Supplementary information for

Facile synthesis of single atom-dispersed silver-modified ultrathin $g-C_3N_4$ hybrid for the enhanced visible-light photocatalytic degradation of sulfamethazine with peroxymonosulfate

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TEXTS

Text S1. Determination of concentration of SMT

The residual concentration of SMT was analyzed on a Shimadzu LC 20A high performance liquid chromatography (HPLC, Japan) equipped with a SPD-M20A diode array detector. Analytical separation was performed by a C18 reversed-phase column (Zorbax Eclipse, 4.6×250 mm, 5 µm). The isocratic elution consisted of 20 % acetonitrile and 80 % water with a flow rate of 1 mL·min⁻¹. The detection wavelength was set as 266 nm.

Text S2. Identification of photocatalytic by-products

Samples for SMT degradation products characterization were concentrated by an HLB cartridge (6 mL, 500 mg, Waters Oasis) through solid phase extraction (SPE) process. The byproducts of SMT was carried out on an Agilent 1100 series HPLC coupled to a 6410 triple quadrupole mass spectrometer (Agilent Technologies, USA). Separation was accomplished using an Agilent C18 column (5 μ m, 4.6 x 150 mm). Elution was performed at a flow rate of 1 mL min-1 with water as eluent A and acetonitrile containing as eluent B, employing a linear gradient from 10% B to 60% B in 0 – 20 min, 60% B to 100% B in the next 2 min. Mass spectral analysis was conducted in positive mode using an electrospray ionization (ESI) source with the following ion source parameters: scan range of m/z = 100-600; fragmentor and capillary voltage were 3.5 kV and 125 V; Nebulizer pressure was 40 psi. Once a potential product was identified, product ion scan MS/MS was performed for structure elucidation.

Parameter	T T '4	Pearl River	WWTP effluent	South China
	Unit	water	water	Sea water
pН	-	7.08	6.93	8.17
UV ₂₅₄	-	0.049	0.0588	0.0103
UV ₄₀₀	-	0.0052	0.0078	0.0022
TOC	mg/l	2.796	4.309	0.2266
Na ^{+a}	ppm	15.65	36.91	10100
K ^{+ a}	ppm	4.4	10.59	374
Cu^{2+a}	ppb	1.04	/	44.07
Mg ^{2+ a}	ppm	3.84	6.38	1340
Al ^{3+ a}	ppb	6.11	8.29	3.12
Cl ^{-b}	ppm	15.61	51.35	18451.87
HCO ₃ -	ppm	4.87	2.77	5.46
SO4 ^{2-b}	ppm	27.25	40.26	2541.96

 Table S1. Primary properties of water samples.

^a Detected by ICP-MS.

^b Detected by anions-ion chromatography.

Numb er	Retention Time/min	[M+H] ⁺	Fragment Peaks	Supposed Structure
1	8.5	124	107, 80, 67	$H_2N \rightarrow H_3$ $H_2N \rightarrow H_3$
2	10.9	215	198, 173, 158, 93	
3	14.5	279	204, 174, 156, 124	$H_2N \xrightarrow{Q} H \xrightarrow{N} H \xrightarrow{V} H_3$
4	18.1	295	215, 186, 124, 108	
5	19.0	309	263, 245, 186, 175, 123	$ \overset{O}{} \overset{H}{} H$

Table S2. LC-MS/MS mass spectrometry pieces information and proposed structure

 of photocatalytic products of SMT.

FIGURES



Fig. S1. FT-IR spectrum of AgTCM/UCN.



Fig. S2. High-resolution XPS spectra of A) C 1s and B) N 1s of AgTCM/UCN.



Fig. S3. Kinetic rate constant of SMT over different A) PMS amounts and B) Ag contents.