Supporting information

## Insight into the dissolution-crystallization strategy towards macro/meso/microporous Silicalite-1 zeolites and their performance in the Beckmann rearrangement of cyclohexanone oxime

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Fig. S1 SEM images of MSS-500 (A) and MSS-150 (B)



Fig. S2 Low-magnification TEM image (A), HRTEM image (B) and SAED pattern (C) of HM-Silicalite-1(500)



Fig. S3  $N_2$  adsorption-desorption isotherms (A) and pore size distributions (B) of the hierarchical sample HM-Silicalite-1(500) calculated by HK method, NLDFT method and Hg intrusion porosimetry, respectively



Fig. S4 SEM image of the conventional microporous sample conv-Silicalite-1



Fig. S5 TGA and DSC curves of as-prepared hierarchical sample HM-Silicalite-1(500).



Fig. S6 Hydrothermal stability tests: (A) XRD pattern of HM-Silicalite-1 before and after hydrothermal treatment; (B) SEM image of HM-Silicalite-1 after hydrothermal treatment.



Fig. S7 Variation of relative crystallinity (RC) of HM-Silicalite-1(500) with the extension of crystallization time.



Fig. S8 TEM images of sample HM-Silicalite-1(500) crystallized at 48 h.



Fig. S9  $N_2$  adsorption-desorption isotherms (A) and pore size distributions (B) of hierarchical sample HM-Silicalite-1(150) calculated by HK method, NLDFT method and Hg intrusion porosimetry, respectively.



Fig. S10 Low-magnification TEM image of HM-Silicalite-1(150).

Sample	Starting gel composition		Crystallization
	TPAOH/SiO <sub>2</sub>	$H_2O/SiO_2$	temperature/K
HM-Silicalite-1(500)	0.17	0.9	383
HM-S1	0.07	0.9	383
HM-S2	0.25	0.9	383
HM-S3	0.17	0.9	403
HM-S4	0.17	0.9	423
HM-S5	0.17	0.4	383
HM-S6	0.17	1.3	383
HM-S7	0.17	1.8	383
HM-Silicalite-1(150)	0.25	1.8	383

## Table S1 Synthesis conditions of the Silicalite-1 samples