

Electronic Supporting Information for

A Macroyclic Silver Polycarbene Complex Based on 1,2,4-Triazole Units: Synthesis and Postsynthetic Modification

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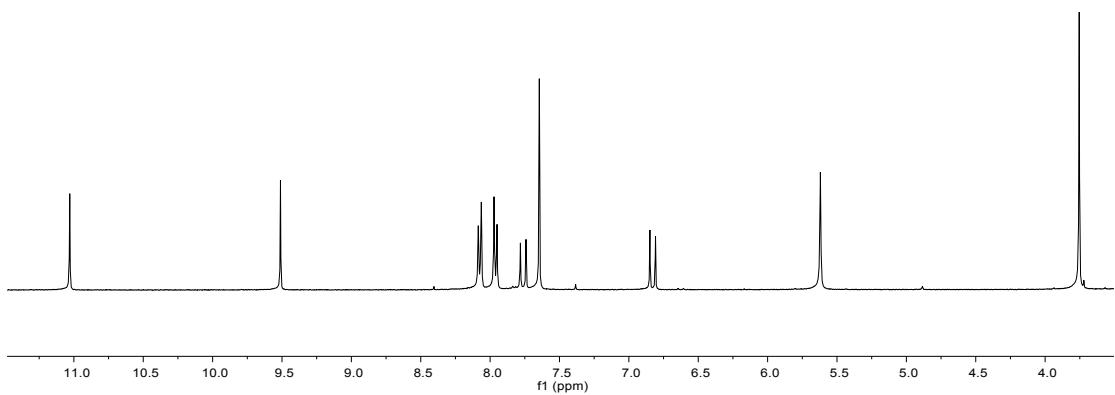


Figure S1. ¹H NMR (400 MHz, DMSO-*d*₆) of H₂-1(BF₄)₂.

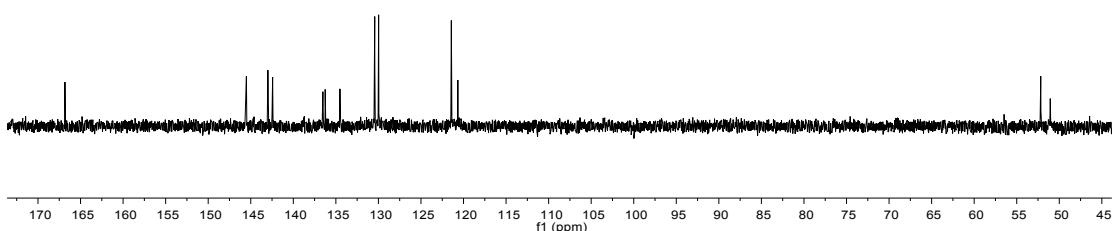


Figure S2. ¹³C NMR (100 MHz, DMSO-*d*₆) of H₂-1(BF₄)₂.

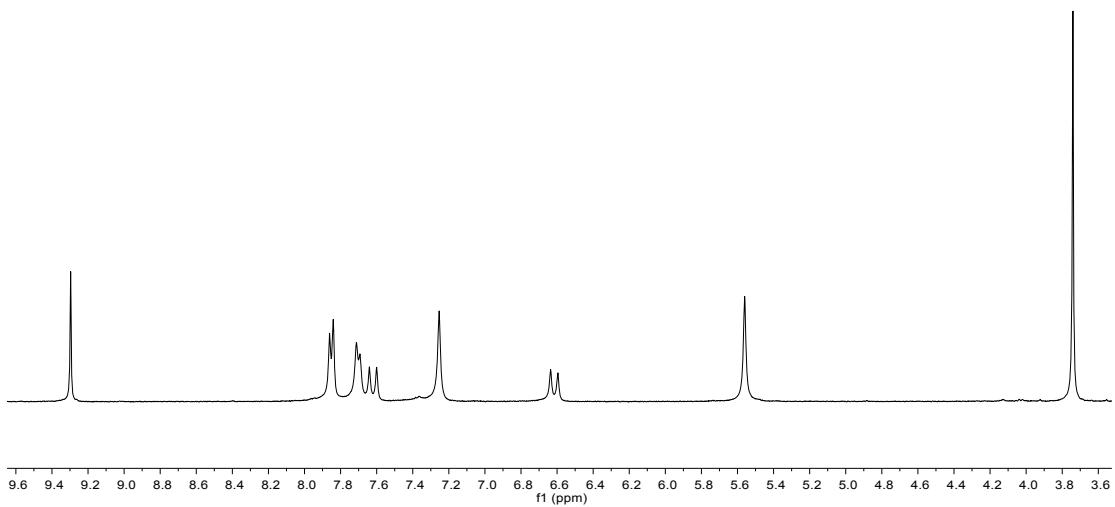


Figure S3. ¹H NMR (400 MHz, DMSO-*d*₆) of $[\text{Ag}_2(\mathbf{1})_2](\text{BF}_4)_2$.

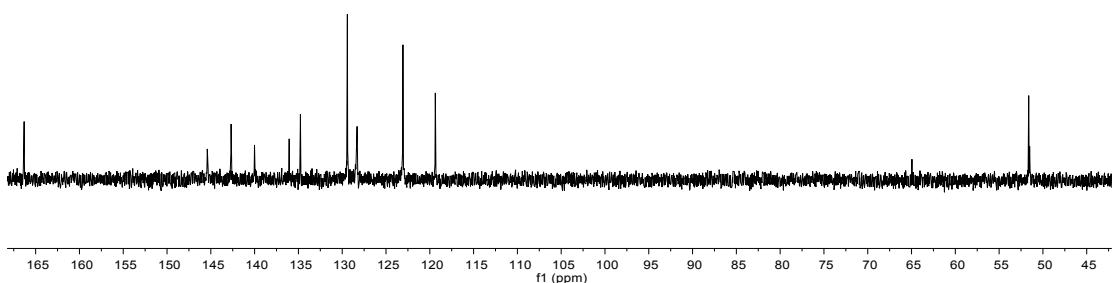


Figure S4. ¹³C NMR (100 MHz, DMSO-*d*₆) of $[\text{Ag}_2(\mathbf{1})_2](\text{BF}_4)_2$.

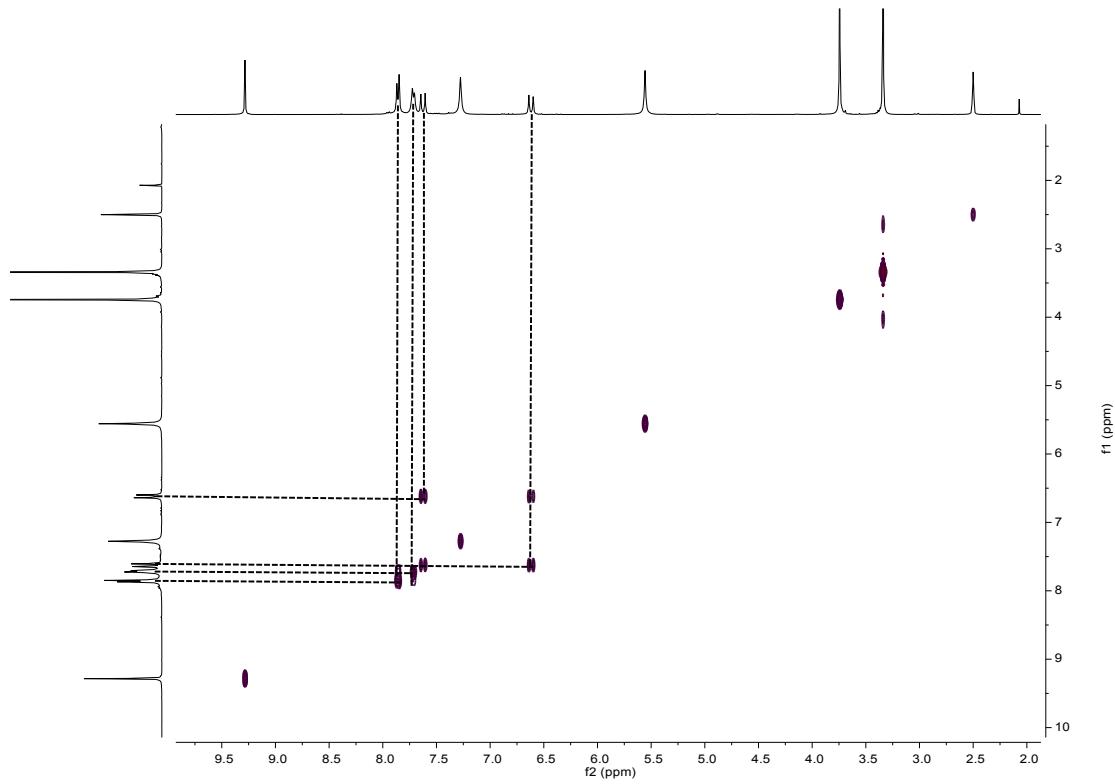


Figure S5. HH-COSY NMR (400 MHz, DMSO-*d*₆) of [Ag₂(1)₂](BF₄)₂.

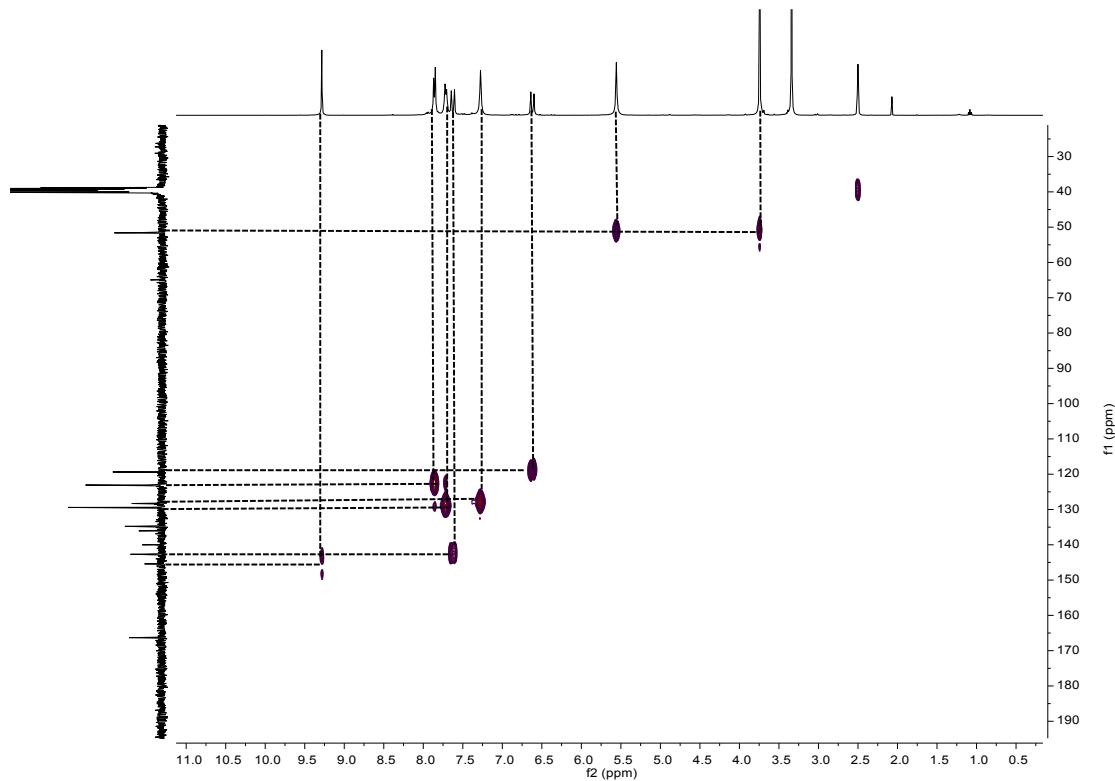


Figure S6. HC-HSQC NMR (400 MHz, DMSO-*d*₆) of [Ag₂(1)₂](BF₄)₂.

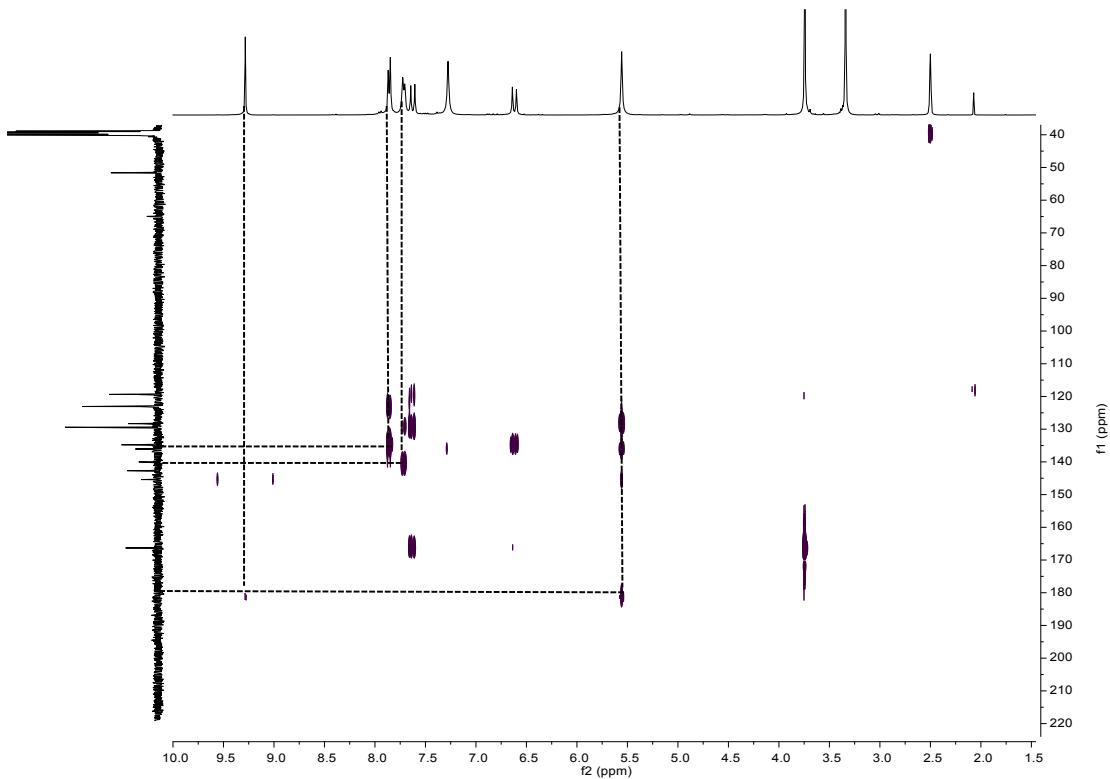


Figure S7. HC-HMBC NMR (400 MHz, $\text{DMSO}-d_6$) of $[\text{Ag}_2(\mathbf{1})_2](\text{BF}_4)_2$.

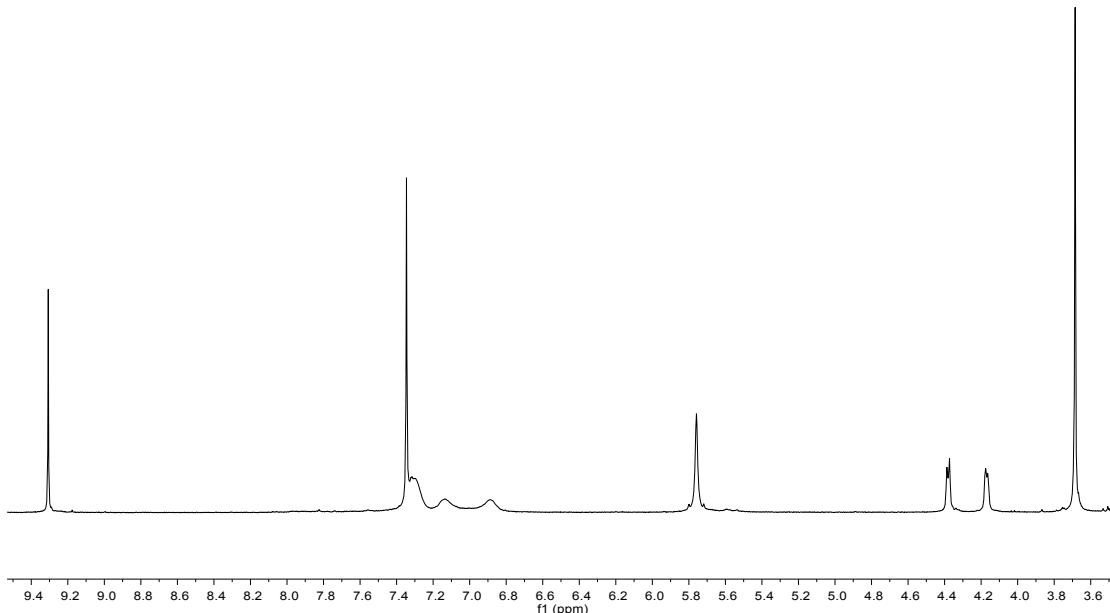


Figure S8. ^1H NMR (400 MHz, $\text{DMSO}-d_6$) of $[\text{Ag}_2(\mathbf{2})_2](\text{BF}_4)_2$.

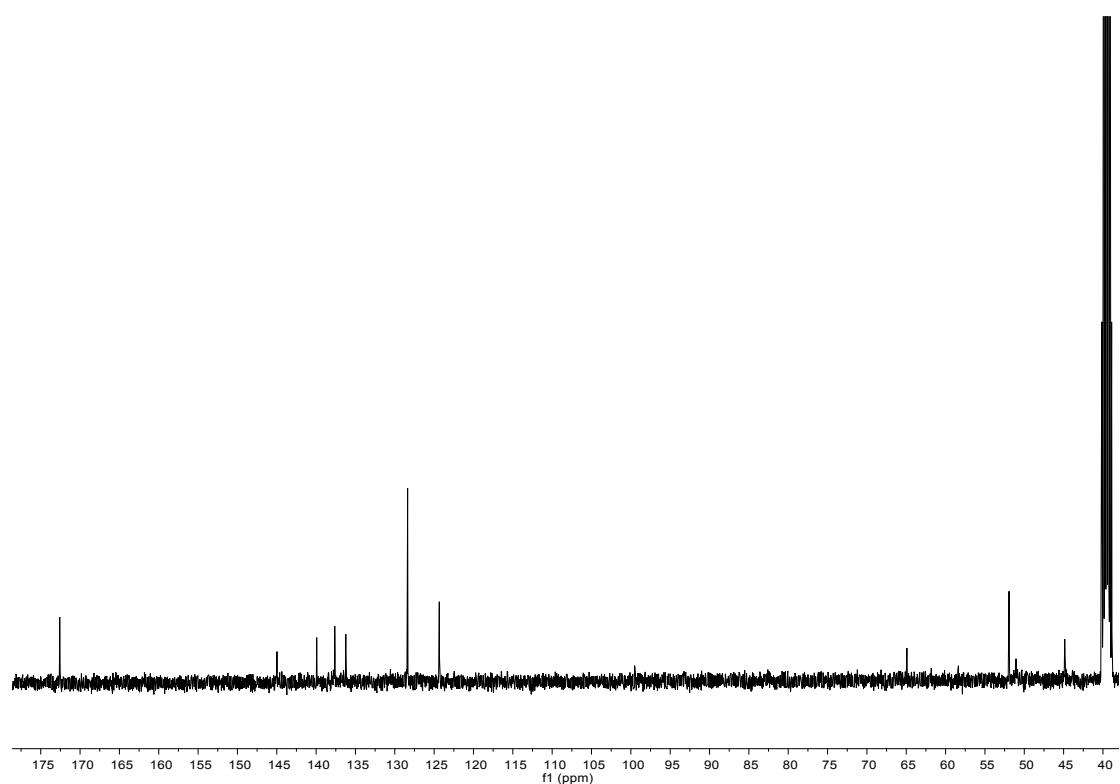


Figure S9. ^{13}C NMR (100 MHz, $\text{DMSO}-d_6$) of $[\text{Ag}_2(\mathbf{2})](\text{BF}_4)_2$.

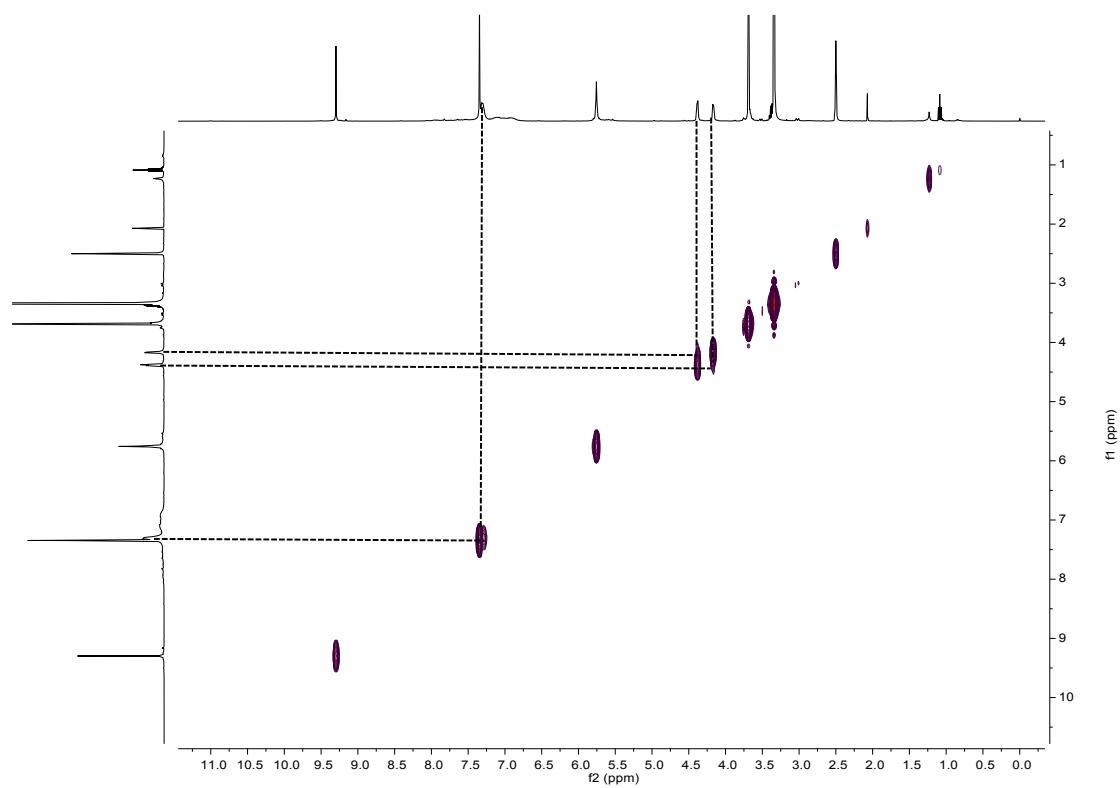


Figure S10. HH-COSY NMR (400 MHz, $\text{DMSO}-d_6$) of $[\text{Ag}_2(\mathbf{2})](\text{BF}_4)_2$.

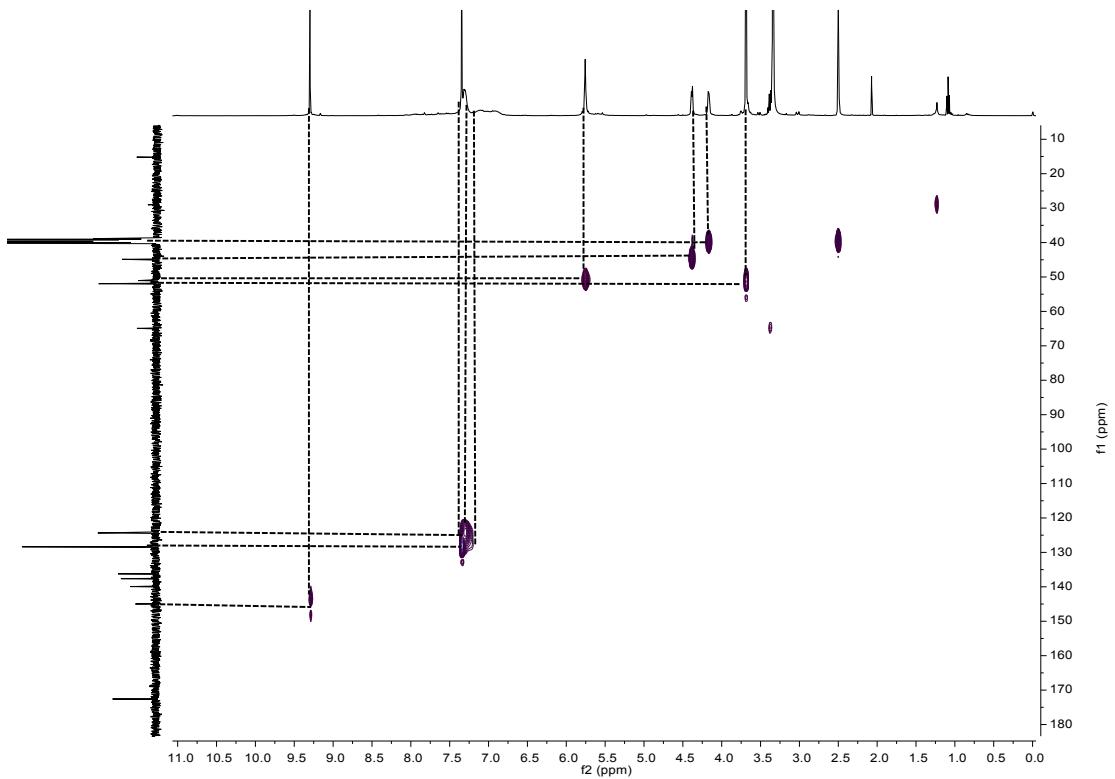


Figure S11. HC-HSQC NMR (400 MHz, DMSO-*d*₆) of [Ag₂(2)](BF₄)₂.

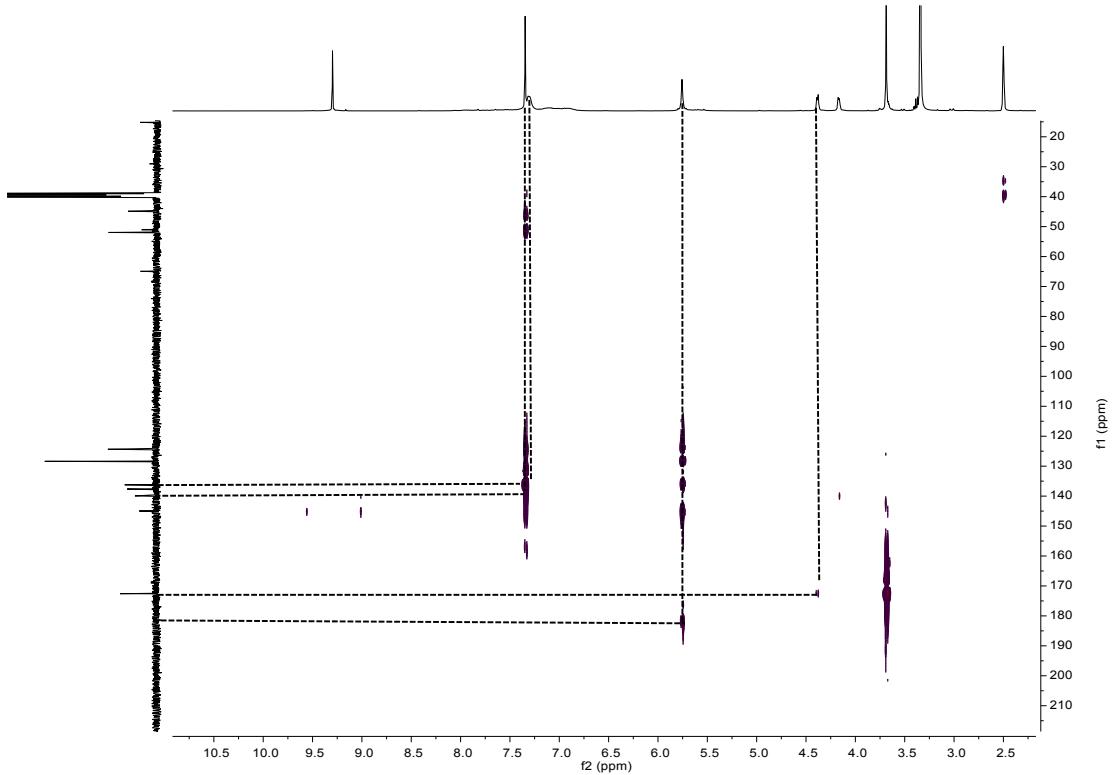


Figure S12. HC-HMBC NMR (400 MHz, DMSO-*d*₆) of [Ag₂(2)](BF₄)₂.

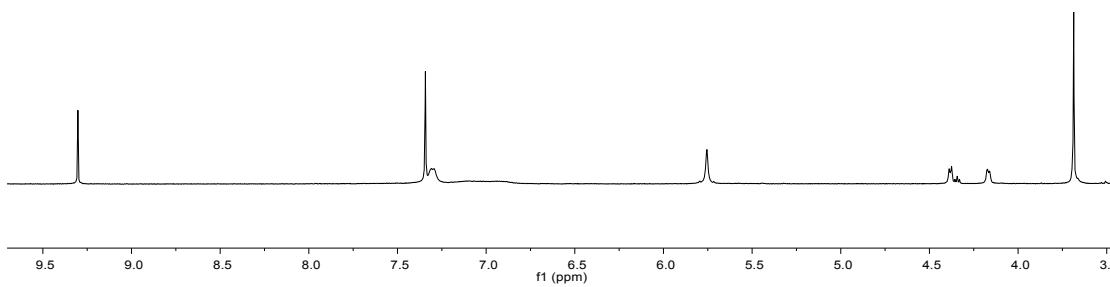


Figure S13. ¹H NMR (400 MHz, DMSO-d₆) of [Ag₂(2)](BF₄)₂ which obtained directly from the reaction of H₄-2(BF₄)₄ and Ag₂O.

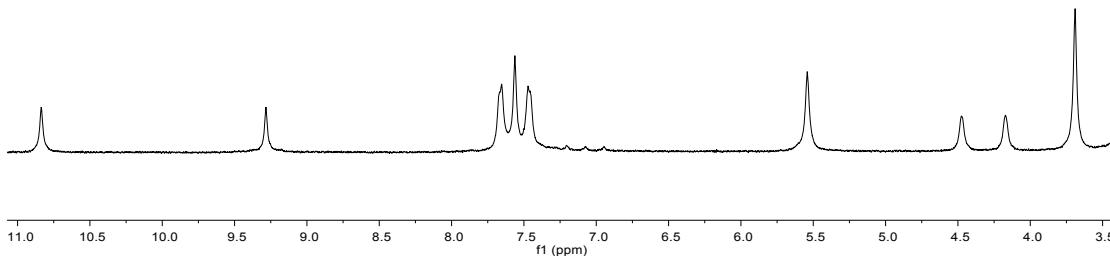


Figure S14. ¹H NMR (400 MHz, DMSO-d₆) of H₄-2(BF₄)₄.

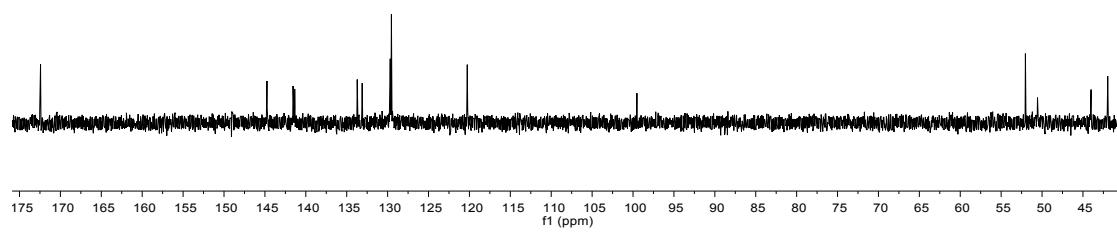


Figure S15. ^{13}C NMR (100 MHz, DMSO- d_6) of $\text{H}_4\text{-2}(\text{BF}_4)_4$.

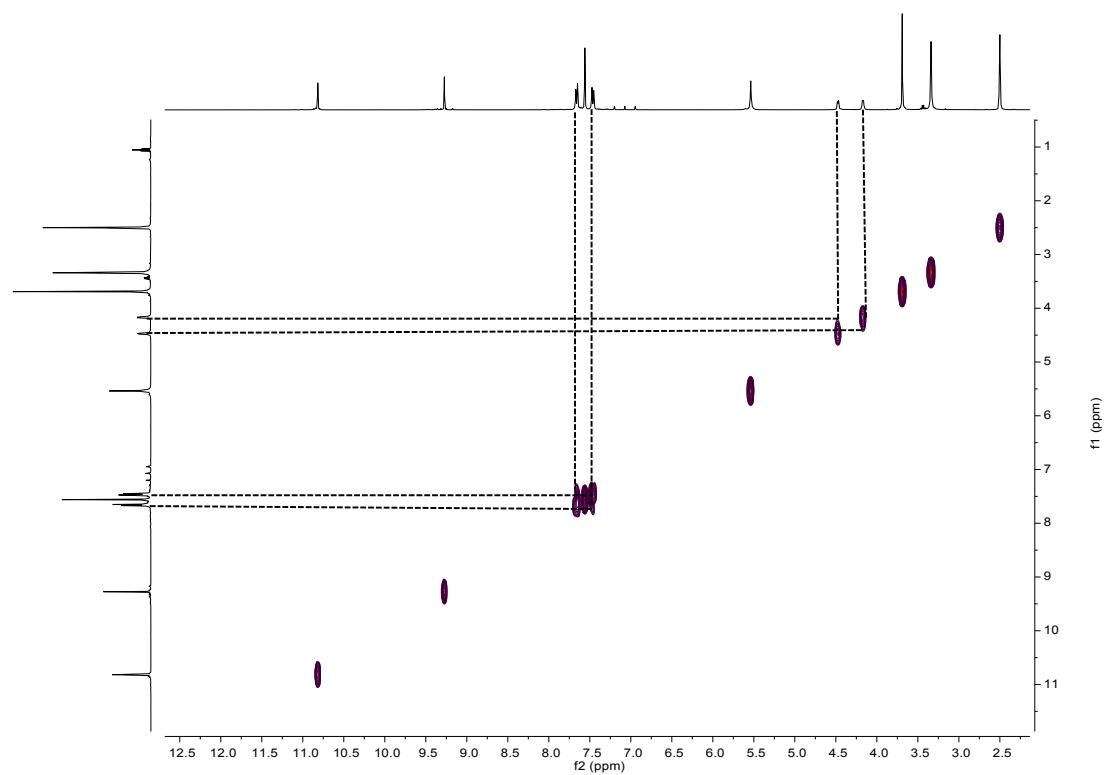


Figure S16. HH-COSY NMR (400 MHz, DMSO- d_6) of $\text{H}_4\text{-2}(\text{BF}_4)_4$.

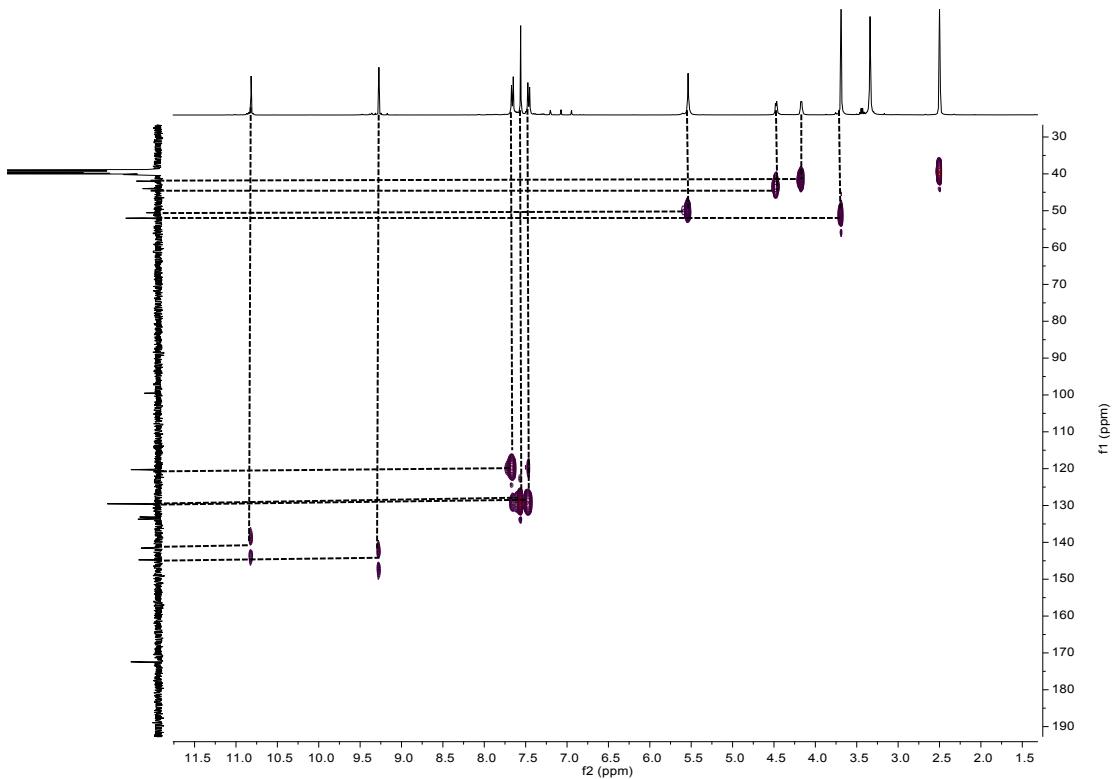


Figure S17. HC-HSQC NMR (400 MHz, DMSO-*d*₆) of H₄-2(BF₄)₄.

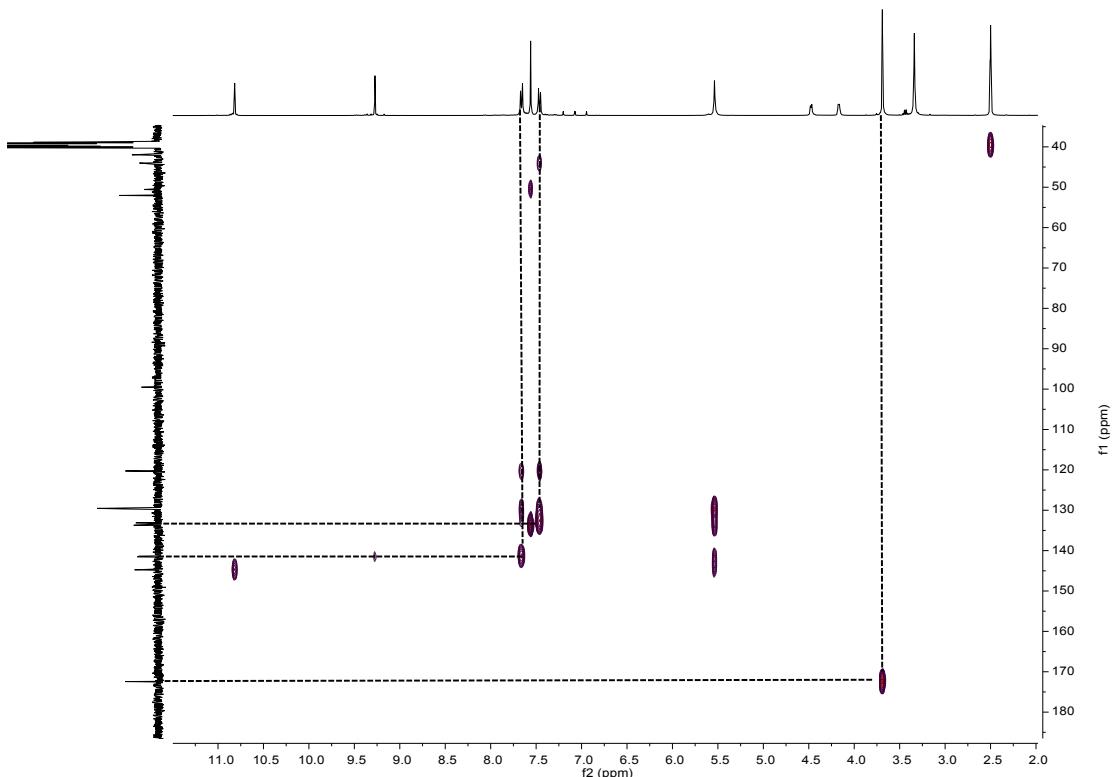
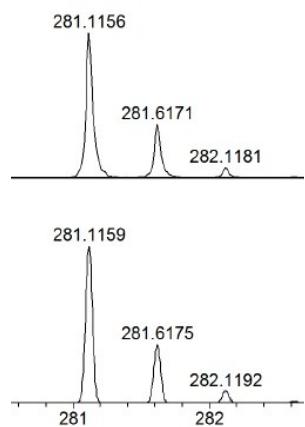
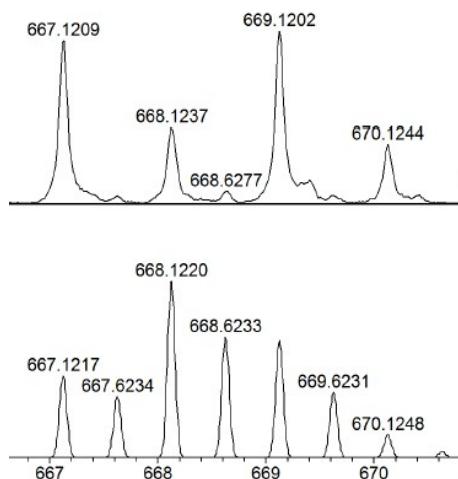


Figure S18. HC-HMBC NMR (400 MHz, DMSO-*d*₆) of H₄-2(BF₄)₄.



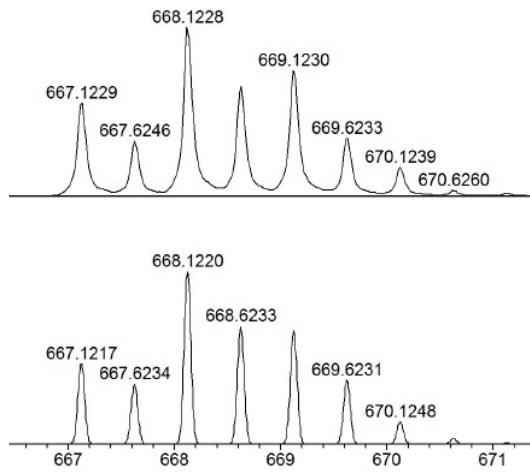
ESI-MS (positive ions) for $\{[H_2-\mathbf{1}]\}^{2+}$: Top (tested) and bottom (Calcd.).

Figure S19. ESI-MS spectra of $H_2\text{-}\mathbf{1}(\text{BF}_4)_2$.



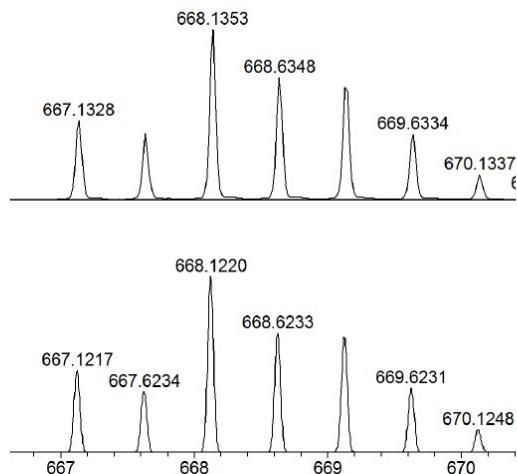
ESI-MS (positive ions) for $\{\text{[Ag}_2(\mathbf{1})_2\}\}^{2+}$: Top (measured) and bottom (Calcd.).

Figure S20. ESI-MS spectra of $[\text{Ag}_2(\mathbf{1})_2](\text{BF}_4)_2$.



ESI-MS (positive ions) for $\{[\text{Ag}_2(\mathbf{2})]\}^{2+}$: Top (measured) and bottom (Calcd.).

Figure S21. ESI-MS spectra of $[\text{Ag}_2(\mathbf{2})](\text{BF}_4)_2$.



ESI-MS (positive ions) for $\{[\text{Ag}_2(\mathbf{2})]\}^{2+}$: Top (measured) and bottom (Calcd.).

Figure S22. ESI-MS spectra of $[\text{Ag}_2(\mathbf{2})](\text{BF}_4)_2$ which obtained directly from the reaction of $\text{H}_4\text{-2}(\text{BF}_4)_4$ and Ag_2O .

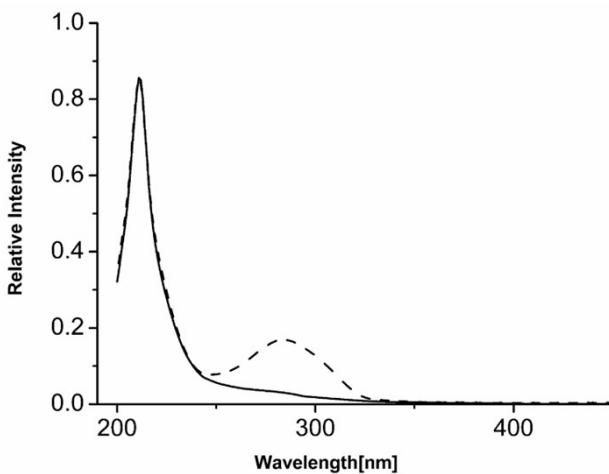


Fig. S23 UV–vis spectra of $[\text{Ag}_2(\mathbf{1})_2](\text{BF}_4)_2$ (5 μM) in CH_3CN before $[\text{Ag}_2(\mathbf{1})_2](\text{BF}_4)_2$ (dash line) and after $[\text{Ag}_2(\mathbf{2})](\text{BF}_4)_2$ (solid line) photoreaction

Table S1. Crystal data of complex $[\text{Ag}_2(\mathbf{1})_2](\text{BF}_4)_2$.

Empirical formula	$\text{C}_{68}\text{H}_{64}\text{Ag}_2\text{B}_2\text{Cl}_8\text{F}_8\text{N}_{12}\text{O}_8$
Formula weight	1850.27
Temperature/K	296(2)
Crystal system	triclinic
Space group	P-1
$a/\text{\AA}$	9.0041(14)
$b/\text{\AA}$	10.5386(16)
$c/\text{\AA}$	20.971(3)
$\alpha/^\circ$	94.863(2)
$\beta/^\circ$	93.285(2)
$\gamma/^\circ$	106.354(2)
Volume/ \AA^3	1895.7(5)
Z	1
$\rho_{\text{calc}}/\text{g/cm}^3$	1.621
μ/mm^{-1}	0.880
F(000)	932.0
Crystal size/ mm^3	$0.130 \times 0.120 \times 0.120$
Radiation	$\text{MoK}_\alpha (\lambda = 0.71073)$
2θ range for data collection/ $^\circ$	3.912 to 50.018
Index ranges	$-10 \leq h \leq 10, -12 \leq k \leq 12, -24 \leq l \leq 24$
Reflections collected	32149
Independent reflections	6594 [$R_{\text{int}} = 0.0332, R_{\text{sigma}} = 0.0265$]
Data/restraints/parameters	6594/0/489
Goodness-of-fit on F^2	1.047
Final R indexes [$I \geq 2\sigma(I)$]	$R_1 = 0.0364, wR_2 = 0.0939$
Final R indexes [all data]	$R_1 = 0.0426, wR_2 = 0.0982$
Largest diff. peak/hole / e \AA^{-3}	1.97/-0.54