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Electronic supplementary information

One-pot hydrothermal synthesis of MoS₂-modified Mn_{0.5}Cd_{0.5}S solid solution for boosted H₂ production activity under visible light

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Fig. S1 N₂ adsorption-desorption isotherms (a) and BJH pore size distribution curves (b) of the the as-prepared $Mn_xCd_{1-x}S$ products ($0.25 \le x \le 0.5$) and 0.5wt% $MoS_2/Mn_{0.5}Cd_{0.5}S$ composite.



Fig. S2 Survey XPS spectra of the obtained 0.5wt% MoS₂/Mn_{0.5}Cd_{0.5}S composite before and after 20 h irradiation for the photoreaction as well as the single MoS₂.



Fig. S3 The bandgap energy of the obtained $Mn_{0.5}Cd_{0.5}S$ solid solution by a relation of $(\alpha hv)^{1/2} vs$. Photo energy (hv).



Fig. S4 Comparisons of XRD patterns of the obtained 0.5wt% MoS₂/Mn_{0.5}Cd_{0.5}S before and after 20 h irradiation for the photoreaction.



Fig. S5 Mott-Schottky plots of the $Mn_{0.5}Cd_{0.5}S$ film electrode obtained at a frequency of 1, 5 or 10 KHz in a Na_2SO_4 solution (1.0 M).



Fig. S6 Photocurrent response curves of the obtained 0.5wt% $MoS_2/Mn_{0.5}Cd_{0.5}S$ and the pristine $Mn_{0.5}Cd_{0.5}S$ under $\lambda \ge 420$ nm light irradiation.



Fig. S7 EIS Nyquist plots of the obtained 0.5wt% $MoS_2/Mn_{0.5}Cd_{0.5}S$ and the pristine $Mn_{0.5}Cd_{0.5}S$ under $\lambda \ge 420$ nm light irradiation.