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Formation of Bimetallic Metal-Organic Frameworks Nanosheets and

Their Derived Porous Nickel-Cobalt Sulfides for Supercapacitors

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Figure S1. SEM image of Ni-MOF nanospheres.



Figure S2. SEM images of (a, b) Ni/Co-MOF-5 for 2h, and (c, d) Ni/Co-MOF-5 for 4h.



Figure S3. EDS spectrum of the Ni/Co-MOF-5.



Figure S4. TG-DSC curves of Ni/Co-MOF-5.



Figure S5. XRD pattern of Ni-Co-S.



Figure S6. Electrochemical performances of Ni/Co-MOF-x as electrodes for supercapacitors: (a) CV curves at 5 mV s⁻¹; (b) GCD curves at 1 A g⁻¹; (c) Specific capacitance of Ni/Co-MOF-x electrode at different current densities; (d) Nyquist plots of Ni/Co-MOF-x.



Figure S7. Electrochemical performances of Ni/Co-MOF-5 at different reaction time as electrodes for supercapacitors: (a) CV curves at 5 mV s⁻¹; (b) GCD curves at 1 A g⁻¹; (c) Specific capacitance at different current densities; (d) Nyquist plots of Ni/Co-MOF-5.



Figure S8. (a) CV curves of AC electrode at different scan rates; (b) GCD curves of AC electrode at various current densities; (c) The corresponding specific capacitance of AC electrode calculated by the GCD curves.



Figure S9. (a) CV curves of Ni/Co-MOF-5 and AC electrodes performed in a three-electrode cell in 3 M KOH electrolyte at a scan rate of 5 mV s⁻¹; (b) CV curves of Ni/Co-MOF-5 electrode at different potentials (50 mV s⁻¹); (c) CV curves of Ni/Co-MOF-5 electrode at various scan rates; (d) GCD curves at different current density of asymmetric supercapacitor devices.

Electrode materials	Specific capacitance	Test condition	Electrolyte	Refs
	(F g ⁻¹)		solution	
2D Ni/Co-MOF	530.4 F g ⁻¹	0.5 A g ⁻¹	1 M LiOH	S1
Ni-MOF	1021 F g^{-1}	$0.7 \mathrm{~A~g^{-1}}$	3 M KOH	S2
Ni/Co-MOF	1049 F g^{-1}	1 A g^{-1}	3 M KOH	S3
N doped Zn-MOF	285.8 F g ⁻¹	0.1 A g^{-1}	6 M KOH	S4
Ni/Co-MOF	1220.2 F g ⁻¹	1 A g ⁻¹	3 М КОН	This
				work

Table S1. Comparison of the capacities of Ni/Co-MOF-5 with the other recently reported MOFs.

References

S1. H. Xia, J. Zhang, Z. Yang, S. Guo, S. Guo and Q. Xu, Nano-Micro Lett, 2017, 9.

S2. Y. Yan, P. Gu, S. Zheng, M. Zheng, H. Pang and H. Xue, J. Mater. Chem. A, 2016, 4, 19078-19085.

S3. H. Gholipour-Ranjbar, M. Soleimani and H. R. Naderi, New J. Chem, 2016, 40, 9187-9193.

S4. S. Zhong, C. Zhan and D. Cao, Carbon, 2015, 85, 51-59.