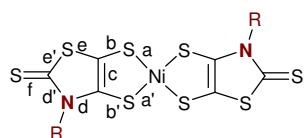


**Neutral, closed-shell nickel bis(2-alkylthio-thiazole-4,5-dithiolate)  
complexes as single component molecular conductors**

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**SUPPLEMENTARY INFORMATION**

**Table S1** Bond distances within complexes  $[C^+][Ni(R\text{-thiazdt})_2]^{2-, 1-, 0}$ .

| Complex                                         | Ni–S<br>(a,a') | S–C<br>(b,b') | C=C (c)   | C–N<br>(d,d') | C–S (e,e') | C=S (f)   | Ref       |
|-------------------------------------------------|----------------|---------------|-----------|---------------|------------|-----------|-----------|
| <i>Neutral complex:</i>                         |                |               |           |               |            |           |           |
| R: Et                                           | 2.158(3)       | 1.686(3)      | 1.388(3)  | 1.378(3)      | 1.733(3)   | 1.640(2)  | 3         |
|                                                 | 2.158(1)       | 1.700(3)      |           | 1.372(3)      | 1.755(3)   |           |           |
| <i>Monoanions:</i>                              |                |               |           |               |            |           |           |
| R: tBu                                          | 2.149(1)       | 1.717(6)      | 1.362(7)  | 1.440(8)      | 1.735(4)   | (a)       | this work |
| C <sup>+</sup> : PPh <sub>4</sub> <sup>+</sup>  | 2.172(3)       | 1.733(5)      |           | 1.398(8)      | 1.731(14)  |           |           |
| R: tBu <sup>(b)</sup>                           | 2.153(1)       | 1.715(4)      | 1.357(5)  | 1.422(4)      | 1.725(4)   | 1.671(4)  | this work |
| C <sup>+</sup> : PPh <sub>4</sub> <sup>+</sup>  | 2.162(1)       | 1.732(4)      |           | 1.367(5)      | 1.732(4)   |           |           |
| R: CH(Me)Ph                                     | 2.161(2)       | 1.708(6)      | 1.357(9)  | 1.399(8)      | 1.747(7)   | 1.645(7)  | 1         |
| C <sup>+</sup> : Et <sub>4</sub> N <sup>+</sup> | 2.155(2)       | 1.709(7)      |           | 1.368(11)     | 1.746(7)   |           |           |
| R: Me                                           | 2.217(5)       | 1.713(6)      | 1.355(6)  | 1.395(5)      | 1.749(5)   | 1.658(3)  | 2         |
| C <sup>+</sup> : PPh <sub>4</sub> <sup>+</sup>  | 2.150(3)       | 1.719(6)      |           | 1.298(4)      | 1.779(3)   |           |           |
| R: Et <sup>(c)</sup>                            | 2.156(1)       | 1.712(3)      | 1.356(3)  | 1.398(3)      | 1.739(2)   | 1.664(3)  | 3         |
| C <sup>+</sup> : PPh <sub>4</sub> <sup>+</sup>  | 2.178(2)       | 1.722(2)      |           | 1.362(4)      | 1.735(3)   |           |           |
| <i>Dianions:</i>                                |                |               |           |               |            |           |           |
| R: Me                                           | 2.199(10)      | 1.735(10)     | 1.345(5)  | 1.405(9)      | 1.75(1)    | 1.678(10) | 2         |
| C <sup>+</sup> : PPh <sub>4</sub> <sup>+</sup>  | 2.195(10)      | 1.732(9)      |           | 1.341(7)      | 1.751(6)   |           |           |
| R: Et                                           | 2.206(13)      | 1.735(11)     | 1.338(14) | 1.409(12)     | 1.751(16)  | 1.692(15) | 3         |
| C <sup>+</sup> : PPh <sub>4</sub> <sup>+</sup>  | 2.198(17)      | 1.741(12)     |           | 1.353(8)      | 1.732(18)  |           |           |

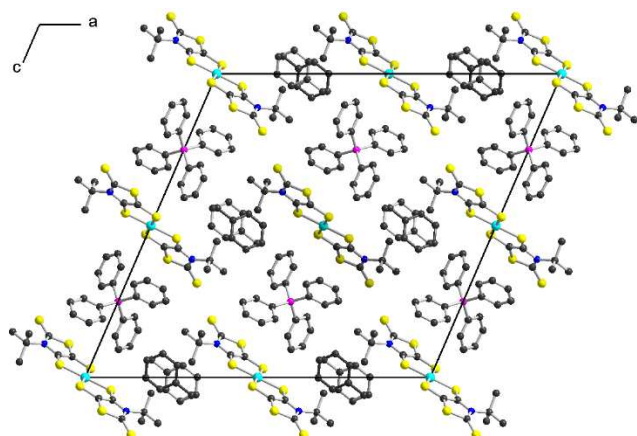
<sup>(a)</sup> The C=S group is disordered. <sup>(b)</sup> Toluene solvate. <sup>(c)</sup> Acetonitrile solvate

## References

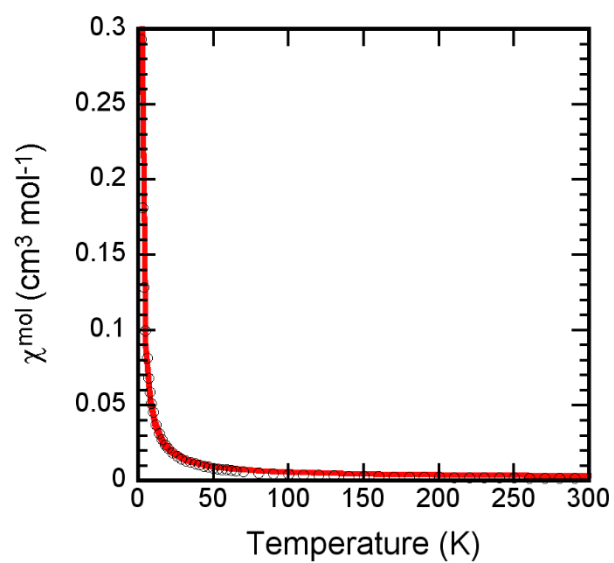
- <sup>1</sup> Y. Le Gal, A. Vacher, V. Dorcet, M. Fourmigué, J. Crassous and D. Lorcy, *New J. Chem.*, 2015, **39**, 122–129.
- <sup>2</sup> S. Eid, M. Fourmigué, T. Roisnel and D. Lorcy, *Inorg. Chem.*, 2007, **46**, 10647–10654.
- <sup>3</sup> A. Filatre-Furcate, N. Bellec, O. Jeannin, P. Auban-Senzier, M. Fourmigué, A. Vacher and D. Lorcy, *Inorg. Chem.*, 2014, **53**, 8681–8690.

**Table S2** Calculated  $\beta$  interaction energies between frontier orbitals within nearest neighbors in crystalline  $[\text{Ni}(\text{RS-tzdt})_2]$  (R = Me, Et) complexes. See Figure

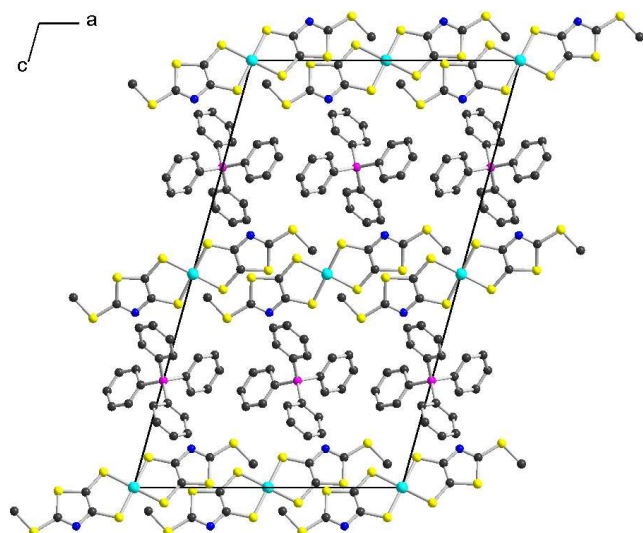
|                                  | $\beta_{\text{HOMO-HOMO}}$ | $\beta_{\text{HOMO-LUMO}}$ | $\beta_{\text{LUMO-HOMO}}$ | $\beta_{\text{LUMO-LUMO}}$ |
|----------------------------------|----------------------------|----------------------------|----------------------------|----------------------------|
| $[\text{Ni}(\text{MeS-tzdt})_2]$ |                            |                            |                            |                            |
| Interaction I                    | −0.1709                    | −0.1131                    | +0.1131                    | +0.0345                    |
| Interaction II                   | −0.0397                    | −0.0343                    | +0.0343                    | +0.0268                    |
| Interaction III                  | −0.1293                    | −0.0050                    | −0.0050                    | +0.0235                    |
| $[\text{Ni}(\text{EtS-tzdt})_2]$ |                            |                            |                            |                            |
| Interaction I                    | −0.1110                    | +0.3583                    | −0.3583                    | +0.2214                    |
| Interaction II                   | −0.0846                    | +0.0378                    | +0.0378                    | −0.0265                    |
| Interaction III                  | −0.1369                    | −0.0140                    | +0.0140                    | +0.0168                    |



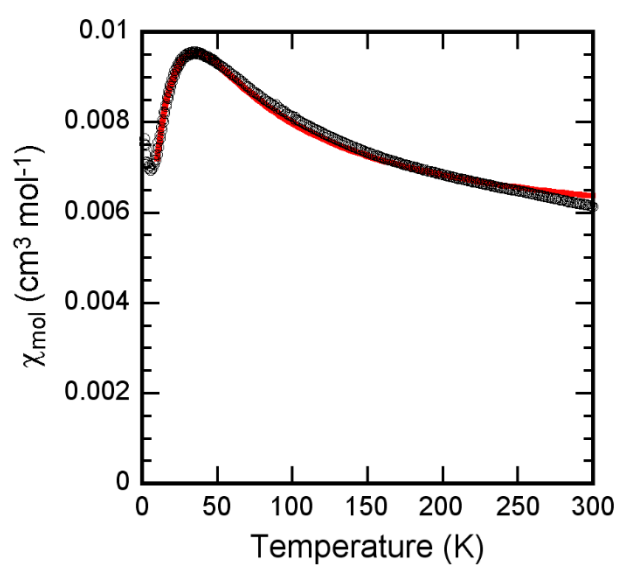
**Fig. S1** Projection view along  $b$  of the unit cell of  $[\text{PPh}_4][\text{Ni}(\text{tBu-thiazdt})_2] \cdot \text{toluene}$ . Hydrogen atoms have been omitted for clarity. Bonds within the inversion centered disordered toluene molecules are highlighted in light and dark grey.



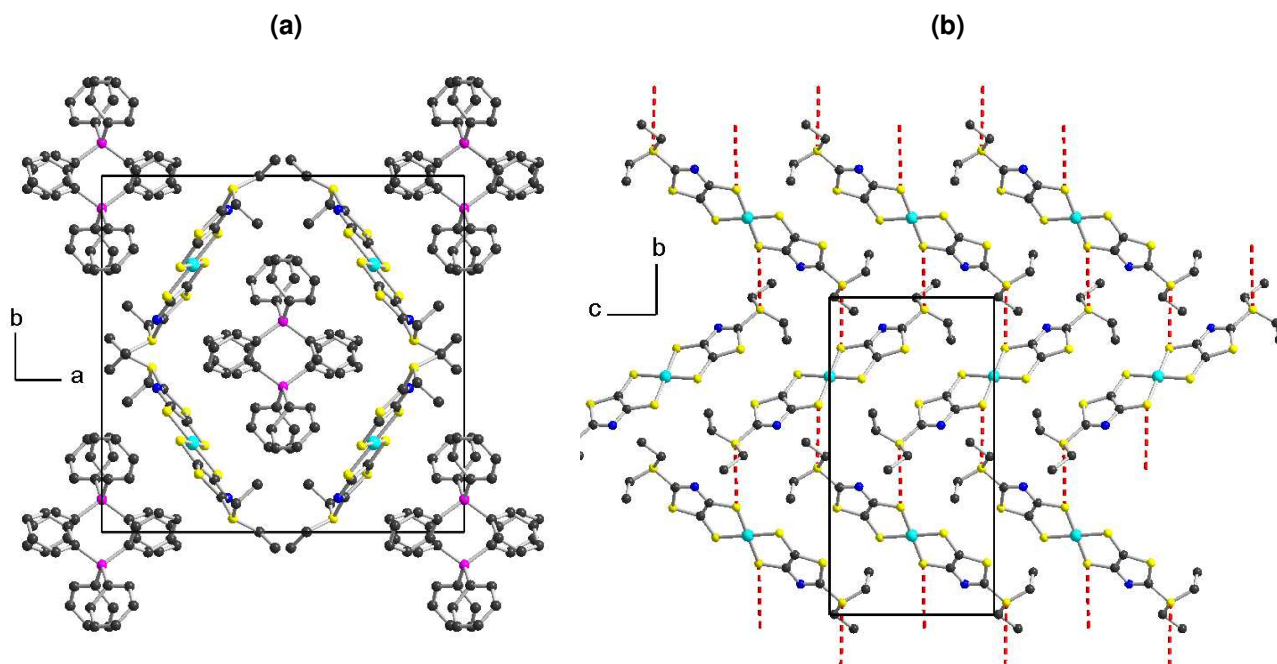
**Fig. S2** Temperature dependence of the magnetic susceptibility of  $[\text{PPh}_4][\text{Ni}(\text{tBu-thiazdt})_2]$ , with the Curie Weiss fit (see text).



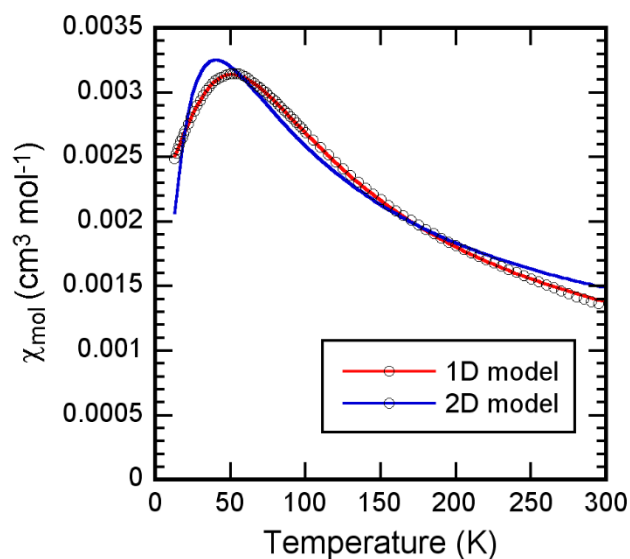
**Fig. S3** Projection view along  $b$  of the unit cell of  $[\text{PPh}_4][\text{Ni}(\text{MeS-tzdt})_2]$ . Hydrogen atoms have been omitted for clarity.



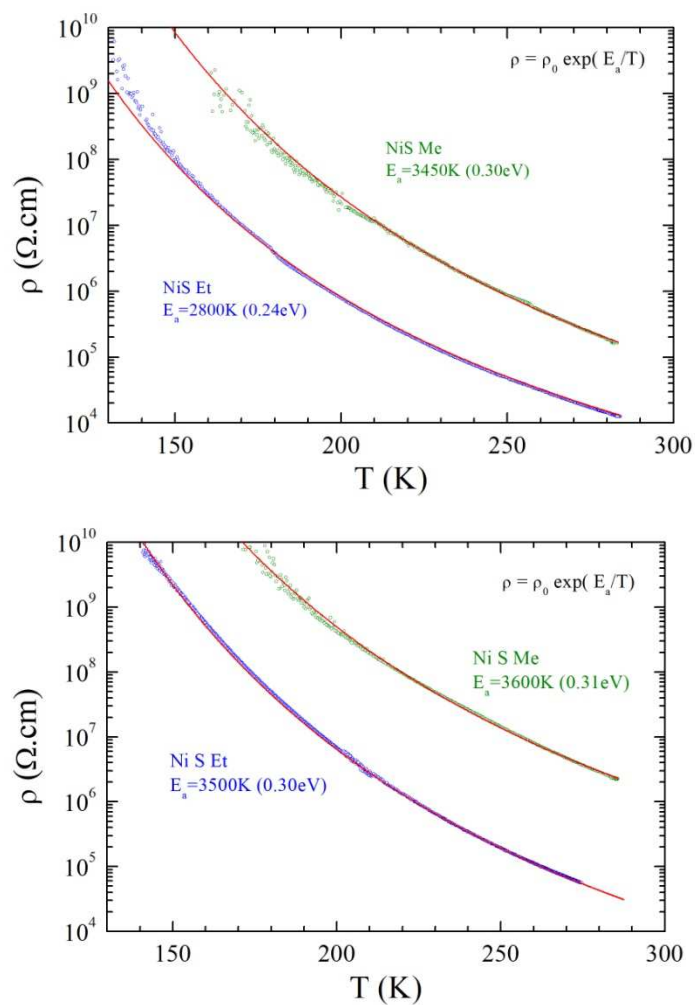
**Fig S4** Temperature dependence of the magnetic susceptibility of  $[\text{PPh}_4][\text{Ni}(\text{MeS-tzdt})_2]$ , with the Bonner-Fisher fit (red curve) for a uniform  $S = \frac{1}{2}$  spin chain. Note a large temperature independent paramagnetism (TIP)  $\chi_0 = 5.3 \times 10^{-3} \text{ cm}^3 \text{ mol}^{-1}$  (see text).



**Fig. S5** (a) Projection view along  $c$  of the unit cell of [PPh<sub>4</sub>][Ni(EtS-tzdt)<sub>2</sub>]. Only one position of the disordered PPh<sub>4</sub><sup>+</sup> cation is shown. (b) Detail of the solid state organization of the radical anions within (bc) planes. The red dotted lines are associated with a short S...S intermolecular contact at 3.823 Å. Hydrogen atoms have been omitted for clarity.



**Fig S6** Temperature dependence of the magnetic susceptibility of [PPh<sub>4</sub>][Ni(EtS-tzdt)<sub>2</sub>], with two different fits, the 1D Bonner-Fisher fit (red curve) for a uniform  $S = 1/2$  spin chain, and the 2D  $S = 1/2$  spin square lattice. The temperature independent paramagnetism (TIP) with the 1D model amounts to  $\chi_0 = 3.1 \times 10^{-4}$  cm<sup>3</sup> mol<sup>-1</sup> (see text).



**Fig S7** Temperature dependence of the resistivity of  $[\text{Ni}(\text{EtS-tzdt})_2]$  and  $[\text{Ni}(\text{MeS-tzdt})_2]$ . The red lines are the Arrhenius fits to the data giving the activation energies. Measurements performed without gold pads (top) and with gold pads (bottom).