

Supporting information for

Insight into the stereoselectivity of TS-1 in epoxidation of *cis/trans*-2-hexene: a computational study

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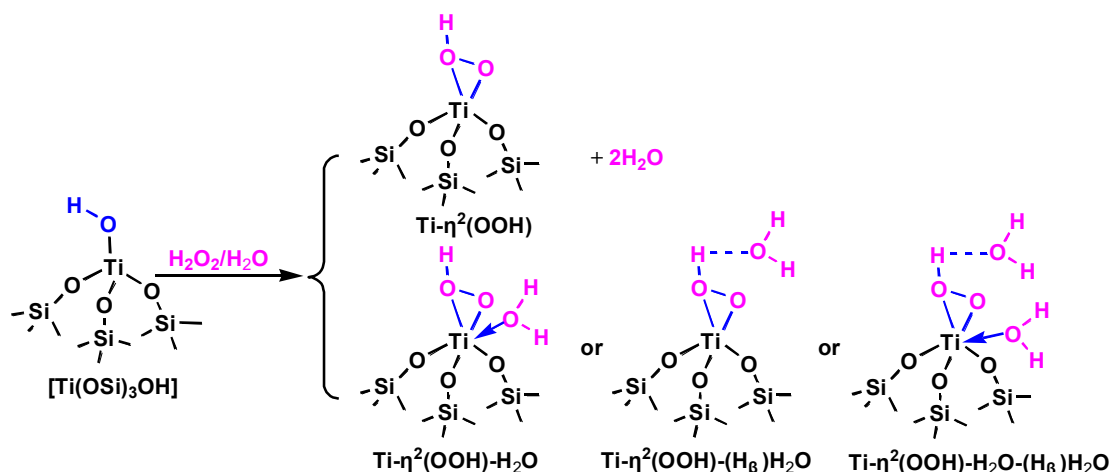
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Fig. S2 Optimized structures of the epoxide products of *cis*-2-hexene (a-d) and *trans*-2-hexene (e-f) adsorbed on different active centers: (a, e) $\text{Ti-}\eta^2(\text{OOH})$, (b, f) $\text{Ti-}\eta^2(\text{OOH})\text{-H}_2\text{O}$, (c, g) $\text{Ti-}\eta^2(\text{OOH})\text{-(H}_\beta\text{)H}_2\text{O}$, and (d, h) $\text{Ti-}\eta^2(\text{OOH})\text{-H}_2\text{O-(H}_\beta\text{)H}_2\text{O}$.



Scheme S1 Generation processes of Ti-hydroperoxo intermediate in TS-1/H₂O₂/H₂O.

The formation energies of the Ti-hydroperoxo intermediates are calculated according to following equations: (1) for Ti-η²(OOH); (2) for Ti-η²(OOH)-H₂O; (3) for Ti-η²(OOH)-(H_β)H₂O; (4) for Ti-η²(OOH)-H₂O-(H_β)H₂O.

$$\Delta E = E[\text{Ti-}\eta^2(\text{OOH})] + E(\text{H}_2\text{O}) - E(\text{Ti-OH}) - E(\text{H}_2\text{O}_2) \quad (1)$$

$$\Delta E = E[\text{Ti-}\eta^2(\text{OOH})\text{-H}_2\text{O}] - E(\text{Ti-OH}) - E(\text{H}_2\text{O}_2) \quad (2)$$

$$\Delta E = E[\text{Ti-}\eta^2(\text{OOH})\text{-(H}_\beta\text{)H}_2\text{O}] - E(\text{Ti-OH}) - E(\text{H}_2\text{O}_2) \quad (3)$$

$$\Delta E = E[\text{Ti-}\eta^2(\text{OOH})\text{-H}_2\text{O-(H}_\beta\text{)H}_2\text{O}] - E(\text{Ti-OH}) - E(\text{H}_2\text{O}_2) - E(\text{H}_2\text{O}) \quad (4)$$

Here, $E(\text{H}_2\text{O})$ and $E(\text{H}_2\text{O}_2)$ are the total electronic energies of optimized H₂O and H₂O₂ molecules, respectively, and $E(\text{Ti-OH})$, $E[\text{Ti-}\eta^2(\text{OOH})]$, $E[\text{Ti-}\eta^2(\text{OOH})\text{-H}_2\text{O}]$, and $E[\text{Ti-}\eta^2(\text{OOH})\text{-(H}_\beta\text{)H}_2\text{O}]$ denote the total electronic energies of optimized models (136T) for Ti-OH, Ti-η²(OOH), Ti-η²(OOH)-H₂O, and Ti-η²(OOH)-(H_β)H₂O, respectively.

Table S1 The total electronic energies of optimized molecules and models at the theoretical level of ωB97XD/6-31G(d, p).

Model	Energy (a.u.)
H ₂ O	-76.39790
H ₂ O ₂	-151.49520
Ti-OH	-57522.97953
Ti-η ² (OOH)	-57598.08494
Ti-η ² (OOH)-H ₂ O	-57674.52100
Ti-η ² (OOH)-(H _β)H ₂ O	-57674.51584
Ti-η ² (OOH)-H ₂ O-(H _β)H ₂ O	-57750.95010

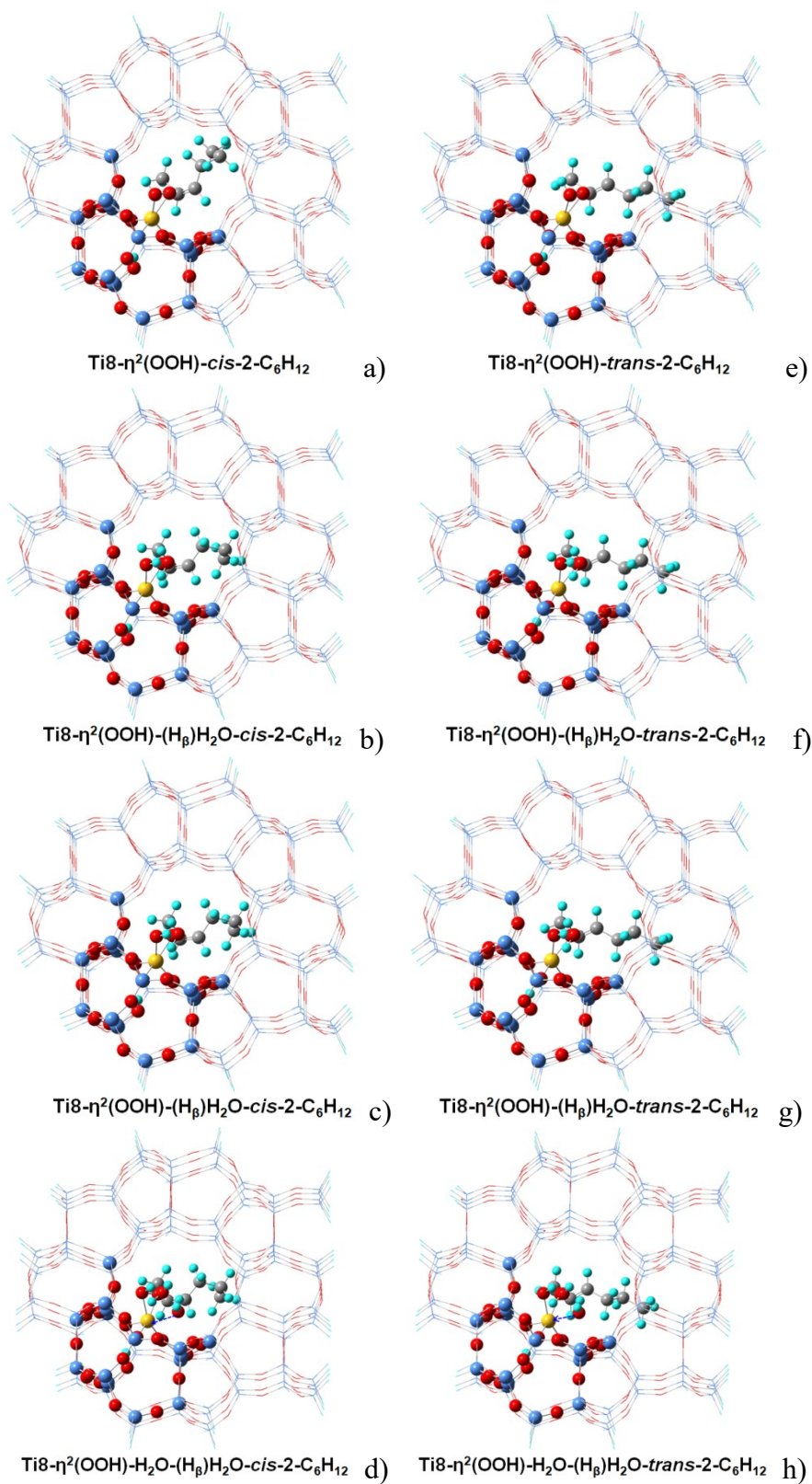


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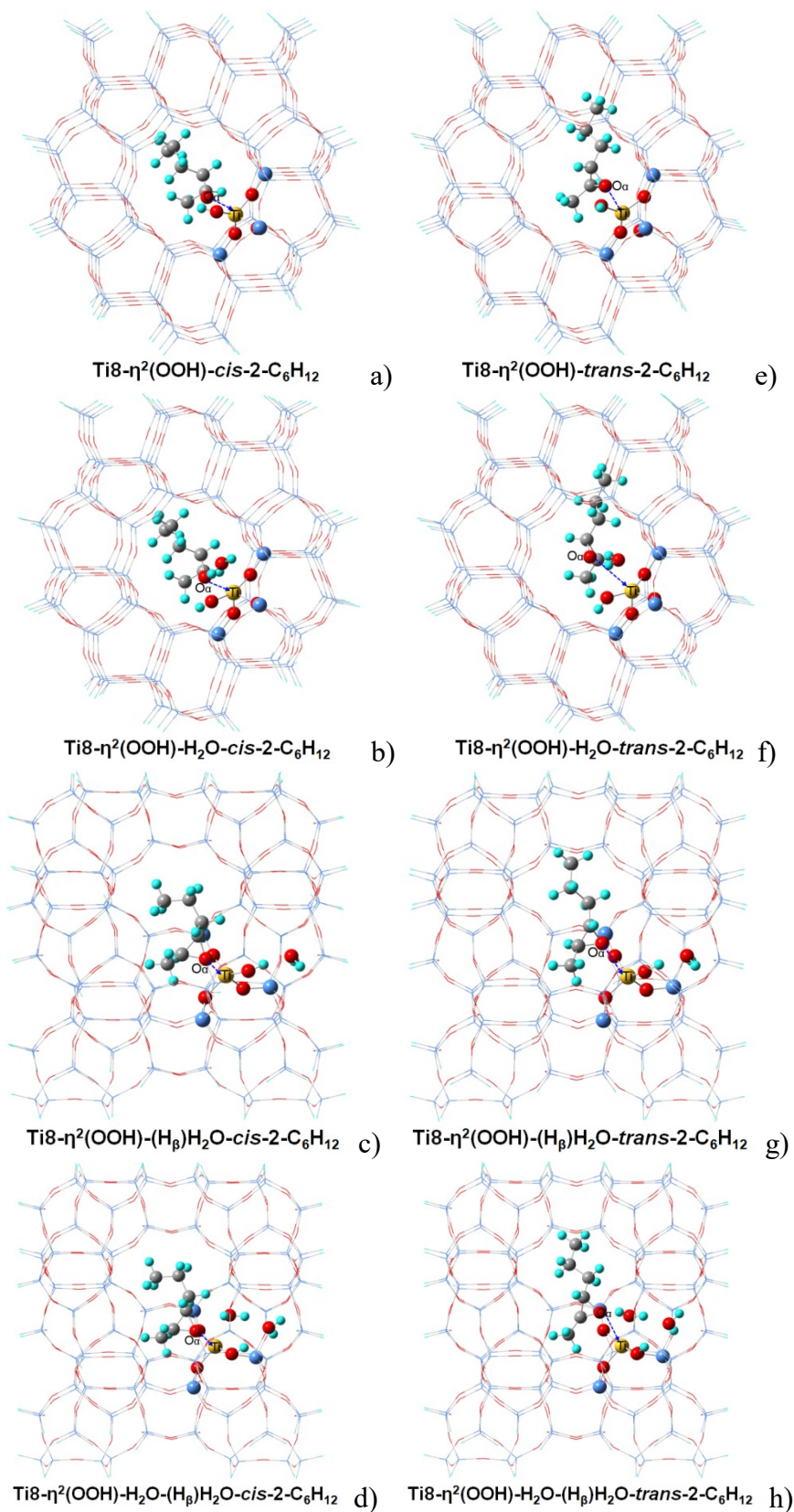


Fig. S2 Optimized structures of the epoxide products of *cis*-2-hexene (a-d) and *trans*-2-hexene (e-f) adsorbed on different active centers: (a, e) $\text{Ti-}\eta^2(\text{OOH})$, (b, f) $\text{Ti-}\eta^2(\text{OOH})\text{-H}_2\text{O}$, (c, g) $\text{Ti-}\eta^2(\text{OOH})\text{-(H}_\beta\text{)H}_2\text{O}$, and (d, h) $\text{Ti-}\eta^2(\text{OOH})\text{-H}_2\text{O-(H}_\beta\text{)H}_2\text{O}$.