

Supporting Information

Design and Catalytic Performance of Molybdenum Active Sites on MCM-41 Framework for the Aerobic Oxidation of 5- Hydroxymethylfurfural to 2,5-Diformylfuran

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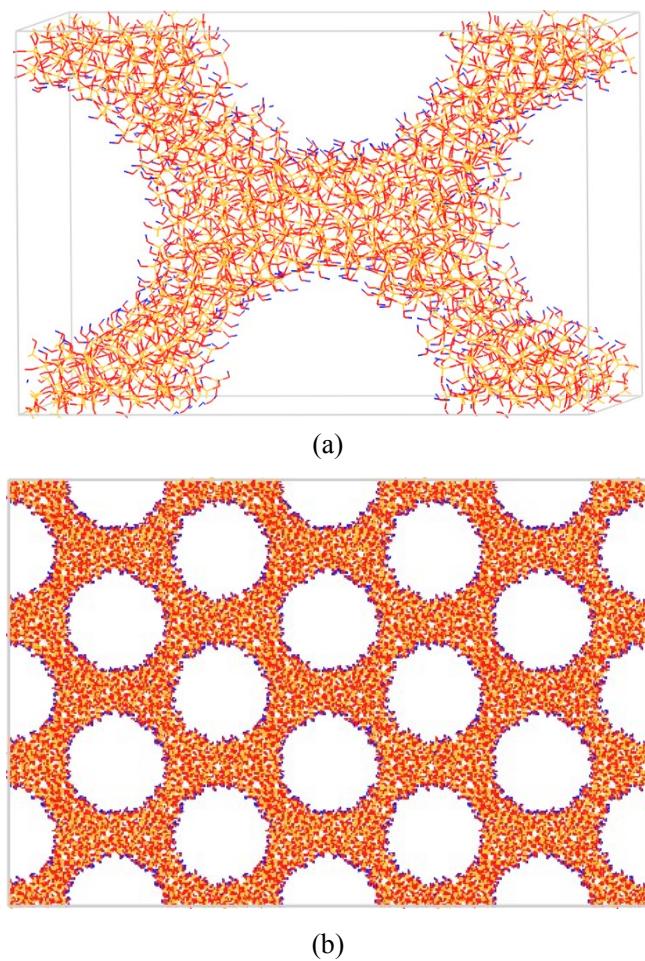


Figure S1: (a) Final structure of MCM-41 after optimization; (b) The structure obtained by replicating the box. Color code: yellow, silicon; red, oxygen; blue, hydrogen.

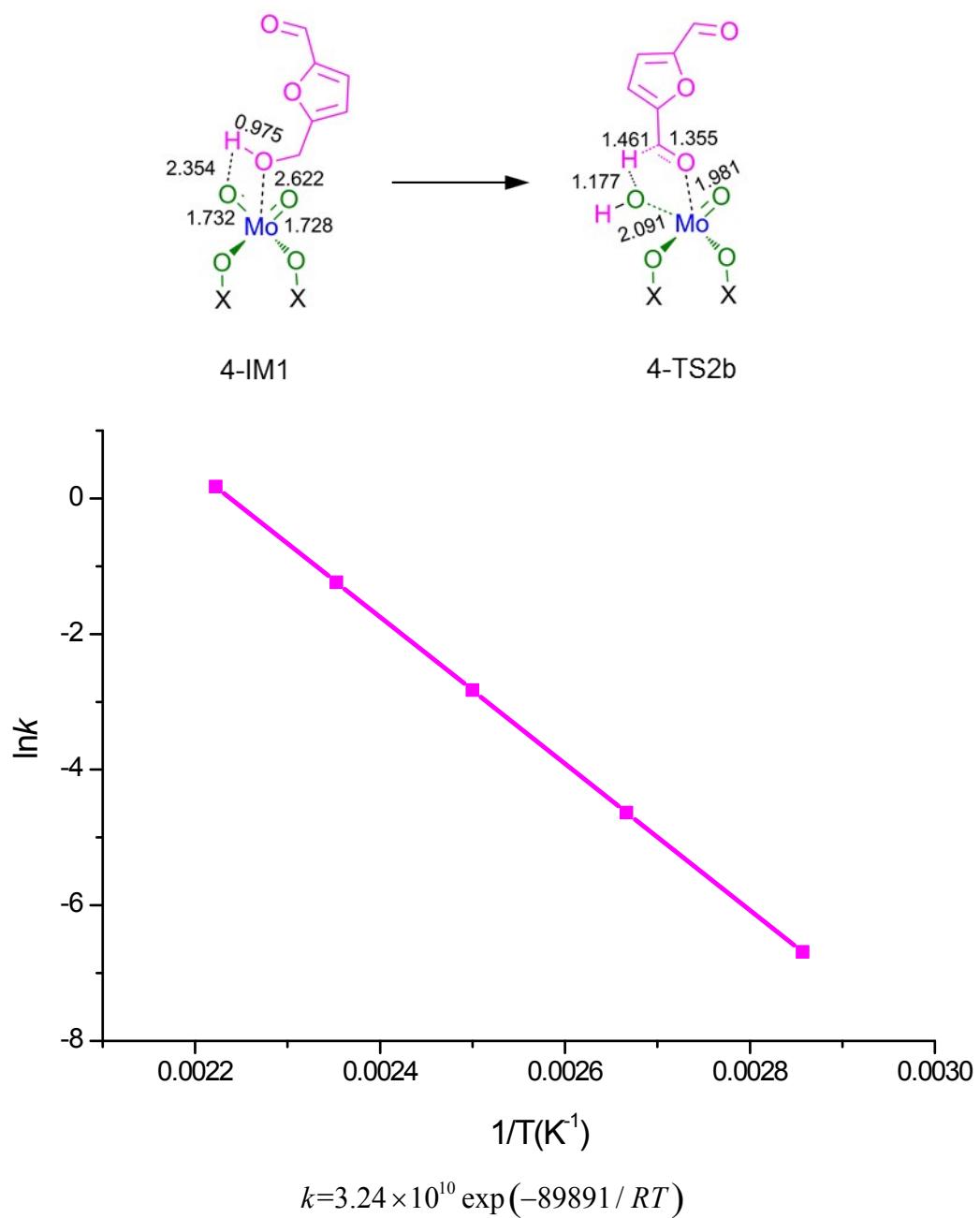


Figure S2: Arrhenius plots of rate constants for the crucial reaction step $4\text{-IM1} \rightarrow 4\text{-TS2b}$ in the catalytic oxidation of HMF to DFF catalyzed over [Mo-4].

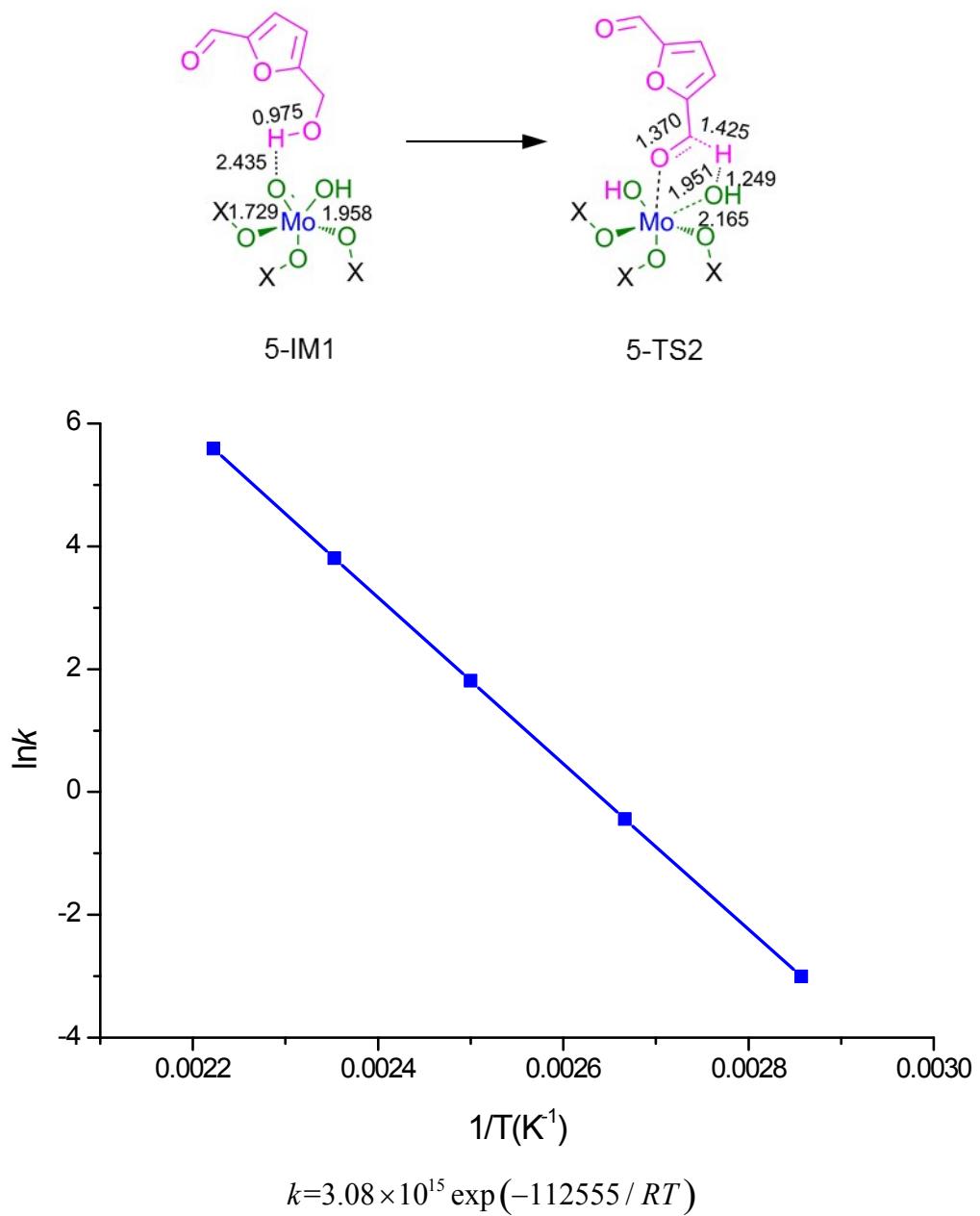
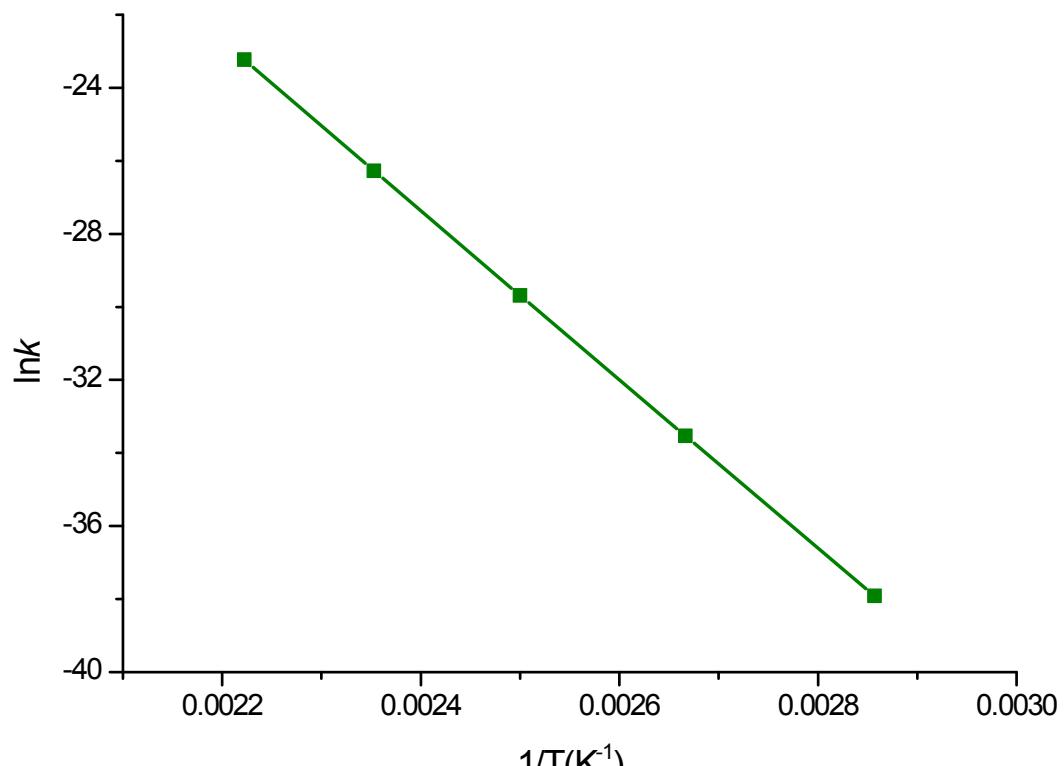
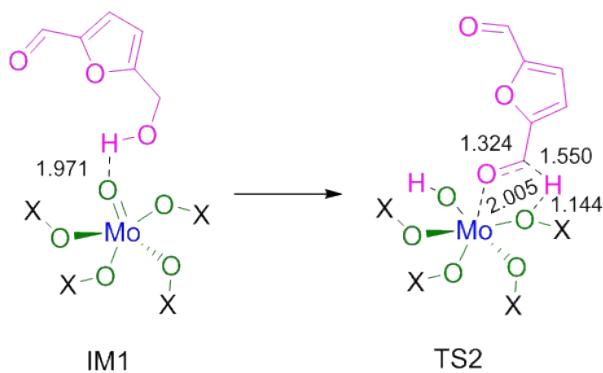


Figure S3: Arrhenius plots of rate constants for the crucial reaction step $5\text{-IM1} \rightarrow 5\text{-TS2}$ in the catalytic oxidation of HMF to DFF catalyzed over [Mo-5].



$$k_{\text{vib, up}} \equiv 9.89 \times 10^{11} \exp(-190648 / RT)$$

Figure S4: Arrhenius plots of rate constants for the crucial reaction step IM1 → TS2 in the catalytic oxidation of HMF to DFF catalyzed over [Mo-HPA].

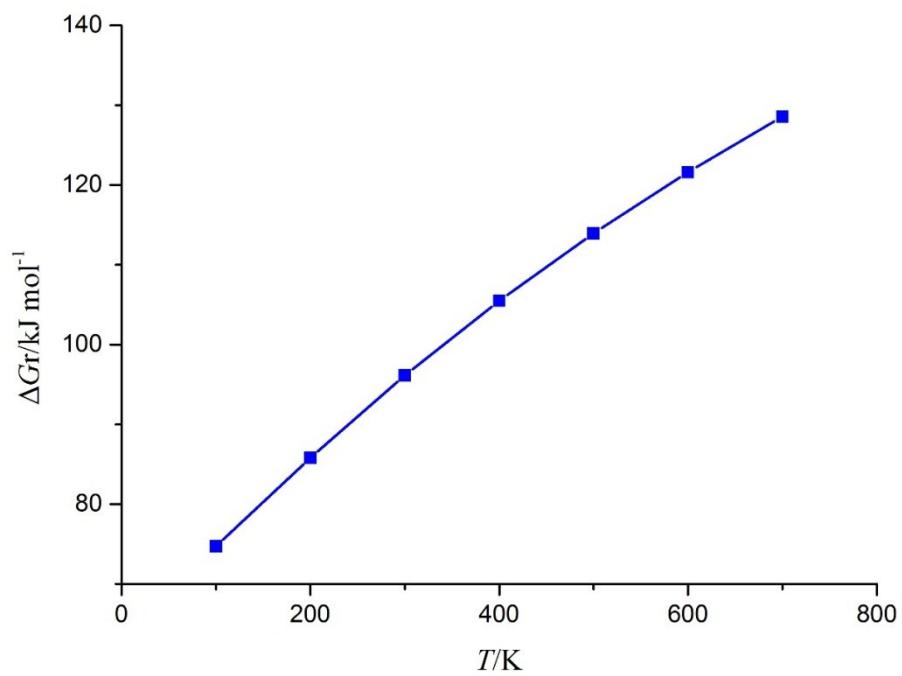


Figure S5: Diagram (relative energies ([Mo-5]-[Mo-4]) vs. temperature) showing the stability ranges for the different molybdenum doped MCM-41 silica

Table S1: Sum of electronic energies (E , hartree) and the relative energies (E_r , kJ mol⁻¹) of various species with respect to the reactants for the reaction of 2HMF + O₂ → 2DFF + 2H₂O catalyzed over [Mo-4] at the hybrid quantum mechanics (QM, GGA-BP/DNP) and molecular mechanics (MM, Universal) calculations.

Species	E	E_r
HMF	-458.02462	
DFF	-456.82137	
³ O ₂	-150.38903	
O ₂	-150.34313	
H ₂ O	-76.45648	
[Mo-4]	-7058.17368	0.0
4-IM1	-7516.20059	-6.0
4-TS1	-7516.16618	84.3
4-IM2	-7516.18688	30.0
4-TS2a	-7516.15152	122.8
4-IM3a	-7516.16384	90.5
4-TS2b	-7516.15766	106.7
4-IM3b	-7516.18517	34.5
[Mo-4+2H _a]	-7059.32141	145.8
[Mo-4+2H _b]	-7059.32862	126.8
³ 4-IM4a	-7209.74869	45.3
4-IM4a	-7209.77235	-16.4
4-TS3a	-7209.75879	19.2
4-IM5a	-7209.78396	-47.3
4-TS4a	-7209.73084	92.2
4-TS3b	-7059.28247	248.0
4-IM5b	-7209.79430	-74.4
³ 4-IM5b	-7209.75413	31.1
4-IM6a	-7209.79025	-63.8
[Mo-4+O]	-7133.32780	-48.1
[³ Mo-4+O]	-7133.28935	52.9
4-IM7	-7591.35264	-48.7
4-TS5	-7591.33832	-11.1
4-IM8	-7591.34422	-26.6
4-TS6a	-7591.31348	54.1
4-IM9a	-7591.33083	8.6
4-TS6b	-7591.29942	91.0
4-IM9b	-7591.45610	-320.3
4-IM10a	-7134.61737	-274.7
4-TS7a	-7134.59537	-217.0
[Mo-4]+ H ₂ O	-7134.63201	-313.2
[Mo-4]	-7058.17368	-308.3

Table S2: Sum of electronic energies (E , hartree) and the relative energies (E_r , kJ mol $^{-1}$) of various species with respect to the reactants for the reaction of 2HMF + O $_2$ → 2DFF + 2H $_2$ O catalyzed over [Mo-4] at the hybrid quantum mechanics (QM, GGA-BP/DNP) and molecular mechanics (MM, Universal) calculations.

Species	E	E_r
[Mo-5]	-7865.57214	0.0
5-IM1	-8323.61608	-50.7
5-TS1	-8323.58013	43.7
5-IM2	-8323.59769	-2.4
5-TS2	-8323.56803	75.4
5-IM3	-8323.60520	-22.2
5-IM4	-7866.78414	-23.0
[Mo-5-O]	-7790.30288	42.1
35-IM5	-7940.71844	-27.6
5-IM5	-7940.74120	-87.3
5-TS3	-7940.73011	-58.2
[Mo-5+O]	-7940.75340	-119.4
5-IM6	-8398.76884	-95.3
5-TS4	-8398.74356	-28.9
5-IM7	-8398.76280	-79.4
5-TS5	-8398.71781	38.7
5-IM8	-8398.86530	-348.5
[Mo-5]	-7865.57214	-308.3