

Supporting Information

Ni-doped CuS as an Efficient Electrocatalyst for Oxygen Evolution Reaction

Joyjit Kundu, Santimoy Khilari, Kousik Bhunia, and Debabrata Pradhan*

Materials Science Centre, Indian Institute of Technology, Kharagpur, W. B. 721 302, India

Corresponding Author

*E-mail: deb@matsc.iitkgp.ernet.in

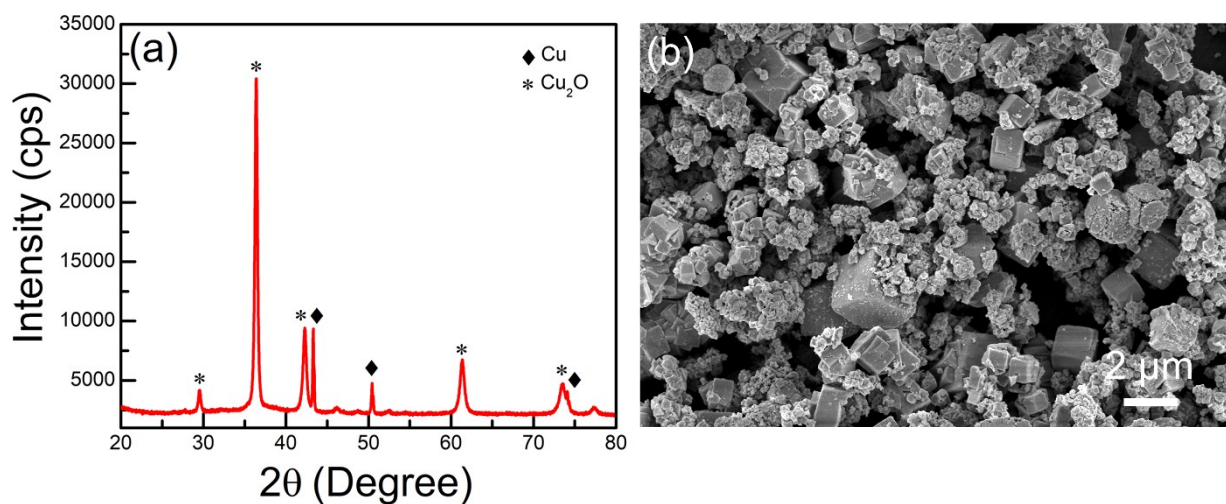
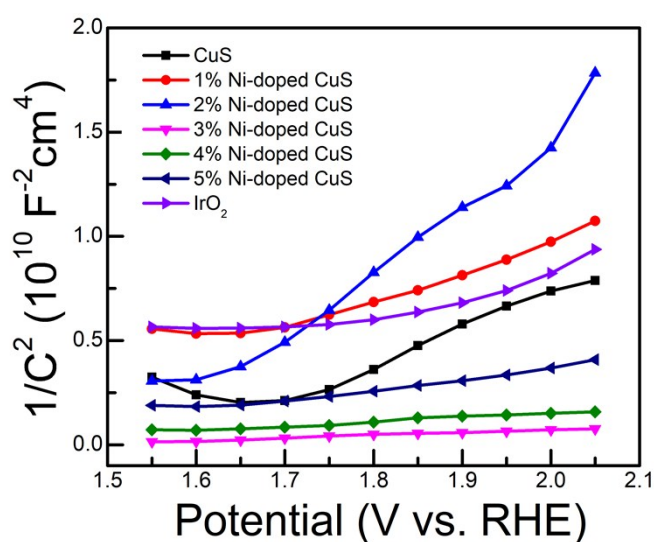


Figure S1. (a) XRD pattern and (b) FESEM image of sample prepared with without ethylenediamine.

Table S1. Comparison on the OER performance of reported electrocatalysts

Catalyst	Electrolyte	Enhancer	Overpotential (V) @10 mA/cm ²	Tafel slope (mV/dec)	Ref
Cu ₂ S NPs	250 mM phosphate buffer, pH = 13	Glycin	0.428	63	1
CuS NPs	250 mM phosphate buffer, pH = 13	Glycin	0.586	82	1
Cu ₂ S NPs	0.25 M phosphate buffer (pH = 13)	-	0.401	52	2
Co-doped Cu ₇ S ₄	1 M KOH	-	0.270	130	3
Cu ₂ S nanosheets	1.0 M KOH	-	0.336@20 mA/cm ²	101	4
CuS _{0.55} hollow NPs	1.0 M KOH	-	0.386@100 mA/cm ²	33	5
CuS Nanosheets	1.0 M KOH	-	0.408	130	6
Co ₉ S ₈ -CuS-FeS	1.0 M KOH	-	0.30	79	7
CuS/NiS ₂	0.1 M KOH	-	0.29	36	8
3% Ni-doped CuS	0.5 M KOH	-	0.39	96.8	This work

**Figure S2.** Mott-Schottky plots of pristine CuS, Ni-doped CuS, and IrO₂ in 0.5(M) KOH at 1000 Hz.

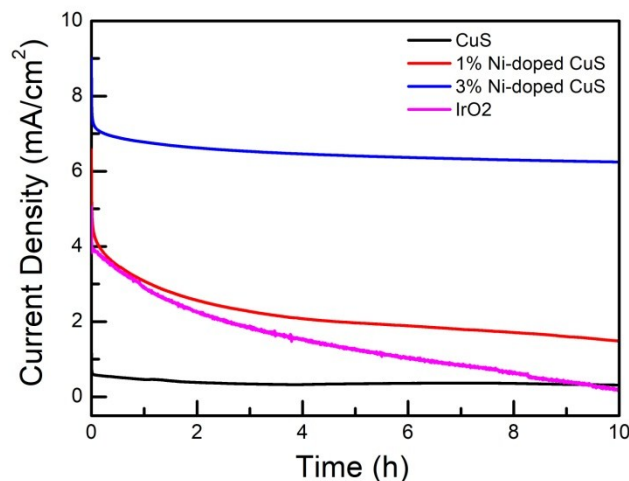


Figure S3. Chronopotentiometry plot of CuS, 1% Ni-doped CuS, 3% Ni-doped CuS and IrO₂.

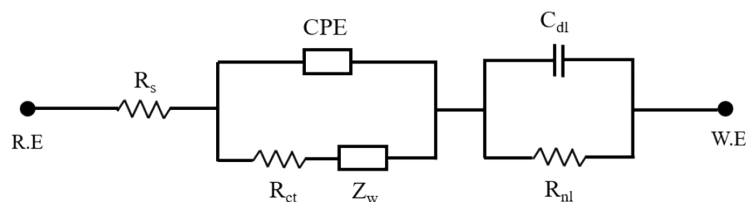


Figure S4. Equivalent circuit diagram of Nyquist plot.

References

- (1) L. An, P. Zhou, J. Yin, H. Liu, F. Chen, H. Liu, Y. Du and P. Xi, Phase Transformation Fabrication of a Cu₂S Nanoplate as an Efficient Catalyst for Water Oxidation with Glycine, *Inorg. Chem.*, 2015, **54**, 3281–3289.
- (2) X. Zhao, L. Liu, Y. Zhang, H. Zhang and Y. Wang, Uniquely Confining Cu₂S Nanoparticles in Graphitized Carbon Fibers for Enhanced Oxygen Evolution Reaction, *Nanotechnology*, 2017, **28**, 345402.
- (3) Q. Li, X. Wang, K. Tang, M. Wang, C. Wang and C. Yan, Electronic Modulation of Electrocatalytically Active Center of Cu₇S₄ Nanodisks by Cobalt Doping for Highly Efficient Oxygen Evolution Reaction, *ACS Nano*, 2017, **11**, 12230–12239.
- (4) L. He, D. Zhou, Y. Lin, R. Ge, X. Hou, X. Sun and C. Zheng, Ultrarapid in situ Synthesis of Cu₂S Nanosheet Arrays on Copper Foam with Room-Temperature-Active Iodine Plasma for Efficient and Cost-Effective Oxygen Evolution, *ACS Catal.*, 2018, **8**, 3859–3864.

- (5) H. Zhang, H. Jiang, Q. Xu, Y. Hu and C. Li, Rapid Low-temperature Synthesis of Hollow $\text{CuS}_{0.55}$ Nanoparticles for Efficient Electrocatalytic Water Oxidation, *Chem. Eng. J.*, doi.org/10.1016/j.ces.2018.10.011.
- (6) H. Liang, W. Shuang, Y. Zhang, S. Chao, H. Han, X. Wang, H. Zhang and L. Yang, Graphene-Like Multilayered CuS Nanosheets Assembled into Flower-Like Microspheres and Their Electrocatalytic Oxygen Evolution Properties, *ChemElectroChem*, 2018, **5**, 494–500.
- (7) S. Zhang, Y. Sun, F. Liao, Y. Shen, H. Shi and M. Shao, Co_9S_8 -CuS-FeS Trimetal Sulfides for Excellent Oxygen Evolution Reaction Electrocatalysis, *Electrochim. Acta*, 2018, **283**, 1695–1701.
- (8) L. An, Y. Li, M. Luo, J. Yin, Y.-Q. Zhao, C. Xu, F. Cheng, Y. Yang, P. Xi and S. Guo, Atomic-Level Coupled Interfaces and Lattice Distortion on CuS/NiS₂ Nanocrystals Boost Oxygen Catalysis for Flexible Zn-Air Batteries, *Adv. Funct. Mater.*, 2017, **27**, 1703779.