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Electronic Supporting Information

ZrO₂-SBA-15 catalysts for the one-pot cascade synthesis of GVL from furfural

J. Iglesias,*^a J. A. Melero,^a G. Morales,^a M. Paniagua,^a B. Hernández,^a A. Osatiashtiani,^b A. F. Lee^c and Karen Wilson^c

^a School of Experimental Sciences and Technology. Universidad Rey Juan Carlos, C/Tulipan. s/n. E-28933, Móstoles, Madrid, Spain.

^b European Bioenergy Research Institute (EBRI), Aston University, Aston Triangle, Birmingham, B4 7ET, UK. ^c School of Science, RMIT University, Melbourne, VIC 3001, Australia

*jose.iglesias@urjc.es

Characterisation



Figure ESI-1. (A) Low-angle, and (B) wide angle powder XRD patterns of ZrO₂-SBA-15, and t-ZrO₂, zircon, and Zr-SBA-15 reference materials.



Figure ESI-2. (A) N_2 adsorption-desorption isotherms, and (B) corresponding BJH- pore size distributions for ZrO_2 -SBA-15 and parent SBA-15 materials.



Figure ESI-3. TEM images of ZrO₂-SBA-15 and parent SBA-15 materials.



Figure ESI-4. SEM images of (A) ZrO₂-SBA-15(1), (B) ZrO₂-SBA-15(2, and (C) ZrO₂-SBA-15(3) materials.



Figure ESI-5. NH₃ TPD profiles for Zr- and ZrO₂-SBA-15 materials.

Sampla	Acid loading	T_{max}^{a}		
Sample	(mmoles _{eq} H+/g)	(°C)		
Zr-SBA-15	0.205	285		
ZrO ₂ -SBA-15(1)	0.227	269		
ZrO ₂ -SBA-15(2)	0.320	264		
ZrO ₂ -SBA-15(3)	0.357	251		

 Table ESI-1. Results from NH3-TPD analysis performed on Zr- and ZrO₂-SBA-15 materials.

^aMaximum desorption temperature in NH₃ TPD



Figure ESI-6. DRIFT spectra of adsorbed pyridine on ZrO₂-coated SBA-15.

 Table ESI-2.
 Lewis:Brønsted acid ratio for ZrO2-coated SBA-15 from pyridine adsorption and subsequent in vacuo DRIFTS analysis.

Sample	B (1550 cm ⁻¹)	L (1445 cm ⁻¹)	B+L	B:L	
ZrO ₂ -SBA-15 (1)	0.067	0.160	0.227	0.42	
ZrO ₂ -SBA-15 (2)	0.023	0.297	0.320	0.08	
ZrO ₂ -SBA-15 (3)	0.004	0.371	0.375	0.01	



Figure ESI-7. ATR-IR spectra of adsorbed pyridine on ZrO₂-coated SBA-15 in i-PrOH.

Table ESI-3. Lewis:Brønsted acid ratio for ZrO₂-coated SBA-15 from pyridine adsorption and ATR-IR analysis in i-PrOH.

Sample	B (1550 cm ⁻¹)	L (1445 cm ⁻¹)	B+L	B:L	
ZrO ₂ -SBA-15 (1)	0.0077	0.0256	0.0333	0.3	
ZrO ₂ -SBA-15 (2)	0	0.0730	0.0730	0	
ZrO ₂ -SBA-15 (3)	0	0.0674	0.0674	0	

Table ESI-4. Basicity loadings calculated for ZrO₂-coated SBA-15 from CO₂-TPD profiles.

Sampla	Basicity loading			
Sample	(µmol∙g⁻¹)			
ZrO ₂ -SBA-15 (1)	2.4			
ZrO ₂ -SBA-15 (2)	4.9			
ZrO ₂ -SBA-15 (3)	11.1			



Scheme ESI-1 Proposed reaction network for the transformation of furfural in alcohol media over ZrO₂-grafted SBA-15. FAL, furfural; FOL, furfuryl alcohol; FE, isopropyl furfuryl ether; i-LEV, isopropyl levulinate; GVL, γ-valerolactone; HUMINS, degradation product

Catalyst		ZrO ₂ -SBA-15 (1)			ZrO ₂ -SBA-15 (2)			ZrO ₂ -SBA-15 (3)		
T/°C		130	150	170	130	150	170	130	150	170
	1	0.881	1.472	1.986	0.913	1.523	2.126	0.949	1.588	2.510
k _i / h⁻¹	2	0.603	1.857	4.622	0.197	0.791	4.548	0.257	1.099	1.193
	3	0.051	0.086	0.219	0.047	0.055	0.192	0.044	0.052	0.064
	4	0.068	0.150	0.268	0.145	0.325	0.421	0.158	0.381	0.793
	5	0.126	0.287	0.703	0.010	0.042	0.412	0.055	0.195	0.125
	6	0.003	0.015	0.150	0.004	0.025	0.107	0.005	0.011	0.025
	7	0.002	0.007	0.019	0.003	0.024	0.095	0.003	0.009	0.021

Table ESI-5 First-order pseudo-homogeneous apparent kinetic constants (k_i) of the modelled overallreaction of furfural in 2-propanol over ZrO_2 -SBA-15 catalysts, at 130, 150 and 170 °C. See SchemeESI-1 for the reaction steps.



Figure ESI-8. Experimental (symbol) and modelled (solid lines) products concentration for furfural transformation to GVL in 2-propanol over ZrO₂-SBA-15 catalysts. Reaction conditions: furfural:2-propanol=1:50 (by mols); furfural:catalyst=2.5:1 (by weight).



Fig ESI-9. TG analysis curves recorded for sample ZrO2-SBA-15 used during 3 consecutive reutilization cycles in the cascade transformation of furfural to GVL. Reaction conditions: 170 °C; 7 h; furfural:2-propanol=1:50 (by mols); furfural:catalyst=2.5:1 (by mass).