aliphatic}}$
1354	1345	88	$\nu(\text{C-C})_{\text{arom}}, \delta(\text{O-H}), \delta(\text{C-H})_{\text{arom}}$
1297	1294	21	$\nu(\text{C-C})_{\text{arom}}, \delta(\text{C-H})_{\text{arom}}$
1236	1249	291	$\nu(\text{C-O})_{\text{arom}}, \delta(\text{C-H})_{\text{arom}}$
1173	1195	107	$\delta(\text{O-H}), \delta(\text{C-H})_{\text{arom}}, \delta(\text{C-H})_{\text{aliphatic}}$
1163	1183	78	$\nu(\text{C-C})_{\text{arom}}, \delta(\text{O-H}), \delta(\text{C-H})_{\text{arom}}, \delta(\text{C-H})_{\text{aliphatic}}$
1046	1053	66	$\nu(\text{C-O})_{\text{aliphatic}}$

^a The scaling factor for frequencies was 0.973.^b Signals below an intensity of 20 were not considered.

4 – Eyring–Polanyi equation

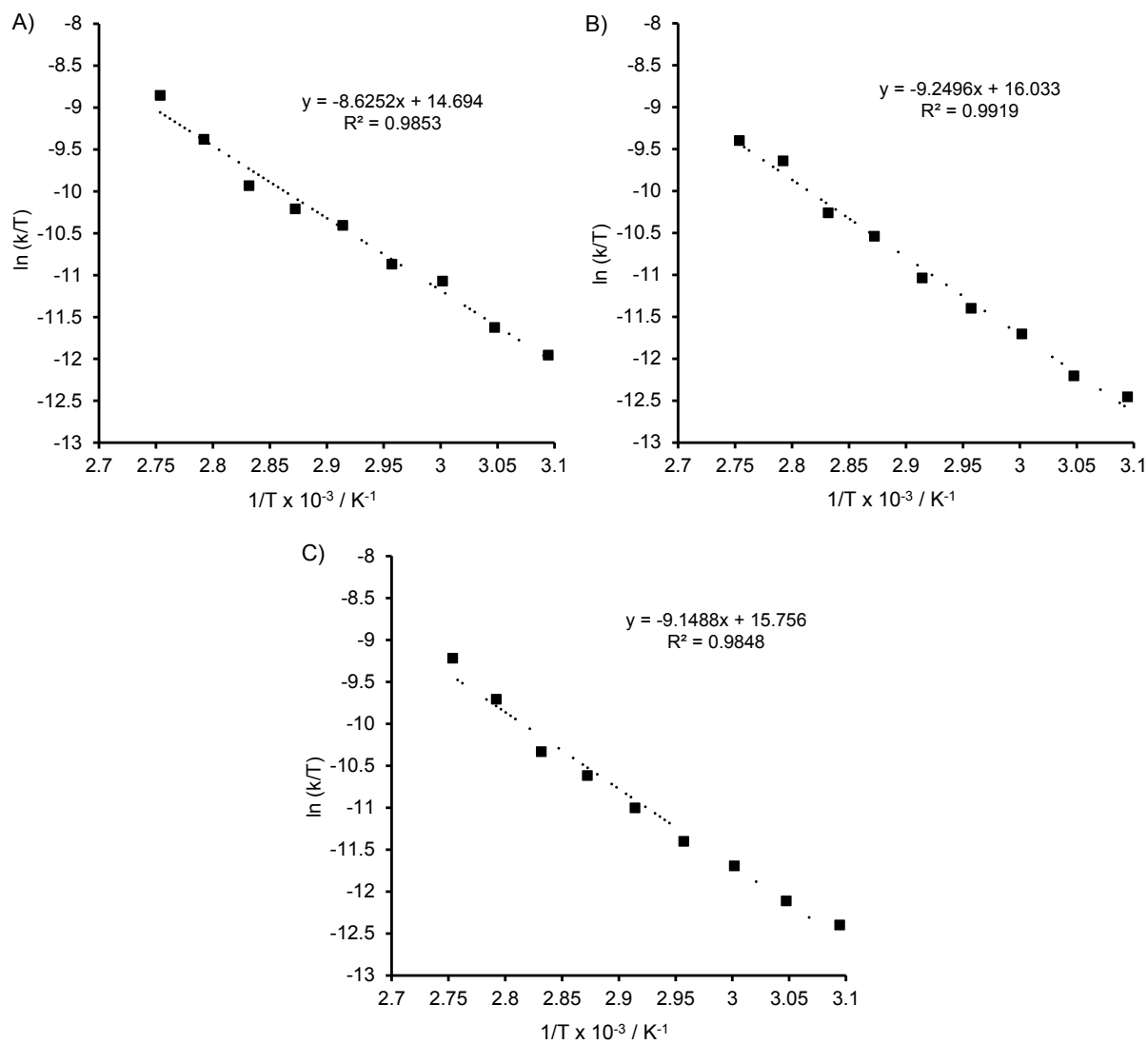


Figure S3. Eyring–Polanyi plots for A) 2-methoxyphenol, B) 3-methoxyphenol and C) 4-methoxyphenol in the presence of RANEY[®] Ni/2-PrOH.