

**MOFs-derived hierarchical hollow spheres composed of carbon-confined Ni nanoparticles for efficient CO<sub>2</sub> methanation**

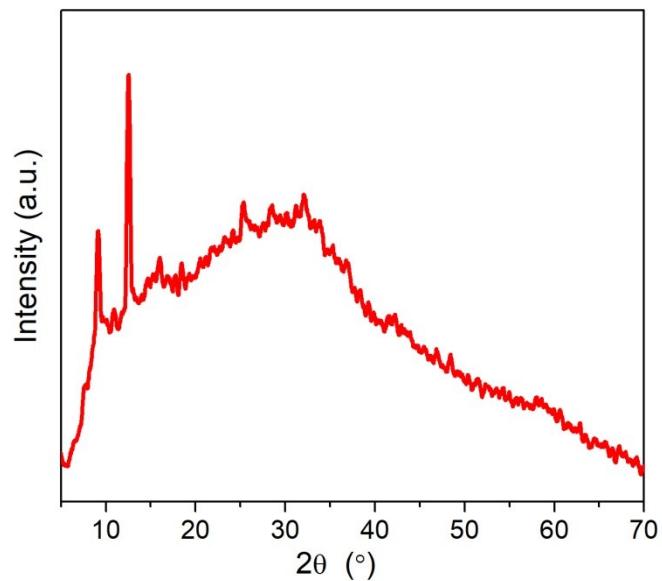
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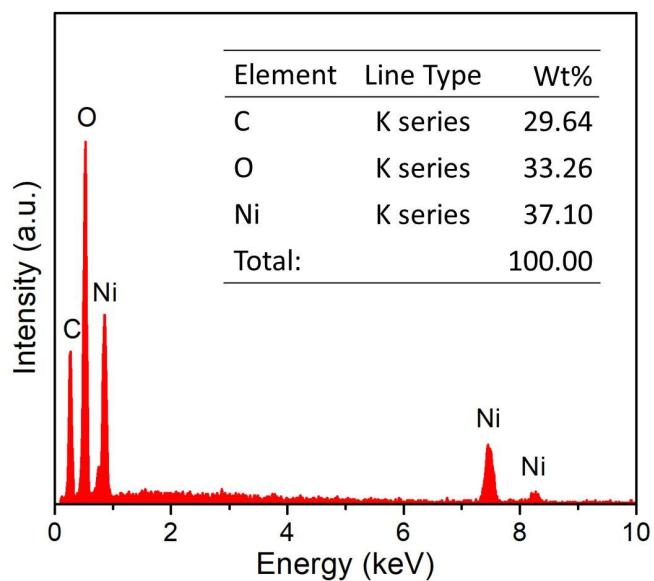
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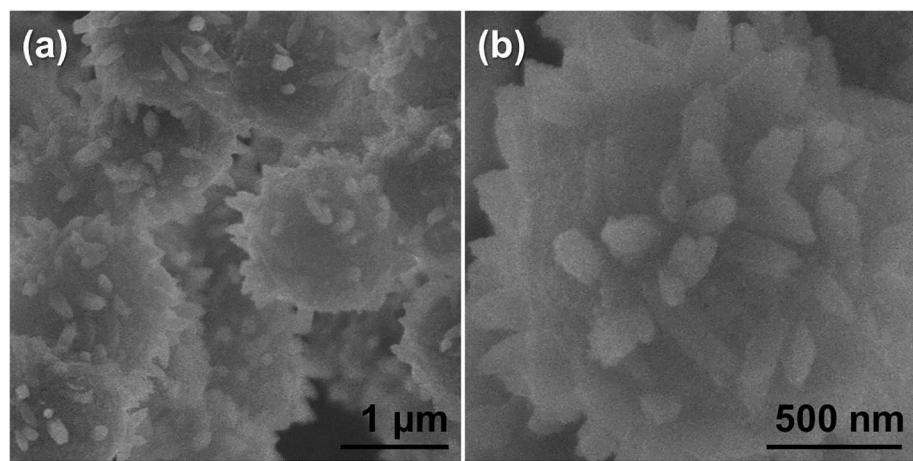
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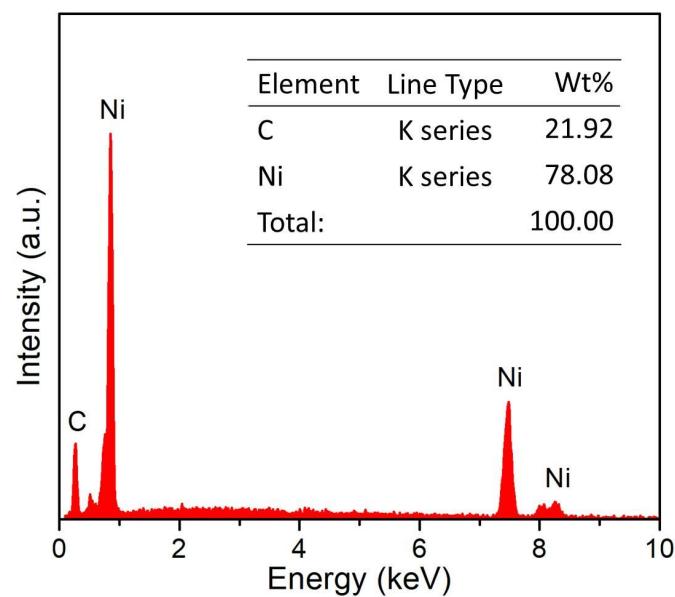
**Fig. S1** XRD pattern of Ni-MOF.



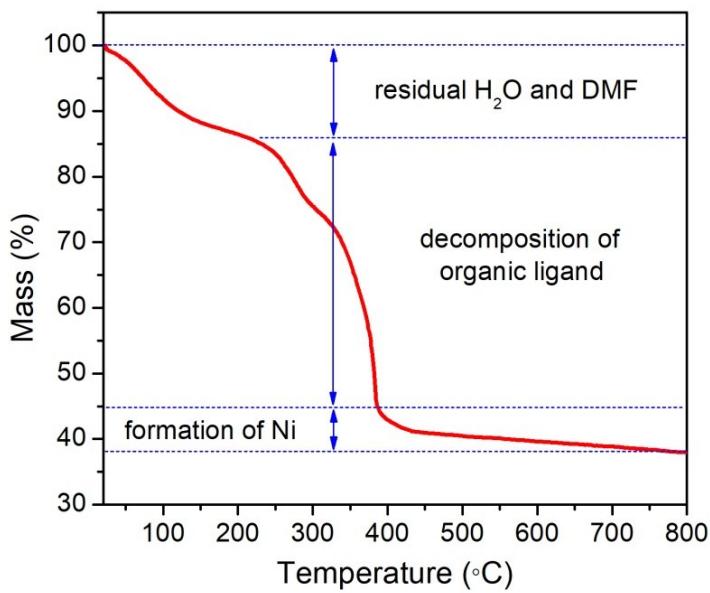
**Fig. S2** EDX spectrum of Ni-MOF. Inset shows the corresponding weight contents of the elements.



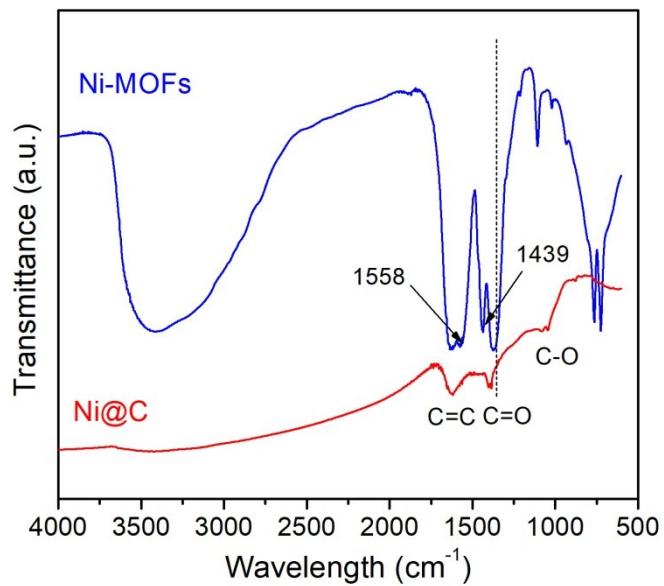
**Fig. S3** FESEM images of hierarchical Ni-MOF microspheres.



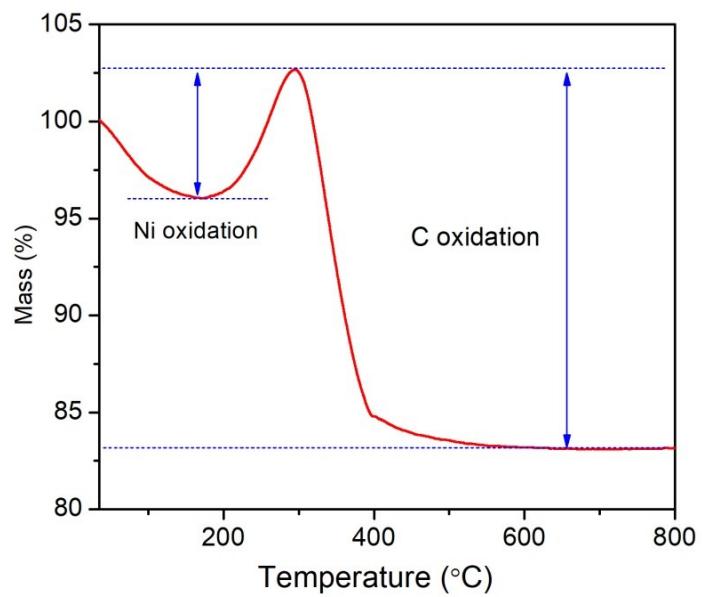
**Fig. S4** EDX spectrum of Ni@C. Inset shows the corresponding weight contents of the elements.



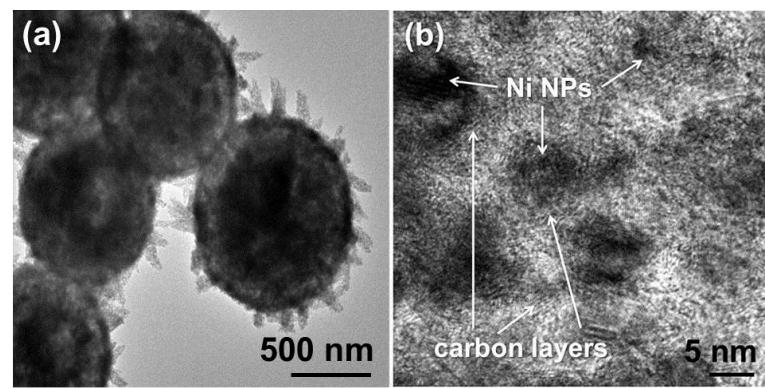
**Fig. S5** TGA results of Ni-MOF in a N<sub>2</sub> atmosphere with a heating rate of 5 °C min<sup>-1</sup>.



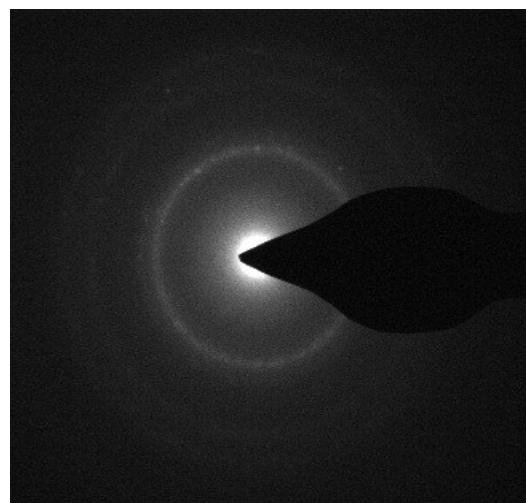
**Fig. S6** FT-IR spectra of Ni-MOF and Ni@NC samples.



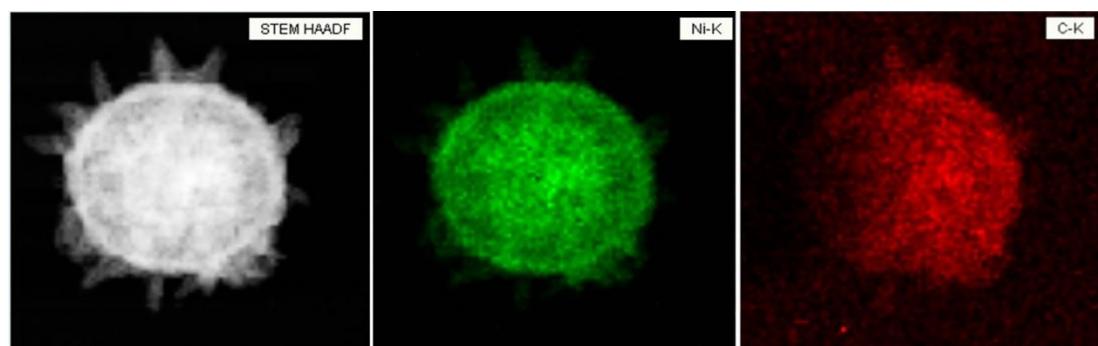
**Fig. S7** TGA results of Ni@C in an air atmosphere with a heating rate of  $5\text{ }^{\circ}\text{C min}^{-1}$ .



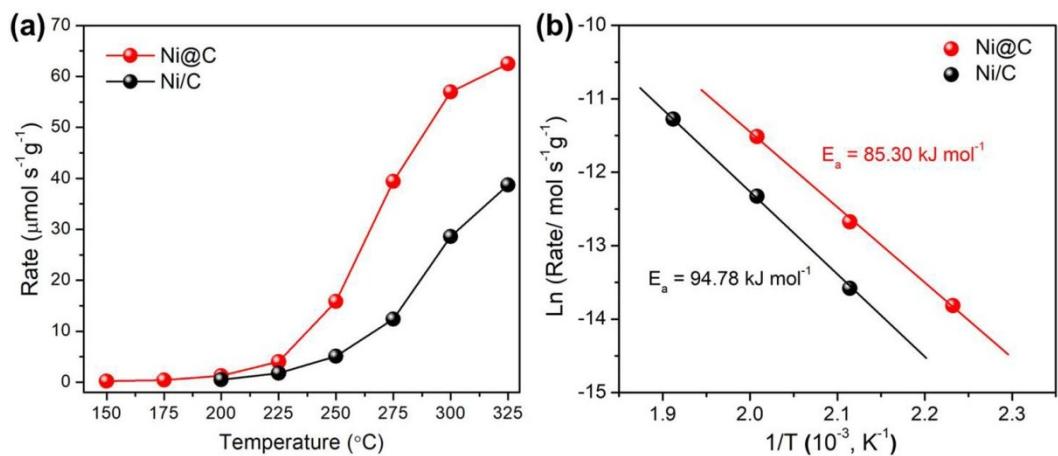
**Fig. S8** (a,b) TEM images of hierarchical Ni@C microspheres.



**Fig. S9** SAED pattern of Ni@C sample.



**Fig. S10** Elemental mappings of an individual hierarchical Ni@C microsphere.



**Fig. S11** (a) CO<sub>2</sub> methanation performance of Ni@C and Ni/C catalysts under different temperatures. (b) Arrhenius plots for CO<sub>2</sub> methanation over Ni@C and Ni/C catalysts.

**Table S1.** Comparison of CO<sub>2</sub> methanation performance of Ni@C with that of other Ni-based catalysts at atmospheric pressure.

Catalyst	Temp. (°C)	GHSV (mL g <sup>-1</sup> h <sup>-1</sup> )	Con.CO <sub>2</sub> (%)	Sel. <sub>CH4</sub> (%)	TOF (10 <sup>-3</sup> s <sup>-1</sup> )	Ref.
Ni@C	300	33000	91.16	99.9	4.28	This work
12Ni4.5Ce/CNT <sup>a</sup>	350	30000	83.8	100	/	1
Ni/MSN <sup>b</sup>	300	50000	64.1	99.9	1.61	2
OMA-2Co8Ni <sup>c</sup>	400	15000	79.9	98.3	/	3
15%Ni/ZSM-5	250	2400	27.1	99	7.56	4
15%Ni/TiO <sub>2</sub>	260	2400	96	99	1.22	5
Ni <sub>0.8</sub> Mg <sub>0.2</sub> @SiO <sub>2</sub>	300	60000	87	99	/	6
Ni(21.7)@S16C <sup>d</sup>	500	36000	33	30	12	7
Ni/CeO <sub>2</sub>	350	10000	100	100	/	8
Ni-Al <sub>2</sub> O <sub>3</sub> -HT <sup>e</sup>	350	75000	82.5	99.5	5.7	9
7.5Ni2.5Fe/Al <sub>2</sub> O <sub>3</sub>	250	32000	22.1	99	5.9	10

<sup>a</sup> CNT: multi-walled carbon nanotubes.

<sup>b</sup> MSN: mesostructured silica nanoparticles.

<sup>c</sup> OMA: ordered mesoporous Al<sub>2</sub>O<sub>3</sub>.

<sup>d</sup> S16C: the cage-type mesopores of –COOH-functionalized mesoporous silica SBA-16.

<sup>e</sup> HT: hydrotalcite.

**Table S2.** The weight contents of the elements of four Ni@C samples (denoted as Ni@C-1<sup>#</sup>, Ni@C-2<sup>#</sup>, Ni@C-3<sup>#</sup> and Ni@C-4<sup>#</sup>) prepared in different batches.

Element	Line Type	Wt%			
		Ni@C-1 <sup>#</sup>	Ni@C-2 <sup>#</sup>	Ni@C-3 <sup>#</sup>	Ni@C-4 <sup>#</sup>
C	K series	20.40	20.92	23.10	22.97
Ni	K series	79.60	79.08	76.90	77.03
Total:		100.00	100.00	100.00	100.00

### Supplementary references

- 1 W. Wang, W. Chu, N. Wang, W. Yang and C. Jiang, *Int. J. Hydrogen Energy*, 2016, **41**, 967-975.
- 2 M. A. A. Aziz, A. A. Jalil, S. Triwahyono, R. R. Mukti, Y. H. Taufiq-Yap and M. R. Sazegar, *Appl. Catal. B Environ.*, 2014, **147**, 359-368.
- 3 L. Xu, X. Lian, M. Chen, Y. Cui, F. Wang, W. Li and B. Huang, *Int. J. Hydrogen Energy*, 2018, **43**, 17172-17184.
- 4 X. Guo, A. Traitangwong, M. Hu, C. Zuo, V. Meeyoo, Z. Peng and C. Li, *Energy & Fuels*, 2018, **32**, 3681-3689.
- 5 J. Liu, C. Li, F. Wang, S. He, H. Chen, Y. Zhao, M. Wei, D. G. Evans and X. Duan, *Catal. Sci. Technol.*, 2013, **3**, 2627-2633.
- 6 Y. Li, G. Lu and J. Ma, *RSC Adv.*, 2014, **4**, 17420-17428.
- 7 C. S. Chen, C. S. Budi, H.-C. Wu, D. Saikia and H. M. Kao, *ACS Catal.*, 2017, **7**, 8367-8381.
- 8 S. Tada, T. Shimizu, H. Kameyama, T. Haneda and R. Kikuchi, *Int. J. Hydrogen Energy*, 2012, **37**, 5527-5531.
- 9 L. He, Q. Lin, Y. Liu and Y. Huang, *J. Energy Chem.*, 2014, **23**, 587-592.
- 10 D. Pandey and G. Deo, *J. Ind. Eng. Chem.*, 2016, **33**, 99-107.