

Supporting Information

Encapsulating CoS₂-CoSe₂ heterostructured nanocrystals in N-doped carbon nanocubes as highly efficient counter electrode for dye-sensitized solar cells

Shoushuang Huang, ^a Haitao Wang, ^a Shangdai Wang, ^a Zhangjun Hu, ^a Ling Zhou, ^b
Zhiwen Chen, ^{*a} Yong Jiang, ^a and Xuefeng Qian ^{*b}

[a] Dr. S. S. Huang, Dr. Z. J. Hu, Dr. Y. Jiang, Prof. Z. W. Chen

School of Environmental and Chemical Engineering, Shanghai University, Shanghai
200444, China;

[b] Prof. X. F. Qian

Shanghai Electrochemical Energy Devices Research Center, School of Chemistry and
Chemical Engineering and State Key Laboratory of Metal Matrix Composites,
Shanghai Jiao Tong University, Shanghai, 200240, P. R. China.

E-mail: xfqian@sjtu.edu.cn;

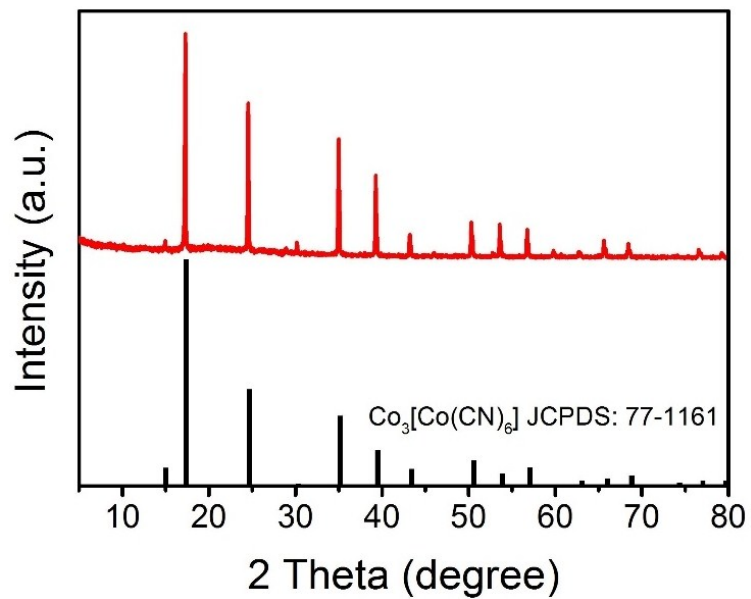


Fig. S1 XRD pattern of the as-synthesized $\text{Co}_3[\text{Co}(\text{CN})_6]_2$ PBA nanocubes.

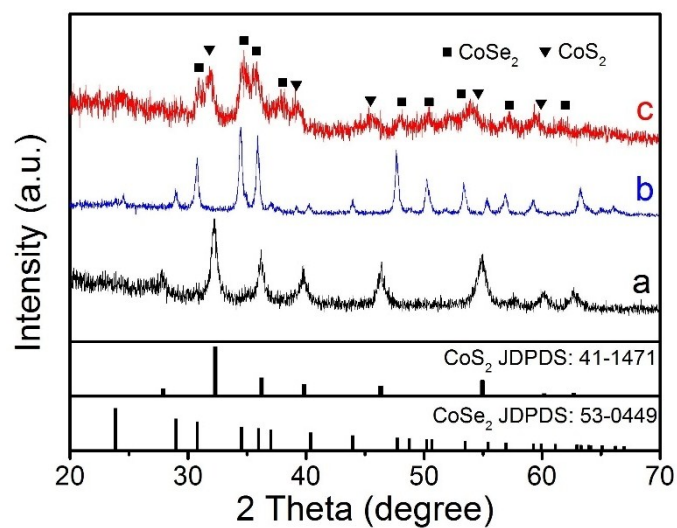


Fig. S2 XRD pattern of the as-prepared $\text{CoS}_2@NC$ (a), $\text{CoSe}_2@NC$ (b) and $\text{CoS}_2\text{-CoSe}_2@NC$ (c) nanocubes.

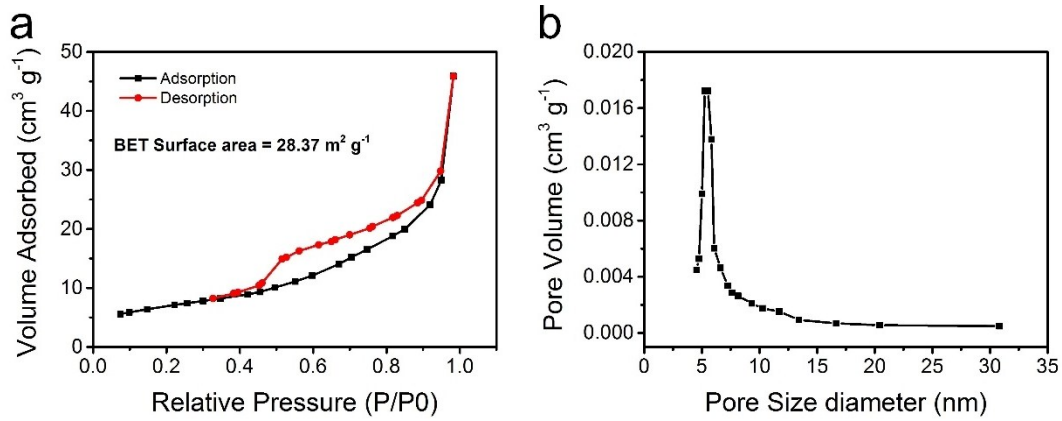


Fig. S3 Nitrogen adsorption/desorption isotherms (a) and pore size distribution of mesoporous CoS₂-CoSe₂@NC nanocubes (b).

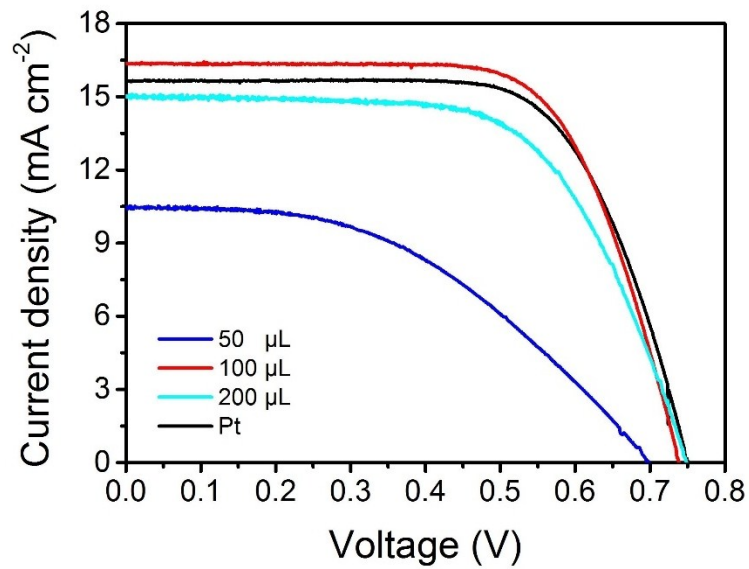


Fig. S4 J-V curve of DSSCs with different thickness of CoS₂-CoSe₂@NC and Pt CEs.

Tab. S1 Photovoltaic performance of DSSCs with different thickness of CoS₂-CoSe₂@NC and Pt CEs.

CEs	V _{oc} [mV]	J _{sc} [mA/cm ²]	FF	η [%]
50 μL	0.697 ± 0.006	10.42 ± 0.15	0.49 ± 0.02	3.56 ± 0.14
100 μL	0.740 ± 0.005	16.35 ± 0.11	0.70 ± 0.01	8.45 ± 0.06
200 μL	0.748 ± 0.013	15.03 ± 0.10	0.64 ± 0.01	7.19 ± 0.11
Pt	0.749 ± 0.003	15.53 ± 0.12	0.69 ± 0.00	8.07 ± 0.07

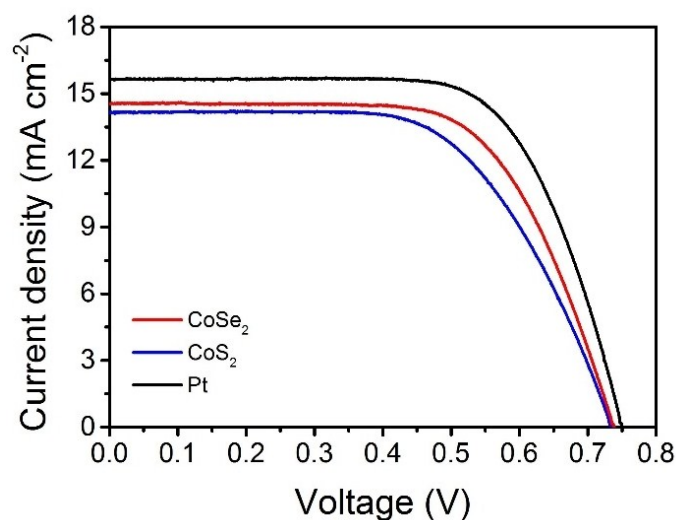


Fig. S5 J-V curves of the DSSCs with CoSe₂, CoS₂ and Pt CEs, measured under AM1.5G illumination (100 mW cm⁻²).

Tab. S2 Photovoltaic performances of DSSCs with CoS₂, CoSe₂ and Pt CEs under AM1.5G illumination.

CEs	V _{oc} [V]	J _{sc} [mA/cm ²]	FF	η [%]
CoSe ₂	0.736 ± 0.007	14.52 ± 0.15	0.65 ± 0.00	6.95 ± 0.13
CoS ₂	0.732 ± 0.008	14.17 ± 0.09	0.63 ± 0.01	6.54 ± 0.08
Pt	0.749 ± 0.003	15.53 ± 0.12	0.69 ± 0.00	8.07 ± 0.07

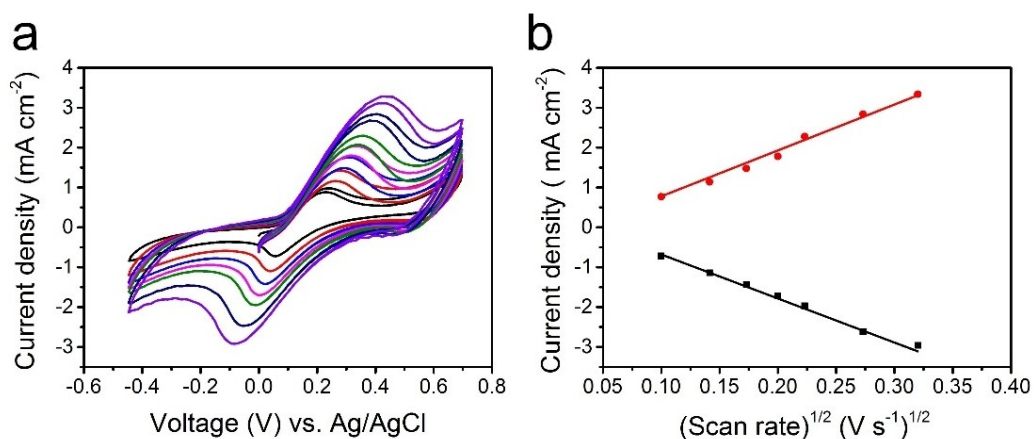


Fig. S6 (a) CV curves of the CoS₂-CoSe₂@NC nanocubes CE at different scan rate; (b) relationship between the peak current density and the square root of scanning rate of CoS₂-CoSe₂@NC nanocubes CEs.

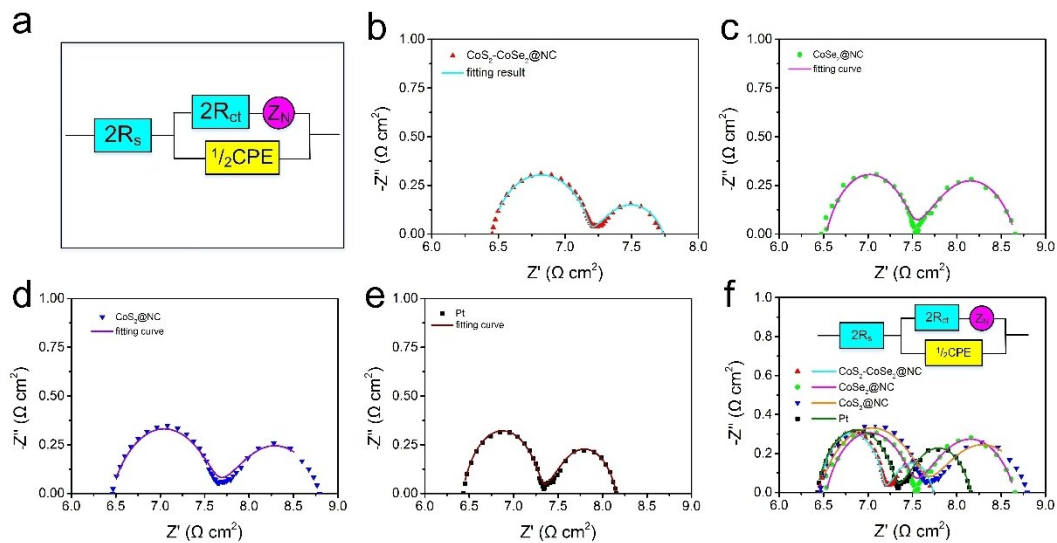


Fig. S7 The equivalent circuit diagram used to fit the impedance spectra in the symmetrical cells (a) and the fitting curves of $\text{CoS}_2\text{-CoSe}_2\text{@NC}$ (a), $\text{CoSe}_2\text{@NC}$ (b), $\text{CoS}_2\text{@NC}$ (c) and Pt (d) CEs.

Tab. S3 Fitted electrochemical parameters from EIS.

CE	R_s $\Omega \text{ cm}^2$	R_{ct} $\Omega \text{ cm}^2$	CPE-T $\mu\text{F cm}^{-2}$	CPE-P	$W_s\text{-R}$ $\Omega \text{ cm}^2$	$W_s\text{-T}$	$W_s\text{-P}$
$\text{CoS}_2\text{-CoSe}_2\text{@NC}$	3.23	0.38	269.16	0.84	0.54	0.35	0.50
$\text{CoSe}_2\text{@NC}$	3.26	0.51	220.52	0.83	1.10	0.68	0.50
$\text{CoS}_2\text{@NC}$	3.23	0.59	217.94	0.86	1.18	0.54	0.50
Pt	3.20	0.45	112.62	0.95	0.83	0.42	0.50

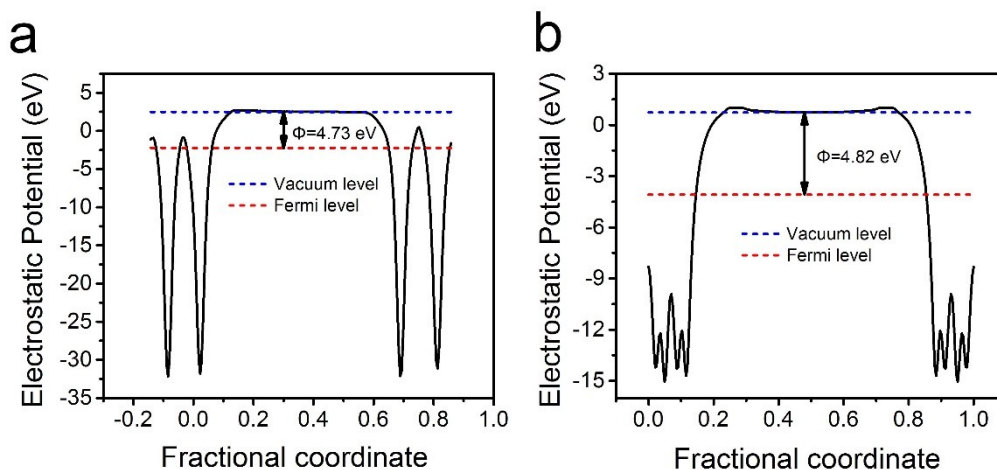


Fig. S8 Average potential profile along X-axis of (a) CoS₂ (001) and (b) CoSe₂ (100). The insert shows the calculated work function.

All of the density functional calculations were performed using the plane-wave pseudopotential method, implemented with the Cambridge Sequential Total Energy Package (CASTEP) code. The local density approximation (LDA) was used to describe the exchange-correlation effects. In order to obtain the work functions of CoS₂ and CoSe₂, two slab modules were created for the calculation. Both of the slabs consisted of a 30 Å vacuum layer and the surface cleaved for bulk material.¹⁻⁴ The atoms in the surface layer were fixed, while all other atoms were fully relaxed with force criteria of 0.025 eV/Å.

1. J. W. Wan, G. J. Fang, H. J. Yin, X. F. Liu, D. Liu, M. T. Zhao, W. J. Ke, H. Tao and Z. Y. Tang, *Adv. Mater.*, 2014, **26**, 8101-8106.
2. Z. X. Qin, Y. B. Chen, Z. X. Huang, J. Z. Su, Z. D. Diao and L. J. Guo, *J. Phys. Chem. C*, 2016, **120**, 14581-14589.
3. J. L. Zheng, W. Zhou, Y. R. Ma, W. Cao, C. B. Wang and L. Guo, *Chem. Commun.*, 2015, **51**, 12863-12866.
4. X. W. Wang, Y. Xie, B. Bateer, K. Pan, Y. T. Zhou, Y. Zhang, G. F. Wang, W. Zhou and H. G. Fu, *Nano Research*, 2016, **9**, 2862-2874.