

*Electronic Supporting Information for*

**Hydroboration of terminal olefins with pinacolborane catalyzed  
by new 2-iminopyrrolyl iron(II) complexes**

Tiago F. C. Cruz,<sup>a</sup> Laura C. J. Pereira,<sup>b</sup> João C. Waerenborgh,<sup>b</sup> Luís F. Veiros<sup>a</sup> and  
Pedro T. Gomes\*<sup>a</sup>

<sup>a</sup> Centro de Química Estrutural, Departamento de Engenharia Química, Instituto Superior Técnico, Universidade de Lisboa, Av. Rovisco Pais 1, 1049-001 Lisboa, Portugal.

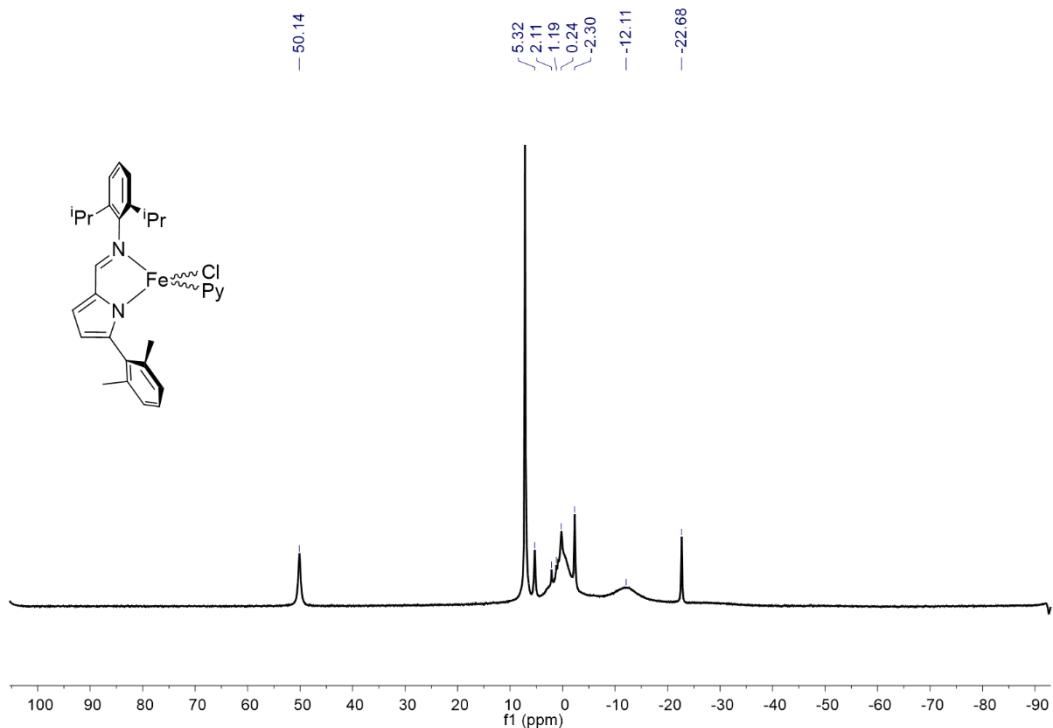
<sup>b</sup> C<sup>2</sup>TN-Centro de Ciências e Tecnologias Nucleares, Instituto Superior Técnico, Universidade de Lisboa, 2695-066 Bobadela LRS, Portugal.

\*Corresponding author. E-mail: [pedro.t.gomes@tecnico.ulisboa.pt](mailto:pedro.t.gomes@tecnico.ulisboa.pt)

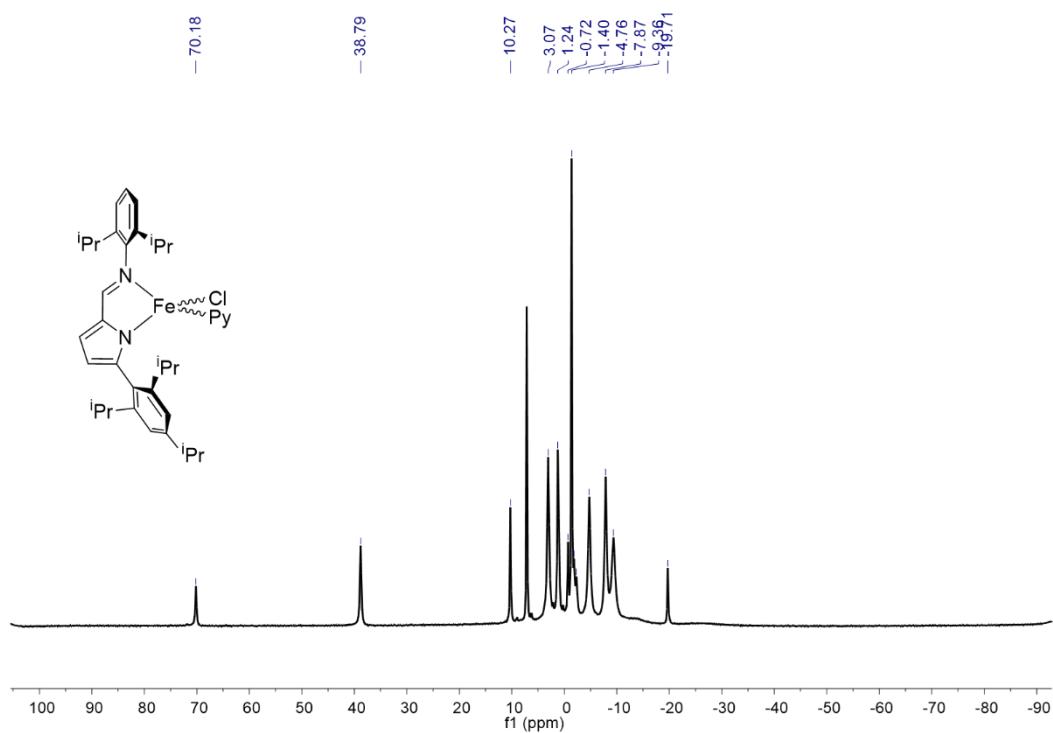
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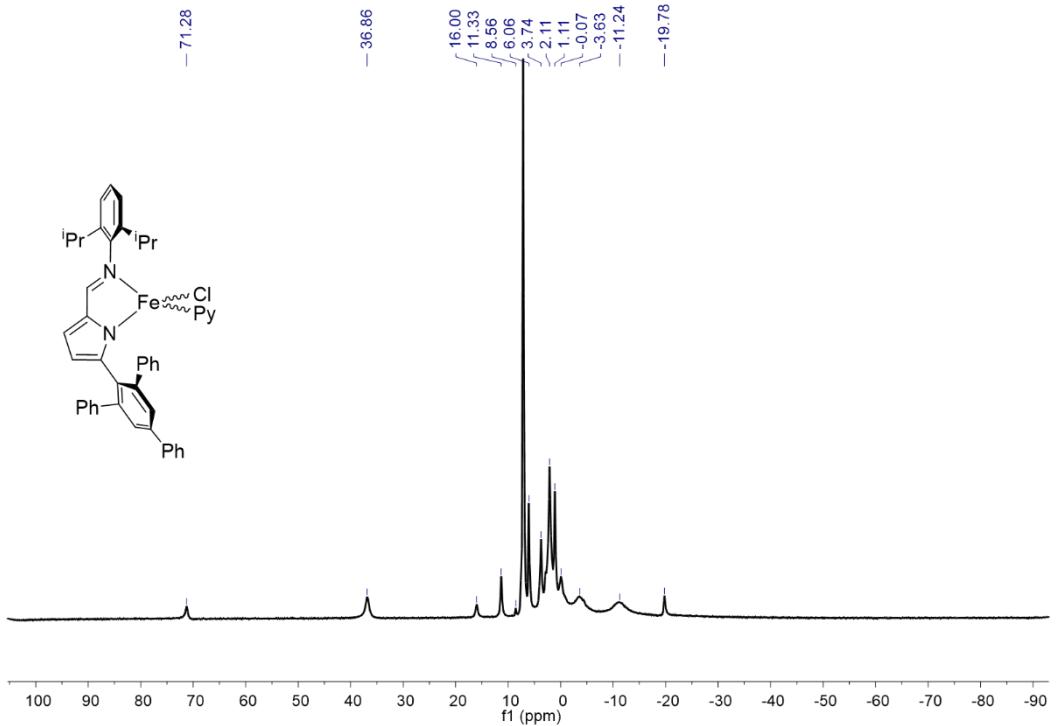
## <sup>1</sup>H NMR spectra of complexes **1a-d** and mixtures **I** and **II**



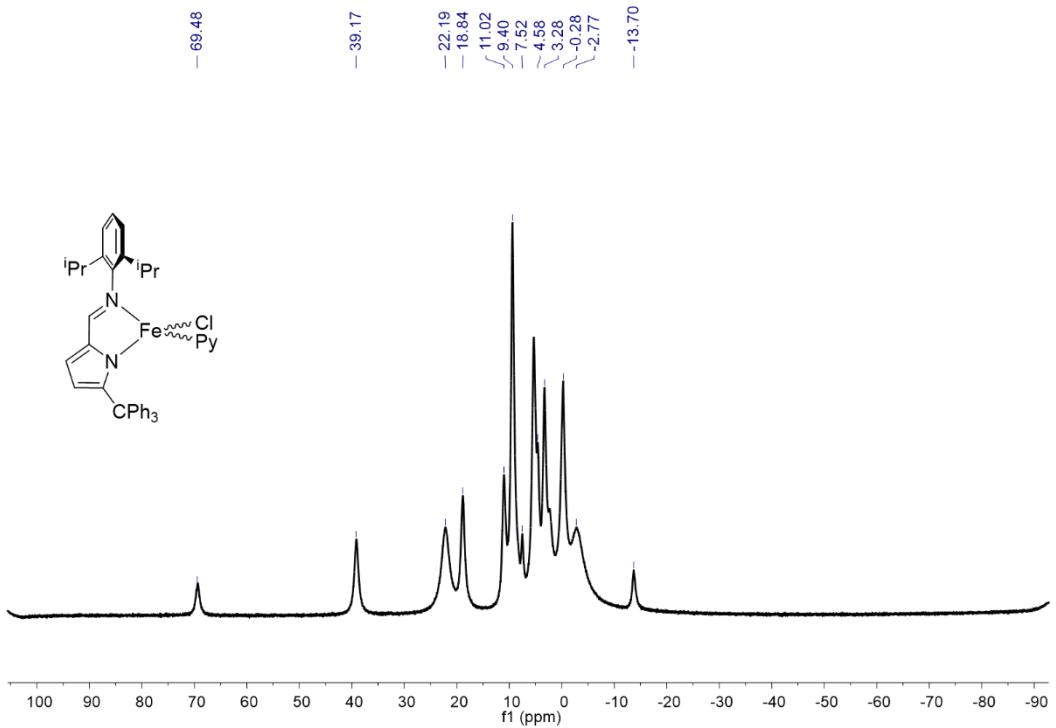
**Figure S1** <sup>1</sup>H NMR spectrum (300 MHz,  $\text{C}_6\text{D}_6$ ) of complex **1a**.



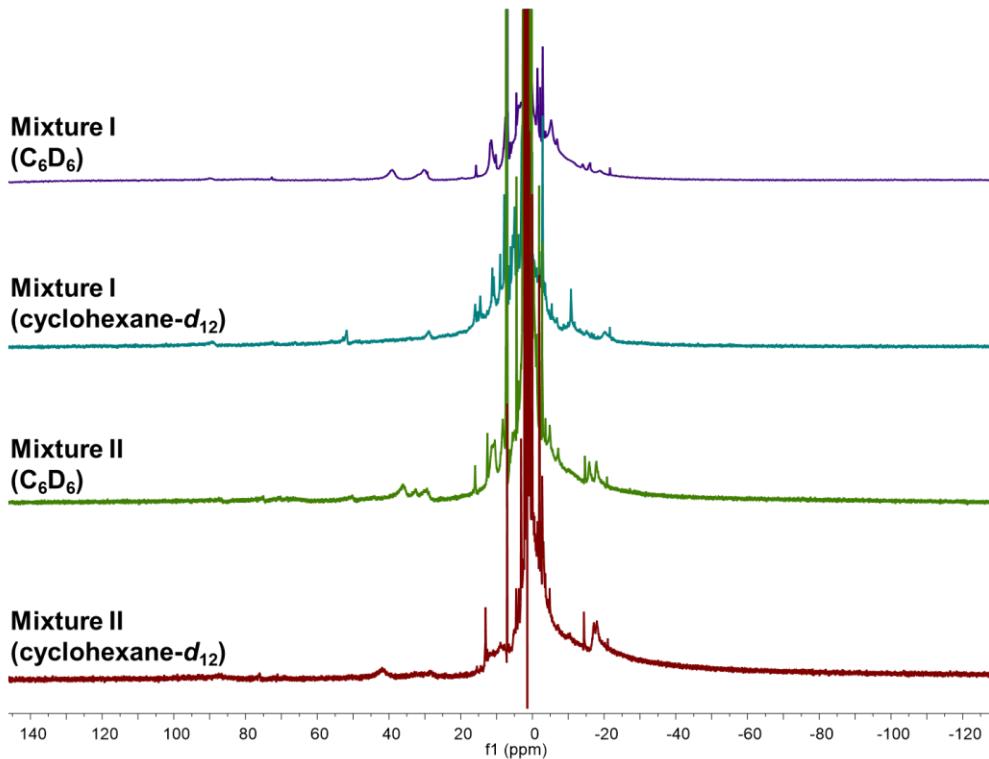
**Figure S2** <sup>1</sup>H NMR spectrum (300 MHz,  $\text{C}_6\text{D}_6$ ) of complex **1b**.



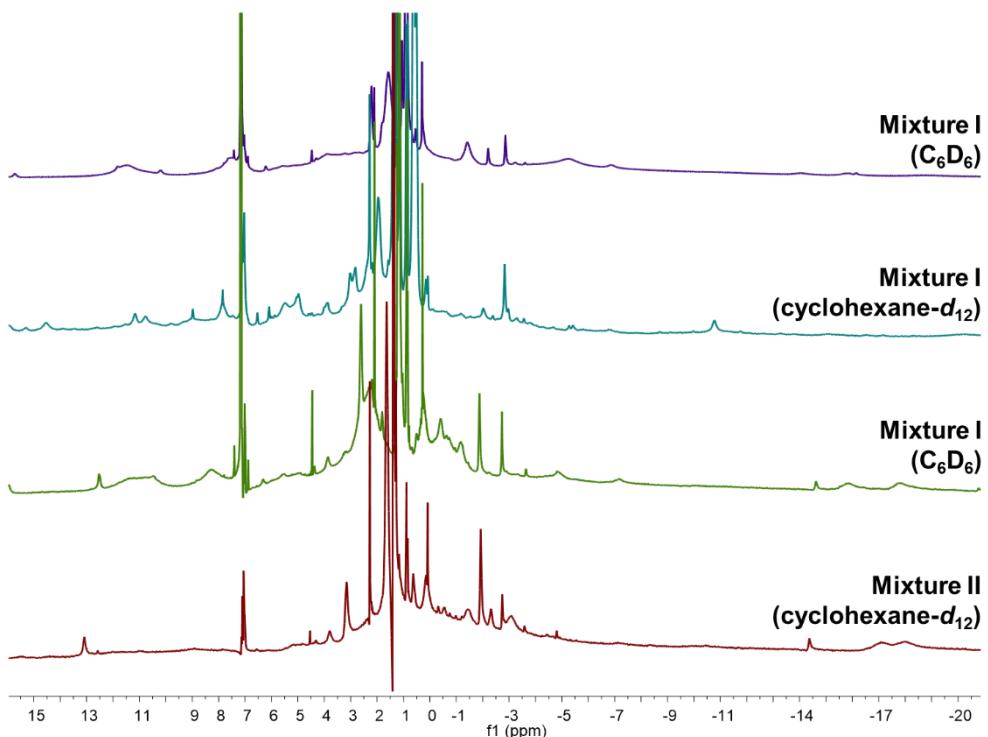
**Figure S3**  $^1\text{H}$  NMR spectrum (300 MHz,  $\text{C}_6\text{D}_6$ ) of complex **1c**.



**Figure S4**  $^1\text{H}$  NMR spectrum (300 MHz,  $\text{CDCl}_3$ ) of complex **1d**.

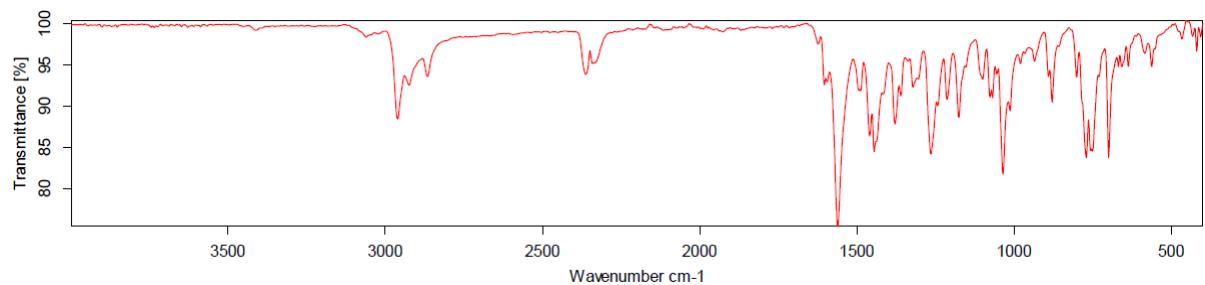


**Figure S5**  $^1\text{H}$  NMR spectra (300 MHz,  $\text{C}_6\text{D}_6$  or cyclohexane- $d_{12}$ ) of fairly concentrated mixtures I and II, demonstrating their rather uninformative nature. The intense resonances observed correspond to the protio-residual resonance(s) of the respective deuterated solvents.

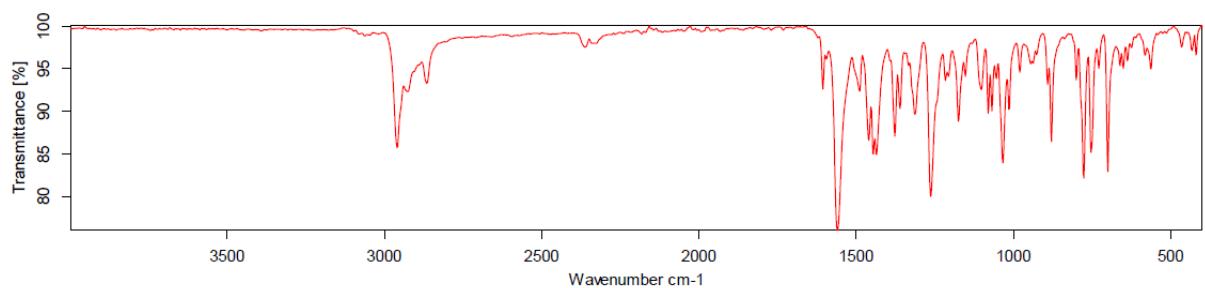


**Figure S6** Magnification of the diamagnetic region of the  $^1\text{H}$  NMR spectra (300 MHz,  $\text{C}_6\text{D}_6$  or cyclohexane- $d_{12}$ ) of fairly concentrated mixtures I and II, demonstrating their rather uninformative nature. The intense resonances observed correspond to the protio-residual resonance(s) of the respective deuterated solvents.

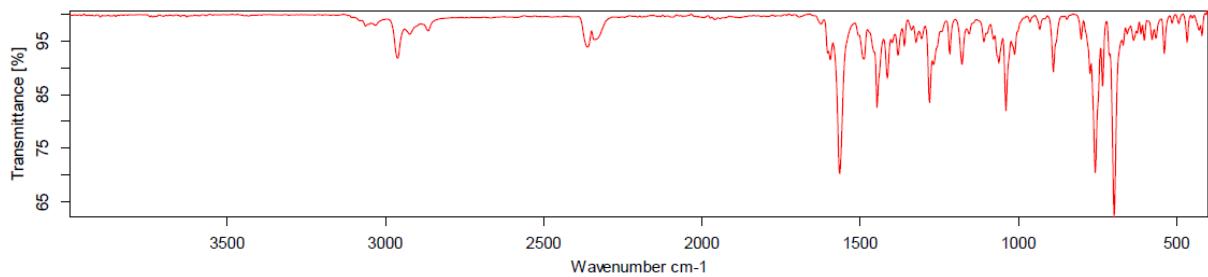
## FTIR spectra of complexes **1a-d** and mixtures I and II



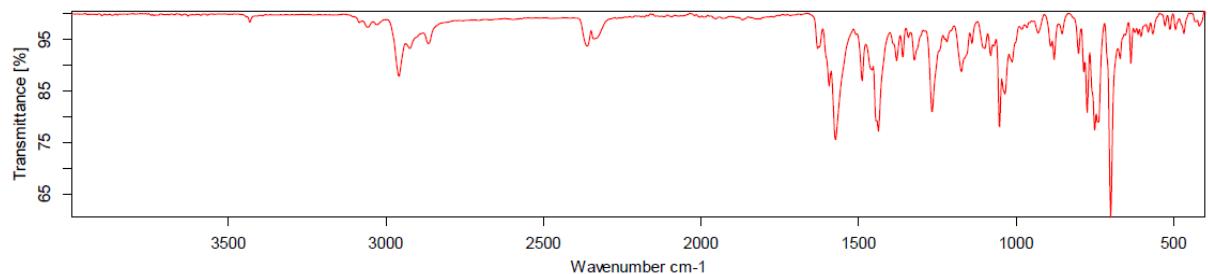
**Figure S7** ATR-FTIR spectrum of complex **1a**.



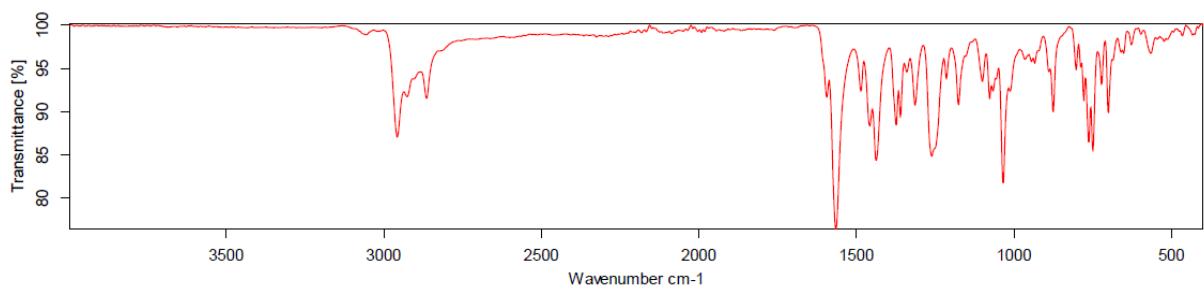
**Figure S8** ATR-FTIR spectrum of complex **1b**.



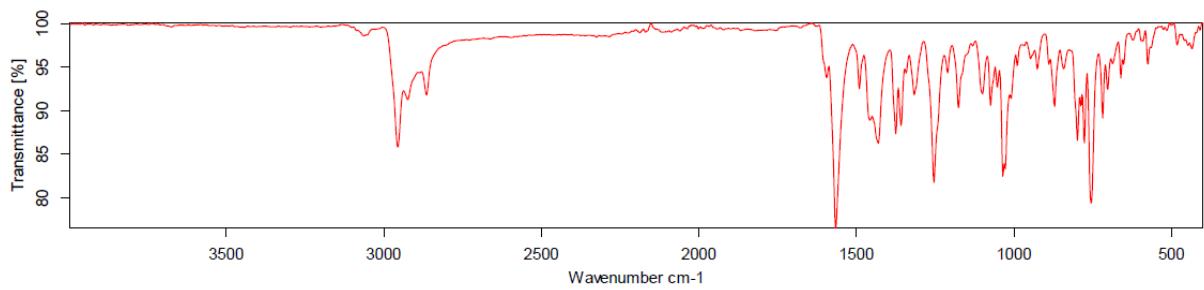
**Figure S9** ATR-FTIR spectrum of complex **1c**.



**Figure S10** ATR-FTIR spectrum of complex **1d**.



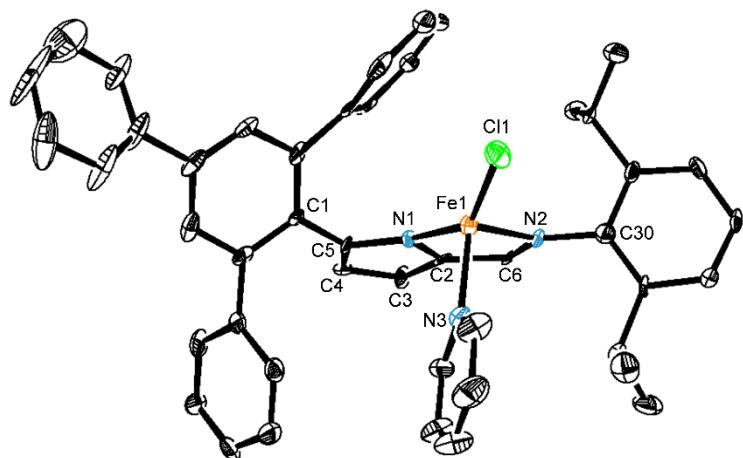
**Figure S11** ATR-FTIR spectrum of the mixture of complexes isolated from the 30 min reaction of complex **1b** with K(HBEt<sub>3</sub>) in toluene (mixture I).



**Figure S12** ATR-FTIR spectrum of the mixture of complexes isolated from the 16 h reaction of complex **1b** with K(HBEt<sub>3</sub>) in toluene (mixture II).

## X-ray diffraction structure of complex **1c**

Even though we obtained crystals of complex **1c** suitable for X-ray diffraction, they revealed to be of poor quality, presenting a relatively low ratio of observed/unique reflections. Although the structure of complex **1c** was refined to a perfect convergence, it is only presented in this section (Figure S6) as a proof of its connectivity (selected bond lengths and angles presented in Table S1).



**Figure S13** ORTEP-3 diagram for complex **1c** showing 30% probability ellipsoids. All hydrogen atoms were omitted for clarity.

## Single crystal X-ray diffraction data

**Table S1** Selected bond lengths ( $\text{\AA}$ ) and angles ( $^\circ$ ) for complexes **1a** and **1c**.

<i>Bond lengths</i>	<b>1a</b>	<b>1c</b>	<i>Bond angles</i>	<b>1a</b>	<b>1c</b>
Fe1–N1	2.0308(19)	2.001(6)	N1–Fe1–N3	103.96(7)	98.3(3)
Fe1–N2	2.112(2)	2.111(8)	N1–Fe1–N2	81.12(8)	82.2(3)
Fe1–N3	2.0917(18)	2.097(6)	N3–Fe1–N2	119.72(8)	105.1(3)
Fe1–C11	2.2345(7)	2.231(3)	N1–Fe1–C11	136.05(6)	144.00(19)
N1–C5	1.354(3)	1.374(9)	N3–Fe1–C11	104.85(5)	106.3(2)
N1–C2	1.378(3)	1.372(10)	N2–Fe1–C11	111.43(5)	115.17(19)
N2–C6	1.289(3)	1.309(9)	C5–N1–C2	107.64(18)	104.1(6)
N2–C <sub>ipso</sub>	1.438(3)	1.446(9)	C6–N2–C <sub>ipso</sub>	118.9(2)	118.3(7)
C4–C5	1.405(4)	1.375(11)	N1–C5–C4	109.1(2)	111.4(7)
C1–C5	1.479(3)	1.471(10)	N1–C5–C1	121.2(2)	119.9(7)
C2–C6	1.419(3)	1.435(10)	C4–C5–C1	129.6(2)	127.7(7)
C2–C3	1.401(3)	1.400(10)	N2–C6–C2	119.2(2)	118.5(8)
C3–C4	1.387(4)	1.388(10)	N1–C2–C3	109.3(2)	111.8(7)
			N1–C2–C6	117.9(2)	117.7(7)
			C3–C2–C6	132.8(2)	129.7(8)
			C4–C3–C2	106.4(2)	105.0(8)
			C3–C4–C5	107.5(2)	107.5(7)

**Table S2** Selected bond lengths ( $\text{\AA}$ ) and angles ( $^\circ$ ) for complex **2<sub>2</sub>**.

	<i>Bond lengths</i>	<i>Bond angles</i>
Fe1–N1	2.059(2)	N1–Fe1–N2
Fe1–N2	2.155(2)	N1–Fe1–N3
Fe1–N3	2.180(3)	N2–Fe1–N3
Fe1–Fe2	2.4874(5)	N1–Fe1–H1A
Fe1–H1A	1.69(3)	N2–Fe1–H1A
Fe1–H1B	1.68(4)	N3–Fe1–H1A
Fe2–N4	1.990(2)	N1–Fe1–H1B
Fe2–N5	2.025(2)	N2–Fe2–H1B
Fe2–H1A	1.55(3)	N3–Fe1–H1B
Fe2–H1B	1.55(4)	H1A–Fe1–H1B
N1–C5	1.354(4)	N4–Fe2–N5
N1–C2	1.399(3)	N4–Fe2–H1A
N2–C6	1.303(4)	N5–Fe2–H1A
N2–C <sub>ipso</sub>	1.435(4)	N4–Fe2–H1B
C1–C5	1.484(4)	N5–Fe2–H1B
C2–C3	1.382(4)	H1A–Fe2–H1B
C2–C6	1.400(4)	C5–N1–C2
C3–C4	1.377(4)	C6–N2–C <sub>ipso</sub>
C4–C5	1.402(4)	C3–C2–N1
		C3–C2–C6
		N1–C2–C6
		C4–C3–C2
		C3–C4–C5
		N1–C5–C4
		N1–C5–C1
		C4–C5–C1
		123.4(3)

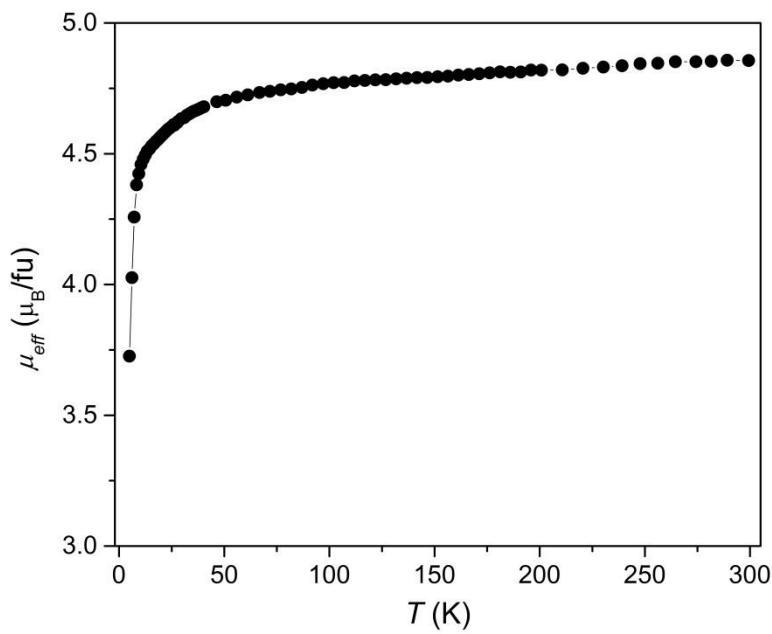
**Table S3** Selected bond lengths (Å) and angles (°) for complex **3**

<i>Bond lengths</i>		<i>Bond angles</i>	
Fe1–N1	1.971(7)	N1–Fe1–N2	82.7(3)
Fe1–N2	1.977(8)	N1–Fe1–C33	129.8(3)
Fe1–C33	2.110(11)	N2–Fe1–C36	125.5(3)
Fe1–C38	2.060(11)	C33–Fe1–C36	85.6(4)
Fe1–C37	2.110(11)	N2–Fe1–C33	99.8(4)
Fe1–C34	2.091(8)	N1–Fe1–C36	119.1(4)
Fe1–C36	2.149(10)	C5–N1–C2	108.2(7)
Fe1–C35	2.062(8)	N1–C2–C3	109.3(8)
N1–C5	1.383(11)	N1–C2–C6	115.1(8)
N1–C2	1.369(12)	C3–C2–C6	135.6(10)
N2–C6	1.293(11)	C4–C3–C2	107.6(9)
N2–C <sub>ipso</sub>	1.443(11)	C3–C4–C5	107.2(8)
C1–C5	1.456(13)	N1–C5–C1	125.2(8)
C2–C3	1.368(12)	N1–C5–C4	107.7(9)
C2–C6	1.423(12)	N2–C6–C2	117.9(10)
C3–C4	1.401(15)		
C4–C5	1.398(12)		

**Table S4** Crystallographic data for compounds **1a**, **1c**, **2<sub>2</sub>** and **3**.

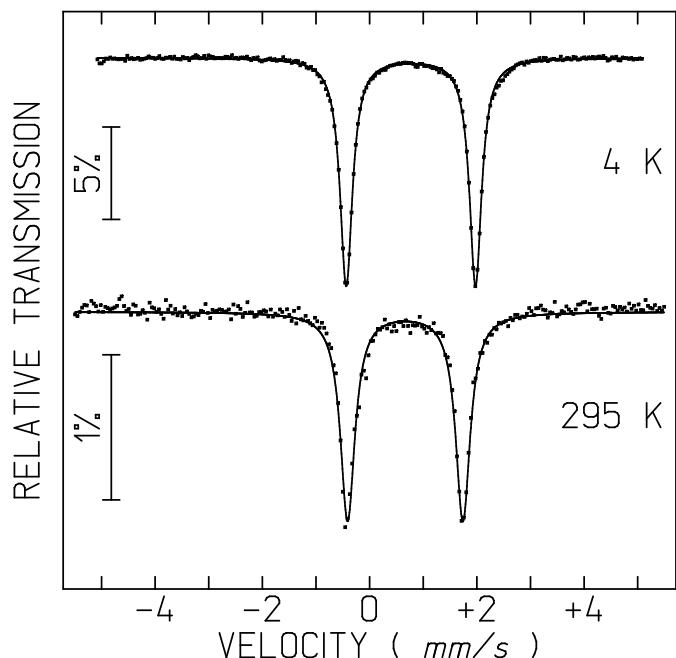
	<b>1a</b>	<b>1c</b>	<b>2<sub>2</sub></b>	<b>3</b>
Formula	C <sub>30</sub> H <sub>34</sub> ClFeN <sub>3</sub>	C <sub>46</sub> H <sub>42</sub> ClFeN <sub>3</sub>	C <sub>69</sub> H <sub>93</sub> Fe <sub>2</sub> N <sub>5</sub>	C <sub>39</sub> H <sub>51</sub> FeN <sub>2</sub>
M	527.90	728.12	1104.18	603.66
λ (Å)	0.71073	0.71073	0.71073	0.71073
T (K)	150	150	150	150
Crystal system	Monoclinic	Monoclinic	Triclinic	Triclinic
Space group	P2 <sub>1</sub>	P2 <sub>1</sub> /c	P-1	P-1
a (Å)	8.9968(3)	20.583(4)	14.0411(13)	9.606(6)
b (Å)	18.4440(8)	11.705(3)	16.3354(14)	10.504(6)
c (Å)	9.3737(4)	19.314(4)	17.4433(18)	17.317(10)
α (°)	90	90	107.001(4)	103.90(2)
β (°)	117.856(2)	116.648(6)	92.917(4)	93.89(3)
γ (°)	90	90	100.939(4)	91.43(3)
V (Å <sup>3</sup> )	1375.21	4158.8(16)	3732.4(6)	1690.7(18)
Z	2	4	2	2
ρ <sub>calc</sub> (g.cm <sup>-3</sup> )	1.275	1.163	0.983	1.186
μ (mm <sup>-1</sup> )	0.668	0.460	0.425	0.474
Crystal size	0.20×0.10×0.10	0.20×0.20×0.20	0.30×0.20×0.20	0.05×0.05×0.1
θ <sub>max</sub> (°)	35.804	25.706	25.61	25.91
Total data	26240	11858	14088	9187
Unique data	12266	7745	9548	5568
R <sub>int</sub>	0.0398	0.1008	0.0483	0.0724
R [I > 2σ(I)]	0.0452	0.1008	0.0562	0.1314
R <sub>w</sub>	0.0805	0.2386	0.1530	0.3289
Goodness of fit	0.980	0.854	1.015	1.122
ρ <sub>min</sub>	-0.445	-0.494	-0.496	-0.770
ρ <sub>max</sub>	0.431	0.515	0.637	1.443

## SQUID Magnetometry data for complex **1a**



**Figure S14** Temperature dependence of the effective magnetic moment,  $\mu_{\text{eff}}$ , of complex **1a** taken at 500 Oe.

## <sup>57</sup>Fe Mössbauer spectroscopy data of complex **1a** and mixtures I and II



**Figure S15** Mössbauer spectra of complex **1a** at 4 and 295 K. Lines over the experimental points are the calculated doublets by the refinement procedure (Table S2).

**Table S5** Estimated parameters from the Mössbauer spectra of complex **1a** taken at 4 and 230 K.<sup>d</sup>

T	IS <sup>a</sup>	QS <sup>b</sup>	Fe species, S <sup>c</sup>
4 K	0.88	2.41	Fe(II) (S=2)
295 K	0.77	2.16	Fe(II) (S=2)

<sup>a</sup> IS, Isomer shift (mm s<sup>-1</sup>) relative to metallic  $\alpha$ -Fe at 295

K. <sup>b</sup> QS, Quadrupole splitting (mm s<sup>-1</sup>). <sup>c</sup> S, spin quantum

number. <sup>d</sup> Estimated errors are  $\leq 0.02$  mm s<sup>-1</sup> for IS and

QS.

**Table S3** Estimated parameters from the Mössbauer spectra of the isolated mixtures of complexes, taken at different temperatures (T), from the reaction of complex **1b** with K(HBET<sub>3</sub>) in toluene at different reaction times.

Catalyst	Reaction time	T	IS <sup>a</sup>	QS ( $\epsilon$ ) <sup>b</sup>	I <sup>c</sup>	Fe species <sup>d</sup>
Mixture I	30 min	4 K	0.69	3.08	33%	Fe(II), CN=4, dimer ( <b>2<sub>2</sub></b> )
			0.96	1.92	32%	Fe(II), CN=5, dimer ( <b>2<sub>2</sub></b> )
			0.63	1.93	35%	Fe(II), CN=4, monomer ( <b>2</b> )
		150 K	0.65	2.84	32%	Fe(II), CN=4, dimer ( <b>2<sub>2</sub></b> )
			0.88	1.80	31%	Fe(II), CN=5, dimer ( <b>2<sub>2</sub></b> )
		230 K	0.65	1.90	37%	Fe(II), CN=4, monomer ( <b>2</b> )
			0.57	2.64	30%	Fe(II), CN=4, dimer ( <b>2<sub>2</sub></b> )
			0.81	1.70	31%	Fe(II), CN=5, dimer ( <b>2<sub>2</sub></b> )
			0.62	1.78	39%	Fe(II), CN=4, monomer ( <b>2</b> )
Mixture II	16 h	4 K	0.69	3.10	28%	Fe(II), CN=4, dimer ( <b>2<sub>2</sub></b> )
			1.06	2.08	29%	Fe(II), CN=5, dimer ( <b>2<sub>2</sub></b> )
			0.60	1.99	34%	Fe(II), CN=4, monomer ( <b>2</b> )
		150 K	0.34	1.66	10%	Fe(I) ( <b>3</b> )
			0.68	2.89	28%	Fe(II), CN=4, dimer ( <b>2<sub>2</sub></b> )
			0.96	1.82	28%	Fe(II), CN=5, dimer ( <b>2<sub>2</sub></b> )
		230 K	0.59	1.89	38%	Fe(II), CN=4, monomer ( <b>2</b> )
			0.34	1.48	6%	Fe(I) ( <b>3</b> )

<sup>a</sup>IS, Isomer shift (mm s<sup>-1</sup>) relative to metallic  $\alpha$ -Fe at 295 K. <sup>b</sup>QS, Quadrupole splitting (mm s<sup>-1</sup>). <sup>c</sup>I, relative areas.

<sup>d</sup>CN, coordination number. <sup>e</sup>Estimated errors are  $\leq 0.02$  mm s<sup>-1</sup> for IS and QS and  $\leq 2\%$  for I.

## Atomic coordinates for the DFT optimized complexes

### **1a ( $S = 2$ )**

Fe	3.023344000	9.420131000	2.770009000	H	6.210379000	14.119305000	3.114095000
Cl	1.189648000	8.748725000	3.882604000	H	5.141869000	14.149501000	1.696809000
N	3.970792000	8.979333000	1.034637000	C	5.850246000	4.713016000	1.859548000
N	4.572156000	9.395279000	4.291862000	H	6.799384000	4.289116000	2.190498000
N	2.864989000	11.364702000	1.825650000	C	4.205357000	13.351772000	3.510214000
C	2.156543000	12.550039000	2.180754000	H	4.505959000	12.325448000	3.268143000
C	4.541387000	7.941822000	0.360825000	C	6.580907000	9.194993000	6.235945000
C	4.592092000	6.562917000	0.909026000	H	7.361888000	9.117112000	6.992117000
C	3.467350000	11.305007000	0.659066000	C	2.076943000	6.315984000	0.504170000
H	3.491718000	12.179105000	-0.001128000	H	2.191425000	6.943247000	-0.386996000
C	2.778564000	13.521404000	3.007249000	H	1.403440000	5.487185000	0.256776000
C	4.072375000	10.095673000	0.223621000	H	1.577995000	6.923334000	1.270362000
C	3.402998000	5.792746000	0.992869000	C	3.471415000	4.498936000	1.531474000
C	0.823740000	12.726196000	1.720974000	H	2.559325000	3.904717000	1.598263000
C	4.278037000	9.037879000	5.559385000	C	4.315463000	13.503330000	5.034214000
H	3.227134000	8.832250000	5.758221000	H	5.334498000	13.265792000	5.368104000
C	0.112176000	11.695187000	0.852198000	H	4.096000000	14.525675000	5.367544000
H	0.748329000	10.802731000	0.812204000	H	3.622714000	12.828286000	5.551094000
C	2.056690000	14.678272000	3.333672000	C	7.131781000	6.758796000	1.185201000
H	2.524874000	15.442364000	3.953447000	H	7.005666000	7.839081000	1.304272000
C	4.723889000	9.759954000	-0.978258000	H	7.864669000	6.410266000	1.922850000
H	4.935277000	10.429808000	-1.806983000	H	7.573175000	6.600720000	0.190595000
C	5.247395000	8.929848000	6.555092000	C	6.891080000	9.559949000	4.923769000
H	4.952087000	8.638402000	7.561713000	H	7.914603000	9.774923000	4.620735000
C	0.756312000	14.876713000	2.875026000	C	-1.230269000	11.249798000	1.451558000
H	0.213622000	15.784851000	3.138640000	H	-1.103049000	10.882933000	2.475751000
C	5.023401000	8.395411000	-0.887938000	H	-1.968868000	12.061594000	1.466284000
H	5.514123000	7.780195000	-1.635411000	H	-1.655234000	10.433500000	0.852581000
C	5.830503000	6.009410000	1.323710000	C	-0.062436000	12.195344000	-0.591043000
C	0.152804000	13.904283000	2.081233000	H	-0.528458000	11.418911000	-1.212564000
H	-0.866199000	14.062665000	1.729971000	H	-0.703252000	13.085912000	-0.634869000
C	5.861570000	9.647968000	3.989191000	H	0.898899000	12.458971000	-1.048986000
H	6.062325000	9.928141000	2.956602000				
C	4.680451000	3.963069000	1.972218000				
H	4.712880000	2.957362000	2.392564000				
C	5.177307000	14.295220000	2.783494000				

### **1b ( $S = 2$ )**

Fe	3.341023000	9.453877000	2.978572000
Cl	1.741551000	8.644358000	4.328396000

N	4.176488000	9.018595000	1.186382000	C	7.092476000	10.277140000	6.144165000
N	4.982718000	9.827182000	4.351808000	H	7.911899000	10.452945000	6.840816000
N	2.990339000	11.366362000	2.003103000	C	2.285034000	6.395926000	0.014200000
C	2.203994000	12.502965000	2.352061000	H	2.441278000	7.477739000	-0.074210000
C	4.732104000	7.999763000	0.468052000	C	3.506731000	4.464649000	1.081741000
C	4.701707000	6.573972000	0.899420000	H	2.594174000	3.890294000	0.928797000
C	3.621781000	11.345324000	0.851799000	C	4.573889000	13.999852000	4.735162000
H	3.612074000	12.220673000	0.192610000	H	5.652983000	13.954528000	4.934139000
C	2.798679000	13.640766000	2.956959000	H	4.227346000	14.994384000	5.043434000
C	4.281402000	10.164614000	0.413142000	H	4.078437000	13.261653000	5.376861000
C	3.517062000	5.812876000	0.701219000	C	7.175953000	6.690565000	1.608271000
C	0.805896000	12.457219000	2.099465000	H	6.975318000	7.752942000	1.415282000
C	4.808293000	9.680311000	5.682293000	C	7.279633000	10.425679000	4.767992000
H	3.810136000	9.376670000	5.994649000	H	8.243733000	10.716987000	4.354141000
C	0.149549000	11.259416000	1.424541000	C	-1.106384000	10.768541000	2.157201000
H	0.873353000	10.434970000	1.453777000	H	-0.893846000	10.565750000	3.212504000
C	1.972882000	14.727847000	3.281450000	H	-1.931874000	11.490030000	2.103309000
H	2.414278000	15.613022000	3.738486000	H	-1.462567000	9.834682000	1.702964000
C	4.936568000	9.871178000	-0.796837000	C	-0.146021000	11.551215000	-0.056143000
H	5.151143000	10.570168000	-1.600336000	H	-0.568365000	10.663834000	-0.546744000
C	5.830341000	9.895738000	6.605022000	H	-0.868680000	12.370689000	-0.165585000
H	5.630208000	9.762701000	7.666995000	H	0.761491000	11.836002000	-0.602357000
C	0.603560000	14.703098000	3.031062000	C	8.214213000	6.234856000	0.570138000
H	-0.017894000	15.560381000	3.290974000	H	9.142857000	6.813652000	0.669705000
C	5.223140000	8.501485000	-0.758717000	H	8.467566000	5.173939000	0.694942000
H	5.703270000	7.910619000	-1.531878000	H	7.841904000	6.368644000	-0.452319000
C	5.855290000	5.947791000	1.435799000	C	7.739129000	6.579882000	3.032596000
C	0.033178000	13.574040000	2.446557000	H	6.997293000	6.880998000	3.781365000
H	-1.037912000	13.561542000	2.249425000	H	8.055928000	5.557080000	3.272770000
C	6.204792000	10.191359000	3.912616000	H	8.620987000	7.224872000	3.146297000
H	6.314213000	10.292088000	2.834205000	C	2.150312000	5.846310000	-1.415653000
C	4.624706000	3.830435000	1.637173000	H	3.052776000	6.046714000	-2.005907000
C	4.994428000	14.752744000	2.351084000	H	1.984243000	4.760979000	-1.417366000
H	6.077343000	14.753524000	2.537000000	H	1.299997000	6.315206000	-1.929441000
H	4.835606000	14.534443000	1.287882000	C	0.990443000	6.195754000	0.814554000
H	4.626128000	15.770172000	2.535925000	H	0.699981000	5.139205000	0.880557000
C	5.786492000	4.592739000	1.791855000	H	1.085562000	6.584358000	1.834324000
H	6.672015000	4.109358000	2.205955000	H	0.161631000	6.726234000	0.326381000
C	4.290463000	13.723983000	3.250590000	C	4.597795000	2.366684000	2.053560000
H	4.719661000	12.742814000	3.019101000	H	5.601723000	2.132504000	2.441828000

C	3.603621000	2.111566000	3.196229000	C	10.114030000	-1.147672000	12.204673000
H	2.569124000	2.303704000	2.884226000	H	9.253912000	-0.716321000	11.694886000
H	3.659872000	1.067804000	3.533764000	C	5.960005000	3.944842000	14.472459000
H	3.815172000	2.758873000	4.056359000	H	6.166911000	3.983720000	13.404003000
C	4.348215000	1.427279000	0.864364000	C	12.158407000	3.396100000	12.814117000
H	5.083224000	1.593339000	0.066847000	H	11.446330000	2.623823000	13.130338000
H	4.420203000	0.377257000	1.178639000	C	12.330798000	-2.224397000	13.514493000
H	3.349844000	1.573731000	0.432681000	H	13.190040000	-2.657510000	14.027642000
				C	12.389898000	5.891058000	13.166951000

### 1c ( $S = 2$ )

Fe	8.573410000	2.325109000	14.275878000	C	12.112489000	7.082482000	13.833557000
Cl	8.057435000	2.481468000	12.094188000	H	12.594947000	8.010265000	13.524917000
N	9.290436000	0.938489000	15.567984000	C	12.030396000	3.516741000	11.289476000
N	10.274061000	3.464038000	15.028041000	H	11.026610000	3.849076000	11.003012000
C	10.811318000	2.766631000	16.001785000	H	12.204657000	2.538943000	10.822674000
H	11.679393000	3.149816000	16.549600000	H	12.760452000	4.215753000	10.861444000
N	6.851068000	3.252704000	15.213863000	C	10.785552000	0.520653000	17.248047000
C	8.186749000	-1.282878000	15.291996000	H	11.589525000	0.655290000	17.965855000
C	9.124856000	-0.350637000	15.977728000	C	8.485996000	6.978062000	16.270925000
C	10.882277000	4.693693000	14.637209000	H	8.861813000	8.008877000	16.255720000
C	6.630506000	3.157331000	16.540751000	H	7.745429000	6.914129000	17.079259000
H	7.355435000	2.575374000	17.106895000	H	7.968805000	6.798195000	15.321110000
C	10.030195000	-0.635373000	17.022595000	C	12.488347000	-1.636285000	12.255774000
H	10.140000000	-1.590551000	17.524795000	H	13.471005000	-1.603013000	11.784329000
C	8.580481000	-1.877606000	14.062369000	C	5.538984000	3.752476000	17.171088000
C	10.579789000	5.900973000	15.319894000	H	5.413444000	3.637613000	18.246079000
C	11.214128000	7.079228000	14.897318000	C	4.557177000	0.050711000	18.323964000
H	11.002673000	8.013244000	15.417058000	H	3.614391000	0.598719000	18.313858000
C	6.377977000	-2.930134000	13.813244000	C	7.674755000	-2.674940000	13.347040000
C	11.788848000	4.681573000	13.543044000	H	7.997775000	-3.098926000	12.397415000
C	6.353316000	-1.056524000	17.091663000	C	4.083852000	-3.409672000	12.901260000
C	6.886063000	-1.561623000	15.792371000	H	3.734776000	-2.476831000	13.343502000
C	9.610445000	5.955742000	16.493690000	C	10.337362000	6.214008000	17.823368000
H	9.137844000	4.969896000	16.568134000	H	11.106658000	5.456396000	18.016313000
C	9.945039000	-1.730507000	13.473310000	H	9.628459000	6.195470000	18.662567000
C	10.314660000	1.475666000	16.330291000	H	10.832642000	7.193686000	17.828915000
C	6.013541000	-2.371701000	15.045871000	C	11.375097000	-1.097202000	11.604455000
H	5.036921000	-2.604493000	15.467857000	H	11.483577000	-0.634830000	10.623025000
C	11.071673000	-2.272130000	14.117053000	C	4.625148000	4.475156000	16.401590000
H	10.955028000	-2.752244000	15.086603000	H	3.761657000	4.953533000	16.863707000

C	13.559966000	2.913203000	13.222219000	H	7.488327000	2.392189000	0.370140000
H	14.335039000	3.630340000	12.920401000	C	6.491332000	8.138033000	0.444813000
H	13.786837000	1.949814000	12.747127000	C	1.859369000	2.508982000	1.159716000
H	13.639728000	2.777364000	14.307767000	H	2.593010000	2.287135000	0.386034000
C	5.866639000	-4.970148000	12.430628000	C	1.205373000	2.557197000	3.473393000
H	6.901975000	-5.289053000	12.548508000	H	1.405629000	2.378380000	4.526577000
C	6.381149000	-0.970020000	19.529901000	C	2.164189000	2.217513000	2.505101000
H	6.867092000	-1.232458000	20.470107000	C	7.049060000	7.874473000	2.839557000
C	4.842346000	4.567213000	15.024610000	C	4.400156000	-0.907169000	2.840110000
H	4.158693000	5.113996000	14.377052000	H	5.213871000	-0.538911000	3.462544000
C	5.177520000	-0.258858000	19.537846000	C	7.660843000	9.135014000	2.808495000
H	4.722522000	0.041895000	20.481987000	H	8.125680000	9.526390000	3.712552000
C	5.139877000	-0.342267000	17.116319000	C	5.170629000	2.110856000	6.272968000
H	4.653535000	-0.090450000	16.173881000	H	6.033447000	2.649124000	6.667982000
C	5.435440000	-3.778019000	13.043304000	C	2.348009000	-0.545429000	1.642871000
C	6.958870000	-1.368897000	18.322063000	H	1.536415000	0.097124000	1.311678000
H	7.872819000	-1.956627000	18.334824000	C	6.466115000	7.375291000	1.643131000
C	4.978652000	-5.763999000	11.702021000	C	3.397538000	-0.015273000	2.407437000
H	5.333359000	-6.687334000	11.243167000	C	6.973794000	7.232336000	-1.869827000
C	3.639626000	-5.384431000	11.568729000	H	7.593900000	8.089753000	-2.163794000
H	2.947010000	-6.003691000	10.998514000	H	6.516857000	6.823671000	-2.781143000
C	3.196508000	-4.203400000	12.171784000	H	7.645256000	6.467324000	-1.461185000
H	2.156816000	-3.892247000	12.065986000	C	7.120946000	9.392391000	0.477436000
				H	7.158527000	9.988951000	-0.433295000
				C	7.075115000	7.046791000	4.118792000
<b>1d (S = 2)</b>				H	6.327668000	6.252388000	3.997600000
Fe	3.851317000	5.323730000	2.110273000	C	4.370693000	-2.259313000	2.496062000
Cl	2.534339000	6.252710000	0.525599000	H	5.166379000	-2.918530000	2.844888000
C	3.806101000	1.482026000	4.344982000	C	-0.319974000	3.374127000	1.771401000
C	6.529037000	5.051103000	1.249736000	H	-1.275293000	3.816555000	1.489255000
H	7.532633000	5.174084000	0.826895000	C	5.885780000	7.639709000	-0.862157000
C	3.490533000	1.485251000	2.823744000	H	5.293238000	6.747073000	-0.632452000
C	5.684159000	1.520529000	1.350441000	C	3.284290000	0.648950000	6.587007000
H	5.790462000	0.456288000	1.183316000	H	2.652957000	0.036656000	7.231892000
C	6.010909000	3.733967000	1.363921000	C	-0.020643000	3.127704000	3.111653000
C	4.628104000	2.155741000	2.041242000	H	-0.742378000	3.378416000	3.890037000
C	4.894155000	2.173467000	4.900613000	C	4.363269000	1.357311000	7.126240000
H	5.546492000	2.756951000	4.255675000	H	4.580794000	1.305639000	8.193352000
C	3.019757000	0.701233000	5.217201000	C	0.632782000	3.067863000	0.796334000
H	2.200986000	0.107831000	4.812950000	H	0.433110000	3.282542000	-0.252756000
C	6.564644000	2.518499000	0.927389000				

C	3.324006000	-2.767846000	1.720722000	C	4.742294000	6.584603000	0.795785000
H	3.294632000	-3.824943000	1.455652000	C	3.610060000	11.343601000	0.734235000
C	7.700913000	9.893240000	1.639920000	H	3.617389000	12.229656000	0.089515000
H	8.185097000	10.870219000	1.635280000	C	3.019249000	13.669603000	2.802875000
C	2.312076000	-1.903410000	1.302978000	C	4.234793000	10.155328000	0.273854000
H	1.478423000	-2.281078000	0.709961000	C	3.583667000	5.771008000	0.668190000
C	6.687558000	7.848898000	5.368755000	C	0.889173000	12.561880000	2.217520000
H	7.446396000	8.595164000	5.636376000	C	3.977841000	9.602215000	5.690817000
H	6.581243000	7.177547000	6.231317000	H	2.907835000	9.395464000	5.724332000
H	5.735680000	8.374483000	5.229688000	C	0.104900000	11.380786000	1.659735000
C	8.438991000	6.362136000	4.309782000	H	0.801668000	10.535472000	1.586064000
H	8.697354000	5.729429000	3.452596000	C	2.284801000	14.804602000	3.180000000
H	8.433324000	5.726799000	5.206369000	H	2.814916000	15.680583000	3.553303000
H	9.241470000	7.102078000	4.428484000	C	4.857472000	9.859882000	-0.954576000
C	4.916772000	8.654681000	-1.485791000	H	5.047014000	10.556107000	-1.766765000
H	4.139783000	8.948204000	-0.771827000	C	4.733742000	9.742279000	6.850912000
H	4.416411000	8.207728000	-2.354401000	H	4.247339000	9.659622000	7.821689000
H	5.427403000	9.562261000	-1.833230000	C	0.896006000	14.841345000	3.085943000
N	4.816556000	3.509749000	2.031785000	H	0.348849000	15.737009000	3.381329000
N	5.850311000	6.089686000	1.675898000	C	5.155669000	8.494736000	-0.915372000
N	3.112365000	6.215724000	3.904997000	H	5.625103000	7.902560000	-1.694236000
C	2.502453000	7.418497000	3.818169000	C	5.940993000	6.021136000	1.303467000
C	3.150151000	5.614948000	5.112822000	C	0.211593000	13.723849000	2.611728000
C	1.923724000	8.050977000	4.915733000	H	-0.874585000	13.757636000	2.534489000
C	2.593356000	6.179867000	6.258473000	C	5.835918000	9.929281000	4.353153000
C	1.967078000	7.424204000	6.164012000	H	6.234279000	10.003415000	3.342637000
H	2.476748000	7.864072000	2.825277000	C	4.806506000	3.864270000	1.617834000
H	3.640356000	4.645106000	5.154817000	C	5.200325000	14.670636000	1.973719000
H	1.446341000	9.020567000	4.783605000	H	6.294536000	14.619915000	2.057569000
H	2.654514000	5.642688000	7.203596000	H	4.933149000	14.451725000	0.932816000
H	1.522179000	7.894072000	7.040930000	H	4.900099000	15.706384000	2.179222000
				C	5.941878000	4.676256000	1.702231000
				H	6.862517000	4.241271000	2.092906000

## 2 ( $S = 2$ )

Fe	3.179530000	9.418875000	2.811365000	C	4.536335000	13.685193000	2.948763000
H	1.869852000	8.726472000	3.508130000	H	4.898647000	12.682149000	2.694450000
N	4.157406000	9.014592000	1.057755000	C	6.108755000	9.975583000	6.742591000
N	4.508303000	9.698656000	4.450871000	H	6.730135000	10.080579000	7.631413000
N	2.996086000	11.367524000	1.897978000	C	2.305042000	6.285787000	0.012830000
C	2.308293000	12.545312000	2.302902000	H	2.416768000	7.369840000	-0.110937000
C	4.705189000	8.000126000	0.332060000	C	3.642583000	4.435825000	1.089055000

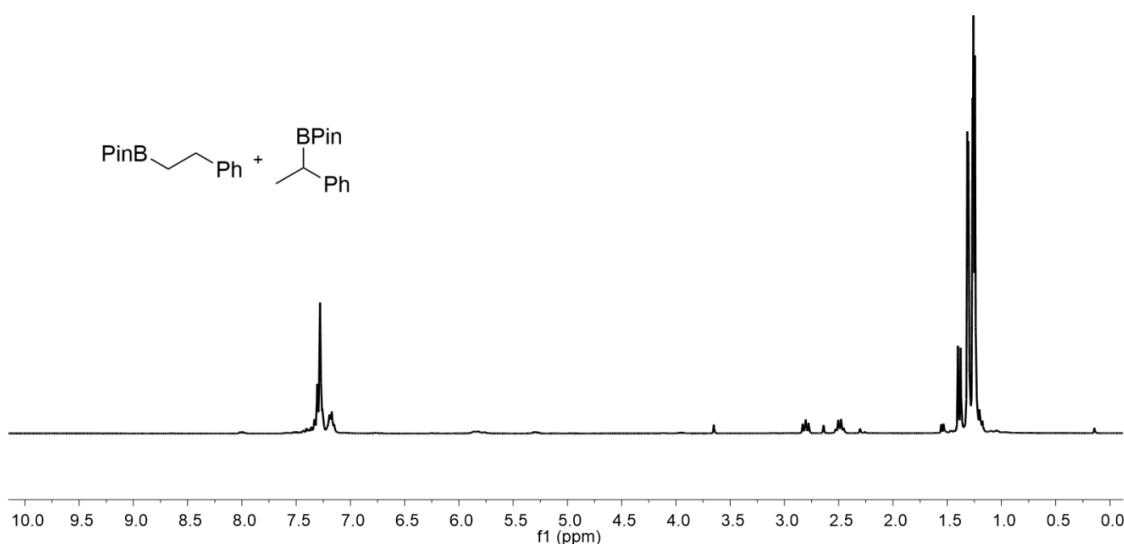
H	2.749199000	3.820865000	0.988892000	H	5.286478000	1.595570000	0.098587000
C	4.971896000	13.964150000	4.395639000	H	4.716512000	0.392022000	1.273809000
H	6.061035000	13.861127000	4.494089000	H	3.571864000	1.524622000	0.537686000
H	4.711031000	14.980704000	4.716748000				
H	4.500574000	13.263326000	5.094385000	<b>2<sub>2</sub> (S = 2)</b>			
C	7.236333000	6.821912000	1.393140000	Fe	2.290535000	-2.632547000	13.745192000
H	6.980969000	7.870260000	1.187460000	Fe	1.552129000	-0.172643000	13.884551000
C	6.665096000	10.066809000	5.464149000	N	2.288862000	1.686174000	13.714380000
H	7.729809000	10.244981000	5.320887000	N	1.472431000	-4.598048000	13.927172000
C	-1.034285000	10.929686000	2.583345000	N	3.769774000	-3.492695000	15.221690000
H	-0.660772000	10.703061000	3.588687000	N	-0.085986000	0.794735000	14.526919000
H	-1.825456000	11.685058000	2.675288000	C	2.406583000	-5.370975000	14.603104000
H	-1.498966000	10.017384000	2.186930000	C	5.733422000	1.920690000	12.744485000
C	-0.405749000	11.673149000	0.239469000	C	1.475856000	2.544339000	14.456061000
H	-0.914185000	10.792208000	-0.175364000	C	4.902392000	-3.059258000	15.974901000
H	-1.121261000	12.506281000	0.232940000	C	3.327615000	2.453130000	13.277329000
H	0.416218000	11.936544000	-0.437491000	C	7.288581000	-3.039898000	16.428559000
C	8.237765000	6.384271000	0.311900000	H	8.304179000	-3.328211000	16.158118000
H	9.142442000	7.006990000	0.345142000	C	3.504726000	-4.778114000	15.276115000
H	8.546057000	5.339574000	0.450225000	H	4.137709000	-5.443236000	15.875389000
H	7.805169000	6.472319000	-0.691423000	C	7.082110000	-2.293100000	17.584217000
C	7.879672000	6.770153000	2.786478000	H	7.925008000	-2.007209000	18.213980000
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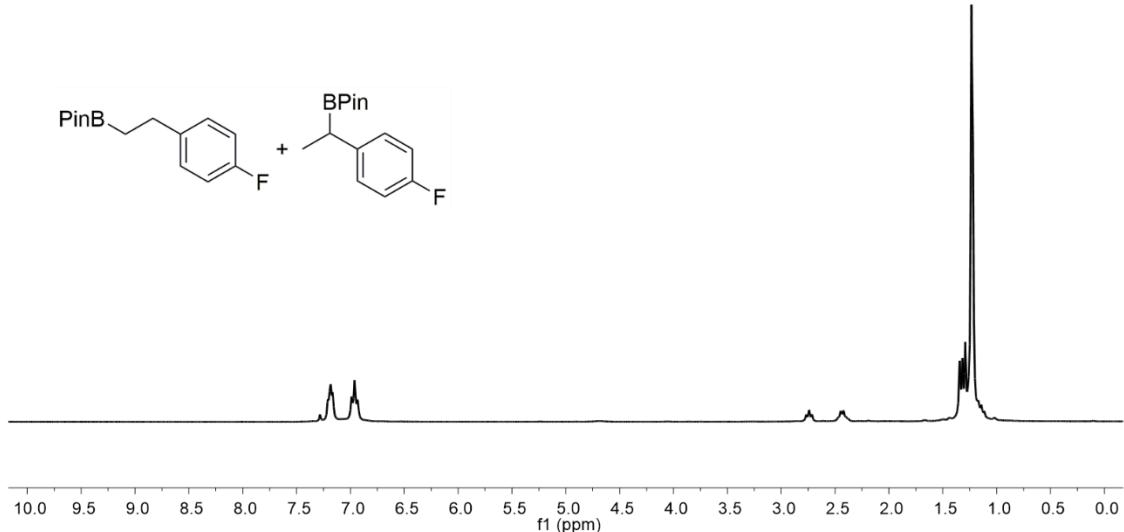
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C	4.237100	4.559800	5.607600				
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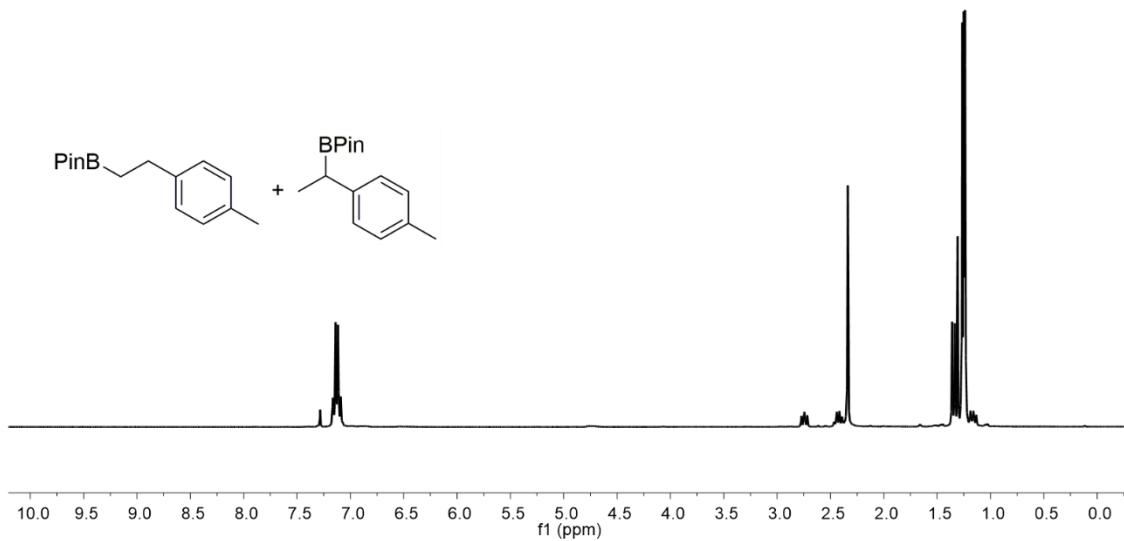
## NMR data for the catalytic hydroboration experiments



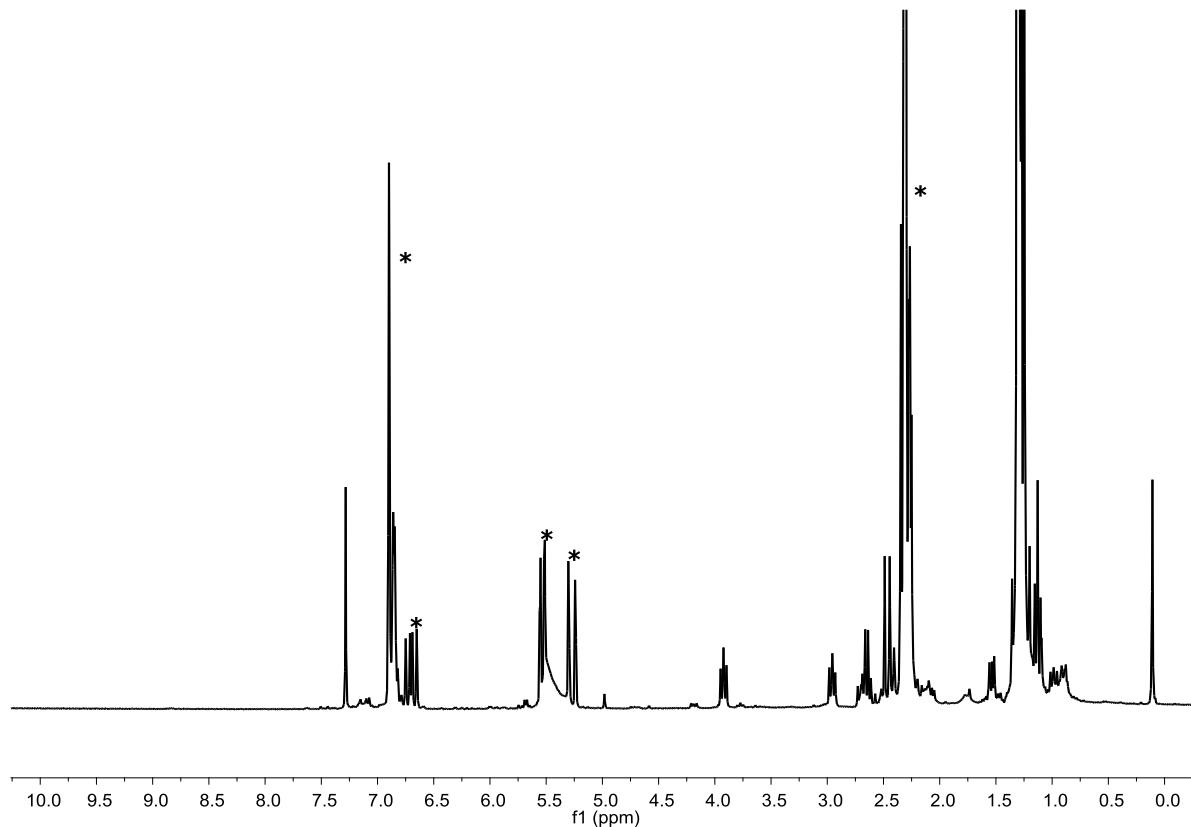
**Figure S16** <sup>1</sup>H NMR spectrum (300 MHz, CDCl<sub>3</sub>) of the hydroboration products of styrene (1 mol% of **1a**, 3 mol% of K(HBEt<sub>3</sub>)), 2 mmol of styrene, 2.5 mmol of HBPin, Reaction time: 16 h, Temperature: 25 °C.



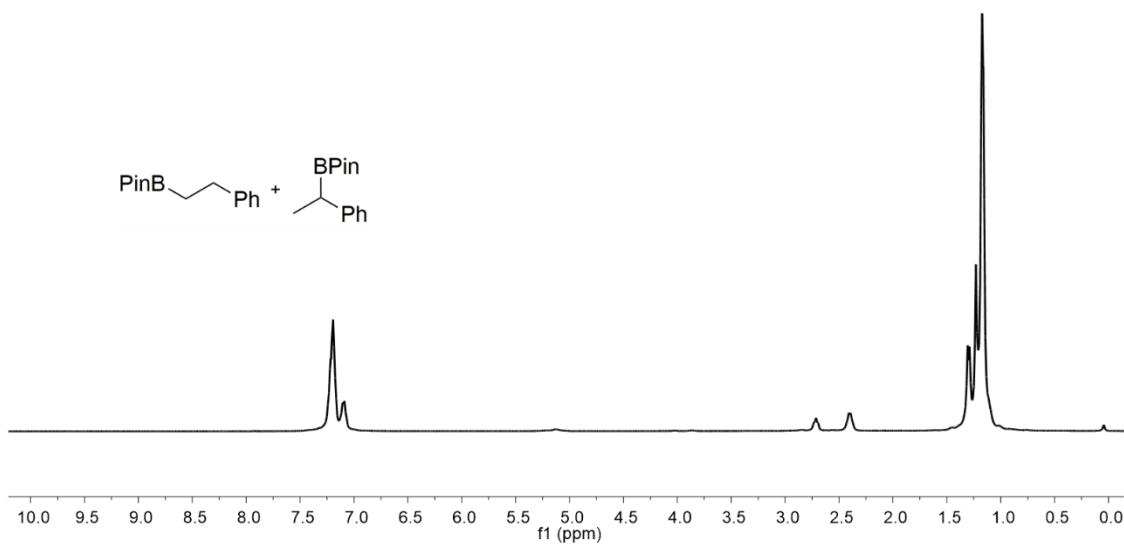
**Figure S17** <sup>1</sup>H NMR spectrum (300 MHz, CDCl<sub>3</sub>) of the hydroboration products of 1-fluoro-4-vinylbenzene (1 mol% of **1a**, 3 mol% of K(HBEt<sub>3</sub>)), 2 mmol of 4-fluorostyrene, 2.5 mmol of HBPin, Reaction time: 16 h, Temperature: 25 °C.



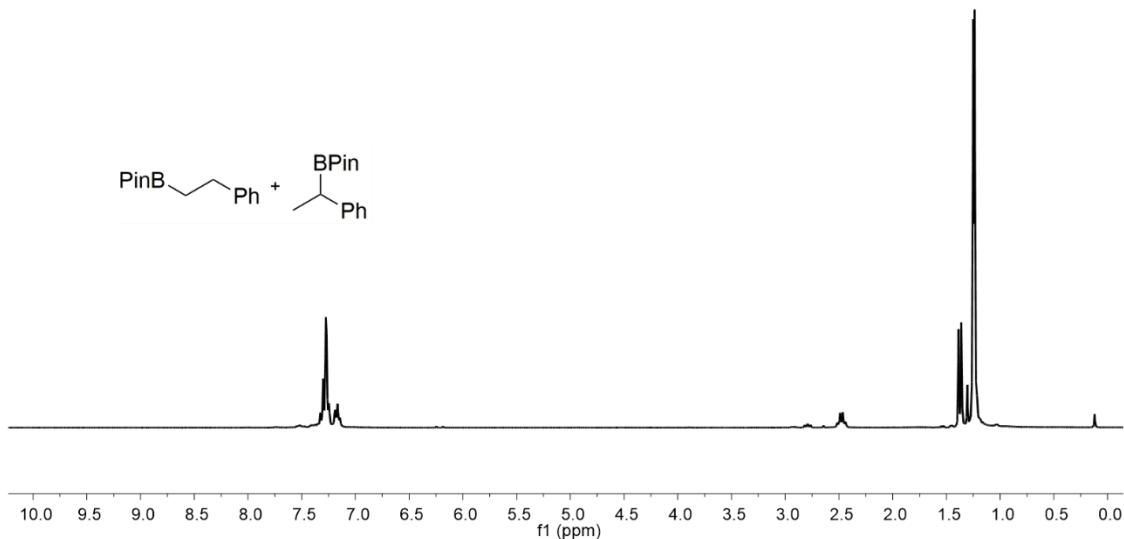
**Figure S18** <sup>1</sup>H NMR spectrum (300 MHz, CDCl<sub>3</sub>) of the hydroboration products of 1-methyl-4-vinylbenzene (1 mol% of **1a**, 3 mol% of K(HBEt<sub>3</sub>)), 2 mmol of 4-methylstyrene, 2.5 mmol of HBPin, Reaction time: 16 h, Temperature: 25 °C.



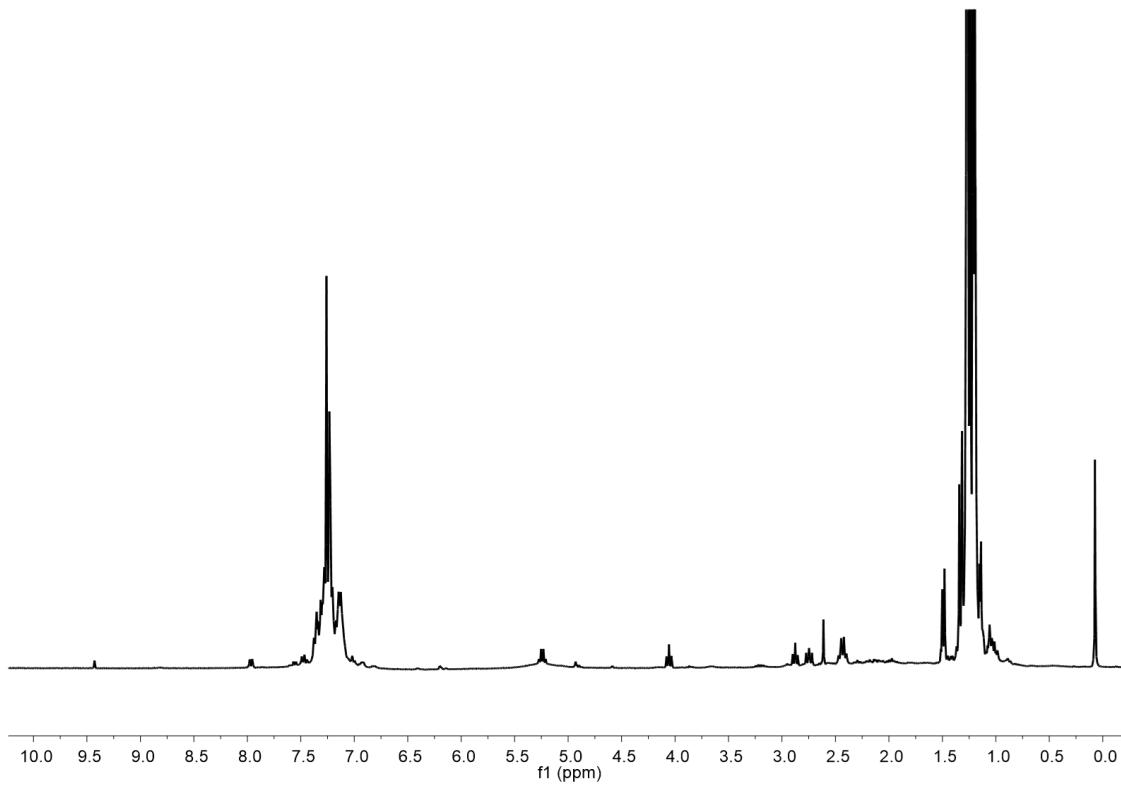
**Figure S19** <sup>1</sup>H NMR spectrum (300 MHz, CDCl<sub>3</sub>) of the isolated mixture of products formed by the catalytic reaction using complex **1a** (1 mol% of **1a**, 3 mol% of K(HBEt<sub>3</sub>)), 2 mmol of 1,3,5-trimethyl-2-vinylbenzene, 2.5 mmol of HBPin, Reaction time: 16 h, Temperature: 25 °C. Asterisks denote unreacted 1,3,5-trimethyl-2-vinylbenzene.



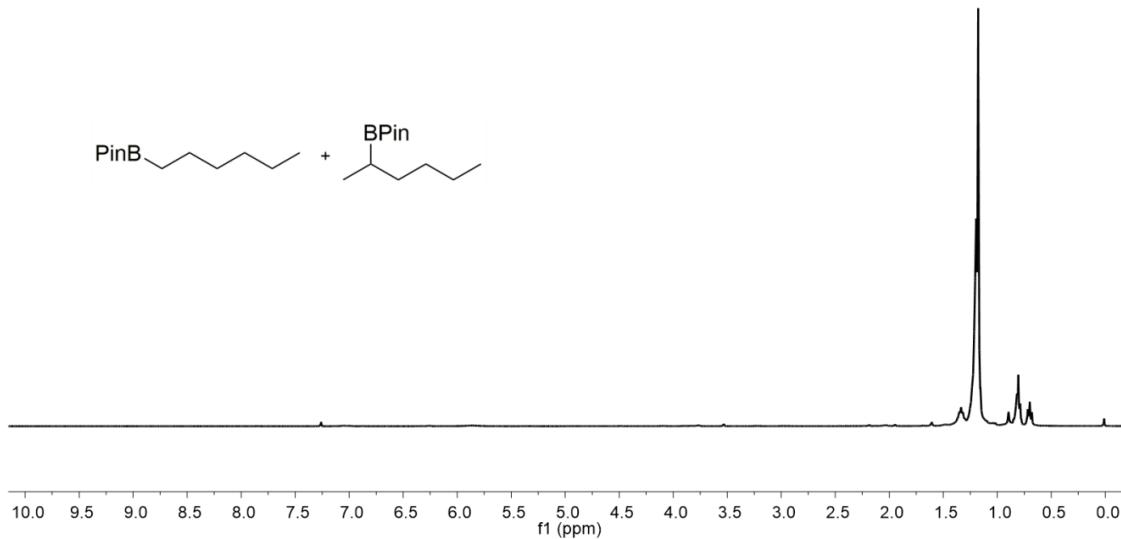
**Figure S20**  $^1\text{H}$  NMR spectrum (300 MHz,  $\text{CDCl}_3$ ) of the hydroboration products of styrene (1 mol% of **1b**, 3 mol% of  $\text{K}(\text{HBET}_3)$ ), 2 mmol of styrene, 2.5 mmol of HBPin, Reaction time: 16 h, Temperature: 25 °C.



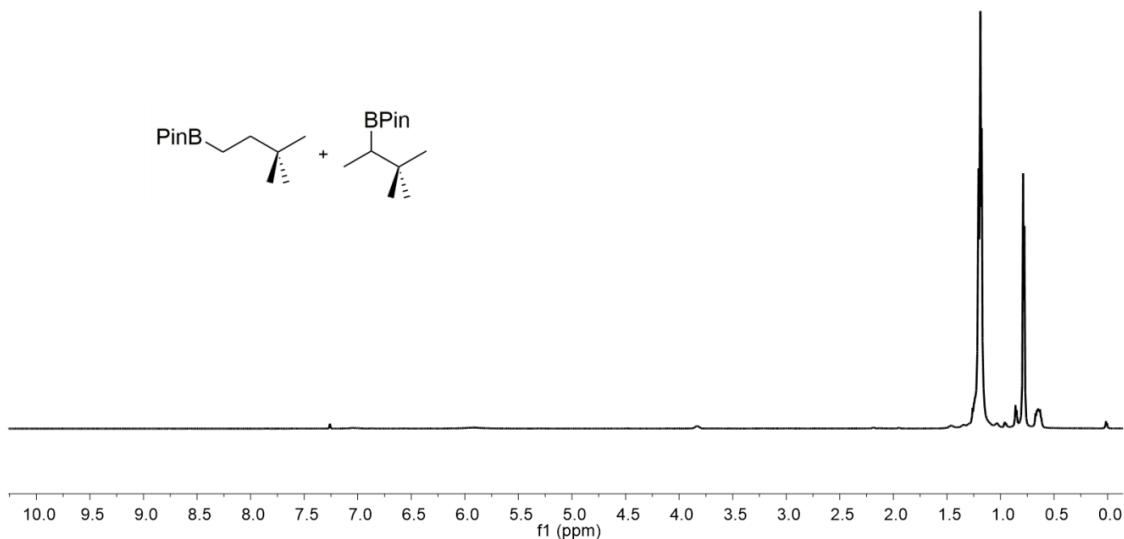
**Figure S21**  $^1\text{H}$  NMR spectrum (300 MHz,  $\text{CDCl}_3$ ) of the hydroboration products of styrene (1 mol% of **1c**, 3 mol% of  $\text{K}(\text{HBET}_3)$ ), 2 mmol of styrene, 2.5 mmol of HBPin, Reaction time: 16 h, Temperature: 25 °C.



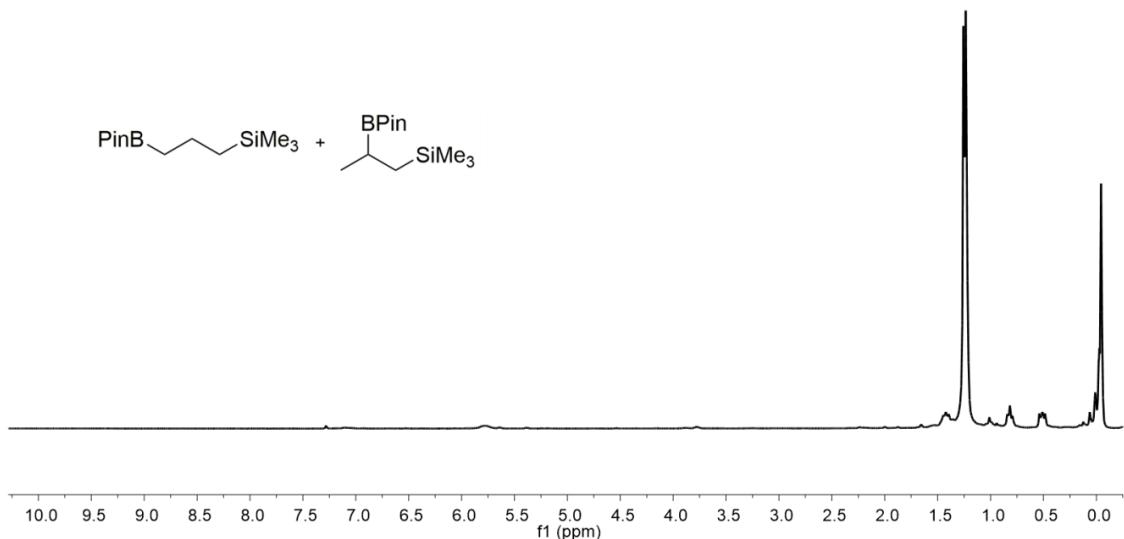
**Figure S22** <sup>1</sup>H NMR spectrum (300 MHz, CDCl<sub>3</sub>) of the isolated mixture of products formed by the catalytic reaction using complex **1d** (1 mol% of **1d**, 3 mol% of K(HBEt<sub>3</sub>)), 2 mmol of styrene, 2.5 mmol of HBPin, Reaction time: 16 h, Temperature: 25 °C).



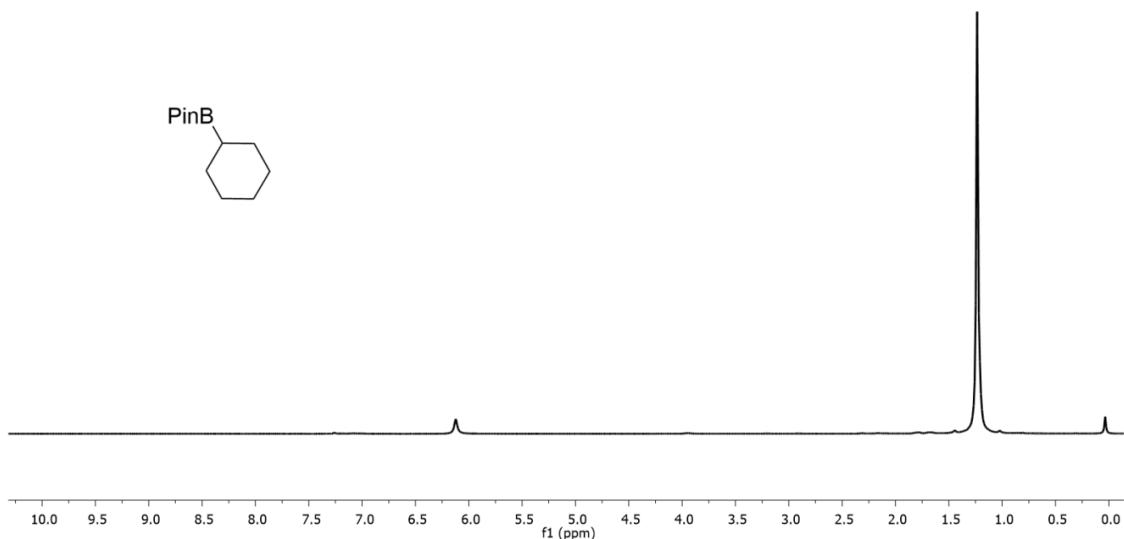
**Figure S23** <sup>1</sup>H NMR spectrum (300 MHz, CDCl<sub>3</sub>) of the hydroboration products of 1-hexene (1 mol% of **1a**, 3 mol% of K(HBEt<sub>3</sub>)), 2 mmol of 1-hexene, 2.5 mmol of HBPin, Reaction time: 16 h, Temperature: 25 °C).



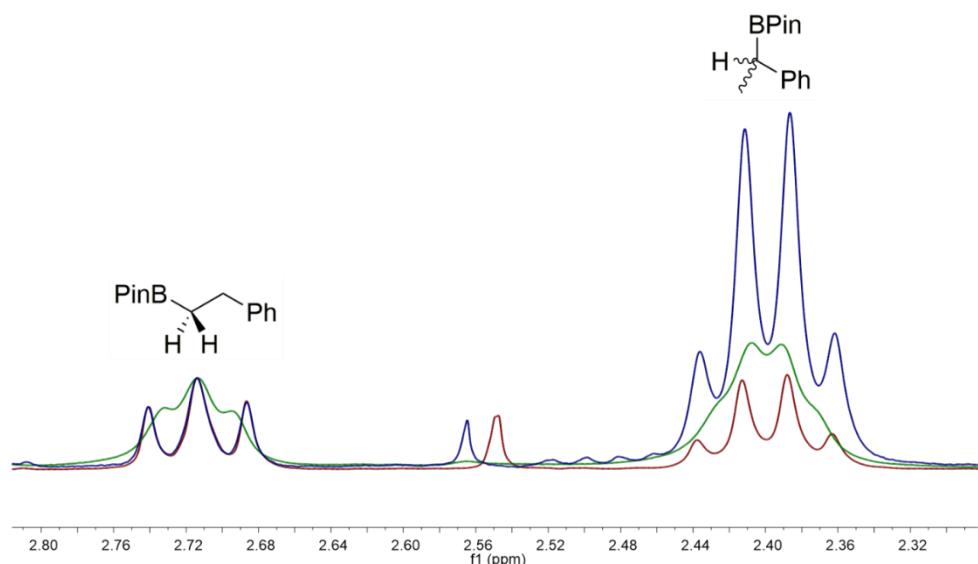
**Figure S24**  $^1\text{H}$  NMR spectrum (300 MHz,  $\text{CDCl}_3$ ) of the hydroboration products of 3,3-dimethylbut-1-ene (1 mol% of **1a**, 3 mol% of  $\text{K}(\text{HBEt}_3)$ , 2 mmol of 3,3-dimethylbut-1-ene, 2.5 mmol of  $\text{HBPin}$ , Reaction time: 16 h, Temperature: 25 °C).



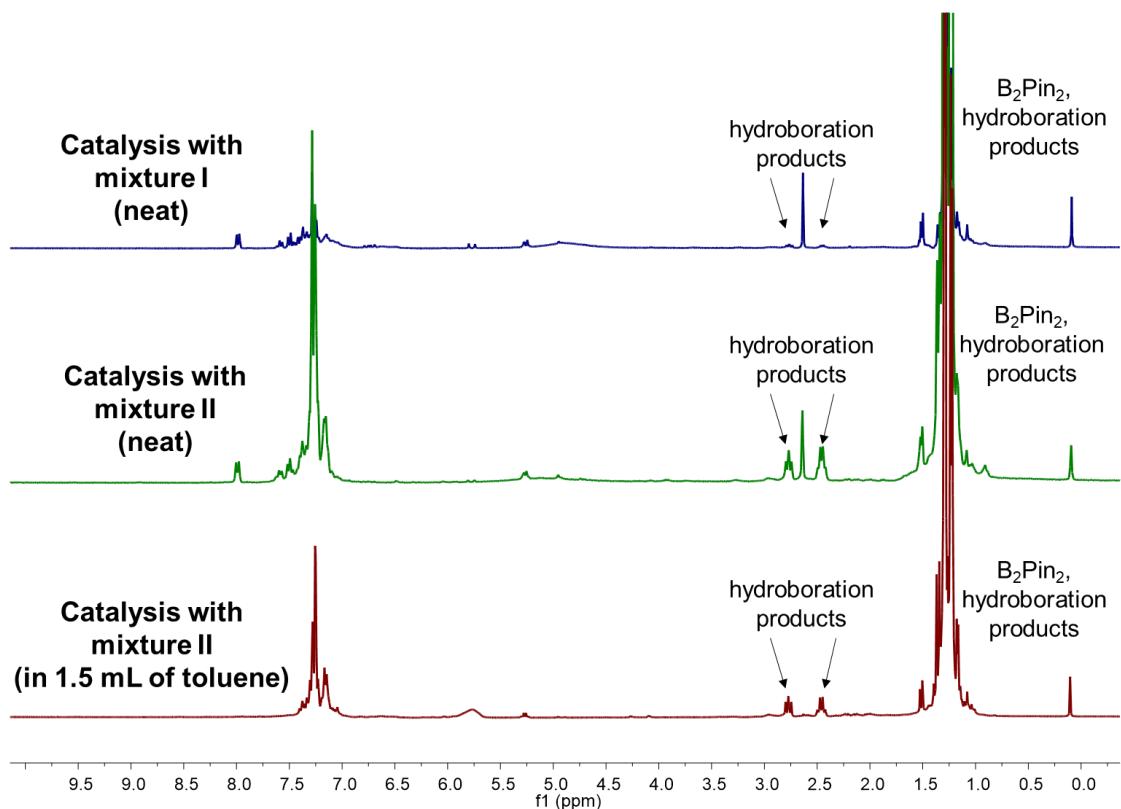
**Figure S25**  $^1\text{H}$  NMR spectrum (300 MHz,  $\text{CDCl}_3$ ) of the hydroboration products of allyltrimethylsilane (1 mol% of **1a**, 3 mol% of  $\text{K}(\text{HBEt}_3)$ ), 2 mmol of allyltrimethylsilane, 2.5 mmol of  $\text{HBPin}$ , Reaction time: 16 h, Temperature: 25 °C).



**Figure S26**  $^1\text{H}$  NMR spectrum (300 MHz,  $\text{CDCl}_3$ ) of the hydroboration product of cyclohexene (1 mol% of **1a**, 3 mol% of  $\text{K}(\text{HBET}_3)$ ), 2 mmol of cyclohexene, 2.5 mmol of  $\text{HBPin}$ , Reaction time: 16 h, Temperature: 25 °C.



**Figure S27** Detail of the superimposition of the  $^1\text{H}$  NMR spectra of the products of the hydroboration of styrene, highlighting the zone corresponding to the  $\alpha$ -protons catalyzed by different complexes activated by  $\text{K}(\text{HBET}_3)$  (red – **1a**; green – **1b**; blue – **1c**), showing the increase in the selectivity of the Mk product in the order: **1a** < **1b** < **1c**.



**Figure S28** Stacking of the <sup>1</sup>H NMR spectra (300 MHz, CDCl<sub>3</sub>) obtained from the catalytic reactions (reaction time: 16 h, temperature: 25 °C) using the mixture I (**2 + 2<sub>2</sub>**; isolated from the reaction of complex **1b** with K(HBET<sub>3</sub>) in toluene over 30 min; 1 mol%), in neat conditions (top spectrum) or in 1.5 mL of toluene (middle spectrum) or with mixture II (**2 + 2<sub>2</sub> + 3**; isolated from the reaction of complex **1b** with K(HBET<sub>3</sub>) in toluene over 16 h; 1 mol%), in neat conditions (bottom spectrum), with styrene (2 mmol) and HBPIn (2.5 mmol). Note that the spectra were normalized in relation to the resonance of the methyl protons of B<sub>2</sub>Pin<sub>2</sub>, highlighting the much higher selectivity in hydroboration products when the reaction is carried either using mixture II as catalyst in neat conditions or using mixture I with toluene as solvent.