

## Electronic Supplementary Information

### **Hydrogen- and noble metal-free conversion of nitro aromatics to amino aromatics having reducible groups over an organically modified TiO<sub>2</sub> photocatalyst under visible light irradiation**

Yuhei Yamamoto,<sup>a</sup> Makoto Fukui,<sup>a</sup> Atsuhiko Tanaka,<sup>b</sup> Keiji Hashimoto<sup>b</sup> and Hiroshi Kominami<sup>b\*</sup>

<sup>a</sup> *Molecular and Material Engineering, Interdisciplinary Graduate School of Science and Engineering, Kindai University, 3-4-1 Kowakae, Higashiosaka, Osaka 577-8502, Japan.*

<sup>b</sup> *Department of Applied Chemistry, Faculty of Science and Engineering, Kindai University, 3-4-1 Kowakae, Higashiosaka, Osaka 577-8502, Japan.*

*E-mail: hiro@apch.kindai.ac.jp*

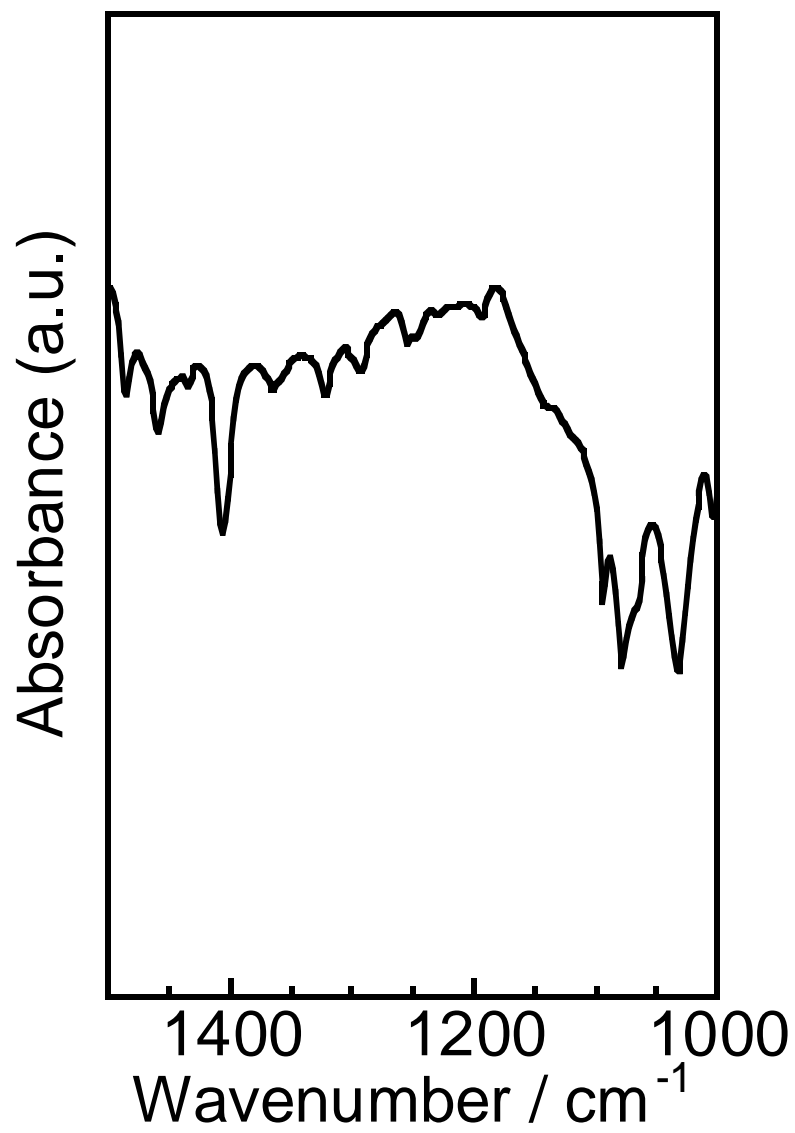


Fig. S1 FT-IR spectrum of TEOA-loaded TiO<sub>2</sub>.

Table S1 summary of various reduction of nitrobenzenes to anilines.

Entry	reference	Catal.	R	Solvent	Initial	Irradiation	Hole scavenger	Temp.	Atm.	Time / h	Conv. / %	Sel. / %
1	10	AgNPs@CeO <sub>2</sub> , 25 mg	vinyl	dodecane	0.5 mmol	×	×	110	H <sub>2</sub> (6 atm)	6	99 <sup>a</sup>	99
2	11	Au/Al-2.5, 0.2 mol%	vinyl	THF	2 mmol	×	×	120	H <sub>2</sub> (3 MPa)	2	100	99
3	11	Au/Al-2.5, 0.2 mol%	Acetyl	THF	2 mmol	×	×	65	H <sub>2</sub> (3 MPa)	4	100	99
4	9	Au/TiO <sub>2</sub> , 0.23 × 10 <sup>2</sup> . <sup>b</sup>	Vinyl	Toluene	<sup>b</sup>	×	×	120	H <sub>2</sub> (15 bar)	6.5	98.5	95.9
5	9	Pt/TiO <sub>2</sub> , 0.75 × 10 <sup>2</sup> . <sup>b</sup>	Chloro	THF	<sup>b</sup>	×	×	45	H <sub>2</sub> (4 bar)	1	99	99.6
6	9	Pt/TiO <sub>2</sub> , 0.24 × 10 <sup>2</sup> . <sup>b</sup>	Acetyl	Toluene	<sup>b</sup>	×	×	40	H <sub>2</sub> (3 bar)	5.6	98.9	90.1
7	2	TiO <sub>2</sub> , 50 mg	vinyl	Water	50 μmol	Hg lamp (> 300 nm)	Oxalic acid, 200 μmol	25	Ar (1 atm)	6	>99	88
8	2	TiO <sub>2</sub> , 50 mg	Chloro	Water	50 μmol	Hg lamp (> 300 nm)	Oxalic acid, 200 μmol	25	Ar (1 atm)	2	>99	>99
9	2	TiO <sub>2</sub> , 50 mg	Bromo	Water	50 μmol	Hg lamp (> 300 nm)	Oxalic acid, 200 μmol	25	Ar (1 atm)	2	>99	>99
10	2	TiO <sub>2</sub> , 50 mg	Acetyl	Water	50 μmol	Hg lamp (> 300 nm)	Oxalic acid, 200 μmol	25	Ar (1 atm)	2	>99	91
11	I	CdS/g-C <sub>3</sub> N <sub>4</sub> , 0.1 g	H	Benzotrifluoride	127.5 μmol	300 W Xe lamp (> 420 nm)	p-Methoxybenzyl alcohol, 382.5 μmol	60	N <sub>2</sub> (0.5 MPa)	4	70	57.1
12	34	DHN/1 wt%Pt-TiO <sub>2</sub> , 50 mg	H	Acetonitrile	0.1 mmol	500 W Xe lamp (> 420 nm)	TEOA, 1 mmol	R. T. <sup>c</sup>	Ar (1 atm)	4	42.7	ca. 75
13	II	Fe(bpy) <sub>3</sub> @rGO, 25 mg	Chloro	Acetonitrile	0.1 mmol	20 W LED (> 400 nm)	hydrazine monohydrate 1 mmol	R. T. <sup>c</sup>	1 atm <sup>c</sup>	10	86	97 <sup>d</sup>
14	II	Fe(bpy) <sub>3</sub> @rGO, 25 mg	Bromo	Acetonitrile	0.1 mmol	20 W LED (> 400 nm)	hydrazine monohydrate 1 mmol	R. T. <sup>c</sup>	1 atm <sup>c</sup>	12	80	94 <sup>d</sup>
15	III	CdS, 20 mg	Chloro	<i>i</i> -PrOH	50 μmol	3 W Blue LED (> 420 nm)	<i>i</i> -PrOH, solvent	R. T. <sup>c</sup>	Ar (1 atm)	20	100	90
16	III	CdS, 20 mg	Acetyl	<i>i</i> -PrOH	50 μmol	3 W Blue LED (> 420 nm)	<i>i</i> -PrOH, solvent	R. T. <sup>c</sup>	Ar (1 atm)	20	100	98
17	This study	DHN/TiO <sub>2</sub> , 50 mg	Vinyl	Acetonitrile	50 μmol	Blue LED (λ <sub>max</sub> = 467 nm)	TEOA, 300 μmol	R. T.	Ar (1 atm)	10	>99	97
18	This study	DHN/TiO <sub>2</sub> , 50 mg	Chloro	Acetonitrile	50 μmol	Blue LED (λ <sub>max</sub> = 467 nm)	TEOA, 300 μmol	R. T.	Ar (1 atm)	10	>99	98
19	This study	DHN/TiO <sub>2</sub> , 50 mg	Bromo	Acetonitrile	50 μmol	Blue LED (λ <sub>max</sub> = 467 nm)	TEOA, 300 μmol	R. T.	Ar (1 atm)	18	>99	>99
20	This study	DHN/TiO <sub>2</sub> , 50 mg	Acetyl	Acetonitrile	50 μmol	Blue LED (λ <sub>max</sub> = 467 nm)	TEOA, 300 μmol	R. T.	Ar (1 atm)	15	>99	>99

<sup>a</sup> Conv. = (Yield / Sel.) × 100, <sup>b</sup> Pt/nitro (mol), <sup>c</sup> No description, <sup>d</sup> Sel. = (Yield / Conv.) × 100, <sup>e</sup> Numbers in parentheses represent results in the presence of 50 mg of the HCO<sub>2</sub>NH<sub>4</sub>.

- I. D. Xia, X., Mengli, M. Sugang, F. Xianliang and C. Shifu, *Appl. Catal. B: Environmental*, 2014, **158**, 382-390.
- II. A. Kumar, P. Kumar, S. Paul and S.L. Jain, *Appl. Surf. Sci.*, 2016, **386**, 103-114.
- III. E. Parvin, K. Foad and Z. Zahra, *J. Photochem. Photobiol.*, 2014, **274**, 7-12.