

Supporting information

Bi₂Mn₄O₁₀: A New Mullite-type Anode Material for Lithium Ion Batteries

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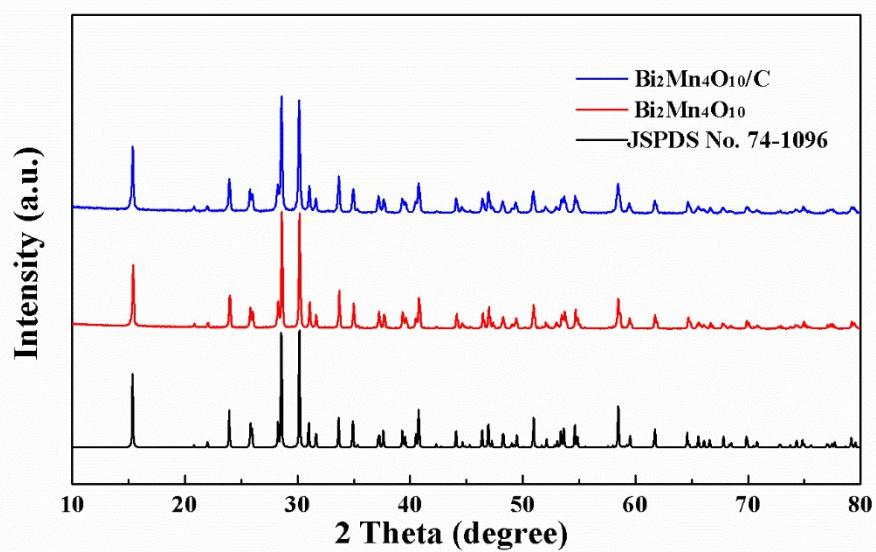


Fig. S1 X-ray diffraction patterns of $\text{Bi}_2\text{Mn}_4\text{O}_{10}$ and $\text{Bi}_2\text{Mn}_4\text{O}_{10}/\text{C}$

Table S1 Rietveld refinement parameters and results obtained for Bi₂Mn₄O₁₀

Atom	Wyck.	x/a	y/b	z/c	U ⁰ (Å ²)
Bi1	4g	0.15932(7)	0.16557(8)	0	0.0136(2)
Mn1	4f	0.5	0	0.2608(4)	0.0128(6)
Mn2	4h	0.40670(23)	0.35065(28)	0.5	0.0078(6)
O1	4e	0	0	0.3017(11)	0.0140(26)
O2	8i	0.3876(7)	0.1998(6)	0.2677(7)	0.0220(17)
O3	4h	0.1431(10)	0.4225(7)	0.5	0.0096(21)
O4	4g	0.1592(11)	0.4466(8)	0	0.0216(26)

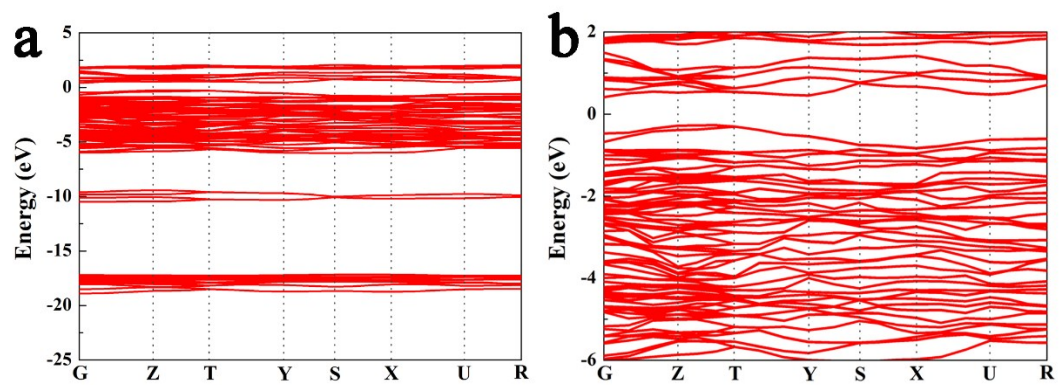


Fig. S2 (a) and (b) Band structure of $\text{Bi}_2\text{Mn}_4\text{O}_{10}$. The band gap is 0.7082 eV.

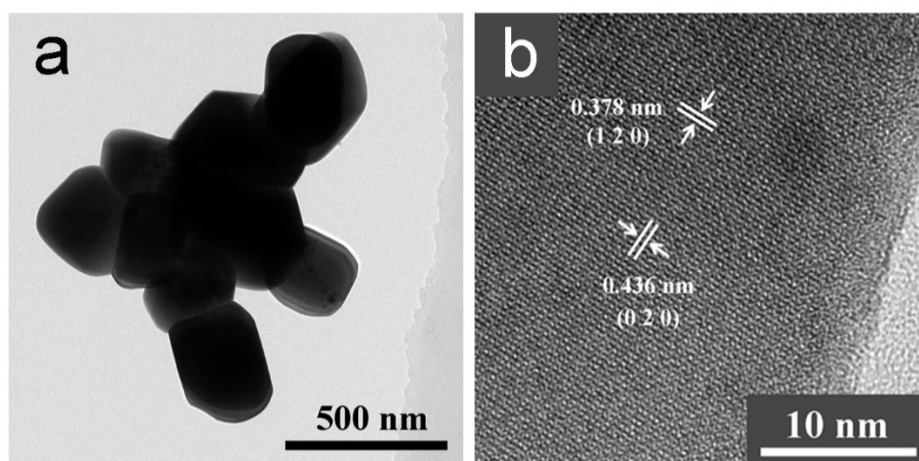


Fig. S3 TEM and HRTEM images of bare BMO.

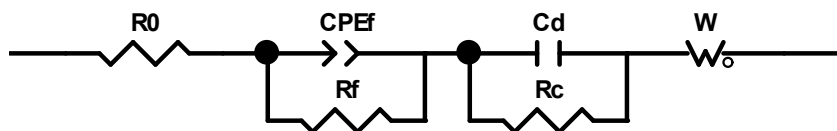


Fig. S4 A proposed equivalent circuit for describing the EIS behavior. The circuit elements are R_0 (electrolyte resistance), CPE_f (constant phase element represents the surface film capacitance) and R_f (surface film resistance), C_d and R_c (double layer capacitance and charge transfer resistance, respectively, in the electrode – electrolyte interface), and W (the Warburg element, which is characterizes diffusion of lithium ions).

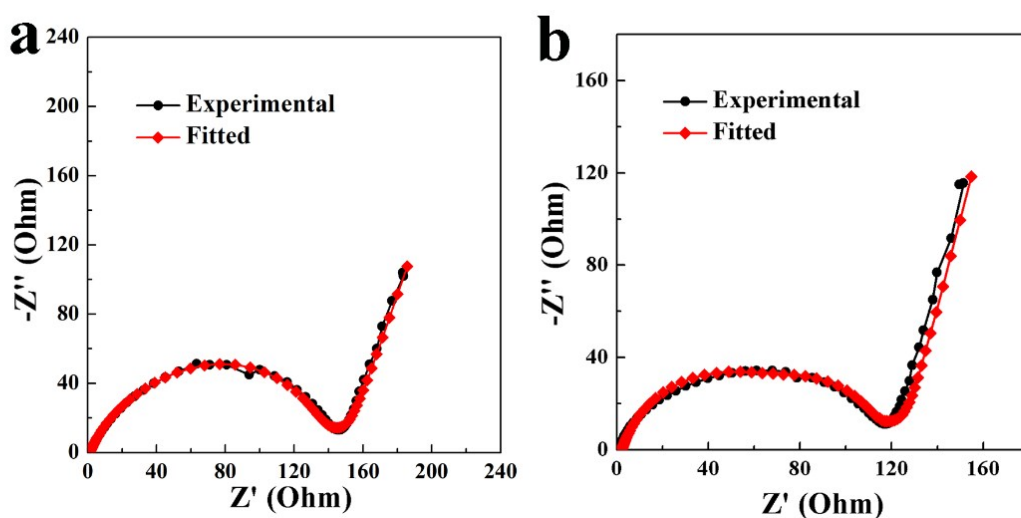


Fig. S5 Experimental and fitted Nyquist plots of bare $\text{Bi}_2\text{Mn}_4\text{O}_{10}$ (a) and $\text{Bi}_2\text{Mn}_4\text{O}_{10}/\text{C}$ composite (b) anodes

Table S2 The fitted impedance parameters of Bi₂Mn₄O₁₀ and Bi₂Mn₄O₁₀/C anodes

Bi ₂ Mn ₄ O ₁₀			Bi ₂ Mn ₄ O ₁₀ /C		
Element	Value	Error%	Element	value	Error%
R ₀	1.286	~3.245%	R ₀	1.084	~2.568%
CPE _f -T	4.1024×10 ⁻⁵	~5.938%	CPE _f -T	1.4048×10 ⁻⁵	~6.512%
CPE _f -P	0.7492	~0.788%	CPE _f -P	0.8286	~0.577%
R _f	83.46	~2.408%	R _f	72.79	~2.891%
C _d	2.6735×10 ⁻⁵	~6.749%	C _d	6.11×10 ⁻⁵	~6.497%
R _c	48.02	~4.523%	R _c	32.84	~3.807%
W-R	51.54	~4.889%	W-R	51.29	~4.081%
W-T	0.6182	~6.754%	W-T	0.5973	~5.445%
W-P	0.3957	~0.716%	W-P	0.4204	~0.795%

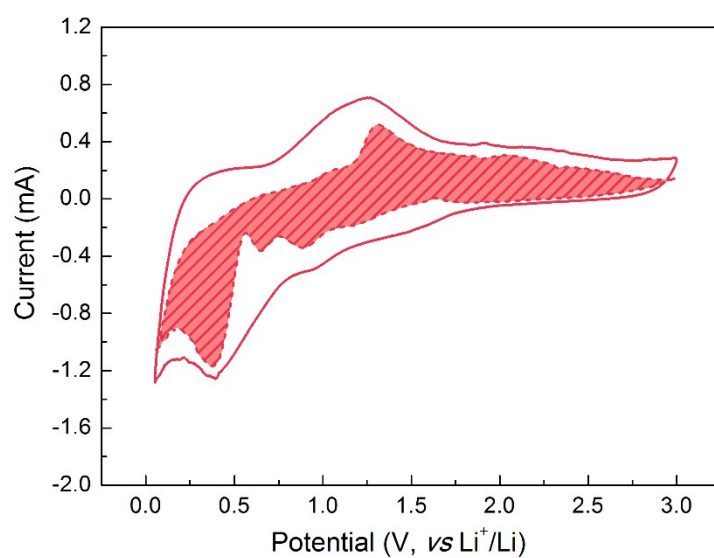


Fig. S6 Current response with voltage at the scan rate of 0.5 mV s^{-1} . The solid line is the current signal obtained from the cyclic voltammetry experiment. The Shaded region with dash line is calculated capacitive current using $i(V) = k_f v$.

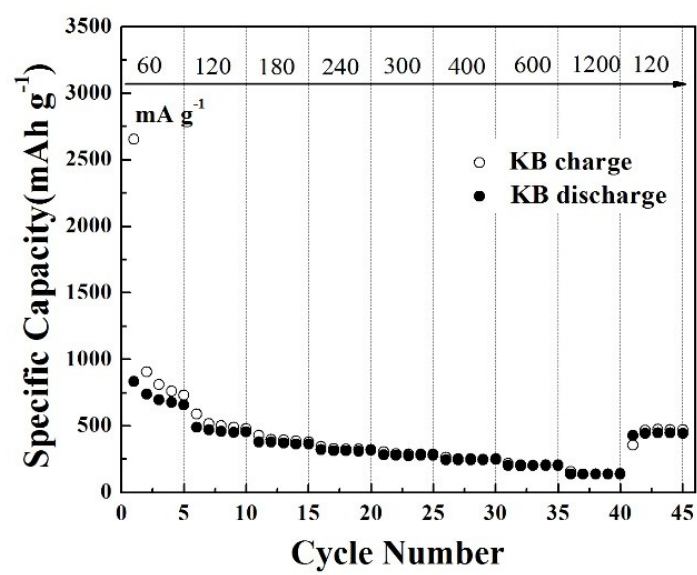


Fig. S7 The rate performance of conductive carbon (KB).

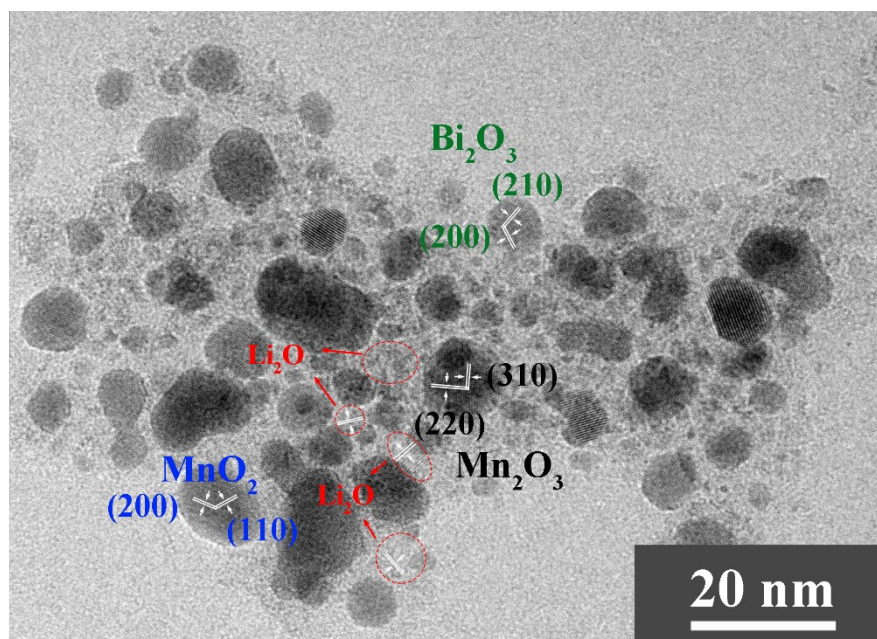


Fig. S8 Ex-situ TEM image of the charged electrode at 3.0 V after 100 discharge-charge cycles. The metal oxide nanoparticles are surrounded by Li_2O buffering matrix.

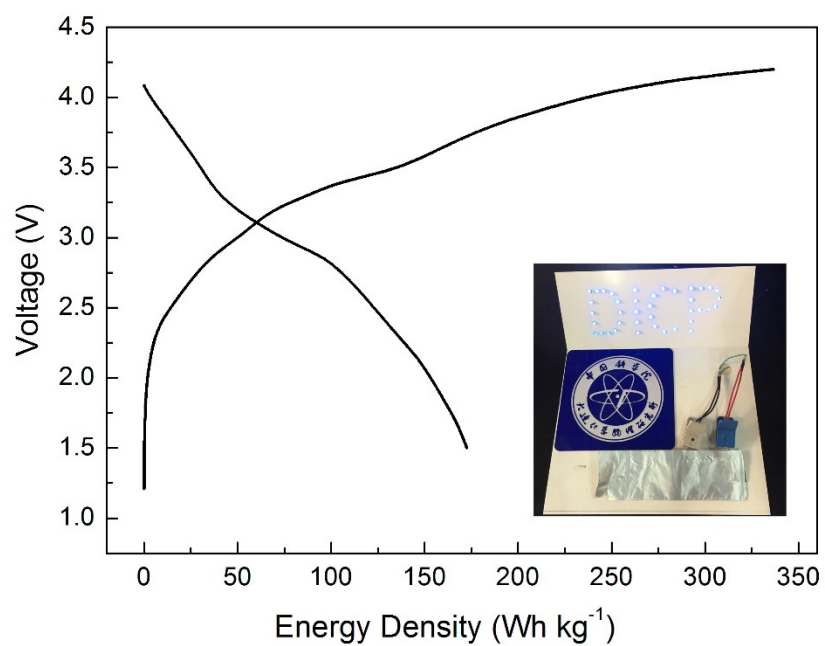


Fig. S9 The charge/discharge performance of $\text{LiMn}_2\text{O}_4|\text{Bi}_2\text{Mn}_4\text{O}_{10}/\text{C}$ full battery at 0.2 C rate. The mass loading of the LiMn_2O_4 in cathode and $\text{Bi}_2\text{Mn}_4\text{O}_{10}/\text{C}$ in anode are $\sim 4.0 \text{ mg cm}^{-2}$ and $\sim 0.4 \text{ mg cm}^{-2}$, respectively.

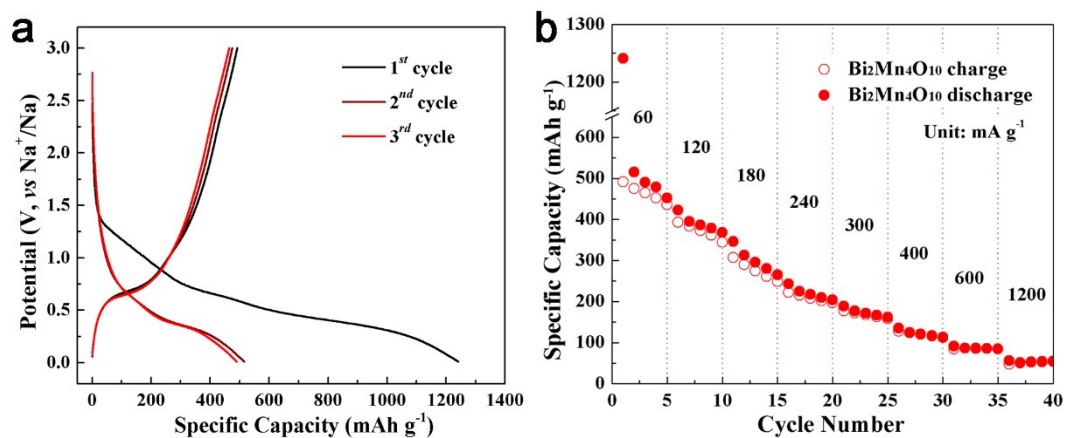


Fig. S10 Discharge-charge curves at 60 mA g⁻¹ (a) and the rate performance (b) of BMO/C anode for sodium-ion battery (SIB).