

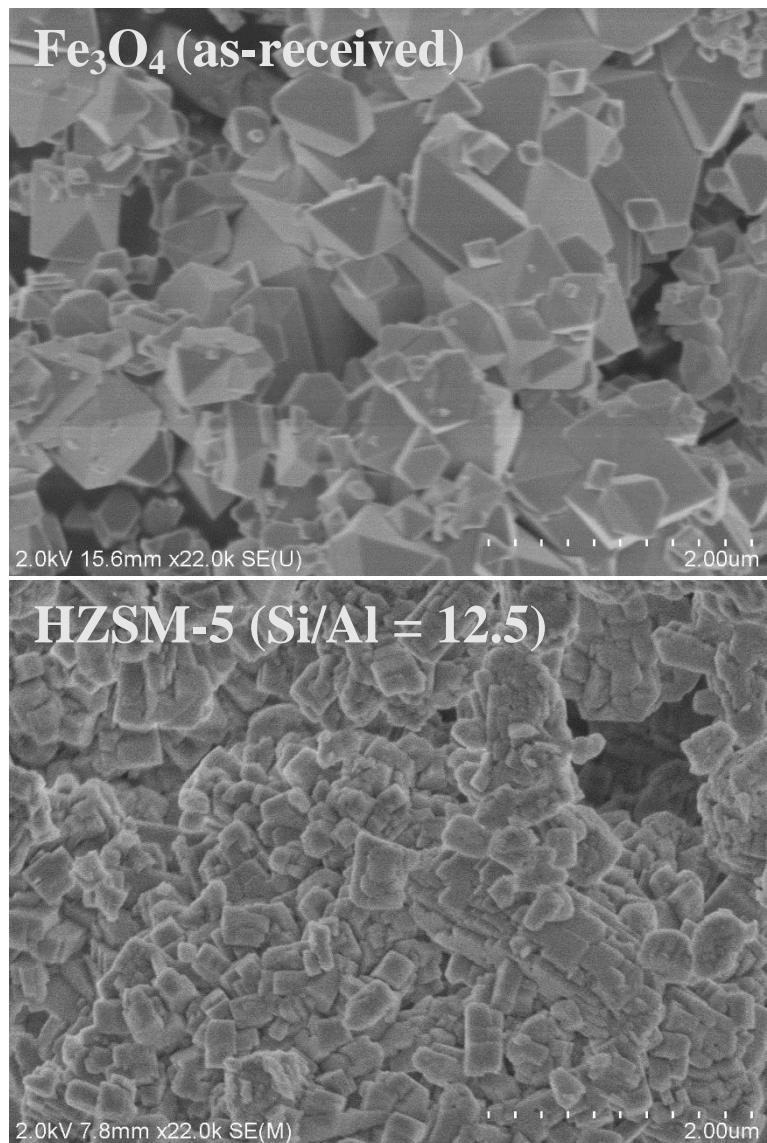
## Supplementary material

### Selective production of aromatics from CO<sub>2</sub>

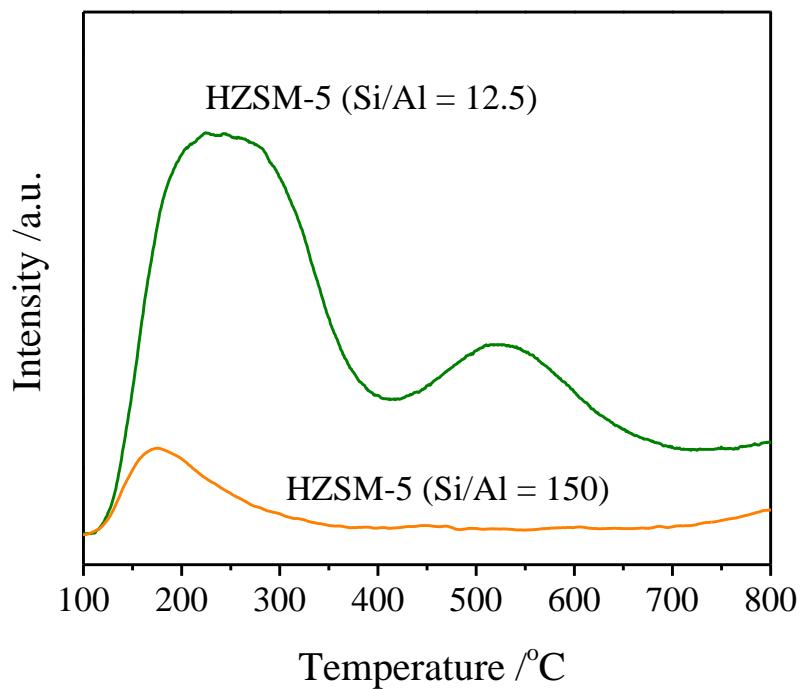
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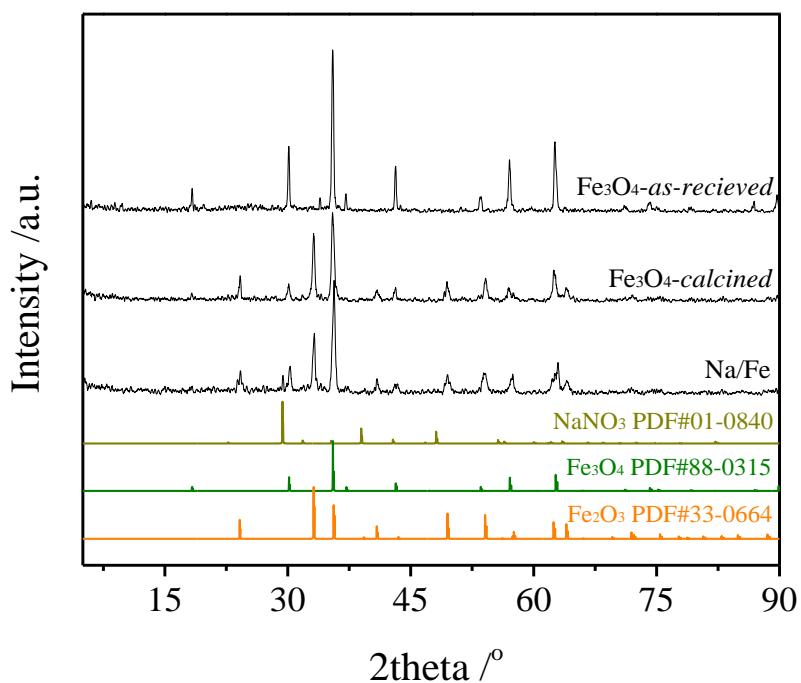
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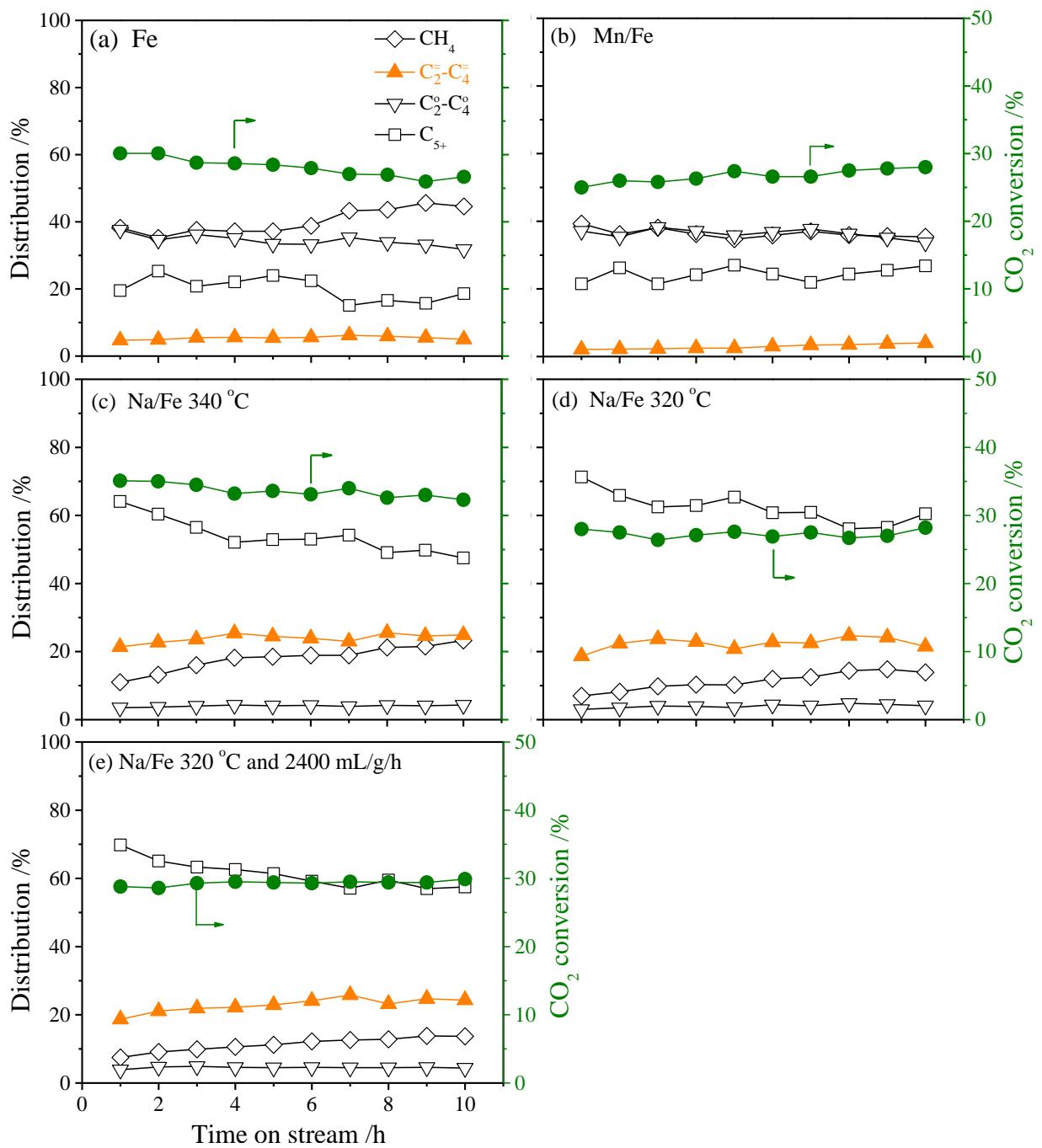
**Fig. S1.** SEM observations of Fe<sub>3</sub>O<sub>4</sub> and HZSM-5 (Si/Al = 12.5).



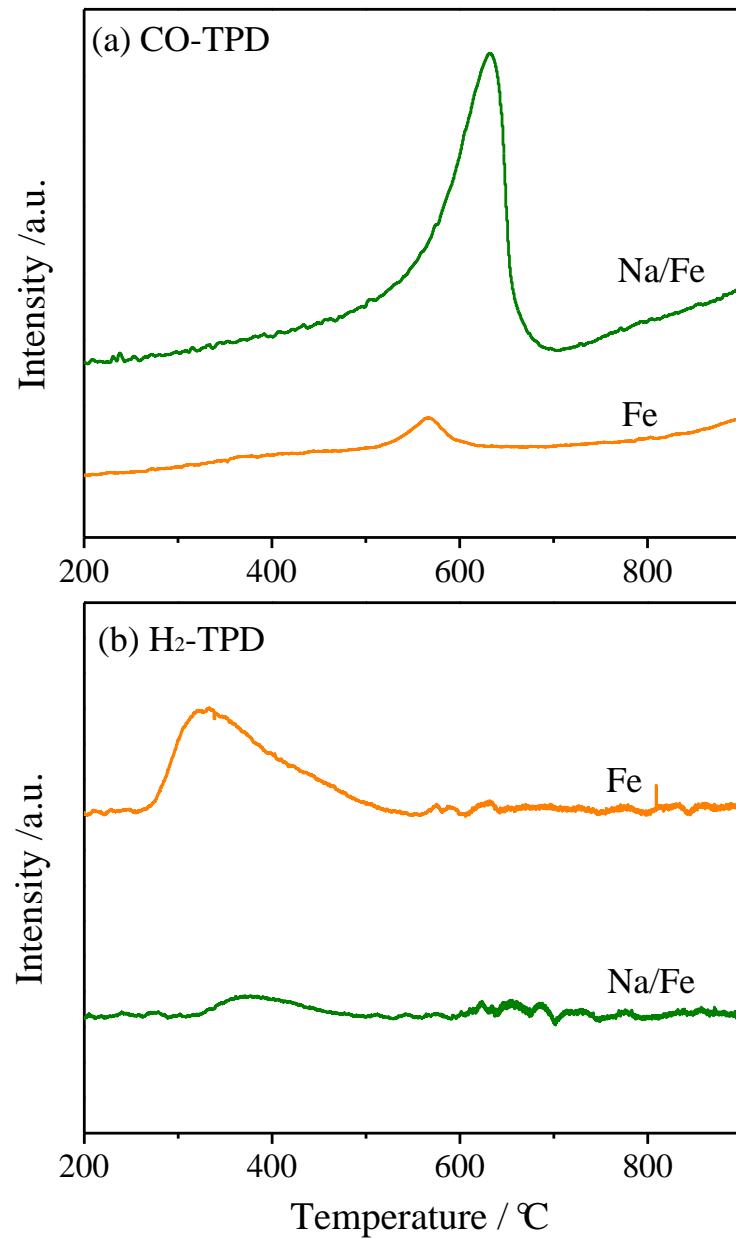
**Fig. S2.**  $\text{NH}_3$ -TPD profiles of fresh HZSM-5 zeolites.



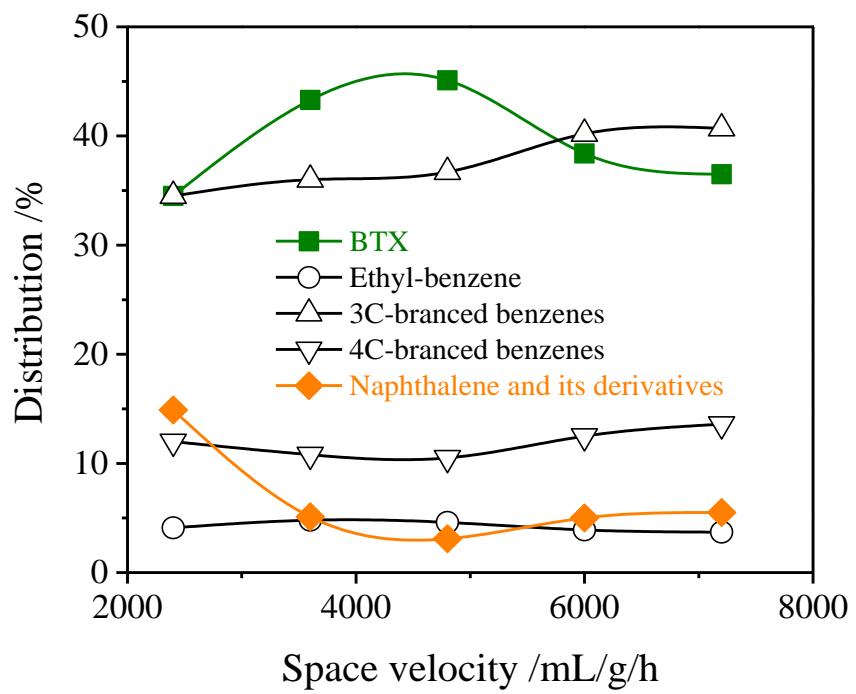
**Fig. S3.** XRD patterns of as-received and calcined  $\text{Fe}_3\text{O}_4$  and  $\text{Na}/\text{Fe}_3\text{O}_4$ .



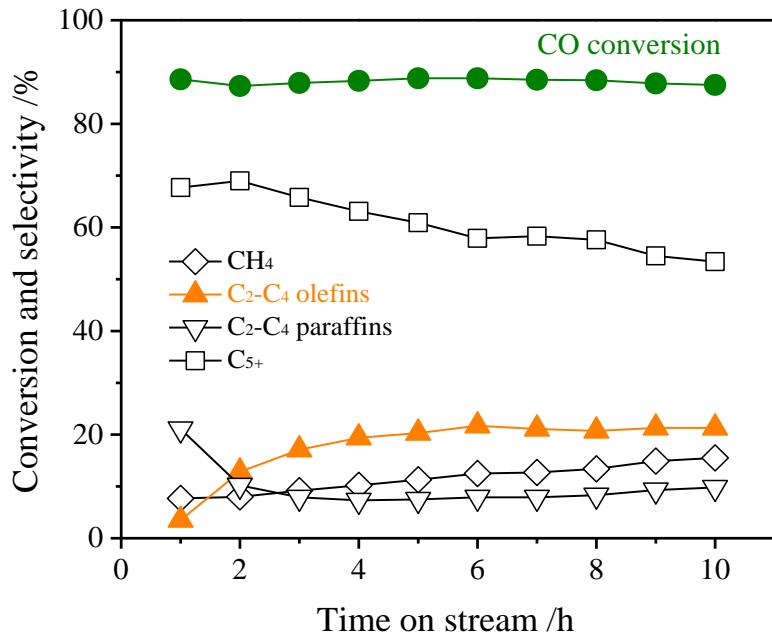
**Fig. S4.** Time-dependences of  $\text{CO}_2$  conversion and hydrocarbon distribution obtained over (a) Fe, (b) Mn/Fe and (c) Na/Fe catalysts at 340 °C, 1 MPa, 4800  $\text{mL g}^{-1} \text{h}^{-1}$  and  $\text{H}_2/\text{CO}_2$  ratio of 3, and (d) Na/Fe catalyst at 320 °C and (e) Na/Fe catalyst at 320 °C and 2400  $\text{mL g}^{-1} \text{h}^{-1}$ .



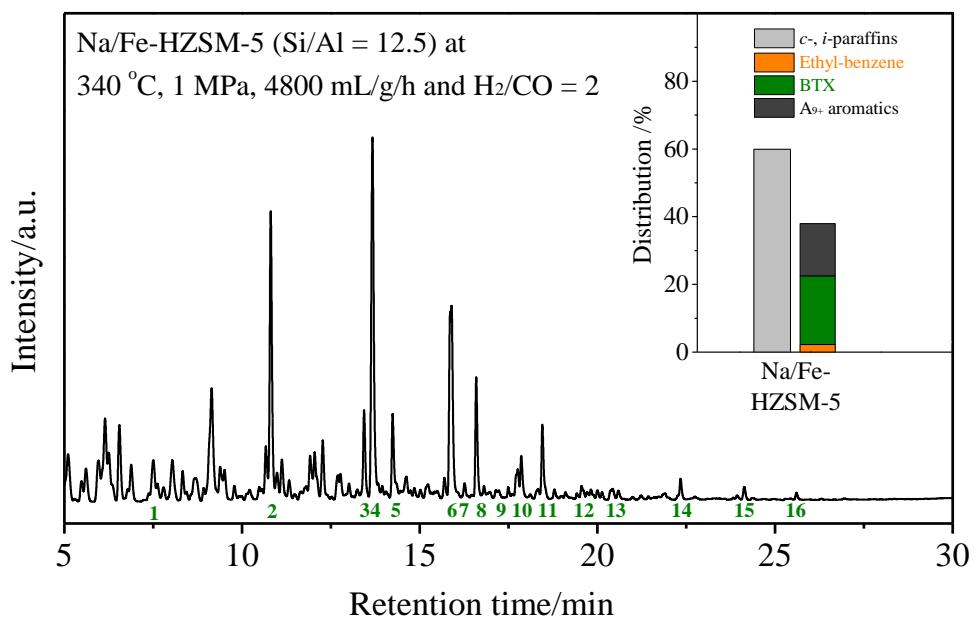
**Fig. S5.** CO-TPD and H<sub>2</sub>-TPD profiles of reduced Fe and Na/Fe catalysts.



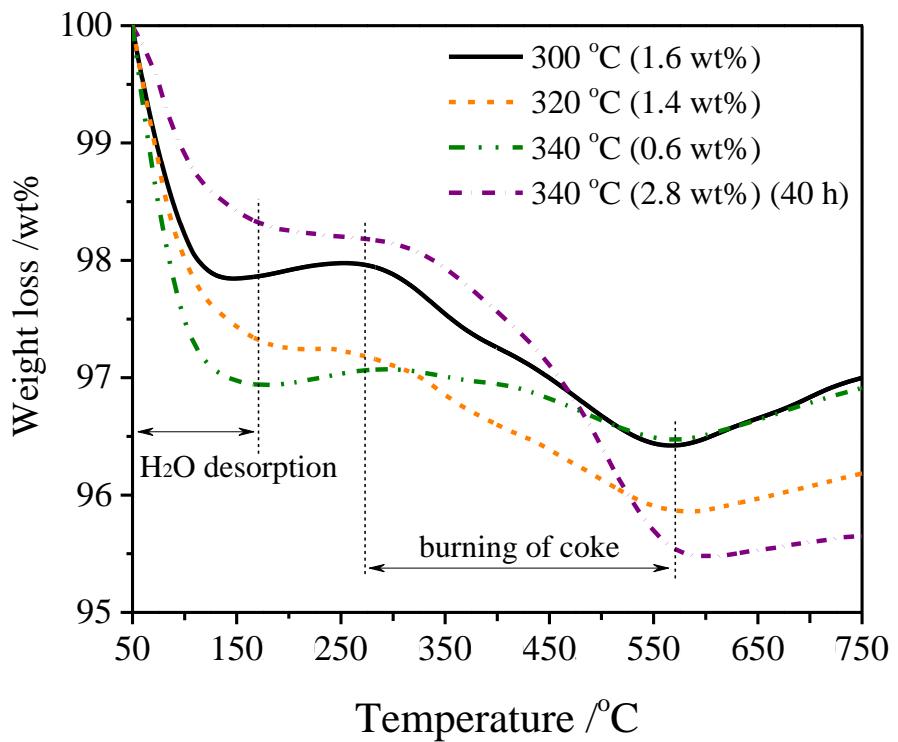
**Fig. S6.** Aromatic distribution obtained at different space velocities over the composite Na/Fe and HZSM-5 (Si/Al = 12.5) catalyst system.



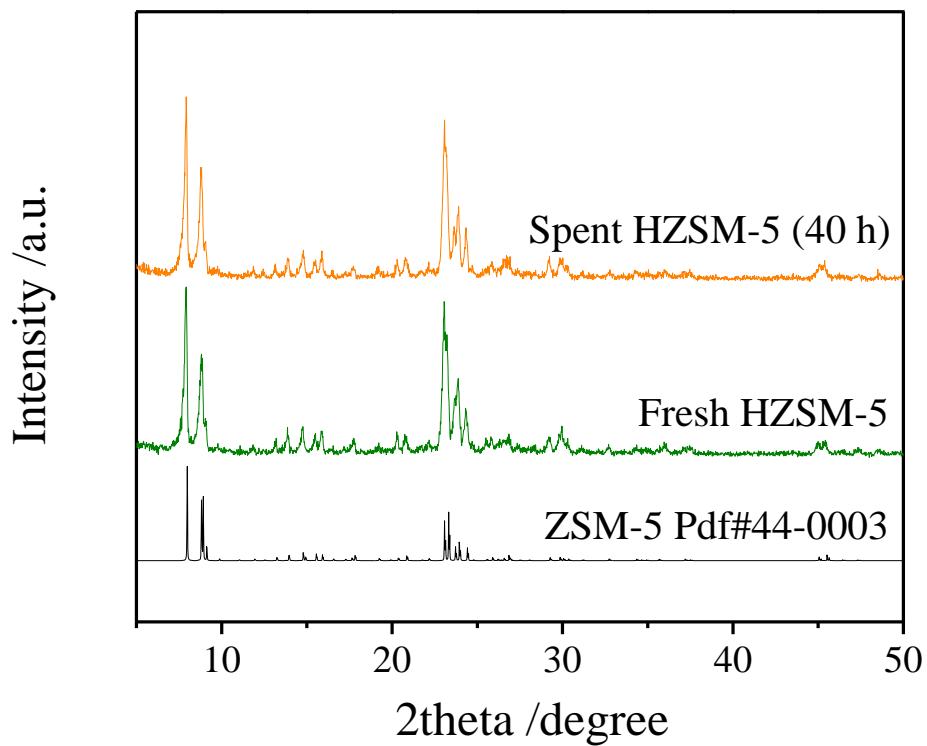
**Fig. S7.** Catalytic hydrogenation of CO (time-dependence of CO conversion and the CO<sub>2</sub>-free selectivities to CH<sub>4</sub>, C<sub>2</sub><sup>=</sup>-C<sub>4</sub><sup>=</sup>, C<sub>2</sub><sup>0</sup>-C<sub>4</sub><sup>0</sup> and C<sub>5+</sub>) over the composite Na/Fe and HZSM-5 (Si/Al = 12.5) catalyst system at the conditions of 340 °C, 1 MPa, 4800 mL g<sup>-1</sup> h<sup>-1</sup> and H<sub>2</sub>/CO = 2.



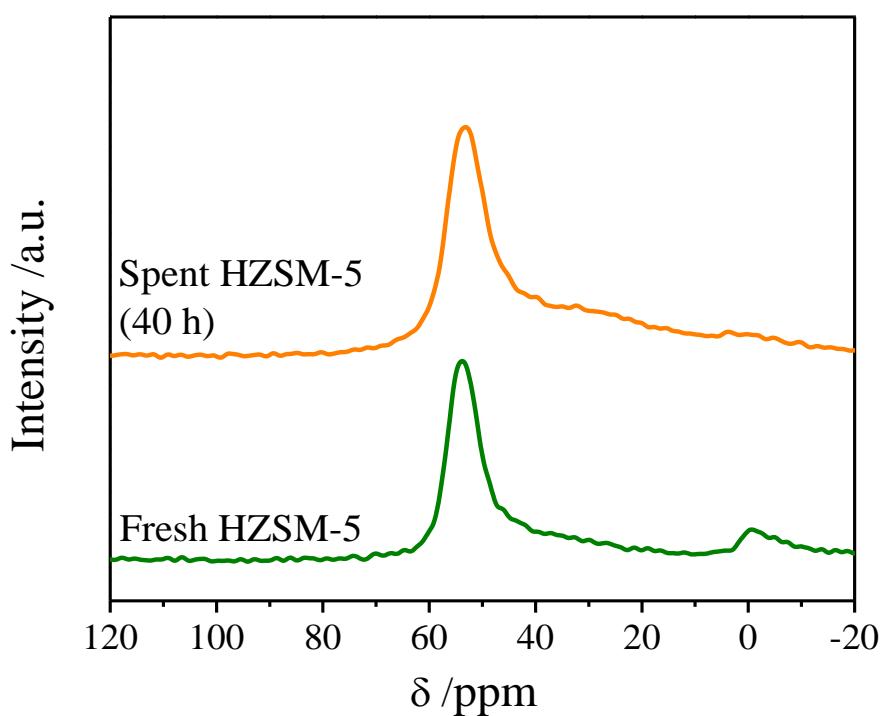
**Fig. S8.** GC spectrum of liquid hydrocarbon obtained from CO hydrogenation over the composite Na/Fe and HZSM-5 catalyst system at 340 °C, 1 MPa, 4800 mL g<sup>-1</sup> h<sup>-1</sup> and H<sub>2</sub>/CO = 2. The numbers of 1-16 present the aromatic hydrocarbons whose names are same with those in Fig. 5 in main text. The inserted figure shows the distribution of hydrocarbons by integration of the GC spectrum.



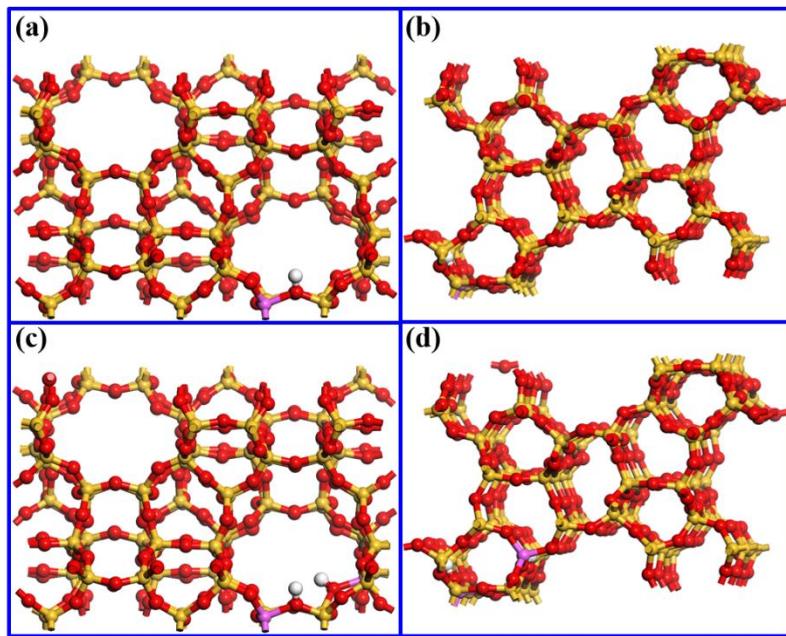
**Fig. S9.** TG profiles of spent HZSM-5 samples collected from tests over the composite Na/Fe and HZSM-5 (Si/Al = 12.5) catalyst system at different reaction temperatures. The first stage of weight loss is attributed to H<sub>2</sub>O desorption and the second one is the burning of coke which is used to calculate the coke content in sample (wt%).



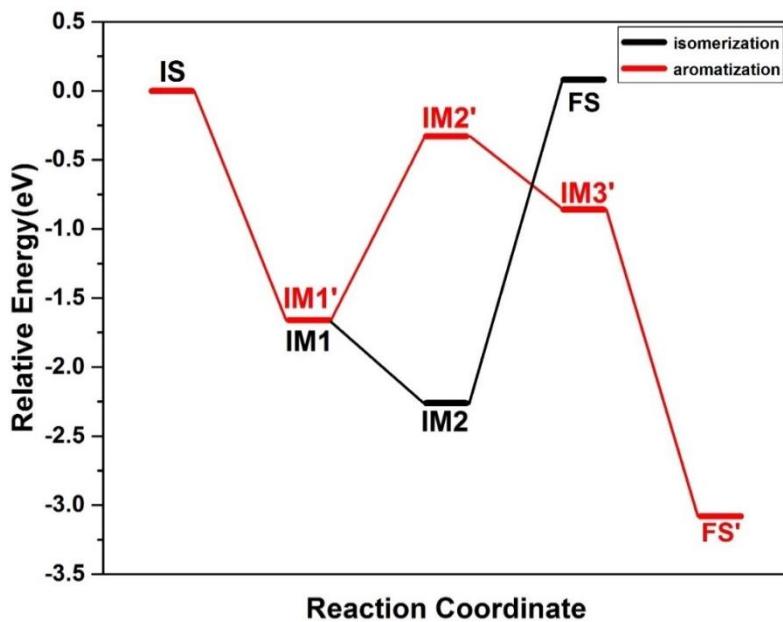
**Fig. S10.** XRD patterns of fresh HZSM-5 and the spent one collected after 40 h reaction (Fig. 8 in main text).



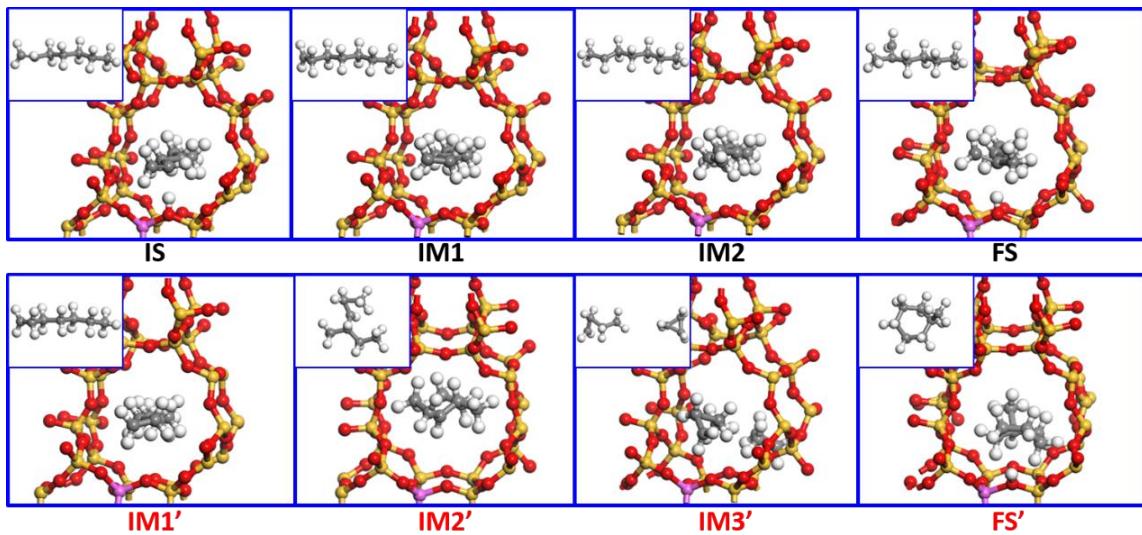
**Fig. S11.**  $^{27}\text{Al}$  NMR spectrum of fresh HZSM-5 zeolite and the spent one collected from the 40-h reaction (Fig. 8 in main text).



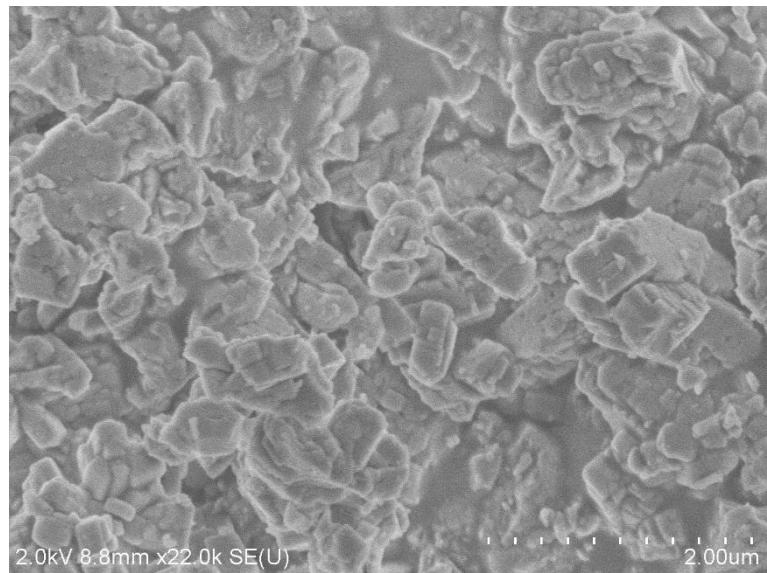
**Fig. S12.** Front and side views of HZSM-5 (a) and (b) and 2HZSM-5 (c) and (d). Red: O atoms; yellow: Si atoms; purple: Al atoms; white: H atoms.



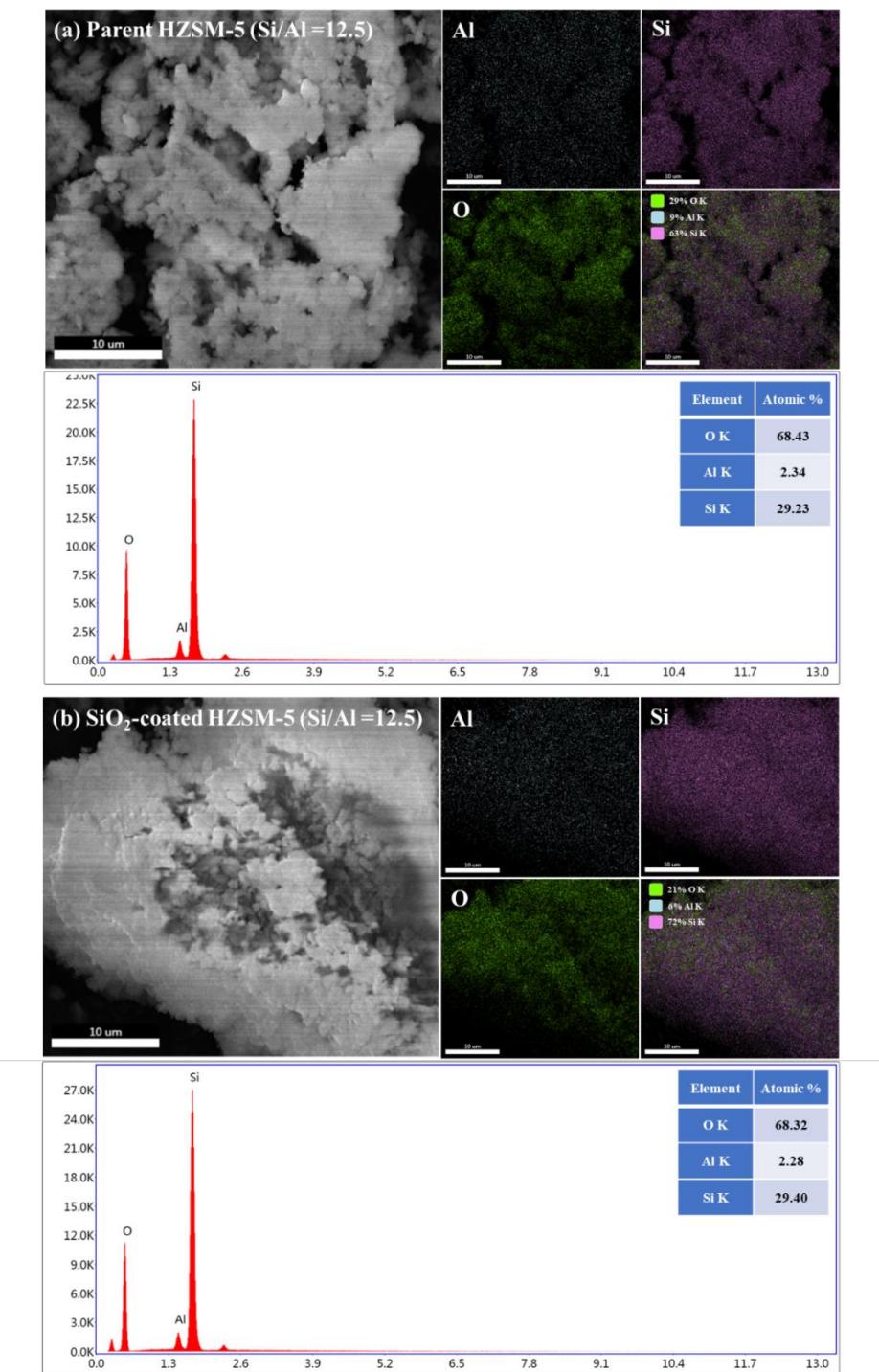
**Fig. S13.** Calculated energy profiles of isomerization and aromatization reactions over HZSM-5.



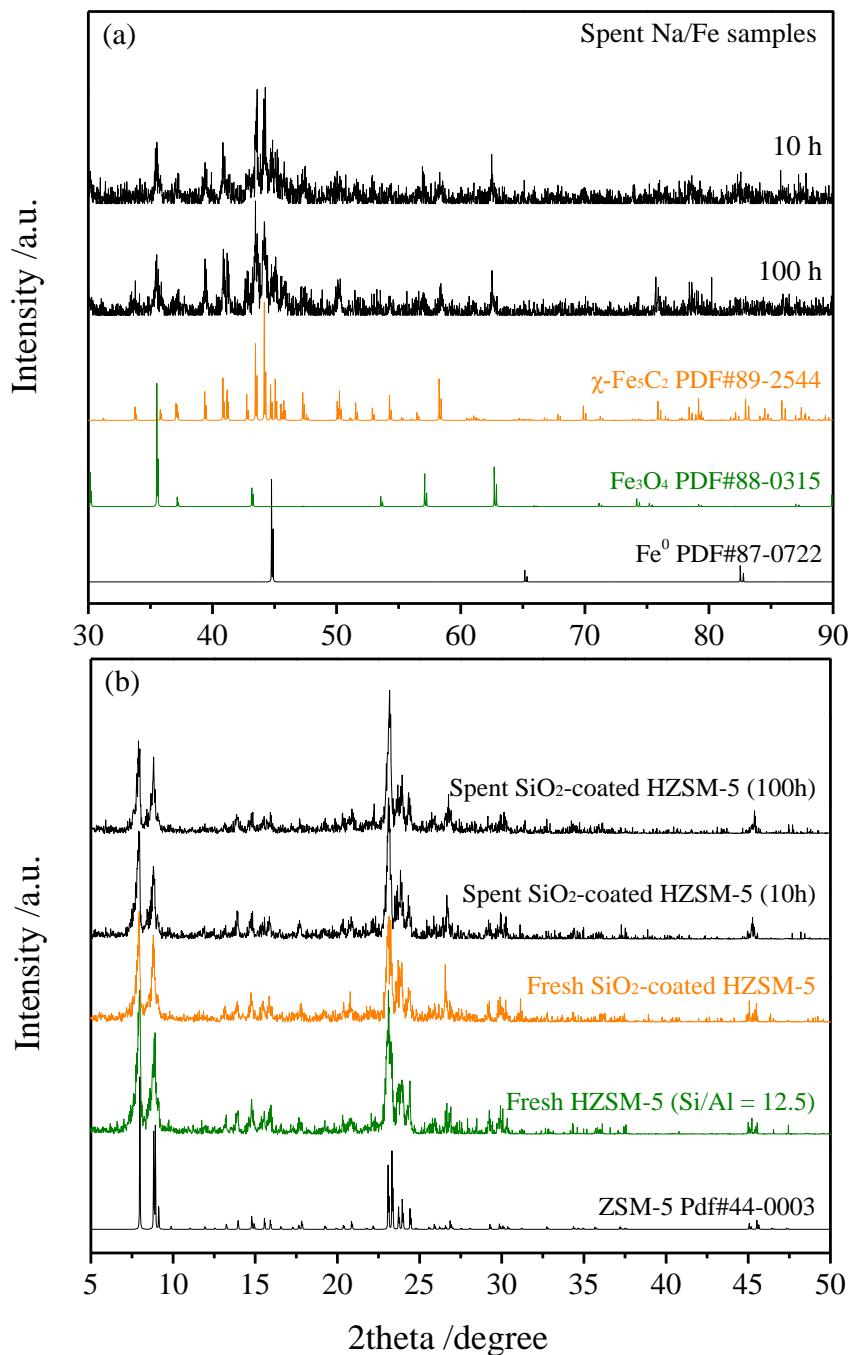
**Fig. S14.** Calculated intermediates in the isomerization (black) and aromatization (red) reactions over HZSM-5. Red: O atoms; yellow: Si atoms; purple: Al atoms; grey: C atoms; white: H atoms.



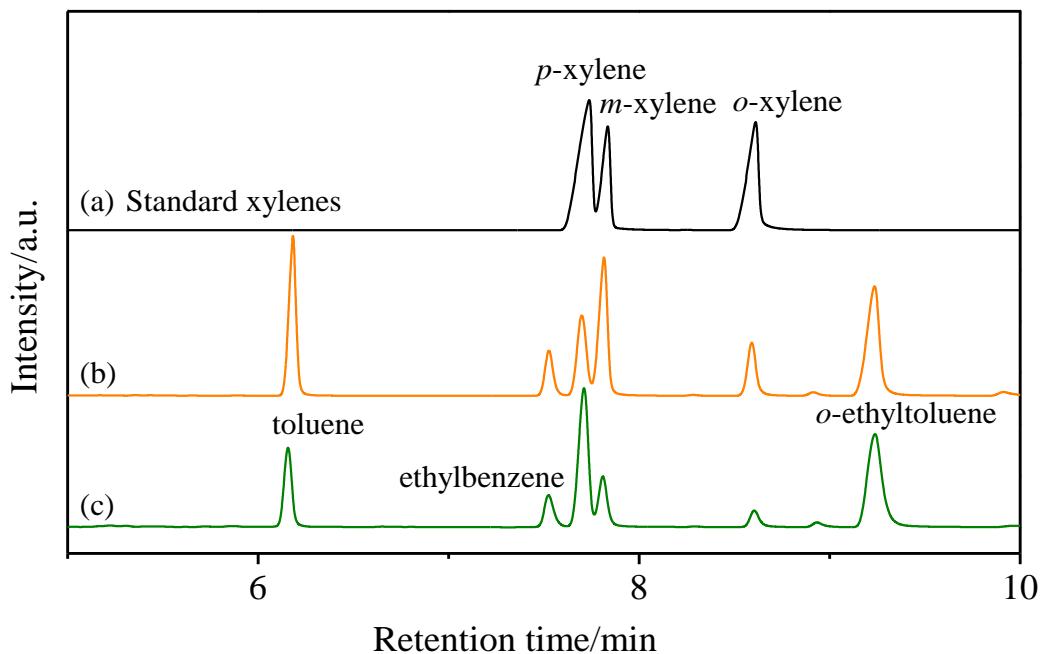
**Fig. S15.** SEM observation of  $\text{SiO}_2$ -coated HZSM-5 ( $\text{Si}/\text{Al} = 12.5$ ) zeolite.



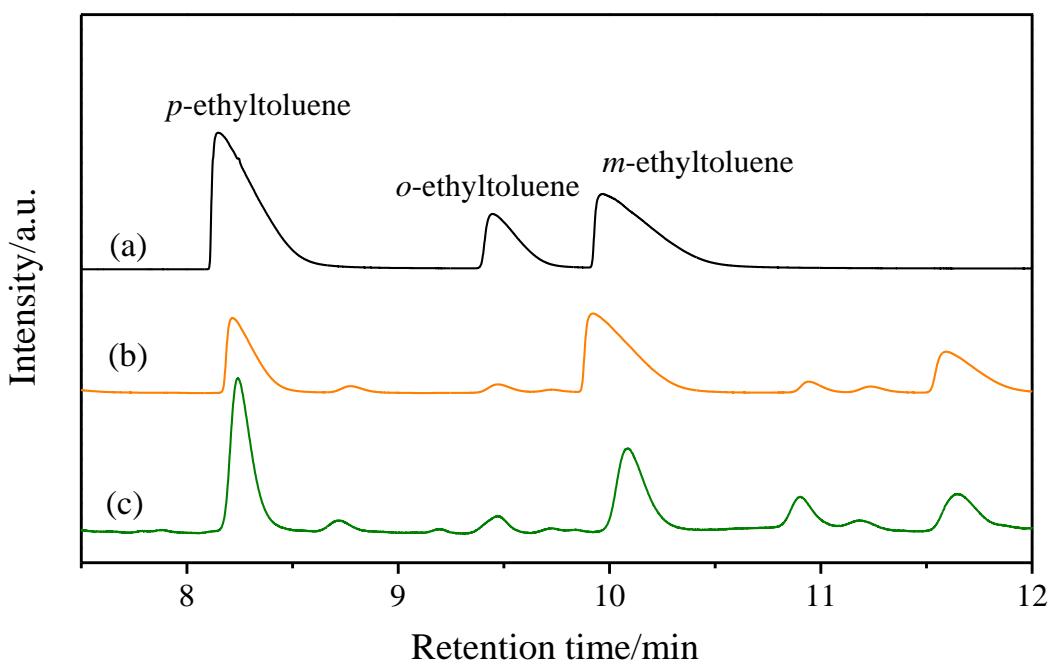
**Fig. S16.** SEM images and EDS analysis of parent HZSM-5 and its  $\text{SiO}_2$ -coated sample.



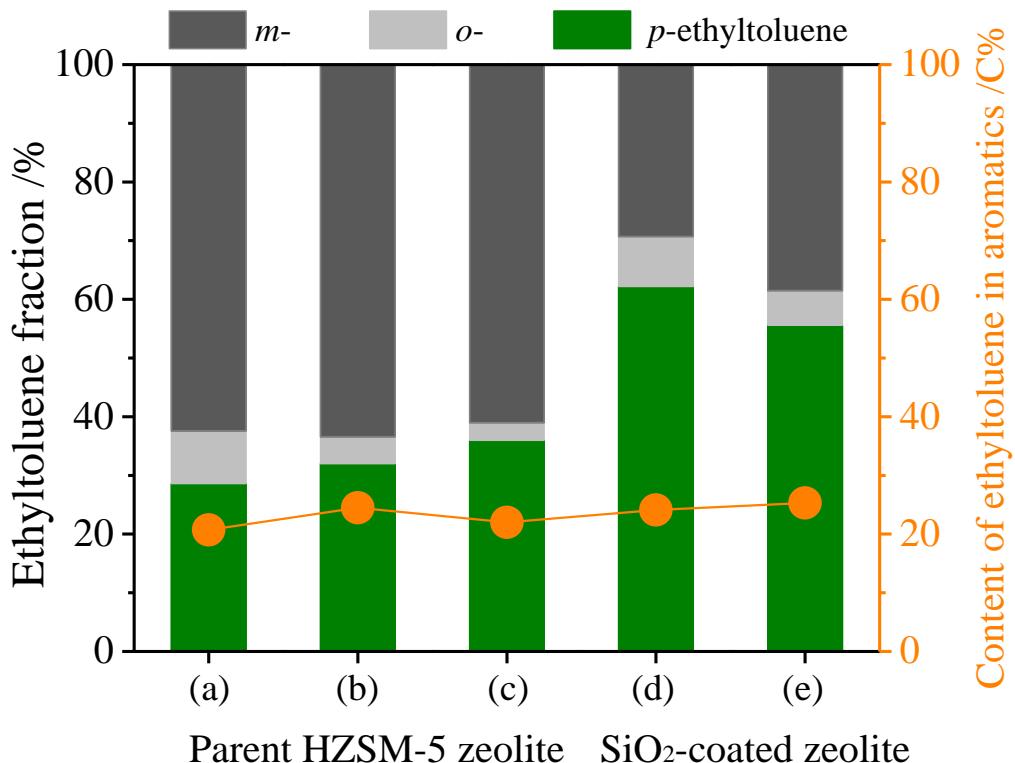
**Fig. S17.** XRD patterns of (a) spent Na/Fe catalyst samples collected from the tests of 10 h and 100 h and (b) fresh HZSM-5 (Si/Al = 12.5), fresh and spent SiO<sub>2</sub>-coated HZSM-5 samples collected from the tests of 10 h and 100 h.



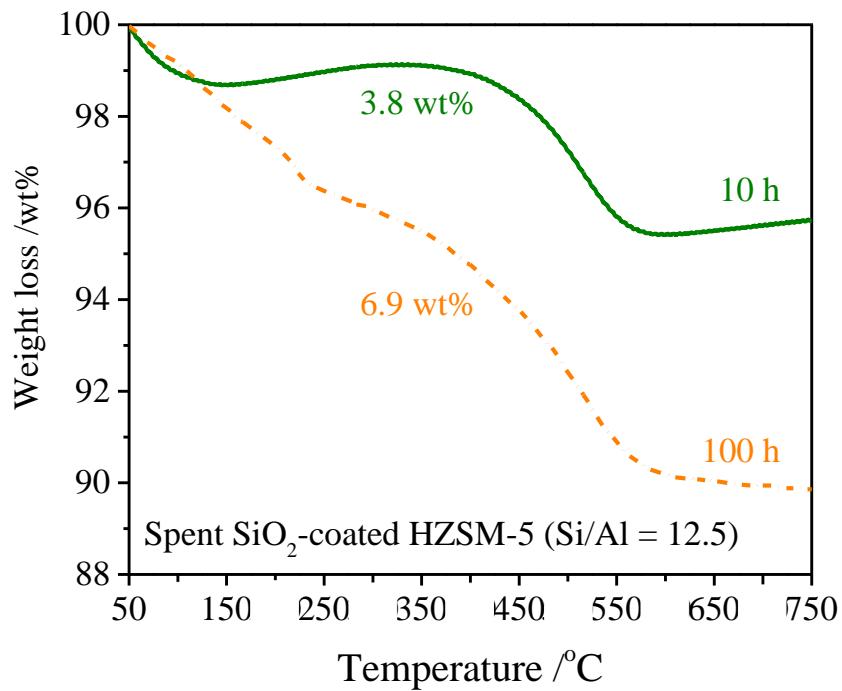
**Fig. S18.** GC spectrum of liquid hydrocarbon product separated by a FFAP capillary column. (a) standard xylenes (black line); (b) sample obtained from the 40-h test over the composite Na/Fe and HZSM-5 catalyst system (orange line); (c) sample obtained from the 10-h test over the Na/Fe and SiO<sub>2</sub>-coated HZSM-5 catalyst system (green line).



**Fig. S19.** GC spectrum of liquid hydrocarbon product separated by a Bentone-34 capillary column under a condition of initial temperature of 50 °C maintaining for 3 min, heating rate of 2 °C min<sup>-1</sup> to final temperature of 90 °C: (a) standard ethyltoluene mixture (black line); (b) sample obtained from the 40-h test over the composite Na/Fe and HZSM-5 catalyst system (orange line); (c) sample obtained from the 10-h test over the composite Na/Fe and SiO<sub>2</sub>-coated HZSM-5 catalyst system (green line).



**Fig. S20.** Distribution of *m*-, *o*- and *p*-ethyltoluene (bar) in total ethyltoluene and the content of the ethyltoluene (●) in total aromatics obtained over the composite Na/Fe and HZSM-5 (Si/Al = 12.5) catalyst system (weight ratio of 1:0.5) at 340 °C, 4800 mL g<sup>-1</sup> h<sup>-1</sup>, H<sub>2</sub>/CO<sub>2</sub> = 3 and (a) 1 MPa; (b) 2 MPa; (c) 1 MPa for 40 h; and over the composite Na/Fe and SiO<sub>2</sub>-coated HZSM-5 (Si/Al = 12.5) catalyst system at the same reaction condition (1 MPa) for (e) 10 h and (f) 100 h.



**Fig. S21.** TG profiles of spent SiO<sub>2</sub>-coated HZSM-5 samples collected from 10 h and 100 h tests over the composite Na/Fe and SiO<sub>2</sub>-coated HZSM-5 (Si/Al = 12.5) catalyst system at 340 °C, 1 MPa, 4800 mL g<sup>-1</sup> h<sup>-1</sup>.

**Table S1.** Hydrogenation of CO or CO<sub>2</sub> over the composite Na/Fe and HZSM-5 catalyst systems.

Feed <sup>a)</sup>	Composite Na/Fe and HZSM-5 system and reaction duration	CO <sub>2</sub> or CO conv. /%	CO or CO <sub>2</sub> sel. /C%	Hydrocarbon distribution /C%					Aromatic content in C <sub>5+</sub> liquid phase /C%	
				CH <sub>4</sub>	C <sub>2</sub> <sup>=</sup> -C <sub>4</sub> <sup>=</sup>	n-C <sub>2</sub> <sup>0</sup> -C <sub>4</sub> <sup>0</sup>	i-C <sub>4</sub> <sup>0</sup>	Liquid phase C <sub>5+</sub> <sup>b)</sup>		
H <sub>2</sub> /CO = 2	Si/Al = 12.5, 10 h	88.2	41.5	13.8	20.8	8.0	0.6	34.2	22.6	39.8
H <sub>2</sub> /CO <sub>2</sub> = 3	Si/Al = 12.5, 40 h	32.3	21.9	28.7	6.9	12.0	7.3	6.6	38.5	85.3
H <sub>2</sub> /CO <sub>2</sub> = 3	Si/Al = 150, 10 h	32.0	23.1	20.9	27.1	5.0	0.3	18.4	28.3	60.5
H <sub>2</sub> /CO <sub>2</sub> = 3	SiO <sub>2</sub> -coated, 10 h	30.9	26.8	22.4	21.9	7.8	2.1	4.1	44.7	91.6
H <sub>2</sub> /CO <sub>2</sub> = 3	SiO <sub>2</sub> -coated, 100 h	30.6	27.1	34.4	25.4	8.0	0.6	9.4	22.2	70.3

<sup>a)</sup> Reaction condition: 340 °C, 1 MPa, 4800 mL g<sup>-1</sup> h<sup>-1</sup>; <sup>b)</sup> Nonaromatic hydrocarbons including *iso*-paraffins mainly and *cyclo*-paraffins.

**Table S2.** Aromatic distribution in total aromatic hydrocarbons (related with Table S1).

Feed <sup>a)</sup>	Composite Na/Fe and HZSM-5 system and reaction duration	Aromatic fraction in total aromatic hydrocarbons /C%								
							3C-branched benzenes	4C-branched benzenes		Branched naphthalene
H <sub>2</sub> /CO = 2	Si/Al = 12.5, 10 h	4.4	17.7	6.1	24.8	6.3	25.7	12.0	0.7	2.3
H <sub>2</sub> /CO <sub>2</sub> = 3	Si/Al = 12.5, 40 h	1.5	17.6	6.1	28.5	8.6	28.2	7.4	0.8	1.3
H <sub>2</sub> /CO <sub>2</sub> = 3	Si/Al = 150, 10 h	1.9	7.9	4.0	21.1	3.9	33.5	24.1	2.1	1.5
H <sub>2</sub> /CO <sub>2</sub> = 3	SiO <sub>2</sub> -coated, 10 h	0.6	13.8	6.3	35.8	3.3	28.2	8.9	1.2	1.9
H <sub>2</sub> /CO <sub>2</sub> = 3	SiO <sub>2</sub> -coated, 100 h	0.9	9.8	6.9	29.5	5.9	30.2	14.3	1.1	1.4

<sup>a)</sup> Reaction condition: 340 °C, 1 MPa, 4800 mL g<sup>-1</sup> h<sup>-1</sup>.